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PDGFB-targeted functional MRI nanoswitch for activatable T₁–T₂ dual-modal ultra-sensitive diagnosis of cancer

Ya'nan Zhang^{1,3}, Lu Liu^{1,2}, Wenling Li^{1,2}, Caiyun Zhang^{1,2}, Tianwei Song^{1,2}, Peng Wang^{1,2}, Daxi Sun^{1,2}, Xiaodan Huang^{1,2}, Xia Qin^{1,2}, Lang Ran^{1,2}, Geng Tian^{1*}, Junchao Qian^{3,4*} and Guilong Zhang^{1,2*}

Abstract

*Correspondence:

tiangeng@bzmc.edu.cn

Geng Tian

As one of the most significant imaging modalities currently available, magnetic resonance imaging (MRI) has been extensively utilized for clinically accurate cancer diagnosis. However, low signal-to-noise ratio (SNR) and low specificity for tumors continue to pose significant challenges. Inspired by the distance-dependent magnetic resonance tuning (MRET) phenomenon, the tumor microenvironment (TME)-activated off–on T_1-T_2 dual-mode MRI nanoswitch is presented in the current study to realize the sensitive early diagnosis of tumors. The tumor-specific nanoswitch is designed and manufactured on the basis of PDGFB-conjugating ferroferric oxide coated by Mn-doped silica (PDGFB-FMS), which can be degraded under the high-concentration GSH and low pH in TME to activate the T_1-T_2 dual-mode MRI signals. The tumor-specific off–on dual-mode MRI nanoswitch can significantly improve the SNR and is used successfully for the accurate diagnosis of early-stage tumors, particularly for orthotopic prostate cancer. In addition, the systemic delivery of the nanoswitch did not cause blood or tissue damage, and it can be excreted out of the body in a timely manner, demonstrating excellent biosafety. Overall, the strategy is a significant step in the direction of designing off–on dual-mode MRI nanoprobes to improve imaging accuracy, which opens up new avenues for the development of new MRI probes.

Keywords Magnetic resonance imaging, TME-activated nanoswitch, Dual-mode contrast agent, Tumor diagnosis, SNR

Junchao Qian qianjunchao@hmfl.ac.cn Guilong Zhang glzhang@bzmc.edu.cn ¹ School of Medical Imaging, Shandong Technology Innovation Center of Molecular Targeting and Intelligent Diagnosis and Treatment, Binzhou Medical University, Yantai 264003, People's Republic of China ² School of Pharmacy, Institute of Aging Medicine, Binzhou Medical University, Yantai 264003, People's Republic of China ³ Hefei Cancer Hospital, Anhui Province Key Laboratory of Medical Physics and Technology, Institute of Health and Medical Technology, Hefei Institutes of Physical Science, Chinese Academy of Sciences, Hefei 230031, People's Republic of China

⁴ Department of Radiation Oncology, School of Medicine, Shandong University, Shandong Cancer Hospital and Institute, Shandong First



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Medical University and Shandong Academy of Medical Sciences, Jinan 250117, Shandong, China



Introduction

In situ accurate and early diagnosis of cancer is essential for selecting appropriate treatment options and determining prognosis without invasive procedures such as tissue biopsies or surgical excision [1-3]. In recent years, clinical molecular imaging devices, such as X-ray computed tomography (CT), optical imaging, positron emission tomography (PET), and single-photon emission computed tomography (SPECT) techniques, have been utilized extensively to analyze and diagnose tumors in vivo [4, 5]. However, due to the superior soft-tissue contrast, deep tissue penetration, high spatial resolution, and lack of ionizing radiation exposure, etc., magnetic resonance imaging (MRI) can provide anatomical and functional information in diagnosing clinical diseases; consequently, it has demonstrated broader application prospects in the diagnosis of tumors [6-8]. Accordingly, contrast agents, such as T_1 or T_2 contrast nanomaterials, have been developed for MRI over the past few decades [9–11]. However, single-mode MR contrast agents (CAs) have inherent drawbacks, such as a short blood circulation time, magnetic susceptibility artifacts, and a clinical accuracy of tumor diagnosis that is not always met [12– 14]. Therefore, it is of the utmost importance to design versatile contrast nanoprobes capable of overcoming limitations to achieve accurate diagnostic results.

