

BRIEF REPORT



Activation studies with amino acids and amines of a β -carbonic anhydrase from *Mammaliicoccus (Staphylococcus) sciuri* previously annotated as *Staphylococcus aureus* (SauBCA) carbonic anhydrase

Andrea Angeli^a , Linda J Urbański^b, Clemente Capasso^c , Seppo Parkkila^{b,d}  and Claudiu T. Supuran^a 

^aDipartimento Neurofarba, Sezione di Scienze Farmaceutiche e Nutraceutiche, Università degli Studi di Firenze, Sesto Fiorentino (Florence), Italy; ^bFaculty of Medicine and Health Technology, Tampere University, Tampere, Finland; ^cDepartment of Biology, Agriculture and Food Sciences, Institute of Biosciences and Bioresources, Napoli, Italy; ^dFimlab Ltd, Tampere University Hospital, Tampere, Finland

ABSTRACT

A β -carbonic anhydrase (CA, EC 4.2.1.1) previously annotated to be present in the genome of *Staphylococcus aureus*, SauBCA, has been shown to belong to another pathogenic bacterium, *Mammaliicoccus (Staphylococcus) sciuri*. This enzyme, MscCA, has been investigated for its activation with a series of natural and synthetic amino acid and amines, comparing the results with those obtained for the ortholog enzyme from *Escherichia coli*, EcoCA β . The best MscCA activators were D-His, L- and D-DOPA, 4-(2-aminoethyl)-morpholine and L-Asn, which showed K_{AS} of 0.12–0.89 μ M. The least efficient activators were D-Tyr and L-Gln (K_{AS} of 13.9–28.6 μ M). The enzyme was also inhibited by anions and sulphonamides, as described earlier. Endogenous CA activators may play a role in bacterial virulence and colonisation of the host which makes this research topic of great interest.

ARTICLE HISTORY

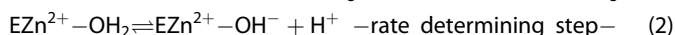
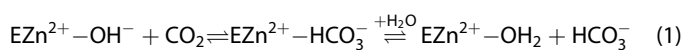
Received 29 August 2022
Revised 27 September 2022
Accepted 28 September 2022

KEYWORDS

Staphylococcaceae; carbonic anhydrase; activator; amine/ amino acid; *Mammaliicoccus (Staphylococcus) sciuri*

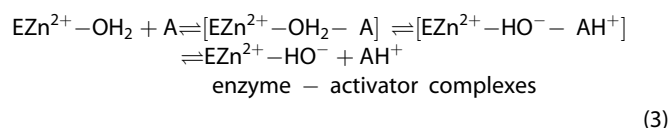
1. Introduction

Carbonic anhydrases (CAs, EC 4.2.1.1), the enzymes which catalyse the interconversion between CO₂ and bicarbonate according to Equations (1) and (2), are widespread in all life kingdoms, including Bacteria^{1–5}. Of the eight genetically distinct CA families known to date, at least four (α -, β -, γ - and ι -CAs) are present in these organisms, in which they play crucial roles related to metabolism, pH regulation, acclimation in different niches in which bacteria grow, but also pathogenesis and virulence in the case of pathogenic species^{4–6}.







Inhibition of CAs belonging to various classes and organisms has been exploited pharmacologically for various applications for the last decades, mainly by targeting human CA (hCA) isoforms, of which 15 are presently known^{7–11}. Many such isoforms are targets for diuretics, antiobesity, antiepileptic, antiglaucoma or antitumor agents^{7–11}. Inhibition of such enzymes from pathogenic bacteria, fungi or protozoans was on the other hand proposed as a new approach to develop anti-infectives with novel mechanisms of action, devoid of the drug resistant problems of the currently used agents^{5,11}. Thus, a large number of drug design studies of CA inhibitors (CAIs) targeting both mammalian and pathogenic CAs are constantly being reported, mainly based on the tail approach developed by one of our groups over the last two decades¹².

On the other hand, activation studies of various classes of CAs have progressed slower compared to the inhibition studies. The CA activation mechanism was definitively demonstrated at the molecular level only in 1997 with the report of the first X-ray crystallographic adduct of a CA – activator complex, more precisely hCA II complexed with histamine¹³. Thus, Briganti et al.¹³ demonstrated that CA activators (CAAs) participate directly in the enzyme catalytic cycle, as shown schematically in Equation (3), binding in a different binding site compared to the classical sulphonamide inhibitors, i.e. at the entrance of the cavity^{6,13}.



Presently, a large number of activation studies of all hCAs are available with many classes of compounds, and several crystallographic and drug design studies were also reported^{14–17}. Furthermore, CAAs may have pharmacological applications for memory therapy as well as for the treatment of cognitive disorders in need of effective therapies¹⁸. Although this field is still in its infancy, crucial advances have been made over the last few years in understanding the connections between fear, extinction/social memory and CA activation/inhibition^{17,18}.

Non-mammalian CAs activation, mainly described in fungal and bacterial pathogens started to be investigated only in the last years, in order to understand whether endo- or exogenic modulators of this enzymatic activity may interfere with virulence, metabolism or pathogenicity of these organisms^{19–21}. Indeed, CAs from

CONTACT Seppo Parkkila  seppo.parkkila@tuni.fi  Faculty of Medicine and Health Technology, Tampere University, Tampere, Finland; Fimlab Ltd, Tampere University Hospital, Tampere, Finland; Claudiu T. Supuran  claudiu.supuran@unifi.it  Faculty of Medicine and Health Technology, Tampere University, Tampere, Finland

© 2022 The Author(s). Published by Informa UK Limited, trading as Taylor & Francis Group.

This is an Open Access article distributed under the terms of the Creative Commons Attribution License (<http://creativecommons.org/licenses/by/4.0/>), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

fungi such as *Malassezia globosa*, *Saccharomyces cerevisiae*, *Candida albicans*, *Cryptococcus neoformans*, etc., or bacteria such as *Vibrio cholerae*, *Mycobacterium tuberculosis*, *Francisella tularensis*, *Brucella suis*, *Escherichia coli*, etc., were recently investigated for their activation profiles with natural and synthetic amines and amino acid derivatives^{19–21}.

