# SCIENTIFIC REPORTS

### **OPEN**

Received: 16 August 2018 Accepted: 5 November 2018 Published online: 27 November 2018

## Isolation, identification, structural elucidation and bioactivity of Heneicos-1-ene from *Coriandrum sativum* L. foliage

Siddharth Priyadarshi<sup>1,2</sup>, Nanishankar V. Harohally<sup>1,2</sup>, C. Roopavathi<sup>3</sup> & Madeneni Madhava Naidu<sup>1,2</sup>

Coriander foliage is a distinctive spice employed on a daily basis in curry and other Indian traditional food preparations mainly for the unique flavour attributes and health benefits. Radical scavenging activity has been demonstrated previously for coriander foliage. However, specific molecules responsible were not identified. A new molecule was isolated via chromatographic technique, and its structure was established by employing multinuclei and multidimensional NMR and HRMS techniques. The identified molecule Heneicos-1-ene was also screened for radical scavenging activity and antimicrobial activity, wherein it displayed radical scavenging activity of 89.6 ± 0.62% at 200 ppm, and also exhibited substantial antimicrobial activity against *E. coli* and *Salmonella typhi*.

*Coriandrum sativum* L. belongs to the family *Apiaceae*. It exists in two categories; small-fruited *C. sativum* L. var. *microcarpum DC* and large-fruited *C. sativum* L. var. *vulgare alef*<sup>1</sup>. The leaves, fruits and stems have a pleasant aromatic aroma hence used as a flavouring substance. The entire parts of the plant are suitable for eating, but mainly the dried seeds and fresh green leaves are used in various culinary preparation. The whole coriander plant is used in chutneys preparation, and the leaves, in particular, are used for flavouring many curries, sauces and soups. Coriander oleoresin and oils are used primarily in the bakery, beverages, chewing gums, condiments, curry mix, sweets, tobacco goods, seasonings for sausages and other meat products<sup>1</sup>. The various phytonutrients present in coriander endow with a range of pharmacological benefits such as stomachic, stimulant, lipolytic, insecticidal, hypolipidemic, hypoglycemic, fungicidal, diuretic diaphoretic cytotoxic, carminative, antispasmodic, antimutagenic, antibacterial, and aflatoxin protective potential<sup>2</sup>.

Previously our group has subjected the coriander foliage for radical scavenging activity<sup>3</sup> and found that ethanol extract had indeed exhibited commendable activity when compared with 95% of butylated hydroxyanisole (BHA). We envisioned to identify the molecule responsible for antioxidant activity. With this background, we set out to extract and identify the antioxidant molecule via chromatographic separation. We present here, isolation, and identification of Heneicos-1-ene from the coriander foliage as well as its radical scavenging and antimicrobial activities.

#### **Reagents and Chemicals**

Chemicals and solvents used for analysis were of analytical grade and procured from Merck, Mumbai, India.

#### **Materials and Methods**

Fresh coriander (*Coriandrum sativum* L.) foliage of *Microcarpum DC* (M.DC) variety was obtained from the local producer (Mysuru, Karnataka, India) on the day of the experiment. The collected green foliage was subjected to a pretreatment; roots and extraneous foreign materials were removed. The foliages were washed with fresh water to remove any dust and other adhering particles and then were spread on filter paper at room temperature to remove excess moisture. Washed foliages (1.5 Kg) were dried for 3.5 hours using low-temperature low humidity (LTLH) dryer maintained at  $52.6 \pm 2.0$  °C air temperature and  $28.0 \pm 3.0$ % relative humidity. The dried foliage was

<sup>1</sup>Academy of Scientific and Innovative Research (AcSIR), CSIR-CFTRI Campus, Mysuru, Karnataka, India. <sup>2</sup>Department of Spices and Flavour Sciences, CSIR-CFTRI, Mysuru, Karnataka, India. <sup>3</sup>Microbiology & Fermentation Technology, CSIR-CFTRI, Mysuru, Karnataka, India. Correspondence and requests for materials should be addressed to M.M.N. (email: mmnaidu@cftri.res.in) powdered in a mixer grinder (Panasonic, Model MX-AC400, Haryana, India) to 52 mesh size. The powder was placed in an airtight aluminium pouch and stored in the dark at room temperature for further analysis.