The concurrent use of T₁-T₂ dual-mode imaging modalities has been reported for the cross-validation of acquired image data [15, 16]. Accordingly, this strategy combines the advantages of two single-mode imaging techniques to provide extremely precise image information. Based on the distance-dependent magnetic resonance tuning (MRET) phenomenon, however, when the enhancer is located close to the quencher, the electron spin fluctuation rate of the enhancer decreases, preventing it from accelerating and relaxing water protons, thereby resulting in a quenched longitudinal relaxivity (r_1) [17–19]. Additionally, Due to direct contact between the two agents, the magnetic field derived from the superparamagnetic T₂ CA disrupts the relaxation process of the paramagnetic T_1 CA, resulting in the attenuation of the T_1 signal and a significant decline in the T_1-T_2 contrast enhancement [20]. Conversely, "always on" MRI contrast agents frequently disregard the specific interaction with tumor tissues exhibiting weak acidity and high GSH concentration, resulting in a low signal-to-noise ratio (SNR) [21-23]. Thus, the development of a smart stimuli-activated MRI off-on nanoswitch [24, 25] to meet the practical requirements for early, tumor-specific diagnosis, with high SNR and off-on properties, presents a significant challenge.

In this study, inspired by the MRET principle, a novel tumor microenvironment (TME)-activated T_1 - and

T₂-dual-mode MRI nanoswitch was designed using platelet-derived growth factor (PDGFB)-conjugated ferroferric oxide coated by Mn-doped silica (FMS) yolk-shell nanostructures (PDGFB-FMS) (Scheme 1), which display a distinct "off-on" T₁-T₂ dual-modal synergistic imaging with an ultrafast response, high sensitivity and specificity toward TME. Under normal tissue conditions, this nanostructure for dual-mode imaging is so stable that the T₁ and T₂ MR signals initially display an "off" state. However, when the tumor is targeted, the weakly acidic and high GSH of TME disrupts the PDGFB-FMS, causing its nanostructure to collapse, resulting in the rapid release of Mn^{2+} ions, which separates from the Fe₃O₄ magnetic core and activates the dual-mode of the T₁ and T₂ MRI signals. In addition, the conjugation of PDGFB cycle peptides makes the nanoswitch recognize the tumor site specifically [26-32], allowing the nanoswitch to be effectively internalized by tumor cells and accumulate preferentially at tumor sites. This smart activatable dualmode MRI nanoswitch not only responds specifically to TME but also exhibits no signal (or a very weak signal) in normal tissues, which can reduce background noise and improve SNR. We, therefore, anticipate that the dualmode nanoprobes will play a major role in the development of a variety of diagnostic nanoplatforms.

Results and discussion

Synthesis of FMS yolk-shell nanoswitch

The preparation of FMS consisted of three steps, as illustrated in Scheme 1a. We first synthesized Fe₃O₄ nanoclusters via the solvent thermal decomposition method using ferric acetylacetonate as the iron source and sodium acrylate as the template. As depicted in Fig. 1a, the Fe_3O_4 nanoclusters were monodispersed and uniformly spherical with a size of approximately 60 nm. Moreover, the hydrodynamic diameter of the Fe₃O₄ nanocluster was approximately 60 ± 19.7 nm, which is comparable to the size obtained from transmission electron microscopy (TEM) studies, indicating that the Fe_3O_4 nanocluster had an excellent colloidal performance. In addition, the high-resolution TEM (HRTEM) image reveals that the Fe_3O_4 nanoclusters were highly crystalline, with a lattice fringe distance of ~ 0.253 nm, which corresponds to the (311) atomic plane of magnetite (Fig. 1b). Subsequently, the well-dispersed spherical Fe₃O₄@SiO₂ (FS) core-shell nanoparticles with a shell thickness of \sim 35 ± 16.0 nm were formed by using TEOS as the silica source to coat the Fe_3O_4 under heating and alkaline conditions (Fig. 1c). Finally, a specific "ammonia-assisted Mn²⁺ etching" strategy was used to etch the outer silica shell to form Mndoped silica-coated ferroferric oxide nanoparticles (FMS) via the Ostwald ripening mechanism [33]. Compared to untreated FS nanoparticles, the representative TEM



Scheme 1. a Schematic illustration of synthetic procedures of PDGFB-FMS and b the mechanism of PDGFB-FMS as the intelligent bimodal MRI contrast agents

image revealed well-defined yolk-shell nanostructures with a larger size (~ 220 ± 100.8 nm) (Fig. 1d). In addition, the FMS exhibited a porous structure that was loose and rough, providing an abundance of active sites for proton exchange. Furthermore, the increasing hydrodynamic size of nanoparticles from FS to FMS also confirmed the TEM observations (Additional file 1: Fig. S1). The highangle annular dark field scanning TEM (HAADF-STEM) image of FS revealed the composition and nanostructure of the core-shell, but FMS further illustated the successful doping of Mn ions and nanostructure changes from core-shell to yolk-shell (Fig. 1e, f). Moreover, the energydispersive X-ray (EDX) spectra showed that all expected elements (Fe, Si, Mn, and O) were detected and their relative positions in yolk-shell nanostructures are well matched (Additional file 1: Fig. S2g–i).