Among the pathogens investigated ultimately for the presence of druggable CAs, was *Staphylococcus aureus*, a bacterium known for its virulence and easy development of drug resistance to a variety of clinically used antibiotics². In 2016 we identified in the NCBI database a sequence annotated as encoding for a β -CA in the genome of *S. aureus*, which we cloned, characterised and showed to be susceptible to inhibition with sulphonamides and anions, two of the most investigated classes of CAIs². This enzyme, denominated SauBCA, showed the typical behaviour of a bacterial β -CA, possessing a significant CO₂ hydrase catalytic activity, similar to those of other such enzymes described earlier in *E. coli*, *M. tuberculosis*, *Salmonella enterica* (serovar *Typhimurium*), and many other pathogenic bacteria by us and other groups^{1–5}. However, a recent reinvestigation of the database showed that the initial

annotation was erroneous, and that the sequence thought to belong to the genome of *S. aureus*, was in fact from another species of this genus, *Staphylococcus sciuri*²². To make things even more complicated, recently *S. sciuri* has been moved to another taxon, *Mammaliococcus sciuri*²³. *Mammaliococcus (Staphylococcus) sciuri*, is known as a Gram-positive, oxidase-positive, coagulase-negative member of these infectious bacteria, provoking disease in humans and animals (it was originally isolated from the squirrel)²². In fact, the taxonomy of the *Staphylococcaceae* family is rather complex, and as mentioned earlier, many genome annotations were inexact or were overlapping between various genetically similar species²³. However, all these bacteria provoke diseases in humans and animals and show variable (usually high) degrees of resistance to clinically used antibiotics²².

Here we report an activation study of the β -CA previously known as SauBCA, and now renamed here as MscCA, with a series of amino acids and amines of types 1–24 (Figure 1) previously investigated as activators of other classes of CAs, including several bacterial such enzymes^{19–21}. We also compare the obtained results with those for a similar β -class enzyme from the model organisms

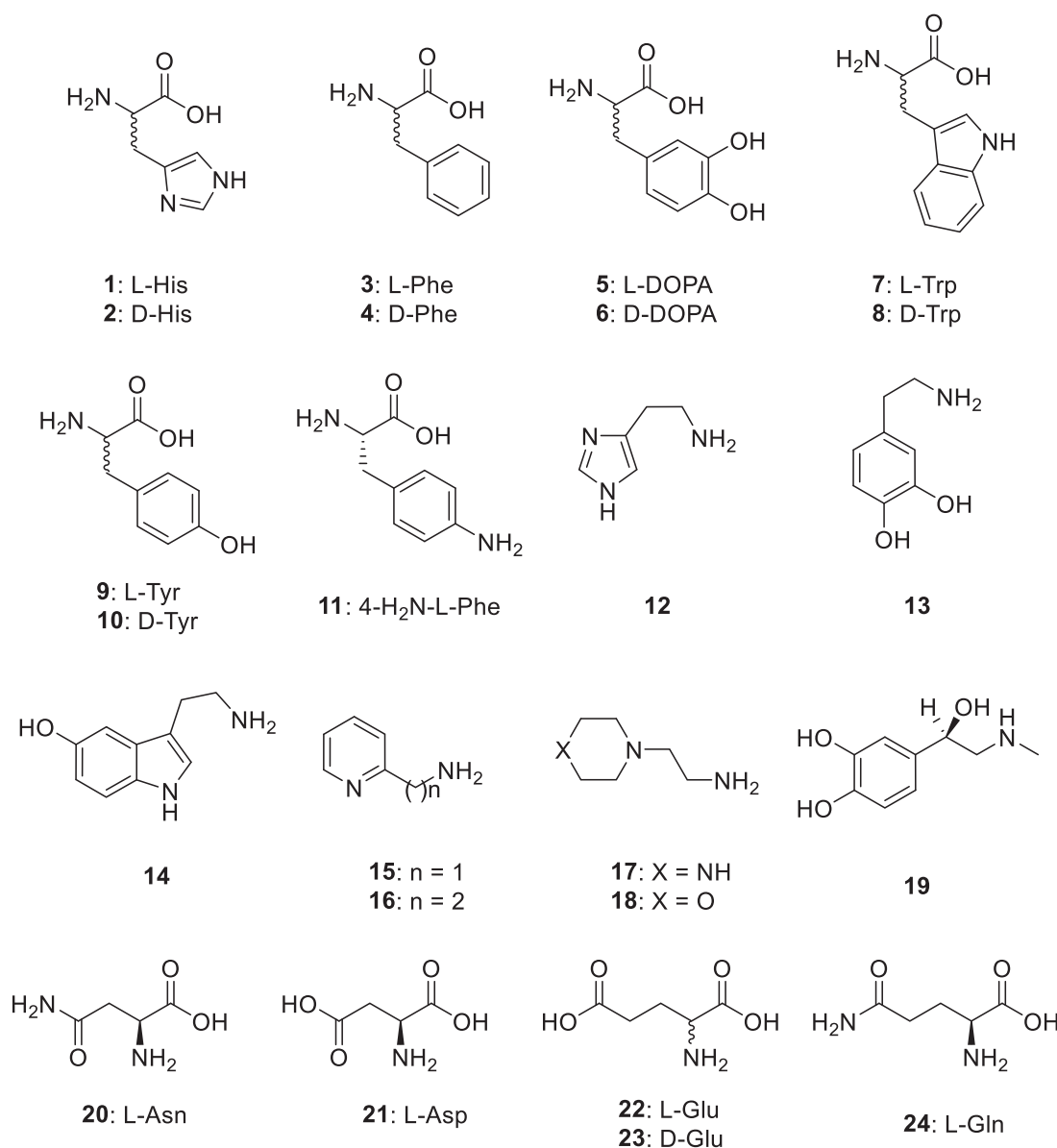


Figure 1. Amino acids and amines 1–24 investigated as CAAs in the current article.