**Isolation, identification and structural elucidation.** Separation using thin layer chromatography. Isolation of the molecule was carried out by thin layer chromatography (TLC) from the extract prepared from dried coriander foliage using 70% ethanol as solvent. TLC aluminium plates  $(20 \times 20 \text{ cm})$  pre-coated with silica gel 60 F<sub>254</sub> were obtained from Merck (Darmstadt, Germany). Developing systems of different composition were tried and optimized to get good separation. Sample solutions were applied to the plates using a capillary tube. All TLC plates were developed using an initial mobile phase of methanol: chloroform (30:70 v/v) followed by twice elution with methanol: chloroform (10:90 v/v). Linear ascending plate development was performed in a suitable chromatographic tank previously saturated for one hour with the developing mobile phase, and the scanning was carried out at 270 nm. The bands obtained were carefully excised from the plate and extracted using 2 mL chloroform. All the chloroform extracts were then pooled; dried in a rotary evaporator (Heidolph, Germany) at 30–35 °C under reduced pressure (40 millibars) and redissolved in a known volume of chloroform.

*Spectroscopic measurement.* The extracts were subjected to high-resolution mass spectrometry (HRMS) and nuclear magnetic resonance (NMR) spectroscopy for structural elucidation and identification of the molecule. Solution NMR spectral data recorded on a Bruker Avance instrument having the <sup>1</sup>H frequency of 500 MHz and <sup>13</sup>C frequency of 125 MHz. NMR spectral analysis was accomplished by employing the combination of experiments consisting of 1D NMR techniques <sup>1</sup>H, <sup>13</sup>C{<sup>1</sup>H}, and 2D NMR techniques involving HSQC (heteronuclear single quantum coherence), HMBC (heteronuclear multiple bond correlation), and TOCSY (total correlation spectroscopy). Mass spectra were recorded in SCIEX 5600 Q-TOF instrument.

**Determination of free radical scavenging activity by DPPH.** The free radical scavenging activity of isolated Heneicos-1-ene from coriander foliage was measured by the 2, 2-diphenyl-1-picrylhydrazyl (DPPH) method<sup>4</sup>. The various concentrations (50, 100, and 200 ppm) of extracts were taken in different test tubes. The volume was made up to 1 mL by using respective solvent. Then 4 mL of 0.1 mM methanol solution of DPPH was added to each test tube and vortexed. The tubes were stored for 20 minutes in the dark at  $21 \pm 2$  °C temperature. Control was prepared as above without Heneicos-1-ene, and the respective solvent was used for the baseline correction. The variation in the absorbance of the samples was measured at 517 nm. BHA was used as a standard antioxidant for comparison. The average values of the three analyses were taken. The free radical scavenging activity was expressed as the inhibition percentage and was calculated using the following formula-

Radical scavenging activity (%) = 
$$\frac{(A_{control} - A_{sample})}{A_{control}} \times 100$$

where, A<sub>sample</sub> and A<sub>control</sub> are the equilibrium absorbance of the test sample and the control, respectively.

**Determination of antimicrobial activity.** *Disc diffusion assay.* The antibacterial activity assays were carried out using disk diffusion method<sup>5</sup> and spot on lawn method using *Staphylococcus aureus, Escherichia coli, Micrococcus luteus, Salmonella typhi* and *Listeria monocytogens* as test organisms. The Muller-Hinton agar plates inoculated with equal to 0.5 McFarland turbidity of pathogen bacteria by sterile swab. Standard blank disk (6.0 mm diameter) was put on a plate and inoculated with 50 µL of the isolated component (10 mg). All Petri plates were incubated for 24 hours at 37 °C. Zones of inhibition measured after 24 hours. The control discs contained DMSO (–ve) and Ampicillin (+ve).

Spot on lawn assay. The assay was carried out on Muller Hinton agar. Muller Hinton agar plates were inoculated with different pathogen bacteria matched to 0.5 McFarland turbidity of by sterile swab.  $10 \,\mu$ L of the isolated component was spotted on the media and incubated at 37 °C.

