

SCIENTIFIC REPORTS

OPEN

Contrasting methane emissions from upstream and downstream rivers and their associated subtropical reservoir in eastern China

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Subtropical reservoirs are an important source of atmospheric methane (CH_4). This study investigated the spatiotemporal variability of bubble and diffusive CH_4 emissions from a subtropical reservoir, including its upstream and downstream rivers, in eastern China. There was no obvious seasonal variation in CH_4 emissions from the main reservoir, which increased slightly from the first half year to the next half year. In the upstream river, CH_4 emissions were low from February to June and fluctuated widely from July to January due to bubble activity. In the downstream river, CH_4 emissions were lowest in February, which was possibly influenced by the low streamflow rate from the reservoir ($275 \text{ m}^3 \text{ s}^{-1}$) and a short period of mixing. There was spatial variability in CH_4 emissions, where fluxes were highest from the upstream river ($3.65 \pm 3.24 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$) and lowest from the main reservoir ($0.082 \pm 0.061 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$), and emissions from the downstream river were $0.49 \pm 0.20 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$. Inflow rivers are hot spots in bubble CH_4 emissions that should be examined using field-sampling strategies. This study will improve the accuracy of current and future estimations of CH_4 emissions from hydroelectric systems and will help guide mitigation strategies for greenhouse gas emissions.

Hydropower has historically been regarded as a clean energy source, however, the view is challenged by a growing body of research that considers hydroelectric reservoirs to be carbon sources. For example, Deemer *et al.* (2016) showed that CH_4 emissions were responsible for the majority of the radiative forcing from reservoir water surfaces, totalling approximately 80% over a 100-year timescale¹. Greenhouse gas emission data are available for 36 Asian reservoirs, of which CH_4 emission flux data have been reported for three reservoirs in China, including Three Gorges^{2,3}, Ertan⁴, and Miyun⁵. However, there are more than 98,000 dams of varying sizes and 142 large-size hydroelectric reservoirs in a range of geographical regions and climate zones in China, excluding dams that are either under construction or planned, from which CH_4 emission fluxes remain to be assessed.

Diffusive flux and gas bubble flux are the primary pathways for CH_4 emissions from open water areas of reservoirs⁶. Ebullition has been shown to be the dominant CH_4 emission pathway, albeit episodic⁷, but pulses of gas bubbles often occur during periods of rapidly falling barometric pressure in lakes, reservoirs, and peatland^{7–10}. Ebullitive CH_4 flux is reported to be 1–3 orders of magnitude greater than diffusive CH_4 flux^{11,12}, and high ebullitive CH_4 flux, observed in shallow water, river deltas, and inflow rivers^{11–13}, is shown to be influenced by allochthonous organic carbon input and burial¹⁴. Chamber methods were used to measure CH_4 emission flux from three large reservoirs in China, where the total CH_4 emission flux (diffusion + ebullition) was measured across the water-air interface^{2–5}, however, it is likely that these studies did not capture the magnitude of bubble CH_4 flux.

Subtropical reservoirs are strong atmospheric CH_4 sources with strong spatial variability¹⁵. Such variations are presumably caused by changes in hydrological characteristics from impoundment. For example, increases in the water level, reduced water velocity, and flooded soils near the bank impact CH_4 emissions from a new reservoir compared with the original river^{7,16}. Similarly, CH_4 emission levels from outlets downstream and inflow rivers upstream of reservoirs were distinct from those of the reservoir water^{7,17,18}, due to variability in hydrological variables, such as water velocity and depth², and dam operation strategy¹⁹. Temporal variability in CH_4 emissions has been attributed to changes in temperature, water column mixing, dissolved oxygen (DO) concentration, and other environmental variables, including retention time and benthic metabolism^{2,20,21}. For example, CH_4

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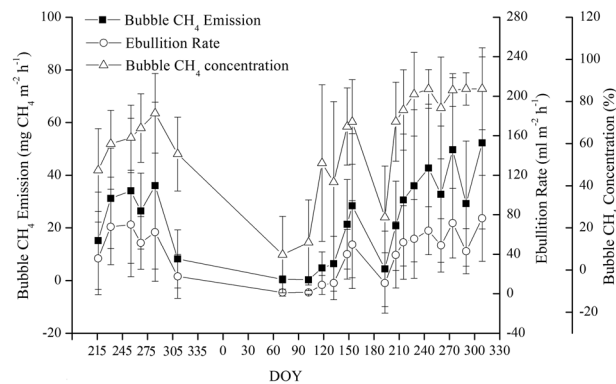


Figure 1. Mean ebullition rates, bubble CH_4 emissions flux, and CH_4 concentrations recorded from the inflow river. DOY: day of year, from 3 August 2016. Bars are \pm SE, $n = 3$.

emissions were greater in summer than in other seasons at the Three Gorges Reservoir and were regulated by temperature, DO, and water velocity², whereas they were only regulated by temperature at three lakes (Följesjön, Erssjön, and Skottenesjön) in southwest Sweden²⁰. Analysis of these differences in effects of environmental factors on spatiotemporal variability in CH_4 emissions from reservoirs may result in more accurate estimates of the total CH_4 emissions than previously determined.

