



Article Spatial Distribution of Polyphenolic Compounds in Corn Grains (Zea mays L. var. Pioneer) Studied by Laser Confocal Microscopy and High-Resolution Mass Spectrometry

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Abstract: Desirable changes in the biochemical composition of food plants is a key outcome of breeding strategies. The subsequent localization of nutritional phytochemicals in plant tissues gives important information regarding the extent of their synthesis across a tissue. We performed a detailed metabolomic analysis of phytochemical substances of grains from *Zea mays* L. (var. *Pioneer*) by tandem mass spectrometry and localization by confocal microscopy. We found that anthocyanins are located mainly in the aleurone layer of the grain. High-performance liquid chromatography in combination with ion trap tandem mass spectrometry revealed the presence of 56 compounds, including 30 polyphenols. This method allows for effective and rapid analysis of anthocyanins by plotting their distribution in seeds and grains of different plants. This approach will permit a more efficient screening of phenotypic varieties during food plant breeding.

Keywords: confocal microscopy; HPLC-MS/MS; tandem mass spectrometry; polyphenolic compounds

1. Introduction

The consumption of corn for 2018–2019 reached 315 million tons in the USA, 276 million tons in China, 63 million tons in the European Union, and 66 million tons in Brazil. In maize breeding, the discovery of genes responsible for the formation of corn endosperm accelerated research on the modeling of nutritional and taste properties of the corn.

The biochemical composition of corn grains, including protein, fatty acid, saccharide, and phenolic content, significantly affect the nutritional quality and taste of corn. The content of essential amino acids, such as valine, isoleucine, leucine, lysine, methionine, threonine, tryptophan, phenylalanine, histidine, and arginine is one of the major factors that determine the nutritional value of corn [1].

Corn grains have the highest polyphenol content (6056.9 mg/kg dry weight or $15.55 \,\mu$ mol/g) among other grains and represent significant interest for phytochemical and metabolomic study [2,3]. Phenolic compounds can have radical scavenging, chelating and



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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). antioxidative activity. Polyphenols can prevent oxidative stress caused by metabolic imbalances between the production and scavenging of free radicals [4]. Phenolic compounds can control oxidative stress by neutralizing or reducing the formation of reactive oxygen species (ROS) or restoration of redox homeostasis by strengthening the endogenous defense system or capturing the ROS [5]. The ability of polyphenolic groups to scavenge free radicals is associated with their aromatic rings and a highly conjugated system with many hydroxyl groups. The spatial position and the number of hydroxyl groups are important reference points for the antioxidant activity of phenolic compounds [6].

Fatty acids affect the palatability and especially the odor of foods. In higher plants, the proportion of essential fatty acids in the composition of vegetable fats is very high (up to 90%). It is mainly composed of palmitic, oleic, and linoleic acids. Analysis of the fatty acid composition of corn grains showed the presence of palmitic acid, linoleic acid, vaccenic acid, oleic acid, stearic acid, and eicosanoic acid.

Monosaccharides are derivatives of polyhydric alcohols and serve as a source for the synthesis of disaccharides (sucrose, maltose, lactose), oligosaccharides, and polysaccharides (cellulose and starch). Many of them have a sweet taste, but there are gradations from tasteless to bitter substances that affect the taste qualities of grains, including corn.

Flavonoids act as exogenous antioxidants and are directly oxidized by radicals with the formation of less reactive species through the following mechanisms: inhibition of xanthine oxidase activity, modulation of channel pathways, and inhibition of nitric oxide synthase activity [7]. The antioxidant potential of flavonoids is associated with the location and the total number of OH groups, or rather, with their molecular structure [8]. The use of flavonoids in biological systems holds great promise for bone tissue engineering. Quercetin, an antioxidant flavonoid, when present in the bloodstream, improves vascular health and reduces the risk of cardiovascular disease in its conjugated form. Quercetin and its derivatives prevent blood clotting or thrombosis and prevent the likelihood of stroke [9].





Figure 1. Structure of the grain of dent corn (with the symbolic designation of parts of the grain) (modified from [10]).

Previous research organized phenolic compounds according to the degree of antioxidant activity: simple phenolic acids < hydroxycinnamic acids < flavonols < flavan-3-ols < dimers of procyanidins [6]. It is known that the antioxidant activity of phenolic acids increases with an increase in the distance separating the carbonyl group and the aromatic

ring, and hydroxycinnamic acid derivatives have stronger antioxidant activity than benzoic acid derivatives [11]. The 7,8-double bond of hydroxycinnamic acids also enhances their antioxidant potential, compared with hydroxybenzoic acids.