*Minimum inhibitory concentration (MIC) and Minimum bactericidal concentration (MBC).* The antimicrobial activity of the isolated compound was determined by adjusting the turbidity of the bacterial pathogen to match a McFarland No. 0.5 standard (~10<sup>7</sup> CFU/mL) and used for the assay<sup>6</sup>. A stock solution of purified compound (2 mg/mL) was prepared and serially diluted in a Muller Hinton broth (100 µL) medium dispensed in a 96-well cell culture plate to obtain a concentration of 100, 50, 25, 12.5, 6.25, 3.125, 1.56 and 0.781 µg/mL. Each well was further added with 100 µL of inoculum. The control wells were maintained with and without samples. Ampicillin (10 µg/mL) was used as reference antibiotic. The plates were sealed using parafilm and incubated for 24 hours at 37 °C. The turbidity of the broth was measured at 600 nm using titer plate reader (Thermo Scientific, USA). The minimum concentration of compound at which it inhibited the visible growth of pathogenic bacteria was taken as minimum inhibitory concentration (MIC). MBC is the lowest concentration of compound preventing the visual growth of the bacterial cultures and reduced the pathogen by ≥99.9%, irrespective of counts of survivors at higher antibiotic concentrations.

#### **Results and Discussion**

**Structural elucidation.** Structural elucidation was carried out by employing high-resolution mass spectrometry (HRMS) and <sup>1</sup>H, <sup>13</sup>C, HMBC, HSQC and TOCSY NMR experiments. HRMS ESI-positive mode spectrum did not reveal the occurrence of molecular ion peak. However, it exhibited a peak pattern corresponding to 302.3049 assigned to  $(M-CH_3 + Na)^+$ , whereas the dominant peak pattern{please refer Figure I under supplementary information} centered at 274.2739 was assigned to  $(M-C_3H_7 + Na)^+$ . To observe the molecular ion peak,



Figure 1. HRMS ESI-negative mode spectrum of Heneicos-1-ene.



Figure 2. <sup>13</sup>C NMR spectrum of Heneicos-1-ene.

we switched to HRMS ESI-negative mode spectrum (Fig. 1) which indicated a strong peak pattern centered at 293.1738 corresponding to  $(M-H)^{-1}$  wherein,  $M = C_{21}H_{42}$ .

<sup>1</sup>H NMR revealed {please refer Figure II under supplementary information} the occurrence of CH<sub>3</sub>, CH<sub>2</sub>, CH and magnetically not equivalent protons (indicated via splitting pattern) of CH<sub>2</sub> groups were observed at chemical shift of 0.88, 1.25, 5.82 and 4.96 ppm, respectively. An additional CH<sub>2</sub> was also seen at chemical shift 2.04 ppm. The integral of 1.25 ppm peak indicated the occurrence of several CH<sub>2</sub> peaks. On the other hand, <sup>13</sup>C NMR displayed (Fig. 2) peaks corresponding to chemical shift of 114.25, 22.84, 29.51, 29.84, 32.08, 33.97, 114.21 and 139.42 ppm. The occurrence of peaks at chemical shift of 114.21 and 139. 42 ppm confirmed the presence of the C=C double bond and indicated the chemical shift values typical of the terminal alkene. HSQC spectrum {please refer Figure III under supplementary information} facilitated the assignment of <sup>13</sup>C NMR for methyl and methylene groups.

Further, HBMC cross peaks (Fig. 3) were observed for the peak of <sup>1</sup>H chemical shift corresponding to 0.88 to 22.84 ppm and 32.94 ppm ( $^{13}$ C NMR) confirming the methyl correlation with adjacent CH<sub>2</sub> groups. Also, <sup>1</sup>H peak corresponding to chemical shift at 2.04 ppm displayed HMBC cross peaks with <sup>13</sup>C NMR peaks of chemical shift 114.21 and 139.42 ppm confirming the correlation of CH<sub>2</sub> group adjacent with CH and terminal CH<sub>2</sub> groups. This was also substantiated by cross peaks of <sup>1</sup>H NMR peaks having the chemical shift of 4.96 and 5.82 ppm with 33.97 ppm of <sup>13</sup>C NMR. Further, TOCSY experiments {please refer Fig. IV under supplementary information} established the correlations via observation of various peaks relating to spin systems and confirmed the molecule to be Heneicos-1-ene (Fig. 4). Heneicos-1-ene is synthetically made by decarbonylation reaction of docosanoic acid (CH<sub>3</sub>(CH<sub>2</sub>)<sub>20</sub>COOH) with Vaska's complex utilizing potassium iodide and acetic anhydride as additive<sup>7</sup>. It has also been made using enzymatic oxidative decarboxylation<sup>8</sup>. This molecule forms the part of essential oil of blue-coloured hybrid tea rose flowers<sup>9</sup>.

