# **Green Synthesis to Develop Iron-Nano Formulations and Its Toxicity Assays**

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Objectives: In the past few years, herbal medicines have gained popularity over synthetic drugs because of their natural source and minimal side effects which has led to a tremendous growth of phytopharmaceuticals usage. With the development of nanotechnology, it provides alternative approaches to overcome several limitations using nano-formulations. In spite of considerable quantity of antianemic preparations with different iron forms available, currently additives are used and represented in modern pharmaceutical market. Iron deficiency anemia is a major global public health problem which particularly affects pregnant women, children and elderly persons. The situation is complicated because of disadvantages and drug side effects from existing antianemic medicines. There is a great demand for the development of new antianemic preparations. Green synthesis of iron oxide nanoparticles, possess high potential in this field.

Methods: Our study focuses on developing green synthesis of iron oxide nanoparticles (IONPs) of 10-50 nm with spherical shape where different dosages were used -1 mg/kg, 10 mg/kg and 100 mg/kg for exposure in Wistar albino female rats for 28 days. The toxicity was assessed using various parameters such as measurements of the rat body and organ mass, hematology, biochemical evaluation and histopathological examinations.

Results: No significant differences were observed in body and organ weights. Hematological indices also indicated no significant differences whereas biochemical factors showed increase in levels of direct bilirubin and globulin of medium as well as high dose and SGPT levels were increased only in high dose. The major organs (heart, kidney and liver) showed histopathological alterations in 10 and 100 mg/kg whereas brain showed only in 100 mg/kg. Conclusion: The toxicity of IONPs was found to be more significant when the concentration was increased; however, low doses can be used for further investigation as an antianemic preparation.

Keywords: iron oxide nanoparticles, biosafety, antianemic preparation

## INTRODUCTION

Iron deficiency anemia (IDA) is one of the most common nutritional disorders in the world prevailing in both developed and underdeveloped countries according to the World Health Organization (WHO) data. It is caused by prolonged low levels of iron in the blood may be due to a variety of factors like irondeficient diet, worm infestations, blood loss due to bleeding, intestinal diseases. This condition is mostly significant for the most vulnerable groups of population among which women of reproductive age, pregnant women, children and elderly persons [1, 2].

Iron deficiency in developing countries; rely mainly on food crops and plant-based diets. These consist of high level of nonheme iron, which is poorly absorbed in the body as compared to heme iron, found in animal tissues is another reason for iron

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deficiency in the developing countries [3]. The presence of absorption inhibitors like phytic acids, phenolic compounds, and calcium in cereals and pulses also results in the low absorption of iron in the blood. Pharmaceutical market has considerable series of antianemic preparations which relies on use of iron salts like ferrous sulphates, ferrous citrate etc. They are highly reactive and are known to cause severe effects on the gastrointestinal tract and also affect the natural microbiota of the digestive system [4, 5].

Hence, there is a requirement for a new kind of a therapy or iron formulation, which can improve the bioavailability of iron in human body by increasing the absorption of iron formulation to treat iron deficiency efficiently. To overcome all the restriction of literature; the researchers, enterprises are passionately investing in nanotechnology. The study of nanotechnology has evolved as a potential to revolutionize idea in the field of science and many more. In addition to this, nanotechnology is one of the leading targeted nutrients, minerals, molecules, or drugs delivery systems, it has a greater credit value in this field. The purpose of our research work, is to focus on the green synthesis of iron oxide nanoparticles which is bio-safe and biocompatible and to study its potential as an antianemic pharmacological substance for new class of drugs. Antianemic preparations development was the benchmark of this work, after studying the toxicity on synthesized iron nanoparticles.

### MATERIALS AND METHOD

#### 1. Synthesis of iron oxide nanoparticles (IONPs)

Plant extracts were formulated with ferric chloride solution for the synthesis of nanoparticles. For the synthesis of iron oxide nanoparticles, hydroponically grown *Spinaciaoleracea* (commonly called as Spinach) leaf extract was used as the reducing agent and mixed in a solution of FeCl<sub>2</sub>, FeCl<sub>3</sub> and NaOH. There was an immediate color change after the addition of leaf extract to the solution. The solution was centrifuged and the supernatant was discarded and followed by pellet formation. Then pellet was washed with ethanol and centrifuged again to remove any impurities (Fig. 1) [6-8].