Escherichia coli, EcoCA β , investigated earlier for its activation with the same class of compounds^{21c}.

2. Materials and methods

2.1. Enzyme production and purification

The protocol described in ref.² has been used to obtain purified recombinant MscCA. EcoCA β was also obtained in-house as reported earlier^{21c}.

2.2. Ca activity/activation measurements

An Sx.18Mv-R Applied Photophysics (Oxford, UK) stopped-flow instrument has been used to assay the catalytic activity of various CA isozymes for CO₂ hydration reaction²⁴. Phenol red (at a concentration of 0.2 mM) was used as indicator, working at the absorbance maximum of 557 nm, with 10 mM Hepes (pH 7.5, for α -CAs)^{14–16} or TRIS (pH 8.3, for β -CAs)^{19–21} as buffers, 0.1 M NaClO₄ (for maintaining constant ionic strength), following the CA-catalyzed CO₂ hydration reaction for a period of 10 s at 25 °C. The CO₂ concentrations ranged from 1.7 to 17 mM for the determination of the kinetic parameters and inhibition constants. For each activator at least six traces of the initial 5–10% of the reaction have been used for determining the initial velocity. The uncatalyzed rates were determined in the same manner and subtracted from the total observed rates. Stock solutions of activators (at 0.1 mM) were prepared in distilled-deionized water and dilutions up to 1 nM were made thereafter with the assay buffer. Enzyme and activator solutions were pre-incubated together for 15 min prior to assay, in order to allow for the formation of the enzyme–activator complexes. The activation constant (K_A), defined similarly with the inhibition constant K_i , can be obtained by considering the classical Michaelis–Menten equation (Equation 4), which has been fitted by non-linear least squares by using PRISM 3:

$$v = v_{\max} / \{1 + (K_M/[S])(1 + [A]_f/K_A)\} \quad (4)$$

where $[A]_f$ is the free concentration of activator.

Working at substrate concentrations considerably lower than K_M ($[S] \ll K_M$), and considering that $[A]_f$ can be represented in the form of the total concentration of the enzyme ($[E]_t$) and activator ($[A]_t$), the obtained competitive steady-state equation for determining the activation constant is given by Equation (5):

$$v = v_0 \cdot K_A / \{K_A + ([A]_t - 0.5\{([A]_t + [E]_t + K_A) - ([A]_t + [E]_t + K_A)^2 - 4[A]_t \cdot [E]_t\}^{1/2})\} \quad (5)$$

where v_0 represents the initial velocity of the enzyme-catalyzed reaction in the absence of activator^{19–21}. Enzyme concentrations in the assay system were of 7.6–12.8 nM.

2.3. Reagents

Amines and amino acid derivatives **1–24** were obtained in the highest purity that was available commercially from Sigma-Aldrich (Milan, Italy).

3. Results and discussion

The catalytic activity of MscCA is significant for the physiologic reaction, i.e. hydration of CO₂ to bicarbonate, with a k_{cat} of $1.46 \times 10^5 \text{ s}^{-1}$ and a Michaelis–Menten constant K_M of 5.7 mM, these kinetic parameters being comparable to those of other α - or

β -CAs investigated earlier^{14,21} (Table 1). The data in Table 1 also indicates that the presence of L-Trp as an activator does not change the K_M for either of the two enzymes belonging to the α -class (hCA I/II) as well as for EcoCA β and MscCA, a situation also observed for all CA classes for which CA activators have been investigated so far^{13–17,19–21}. In fact, as proven by kinetic and crystallographic data^{13–18}, the activator binds in a different region of the active site than the site of substrate binding. Thus, the activator does not influence K_M but has an effect only on k_{cat} . Indeed, a 10 μM concentration of L-Trp leads to a 7.5-fold enhancement of the kinetic constant of MscCA compared to the same parameter in the absence of the activator (Table 1). For hCA I and II, the enhancement of the kinetic constant in the presence of L-Trp was rather modest, as these enzymes have a weaker affinity for this activator (Table 1). On the other hand, L-Trp has a low micromolar affinity for MscCA which explains its effective activating effect on this bacterial enzyme.

Thus, we proceeded with the investigation of activators **1–24** (Figure 1) belonging to the amino acid and amine chemotypes for understanding their ability to activate MscCA as well as the structure–activity relationship profiles. In Table 2, the activation constants of these compounds against the target enzyme MscCA as well as hCA II and I (α -CA enzymes) and EcoCA β (a bacterial β -CA) are shown, for comparative reasons. The following SAR was observed for the activation of MscCA:

- All investigated amines and amino acids showed activating effects against MscCA, with K_A s ranging between 0.12 and 28.6 μM . It has been demonstrated earlier that the activator binds at the entrance of the CA active site (for α -class CAs^{6,13–17}) and participates in the rate determining step of the catalytic cycle, the shuttling of the protons from the zinc coordinated water molecule to the reaction medium. In this way the nucleophilic metal hydroxide species of the enzyme is formed, which enhances the overall catalytic process^{6,13–17}. Although no X-ray crystal structures of β -CA – activator complexes are known to date, we hypothesise that the activation mechanism is similar for all CA classes. This is also the reason why the CAAs possess protonatable moieties of the amino, imidazole and other heterocycles, or even carboxylate type¹⁷, all of them present also in compounds **1–24** investigated here.
- The most effective activators were D-His, 4-amino-L-Phe, L- and D-DOPA, 4-(2-Aminoethyl)-morpholine and L-Asn, which showed K_A s of 0.12–0.89 μM . These derivatives belong to three different chemotypes: aromatic amino acids based on the His/Phe chemotype (**3**, **5**, **6** and **11**); heterocyclic amines

Table 1. Activation of human carbonic anhydrase (hCA) isozymes I, II, EcoCA β and MscCA with L-Trp, at 25 °C, for the CO₂ hydration reaction²⁵.