Jigh-performance liquid chromatography (HPLC) was predominantly used to identify carotenoids [12–14] and polyphenols [15,16] in corn grains. A review by Ranilla (2020) summarized the application of metabolomics for the characterization of metabolites in corn grains and emphasized the importance of phenotype–genotype studies aimed to explore corn genetic diversity [17]. The application of electrospray ionization mass spectrometry (ESI–MS) in combination with HPLC is a cost-effective and statistically robust method for high-throughput phenotypic characterization of corn [18]. The HPLC–ESI–MS/MS analytical configuration is widely used for the characterization of phenolic bioactive compounds in worldwide corn biodiversity. Montilla et al. (2011) characterized 10 corn landraces based on the content of phenolic fractions [19]. Das and Singh (2016) characterized four corn hybrids based on the content of phenolic acids, anthocyanins, and flavonols [20].

Another important problem is the study of spatial distribution and composition of phytochemicals in corn grains. Microscopic images are widely used as important sources of information on morphometric characteristics of cells and the architecture of plant tissue [21]. Confocal laser scanning microscopy was previously used to localize the phenolic compounds in different plants [22]. Morphological and biochemical changes in roots of corn *Zea mays* L. were previously studied by confocal microscopy [23,24]. To the best of the authors' knowledge, no published studies report an application of confocal microscopy for the identification of phytochemicals in the grains of corn *Zea mays* L.

Considering the qualitative data of phytochemical composition obtained by HPLC– MS and literature information regarding the optical properties of identified chemicals, the combination of HPLC data with fluorescence microscopy is a good opportunity to explore the localization of phenolic compounds in plants. The combination of these methods is important for breeding since it allows us to assess whether the genes involved in the synthesis of these substances are expressed only in certain tissues (e.g., the aleurone layer, the germ layer, the vitreous endosperm) or in all grain glutes uniformly. In addition, this approach makes it possible to estimate the number and size of storing organelles (granules, chloroplasts, vesicles), since selection is important in both increasing their number and increasing their size. Thus, the combination of these methods allows us to obtain more complex information about the studied plants.

In this study, we used combined mass spectrometry and confocal laser microscopy to determine the structural properties and phytochemical composition of corn grains. In our case, the combination of HPLC–MS and fluorescence microscopy allowed us to demonstrate the localization of polyphenolic compounds in the grains of corn *Zea mays* L. However, the interpretation of the results of this study requires taking into account the limitations of the study design. The application of combined HPLC and fluorescent microscopy includes the possibility of spatial localization of different groups of plant chemicals in general but not the individual compounds.

2. Results

2.1. Tandem Mass Spectrometry

The extracts of corn grains were analyzed using liquid chromatography–electrospray ionization mass spectrometry (LC–ESI MS) to explore the diversity of available phytochemicals. The structural identification of each compound was carried out based on their accurate mass and MS/MS fragmentation using LC–ESI MS. In total, 56 compounds were successfully identified and characterized by comparing fragmentation patterns with those available in the literature. The results of a preliminary study showed the presence of 56 compounds corresponding to the genus *Zea*, some of which were identified for the first time in *Zea mays* L. The identified compounds, with molecular formulas m/z calculated and observed MS/MS data, and their comparative profile for corn grains are summarized



in Table A1. The chromatograms of total compounds in the grain extract in positive and negative ionization modes are presented in Figure 2.

Figure 2. The total compounds chromatogram of Zea mays L. (var. Pioneer) extract.

In the present study, 30 polyphenol compounds were identified and characterized. In addition, 26 compounds of other classes were identified, including identified for the first time in corn grains oxylipins 13-trihydroxy-octadecenoic acid and 9,12,13-Trihydroxy-*trans*-10-octadecenoic acid.

Figures 3 and 4 show examples of the decoding spectra (collision-induced dissociation (CID) spectrum) of the ion chromatogram obtained using tandem mass spectrometry. The $[M-H]^-$ ion produced three fragment ions at m/z 171, m/z 211, and m/z 293 (Figure 3). The fragment ion at m/z 171 yields a daughter ion at m/z 153. This compound was identified in the bibliography as 13-trihydroxy-octadecenoic acid (THODE) in extracts from *Bituminaria* [25], *Broccoli* [26], *Sasa veitchii* [27].