In addition, X-ray diffraction (XRD) spectra of Fe_3O_4 revealed the characteristic spinel structure of magnetite (Fig. 1g) [34]. Upon growth of the SiO₂ shell layer, a characteristic broad peak of amorphous silica is produced, whereas it nearly disappears after Mn^{2+} ions etch the FMS. This result indicates that the silica shell is successfully coated on the surface of Fe₃O₄ nanoparticles, and it is effectively etched by Mn²⁺ in the subsequent process. After Mn²⁺ etching, significant increases in specific surface area (342.92 m²/g) and porosity are observed for FMS nanostructures with a pore size distribution dominated by 2.0 nm (Fig. 1h). As is well known, the weight loss of the thermogravimetric (TG) curves below 200 °C is attributable to the loss of water molecules, including physically adsorbed and bound water. Accordingly, the TG analysis suggests that FMS adsorbs significantly more water (7.98%) than FS (6.58%) and Fe_3O_4 (2.35%) (Additional file 1: Fig. S2a). These findings indicate that FMS has a strong affinity for water molecules and facilitates the interaction with the proton, which can improve the relaxation properties.



Fig. 1 a TEM image and **b** HRTEM image of Fe₃O₄. TEM images of **c** FS and **d** FMS. HAADF-STEM and elemental distribution images of **e** FS and **f** FMS. **g** XRD patterns of Fe₃O₄, FS, and FMS. **h** N₂ adsorption/desorption isotherms of FS and FMS (inset: pore size distribution), **i** XPS full spectra of Fe₃O₄, FS, and FMS

In the Fourier transform infrared (FT-IR) spectrum of Fe_3O_4 , the peak of the Fe–O bond appears at 570 cm⁻¹ [34], while the typical peaks of the -C=O and C–H bonds appear at 1635 cm⁻¹ and 2920 cm⁻¹, respectively (Additional file 1: Fig. S2e). In addition, the FT-IR spectrum of FS exhibits a distinct Si–O peak at 1100 cm⁻¹ [35], indicating that silica has been successfully coated on the surface of Fe₃O₄. Moreover, the Si–O peak of FMS shifts significantly from 1100 to 1000 cm⁻¹ as a result of the transformation of the Si–O-Si bond to the Mn–O-Si bond, indicating the successful etching of the silica layer. In the PDGFB-FMS FT-IR curve, the new peak at 1450 cm⁻¹ indicates that PDGFB is successfully anchored to the surface of FMS. The formation of PDGFB-FMS nanoswitch is further supported by ζ potential changes of

particles (Additional file 1: Fig. S2f). The outer SiO₂ layer of the FS exhibits a negative ζ potential of approximately – 46.3 mV. In contrast, the ζ potential of FMS decreased to – 15.5 mV, confirming the presence of Mn²⁺ ions. Furthermore, the ζ potential of PDGFB-FMS was found to be – 18 mV, indicating that the conjugation of PDGFB was successful. The presence of Fe, Si, Mn, C, and O was further confirmed from the full XPS spectrum of FMS (Fig. 1i). Due to the dense, thick SiO₂ shell that obscured the Fe signal in the Fe2p high-resolution spectrum of FS, Fe2p peaks at 711.8 and 724.3 eV vanished. Notably, the Fe peaks of the FMS yolk-shell nanostructure recovered, and strong Mn peaks emerged (Fig. 1i and Additional file 1: Fig. S2b, d), indicating the formation of a loose Mndoped silica shell layer. Remarkably, the Si2p peak for FS shifted significantly from 103.3 to 102.4 eV (Additional file 1: Fig. S2c), further confirming the transformation of Si from a dense silica layer to a loose structure. The hydrodynamic size of PDGFB-FMS had no significant variation in blood, phosphate-buffered saline (PBS), and 10% fetal bovine serum (FBS) within 3 days, demonstating excellent stability (Additional file 1: Fig. S3).

