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Review paper

Chemical derivatization strategies for enhancing the HPLC analytical performance of natural active triterpenoids

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ABSTRACT

Triterpenoids widely exist in nature, displaying a variety of pharmacological activities. Determining triterpenoids in different matrices, especially in biological samples holds great significance. Highperformance liquid chromatography (HPLC) has become the predominant method for triterpenoids analysis due to its exceptional analytical performance. However, due to the structural similarities among botanical samples, achieving effective separation of each triterpenoid proves challenging, necessitating significant improvements in analytical methods. Additionally, triterpenoids are characterized by a lack of ultraviolet (UV) absorption groups and chromophores, along with low ionization efficiency in mass spectrometry. Consequently, routine HPLC analysis suffers from poor sensitivity. Chemical derivatization emerges as an indispensable technique in HPLC analysis to enhance its performance. Considering the structural characteristics of triterpenoids, various derivatization reagents such as acid chlorides, rho-damines, isocyanates, sulfonic esters, and amines have been employed for the derivatization analysis of triterpenoids. This review comprehensively summarized the research progress made in derivatization strategies for HPLC detection of triterpenoids. Moreover, the limitations and challenges encountered in previous studies are discussed, and future research directions are proposed to develop more effective derivatization methods.

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1. Introduction

Triterpenoids are a class of botanical secondary metabolites that are condensed from six isoprene units. They are widely existing in nature in the form of aglycones or glycosides. According to the type of carbon skeleton, they can be divided into tetracyclic triterpenoids (including dammarane-type, lanostane-type, cycloartanetype, tirucallane-type and cucurbitane-type triterpenoids) and pentacyclic triterpenoids (including oleanane-type, ursane-type, lupane-type, and friedelane-type triterpenoids), as shown in Fig. 1. The chemical structures of triterpenoids are very similar, with the characteristic of no conjugated system or low conjugation degree. It is noteworthy that C-3, C-28, and other positions of the carbon skeleton contain free hydroxyl and carboxyl groups (for aglycones), which are also the glycoxidation sites (for glycosides). Studies have shown that triterpenoids have significant anti-HIV [1], anti-cancer [2], antiviral [3], anti-inflammatory [4], and metabolic regulation [5] activities with amazing applications in medicine, food, and cosmetics industries.

Given the medicinal importance of triterpenoids, the detection of triterpenoids has always been a research hotspot in the field of analytical chemistry. A variety of methods have been reported for triterpenoid analysis, such as colorimetry [6], capillary electrophoresis (CE) [7], infrared spectroscopy (IR) [8], thin-layer chromatography (TLC) [9], gas chromatography (GC) [10], highperformance liquid chromatography (HPLC) [11–13] and immunoassay [14]. Among them, HPLC has become the predominant technology for triterpenoid analysis with fast analysis speed, high sensitivity, high selectivity, good repeatability, and high sample universality. Unfortunately, achieving effective separation of triterpenoids by conventional HPLC methods is challenging due to the similar structure and polarity, necessitating significant







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Fig. 1. Distinct skeletons of triterpenoids. (A) Dammarane-type, (B) lanostane-type, (C) cycloartane-type, (D) tirucallane-type, (E) cucurbitane-type, (F) oleanane-type, (G) ursane-type, (H) lupane-type, and (I) friedelane-type.

improvements in analytical methods. Moreover, triterpenoids often lack characteristic ultraviolet (UV) absorption groups and chromophores, along with low ionization efficiency in mass spectrometry, resulting in an extremely limited detection response. Especially for the low abundance of triterpenoids in biological samples, accurate quantitative detection is more difficult.

To address these issues, derivatization techniques have been introduced to HPLC-based detection of triterpenoids. Chemical derivatization, as an indispensable technique in the field of chromatographic analysis, has the advantages of improving separation, sensitivity, and selectivity. The combination of derivatization technology with GC [15] and LC [16] plays a vital role in the fields of pharmaceutical analysis [17], environmental monitoring [18], metabolomics [19], and food safety [20]. Appropriate derivatization of triterpenoids not only improves the chromatographic behavior but also significantly enhances the instrument response, thereby greatly enhancing its application potential in trace analysis.

To date, there have been many research articles but only three reviews on the analysis of triterpenoids [21–23]. However, the published reviews lacked a comprehensive description of the derivatization technology and were rather general in nature. This review focuses on various derivatization strategies that can improve chromatographic performance and detection sensitivity during high performance liquid chromatography-ultraviolet-fluorescence detector-mass spectromatry (HPLC-UV/FLD/MS) analysis of triterpenoids. It should be noted that chemical derivatization is not applicable in prep-HPLC purification and HPLC-evaporative light scattering detection (ELSD) analysis. Although ELSD is a common detector for weak-UV-absorbing analytes under equal or gradient elution modes [24], chemical derivatization cannot change its inherent disadvantages, such as low sensitivity, poor baseline stability, cumbersome parameter setting, and nonlinear response behavior [25]. Thus, neither prep-HPLC purification nor HPLC-ELSD analysis of triterpenoids has been discussed in this review. The derivatization strategies are based on the following three principles: (1) introduce chromophores or conjugate units to meet the requirements of HPLC-UV detection, (2) introduce fluorophore units for HPLC-FLD detection, and (3) introduce ammonia nitrogen fragments for HPLC-MS detection (Table S1) [26–68]. Detailed examples are systematically listed and commented according to different sites of derivatization reactions, including hydroxyl, carboxyl, and other functional groups (e.g., double bond and carbonyl group) at positions C-3, C-28, etc., as shown in Fig. 2. The principles, advantages, and limitations of each method are thoroughly discussed, hoping to inspire further research into the development of superior derivatization reagents and procedures.