Figure 3. Mass spectrum of 13-trihydroxy-octadecenoic acid (THODE) from the extract of corn grains, m/z 329.20.

The mass spectrum in the positive ion mode of pelargonidin-3-*o*-glucoside from extracts of corn grains is shown in Figure 4. The $[M + H]^+$ ion produced three fragment ions at m/z 271, m/z 415, and m/z 186 (Figure 4). The fragment ion at m/z 271 yields two daughter ions at m/z 253 and m/z 121. The fragment ion at m/z 253 yields one daughter ion at m/z 235. To our knowledge, pelargonidin-3-*o*-glucoside was reported in *Triticum aestivum* L. [28,29], strawberry [30].



Figure 4. Mass spectrum of pelargonidin-3-O-glucoside from extracts of corn grains, *m*/*z* 432.77.

2.2. Confocal Microscopy

Confocal microscopy, coupled with Airyscan technology, demonstrated blue (Figures 5b, 6b and 7b) and red fluorescence (Figures 5c, 6c and 7c) in the longitudinal and transverse sections, and in the aleurone layer of the grain, respectively.

According to the literature data, strong blue fluorescence of plant grains under UV excitation could be explained by the presence of phenolic compounds such as hydroxycinnamic [31] or ferulic acid [32], and lignin [33]. The endosperm reveals very low blue autofluorescence (Figures 6 and 7) due to the very low amount of phenolic substances in the endosperm cells of seeds and grains [34]. It was reported that the pericarp of *Zea mays* had a total phenolic content 30–34 fold higher than endosperm [35]. Our results demonstrated that the aleurone cells (Figures 5b, 6b and 7b) and embryo (Figure 7b) were enriched with blue autofluorescence substances. At the same time, it is known that no lignin is present in aleurone [36], but hydroxycinnamic, ferulic, and coumaric acids were reported in aleurone cells of cereals [37,38]. Therefore, the observed blue fluorescence might be caused by hydroxycinnamic, ferulic, and coumaric acids. The main blue fluorescent compound in the pericarp is lignin, which is a heterogeneous mixture of randomly polymerized phenolic monolignols [39].

The emission in the red spectrum mainly occurs due to the presence of various polyphenolic compounds, including anthocyanins and anthocyanidins [40].



(a)



(b)

Figure 5. Cont.



Figure 5. The longitudinal section of the grain (grain margin in the embryo area), $63 \times$ magnification: (a) multispectral image, excitation 405 nm with the emission in 400–470 nm (blue), excitation 488 nm with the emission in 620–700 nm (red); (b) hydroxycinnamic and ferulic acids, and lignin content in the corn grain indicated in blue spectra; (c) anthocyanin content in the grain indicated in red spectra; p, pericarp; al, aleurone.



Figure 6. Cont.



Figure 6. The transverse section of the grain, a border between endosperm (left) and embryo (right), 20× magnification: (**a**) multispectral image, excitation 405 nm with the emission in 400–470 nm (blue), excitation 488 nm with the emission in 620–700 nm (red); (**b**) hydroxycinnamic and ferulic acids, and lignin content in the corn grain indicated in blue spectra; (**c**) anthocyanin content in the grain indicated in red spectra; p, pericarp; al, aleurone; en, endosperm; em, embryo.

(c)

100 µm

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(a)



(b)





Figure 7. The aleurone layer of the grain (upper margin of the grain, the longitudinal section), $63 \times$ magnification: (**a**) multispectral image, excitation 405 nm with the emission in 400–470 nm (blue), excitation 488 nm with the emission in 620–700 nm (red); (**b**) hydroxycinnamic and ferulic acids, and lignin content in the corn grain indicated in blue spectra; (**c**) anthocyanin content in the grain indicated in red spectra; p, pericarp; al, aleurone; en, endosperm.

3. Discussion

It is known that polyphenols have strong antioxidation, anticancer, anti-infection, and other valuable activities [41]. The knowledge of polyphenol distribution in plants will benefit the development of the methods of their direct extraction and further application in the food, pharmaceutical, and cosmetic industries.