TME-responsive "off-on" T_1-T_2 imaging of FMS nanoswitch

We also analyzed the degradation of FMS under GSH and weak acid conditions. In the presence of weak acidity and GSH, Fig. 2a, b illustrates the significant effects of FMS on the hydrodynamic size distribution and morphology of corresponding particles. Most FMS particles disassemble from regular spherical structures into scattered dots over time. In contrast, no discernible change is observed for FMS in the absence of GSH and under neutral conditions. In addition, the degree of dissociation of FMS increased significantly with increasing GSH concentration and decreasing pH, demonstrating excellent pH- and GSH-responsive degradability (Additional file 1: Fig. S4). To gain more insight into the evolution process, the release of Mn^{2+} ions from FMS with the change of GSH concentration and pH value was investigated. Accordingly, the results confirm that the behavior of Mn²⁺ ion release from FMS is pH- and GSH-dependent (Fig. 2c, d). These results indicate that FMS can be responsively dissociated in TME with an abundance of GSH and an acidic environment by breaking the Mn-O bond and the subsequent collapse of the FMS backbone within one hour (Fig. 2c, d). Based on the excellent degradation of FMS triggered by TME, we further investigated the off-on switchable properties of the dual-mode T₁-T₂ signal in FMS. Under physiological conditions (in pH 7.4 PBS), both T₁- and T₂-weighted imaging (T₁WI and T₂WI, respectively) exhibited an apparent quenching phenomenon (Fig. 2e, g). In addition, the corresponding T_1 relaxivity (r_1) and T_2 relaxivity (r_2) were measured to be 5.1 and 126.46 mM⁻¹ s⁻¹, respectively (Fig. 2f, h). When exposed to GSH (20 mM) and pH 4.5, the FMS nanostructure decomposed rapidly into Fe₃O₄ and Mn²⁺ ions, resulting in a rapid recovery of T_1 and T_2 signals as the distance between the Mn^{2+} ions and Fe_3O_4 magnetic core increased (Fig. 2e, g). After treatment with PBS containing 20 mM GSH and pH 4.5, the calculated r_1 and r_2 values were found to be 11.06 and 304.22 mM⁻¹ s⁻¹, respectively (Fig. 2f, h). In addition, as depicted in Additional file 1: Figs. S5, S6, the "off-on" characteristics of



Fig. 2 a TEM images and **b** corresponding hydrodynamic size changes (blue, 15 min; red, 30 min; green, 60 min) of FMS after treated with different pHs and GSH conditions. **c**, **d** Cumulative Mn^{2+} release curves of FMS at different **c** pH and **d** GSH concentrations. **e** T_1 Wl and **f** corresponding r_1 values of the FMS phantoms after different treatments under 3.0 T. **g** T_2 Wl and **h** corresponding r_2 values of the FMS phantoms after different treatments under 3.0 T.

dual-mode T_1-T_2 nanoprobes were highly dependent on GSH concentrations and pH values. These results, therefore, support our hypothesis that low pH and high GSH concentrations triggered the release of Mn^{2+} , which reduces the mutual magnetic resonance interference between Mn^{2+} and Fe_3O_4 and activates the T_1-T_2 MRI signals (Additional file 1: Tables S1 and S2).

Quenching mechanism of T₁ and T₂ imaging

The T₁-T₂ dual-quenching mechanism of FMS was also investigated methodically. Accordingly, Mn²⁺ ions with a single T_1 contrast and Fe_3O_4 with a single T_2 contrast were used as controls (Additional file 1: Fig. S7a and Fig. 3). For the T_1 quenching effect, we proposed that the strong local magnetic field originating from the large magnetic moment of the Fe₃O₄ magnetic core would interfere with the spin-lattice relaxation process of Mn-doped silica shell with much weaker paramagnetism surrounding it, thereby quenching the T_1 signal of paramagnetic Mn^{2+} . Accordingly, the M-H curves of FMS exhibited a lower saturation magnetization (25 emu/g) at 300 K in comparison to Fe_3O_4 (112.4 emu/g). In addition, the magnetic field generated by the superparamagnetic Fe_3O_4 core significantly disrupts the relaxation process of the paramagnetic Mn-doped silica shell, resulting in the quenching of the T_1 signal during the etching procedure (Fig. 3a). Meanwhile, the grayscale and corresponding pseudo-color T_2WI of Fe₃O₄, FS, and FMS exhibited significant T_2 quenching effects (Fig. 3b, c), and their T₂ relaxivities (r_2) were 317.4 mM⁻¹ s⁻¹, 203.2 mM⁻¹ s⁻¹, and 126.5 mM^{-1} s⁻¹, respectively (Fig. 3d). Therefore, T₁ and T₂ MRI signals can be interfered with when Fe₃O₄ and Mn-doped silica are in direct contact. To further validate the MRET hypothesis, we synthesized a series of FMS (FMS-1, FMS-2, FMS-3, FMS-4, FMS-5) with different separation distances between the paramagnetic Mn^{2+} and the Fe₃O₄ magnetic core by varying the thickness of the silica separation layer from 40 ± 2.6 to 0 ± 1.2 nm (Fig. 3e). Notably, the hydrodynamic size of FMS increased gradually as the etching process progressed (Additional file 1: Fig. S7b), while simultaneously increasing the specific surface area and porosity (Additional file 1: Fig. S8), facilitating the energy exchange with water molecules. The nanoswitch showed obvious quenching in T₁ and T₂ signals with the Mn²⁺ doping ratio from 33 to 88% (Additional file 1: Fig. S9). This was because that high ratio of Mn-doped silica significantly reduced the distance between Fe_3O_4 core and Mn^{2+} , resulting in the increase of quenching effect in T_1 and T_2 imaging. As shown in Fig. 3f, the T₁ MRI signal of the Mn-doped silica shell in FMS decreases as the Mn approaches the Fe₃O₄ core. Furthermore, as the separation distance decreases from 40 to 30, 20, 10, and 0 nm, the r_1 values of the Mn-doped silica shell decrease from 9.7