2. Derivatization of hydroxyl groups in triterpenoids

Due to the weak UV absorption and electrospray ionization efficiency of C-3 hydroxyl group, it is difficult to effectively detect triterpenoids by conventional chromatography. Thus, it has to be determined after chemical derivatization. The nucleophilicity of hydroxyl group is still too weak to be derivatized, especially in the aqueous phase. Therefore, it needs to be labeled with strong electrophile reagents. The derivatization of hydroxyl groups in triterpenoids reported in the literature mainly includes acyl chlorides [27,31,32], rhodamines [37–39], isocyanates [40], and other strong acylation reagents.

2.1. Acyl chloride-type derivatization reagents

Acyl chlorine-derivatized reagents, such as benzoyl chloride (BC), *p*-nitrobenzoyl chloride (PNBC), 3,5-dinitrobenzoyl chloride (3,5-DNB), etc., contain benzoyl chromophores, which can significantly improve the instrument response at specific wavelength. With the increase in the p- π conjugate system of the derivatization

reagent, the instrument response of the derivative increases gradually. In addition, the stronger electron-withdrawing of the group linked to the carbonyl group in the derivatization reagent will reduce the electron cloud density of the carbonyl group more, thus enhancing the reaction activity with the nucleophile and finally obtaining a higher derivative yield. The reactions are usually carried out in pyridine solution. On the one hand, the esterification reaction requires a weak alkaline environment, and on the other hand, the derivatization reagent can be avoided from hydrolysis in aqueous solution. In practice, it is necessary to add excessive derivatization reagent to obtain a higher yield.

2.1.1. Benzoyl chloride-type derivatization reagents

Ginsenosides are dammarane-type tetracyclic triterpenoid saponins that play an important role in immune regulation, central nervous system disease, and cerebrovascular diseases [69-71]. Panaxadiol (PD), as the hydrolysate of ginsenoside, is often determined as an indicator of total ginsenosides. It was reported that PD could be fully derivatized by BC in a water bath at 80 °C for 2 h [26]. In addition to the derivatization of the hydroxyl group of the aglycone moiety, the hydroxyl group of the sugar chain can also directly participate in the reaction. By comparing the derivatization reagents BC, 4-nitrobenzoyl chloride, 3,5-DNB and β -naphthyl chloride. BC was selected as the best derivatization reagent for ginsenoside to generate a single derivative [27]. More than 10-fold amounts of BC were used in the pyridine solution. There was no hydroxyl characteristic peak observed in the infrared spectrum, indicating that all the hydroxyl groups of ginsenoside were benzoylated. The method is simple to operate and can be used to distinguish white ginseng from the root hairs of ginseng by HPLC pattern analysis.

Astragaloside IV (AS-IV) is a cycloartane-type triterpenoid saponin that is considered the primary active ingredient of Astragali Radix [72]. Due to the lack of characteristic absorption groups in its molecular structure, it is difficult to analyze directly by UV detection. Based on this, the use of excessive BC to derivatize AS-IV not only improves the sensitivity of AS-IV but also significantly improves the chromatographic behavior [28]. A comparative detection of AS-IV was performed between HPLC-ELSD and derivative HPLC-UV. After derivatization by BC, the results of the derivative HPLC-UV method were consistent with those of the HPLC-ELSD method, which overcame the limitations of equipment [29]. It is worth noting that certain sample pretreatment is needed due to the complexity of natural products. For example, liquid-liquid extraction (LLE), macroporous adsorption resin column chromatography. and solid-phase extraction were successively used to purify the AS-IV extract to reduce the influence of the complex matrix [30].

Other benzoyl chloride-type reagents containing more chromophores can further enhance the sensitivity of triterpenoid detection, including PNBC, 3,5-DNB and 4-(4,5-diphenyl-1H-imidazole-2-yl) benzoyl chloride (DIB-Cl). Some derivative HPLC-UV studies have been reported with effective diminishment of matrix interference and great improvement of detection sensitivity, such as the determination of eruboside B at 260 nm in garlic after derivatization by PNBC [31], the determination of oleanolic acid at 254 nm in Achyranthes bidentata Bl after derivatization by PNBC [32], the determination of PD and panaxatriol (PT) at 230 nm in red ginseng and Shengmai injection after derivatization by 3,5-DNB [33], and the determination of AS-IV at 230 nm in Astragali Radix after derivatization by 3,5-DNB [34]. What is particularly interesting is that DIB-Cl has been used as a derivatization reagent for the determination of betulinic acid, ursolic acid, and oleanolic acid in rosemary extract [35]. The derivatization reaction was completed after standing for 5 min at room temperature. The larger conjugated structure originating from DIB-Cl made the detection sensitivity of the derivative HFLC-FLD method more than 2000 times higher than that of the nonderivative HPLC-UV method [73], reaching a low detection limit comparable to that of liquid chromatography tandem mass spectometry (LC-MS/MS).