Isozyme	k_{cat}^* (s^{-1})	K_M^* (mM)	$(k_{\text{cat}}^{\text{L-Trp}})^{**}$ (s^{-1})	K_A^{***} (μM) L-Trp
hCA I ^a	2.0×10^5	4.0	3.4×10^5	44.0
hCA II ^a	1.4×10^6	9.3	4.9×10^6	27.0
EcoCA β ^b	5.3×10^5	12.9	1.8×10^6	18.3
MscCA ^c	1.46×10^5	5.7	1.10×10^6	1.02

*Observed catalytic rate without activator. K_M values in the presence and the absence of activators were the same for the various CAs (data not shown), **Observed catalytic rate in the presence of 10 μM activator; ***The activation constant (K_A) for each enzyme was obtained by fitting the observed catalytic enhancements as a function of the activator concentration. All data are mean from at least three determinations by a stopped-flow, CO₂ hydrase method²⁴. Standard errors were in the range of 5–10% of the reported values (not shown). ^aHuman recombinant isozymes, from ref.¹⁴; ^bBacterial recombinant enzyme, from ref.^{21c}; ^cThis work.

- incorporating 2-aminoethyl side chains (**18**) and aliphatic dicarboxylic amino acid derivative (**20**). However, other investigated compounds structurally similar to these derivatives showed weaker CA activating effects, proving that the molecular recognition between the MscCA active site and the modulator is governed by many factors, and that small structural changes in the activator molecule leads to drastically different activating effects (Table 2). For example D-DOPA is an effective MscCA activator (K_A of 0.40 μM) whereas the structurally related D-Tyr and D-Phe (with one and no phenolic OH moieties, respectively) showed weaker such properties, with K_A s of 13.9 and 8.62 μM , respectively. The same differences can be observed between the structurally related amines **17** and **18**, which differ only by the endocyclic X group, with the morpholine derivative **18** being 29.3-times a better activator compared to the piperazine **17**.
- iii. Most of the investigated activators were effective, low micromolar activators of MscCA (K_A s of 1.02–8.62 μM) – Table 2. They include compounds **1**, **3**, **4**, **7–9**, **12–17**, **19**, **21–23**, belonging both to the amino acid and amine series. There seem to be no preference for L- or D-amino acids, since in some cases the D-enantiomer was a better activator, whereas in other cases the L-enantiomer showed more effective activating effects. Furthermore, these activators, as the ones discussed above, belong to heterogeneous chemotypes, making the SAR rather difficult to dissect. What is important on the other fact is that many diverse structural motifs incorporating proton shuttling moieties of the amino, heterocyclic or carboxylate type show these effects.
 - iv. The least effective activators against MscCA were D-Tyr and L-Gln, with K_A s of 13.9–28.6 μM .

Table 2. Activation constants of hCA I, hCA II and the bacterial enzymes EcoCA β (*E. coli*) and MscCA with amino acids and amines 1–24, by a stopped-flow CO₂ hydrase assay²⁴.

No.	Compound	K_A (μM) [*]			
		hCA I ^a	hCA II ^a	EcoCA β ^b	MscCA ^c
1	L-His	0.03	10.9	36.0	5.24
2	D-His	0.09	43	23.7	0.47
3	L-Phe	0.07	0.013	12.0	1.25
4	D-Phe	86	0.035	15.4	8.62
5	L-DOPA	3.1	11.4	10.7	0.89
6	D-DOPA	4.9	7.8	3.14	0.40
7	L-Trp	44	27	18.3	1.02
8	D-Trp	41	12	11.5	3.45
9	L-Tyr	0.02	0.011	9.86	3.81
10	D-Tyr	0.04	0.013	17.9	13.9
11	4-H ₂ N-L-Phe	0.24	0.15	7.34	0.73
12	Histamine	2.1	125	18.5	1.15
13	Dopamine	13.5	9.2	11.3	6.23
14	Serotonin	45	50	2.76	1.08
15	2-Pyridyl-methylamine	26	34	48.7	2.69
16	2-(2-Aminoethyl)pyridine	13	15	17.2	7.94
17	1-(2-Aminoethyl)-piperazine	7.4	2.3	14.1	3.52
18	4-(2-Aminoethyl)-morpholine	0.19	17.4	0.12	
19	L-Adrenaline	0.09	96.0	9.15	5.26
20	L-Asn	11.3	>100	49.5	0.88
21	L-Asp	5.20	>100	18.9	4.67
22	L-Glu	6.43	>100	18.0	3.75
23	D-Glu	10.7	>100	11.4	4.93
24	L-Gln	>100	>50	49.2	28.6

^{*}Mean from three determinations by a stopped-flow, CO₂ hydrase method²⁵. Standard errors were in the range of 5–10% of the reported values (data not shown).

^aHuman recombinant isozymes, from ref.¹⁴; ^bBacterial recombinant enzyme, ref.²¹; ^cBacterial recombinant enzyme, this work.

- v. The activation profile of MscCA is very different from that of other bacterial β -CAs, as the *E. coli* enzyme showed in Table 2, as well as the human isoforms hCA I and II.

4. Conclusions

The β -CA from *M. sciuri*, previously considered to be present in the genome of *S. aureus*, is effectively activated by amines and amino acids. Furthermore, as described earlier, this enzyme is also inhibited by anions and sulphonamides². Recently, Götzt's group²⁵ performed a thorough analysis regarding the presence of CAs in the genome of *S. aureus* and related species, expressing a rather critical vision regarding our earlier work on SauBCA² and bacterial CAs in general³. It is true that we did not investigate in detail whether the *S. aureus* genome sequences present in the NCBI database are all correct, as this is not our main research interest. However, the experiments and statements in which the N-cyanosulphonamide S-0859 is considered as a selective inhibitor of sodium-bicarbonate cotransporters by Götzt's group in order to definitely demonstrate the absence of CAs in this bacterium²⁵ are inconclusive, since N-cyanosulfonamides also act as rather effective CAIs^{26,27}. Whether CAs are present only in some members of the *Staphylococcaceae* and not in others, is of course highly relevant, but it should be noted that bacteria may encode also for γ -CAs³, which were not searched for in the above-mentioned study²⁵. What is more relevant according to us, is the fact that our study and the preceding ones², although performed on an enzyme thought to belong to *S. aureus* but which is actually *M. sciuri*, may bring to attention druggable targets which may lead to antibiotics with a novel mechanism of action. In fact, several groups showed that inhibition of bacterial CAs represents an effective and innovative way for fighting drug resistant bacteria^{4,5}, with all the scepticism from groups as the one mentioned above that these enzymes could be considered anti-infective drug targets. As far as we know, resistance to sulphonamide CAIs has not been registered for any of the investigated bacterial species, although this phenomenon is erroneously mentioned in ref.²⁵

