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2-(1,3-Benzodioxol-5-yl)-3-phenylquinazolin-4(3H)-one

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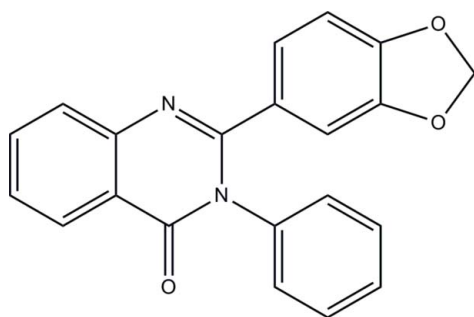
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 Key indicators: single-crystal X-ray study; $T = 273$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.040; wR factor = 0.101; data-to-parameter ratio = 16.0.

In the molecule of the title compound, $\text{C}_{21}\text{H}_{14}\text{N}_2\text{O}_3$, the quinazoline ring system [maximum deviation = 0.076 (1) Å] makes dihedral angles of 40.57 (9) and 42.31 (11)°, respectively, with the phenyl and 1,3-benzodioxole rings. The dihedral angle between the phenyl ring and the 1,3-benzodioxole ring is 4.34 (10)°. In the crystal, $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules into infinite zigzag chains extending along [100].

Related literature

For the biological and pharmaceutical importance of quinazolines, see: Arfan *et al.* (2008); Bartroli *et al.* (1998); Kung *et al.* (1999); Mannschreck *et al.* (1984).



Experimental

Crystal data

 $\text{C}_{21}\text{H}_{14}\text{N}_2\text{O}_3$
 $M_r = 342.34$

 Monoclinic, $P2_1$
 $a = 8.984$ (4) Å
 $b = 6.056$ (3) Å
 $c = 15.248$ (6) Å
 $\beta = 95.357$ (6)°
 $V = 826.0$ (6) Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 273$ K
 $0.30 \times 0.25 \times 0.20$ mm

Data collection

 Bruker APEXII CCD area-detector diffractometer
 9157 measured reflections

 3751 independent reflections
 3163 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.023$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.101$
 $S = 1.04$
 3751 reflections
 235 parameters

 1 restraint
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.16$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.20$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C17}-\text{H17}\cdots\text{O5}^i$	0.93	2.37	3.185 (3)	146
$\text{C25}-\text{H23A}\cdots\text{O5}^ii$	0.97	2.49	3.416 (3)	160

 Symmetry codes: (i) $x, y + 1, z$; (ii) $x - 1, y + 1, z$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FK2072).

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supplementary materials

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2-(1,3-Benzodioxol-5-yl)-3-phenylquinazolin-4(3H)-one

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Comment

Quinazoline and their derivatives are an interesting class of heterocyclic compounds that have drawn much attention because of their biological and pharmaceutical activities; such as anti-bacterial (Kung *et al.*, 1999), antimicrobial (Arfan *et al.*, 2008) antifungal (Bartoli *et al.*, 1998) and anticonvulsant activities (Mannschreck *et al.*, 1984). In view of their importance, the crystal structure determination of the title compound was carried out and the results are presented herein.

In the molecular structure of the title compound (Fig. 1), the dihedral angles between the quinazoline moiety (N3–C4/C6–C11/N1–C2) and the phenyl ring (C12/C13/C14/C15/C16/C17) as well as the 1,3-benzodioxole ring (C18–C19/C20–C24/C25–C26) are 40.57 (9)° and 42.31 (11)°, respectively. The dihedral angle between the phenyl ring (C12/C13/C14/C15/C16/C17) and 1,3-benzodioxole ring (C18–C19/C20–C24/C25–C26) is 4.34 (10)°. The crystal packing exhibits intermolecular C—H···O interactions (Fig. 2) that link molecules into endless zig-zag chains extended along [100].

Experimental

To a solution of 2-amino-*N*-phenylbenzamide (1 mmol) and benzo[*d*][1,3]dioxole-5-carbaldehyde (1 mmol) in ethyl acetate (2 ml) was added propyl phosphonic anhydride (1 mmol) and the reaction mixture was stirred for about 2 hrs at room temperature, then 2,3-Dichloro-5,6-dicyano-1,4-benzoquinone (DDQ) (1 mmol) was added and stirred for about 30 minutes. The reaction mixture was diluted with water and extracted to ethyl acetate and it was washed with 10% NaHCO₃, water, brine solution and dried over anhydrous sodium sulfate and concentrated under reduced pressure to get a crude product which was recrystallized by slow evaporation in ethyl acetate at room temperature to get the title compound.