2.1.2. Dansyl chloride

In addition to the benzoyl chlorides mentioned above, dansyl chloride (DC) is commonly used in amine derivatization and plays an important role in protein detection [74]. It is often used for the derivatization of phenolic hydroxyl and alcohol hydroxyl groups [75]. A novel HPLC-FLD method for the detection of triterpenoids was developed [36]. DC was used in pyridine solution to derivatize 12 triterpenoids in plant secretions. The linear ranges of triterpenoids were $0.12-2500 \mu g/mL$ (r = 0.9969-0.9995), and the limit of detections (LOD) was less than 0.45 ng/mL.

2.2. Rhodamine-type derivatization reagents

Since the conjugated system is introduced by the acyl chloride derivatization reagent, the LODs of the analytes are greatly improved by UV or FLD detectors [35,73]. However, the derivative is still not suitable for MS detection. This is due to the low ionization efficiency of the derivative without the ammonia nitrogen fragment, resulting in limited signal enhancement on the MS detector. To achieve lower detection limits by LC-MS, especially for trace analysis in biological samples, the use of rhodamine-type reagents for derivatization came into being.

Rhodamine is a synthetic dye with high fluorescence and photostability that was originally used as a fluorescent stain for cells [76,77]. As a guaternary ammonium, rhodamine has high ionization efficiency in electrospray ionization (ESI) positive ion mode. Thus, it can be used as a derivatization reagent for amine and hydroxyl detection with a clear signal improvement in LC-MS analysis [78,79]. To enhance the reactivity of the derivatization reagent, the carboxyl group of rhodamine is often first converted into a more electrophilic amide with N, N-carbonyl diimidazole (CDI) in the presence of catalyst 4-dimethylaminopyridine (DMAP). In addition, microwave-assisted derivatization has been adopted to obtain a higher yield. For example, a ultra high performance liquid chromatography coupled with mass spectrometry (UHPLC-MS/MS) method was reported for PD detection after derivatization by rhodamine B (RB). The derivatization reaction was carried out in two steps (Fig. 3A) [37]. RB was first activated by CDI/DMAP and then reacted with the ginseng powder extract pretreated by ultrasonic assisted dispersive liquid-liquid microextraction (UA-DLLME). The derivatization reaction could be completed in 30 min with the assistance of microwaves, which greatly increased the yield rate and shortened the reaction time compared with conventional water bath heating. The quantitative analysis of PD could be completed within 3 min after chromatographic separation. The LOD was as low as 4.0 ng/L, which improved the sensitivity by at least 250 times compared to that without derivatization [80].

There were two kinds of cleavage pathways to produce different fragment ions of rhodamine derivative. One was the cleavage of the bond outside the oxygen atom in the ester bond, resulting in the rhodamine main structure fragment ion; the other was the cleavage of the bond between the benzene ring and the carbonyl group (Fig. 3B) [39].

RB not only has a slow reaction rate due to the steric hindrance of the 2'-carboxyl group but is also easy to convert into a lactone structure by the loss of a positive intramolecular charge, resulting



Fig. 2. Common derivatization strategies based on hydroxyl and carboxyl reactions and corresponding derivatization reagents. BC: benzoyl chloride; PNBC: *p*-nitrobenzoyl chloride; 3,5-DNB: 3,5-dinitrobenzoyl chloride; DIB-CI: 4-(4,5-diphenyl-1H-imidazole-2-yl) benzoyl chloride; DC: dansyl chloride; RB: rhodamine B; CSR: 4'-carboxy-substituted rosamine; ACCR6G: 3-*N*-(*Do*-/*D*₃-methyl-, and *Do*-/*D*₅-ethyl-)-2'-carboxyl chloride rhodamine 6 G; PTSI: *p*-toluenesulfonyl isocyanate; BAETS: 2-(5-benzoacridine) ethyl-*p*-toluenesulfonate; AETS: acridone-9-ethyl-*p*-toluenesulfonate; APETS: 2-(2-(anthracen-10-yl)-1H-phenanthro[9,10-d] imidazole-1yl) ethyl 4-methylbenzenesulfonate; DDCETS: 2-(7H-dibenzo [a, g] carbazol7-yl) ethyl4-methylbenzenesulfonate; PBIOTS: 2-(2-(pyren-1-yl)-1H-benzo[d]imidazole-1-yl)-ethyl-*p*-toluenesulfonate; CPMS: 1-(9H-carbazol-9yl) propan-2-yl)



Fig. 3. Derivatization of triterpenoids with rhodamine. (A) Derivatization of panaxadiol (PD) by rhodamine B. (B) Fragmentation pathway of rhodamine derivatives in electrospray ionization (ESI) positive ion mode. Reprinted from Refs. [37,39] with permission.

in decreased reactivity. 4'-Carboxy-substituted rosamine (CSR), another rhodamine derivatization reagent, is a good substitute for RB. A simple and environmentally friendly UHPLC-MS/MS method was developed to determine 20(S)-protopanaxadiol (PPD) and 20(S)-protopanaxatriol (PPT) in plasma after derivatization by CSR [38]. This reaction had two obvious advantages: (1) it could be quickly completed in 5 min with the assistance of microwaves; (2) it was a one-step reaction, which no longer required the participation of CDI and DMAP.