Disclosure statement

The authors have no relevant affiliations of financial involvement with any organisation or entity with a financial interest in or financial conflict with the subject matter or materials discussed in the manuscript. CT Supuran is Editor-in-Chief of the *Journal of Enzyme Inhibition and Medicinal Chemistry*. He was not involved in the assessment, peer review, or decision-making process of this paper. The authors have no relevant affiliations of financial involvement with any organisation or entity with a financial interest in or financial conflict with the subject matter or materials discussed in the manuscript. This includes employment, consultancies, honoraria, stock ownership or options, expert testimony, grants or patents received or pending, or royalties.

Funding

The Italian Ministry for University and Research (MIUR) is thanked for grant no. PRIN: rot. 2017XYBP2R (to CTS) and the project FIS2019_04819 BacCAD (to CTS and CC). This research program was also funded by the Academy of Finland (to SP).

ORCID

Andrea Angeli  <http://orcid.org/0000-0002-1470-7192>
 Clemente Capasso  <http://orcid.org/0000-0003-3314-2411>
 Seppo Parkkila  <http://orcid.org/0000-0001-7323-8536>
 Claudiu T. Supuran  <http://orcid.org/0000-0003-4262-0323>

References

- (a) Guilloton MB, Korte JJ, Lamblin AF, et al. Carbonic anhydrase in *Escherichia coli*. A product of the cyn operon. *J Biol Chem* **1992**;267(6):3731–4. (b) Guilloton MB, Lamblin AF, Kozliak EI, et al. A physiological role for cyanate-induced carbonic anhydrase in *Escherichia coli*. *J Bacteriol* **1993**;175(5):1443–51. (c) Cronk JD, Endrizzi JA, Cronk MR, et al. Crystal structure of *E. coli* beta-carbonic anhydrase, an enzyme with an unusual pH-dependent activity. *Protein Sci* **2001**;10(5):911–22. (d) Merlin C, Masters M, McAteer S, Coulson A. Why is carbonic anhydrase essential to *Escherichia coli*? *J Bacteriol* **2003**;185(21):6415–24. (e) Supuran CT. Bacterial carbonic anhydrases as drug targets: toward novel antibiotics? *Front Pharmacol* **2011**;2:34. (f) Di Fiore A, De Luca V, Langella E, et al. Biochemical, structural, and computational studies of a γ -carbonic anhydrase from the pathogenic bacterium *Burkholderia pseudomallei*. *Comput Struct Biotechnol J* **2022**;20:4185–94.
- (a) Urbanski LJ, Bua S, Angeli A, et al. Sulphonamide inhibition profile of *Staphylococcus aureus* β -carbonic anhydrase. *J Enzyme Inhib Med Chem* **2020**;35(1):1834–39. (b) Urbanski LJ, Vullo D, Parkkila S, Supuran CT. An anion and small molecule inhibition study of the β -carbonic anhydrase from *Staphylococcus aureus*. *J Enzyme Inhib Med Chem* **2021**;36(1):1088–92.
- (a) Ferraroni M, Del Prete S, Vullo D, et al. Crystal structure and kinetic studies of a tetrameric type II β -carbonic anhydrase from the pathogenic bacterium *Vibrio cholerae*. *Acta Crystallogr D Biol Crystallogr* **2015**;71(Pt 12):2449–56. (b) Capasso C, Supuran CT. Anti-infective carbonic anhydrase inhibitors: a patent and literature review. *Expert Opin Ther Pat* **2013**;23(6):693–704. (c) Nishimori I, Vullo D, Minakuchi T, et al. Anion inhibition studies of two new β -carbonic anhydrases from the bacterial pathogen *Legionella pneumophila*. *Bioorg Med Chem Lett* **2014**;24(4):1127–32. (d) Capasso C, Supuran CT. An overview of the alpha-, beta- and gamma-carbonic anhydrases from Bacteria: can bacterial carbonic anhydrases shed new light on evolution of bacteria? *J Enzyme Inhib Med Chem* **2015**;30(2):325–32.
- (a) De Luca V, Carginale V, Supuran CT, Capasso C. The gram-negative bacterium *Escherichia coli* as a model for testing the effect of carbonic anhydrase inhibition on bacterial growth. *J Enzyme Inhib Med Chem* **2022**;37(1):2092–8. (b) Supuran CT, Capasso C. Antibacterial carbonic anhydrase inhibitors: an update on the recent literature. *Expert Opin Ther Pat* **2020**;30(12):963–82. (c) Giovannuzzi S, Hewitt CS, Nocentini A, et al. Inhibition studies of bacterial α -carbonic anhydrases with phenols. *J Enzyme Inhib Med Chem* **2022**;37(1):666–71.