Emissions occur from rivers downstream of reservoirs, due to degassing fluxes at turbines and spillways. A large quantity of CH_4 emits in the downstream river when the hypolimnion water passes through turbines and spillways because of the differences in temperature and pressure. The rapid stream of water increases the water current velocity, which enhances the gas transfer velocity at the air-water interface and improves downstream CH_4 emission flux²². 50% of the total CH_4 emissions recorded downstream from the Balbina Reservoir in Brazil¹⁸ represented approximately 30% of the total greenhouse gas emissions from the eight reservoirs in the dry tropical biome region of the country²³, whereas downstream emissions accounted for 10% of the total CH_4 emissions from the Nam Theun 2 Reservoir in Laos²⁴.

In this study, we compared CH_4 emissions from a reservoir with sites upstream and downstream to quantify spatial variations in CH_4 emissions to ensure a more accurate estimation of CH_4 emissions from hydroelectric reservoir systems. Specifically, we tested the hypothesis that upstream and downstream CH_4 emissions are greater than from a reservoir.

Results

Temporal variation in ebullitive CH_4 emissions. There were similar seasonal patterns of ebullition rate, bubble CH_4 emission flux, and bubble CH_4 concentration, all of which were lower in spring than in summer and autumn (Fig. 1). Mean ebullition rate from the upstream river was $39.93 \pm 24.3 \text{ ml m}^{-2} \text{ h}^{-1}$ (range: 1.17–76.4 $\text{ml m}^{-2} \text{ h}^{-1}$), mean bubble CH_4 flux rate was $22.62 \pm 15.1 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$ (range: 0.31–52.27 $\text{mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$), and mean CH_4 concentration by volume in the collected gas was $59.04 \pm 23.3\%$ (range: 7.32–86.03%). Ebullitive CH_4 flux positively correlated with the ebullition rate ($R^2 = 0.92$, $P < 0.001$) and bubble CH_4 concentration ($R^2 = 0.76$, $P < 0.001$, see Supplementary Fig. S3).

Temporal variation in diffusive CH_4 emissions. CH_4 emissions from the upstream river (NW) were low from February to June, but increased and fluctuated from July to January (Fig. 2). Furthermore, on a monthly scale, mean diffusive CH_4 fluxes during the sampling period were similar and generally constant over time among the three areas of the main reservoir; however, fluxes peaked in the southwest (SW) lake on 1 August (DOY: 213) and 8 February (DOY: 39; Fig. 2). On a seasonal scale, there were similar seasonal patterns in CH_4 fluxes among the three areas of the reservoir, where they were lowest in the spring and highest in the autumn (see Supplementary Fig. S4). Mean CH_4 fluxes on the northeast (NE), SW, and southeast (SE) lakes in the next half year were 1.72, 1.54, 1.57 times as many as those in the first half of year, respectively. In addition, there was some temporal variation in mean CH_4 emissions downstream of the reservoir (DR), where it was highest in December 2014 and lowest in February 2015; otherwise, emissions were generally constant (Fig. 2).

Spatial variation in CH_4 emissions. Mean flux in CH_4 emissions from the upstream river was $3.65 \pm 3.24 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$, whereas mean bubble CH_4 flux (NW-B) was $2.73 \pm 2.02 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$ and diffusive CH_4 flux (NW-D) was $0.92 \pm 1.22 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$ (Fig. 3). Although there were no bubble CH_4 emissions in the reservoir or the downstream river, the mean diffusive CH_4 emission flux in the reservoir was $0.082 \pm 0.061 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$ (NE: $0.076 \pm 0.049 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$, SW: $0.106 \pm 0.083 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$, and SE: $0.064 \pm 0.034 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$), which was lower than in the downstream river, where it was $0.49 \pm 0.20 \text{ mg m}^{-2} \text{ h}^{-1}$ (Fig. 3). Mean diffusive CH_4 emissions from the upstream and downstream rivers were higher than those from the reservoir by a factor of 11 and 6, respectively (Fig. 3).

There was no significant difference in mean CH_4 emissions from the marginal to pelagic zones among the three sampling areas of the reservoir (see Supplementary Fig. S5C–E); however, the mean CH_4 emissions from the nearest sampling-point in the downstream river (DRP1: $0.78 \pm 0.44 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$) were significantly

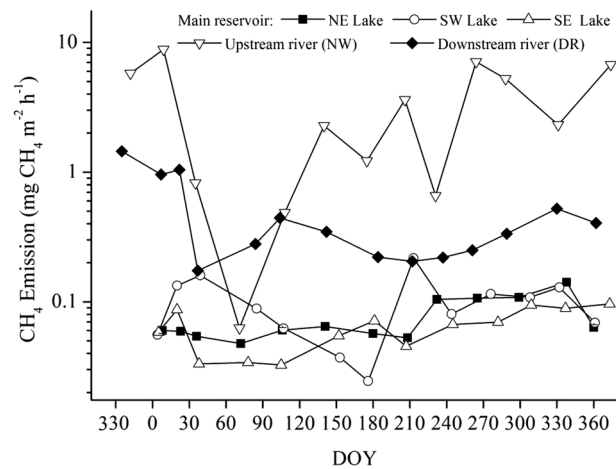


Figure 2. Seasonal dynamics in the average diffusive CH_4 emissions, measured using floating chambers, from the different regions of Xin'anjiang Reservoir.