**NMR Data.** <sup>1</sup>H NMR(CDCl<sub>3</sub>): 0.88 (t, J = 6.9 Hz, 3 H, C(21)H<sub>3</sub>), 1.25(m, C(3)H-C(19)H<sub>2</sub>), 2.04(m, 2 H,C(20) H<sub>2</sub>), 4.96(m, 2 H,C(1)H<sub>2</sub>), 5.82(m, 1 H,C(2)H).

<sup>13</sup>C NMR(CDCl<sub>3</sub>): 14.25(C(21)), 22.54(C(20)), 32.08(C(19)), 29.51(C(18)), 29.84(C(4)-C(17), 33.97(C(3)), 114.21(C(2)), 139.42(C(1)).



Figure 3. The HMBC spectrum of Heneicos-1-ene.



Figure 4. Structure of molecule identified: Heneicos-1-ene.





Bacterial pathogens	Chloroform extract (10 mg/mL)	Ampicillin§ (25µg/mL)
E. coli	$8.9 \pm 0.079  \text{mm}$	$23.3 \pm 0.9  mm$
Salmonella typhi	$11\pm0.025mm$	$22.8\pm0.7mm$

**Table 1.** The zone of inhibition of Heneicos-1-ene and antibiotic ampicillin tested against bacterial pathogens.

 §6 mm disc as standard

**Radical Scavenging Activity.** The isolated non-volatile compound Heneicos-1-ene from coriander foliage and BHA at various concentrations (50, 100 and 200 ppm) were prepared by dissolving in the chloroform. As can be observed from Fig. 5; the maximum radical scavenging activity of  $89.6 \pm 0.62\%$  was attained at 200 ppm, and activity decreased with decreasing concentration of compound when compared with 95% of BHA. Heneicos-1-ene exhibited comparable radical scavenging activity when compared with the artificial antioxidant BHA.

*In-vitro* **antibacterial activity.** Antibacterial activity of both aqueous crude extract and pure compound (Heneicos-1-ene) dissolved in chloroform were tested *in-vitro* by using disc diffusion method, spot on the lawn and minimal inhibition concentration method. Heneicos-1-ene has inhibited the growth of *E. coli* and *Salmonella typhi* at the concentration of 10 mg with a zone of inhibition  $8.9 \pm 0.079$  mm and  $11 \pm 0.025$  mm, respectively (Table 1). The zone of inhibition of the Heneicos-1-ene was more prominent against *Salmonella typhi* than *E. coli* (Fig. 6). There was no significant antibacterial activity observed against the other three tested bacterial pathogens





.....

	Test Microorganisms MIC/MBC (µg/mL)	
Sample	E. coli	Salmonella typhi
Ampicillin	0.312/0.625	1.25/0.625
Chloroform extract	50/100	25/50

Table 2. The MIC and MBC values of Heneicos-1-ene and antibiotics tested against bacterial pathogens.

such as *Staphylococcus aureus*, *Listeria monocytogens* and *Micrococcus luteus*. However, the aqueous crude extract was not effective in inhibiting the growth of any pathogens used in the study.

With regards to the antibacterial activity, the Minimal Inhibition Concentration (MIC) values of Heneicos-1-ene against *E. coli* and *Salmonella typhi* of the compound were confirmed using broth dilution microtiter system and expressed data as the mean  $\pm$  standard deviation (SD, n = 3). The antimicrobial activity of Heneicos-1-ene was found in the range of 25 to 50 µg/mL and the minimal bacterial concentration (MBC) was in the range of 50–100 µg/mL (Table 2) for *E. coli* and *Salmonella typhi*. MBC were generally within a two-fold dilution of the MICs. Since the Heneicos-1-ene did not show any inhibition zone against other pathogens, the MIC and MBC were not determined.

Several bioactive peptides, volatile compounds<sup>10</sup> and essential oils<sup>11,12</sup> isolated from coriander have been accounted for a broad range of antimicrobial activity. However, our study for the first time characterizes and confirms the antimicrobial activity of Heneicos-1-ene against *E. coli* and *Salmonella typhi* and possesses a promising role in maintaining the shelf life of the food products by preventing spoilage.