In recent years, sample derivatization based on stable isotope labelling has emerged for LC–MS analysis, showing great potential in pharmaceutical analysis and food safety monitoring [81–83]. A

series of quadruple stable isotopes, $3-N-(D_0-/D_3-methyl)$, and D_0-/D_5 -ethyl-)-2'-carboxyl chloride rhodamine 6 G (ACCR6G), were designed and synthesized to improve the chromatographic performance and MS response of PD and PT. This strategy first introduced an isotopic methyl or ethyl tag to PD and PT from samples and an analogue propyl tag to PD and PT from the standards, respectively. Then two categories of derivatives were mixed equally for the subsequent magnetic dispersive solid-phase extraction before UPLC-MS/MS determination (Fig. 4) [39]. In this method, the detection sensitivity of PD and PT derivatives was increased by more than 200 times compared with that of unlabeled forms.

methanesulfonate; PTD: *p*-toluidine; BAAH: 2-(12-benzo[b]acridin-5-(12H)-yl)-acetohydrazide; BCEHC: 2-(5H-benzo[a]-carbazol-11(6H)-yl)-ethyl-hydrazine-carboxylate; CPR: 2'carbonyl-piperazine rhodamine B; PA: 2-picolylamine; DMED: *N*,N-dimethylethylenediamine; CMP: 3-carbinol-1-methylpyridinium iodide; HIQB: (2-(2-hydrazinyl-2-oxoethyl) isoquinolin-2-ium bromide; ADAM: 9-anthryldiazomethane; 4-NBB: 4-nitrobenzyl bromide; PMP: 1-phenyl-3-methyl-5-pyrazolone.

2.3. Isocyanate-type derivatization reagents

Isocyanates are another derivatization reagent that can be used for MS detection. Different from the positive ion detection mode of rhodamine derivatives, isocyanate derivatives are usually detected in negative ion mode. *P*-toluene sulfonyl isocyanate (PTSI) was reported to label hydroxyl groups in triterpenoids, and the resulting esters can improve ionization. The derivatization reaction was rapid and quantitative, completed within a few minutes at room temperature, followed by the addition of ammonium acetate to terminate the reaction [40,41].

Betulin is a neutral pentacyclic triterpene with two hydroxyl groups. It is difficult to directly detect because of its low ionization efficiency in ESI/MS. The researchers attempted a direct analysis without derivatization, and unfortunately no detection signal was observed. After comparing several different derivatization reagents, including acetic anhydride, N, O-bis (trimethysilyl) trifluoroacetamide, *o*-phthalic anhydride, perfluoropropionic anhydride, and PTSI, PTSI was selected as the best one to establish an LC-MS/MS method for the determination of betulin in plasma, since it could generate a derivative with higher ionization efficiency [40]. Similarly, another derivatization determination of lupeol was also performed by PTSI to monitor its dynamic change in rat plasma [41].

3. Derivatization of carboxyl groups in triterpenoids

Most triterpenoids also have carboxyl groups in addition to hydroxyl groups. Thus, it is a good choice to prepare carboxylic ester derivatives of triterpenoids by introducing high response units for FLD and MS detection. Sulfonate derivatization design is most commonly reported [43,44,51]. This method can not only enhance sensitivity but also improve chromatographic separation, making it more advantageous for simultaneous quantification of multiple triterpenoids with similar structure. In addition, aromatic amines, hydrazides, and quaternary ammonium salts are also used for carboxyl derivatization [58,60,64]. Due to the introduction of ammonia nitrogen, the signal of derivatives in MS analysis is also greatly improved, which is particularly beneficial for the detection of trace analytes in biological samples.

3.1. Sulfonic ester-type derivatization reagents

Sulfonic esters are common alkylation reagents [84]. The common structural feature of these reagents is that an aromatic ring planar group is connected to a large conjugate structure through an ethyl sulfonate link chain. The sulfonate part, as a good leaving group, can generate a carbocation nucleophilic substitution reaction with the carboxyl group, thus introducing a high fluorophore. In other words, the reaction of the sulfonates with the carboxylic acid is actually the alkylation of the carboxyl group. Due to the introduction of high fluorophores, sulfonate derivatives can be easily detected by FLD with improved sensitivity. This is a simple and quick derivatization reaction, which can be completed by heating at 90 °C for approximately 30 min. The derivatization reaction is usually carried out in the cosolvent DMF, which can not only yield a higher fluorescence response but also avoid the precipitation of lower polarity analytes. To obtain higher reaction efficiency, K₂CO₃ is usually added as a catalyst.