Fig. 3 a M-H hysteresis loops of Fe_3O_4 , FS, and FMS at 300 K. b Grayscale T_2WI , and c pseudocolor T_2WI , and d corresponding r_2 values of Fe_3O_4 , FS, and FMS. e TEM images, $\mathbf{f} T_1WI$ and T_2WI , $\mathbf{g} r_1$ values, $\mathbf{h} r_2$ values, and \mathbf{i} M-H hysteresis loops at 300 K of FMS-1 to FMS-5

to 8.4, 7.8, 7.3, and 5.1 mM⁻¹ s⁻¹ (Fig. 3g and Additional file 1: Fig. S7c), which is lower than that of free Mn^{2+} (r₁ = 15.9 mM⁻¹ s⁻¹) (Additional file 1: Fig. S7a). Concurrently, a similar T_2 MRI signal quenching phenomenon was observed in T₂WI and corresponding pseudo-color MR images (Fig. 3f). As the etching depth increases, the r₂ value of Fe₃O₄ decreases from 190.85 to 177.66, 149.15, 134.67, and 126.5 $mM^{-1} s^{-1}$ (Fig. 3h and Additional file 1: Fig. S7d). In addition, the saturated magnetization of FMS decreased gradually with increasing Mn content, which could be attributed to the disruption of Fe₃O₄ magnetic moments induced by paramagnetic Mn^{2+} (Fig. 3i). Thus, the aforementioned results demonstrate that the T_1 and T_2 dualquenching effect of FMS is inversely proportional to the separation distance, which can be precisely controlled by the Mn etching process.

Evaluation of internalization and cell MRI of PDGFB-FMS

The hypoxic TME stimulates the overexpression of PDGF-β receptors in breast cancer cells and prostate cancer cells [36, 37]. To obtain tumor-specific delivery, the PDGFB targeting ligand is, therefore, conjugated to FMS to create a PDGFB-FMS nanoswitch that targets tumors. The internalization efficacy of a nanoprobe is essential for evaluating its targeting capability and contrast performance. Using confocal laser scanning microscopy (CLSM) and inductively coupled plasma mass spectrometry (ICP-MS) analysis, the uptake of FMS and PDGFB-FMS was, therefore, investigated in 4T1 cells and PC-3 cells based on this information (Fig. 4a-d, and Additional file 1: Figs. S10, S11). The results of CLSM indicated that the internalization of FMS and PDGFB-FMS is concentration- and time-dependent. In addition, PDGFB-FMS treated cells exhibited a stronger green fluorescence



Fig. 4 a, b CLSM observation: the internalization process of 4T1 cells treated with FMS and PDGFB-FMS at a certain concentration of 10 μ g/mL for 1, 2, and 4 h. c, d CLSM observation: the internalization process of 4T1 cells treated with different concentrations of FMS and PDGFB-FMS for 4 h. e T₁Wl and f T₂Wl of the cells treated with different samples (* and # represent PDGFB-FMS and FMS, respectively) at different time points, and h, i the corresponding T₁ and T₂ MRI SNR analysis. g T₁Wl and T₂Wl of 4T1 cells treated with PDGFB-FMS at the presence of different concentrations of GSH, and j the corresponding T₁ and T₂ MRI SNR analysis

than FMS-treated cells, confirming their superior targeting ability to breast cancer and prostate cancer. We also calculated the colocalization and particle uptake ratios corresponding to the CLSM images. We found that the differences in cell internalization for FMS and PDGFB-FMS treatments are statistically significant (Additional file 1: Fig. S12). In addition, Fe content was used as an indicator for ICP-MS analysis to determine the internalization efficacy of PDGFB-FMS. Results indicate that Fe content in 4T1 cells increased gradually with increasing time and concentration, and cells treated with PDGFB-FMS accumulated more Fe than those treated with FMS. These results are consistent with the CLSM observation, further validating the performance of PDGFB-FMS in targeting tumors.

The imaging performance of PDGFB-FMS was further investigated at the cellular level, in light of the aforementioned encouraging findings. The T₁ and T₂-weighted signals of 4T1 cells treated with FMS and PDGFB-FMS were found to be significantly activated (Fig. 4e, f, h, i) and demonstrate a time-activation relationship. Notably, the T_1 and T_2 signal almost remained unchanged in THLE-3 cells incubated with FMS and PDGFB-FMS, which can be attributed to the lower cellular GSH level and increased internalization efficacy. To determine whether the activation of PDGFB-FMS correlates positively with intracellular GSH levels, 4T1 cells were pretreated with various concentrations of GSH and then incubated continuously with PDGFB-FMS. The results revealed that both the T_1 and T_2 signals of cells were significantly improved (Fig. 4g, j). Moreover, the corresponding relaxation rate change ΔR_1 ($R_1 = 1/T_1$) and ΔR_2 ($R_2 = 1/T_2$) increased gradually after GSH pretreatment (Additional file 1: Table S3), suggesting that the intracellular high level of GSH can be used as a stimulus to activate the dual-mode MRI contrast enhancement.