Another important problem is the influence of environmental conditions on the polyphenol composition of the plants. The significant genotypic effects and interactions of the genotype with the environment suggest that breeding methodology will require careful site selection and accounting for changes in genotype rank with changes in cultivation sites.

The important characteristics such as grain color, protein, and polyphenol distribution represent significant interest for breeding. In the grain images, the fluorescence signal under UV excitation (405 nm) comes from ferulic acid [42] and lignin [33]. It should be noted that lignin is absent in aleurone, while coumaric and diferulic acids are present in the walls of aleurone cells. These acids can contribute to the autofluorescence of these cell walls [43,44].

Autofluorescence in the aleurone cell walls was not uniform, which is consistent with the studies presented below. Saadi et al. (1998) showed that autofluorescence was more intense in the anticline than in the periclinal cell walls of the corn grains [45]. Moreover, studies have shown that the content of ferulic acid in the anticlinal cell wall of the corn was twice as high as in the periclinal cell wall [46]. However, research by Phillippe et al. [34] argues that anticlinal and periclinal cell walls contain equal amounts of feruloylated arabinoxylan. Therefore, it seems that autofluorescence in the walls of anticlinal aleurone cells can additionally be caused by other substances, for example, coumaric and diferulic acids, which were found in aleurone cells [37].

Our study showed the metabolic profile of the corn *Zea mays* L. (var. *Pioneer*) represented as 56 compounds including 2 compounds identified in corn grains for the first time—namely, oxylipins 13-trihydroxy-octadecenoic acid and 9,12,13-trihydroxy-*trans*-10-octadecenoic acid. Laser microscopy showed the presence of polyphenolic compounds and, in particular, hydroxycinnamic and ferulic acids, and anthocyanins, in the tissues of corn grain.

The method used in this study is effective for rapid analysis of the distribution of polyphenolic compounds in seeds and grains of different plants. This approach allows the study of plant morphology and the characterization of relevant bioactive phytochemicals using an inexpensive and fast methodology. The characterization of novel corn hybrid genotypes harvested from different geographical areas is a strategic problem and addressing this problem would allow sustainable development of local agriculture.

4. Materials and Methods

4.1. Materials and Chemicals

As an object of research, we used corn grains *Zea mays* L., variety *Pioneer* P1467. The sample was harvested in 2020 in urban-type settlement Kirovsky (Primorsky Krai, Russian Far East) and obtained from a local farmer.

HPLC-grade acetonitrile was purchased from Fisher Scientific (Southborough, UK), MS-grade formic acid was from Sigma-Aldrich (Steinheim, Germany). Ultra-pure water was prepared from SIEMENS ULTRA clear (SIEMENS Water Technologies, Munich, Germany), and all other chemicals were analytical grade.

4.2. Fractional Maceration

Fractional maceration technique was applied to obtain highly concentrated extracts [47]. From 500 g of the sample, 4 g of corn seeds was randomly selected for maceration. The total amount of the extractant (ethyl alcohol of reagent grade) was divided into 3 parts, and the grains were consistently infused with the first, second, and third parts. The solid–solvent ratio was 1:20. The infusion of each part of the extractant lasted 7 days at room temperature.

4.3. Liquid Chromatography

HPLC was performed using Shimadzu LC-20 Prominence HPLC (Shimadzu, Kyoto, Japan) equipped with a UV sensor and C18 silica reverse phase column (4.6×150 mm, particle size: 2.7 µm) to perform the separation of multicomponent mixtures. The gradient elution program with two mobile phases (A, deionized water; B, acetonitrile with formic acid 0.1% v/v) was as follows: 0–2 min, 0% B; 2–50 min, 0–100% B; control washing 50–60 min 100% B. The entire HPLC analysis was performed with a UV–vis detector SPD-20A (Shimadzu, Kyoto, Japan) at a wavelength of 230 nm for identification of catechin, epicatechin, quercetin, and other compounds [48]; the temperature was 50 °C, and the total flow rate 0.25 mL min⁻¹. The injection volume was 10 µL. Additionally, liquid chromatography was combined with a mass spectrometric ion trap to identify compounds.

4.4. Mass Spectrometry

MS analysis was performed on an ion trap amaZon SL (Bruker Daltoniks, Bremen, Germany) equipped with an ESI source in negative ion mode. The optimized parameters were obtained as follows: ionization source temperature: 70 °C, gas flow: 9 L/min, nebulizer gas (atomizer): 7.3 psi, capillary voltage: 4500 V, endplate bend voltage: 1500 V, fragmentary: 280 V, collision energy: 60 eV. An ion trap was used in the scan range m/z 100–1.700 for MS and MS/MS. All experiments were repeated three times. A four-stage ion separation mode (MS/MS mode) was implemented.