There are abundant triterpene acids with similar structures in plants, especially some isomers that appear in pairs [85]. Due to the small structural differences, these compounds are not well separated under conventional chromatographic conditions. Sulfonation derivatization provides an excellent solution to improve sensitivity and chromatographic separation performance. For example, oleanolic acid/ursolic acid and maslinic acid/corosolic acid are both very common pairs of epimers with low chromatographic resolution. After derivatization by 2-(5-benzoacridine) ethyl-p-toluenesulfonate (BAETS), not only was the response of the detector greatly improved, but their chromatographic behaviors were also significantly improved [42]. Another HPLC-FLD method was reported for the determination of five triterpene acids (betulinic acid, betulonic acid, maslinic acid, ursolic acid, and oleanolic acid) within 16 min after labelling with the fluorescent derivatization reagent acridone-9-ethyl-p-toluenesulfonate (AETS) [43]. Similarly, a large number of studies have been reported using other derivatization reagents to achieve simultaneous detection of multiple triterpene acids, such as 2-(2-(anthracen-10-yl)-1H-phenanthro[9,10-d] imidazole-1yl) ethyl 4-methylbenzenesulfonate (APIETS) [44], 2-(12,13-dihydro-7H-dibenzo [a, g] carbazol7-yl) ethyl4-methylbenzenesulfonate (DDCETS) [45], benzimidazo-[2,1-b] quinazolin-12(6H)-one-5ethyl-p-toluenesulfonate (BQETS) [46], BAETS [47,48] and 2-(7H-



Fig. 4. Derivatization of panaxadiol (PD) and panaxatriol (PT) with 3-N-(D_0 -/ D_3 -methyl-, and D_0 -/ D_5 -ethyl-)-2'-carboxyl chloride rhodamine 6 G (ACCR6G) based on stable isotope labelling. Reprinted from Ref. [39] with permission.

dibenzo [a, g] carbazole-7-yl) ethyl-4-methylbenzenesulfonate (DBCETS) [49,50]. These works fully demonstrated the great advantages of the sulfonic ester derivatization strategy in improving the chromatographic separation, sensitivity, and universality of triterpene acids.

In addition to the above analysis of plant extracts, semiautomatic extraction derivatization has been used to study the pharmacokinetics of seven triterpene acids (tormentic acid. ursolic acid, oleanolic acid, betulonic acid, betulinic acid, corosolic acid, maslinic acid) with only 20 µL plasma required [51]. The protocol of this method was that the blood was pumped into a modified micro-syringe with cosolvent, catalyst and derivatization reagent through a tail vein puncture. The circulating water was passed through the outer cannula of the syringe to provide heat for the reaction. After the reaction, acetonitrile was added to precipitate the protein. The triterpene acids were derivatized by 2-(2-(pyren-1-yl)-1H-benzo[d]imidazole-1-yl)-ethyl-p-toluenesulfonate (PBIOTs). The whole pretreatment procedure was more conveniently finished within 26 min. Seven analytes were rapidly separated on an RP-18 analytical column within 30 min and quantified by a fluorescence detector. An artificial neural network combined with a genetic algorithm was applied to optimize the derivatization conditions, achieving high sensitivity (LODs: 0.67-1.08 ng/mL). This semiautomatic extraction derivatization method overcame the shortcomings of a large blood sample volume and strong injury to animals in the traditional pharmacokinetic blood collection model [86,87], which was especially suitable for a large number of subbiosample analyses.

According to chemical reaction principles, other types of natural products containing carboxyl groups (e.g., fatty acids and phenolic acids) can also react with derivatization reagents. To avoid interference, an appropriate identification capacity is needed. The most accurate identification is mediated via comparison to retention time and MSⁿ fragment with standards [88–90]. However, the penetration rate of high-end instruments is not high, such as high-resolution MS (e.g., Orbitrap, quadrupole time-of-flight (Q-TOF), and Fourier transform ion cyclotron resonance (FT-ICR)), which to a certain extent increases the difficulty of analysis. To solve this problem, a chiral derivatization reagent, 1-(9H-carbazol-9yl) propan-2-yl-methanesulfonate (CPMS), was easily synthesized and applied to the detection of six triterpenoid acids (maslinic acid, corosolic acid, betulinic acid, oleanolic acid, ursolic acid, and betulonic acid), which could realize the effective separation of isomers and avoid the interference of chromatographic peaks of fatty acid derivatives. It was revealed that each triterpene acid produced two chromatographic derivative peaks with similar peak areas (Fig. 5) [52], while the fatty acid produced only one peak. The paired peaks confirmed each other to ensure the accuracy of analytes. The quantitative analysis using the bimodal phenomenon needs to meet the following three elements, none of which is dispensable: (1) the derivatization reagent contains chiral carbon atoms, (2) the analyte has chiral carbon atoms (tetracyclic and above), and (3) the analyte has good hydrophobicity.

3.2. Amine-type derivatization reagents

The reaction using sulfonic esters to derivatize triterpenoid acids must occur in an organic environment. However, it is not suitable for the direct derivatization of aqueous biological samples (plasma, urine, bile, etc.). Thus, the amine derivatization method based on carbodiimide-mediated condensation is a good



Fig. 5. Chromatogram of triterpene acids derivatized by chiral derivatization reagent. Peaks 1, 2, 3, 4, 5 and 6 are maslinic acid, corosolic acid, betulinic acid, oleanolic acid, ursolic acid, and betulonic acid, respectively. Reprinted from Ref. [52] with permission.

alternative to directly react in the aqueous phase, greatly simplifying the sample preparation process. Highly visible derivatization reagents with amino groups, including hydrazine and amine reagents, are often used to label triterpenoid acids from amide bonds for LC-UV or LC-FLD detection. Furthermore, the introduction of ammonia nitrogen makes it particularly suitable for detecting trace metabolites in biological samples by LC-MS.