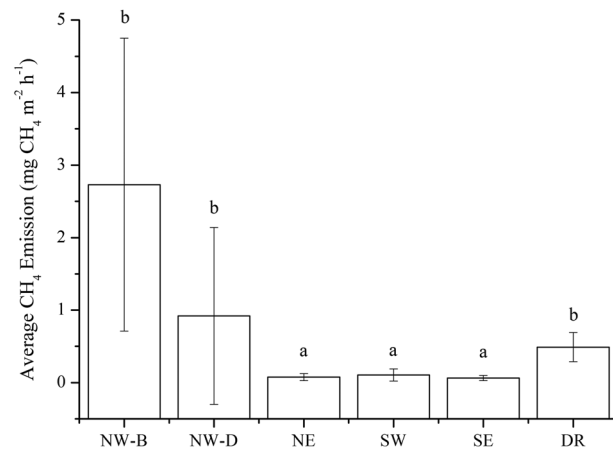


Figure 3. Mean CH_4 emissions from the reservoir and the upstream and downstream rivers. NW-B, bubble emissions from the northwest transect (upstream); NW-D: diffusive emissions from the northwest transect. NE, northeast lake; SW, southwest lake; SE, southeast lake; DR, downstream river. Different small letters indicate the differences in mean CH_4 emissions flux among the sampling areas at $P=0.05$.

higher than those from the second nearest sampling-point (DRP2: $0.34 \pm 0.30 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$; $P < 0.001$; see Supplementary Fig. S5A), and the average CH_4 emissions from the pelagic zones of the upstream river were significantly higher than those from the marginal zone (see Supplementary Fig. S5B).

Effects of temperature and wind speed on CH_4 emissions. CH_4 flux from the reservoir was positively correlated with wind speed and air-water temperature difference, whereas CH_4 flux from the downstream river was positively correlated with air-water temperature difference (see Supplementary Tables S1 and S2).

Discussion

Comparison of CH_4 emissions with other reservoirs. Average CH_4 emissions from the main reservoir ($0.082 \pm 0.061 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$) are lower than those from the other temperate and subtropical reservoirs listed in Table 1, except for Douglas Lake, which is presumably due to the deep, oxic conditions and clean water quality in Xin'anjiang Reservoir^{25,26}. The mean CH_4 emissions from the upstream river in the study are comparable to that from Three Gorges Reservoir ($2.72 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$), which is one order of magnitude greater than that from Eguzon Reservoir ($0.24 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$), but significantly lower than those from Australian reservoirs and an agriculturally impacted reservoir in the United States, due to the differences in bubble activity (Table 1). The heterogeneity, specifically, differences in ebullition frequency and ebullition magnitudes, contribute to the variability in average CH_4 fluxes observed among the reservoirs¹². Frequency of bubble occurrence upstream of the reservoir is low (16.2%), and the average ebullitive CH_4 emission level is $16.83 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$ (Table S4), which is one order of magnitude lower than the ebullition magnitudes in William H. Harsha Lake ($130.7 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$), Gold Creek ($172.4 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$), and Little Nerang Dam ($165.7 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$; Table 1). A large number of bubbles contribute to the extremely high CH_4 emissions from the inflow rivers in the three reservoirs^{15,27–29}.

Country	Reservoir	CH ₄ Flux (mg CH ₄ m ⁻² h ⁻¹)			Refs
		Upstream river*	Open water area	Downstream river	
China	Xin'anjiang	2.73 ± 2.02 (B) 0.92 ± 1.22 (D)	0.082 ± 0.061	0.49 ± 0.20	This study
	Three Gorges	2.72 ± 1.98	0.23 ± 0.40	0.26 ± 0.16	2,3
	Ertan		0.12 ± 0.063		4
	Miyun		0.30 ± 0.31		5
	16 reservoirs in Chongqing		0.63 ± 0.89		50
USA	William H. Harsha Lake	130.72 ± 27.50	9.77 ± 2.00		27
	Douglas Lake	0.018 (D)	0.017 ± 0.012		51
	Eagle Creek		0.44 ± 0.73		52
	6 reservoirs in Western US		0.13–0.40		53
Australia	Gold Creek	172.36 ± 24.72	12.35 ± 6.36		28
	Little Nerang Dam	165.70 ± 236.43	7.70 ± 19.38		29
Laos	Nam Leuk		1.68 ± 2.68		54
	Nam Ngum		0.13 ± 0.13		54
	Nam Theun 2		1.0–2.67	Below the powerhouse: 8.0 ± 14.7 Below the Nakai Dam: 0.93–2.2	24,33
France	Eguzon	0.24 ± 0.56 (B) 2.2 ± 3.2 (D)	0.4 (0–2.67)	0.68 ± 0.68	55

Table 1. Literature review of CH₄ emissions from temperate and subtropical reservoirs. *CH₄ flux in upstream river: B: Bubble emission, D: Diffusive emission.