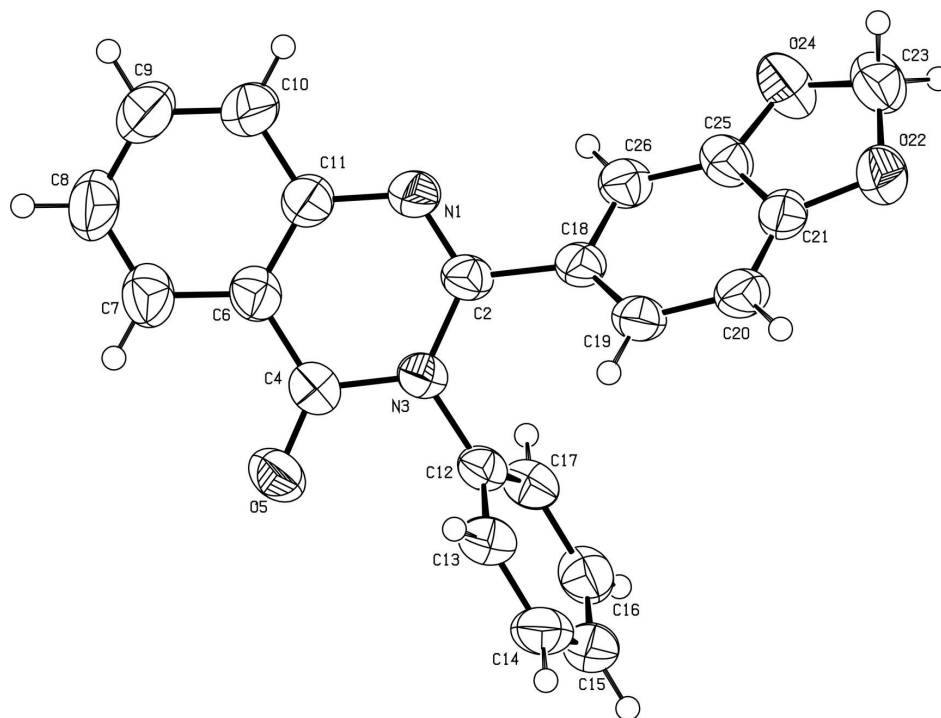
Refinement

H atoms were placed at idealized positions and allowed to ride on their parent atoms with C—H distances in the range of 0.93 to 0.97 Å; $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

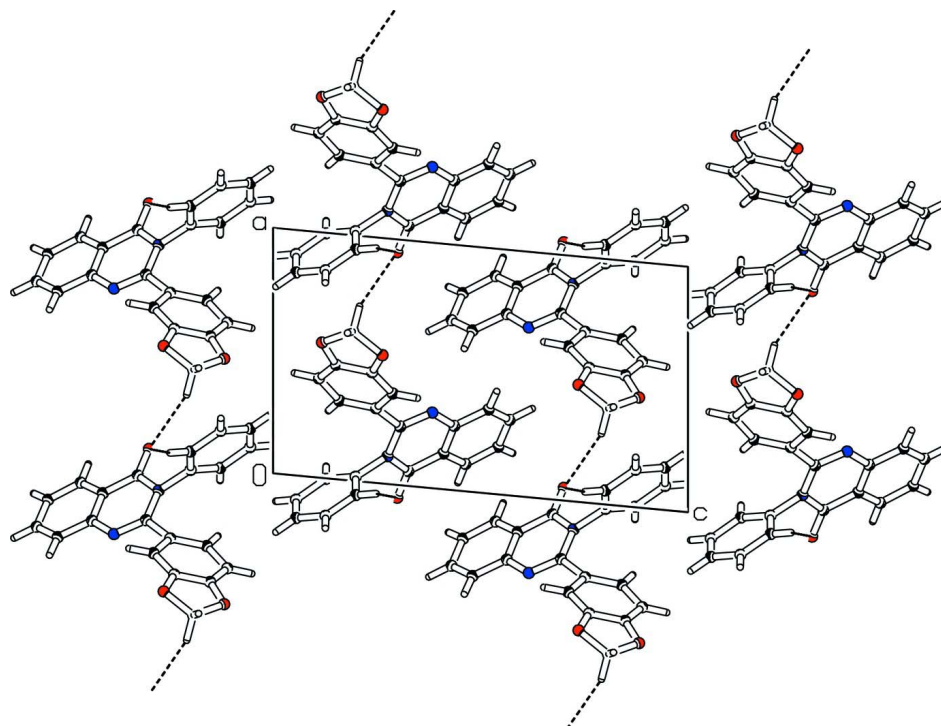
The title compound crystallizes in the non-centrosymmetric space group $P2_1$; however, in the absence of significant anomalous scattering effects, the Flack parameter is essentially meaningless.