Carbodiimide-mediated condensation reaction is a simple method to synthesize amide in aqueous solution [91–94]. Derivatization reagents can rapidly label carboxylic acids under mild conditions by carbodiimide (i.e., EDC and DCC) in the presence of acylation activators, (i.e., DMAP [95] and HOBt [96]). Otherwise, the derivatization reaction will need to be assisted by heating or more than 1 h to complete without acylation activators. For example, *p*-toluidine is often used to derivatize triterpene acids without DMAP or HOBt, yielding derivatives with aromatic UV absorption for HPLC-UV detection [53–57]. The derivatization reaction was usually performed in a water bath heated at 30 °C for 3 h or 80 °C for 1 h.

In addition to derivative HPLC-UV detection [53-57], an HPLC-FLD method for the determination of four triterpenoid acids (oleanolic acid, ursolic acid, maslinic acid, and betulinic acid) was established by using 2-(12-benzo[b]acridin-5-(12H)-yl)-acetohydrazide (BAAH) as the derivatization reagent [58]. Plasma samples could be directly reacted with BAAH without additional extraction or precipitation pretreatment. This method overcame the limitation of 'anhydrous conditions' and greatly simplified the experimental operation, making it suitable for high-throughput serum sample screening. After that, a comparison HPLC-FLD test for the determination of five triterpenoid acids (asiatic acid, betulinic acid, corosolic acid, maslinic acid and oleanolic acid) was reported by precolumn derivatization with two different labelling reagents, 2-(5H-benzo[a]-carbazol-11(6H)-yl)-ethyl-hydrazine-carboxylate (BCEHC) and BAAH [59]. The results revealed that both labelling reagents could favorably satisfy the analytical requirements with their own advantages. BCEHC had lower limits of detection $(0.42-1.35 \ \mu g/L \ vs \ 0.81-3.03 \ \mu g/L)$, while BAAH had a shorter retention time (8 min vs 11.5 min).

LC-MS is the most sensitive detection method for ammonia derivatives. Accurate and sensitive LC-MS/MS determination of oleanolic acid and maslinic acid in human urine and serum was achieved after derivatization with 2-picolylamine (PA) [60]. Another UPLC-MS/MS determination of oleanolic acid and ursolic acid in a pharmacokinetic study was also reported using the labelling reagent 2'-carbonyl-piperazine rhodamine B (CPR) [61].

When LC-MS analysis is performed, it is usually necessary to add an appropriate internal standard to correct errors in sample preparation and overcome possible ion suppression in MS detection. The use of isotope labeled derivatization reagent to generate internal standard is the optimal method. For example, an LC-MS/MS method was developed to determine two pentacyclic triterpenoids (oleanolic acid and ursolic acid) using deuterated-N, N-dimethylethylenediamine (DMED- d_4)-derivatized standards as internal standards [62]. Similarly, 3-carbinol-1methylpyridinium iodide (CMP) and CMP- d_3 were used to label triterpene acid in another study using stable isotope coding techniques [63]. Isotope labeling method has unparalleled advantages in correcting matrix effects, eliminating experimental errors, and improving accuracy and precision. However, the drawbacks are also evident: the isotope labeled derivatization reagents are expensive and even not commercially available, which limits its application.

Reference standards are necessary for quantitative analysis. However, in practical work, some reference standards often cannot be obtained or are expensive. Thus, the development of analytical methods free of reference standards has always been a hot topic. Quantitative analysis of multiple components via a single marker method provides one solution [97,98], while the detection of equivalent ions after derivatization provides another solution. Based on the ion equivalence strategy, a quaternary ammonium cationic derivatization reagent, (2-(2-hydrazinyl-2oxoethyl) isoquinolin-2-ium bromide (HIQB), was synthesized to derivatize seven available triterpenoid acids (pachymic acid, polyporenic acid C, dehydrotumulosic acid, 16a-hydroxytrametenolic acid, dehydrotrametenolic acid, 3-o-acetyl-16ahydroxydehydrotrametenolic acid, and poricoic acid B). Then, a novel LC-MS/MS method was developed using oleanolic acid as an internal standard (Fig. 6) [64]. Two kinds of fragment ions, i.e., m/z 130 and 170, were generated by the derivatives and were thus used as the quantitation ion and qualitative ion, respectively. The results showed that the sensitivity of each of the seven triterpene acids after derivatization was significantly increased by 20-160 times. This derivatization method could determine a total of 53 and 19 triterpene acids in the Poria cocos extracts and madin-darby canine kidney (MDCK) cell bidirectional transport samples with an limit of quantification (LOQ) of 0.1 nM, respectively. It can be expected that this novel ion equivalence strategy has great potential for analyzing other analogues of triterpenoid acids under the following two preconditions: one is to ensure the same MS quantitative ion fragments produced by internal standard and analytes; the other is to prove that the ionization efficiency of derivatives is only related to the concentration of analytes.