- (a) Supuran CT, Capasso C. Biomedical applications of prokaryotic carbonic anhydrases. *Expert Opin Ther Pat* **2018**;28(10):745–54. (b) Flaherty DP, Seleem MN, Supuran CT. Bacterial carbonic anhydrases: underexploited antibacterial therapeutic targets. *Future Med Chem* **2021**;13(19):1619–22. (c) Hewitt CS, Abutaleb NS, Elhassanny AEM, et al. Structure-activity relationship studies of acetazolamide-based carbonic anhydrase inhibitors with activity against *Neisseria gonorrhoeae*. *ACS Infect Dis* **2021**;7(7):1969–84. (d) Abutaleb NS, Elhassanny AEM, et al. Repurposing FDA-approved sulphonamide carbonic anhydrase inhibitors for treatment of *Neisseria gonorrhoeae*. *J Enzyme Inhib Med Chem* **2022**;37(1):51–61. (e) An W, Holly KJ, Nocentini A, et al. Structure-activity relationship studies for inhibitors for vancomycin-resistant *Enterococcus* and human carbonic anhydrases. *J Enzyme Inhib Med Chem* **2022**;37(1):1838–44.
- (a) Supuran CT. Carbonic anhydrases: novel therapeutic applications for inhibitors and activators. *Nat Rev Drug Discov* **2008**;7(2):168–81. (b) Nocentini A, Supuran CT. Advances in the structural annotation of human carbonic anhydrases and impact on future drug discovery. *Expert Opin Drug Discov* **2019**;14(11):1175–97. (c) Supuran CT. How many carbonic anhydrase inhibition mechanisms exist? *J Enzyme Inhib Med Chem* **2016**;31(3):345–60. (d) Supuran CT. Advances in structure-based drug discovery of carbonic anhydrase inhibitors. *Expert Opin Drug Discov* **2017**;12(1):61–88. (e) Supuran CT. Structure and function of carbonic anhydrases. *Biochem J* **2016**;473(14):2023–32.
- (a) Supuran CT. Applications of carbonic anhydrases inhibitors in renal and central nervous system diseases. *Expert Opin Ther Pat* **2018**;28(10):713–21. (b) Supuran CT. Carbonic anhydrase inhibitors and their potential in a range of therapeutic areas. *Expert Opin Ther Pat* **2018**;28(10):709–12. (c) Supuran CT. Carbonic anhydrase inhibitors as emerging agents for the treatment and imaging of hypoxic tumors. *Expert Opin Investig Drugs* **2018**;27(12):963–70. (d) Nocentini A, Supuran CT. Carbonic anhydrase inhibitors as antitumor/antimetastatic agents: a patent review (2008–2018). *Expert Opin Ther Pat* **2018**;28(10):729–40. (e) Dogné JM, Hanson J, Supuran C, Pratico D. Coxibs and cardiovascular side-effects: from light to shadow. *Curr Pharm Des* **2006**;12(8):971–975.
- (a) Ozensoy Guler O, Capasso C, Supuran CT. A magnificent enzyme superfamily: carbonic anhydrases, their purification and characterization. *J Enzyme Inhib Med Chem* **2016**;31(5):689–94. (b) De Simone G, Supuran CT. (In)organic anions as carbonic anhydrase inhibitors. *J Inorg Biochem* **2012**;111:117–29. (c) Supuran CT. Carbonic anhydrase inhibition and the management of hypoxic tumors. *Metabolites* **2017**;7:E48. (d) Bertol E, Vaiano F, Mari F, et al. Advances in new psychoactive substances identification: the U.R.I.To.N. Consortium. *J Enzyme Inhib Med Chem* **2017**;32(1):841–9. (e) Boztaş M, Çetinkaya Y, Topal M, et al. Synthesis and carbonic anhydrase isoenzymes I, II, IX, and XII inhibitory effects of dimethoxybromophenol derivatives incorporating cyclopropane moieties. *J Med Chem* **2015**;58(2):640–50.
- (a) Supuran CT. Carbonic anhydrases and metabolism. *Metabolites* **2018**;8(2):25. (b) Supuran CT. Exploring the multiple binding modes of inhibitors to carbonic anhydrases for novel drug discovery. *Expert Opin Drug Discov* **2020**;15(6):671–86.
- (a) Alterio V, Di Fiore A, D'Ambrosio K, et al. Multiple binding modes of inhibitors to carbonic anhydrases: how to design specific drugs targeting 15 different isoforms? *Chem Rev* **2012**;112(8):4421–68. (b) Briganti F, Pierattelli R, Scozzafava A, Supuran CT. Carbonic anhydrase inhibitors. Part 37. Novel classes of carbonic anhydrase inhibitors and their interaction with the native and cobalt-substituted enzyme: kinetic and spectroscopic investigations. *Eur J Med*