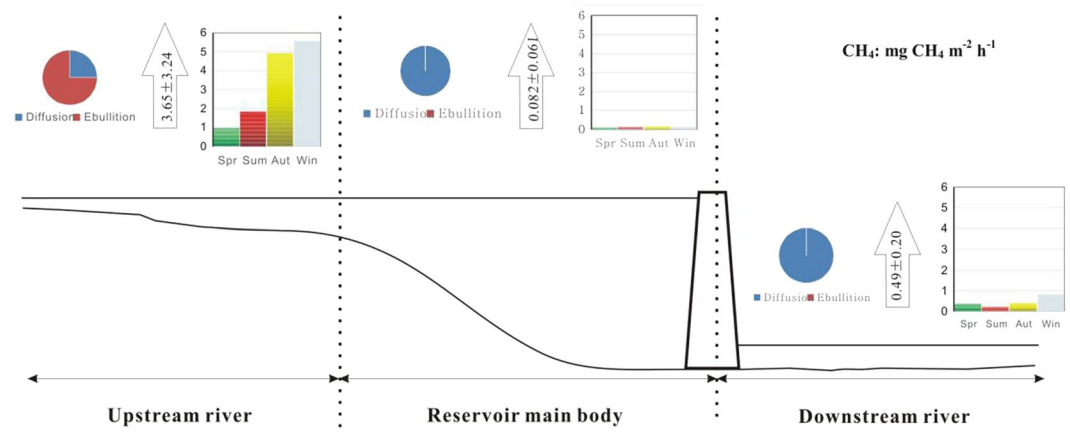


Figure 4. Schematic diagram of the spatiotemporal variability in CH₄ emissions from Xin'anjiang Reservoir.

In regard to the downstream river of the reservoir, it has comparable CH₄ emissions levels with the other listed reservoirs in Table 1.

Seasonal variation in CH₄ emissions. In the upstream river, CH₄ emissions in autumn and winter are higher than those in spring and summer (Fig. 4), due to the differences in the frequency of bubbles (22.6% versus 8%), but the differences do not reach a significant level ($p > 0.05$) after performing a one-way ANOVA test. However, the results measured by the bubble traps indicate that the bubble CH₄ emissions in summer and autumn are significantly higher than those in spring (Fig. 1). One of the major differences between the two methods is the duration of the measurement. The measurements using bubble traps were performed over 20–33-h periods, whereas chamber measurements were conducted for 20–30 min only. The floating chambers captured both ebullition and diffusive gas emissions²⁷, whereas only CH₄ ebullition fluxes were collected using bubble traps⁸. However, the average ebullitive CH₄ flux (22.62 ± 15.1 mg CH₄ m⁻² h⁻¹) measured using bubble traps was approximately 5 times higher than that measured using floating chambers (3.65 ± 3.24 mg CH₄ m⁻² h⁻¹). These differences can be explained by the sudden release of bubbles on these rare occasions, which reveals strong spatiotemporal heterogeneities of the ebullition process because ebullition is highly sporadic and occurs during a very short period of time⁷. The measurements using floating chambers are conducted over a short period of time and a small surface might lead to an underestimation of this emission pathway if hot spots and hot moments are missed during the deployment of the chambers. Such a phenomenon is strongly smoothed when using bubble traps over longer periods of time than the typical floating chamber deployment time (20–33 h versus 20–30 min)³⁰.

Another explanation for the differences in CH₄ emissions from the upstream river is that they were measured in different years (2014–2015 versus 2016–2017). Admittedly, interannual variability in upstream CH₄ emissions presumably caused unnecessary errors. However, if the average CH₄ flux is calculated only from these bubble-captured chambers in the NW transect in 2014 and 2015, it is $16.83 \pm 12.48 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$, which is approximately 25% less than that measured by the bubble traps ($22.62 \pm 15.1 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$) in 2016 and 2017. Although both the interannual variability and different methods contributed to variances, differences remained in CH₄ emissions when the diffusive and ebullitive CH₄ fluxes were synchronously measured due to episodic bubbles. Nevertheless, the partitioning of bubble and diffusive CH₄ emissions is an uncertainty in this study.

We observed that mean CH₄ emissions from the reservoir in the second half of the year were higher than in the first half of the year (Fig. 2); this was due, in part, to seasonal hydrological dynamics. Hydrology mediates many biogeochemical processes, such as O₂ concentration and thermal stratification, in aquatic systems. Zhang *et al.* (2015) found that oxycline and thermocline progressively sank in Xin'anjiang Reservoir in the second half of the year²⁵. Vertical transport of CH₄ in the water column is typically limited by slow rates of diffusion through the thermocline or oxycline³¹, and thermal and DO stratification typically become weaker in the second half of year, presumably resulting in increases in CH₄ flux at the air–water interface.

Xin'anjiang Reservoir is a thermal stratification lake characterised by a short mixing period in February and March²⁵. However, the vertical distributions of CH₄ and O₂ concentrations, and temperature were not measured in this study, which failed to illuminate the temporal variability in CH₄ emissions. Lake overturn is a hot moment that exhibits disproportionately high CH₄ emissions and CH₄ oxidation³². Many studies indicate that CH₄ storage sharply decreases during seasonal overturn periods^{32–35}, emitting 12–46% of the total CH₄ to the atmosphere, whereas the remainder (54–88%) is consumed by methane-oxidizing bacteria^{32,34,35}. Although a minor proportion of the storage CH₄ was emitted to the atmosphere, the contribution to the annual diffusive CH₄ emissions was still great^{32,34,35}, and even extremely diffusive CH₄ fluxes occurred³³. However, CH₄ emissions from the main reservoir did not show a pulse in February and May (Fig. 2), presumably because the low measurement frequency in our study did not capture the CH₄ emission peaks. Thermal stratification and its impact on CH₄ emissions is important to understanding the mechanisms of the spatial and temporal variability of CH₄ emissions from reservoirs, which should be examined in the near future.