In vivo MRI studies of PDGFB-FMS

A series of in vivo MRI experiments were also designed and conducted using a mouse model bearing the 4T1 tumor to validate the activable MRI diagnosis for the tumor. FDA-approved "Magnevist" (Gd-DTPA) and "Feridex" (Fe₃O₄) were selected as controls for T₁WI and T₂WI, respectively. Accordingly, the 4T1 tumorbearing mice were intravenously administrated using Gd-DTPA, Fe₃O₄, FMS, and PDGFB-FMS at the dosage of 5 mg/kg, respectively, following which, they were scanned using a 7.0 T MRI scanner, and the T₁WI and T₂WI of the tumor in the axial plane were obtained at various time points. As shown in Fig. 5a, the T₁WI of tumors treated with PDGFB-FMS brightens significantly at 0.5 h post-injection (p.i.) and then darkens over time. In addition, the tumor site of the PDGFB-FMS group in T₁WI was observed to be the clearest and brightest compared to those of FMS and Gd-DTPA. Similarly to T₁WI, the T₂WI of the tumor treated with PDGFB-FMS was observed to be darkest at p.i. 0.5 h (Fig. 5c), following the same pattern as T_1 WI. On the contrary, the contrast enhancement of FDA-approved Gd-DTPA and Fe₃O₄ for T₁WI and T₂WI, respectively, was found to be weak. In addition, the quantitative analysis of tumor regions is performed by measuring the change in SNR (ΔSNR) before and after administration. As shown in Fig. 5b, the maximum Δ SNR of the tumor regions in the PDGFB-FMS group was greater than that of the FMS and Gd-DTPA groups, demonstrating the most pronounced contrast enhancement. In addition, T₂WI of mice treated with FMS exhibited a low Δ SNR, whereas, PDGFB-FMS exhibited a comparable Δ SNR (Fig. 5d), which may be a result of efficient targeting. In addition, the respective T_1 and T₂ map images confirmed the aforementioned findings. As shown in Additional file 1: Fig. S13, there was a 30.7% decrease in the T_1 map value in the tumor region treated with PDGFB-FMS at p.i. 0.5 h, whereas the Gd-DTPA and FMS groups only led to a 16.5% and 16.6% decrease, respectively. Similarly, a significant decrease in the T_2 map value is observed at p.i. 0.5 h, and it follows the same pattern as the T_1 map value. The aforementioned results demonstrate that the PDGFB-FMS dualactivated nanoswitch has the desired dual-modal imaging capability for accurate tumor diagnosis.

To further illustrate GSH tuning nanoswitch, we examined the T_1 and T_2 tumor map images following PDGFB-FMS treatment in the absence and presence of GSH and GSH inhibitor (LBS). Briefly, mice carrying 4T1 tumors were intratumorally pretreated with 10 mM GSH, 2.5 mM LBS (LBS-1), or 5 mM LBS (LBS-2), respectively. The PDGFB-FMS was then administered intravenously to these 4T1 tumor-bearing mice. As anticipated, the GSHpretreated tumor exhibited lower T_1 and T_2 map values at the same PDGFB-FMS injection as compared to the non-GSH-pretreated tumor. In contrast, T₁ and T₂ map value attenuation of the tumor after LBS pretreatment was significantly inhibited, and their inhibition effect was found to be concentration-dependent (Fig. 5e-g). This phenomenon indicates that the decreased intratumoral GSH concentration caused by LBS prevented the activation of GSH-activated T₁-T₂ signals. The findings, therefore, clearly demonstrate the clear positive correlation between PDGFB-FMS activation and intracellular GSH concentrations.