4.5. Optical Microscopy

Before the microscopic examination, a longitudinal and transverse dissection of corn grains was performed with MS-2 sled microtome (Tochmedpribor, Ukraine). The obtained

sliced corn grains were placed on microscopic cover glass through immersion oil to reduce light refraction by air gaps.

The autofluorescence parameters of a slice of corn grain were determined using an inverted confocal microscope (confocal laser scanning microscopy—CLSM, LSM 800, Carl Zeiss Microscopy GmbH, Berlin, Germany). The autofluorescence spectrum was chosen using lambda scan mode of the confocal microscope, which allows to determine the emission maximum in a specific sample and obtain spectral acquisition. The specimen was excited by each laser separately and two main peaks of autofluorescence were revealed: excitation by a UV laser, 405 nm (solid state, diode, 5mW) with the emission maxima in the ranges 400–470 nm (blue); excitation by a blue laser, 488 nm (solid state, diode, 10 mW) with the emission maximum in 620–700 nm (red). The used power and detector gain for blue and red channels were 5% and 750 V, and 7% and 850 V, respectively.

The images were obtained using objectives Plan-Apochromat $20 \times /0.8$ M27 and Plan-Apochromat $63 \times /1.40$ Oil DIC M27 with $20 \times$ and $63 \times$ magnification, correspondingly. The zoom factor was 0.5. Airyscan at the SR mode was used to increase resolution. The software ZEN 2.1 (Carl Zeiss Microscopy GmbH, Berlin, Germany) was used for image acquisition.

5. Conclusions

We determined the qualitative characteristics of secondary metabolites in the tissues of corn *Zea mays* L. (var. *Pioneer*). In total, 56 compounds were identified, including 2 compounds identified in corn grains for the first time—namely, oxylipins 13-trihydroxy-octadecenoic acid and 9,12,13-trihydroxy-*trans*-10-octadecenoic acid.

The combination of these data with fluorescence microscopy data revealed the most probable localization of phenolic and polyphenolic compounds. In addition, confocal microscopy allowed us to assess the localization of hydroxycinnamic and ferulic acids in aleurone cells and embryos and anthocyanin content in pericarp and aleurone cells. The combination of these methods is important for breeding since it allows us to assess whether the genes involved in the synthesis of these substances are expressed only in certain tissues (the aleurone layer, the germ layer, the vitreous endosperm) or in all grain glutes uniformly. In addition, this approach makes it possible to estimate the number and size of storing organelles (granules, chloroplasts, vesicles), since selection is important both in the area of increasing their number and increasing their size. Thus, the combination of these methods allows us to obtain more complete information about the variables under study. In addition, it shows that confocal microscopy can be used to obtain preliminary information during volumetric screenings of varietal samples, which will allow selecting target groups for more detailed analysis much faster and without the use of expensive reagents.

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Appendix A

Table A1. The list of compounds identified in ethanolic extracts of Zea mays L. (var. Pioneer) grains.