Computing details

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT* (Bruker, 2009); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008).

**Figure 1**

Molecular structure of the title compound with anisotropic displacement ellipsoids drawn at the 50% probability level.

**Figure 2**

Packing diagram of the molecule viewed along b-axis. Intermolecular hydrogen bonding drawn as dotted lines.

2-(1,3-Benzodioxol-5-yl)-3-phenylquinazolin-4(3H)-one

Crystal data

$C_{21}H_{14}N_2O_3$	$F(000) = 356$
$M_r = 342.34$	$D_x = 1.377 \text{ Mg m}^{-3}$
Monoclinic, $P2_1$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
Hall symbol: P 2yb	Cell parameters from 3751 reflections
$a = 8.984 (4) \text{ \AA}$	$\theta = 1.3\text{--}28.0^\circ$
$b = 6.056 (3) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$c = 15.248 (6) \text{ \AA}$	$T = 273 \text{ K}$
$\beta = 95.357 (6)^\circ$	Block, yellow
$V = 826.0 (6) \text{ \AA}^3$	$0.30 \times 0.25 \times 0.20 \text{ mm}$
$Z = 2$	

Data collection

Bruker APEXII CCD area-detector diffractometer	$R_{\text{int}} = 0.023$
ω and φ scans	$\theta_{\text{max}} = 28.0^\circ$, $\theta_{\text{min}} = 1.3^\circ$
9157 measured reflections	$h = -11 \rightarrow 11$
3751 independent reflections	$k = -8 \rightarrow 7$
3163 reflections with $I > 2\sigma(I)$	$l = -19 \rightarrow 19$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.040$	H-atom parameters constrained
$wR(F^2) = 0.101$	$w = 1/[\sigma^2(F_o^2) + (0.0558P)^2 + 0.0215P]$
$S = 1.04$	where $P = (F_o^2 + 2F_c^2)/3$
3751 reflections	$(\Delta/\sigma)_{\text{max}} < 0.001$
235 parameters	$\Delta\rho_{\text{max}} = 0.16 \text{ e \AA}^{-3}$
1 restraint	$\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O5	1.05179 (14)	0.2056 (2)	0.70071 (8)	0.0591 (4)
O24	0.44816 (14)	1.2103 (3)	0.88187 (8)	0.0637 (4)
O26	0.47712 (16)	1.2832 (2)	0.73629 (9)	0.0690 (5)
N1	0.69195 (15)	0.5808 (3)	0.61645 (9)	0.0552 (5)
N3	0.88645 (14)	0.4809 (2)	0.72424 (8)	0.0416 (4)
C2	0.76074 (16)	0.6048 (3)	0.69362 (10)	0.0435 (5)