#### 2. Characterization of iron oxide nanoparticles

The physicochemical properties of nanoparticles are important for their bio-distribution, safety, and efficacy. Character-



Figure 1. Green synthesized iron oxide nanoparticles.

ization is performed using a variety of analytical techniques, including Scanning Electron Microscopy (SEM – JEOL JSM-7600F) for surface morphology and size of the nanoparticles. The X-ray diffraction (XRD – D8 ADVANCE) measurement was performed on X-ray diffractometer (Panalytical Xpert-PRO 3050/60) operated at 30 kV and 100 mA and spectrum was recorded by CuK $\alpha$  radiation with wavelength of 1.5406 Å in the 2 $\theta$  range of 20°-80° andiron content determination was done using Atomic Absorption Spectroscopy (AAS – PEAnalyst800) [6-12].

#### 3. Treatment of rats

180-200 gram of Wistar female albino rats were purchased from 'Bombay Veterinary College', Parel, Mumbai and were acclimatized in our institutional animal house for 1 week before the experiment. They were maintained under standard conditions of 12/12 h light/dark cycle at  $21^{\circ}C \pm 3^{\circ}C$  temperature and 30-70% relative humidity and fed with standard feed and water. All studies were approved by the Institutional Animal Ethics Committee – Approval No.: 2016/7/12.Sub-acute toxicity study of biosynthesized Fe<sub>2</sub>O<sub>3</sub> nanoparticles was carried out for 28day with repeated oral dose of 1 mg/kg, 10 mg/kg and 100 mg/ kg. The rats were analyzed for various parameters such as body weight, organ weight, hematological indices, biochemical analysis and histopathological studies which were compared with the control group of rats.

#### 4. Toxicity assessment of IONPs on rat model

Wistar albino female rats were randomly divided into five groups, including control, vehicle control, low dose (1 mg/kg),

medium dose (10 mg/kg) and high dose (100 mg/kg) (n = 6 for each group). Here, control rats were fed with saline and vehicle control group with the solution of CMC, FeCl<sub>3</sub>, FeCl<sub>2</sub> and NaOH. IONPs were fed with three different doses low (1 mg/kg), medium (10 mg/kg) & high (100 mg/kg) for 28 days.

## 5. Statistical analysis

Data represent mean and standard deviation (SD) of each group of experimental animals. Group mean values of each experiment were compared by the paired *t*-test. p < 0.05 was considered to be the level of significance.

## RESULTS

The obtained results using scanning electron microscopy analysis clearly show that IONPs have spherical shape with average diameter of about 10-50 nm (Fig. 2A). The IONPs synthesized from hydroponically grown spinach extract has estimated 40.34% of Fe content whereas the corresponding plant extract has estimated Fe content in 0.0007% ppm level using Atomic absorption spectroscopy. XRD analysis was carried out for structural characterization of biosynthesized nanoparticles. XRD of IONPs has showed two peaks – 33.09°20 and 35.56°20. Both of these peaks denote presence of hematite (Fe<sub>2</sub>O<sub>3</sub>) structure (Fig. 2B).

One of the main purposes of the current study was to evaluate the potential toxicity of IONPs. In order to obtain detailed information of toxicity, we performed *in vivo* serum biochemical analysis, blood chemistry and histopathological examination.

#### 1. Body & organ weight

The weight of rats from each group were recorded before starting the treatment i.e. on day 0 and after treatment i.e. on day 28. On sacrifice, no significant differences in body weights and organ weights were recorded compared to control ones (n = 6) (Fig. 3).

#### 2. Hematological analysis

Several important hematology markers including the white blood cell count (WBC), red blood cell count (RBC), hemoglobin (HB), mean corpuscular hemoglobin (MCH) and mean corpuscular hemoglobin concentration (MCHC) etc were recorded at day 0 and 28, which were selected for further toxicity and compared with the control rat observations presented in Table 1.

#### **3. Biochemical examination**

Oral administration of IONPs induced marked increase of direct bilirubin and globulin levels of samples administered with medium as well as high dose whereas SGPT levels show significant differences only in high dose. No significant differences were observed in renal function test (Table 2) [14].

#### 4. Histopathological studies

Rat tissue sections were prepared and observed under a light microscope for morphological changes following exposure to 1 mg/kg, 10 mg/kg and 100 mg/kg of IONPs. The histological examinations of rat when exposed with 10 mg/kg revealed



Figure 2. (A) Representing SEM images of green synthesized nanoparticles showing 10-50 nm in size and spherical in shape and (B) showing the state of nanoparticles as hematite (IONPs) [8, 13].



**Figure 3.** (A-D) Body weight and relative organ weight before and after oral administration of IONP at 0 and 28 days. Data are expressed as means  $\pm$  SD of the mean (n = 6).  $\alpha$  is significant at p < 0.05 relative to the control.