No.	Class	Compound	Molecular Formula	Calculated Mass	Molecular Ion [M-H] ⁻	Molecular Ion [M+H] ⁺	Fragmenation Ion MS2	Fragmentation Ion MS3	Fragmentation Ion MS4	References
		POLYPHENOLS								
1	Phenolic acid	Caffeic acid [(2E)-3-(3,4-Dihydroxyphenyl)acrylic acid]	C ₉ H ₈ O ₄	180.1574		181	135	119		Dracocephalum palmatum [49]; Eucalyptus [50]; Triticum [51]; Salvia miltiorrhiza [52]
2	Phenolic acid	Hydroxy methoxy dimethylbenzoic acid	$C_{10}H_{12}O_4$	196.1999		197	177; 153	125		F. herrerae; F. glaucescens [53]
3	Phenolic acid	Hydroxyferulic acid	$C_{10}H_{10}O_5$	210.1834		211	193; 125			Andean blueberry [54];
4	Stilbene	Resveratrol [trans-Resveratrol; 3,4',5-Trihydroxystilbene; Stilbentriol]	C ₁₄ H ₁₂ O ₃	228.2433		229	209	163	146	A. cordifolia; F. glaucescens; F. herrerae [53]; Radix polygoni multiflori [55]
5	Dihydroxybenzoic acid	3,4-Diacetoxybenzoic acid	C ₁₀ H ₁₁ O ₆	238.1935	237		119			Potato leaves [56]; Triticum aestivum L. [57];
6	Flavan-3-ol	Epiafzelechin [(epi)Afzelechin]	$C_{15}H_{14}O_5$	274.2687		275	245; 176	175		Cassia granidis [58]; Cassia abbreviata [59,60]; A. cordifolia; F. glaucescens; F. herrerae [53]
7	Flavonol	Kaempferol [3,5,7-Trihydroxy-2-(4-hydro- xyphenyl)-4H-chromen-4-one]	$C_{15}H_{10}O_{6}$	286.2363	285		185; 117; 257	117		Rhus coriaria (Sumac) [61]; Lonicera japonicum [62]; Andean blueberry [54]; Potato [63]; Potato leaves [56];
8	Flavan-3-ol	Catechin [D-Catechol]	$C_{15}H_{14}O_{6}$	290.2681		291	261; 189	173; 242	191; 143	Potato [64]; Triticum [51]; millet grains [65]; Solanaceae [66]; Beer [67]; V. edulis [53]; Vigna inguiculata [68]; Radix polygoni multiflori [55]; Senna singueana [69]; Camellia kucha [70];
9	Flavan-3-ol	(epi)catechin	$C_{15}H_{14}O_{6}$	290.2681		291	261; 173	243; 173		C. edulis [53]; Radix polygoni multiflori [55]; Camellia kucha [70];
10	Hydroxycinnamic acid	Caffeoylmalic acid	$C_{13}H_{12}O_8$	296.2296	295		277; 171	233; 113		Potato leaves [56]; Strawberry [71]

Table A1. Cont.

Calculated Molecular Ion Molecular Ion Molecular Fragmenation Fragmentation Fragmentation No. Class References Compound Formula Mass [M-H][M+H]+ Ion MS2 Ion MS3 Ion MS4 Rhus coriaria [61]; Potato leaves [56]; Vigna sinensis [72] Impatiens glandulifera Royle 275; 245; 203; 303 [73]; Eucalyptus [50]; Triticum [51]; millet grains [65]; 11 C15H10O7 302.2357 175 Flavonol Quercetin 175 Tomato [74]; Bougainvillea [75] millet grains [65]; Solanaceae [66]; Licania ridigna [76]; G. linguiforme [53]; Senna singueana [69]; 12 Flavan-3-ol Gallocatechin [+(-) Gallocatechin] C15H14O7 306.2675 307 277; 207 207; 159 Vaccinium myrtillus [77] Dracocephalum palmatum [49]; Potato [63,64]; Perilla frutescens [78]; Tomato [74]; 13 C15H10O8 318.2351 319 291; 219; 174 259; 191 243; 161 Mentha [79]; Flavonol Myricetin Salvia miltiorrhiza [52]; Rubus occidentalis [80]; Sanguisorba officinalis [81]; Radix polygoni multiflori [55] Cirsiliol 330.2889 329 14 Flavone $C_{17}H_{14}O_7$ 229; 171; 293 211; 155 183 Ocimum [82] 15 Flavone 5,7-Dimetoxy-3,3',4'-trihydroxyflavone C17H14O7 330.2889 331 315; 270 313 285; 257 Oxalis corniculata [83] 447 287 152 16 Flavone Luteolin 7,3'-disulphate C15H10O12S2 446.3627 Zostera marina [84] G. linguiforme [53]; 17 Flavone Apigenin 7-sulfate $C_{15}H_{10}O_8S$ 350.3001 351 337; 308 308; 291 sulphates [85] Punica granatum [86]; Wheat [87]; Matairesinol [(–)-Matairesinol; Artigenin 18 Lignan C20H22O6 358.3851 359 324; 289; 127 144 127 Congener] Lignans [88] Hydroxycinnamic 19 Caffeic acid derivative $C_{16}H_{18}O_9Na$ 377.2985 377 341; 215 179; 113 Bougainvillea [75] acid derivative Gallate ester, 20 426.3729 427 301; 171; 382 171 derivative of Epiafzelechin 3-O-gallate C22H18O9 *Camellia kucha* [70]; epiafzelechin 418; 314; 265; Triticum durum [89]; 21 Flavone $C_{21}H_{20}O_{10}$ 432.3775 433 257; 169 Apigenin-C-hexoside 219; 155 Beer [67] Triticum aestivum L. [28,29]; 22 Anthocyanidin Pelargonidin-3-O-glucoside (callistephin) C21H21O10 433.3854 433 271; 185 253; 121 235 strawberry [30] Triticum [29,51]; Cyanidin-3-O-glucoside [Cyanidin acerola [90]; rice [91]; Vigna 23 Anthocyanidin $C_{21}H_{21}O_{11} +$ 449.3848 447 285 199 sinensis [72]; Rapeseed petals 3-O-beta-D-Glucoside; Kuromarin] [92]