C4	0.93930 (17)	0.3085 (3)	0.67377 (10)	0.0443 (5)
C6	0.85398 (17)	0.2701 (3)	0.58978 (11)	0.0490 (5)
C7	0.8883 (2)	0.0937 (4)	0.53638 (13)	0.0621 (7)
C8	0.8062 (2)	0.0598 (5)	0.45769 (15)	0.0835 (9)
C9	0.6918 (3)	0.2064 (6)	0.42996 (16)	0.1048 (12)
C10	0.6579 (3)	0.3796 (6)	0.48092 (14)	0.0923 (9)
C11	0.73748 (19)	0.4136 (4)	0.56322 (11)	0.0569 (6)
C12	0.98242 (16)	0.5453 (3)	0.80117 (9)	0.0413 (5)
C13	1.00197 (18)	0.4050 (3)	0.87253 (10)	0.0487 (5)
C14	1.09842 (19)	0.4651 (4)	0.94460 (12)	0.0591 (6)
C15	1.1735 (2)	0.6638 (4)	0.94456 (12)	0.0608 (7)
C16	1.15312 (19)	0.8023 (3)	0.87348 (13)	0.0588 (6)
C17	1.05718 (18)	0.7432 (3)	0.80046 (11)	0.0490 (5)
C18	0.69538 (15)	0.7673 (3)	0.75203 (10)	0.0423 (5)
C19	0.67842 (17)	0.7258 (3)	0.84001 (10)	0.0484 (5)
C20	0.59554 (18)	0.8651 (3)	0.88969 (11)	0.0526 (6)
C21	0.53344 (17)	1.0466 (3)	0.84817 (11)	0.0476 (5)
C22	0.55001 (18)	1.0905 (3)	0.76149 (11)	0.0490 (5)
C23	0.62930 (17)	0.9549 (3)	0.71145 (10)	0.0471 (5)
C25	0.3916 (2)	1.3373 (4)	0.80727 (13)	0.0644 (7)
H7	0.96700	-0.00050	0.55440	0.0750*
H8	0.82650	-0.06040	0.42280	0.1000*
H9	0.63780	0.18510	0.37560	0.1260*
H10	0.58160	0.47610	0.46110	0.1110*
H13	0.95090	0.27130	0.87220	0.0580*
H14	1.11260	0.37170	0.99310	0.0710*
H15	1.23820	0.70390	0.99310	0.0730*
H16	1.20370	0.93640	0.87410	0.0710*
H17	1.04380	0.83620	0.75180	0.0590*
H19	0.72360	0.60150	0.86660	0.0580*
H20	0.58330	0.83540	0.94840	0.0630*
H23A	0.28690	1.30300	0.79170	0.0770*
H23B	0.40020	1.49370	0.82050	0.0770*
H26	0.63900	0.98610	0.65250	0.0570*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O5	0.0584 (7)	0.0551 (7)	0.0635 (7)	0.0186 (6)	0.0045 (6)	0.0065 (6)
O24	0.0545 (7)	0.0728 (9)	0.0645 (7)	0.0097 (7)	0.0101 (6)	-0.0164 (7)
O26	0.0705 (9)	0.0601 (8)	0.0795 (8)	0.0209 (7)	0.0228 (7)	0.0092 (7)
N1	0.0408 (7)	0.0766 (11)	0.0470 (7)	0.0126 (8)	-0.0016 (6)	-0.0096 (8)
N3	0.0370 (6)	0.0445 (7)	0.0430 (6)	0.0033 (6)	0.0025 (5)	0.0029 (6)
C2	0.0338 (7)	0.0526 (10)	0.0441 (8)	0.0033 (7)	0.0035 (6)	0.0023 (7)
C4	0.0430 (8)	0.0414 (9)	0.0497 (9)	0.0006 (7)	0.0105 (7)	0.0053 (7)
C6	0.0392 (8)	0.0560 (11)	0.0533 (9)	-0.0022 (8)	0.0125 (7)	-0.0040 (8)
C7	0.0522 (10)	0.0674 (13)	0.0688 (11)	-0.0017 (9)	0.0163 (9)	-0.0163 (10)
C8	0.0590 (12)	0.108 (2)	0.0847 (14)	0.0021 (14)	0.0134 (11)	-0.0491 (15)
C9	0.0626 (13)	0.169 (3)	0.0792 (14)	0.0229 (17)	-0.0125 (11)	-0.0633 (17)
C10	0.0616 (12)	0.142 (2)	0.0694 (13)	0.0328 (15)	-0.0145 (10)	-0.0415 (15)

C11	0.0403 (8)	0.0769 (13)	0.0534 (9)	0.0035 (9)	0.0033 (7)	-0.0150 (9)
C12	0.0325 (7)	0.0455 (9)	0.0458 (8)	0.0052 (7)	0.0040 (6)	0.0019 (7)
C13	0.0408 (8)	0.0539 (10)	0.0514 (9)	0.0014 (8)	0.0039 (6)	0.0096 (8)
C14	0.0481 (9)	0.0804 (14)	0.0480 (9)	0.0125 (10)	0.0005 (7)	0.0101 (9)
C15	0.0433 (9)	0.0767 (15)	0.0607 (11)	0.0098 (9)	-0.0041 (8)	-0.0119 (10)
C16	0.0437 (9)	0.0525 (10)	0.0799 (12)	-0.0023 (9)	0.0036 (8)	-0.0120 (10)
C17	0.0425 (8)	0.0440 (9)	0.0601 (10)	0.0036 (7)	0.0026 (7)	0.0071 (8)
C18	0.0303 (7)	0.0518 (9)	0.0444 (8)	0.0002 (7)	0.0013 (6)	-0.0016 (7)
C19	0.0390 (8)	0.0597 (10)	0.0458 (8)	0.0077 (8)	0.0001 (6)	0.0028 (8)
C20	0.0425 (8)	0.0740 (13)	0.0413 (8)	0.0036 (9)	0.0035 (6)	-0.0016 (8)
C21	0.0345 (7)	0.0593 (11)	0.0487 (8)	-0.0016 (7)	0.0031 (6)	-0.0131 (8)
C22	0.0395 (8)	0.0478 (9)	0.0599 (10)	0.0020 (8)	0.0064 (7)	-0.0001 (8)
C23	0.0395 (8)	0.0580 (10)	0.0443 (8)	0.0009 (8)	0.0063 (6)	0.0022 (8)
C25	0.0517 (10)	0.0643 (12)	0.0785 (12)	0.0096 (9)	0.0130 (9)	-0.0074 (10)