- Chem 1996;31(12):1001–10. (c) Neri D, Supuran CT. Interfering with pH regulation in tumours as a therapeutic strategy. *Nat Rev Drug Discov* 2011;10(10):767–77. (d) Supuran CT, Vullo D, Manole G, Casini A, Scozzafava A. Designing of novel carbonic anhydrase inhibitors and activators. *Curr Med Chem Cardiovasc Hematol Agents* 2004;2(1):49–68.
11. (a) Supuran CT, Capasso C. An overview of the bacterial carbonic anhydrases. *Metabolites* 2017;7(4):56. (b) Supuran CT. Carbon- versus sulphur-based zinc binding groups for carbonic anhydrase inhibitors? *J Enzyme Inhib Med Chem* 2018;33(1):485–95. (c) Rahman MM, Tikhomirova A, Modak JK, et al. Antibacterial activity of ethoxzolamide against *Helicobacter pylori* strains SS1 and 26695. *Gut Pathog* 2020;12:20. (d) Supuran CT. Emerging role of carbonic anhydrase inhibitors. *Clin Sci (Lond)* 2021;135(10):1233–49.
 12. (a) Kumar A, Siwach K, Supuran CT, Sharma PK. A decade of tail-approach based design of selective as well as potent tumor associated carbonic anhydrase inhibitors. *Bioorg Chem* 2022;126:105920. (b) Bonardi A, Bua S, Combs J, et al. The three-tails approach as a new strategy to improve selectivity of action of sulphonamide inhibitors against tumour-associated carbonic anhydrase IX and XII. *J Enzyme Inhib Med Chem* 2022;37(1):930–9. (c) Abdel-Mohsen HT, El Kerdawy AM, Omar MA, et al. Application of the dual-tail approach for the design and synthesis of novel Thiopyrimidine-Benzene-sulfonamide hybrids as selective carbonic anhydrase inhibitors. *Eur J Med Chem* 2022;228:114004. (d) Nocentini A, Angeli A, Carta F, et al. Reconsidering anion inhibitors in the general context of drug design studies of modulators of activity of the classical enzyme carbonic anhydrase. *J Enzyme Inhib Med Chem* 2021;36(1):561–80.
 13. Briganti F, Mangani S, Orioli P, et al. Carbonic anhydrase activators: X-ray crystallographic and spectroscopic investigations for the interaction of isozymes I and II with histamine. *Biochemistry* 1997;36(34):10384–92.
 14. (a) Temperini C, Scozzafava A, Vullo D, Supuran CT. Carbonic anhydrase activators. Activation of isozymes I, II, IV, VA, VII, and XIV with L- and D-histidine and crystallographic analysis of their adducts with isoform II: engineering proton-transfer processes within the active site of an enzyme. *Chemistry* 2006;12(27):7057–66. (b) Temperini C, Scozzafava A, Vullo D, Supuran CT. Carbonic anhydrase activators. Activation of isoforms I, II, IV, VA, VII, and XIV with L- and D-phenylalanine and crystallographic analysis of their adducts with isozyme II: stereospecific recognition within the active site of an enzyme and its consequences for the drug design. *J Med Chem* 2006;49(10):3019–27. (c) Temperini C, Innocenti A, Scozzafava A, Supuran CT. Carbonic anhydrase activators: kinetic and X-ray crystallographic study for the interaction of D- and L-tryptophan with the mammalian isoforms I–XIV. *Bioorg Med Chem* 2008;16(18):8373–78. (d) Temperini C, Innocenti A, Scozzafava A, et al. Carbonic anhydrase activators: L-Adrenaline plugs the active site entrance of isozyme II, activating better isoforms I, IV, VA, VII, and XIV. *Bioorg Med Chem Lett* 2007;17(3):628–35.
 15. (a) Akocak S, Lolak N, Vullo D, et al. Synthesis and biological evaluation of histamine Schiff bases as carbonic anhydrase I, II, IV, VII, and IX activators. *J Enzyme Inhib Med Chem* 2017;32(1):1305–12. (b) Akocak S, Lolak N, Bua S, et al. α -Carbonic anhydrases are strongly activated by spinaceamine derivatives. *Bioorg Med Chem* 2019;27(5):800–4. (c) Akocak S, Lolak N, Bua S, Nocentini A, Supuran CT. Activation of human α -carbonic anhydrase isoforms I, II, IV and VII with bis-histamine Schiff bases and bis-spinaceamine substituted derivatives. *J Enzyme Inhib Med Chem* 2019;34(1):1193–98. (d) Dave K, Scozzafava A, Vullo D, et al. Pyridinium derivatives of histamine are potent activators of cytosolic carbonic anhydrase isoforms, I, II and VII. *Org Biomol Chem* 2011;9(8):2790–800. (e) Dave K, Ilies MA, Scozzafava A, et al. An inhibitor-like binding mode of a carbonic anhydrase activator within the active site of isoform II. *Bioorg Med Chem Lett* 2011;21(9):2764–68.
 16. (a) Bhatt A, Mondal UK, Supuran CT, et al. Crystal structure of carbonic anhydrase II in complex with an activating ligand: implications in neuronal function. *Mol Neurobiol* 2018;55(9):7431–37. (b) Temperini C, Scozzafava A, Supuran CT. Carbonic anhydrase activators: the first X-ray crystallographic study of an adduct of isoform I. *Bioorg Med Chem Lett* 2006;16(19):5152–6. (c) Temperini C, Scozzafava A, Puccetti L, Supuran CT. Carbonic anhydrase activators: X-ray crystal structure of the adduct of human isozyme II with L-histidine as a platform for the design of stronger activators. *Bioorg Med Chem Lett* 2005;15(23):5136–41. (d) Clare BW, Supuran CT. Carbonic anhydrase activators. 3: Structure-activity correlations for a series of isozyme II activators. *J Pharm Sci* 1994;83(6):768–73.
 17. (a) Supuran CT. Carbonic anhydrase activators. *Future Med Chem* 2018;10(5):561–73. (b) Temperini C, Scozzafava A, Supuran CT. Carbonic anhydrase activation and the drug design. *Curr Pharm Des* 2008;14(7):708–15. (c) Akocak S, Supuran CT. Activation of α -, β -, γ -, δ -, ζ - and η - class of carbonic anhydrases with amines and amino acids: a review. *J Enzyme Inhib Med Chem* 2019;34(1):1652–1659. (d) Angeli A, Vaiano F, Mari F, et al. Psychoactive substances belonging to the amphetamine class potentially activate brain carbonic anhydrase isoforms VA, VB, VII, and XII. *J Enzyme Inhib Med Chem* 2017;32(1):1253–1259.
 18. (a) Canto de Souza L, Provensi G, Vullo D, et al. Carbonic anhydrase activation enhances object recognition memory in mice through phosphorylation of the extracellular signal-regulated kinase in the cortex and the hippocampus. *Neuropharmacology* 2017;118:148–56. (b) Sanku RKK, John JS, Salkovitz M, et al. Potential learning and memory disruptors and enhancers in a simple, 1-day operant task in mice. *Behav Pharmacol* 2018;29(6):482–92. (c) Schmidt SD, Nachtigall EG, Marcondes LA, et al. Modulation of carbonic anhydrases activity in the hippocampus or prefrontal cortex differentially affects social recognition memory in rats. *Neuroscience* 2022;497:184–95. (d) Provensi G, Nocentini A, Passani MB, et al. Activation of carbonic anhydrase isoforms involved in modulation of emotional memory and cognitive disorders with histamine agonists, antagonists and derivatives. *J Enzyme Inhib Med Chem* 2021;36(1):719–26. (e) Schmidt SD, Costa A, Rani B, et al. The role of carbonic anhydrases in extinction of contextual fear memory. *Proc Natl Acad Sci USA* 2020;117(27):16000–8. (f) Blandina P, Provensi G, Passani MB, et al. Carbonic anhydrase modulation of emotional memory. Implications for the treatment of cognitive disorders. *J Enzyme Inhib Med Chem* 2020;35(1):1206–14.
 19. (a) Innocenti A, Zimmerman SA, Scozzafava A, et al. Carbonic anhydrase activators: activation of the archaean β -class (Cab) and γ -class (Cam) carbonic anhydrases with amino acids and amines. *Bioorg Med Chem Lett* 2008;