Table A1. Cont.

No.	Class	Compound	Molecular Formula	Calculated Mass	Molecular Ion [M-H]-	Molecular Ion [M+H] ⁺	Fragmenation Ion MS2	Fragmentation Ion MS3	Fragmentation Ion MS4	References
24	Flavone	Luteolin-7-0-beta-glucuronide	$C_{21}H_{18}O_{12}$	462.3604		463	447; 395; 359; 285; 199; 149	287; 199		Mentha [93,94]; rat plasma [95]; Newbouldia laevis [60]
25	Flavonol	Kaempferol-3-O-glucuronide	$C_{21}H_{18}O_{12}$	462.3604		463	287; 198	269; 198		Strawberry [71]; A.cordifolia; G. linguiforme [53]; Rhus coriaria [61]
26	Anthocyanidin	Delphinidin malonyl hexoside	$C_{24}H_{23}O_{15}$	551.4304		551	465; 287; 185	287; 115		F. glaucescens [53]
27	Flavone	Chrysoeriol C-hexoside-C-pentoside	$C_{27}H_{30}O_{15}$	594.5181		595	578; 536; 509; 425	294		Triticum aestivum L. [57,96]; T. durum [89]
28	Flavonol	Quercetin 3,4'-di-O-beta-glucopyranoside [Quercetin diglucoside]	$C_{27}H_{30}O_{17}$	626.5169		627	465	447; 405; 303		Potato leaves [56]; Potato [63]; Rapeseed petals [92];
29	Flavone	Tricin trimethyl ether 7-O-hexosyl-hexoside	C ₃₀ H ₃₆ O ₁₇	668.5966		669	345; 387; 283			Triticum aestivum L. [97]
30	Flavan-3-ol	(Epi)fisetinidol-(epi)catechin-A- (epi)fisetinidol	$C_{45}H_{36}O_{16}$	832.7577	831		721; 693; 609; 575; 537; 506			Chamaecrista nictitans [98]
		OTHER COMPOUNDS								
31	Amino acid	L-Lysine	$C_6H_{14}N_2O_2$	146.1876		147	119			Lonicera japonica [62];
32	Amino acid	L-threanine	$C_7 H_{14} N_2 O_3$	174.1977		175	159			Camellia kucha [70]
33	Amino acid	L-Tryptophan [Tryptophan; (S)-Tryptophan]	$C_{11}H_{12}N_2O_2$	204.2252		205	161; 159	143		Passiflora incarnata [99]; Vigna unguiculata [100]; Camellia kucha [70];
34	Omega-5 fatty acid	Myristoleic acid [Cis-9-Tetradecanoic acid]	$C_{14}H_{26}O_2$	226.3550		227	209; 168	127		F. glaucescens [53]
35	Monobasic saturated carboxylic acid	Myristic acid [Tetradecanoic acid; N-Tetradecanoic acid]	$C_{14}H_{28}O_2$	228.3709		229	142; 205	114		Rhododendron adamsii [101]
36	Medium-chain fatty acid	Hydroxy dodecanoic acid	$C_{12}H_{22}O_5$	246.3001		247	238	203	174	F. glaucescens [53]
37	Ribonucleoside composite of adenine (purine)	Adenosine	$C_{10}H_{13}N_5O_4$	267.2413		268	136			Lonicera japonica [62]
38	Omega-3 fatty acid; octadecate- traenoic acid	Stearidonic acid [6,9,12,15-Octadecatetraenoic acid; Moroctic acid]	$C_{18}H_{28}O_2$	276.4137		277	259; 177	177		Salviae Miltiorrhizae [102]; G. linguiforme [53]; Rhus coriaria [61]
39	Omega-3 fatty acid	Linolenic acid (Alpha-Linolenic acid; Linolenate)	C ₁₈ H ₃₀ O ₂	278.4296		279	243; 173	173	131	Salviae [102]; rice [91]; Pinus silvestris [103]
40	Diterpenoid	Isocryptotanshinone II	$C_{19}H_{20}O_3$	296.3603		297	279; 197	173		Salviae Miltiorrhizae [102]

Table A1. Cont.