Geometric parameters (Å, °)

O5—C4	1.225 (2)	C15—C16	1.369 (3)
O24—C21	1.381 (2)	C16—C17	1.390 (3)
O24—C25	1.427 (3)	C18—C19	1.387 (2)
O26—C22	1.375 (2)	C18—C23	1.399 (3)
O26—C25	1.423 (2)	C19—C20	1.395 (2)
N1—C2	1.285 (2)	C20—C21	1.362 (3)
N1—C11	1.383 (3)	C21—C22	1.370 (2)
N3—C2	1.399 (2)	C22—C23	1.366 (2)
N3—C4	1.406 (2)	C7—H7	0.9300
N3—C12	1.443 (2)	C8—H8	0.9300
C2—C18	1.485 (2)	C9—H9	0.9300
C4—C6	1.449 (2)	C10—H10	0.9300
C6—C7	1.395 (3)	C13—H13	0.9300
C6—C11	1.391 (3)	C14—H14	0.9300
C7—C8	1.364 (3)	C15—H15	0.9300
C8—C9	1.394 (4)	C16—H16	0.9300
C9—C10	1.357 (5)	C17—H17	0.9300
C10—C11	1.400 (3)	C19—H19	0.9300
C12—C13	1.379 (2)	C20—H20	0.9300
C12—C17	1.374 (3)	C23—H26	0.9300
C13—C14	1.383 (3)	C25—H23A	0.9700
C14—C15	1.380 (3)	C25—H23B	0.9700
C21—O24—C25	105.01 (14)	O24—C21—C22	109.49 (15)
C22—O26—C25	105.05 (14)	C20—C21—C22	121.91 (16)
C2—N1—C11	118.47 (15)	O26—C22—C21	110.03 (15)
C2—N3—C4	121.26 (13)	O26—C22—C23	127.86 (15)
C2—N3—C12	121.91 (13)	C21—C22—C23	122.12 (16)
C4—N3—C12	116.04 (12)	C18—C23—C22	117.59 (14)
N1—C2—N3	123.30 (15)	O24—C25—O26	107.75 (16)
N1—C2—C18	116.16 (14)	C6—C7—H7	120.00
N3—C2—C18	120.49 (13)	C8—C7—H7	120.00
O5—C4—N3	120.30 (14)	C7—C8—H8	120.00