3.3. Other reagents for derivatizing carboxylic acids

Benzyl bromide as a widely used alkylation reagent, it can generate electrophilic carbocations that react with the nucleophile carboxylic acid to form esters [99]. 4-Nitrobenzyl bromide (4-NBB) was reported to label carboxyl groups in triterpenoids. The electrophilicity of carbocation can be further enhanced by the electronabsorbing action of nitro group. For example, an HPLC-UV method was established by using 4-NBB to derivatize glycyrrhizic acid [65].

Diazomethane is another class of alkylating reagents for carboxylic acid derivatization. Under acidic conditions, the methylene group of diazomethanes is protonated and reacts with carboxylic acids to form esters [100,101]. Based on this, a fluorescent derivatization reagent, 9-anthraquinone diazomethane (ADAM), was synthesized to label triterpene acids (ursolic acid, oleanolic acid, and betulinic acid) in toluene medium [66]. The LOD measured by HPLC-FLD after derivatization was nearly 100 times lower than that measured by HPLC-UV before derivatization.

4. Derivatization of other functional groups

Triterpenoids containing carbonyl groups in the structure can directly react with quaternary ammonium reagents (i.e., Girard's reagent T and Girard's reagent P) to form oximes and hydrazones, thus enhancing the detection signal intensity [102,103]. Due to the influence of the spatial structure of the derivatization reagents. small molecules and monocarboxylic compounds are easier to be derivatized [104]. NH₂OH·HCl is a nucleophile that can react rapidly with carbonyl groups to form easily ionized hydroxylase oxime under mild conditions [105]. The carbonyl group in triterpenoids was derivatized using NH₂OH · HCl for ESI/MS detection in our team's previous work [106]. This method enhances the MS sensitivity of triterpenoids with low polarity, poor ionization and even no ionization. It can be used not only to detect carbonyl groups but also to detect hydroxyl groups, provided that the primary and secondary hydroxyl groups are oxidized to carbonyl groups in advance. If there is no carbonyl group in the analyte itself, it is also possible to obtain its carbonyl intermediate for derivatization to achieve quantitative analysis. For example, the double bond on the side chain of ginsenosides at C-17 could be oxidized to a carbonyl group by ozone and then reacted with FMOC-hydrazine to form the corresponding hydrazone under weakly acidic conditions (Fig. 7) [67].

In addition to the functional groups of the aglycone part, some special sugar groups of saponins can also be used as derivatization sites. 1-Phenyl-3-methyl-5-pyrazolone (PMP) is a typical reducing sugar derivatization reagent for LC-UV and LC-MS detection [107,108]. The advantage of the PMP-derivatized reaction is that it can be quantitatively reacted with reducing sugar under mild conditions to obtain a unique detectable product. For example, *D*-quinovose (6-deoxyglucose) is a specific monosaccharide in the triterpenoid glycosides of sea cucumber. Using *D*-quinovose as the measurement standard, triterpenoid glycosides were indirectly determined by HPLC-UV detection after PMP derivation in NaOH solution [68].

5. Sample pretreatment

Appropriate sample pretreatment of botanical extracts can not only enrich trace analytes, but also reduce matrix interference, thereby achieving the goal of improving the analytical performance. On the other hand, the derivatization process may generate new impurities. Moreover, excessive derivatization reagents and catalysts will not only decrease chromatographic performance but also pollute the analytical column. Therefore, it is very important to preprocess the samples in an appropriate way before and after derivatization.

The basic requirement of chromatographic analysis is to remove possible interfering impurities and simplify the chromatogram as much as possible. LLE, as a classic method, is widely used for sample pretreatment, such as the extraction of ginsenosides and astragaloside IV from crude extract using *n*-butanol [27,28]. LLE is highly universal and free of any special equipment. However, it requires a large number of toxic chemical solvents and is tedious, timeconsuming and laborious to perform. Solid-phase extraction (SPE) is another commonly used sample pretreatment before derivatization using various fillers, including octadecyl-silylated silica gel, polystyrene/divinyl benzene polymer, alumina, hydrophilic adsorbents, and silica gel bonded with quaternary ammonium ions. For example, the use of a C₁₈ column to purify saponins in garlic [31] and ginsenosides in plasma [67] and the use of Supelclean LC-SAX to purify triterpene acids in raspberry [66] have the advantages of



Fig. 6. Equivalent quantitative ion strategy with quaternary ammonium cation derivatization. Reprinted from Ref. [64] with permission. MS: mass spectromatry; MRM: multiple reaction monitoring.

organic solvent savings and convenient operation. However, the expensive price of SPE cartridges hinders their extensive use. In recent years, the application of dispersive liquid-liquid microextraction (DLLME) has overcome the problems of LLE and SPE in sample preparation [109,110]. A large amount of triterpenoids can be enriched in a short time with a small amount of organic solvent through a simple operation. In addition, the ultrasonic assistance technique is often used to improve the extraction efficiency of DLLME, for example, the extract of panaxadiol [37,39] and several other triterpene acids [45,46] from some Chinese medicinal herbs.