We recorded a clear peak in CH₄ emissions ($0.25 \pm 0.15 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$) on 1 August (DOY: 213) in the SW lake (Fig. 2), which was presumably due to the fluxes from the two marginal sampling points (SWP1 and SWP2) of $0.47 \pm 0.11 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$ and $0.33 \pm 0.061 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$, respectively (see Supplementary Table S7). The high CH₄ fluxes from the marginal zone may be attributed to the decomposition of vegetation in the littoral zone when the water level increased to its highest point (104.4 m) in July (see Supplementary Fig. S2). It is likely that the gentle slopes that had adequate levels of soil on the banks of the SW transect permitted the growth of vegetation in the littoral zone during the spring when water levels were low, whereas the banks of the NE and SE lakes were steep and rocky and presumably less well vegetated. Similar peaks in CH₄ emissions have also been reported from littoral zones of the Miyun and Three Gorges Reservoirs^{5,36}.

We recorded another peak, albeit low ($0.16 \pm 0.097 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$) on 8 February (DOY: 39) in the SW lake (Fig. 2), which was caused by strong winds. Gas samples were only collected from three of the five sampling points due to unstable safety conditions on the surface of the reservoir. Mean CH₄ fluxes were 0.23 and 0.20 mg CH₄ m⁻² h⁻¹ at SWD2 and SWD4 (see Materials and Methods), respectively, when the wind speed reached 8–10 m s⁻¹, whereas the lowest CH₄ flux at SWD5 ($0.049 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$) occurred in the central area of the reservoir due to the low wind speed (2.63 m s⁻¹). Many studies support the opinion that CH₄ emissions from water surfaces can be enhanced by strong wind speeds^{15,22,33,37}.

Downstream CH₄ emissions (including degassing at the turbines) have been found to be proportional to streamflow¹⁹. It is impossible to calculate the degassing emissions from the turbines at Xing'anjiang Dam based on the differences in CH₄ concentrations between the water intake and water outlet below the dam because access is forbidden 500 m upstream and downstream of the dam due to safety concerns. However, measurements of CH₄ emissions at four distances downstream of the dam, taken 13 times in 2015 (see Supplemental Table S9), were found to be at their lowest ($0.17 \pm 0.11 \text{ mg CH}_4 \text{ m}^{-2} \text{ h}^{-1}$) in February, which is presumably a result of a low discharge flow rate ($275 \text{ m}^3 \text{ s}^{-1}$). Another possible explanation for the low flux is related to the lake overturn phenomenon in February²⁵. Most of the CH₄ stored in the hypolimnion is oxidized or released to the atmosphere during overturn periods^{32–35}, and a very small fraction of the original quantity of CH₄ remains in the water column³²; thus, a low CH₄ flux level was measured in the downstream river in February (Fig. 2).

Spatial variation in CH₄ emissions. Upstream CH₄ emissions are hot spots because they exhibit disproportionately high ebullitive CH₄ emissions relative to the surrounding matrix³⁸. Upstream river CH₄ emission dynamics are predominantly influenced by bubbles since the peaks in the CH₄ emissions flux (Fig. 2) are driven by bubbles (see Supplementary Table S4). In contrast to other studies^{21,24}, we found that bubbles occurred in the deep-water zone (>10 m) rather than in the shallow zone (<5 m), and we suggest that the high ebullitive CH₄ emissions from deep water zones are related to heterogeneous sediment accumulation^{12,13} because little or no sediment accumulates along reservoir margins³⁹.

The average CH₄ emission rate at the upstream site (NW) was one to 2 orders of magnitude greater than the other sites (Figs 2 and 3), highlighting the importance of identifying ebullition hot spots to improve total emissions estimates²⁷. The results supported our hypothesis that CH₄ emissions are higher in rivers upstream and downstream of the reservoir than in the main reservoir (Fig. 3), where high CH₄ emissions from the upstream river were mediated by bubbles (Figs 1 and 3, see Supplementary Table S4). The CH₄ in the gas bubbles can escape oxidation during transport through the water column as CH₄ moves faster through the water column by ebullition than by diffusion⁴⁰.

location	observations	ref
Three Gorges Reservoir, China	Upstream, reservoir tail waters and tributary sites had higher CH ₄ fluxes than the mainstream of the reservoir.	3
Lake Kariba, Zambia/Zimbabwe	Higher fluxes in river deltas (~10 ³ mg CH ₄ m ⁻² d ⁻¹) than nonriver bay (less than 100 mg CH ₄ m ⁻² d ⁻¹) due to the high ebullition frequency and ebullition magnitudes.	27
Little Nerang Dam, Lake Wivenhoe, Lake Baroon, Australia	CH ₄ saturation was higher in inflow zones than in the main body.	27
William H. Harsha Lake, USA	Extreme high CH ₄ emission (mean: 3137 ± 660 mg CH ₄ m ⁻² d ⁻¹) at the most upstream site; 1 to 2 order of magnitude greater than the other sites.	27
Glod Creek Reservoir, Australia	Highest CH ₄ water-air fluxes were found at the main water inflow areas of the reservoir.	28
Little Nerang Dam, Australia	1.8–7.0% of the upstream surface area called “ebullition zone”; 97% of the total methane occurred in the ebullition zones.	29
Chapéu D’Uvas, Curuá-Una, Furnas, Brazil	Elevated pCH ₄ and CH ₄ concentrations in river inflow areas and decreasing values toward the dam; River inflows are hot spots of diffusive C gas flux.	37

Table 2. Some examples of studies reporting high methane emissions from the upstream inflow areas of reservoir.