#### Conclusion

We isolated Heniocose-1-ene from a natural food source coriander foliage for the first time. Heniocose-1-ene exhibited excellent radical scavenging and antimicrobial activity.

#### **Data Availability Statement**

Complete data of the manuscript is available with Corresponding author.

#### References

- 1. Rajeshwari, U. & Andallu, B. Medicinal benefits of coriander (Coriandrum sativum L). Spatula DD 1, 51-58 (2011).
- Benjumea, D., Abdala, S., Luis, F. H., Paz, P. P. & Herrera, D. M. Diuretic activity of Artemisia thuscula, an endemic canary species. Journal of Ethnopharmacology 100, 205–209 (2005).
- Priyadarshi, S., Khanum, H., Ravi, R., Borse, B. B. & Naidu, M. M. Flavour characterisation and free radical scavenging activity of coriander (*Coriandrum sativum* L.) foliage. *Journal of Food Science and Technology* 53, 1670–1678 (2016).
- Oktay, M., Culcin, I. & Kufrevioglu, O. I. Determination of *in vitro* antioxidant activity of fennel (*Foeniculum vulgare*) seed extracts. *LWT - Food Science and Technology* 36, 263–271 (2003).
- Balachandran, C. et al. Synthesis and medicinal properties of plant-derived vilangin. Environmental Chemistry Letters 11, 303–308 (2013).
- Sarker, S. D., Nahar, L. & Kumarasamy, Y. Microtitre plate-based antibacterial assay incorporating resazurinas an indicator of cell growth, and its application in the *in vitro* antibacterial screening of phytochemicals. *Methods* 42, 321–324 (2007).
- 7. Maetani, S., Fukuyama, T., Suzuki, N., Ishihara, D. & Ryu, I. Efficient Iridium-Catalyzed Decarbonylation Reaction of Aliphatic Carboxylic Acids Leading to Internal or Terminal Alkenes. *Organometallics* **30**, 1389–1394 (2011).
- Dennig, A. et al. Oxidative Decarboxylation of Short-Chain Fatty Acids to 1-Alkenes. Angewandte Chemie International Edition 54, 8819–8822 (2015).
- 9. Joichi, A. et al. Volatile constituents of blue-coloured hybrid tea rose flowers. Flavour and Fragrance Journal 28, 180-187 (2013).

- Kubo, I., Fujita, K. I., Kubo, A., Nihei, K. I. & Ogura, T. Antibacterial Activity of Coriander Volatile Compounds against Salmonella choleraesuis. Journal of Agricultural and Food Chemistry 52, 3329–3332 (2004).
- 11. Hammer, K. A., Carson, C. F. & Riley, T. V. Antimicrobial activity of essential oils and other plant extracts. *Journal of applied microbiology* **86**, 985–990 (1999).
- 12. Mandal, S. & Mandal, M. Coriander (*Coriandrum sativum* L.) essential oil: Chemistry and biological activity. Asian Pacific Journal of Tropical Biomedicine 5, 421–428 (2015).

#### Acknowledgements

The authors acknowledge Director, CSIR-CFTRI, Mysuru for the facilities provided to carry out the study. This investigation was monetarily upheld under 12<sup>th</sup> plan project (Agropathy, BSC-0105). First author Siddharth Priyadarshi would like express gratitude toward Academy of Scientific and Innovative Research (AcSIR), New Delhi, India for registering his Ph.D and University Grants Commission (UGC), New Delhi, India for granting Rajiv Gandhi National Fellowship (RGNF), 2012–13.

#### **Author Contributions**

Siddharth Priyadarshi. Carried out the work of Isolation, identification, and bioactivity of Heneicos-1-ene from *Coriandrum sativum* L. foliage; Nanishankar V. Harohally. Structural elucidation by Spectroscopy; Roopavathi C. Antimicrobial properties; Madeneni Madhava Naidu. Experimental Design, Overall co-ordination, manuscript preparation and communication.

#### **Additional Information**

Supplementary information accompanies this paper at https://doi.org/10.1038/s41598-018-35836-z.

Competing Interests: The authors declare no competing interests.

**Publisher's note:** Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

**Open Access** This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit http://creativecommons.org/licenses/by/4.0/.

© The Author(s) 2018