O5—C4—C6	124.52 (16)	C9—C8—H8	120.00
N3—C4—C6	115.17 (14)	C8—C9—H9	119.00
C4—C6—C7	120.76 (16)	C10—C9—H9	119.00
C4—C6—C11	118.60 (16)	C9—C10—H10	120.00
C7—C6—C11	120.64 (16)	C11—C10—H10	120.00
C6—C7—C8	119.9 (2)	C12—C13—H13	120.00
C7—C8—C9	119.6 (2)	C14—C13—H13	120.00
C8—C9—C10	121.2 (2)	C13—C14—H14	120.00
C9—C10—C11	120.3 (3)	C15—C14—H14	120.00
N1—C11—C6	122.83 (15)	C14—C15—H15	120.00
N1—C11—C10	118.6 (2)	C16—C15—H15	120.00
C6—C11—C10	118.4 (2)	C15—C16—H16	120.00
N3—C12—C13	119.83 (15)	C17—C16—H16	120.00
N3—C12—C17	118.99 (14)	C12—C17—H17	120.00
C13—C12—C17	121.15 (14)	C16—C17—H17	120.00
C12—C13—C14	119.28 (17)	C18—C19—H19	119.00
C13—C14—C15	119.95 (18)	C20—C19—H19	119.00
C14—C15—C16	120.36 (17)	C19—C20—H20	122.00
C15—C16—C17	120.23 (17)	C21—C20—H20	122.00
C12—C17—C16	119.03 (16)	C18—C23—H26	121.00
C2—C18—C19	123.05 (16)	C22—C23—H26	121.00
C2—C18—C23	116.63 (14)	O24—C25—H23A	110.00
C19—C18—C23	119.63 (15)	O24—C25—H23B	110.00
C18—C19—C20	121.88 (16)	O26—C25—H23A	110.00
C19—C20—C21	116.88 (15)	O26—C25—H23B	110.00
O24—C21—C20	128.60 (15)	H23A—C25—H23B	108.00
C25—O24—C21—C20	-170.35 (18)	C11—C6—C7—C8	-0.6 (3)
C25—O24—C21—C22	9.67 (19)	C4—C6—C11—N1	-5.3 (3)
C21—O24—C25—O26	-15.83 (19)	C4—C6—C11—C10	178.3 (2)
C25—O26—C22—C21	-10.09 (19)	C7—C6—C11—N1	175.08 (18)
C25—O26—C22—C23	170.00 (18)	C7—C6—C11—C10	-1.4 (3)
C22—O26—C25—O24	15.97 (19)	C6—C7—C8—C9	2.2 (3)
C11—N1—C2—N3	4.8 (3)	C7—C8—C9—C10	-1.7 (4)
C11—N1—C2—C18	-172.49 (16)	C8—C9—C10—C11	-0.4 (4)
C2—N1—C11—C6	0.3 (3)	C9—C10—C11—N1	-174.8 (2)
C2—N1—C11—C10	176.7 (2)	C9—C10—C11—C6	1.9 (4)
C4—N3—C2—N1	-4.8 (2)	N3—C12—C13—C14	177.68 (15)
C4—N3—C2—C18	172.43 (14)	C17—C12—C13—C14	0.1 (3)
C12—N3—C2—N1	164.67 (16)	N3—C12—C17—C16	-178.07 (15)
C12—N3—C2—C18	-18.1 (2)	C13—C12—C17—C16	-0.5 (2)
C2—N3—C4—O5	178.13 (15)	C12—C13—C14—C15	0.1 (3)
C2—N3—C4—C6	-0.4 (2)	C13—C14—C15—C16	0.0 (3)
C12—N3—C4—O5	8.1 (2)	C14—C15—C16—C17	-0.4 (3)
C12—N3—C4—C6	-170.47 (14)	C15—C16—C17—C12	0.6 (3)
C2—N3—C12—C13	120.18 (17)	C2—C18—C19—C20	-169.50 (15)
C2—N3—C12—C17	-62.2 (2)	C23—C18—C19—C20	0.7 (2)
C4—N3—C12—C13	-69.84 (18)	C2—C18—C23—C22	170.86 (14)
C4—N3—C12—C17	107.81 (17)	C19—C18—C23—C22	0.1 (2)

N1—C2—C18—C19	135.61 (17)	C18—C19—C20—C21	-1.0 (2)
N1—C2—C18—C23	-34.8 (2)	C19—C20—C21—O24	-179.44 (16)
N3—C2—C18—C19	-41.8 (2)	C19—C20—C21—C22	0.5 (2)
N3—C2—C18—C23	147.76 (15)	O24—C21—C22—O26	0.25 (19)
O5—C4—C6—C7	6.2 (3)	O24—C21—C22—C23	-179.83 (16)
O5—C4—C6—C11	-173.46 (17)	C20—C21—C22—O26	-179.74 (15)
N3—C4—C6—C7	-175.32 (16)	C20—C21—C22—C23	0.2 (3)
N3—C4—C6—C11	5.0 (2)	O26—C22—C23—C18	179.41 (16)
C4—C6—C7—C8	179.72 (19)	C21—C22—C23—C18	-0.5 (2)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C17—H17 \cdots O5 ⁱ	0.93	2.37	3.185 (3)	146
C25—H23 <i>A</i> \cdots O5 ⁱⁱ	0.97	2.49	3.416 (3)	160

Symmetry codes: (i) $x, y+1, z$; (ii) $x-1, y+1, z$.