Table 1. Effect of different dosage of IONPs on hematological indices (WBC, RBC, HGB, HCT, MCV, MCH, MCHC and PLT) in Wista	r
rat at day 28	

Tests	Control	Vehicle control	Low dose	Medium dose	High dose
WBC (*10 <sup>3</sup> /µL)	8.4 ± 1.1	9.2 ± 4.3	6.7 ± 1.9	8.2 ± 1.5	12 ± 6.7
RBC (*10 <sup>6</sup> / $\mu$ L)	7.7 ± 0.3	7.6 ± 0.3	7.2 ± 0.8	8.2 ± 1.2	8.1 ± 1.9
HGB (g/dL)	16.2 ± 1.5	13.5 ± 0.7	13.3 ± 1.8	14.5 ± 2.3	13.8 ± 1.5
HCT (%)	46.7 ± 1.8	50.5 ± 2.6	49.9 ± 6.1	56.2 ± 8.0	47.2 ± 5.4
MCV (fL)	60.3 ± 2.7	66.8 ± 0.8	69.2 ± 1.8	68.2 ± 0.2	62.7 ± 0.6
MCH (pg)	18.4 ± 0.3	17.8 ± 0.2	18.4 ± 0.8	17.6 ± 0.4	18.6 ± 0.5
MCHC (g/dL)	31.8 ± 1.6	26.7 ± 0.1	26.7 ± 0.6	28.1 ± 0.6	29.2 ± 0.4
PLT (*10 <sup>3</sup> /μL)	871 ± 5.4	789 ± 4.6	939 ± 2.3	777 ± 5.2	638 ± 3.1

vacuolar degeneration of cardiomyocytes in the heart (Fig. 4N), Kidney shows dilated and congested blood vessels (Fig. 4O) and liver shows nuclear degeneration of hepatocytes (Fig. 4P). These inimical cellular alterations were non-incident in the control group as well as in the vehicle group (Fig. 4A-H). Whereas exposure of 100 mg/kg shows abnormal histopathological gross lesions in all major organs like brain shows congested blood vessels and increased nuclear condensation, heart and kidney shows congested blood vessels and renal tissues. Liver shows congested blood vessels and mononuclear cell infiltration (Fig. 4Q-T). The results of this experimental study indicated that IONPs in mentioned concentration (10 mg/kg, 100 mg/kg)

Tests	Control	Vehicle control	Vehicle control Low dose Med		High dose	
LFT						
Total bilirubin (mg/dL)	0.4 ± 0.05	$0.4 \pm 0.1$	$3.2 \pm 0.1$	0.9 ± 0.3	0.9 ± 0.5	
Direct bilirubin (mg/dL)	0.4 ± 0.09	$0.5 \pm 0.05$	2.1 ± 3.0	0.6 ± 0.2	0.5 ± 0.1	
SGOT (mg/dL)	237.2 ± 85.1	248 ± 18.4	272.6 ± 2.1	336 ± 33.0	262.5 ± 34.8	
Total protein (g/dL)	7.9 ± 0.4	8.5 ± 0.6	9.4 ± 2.1	7.5 ± 0.7	8.2 ± 0.3	
Albumin (g/dL)	4 ± 0.5	4.3 ± 0.2	6.9 ± 0.5	6.6 ± 0.2	5.5 ± 0.2	
Globulin (g/dL)	3.7 ± 0.2	$4.2 \pm 0.5$	$2.5 \pm 0.4$	2.64 ± 0.5	2.6 ± 0.1	
Alkaline phosphate (mg/dL)	168.1 ± 33.8	93.9 ± 21.5	50.9 ± 51.8	129.5 ± 31.4	99.5 ± 26.6	
SGPT (mg/dL)	55.8 ± 2.9	52 ± 3.6	64.5 ± 4.4	72.3 ± 6.5	42.3 ± 2.5	
RFT						
BUN (mg/dL)	16.3 ± 2.9	15.3 ± 1.5	20.7 ± 0.6	17.3 ± 1.0	21.7 ± 1.6	
Urea (mg/dL)	2.6 ± 6.4	31.4 ± 1.2	44.3 ± 1.2	37 ± 2.0	46.5 ± 3.4	
Creatinine (mg/dL)	1.1 ± 0.2	$1.1 \pm 0.2$	$0.9 \pm 0.1$	$0.9 \pm 0.1$	$1.2 \pm 0.1$	
Uric acid (mg/dL)	3.6 ± 1.1	3.8 ± 0.3	8.7 ± 2.0	6.4 ± 1.5	3 ± 1.1	

Table 2.	Effect of	different	dosage of	f IONPs or	n the activ	itv of LF1	& RF	T in Wist	ar rat a <sup>r</sup>	t dav	28
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shows significant effect on the histological analysis.