Given the excessive derivatization reagents and catalysts used, the chemicals have a serious impact on the detection results. Especially for mass spectrometry detection, excessive derivatization reagents would strongly suppress the signal intensity of the analytes [20]. Therefore, the pretreatment of triterpenoid derivatives is also crucial for HPLC analysis. The interference is significantly weakened by LLE treatment, such as the ginsenoside derivative extracted by chloroform [27]. Dual-ultrasonic-assisted dispersive liquid-liquid microextraction (Dual-UADLLME) was used for sample treatment before and after the derivatization of PPD and PPT. In view of the disadvantages of chlorine solvent used in traditional DLLME with high toxicity and low degradability, brominated solvent was used as a good substitute for sample extraction in this new Dual-UADLLME. Furthermore, a back-UADLLME procedure was designed to remove excess derivatization reagents and catalysts to avoid ion inhibition effects as much as possible (Fig. 8A) [38]. In recent years, dispersed solid-phase extraction (DSPE) has attracted extensive attention due to its easy operation, time savings, and environmental friendliness. Magnetic DSPE has been applied to extract PD derivative and PT derivative [39] and oleanolic acid derivative and ursolic acid derivative [61]. The principle of the method is to use magnetic functional nanomaterials (e.g., Fe₃O₄/graphene oxide) as the carrier of the derivatization reaction. After the reaction, the derivative is separated by an external magnet accompanying solvent cleaning for LC analysis (Fig. 8B) [111]. The use of magnetic dispersants can improve the specificity and selectivity of the separation.

6. Conclusions and outlook

Effective monitoring of triterpenoids is of great significance to guide natural medicine R&D and their clinical application. HPLC is



Fig. 7. Derivatization of ginsenoside Rb1 by N-(9-Fluorenylmethoxycarbonyl)-hydrazine (FMOC-hydrazine). Reprinted from Ref. [67] with permission.



Fig. 8. Sample pretreatment operations. (A) Dual-ultrasonic-assisted dispersive liquid-liquid microextraction (Dual-UADLLME) used for sample pretreatment. (B) Magnetic dispersive solid-phase extraction (MDSPE) for sample pretreatment. (a) Preparation of magnetic graphene oxide (Fe₃O₄/GO) from graphene oxide (GO). (b) MDSPE sample processing process (By Figdraw, www.figdraw.com). Reprinted from Refs. [38,111] with permission. IS: internal standard; PPD: 20(S)-protopanaxadiol; PPT: 20(S)-protopanaxatriol; UHPLC-MS/MS: ultra high performance liquid chromatography coupled with mass spectrometry; LC-MS/MS: liquid chromatography tandem mass spectometry; MAD: microwave-assisted derivatization.

the preferred method for the analysis of triterpenoids. This review comprehensively summarizes multiple derivatization methods of triterpenoids for HPLC-UV/FLD/MS detection.

The specific derivatization reaction process is described in this review. It can be clearly observed that there are significant differences in reaction conditions among different studies. For example, for the derivatization of triterpenoid saponins, AS-IV could be completed at 4 °C for 12 h [29], while ginsenoside needed to be reacted at room temperature for 3 days [27]. Besides, the structures of some derivatives are not detailed. For example, when AS-IV was derivatized by excessive BC, there have been report of one hydroxyl group being derivatized [28] and there have also been reports of all hydroxyl groups being derivatized [30]. When using BC to derivatize ginseng triterpenes, all hydroxyl groups were derivatized [26,33]. In the derivatization by rhodamine, it was reported that only 3-OH other than 6-OH or 12-OH was involved in the reaction [37,39]. However, another work reported that only 20-OH participated in the reaction, while 3-OH did not, unless the amount of derivatization reagent was far excessive [38]. These phenomena indicate that the reactive sites and derivatives vary greatly with multiple factors, including the structures of derivatization reagent, the types of triterpenoids, the actual dosages, and the reaction time, etc. Relevant researchers have also noticed this aspect and usually optimize the reaction conditions through reasonable experimental design, mainly including single-factor experiments and response surface methodology (RSM). For example, the reaction parameters, including derivatization temperature, reaction time and reagent dosage, were optimized by RSM for the determination of four triterpene acids [44].

In recent years, stable isotope labelling derivatization has been developing increasing rapidly. It is a new type of sample pretreatment technique for LC-MS/MS with powerful capabilities in terms of sensitivity, accuracy, and qualitative and quantitative performance. However, limited by the synthesis and commercial sales of reagents, their application is still in a bottleneck period. It shows great potential applications in complex triterpenoids analysis and will be an interesting research field and direction of exploration. In addition, although a derivatization program has been developed for the analysis of triterpenoids by virtue of the 'bimodal phenomenon' of chromatography [52], mainstream derivatization methods still need to rely on mass spectrometry to ensure the accuracy of experimental results. A suggestion can be made in the selection of derivatization reagents: derivatization reagents containing ammonia nitrogen in the molecule should be selected as much as possible. Overall, with the continuous progress of basic research, more effective derivatization strategies are expected to be developed in the near future to improve the performance of triterpenoids in HPLC analysis.

CRediT author statement

Xiao-Feng Huang: Methodology, Validation, Investigation, Data curation, Writing - Original draft preparation; Ying Xue: Methodology, Validation, Data curation; Li Yong: Validation; Tian-Tian Wang: Data curation; Pei Luo: Writing - Reviewing and Editing, Supervision, Funding acquisition; Lin-Sen Qing: Writing -Reviewing and Editing, Investigation, Conceptualization, Supervision, Funding acquisition.

Declaration of competing interest

The authors declare that there are no conflicts of interest.

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Appendix A. Supplementary data

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