The imaging potential of the PDGFB-FMS nanoswitch for in-situ carcinoma was also investigated (Fig. 6). This study establishes the transgenic adenocarcinoma of mouse prostatic (TRAMP) model to simulate the progression of orthotopic prostate cancer [38, 39]. We first



Fig. 5 a, c Representative T_1WI and T_2WI of 4T1 tumor-bearing mouse at axial planes after intravenous injection of different samples and b, d Δ SNR analysis corresponding to a, c, respectively. Tumor region are marked by white dotted lines. e color-coded T_1 and T_2 map images, f corresponding T_1 map value, and g T_2 map values of 4T1 tumor-bearing mice treated with PDGFB-FMS at the presence of GSH or GSH inhibitor (LBS)

assessed the cytotoxicity of PDGFB-FMS nanoswitch via cell counting kit-8 (CCK-8) assays before in vivo MRI studies. No apparent cytotoxicity against PC-3 cells was observed for PDGFB-FMS nanoswitch, indicating that PDGFB-FMS nanoswitch has good biocompatibility (Additional file 1: Fig. S14). Then, transgenic adenocarcinoma mouse prostate (TRAMP) mice were administered Gd-DTPA, Fe₃O₄, FMS, and PDGFB-FMS intravenously, respectively. Correspondingly, the T₁ and T₂ MR contrasts of the prostate tumor site were significantly enhanced 1 and 4 h post-injection of PDGFB-FMS, respectively. However, the modest dynamic contrast enhancement of T₁ and T₂ in the Gd-DTPA, Fe₃O₄, and FMS groups was visible 8 h after injection (Fig. 6a, b, g, h). The Δ SNR was then used to quantify the MR signals of tumors at different time points (Fig. 6c, i). Accordingly, for T₁-weighted MRI, the maximum Δ SNR of tumors treated with PDGFB-FMS reached up to 75.6±8.8%, which was significantly greater than that of tumors treated with FMS (39.9±6.3%) and Gd-DTPA (32.7±4.5%) (Fig. 6c). Using T₂-weighted MRI, the maximal Δ SNR of tumors treated with PDGFB-FMS, FMS, and Fe₃O₄ is 33.3%, 25.7%, and 39.7%, respectively (Fig. 6i). These results suggest that PDGFB-FMS targets tumor sites and that T₁ and T₂ imaging are effectively illuminated by TME. Furthermore, we collected T₁ and T₂ map images to verify this conclusion (Fig. 6d, e, j, and k). At the tumor treated with PDGFB-FMS, approximately 33.3% of reduction in T₁ map value was detected, whereas FM and Gd-DTPA only led to 19.4% and 10.8%



Fig. 6 MRI visualization of early orthotopic prostate tumor-bearing mice treated with different samples. **a** Grayscale and **b** pseudocolor T_1 WI. **c** Δ SNR corresponding to **a**. **d** Grayscale and **e** color-coded T_1 map images. **f** T_1 map values analysis corresponding to **d**. **g** Grayscale and **h** pseudocolor T_2 WI. **i** Δ SNR corresponding to **g**. **j** Grayscale and **k** color-coded T_2 map images. **I** T_2 map values analysis corresponding to **j**. Tumor regions are indicated by white dotted lines

reduction in T₁ map value, respectively (Fig. 6f). Moreover, PDGFB-FMS exhibited a significant decrease in T₂ map value (41.7% at p.i. 4 h), whereas T₂ map value changes for FMS and Fe₃O₄ were 31.9% and 53.0%, respectively (Fig. 6l). Considering the significantly higher T₁ and T₂ Δ SNR and the sensitive "off–on" dual-mode MRI signals, the PDGFB-FMS nanoswitch demonstrates its superiority in the area of early-stage cancer detection.

In vivo biosafety assessment of PDGFB-FMS nanoswitch

As is well-known, the biosafety of a nanoagent is a crucial factor to consider when assessing its clinical application potential [40, 41]. Based on this, we first evaluate the biocompatibility of PDGFB-FMS with blood using a hemolysis assay and a routine blood examination. As shown in Fig. 7a and b, the hemolysis rate of PDGFB-FMS is extremely low and negligible, indicating that PDGFB-FMS cannot damage red cells. In addition, major blood routine indicators such as WBC, RBC, PLT, HGB, and HCT (Additional file 1: Fig. S15) at p.i. PDGFB-FMS did not demonstrate any significant changes. These results demonstrate the biocompatibility of PDGFB-FMS with blood. Subsequently, we also investigated cytoxicity of PDGFB-FMS on normal cells (THLE-3, 293T, PC-12, and SH-SY5Y cells). As shown in Additional file 1: Fig. S16a-c, the PDGFB-FMS is biocompatible with THLE-3,



Fig. 7 a Hemolysis photos and b hemolysis rate of PDGFB-FMS. c Bio-distribution of Mn element after i.v. injection of PDGFB-FMS. d H&E staining of major organs in the mice treated with FMS and PDGFB-FMS for 7 days. The scale is 50 µm. e Bio-distributions of Mn element in major organs after treated with PDGFB-FMS and FMS for 24 h. Pharmacokinetic curves of f PDGFB-FMS, g FMS, and h Gd-DTPA in vivo