## DISCUSSION

The aim of the present study was to assess and investigate the toxic impacts of different doses of our biosynthesized IONPs on Wistar albino female rats. The IONPs were successfully synthesized by biological method using *Spinacia oleracea* leaves extract and characterized using sophisticated instruments which revealed 10-50 nm sized particles with spherical shape by SEM. XRD revealed hematite (Fe<sub>2</sub>O<sub>3</sub>) state of nanoparticles and AAS estimated 40.34% Fe content in the sample. Whenever, there is usage of nanoparticles, the main concern is its toxicity influence. Hence, in this study, the synthesized IONPs were fed in three different doses – 1, 10 and 100 mg/kg for 28 days. To evaluate the impact of toxicity in rats various parameters were studied such as body and organ weight, hematological indices, biochemical studies and histopathological studies.

The effects of IONPs on body and organ weight of rats does not show any loss or gain in weight. Kumari et al, revealed neither loss in body weight nor any abnormal signs or symptoms on repeated administration of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> NP for 28 days similar to that of our results [15]. However, another study showed contrast findings to ours as Wang et al observed significant weight loss and gastro-intestinal damage with severe symptoms [16]. With respect to hematological indices, the evaluation did not show any significant differences in either of the doses. These findings are in accordance with observations of our previous study [13]. No obvious signs of inflammatory responses or toxic reactions that could be attributed to these doses were found [17-19]. Whereas for biochemical factors, present findings are in contrast with observations of our previous work [13]. In accordance to our findings, Gillian Federici et al revealed no significant alterations were found in hematological profile but contrast with our findings with respect to biochemical parameters such as blood cells count and enzymes activity [20].

Studies of nanotoxicology performed in vivo using correct doses and exposure routes are of greater significance due to the complexity of systemic phenotypic reaction and the physiological/anatomic effect than can be translated from animal model to human exposure. Hence, it was also point of concern to investigate the effects of toxicity at the tissue level. The histopathological studies of major organs - heart, kidney and liver showed abnormal alterations when compared to the control rats in 10 and 100 mg/kg doses and brain in 10 mg/kg dose. In contrast to our findings, higher accumulation of IONPs (dose of 1000 mg/kg for 28 days) was found in liver of rats causing congestion of portal tract whereas kidneys and brain were not affected [21]. Other literature demonstrated the assessment of superparamagnetic iron oxide nanoparticles on SD rats that revealed no alterations in histopathological and biochemical parameters [22].

Hematological profile and biochemical parameters such as blood cells count and enzymes activity are used to provide use-



Figure 4. Histological evaluation of IONPs on various tissues. Representative images of brain, heart, kidney and liver of control, vehicle control and IONPs (1 mg/kg, 10 mg/kg, 100 mg/kg) treated groups where organs collected after 28 days of oral administration. Sections were stained with H&E and observed under a light microscope at 10X & 40X magnification.

ful information for diagnosis in routine clinical evaluation of the state of health of a patient and animals. There were no abnormal findings at hematological parameters and biochemical parameters except direct bilirubin, globulin and SGPT levels in treated-rats. The unchanged levels of BUN, urea, creatinine and uric acid in the present study indicated normal renal functions in the IONPs exposed group. However, the oral administration of a moderate and high dose of IONPs (10 mg/kg, 100 mg/kg) induced a marked increase of toxicity at tissue level [23, 24].

# CONCLUSION

The present study focused on the green synthesis and characterization of IONPs from spinach and its toxicity assessment with the exposure of different dosages to experimental rats. An effective method for green synthesis of metallic IONPs using leaf extracts of spinach was established and was characterized by SEM, XRD and AAS. SEM results revealed that the obtained particles were spherical in shape with an average diameter of 10-50 nm. XRD peaks indicated that the synthesized nanoparticles were found to be in  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> state and crystalline in nature. AAS analysis confirms content of iron. In the present study, different dosages were used 1 mg/kg, 10 mg/kg and 100 mg/kg for exposure in Wistar albino female rats for 28 days. Based on the measurements of the rat body and organ mass, haematology, biochemical evaluation and histopathological examinations indicated toxicity for 10 mg/kg and 100 mg/kg. Therefore, we can conclude that though, the toxicity of IONPs is more significant when the concentration is increased; less than 10 mg/kg dose can be used for further investigation as an antianemic preparation at the pharmaceutical industry in the near future. The derived green biosynthesis of IONPs protocol by utilizing plant extracts has several advantages over chemical synthesis: (1) sustainable availability of materials for large scale synthesis of NPs; (2) plant extract does not contain toxic substances, and (3) various bioactive molecules availability in different plant species.

# **CONFLICT OF INTEREST**

The author declares no conflict of interest.

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