Fluxes in CH₄ ebullition in inflow water systems are common in the other reservoirs (Table 2), which may be attributable to the fact that water slows down in these areas and sediments have higher chances for deposition¹⁴. Sediment accumulation rates are positively correlated to the areal organic carbon burial rates³⁹, and rapid burial of fresh sediments and organic matter made upstream sites more carbon rich and prime for CH₄ production by anaerobic metabolism compared to other parts of the reservoirs^{12,15,27,37}, as CH₄ production in reservoirs is strongly driven by organic carbon availability⁴¹. Thus, ebullitive CH₄ emissions are often reported to be exponentially increased with corresponding sediment accumulation rates^{14,42}. The upstream reaches of Xin’anjiang Reservoir directly receive the catchment and stream inflow of industrial and domestic pollution⁴³, which presumably fostered high rates of sediment CH₄ production in the upstream rivers of the reservoir, causing ebullition zones to subsequently appear^{27,29}. Moreover, ebullition rates tend to be highest in shallow areas because short water residence times limit the dissolution of CH₄-rich bubbles released from the sediment⁴⁴. The upstream river is the shallowest area compared with other regions (see Supplementary Table S3), which is beneficial for bubbles transport from the sediment to the atmosphere because of the small proportion of dissolved gas bubbles during ascent^{15,27,37}. Additionally, CH₄ imported from the Xin’anjiang catchment may further contribute to the observed pattern at river inflow areas.

We recorded higher CH₄ emissions from the downstream river than from the surface of the reservoir adjacent to the dam (Fig. 2) that had presumably been released from dissolved CH₄ in the hypolimnion layer of the reservoir¹⁷ because water inlets of turbines located in the hypolimnion layer (26–37 m under water surface)⁴³ and the discharged water derived from the hypolimnion layer almost year round (except February, due to mixing periods). The water adjacent to the dam is thermally stratified, where water in the warmer, upper layer (epilimnion <33 m) is in contact with the atmosphere and is more oxygen-rich, whereas the deeper, colder layer (hypolimnion) contains relatively low levels of O₂ concentration²⁵. We suggest that CH₄ produced in the reservoir is easily stored in the hypolimnion⁴⁵, and the release of dissolved CH₄ to the atmosphere occurs due to differences in pressure, temperature, and turbulence when water passes through the turbines and spillways¹⁹. Water passing through the turbines and spillways is drawn from the hypolimnion, and downstream CH₄ emissions are released under decreased pressure below the dam¹⁹.

The explanation for the low CH₄ emissions from the main reservoir is that the deep, oxic waterbody slows emissions by offering more options for CH₄ oxidation. Water depths of the sampling points range from 10–69 m, except for those on the margin (Table S3), and it is possible that such reservoir depths increase the possibility of oxidization for diffusive CH₄ molecules. Moreover, Zhang *et al.* (2015) reported that the DO concentration never fell below 2 mg/L, the critical value for anoxia, in Xin’anjiang Reservoir²⁵. The lack of an anoxic layer permits the oxidization of dissolved CH₄ under aerobic conditions by methanotrophic bacteria²⁷. Furthermore, biomass clearing before flooding limited the availability of organic carbon^{26,43}, which is important for CH₄ production in sediments⁴¹. Chlorophyll a is a significant predictor of CH₄ emissions from reservoir water surfaces^{1,37}, and Xin’anjiang Reservoir is presently in an oligotrophic state, with a low concentration of chlorophyll a (1–3 µg L⁻¹)²⁶, which limits CH₄ emissions from the reservoir. Moreover, the dendritic shape of Xin’anjiang Reservoir facilitates the deposition of allochthonous organic carbon in the sediment of the NW lake (see Supplementary Fig. S1)⁴⁶, and limited fresh sediments are deposited in the main reservoir.

Mitigation strategies for CH₄ emissions. Management strategies should increase CH₄ oxidation in the sediments and water columns and decrease CH₄ production, ebullition, and degassing emission at the dam to mitigate CH₄ emissions from reservoirs. Extremely allochthonous organic material and organic carbon burial stimulated ebullition in the upstream rivers and river deltas^{12,14}, therefore, periodical dredge campaigns²⁷, reducing watershed soil erosion¹⁴ and nutrient input⁴⁷, can efficiently reduce ebullitive CH₄ emissions. Moreover, the location of spillways and turbines have an impact on CH₄ emissions from reservoirs²⁷.

Previous studies have shown that extreme CH₄ ebullitive emissions are ultimately attributable to very high sedimentation rates¹⁴, as well as exhibiting an exponentially increasing relationship between CH₄ ebullitive

emissions and the sediment accumulated rates in the 6 small reservoirs of the Saar River⁴². The mechanism is characterised by deeper sediment layers contributing to CH₄ formation⁴², and the deeply accumulated CH₄ causes supersaturation and consequent bubble formation and release¹⁴. In the study, inflow rivers are ebullition hot spots, thus policymakers should take effective measures to control substantial CH₄ emissions. The sediment is dredged periodically to reduce deposited organic matter, which presumably decreases the magnitude of ebullitive CH₄ emissions efficiently, although carbon leakage occurs during the process²⁷.