293T, PC-12, and SH-SY5Y cells over a wide range of concentrations, exhibiting no toxicity. We also explored tissue toxicity of PDGFB-FMS using hematoxylin and eosin (H&E) analysis. After 7 days of PDGFB-FMS treatment, no obvious abnormalities in the vital organs were observed, confirming the excellent biosafety of the tissue (Fig. 7d). In addition, the biodistribution analysis revealed that the reticuloendothelial system was responsible for the majority of PDGFB-FMS accumulation in the liver (Fig. 7c). Notably, PDGFB-FMS levels in the body decreased significantly over time, indicating that it can be excreted effectively. Except for the accumulation in the liver at the early stages, PDGFB-FMS shows higher accumulation in tumors than non-targeted FMS, demonstrating excellent tumor-targeting (Fig. 7e). Further research was also conducted into the pharmacokinetics of PDGFB-FMS, FMS, and Gd-DTPA. At p.i. 10 h, the residual concentrations of FMS and Gd-DTPA were only 18.9% and 4.6%, respectively, while PDGFB-FMS was 23.8%. Moreover, PDGFB-FMS had the longest half-life in the blood ($t_{1/2}$ =3.68 h) compared to FMS ($t_{1/2}$ =2.88 h) and Gd-DTPA ($t_{1/2}$ =0.97 h) (Fig. 7f–h). Notably, at p.i. 72 h, PDGFB-FMS could be completely eliminated from the body, thereby avoiding the potential risks to the body posed by long-term residual [42]. To summarize, the superior biocompatibility of PDGFB-FMS and its prompt in vivo clearance facilitate its use in tumor molecular MRI.

Conclusion

In conclusion, TME-activated "off-on" T_1-T_2 dualmode MRI nanoswitch (PDGFB-FMS) has been successfully designed and fabricated. This novel nanoswitch structure is composed of a superparamagnetic Fe_3O_4 core and a paramagnetic Mn-doped silica shell, in which the T_1 and T_2 signals are suppressed due to the phenomenon of distance-dependent magnetic resonance tuning. When exposed to GSH and a low pH value, the T_1-T_2 dual-mode MRI signals are activated with significant T_1 and T_2 contrast enhancement. PDGFB-FMS has been utilized for the precise diagnosis of heterotopic and orthotopic tumors due to its increased SNR and tumor-specificity. Biosafety assessments further confirm the low toxicity of PDGFB-FMS. In addition, due to its rapid biodegradability, PDGFB-FMS can be excreted out of the body in a timely manner, posing no potential risk to the body. This study, therefore, provides a promising strategy for the development of intelligent MRI contrast agents to achieve accurate diagnosis of tumors.

Supplementary Information

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Additional file 1: Figure S1. The hydrodynamic sizes of different samples. Figure S2. TG curves, XPS spectra, FT-IR spectra, and zeta potential of different samples. STEM image of FMS. Figure S3. Hydrodynamic size variation of PDGFB-FMS in different media. Figure S4. TEM images of FMS after different treatments and corresponding hydrodynamic size distribution. Figure S5. T₁WI and T₂WI of FMS. Figure S6. r₁ and r₂ values of FMS. Figure S7. r₁ value of free Mn²⁺ ions at 3.0 T, hydrodynamic size distribution of FMS-1 to FMS-5, r1 linear fit curves and r2 linear fit curves of FMS-1 to FMS-5. Figure S8. N2 adsorption-desorption isotherms and corresponding pore size distribution of FS, and FMS-1 to FMS-5. Figure **S9.** T_1-T_2WI of FMS with different Fe_3O_4 to Mn^{2+} ratios. **Figure S10.** Fe content of 4T1 cells. Figure S11. CLSM observation: the internalization of PC-3 cells. Figure S12. Colocalization and particle uptake ratios of 4T1 cells. Figure S13. Representative color-coded T₁ and T₂ map images of 4T1 tumor-bearing mice and T_1 and T_2 map values analysis. Figure S14. Cytotoxicity of PDGFB-FMS against PC-3 cells. Figure S15. Blood routine examination of mice. Figure S16. The viabilities of THLE-3, 293 T, PC-12, and SH-SY5Y cells. Table S1. Δr_1 and Δr_2 value of FMS under different pH conditions. Table S2. Δr_1 and Δr_2 value of FMS under different GSH conditions. **Table S3.** In vitro ΔR_1 and ΔR_2 of 4T1 cells treated with PDGFB-FMS at the presence of different concentrations of GSH.

Author contributions

JQ and GZ designed this study. YZ and CZ prepared materials and carried out the corresponding characterization. WL and TS carried out the cell and the animal experiments. PW, DS, XQ, and XH also performed some animal experiments. LL and GZ are all responsible for writing the manuscript. Correspondence and requests for materials should be addressed to GT and GZ. All authors read and approved the final manuscript.

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Declarations

Ethics approval and consent to participate

Mice used in this study were treated in accordance with the ethics committee guidelines at the Binzhou Medical University.

Competing interests

The authors declare no competing financial interest.

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