No.	Class	Compound	Molecular Formula	Calculated Mass	Molecular Ion [M–H] [_]	Molecular Ion [M+H] ⁺	Fragmenation Ion MS2	Fragmentation Ion MS3	Fragmentation Ion MS4	References
41	Alpha-omega dicarboxylic acid	Octadecanedioic acid [1,16-Hexadecanedicarboxylic acid]	$C_{18}H_{34}O_4$	314.4602	313		295; 183	293; 179	275; 177	F. glaucescens [53]
42	Unsaturated essential fatty acid	Oxo-eicosatetraenoic acid	$C_{20}H_{30}O_3$	318.4504		319	301	186		F. potsii [53]
43	Oxylipin	13- Trihydroxy-Octadecenoic acid [THODE]	$C_{18}H_{34}O_5$	330.4596	329		171; 211; 293	153		<i>Bituminaria</i> [25]; <i>Broccoli</i> [26]; Sasa veitchii [27]
44	Oxylipin	9,12,13- Trihydroxy- <i>trans</i> -10-octadecenoic acid	C ₁₈ H ₃₄ O ₅	330.4596	329		171; 229	127		Potato leaves [56]
45	Unsaturated essential fatty acid	Eicosatetraenedioic acid	$C_{20}H_{30}O_4$	334.4498		335	321; 124	291		G. linguiforme [53]
46	Isoquinoline alkaloid	Berberine [Berberin; Umbelletine; Berbericine]	C ₂₀ H ₁₈ NO ₄	336.3612		337	321; 225	291	291	Tinospora cordifolia [104,105]
47	Pentacyclic diterpenoid	Gibberellic acid	C ₁₉ H ₂₂ O ₆	346.3744		347	345; 259	329; 173	289	Triticum aestivum [106]
48	Berberine alkaloid	Palmatine [Berbericinine; Burasaine]	C ₂₁ H ₂₂ NO ₄	352.4037		353	337; 163	308	293	Ocotea [107,108]
49	Androgen; anabolic steroid	Vebonol	C ₃₀ H ₄₄ O ₃	452.6686		453	435; 336; 209	336; 226	209	Rhus coriaria [61]; Hylosereus polyrhizus [109]
50	Triterpenoid	Oleanoic acid	C ₃₀ H ₄₈ O ₃	456.7003		457	411; 249; 183	227; 169		Pear [110]; Ocimum [82]
51	Triterpenoid	Maslinic acid	$C_{30}H_{48}O_4$	472.6997		473	425; 319; 201	291		Pear [110]; Folium Eriobotryae [111]; Malus domestica [112]
52	Thromboxane receptor antagonist	Vapiprost	C ₃₀ H ₃₉ NO ₄	477.6350		478	460; 337; 263; 155	263; 155	245; 189; 111	Rhus coriaria [61]; Hylosereus polyrhizus [109]
53	Indole sesquiterpene alkaloid	Sespendole	C ₃₃ H ₄₅ NO ₄	519.7147		520	184	184; 125		Rhus coriaria [61]; Hylosereus polyrhizus [109]
54	2- arylbenzofuran flavonoid	Lithospermic acid A	C ₂₇ H ₂₂ O ₁₂	538.4564		539	521; 409; 340; 241	395; 252; 167		Mentha [79,93,94]; Salvia multiorrizae [52]
55	Carotenoid	(all-E)-lutein 3'-O-myristate	C ₄₀ H ₅₄ O	550.8562		551	533; 505; 469;429; 373; 345	453; 410		Carotenoids [113]
56	Triterpenoid	3-O-glucuronide-29-hydroxyoleanolic acid	$C_{35}H_{52}O_{11}$	648.7808		649	473; 367; 291; 229	456; 385; 269	408; 305; 262; 187	Bougainvillea [75]

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