Another practical measure is to prevent the excessive input of nutrients and pollution to the reservoir^{30,47,48}, which would reduce the available organic carbon for CH₄ production⁴⁷. Cage culture is an important nutrient input, which enhances N, P, and TOC accumulation in the sediments of the lacustrine zone⁴⁸. Moreover, the NW lake received more soil erosion, sewage input, and industrial pollution from the upstream rivers in Anhui Province. In response, authorities have taken measures to decrease the inputs of all types of pollution, such as cage culture prohibition and inter-provincial ecological compensation, which improved the water quality in the upstream river from a eutrophic to mesotrophic state, presumably decreasing CH₄ production and emissions from the reservoir³⁷.

Dam design is also important for CH₄ emissions, especially the location of water intakes. CH₄ concentrations are higher in the hypolimnion than in the epilimnion during thermal stratification periods¹⁹. The degassing that occurs as hypolimnion water is routed through a dam accounts for a large fraction (>50%) of the total CH₄ emissions in some Amazon tropical reservoirs^{17,18}. However, if turbine intakes are located in the upper layer of a dam, shallow waters will be withdrawn during thermal stratification to avoid substantial CH₄ degassing from the CH₄-rich water in the hypolimnion¹⁹, for example, only 0.8% from Harsha Lake²⁷. Moreover, a significant increase in CH₄ emissions was reported 3 km upstream from Nam Theum 2 Dam due to the artificial mixing induced by water intakes³³, and CH₄-rich water from the reservoir's hypolimnion reached the surface and resulted in a high CH₄ diffusive flux. Therefore, the water intake in the hypolimnion not only increased the degassing flux at the dam but also risked enhancing the CH₄ diffusive flux upstream of the dam.

In summary, upstream rivers are hot spots in bubble CH₄ emissions, significantly contributing to the total CH₄ emissions from hydroelectric reservoir systems. If upstream sites are ignored in field-sampling strategies, entire-system CH₄ emissions will be underestimated. CH₄ emissions from a main reservoir are lower than that from a downstream river. Capturing the spatial heterogeneity of CH₄ emissions is vital to estimating the total CH₄ emissions in a hydroelectric system. Seasonal variation in CH₄ emissions exhibited a high value in autumn and winter and a low value in spring and summer. A thorough investigation should be conducted for the entire reservoir region over a long period because bubbles are episodic and diffusive CH₄ emission flux exhibits a strong spatiotemporal variability.

Materials and Methods

Study sites. Xin'anjiang Reservoir (118°42'–118°59'E, 29°28'–29°58'N) is located in China's north subtropical zone. The mean annual air temperature, precipitation, and evaporation are 17.7 °C, 2015.1 mm, and 712.9 mm, respectively (see Supplementary Fig. S2). Constructed in 1959, the reservoir has a water surface area of 580 km² and mean depth of 37 m, with a capacity of approximately 1.78×10^{10} m³⁴³, and an annual average inflow and outflow of 9.4×10^9 m³ and 9.1×10^9 m³, respectively. Water retention time is approximately 2 years, and in 2015, the water level fluctuated between 98 and 104 m above elevation (see Supplementary Fig. S2). According to China's surface water classification standards, the water quality of Xin'anjiang Reservoir is grade I, serving as an important water source in eastern China that presently provides drinking water.

The reservoir consists of a series of connected lakes in all cardinal directions around a central lake that serves as the main waterbody (see Supplementary Fig. S1). The watercourse of the northwest lake is the dominant source of upstream inflow, contributing 60–80% of the total inflow. The downstream river is the watercourse below Xin'anjiang Dam.

The four sub-lakes and downstream river were sampled at points along transects (see Supplementary Fig. S1). The NW lake transect (118°43'04"E, 29°44'03"N), located in the main upstream inflow inlet, has a width of 0.3 km and three sampling points extending 10, 50, and 120 m (NWP1, NWP2, and NWP3, respectively) from the southern bank marginal zone to the pelagic zone, whereas the NE (119°03'03"E, 29°38'44"N), SW (118°44'39"E, 29°28'18"N), and SE (118°45'20"E, 29°28'39"N) lake transects are located in the open water and have five sampling points (P1 to P5) extending from the marginal to pelagic zones (Table S3). Four sampling points in the downstream river below the dam are located 0.35, 1, 4, and 7 km from Xin'anjiang Dam (DRP1, DRP2, DRP3, and DRP4, respectively).

CH₄ flux measurement. Floating static chambers were used to collect gas samples at all sampling points between 08:30 and 11:30 hrs, monthly from December 2014 to December 2015, and bubble traps were used to collect bubbles from the upstream river from August 2016 to November 2017, where samples were collected once or twice per month, except November 2016, and January and February 2017. Air and water temperatures were measured using an alcohol thermometer, and wind speed in the field was measured using an anemometer (Kestrel 1000, Nielsen-Kellerman Co., USA).

Flux of diffusive CH₄ emissions was collected using floating static chambers and analysed by gas chromatograph. Three floating static chambers (basal area of 0.29 m² and volume of 0.117 m³) at each sampling point comprised a non-covered plastic box wrapped in light-reflecting and heatproof materials to minimize internal temperature variation, with plastic foam collars fixed to opposite sides. The headspace height inside the chamber was approximately 35 cm. A silicone tube (0.6 and 0.4 cm outer and inner diameters, respectively) was inserted into the upper central side of the chamber to collect gas samples that were then dried to prevent biological reactions in plexiglass tubes filled with calcium chloride (anhydrous, analytical reagent). Another silicone tube was inserted into the upper corner of the chamber to maintain a balance in air pressure between the inside and outside

of the chamber. Static chambers drifted freely behind a boat to reduce measurement bias⁴⁹. Samples of gas were collected from the static chamber in air-sampling bags (0.5 L, Hedetech, Dalian, China) four times every 7 min over a 21-min period using a hand-driven pump (NMP830KNDC, KNF Group, Freiburg, Germany) and were stored until analysis². The air-sampling bags made of aluminium are suitable for gas storage for 7 days and do not absorb or react with CH₄. Leakage and memory effects of the air-sampling bags were tested in earlier experiments.

We placed 16–26 bubble traps 10–15 m apart in a river crossing rope in the upstream river, where water depth ranged from 5–25 m. The traps consisted of an inverted 30-cm diameter circular funnel fixed to the neck of a 0.56-L plastic bottle, and an additional skirt (50-cm diameter) was fixed to the funnel aperture to enlarge the area over which bubbles were collected⁸. Each funnel was stabilized with three equally sized weights to ensure no tiny bubbles remained in the traps at the initial stage. Trapped gas bubbles liberated from water were collected in the bottles after 24 hours, and then the remaining volume of water was measured to calculate the volume of liberated gas bubbles. The trapped gas was diluted 1000 times by injecting 1×10^{-3} -L of trapped gas into 1- or 0.5-L gas bags that had been filled with N₂ to facilitate analysis of CH₄ concentration by gas chromatography. Trapped gas within these bags was analysed within 3 days using a gas chromatograph (Agilent 7890 A, Agilent Technologies, Santa Clara, USA) equipped with a flame ionization detector (FID). The oven, injector, and detector temperatures were set at 70, 25, and 200 °C, respectively. Standard mixed gas (CH₄: 1.83 ppm, provided by the China National Research Centre for Certified Reference Materials, Beijing) was used to quantify the CH₄ concentration in one of every 10 samples, and the coefficient of variation of CH₄ concentration in the replicated samples was <1%.

The increasing rate of gas concentration (dc/dt) within the static chamber was calculated as the slope of the linear regression of the gas concentration versus time. Diffusion chambers collect diffusive emissions as well as ebullitive emissions if they are present. Therefore, if the slope of the linear regression of the gas concentration in the chamber versus time was linear, with $R^2 > 0.9$, then the chamber was assumed to collect only diffusive emissions. If $R^2 < 0.9$, then the chamber was assumed to collect total (diffusive + ebullitive) emissions³⁰.

The flux of diffusive CH₄ emissions (F_1 , mg CH₄ m⁻² h⁻¹) is calculated as (Eq. 1):

$$F_1 = \rho \times \frac{dc}{dt} \times \frac{273.15}{273.15 + T} \times H \quad (1)$$

where ρ is the density of gas under the standard conditions (0.714 kg m⁻³ for CH₄), H is the height from the top of the inverted chamber to the water surface (here, 0.35 m), 273.15 is the absolute temperature at 0 °C, and T is the air temperature (°C).

The flux of CH₄ via ebullition (F_2 ; mg CH₄ m⁻² h⁻¹), measured as the bubble CH₄ flux by bubble traps, is calculated as (Eq. 2):

$$F_2 = \frac{C_{CH_4} \times V \times M}{A_f \times t \times V_m} \times \frac{1}{1000} \quad (2)$$

where C_{CH_4} is the CH₄ concentration (μL L⁻¹), V is the accumulated headspace gas volume (L), M is the molar weight of CH₄ (16.04 g mol⁻¹), A_f is the funnel area (0.14 m²), t is the measurement duration (h), and V_m is the molar volume of gas at room temperature (22.4 L mol⁻¹)⁸.

The ebullition rate (ER ; mL m⁻² h⁻¹), which reflects the volume rate of released accumulated bubbles, is calculated as (Eq. 3).

$$ER = \frac{V}{A_f \times t} \quad (3)$$

where the parameters V , A_f , and t are provided in Eq. (2).

Statistical analysis. The flux in CH₄ emissions data that did not meet the test for normality (Kolmogorov-Smirnov) were transformed to trigonometric or logarithmic functions prior to testing for seasonal and spatial variability using one-way analysis of variance (ANOVA) and Tukey's HSD test. Data were analysed using the SPSS statistical package (v. 18.0, Chicago, IL, USA).

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Acknowledgements

The study was funded by the National Natural Science Foundation of China (41303065) and Zhejiang Hangzhou Urban Forest Ecosystem Research Station. We thank Li Hepeng and Yue Chunlei for financial support for the manuscript publication, the Xin'anjiang hydropower plant for provision of streamflow data below the dam, Sun Bin Feng for the photograph in Supplementary Fig. S1, Xu Gaofu for provision of meteorology data presented in Supplementary Fig. S2. Data presented in this work can be found in the supporting information.

Author Contributions

Y. L. designed the experiments, collected the data, and wrote the manuscript.

Additional Information

Supplementary information accompanies this paper at <https://doi.org/10.1038/s41598-019-44470-2>.

Competing Interests: The author declares no competing interests.

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