- 18(23):6194–8. (b) Vullo D, Del Prete S, Capasso C, Supuran CT. Carbonic anhydrase activators: activation of the β -carbonic anhydrase from *Malassezia globosa* with amino acids and amines. *Bioorg Med Chem Lett* 2016;26(5):1381–85. (c) Isik S, Kockar F, Aydin M, et al. Carbonic anhydrase activators: activation of the β -carbonic anhydrase Nce103 from the yeast *Saccharomyces cerevisiae* with amino acids and amines. *Bioorg Med Chem Lett* 2009;19(6):1662–5. (d) Innocenti A, Hall RA, Scozzafava A, et al. Carbonic anhydrase activators: activation of the beta-carbonic anhydrases from the pathogenic fungi *Candida albicans* and *Cryptococcus neoformans* with amines and amino acids. *Bioorg Med Chem* 2010;18(3):1034–7.
20. (a) Angeli A, Del Prete S, Osman SM, et al. Activation studies of the α - and β -carbonic anhydrases from the pathogenic bacterium *Vibrio cholerae* with amines and amino acids. *J Enzyme Inhib Med Chem* 2018;33(1):227–33. (b) Angeli A, Del Prete S, Donald WA, et al. The γ -carbonic anhydrase from the pathogenic bacterium *Vibrio cholerae* is potentially activated by amines and amino acids. *Bioorg Chem* 2018;77:1–5. (c) Angeli A, Del Prete S, Osman SM, et al. Activation studies with amines and amino acids of the β -carbonic anhydrase encoded by the Rv3273 gene from the pathogenic bacterium *Mycobacterium tuberculosis*. *J Enzyme Inhib Med Chem* 2018;33(1):364–69.
21. (a) Angeli A, Del Prete S, Pinteala M, et al. The first activation study of the β -carbonic anhydrases from the pathogenic bacteria *Brucella suis* and *Francisella tularensis* with amines and amino acids. *J Enzyme Inhib Med Chem* 2019;34(1):1178–85. (b) Stefanucci A, Angeli A, Dimmito MP, et al. Activation of β - and γ -carbonic anhydrases from pathogenic bacteria with tripeptides. *J Enzyme Inhib Med Chem* 2018;33(1):945–50. (c) Nocentini A, Del Prete S, Mastrolorenzo MD, et al. Activation studies of the β -carbonic anhydrases from *Escherichia coli* with amino acids and amines. *J Enzyme Inhib Med Chem* 2020;35(1):1379–86.
22. (a) Gramoli JL, Wilkinson BJ. Characterization and identification of coagulase-negative, heat-stable deoxyribonuclease-positive staphylococci. *J Gen Microbiol* 1978;105(2):275–285. (b) Fungwithaya P, Boonchuay K, Narinthorn R, et al. First study on diversity and antimicrobial-resistant profile of staphylococci in sports animals of Southern Thailand. *Vet World* 2022;15(3):765–74. (c) Cai Y, Zheng L, Lu Y, et al. Inducible resistance to β -lactams in oxacillin-susceptible mecA1-positive *Staphylococcus sciuri* isolated from retail pork. *Front Microbiol* 2021;12:721426. (d) Chen S, Wang Y, Chen F, et al. A highly pathogenic strain of *Staphylococcus sciuri* caused fatal exudative epidermitis in piglets. *PLoS One* 2007;2(1):e147.
23. Madhaiyan M, Wirth JS, Saravanan VS. Phylogenomic analyses of the *Staphylococcaceae* family suggest the reclassification of five species within the genus *Staphylococcus* as heterotypic synonyms, the promotion of five subspecies to novel species, the taxonomic reassignment of five *Staphylococcus* species to *Mammaliococcus* gen. nov., and the formal assignment of *Nosocomiicoccus* to the family *Staphylococcaceae*. *Int J Syst Evol Microbiol*. 2020;70(11):5926–5936.
24. Khalifah RG. The carbon dioxide hydration activity of carbonic anhydrase. I. Stop-flow kinetic studies on the native human isoenzymes B and C. *J Biol Chem* 1971;246(8):2561–2573.
25. (a) Fan S-H, Matsuo M, Huang L, et al. The MpsAB bicarbonate transporter is superior to carbonic anhydrase in biofilm-forming bacteria with limited CO₂ diffusion. *Microbiol Spectr* 2021;9(1):e0030521. (b) Fan SH, Liberini E, Götz F. *Staphylococcus aureus* genomes harbor only MpsAB-like bicarbonate transporter but not carbonic anhydrase as dissolved inorganic carbon supply system. *Microbiol Spectr* 2021;9(3):e0097021.
26. Supuran CT, Scozzafava A, Briganti F. Carbonic anhydrase inhibitors: N-cyanosulfonamides, a new class of high affinity isozyme II and IV inhibitors. *J Enzyme Inhib*. 1999;14(4):289–306.
27. (a) Amedei A, Capasso C, Nannini G, Supuran CT. Microbiota, bacterial carbonic anhydrases, and modulators of their activity: links to human diseases? *Mediators Inflamm* 2021;2021:6926082. (b) Nocentini A, Supuran CT, Capasso C. An overview on the recently discovered iota-carbonic anhydrases. *J Enzyme Inhib Med Chem* 2021;36(1):1988–95. (c) Campestre C, De Luca V, Carradori S, et al. Carbonic anhydrases: new perspectives on protein functional role and inhibition in *Helicobacter pylori*. *Front Microbiol* 2021;12:629163. (d) Del Prete S, Nocentini A, Supuran CT, Capasso C. Bacterial ι -carbonic anhydrase: a new active class of carbonic anhydrase identified in the genome of the Gram-negative bacterium *Burkholderia territorii*. *J Enzyme Inhib Med Chem* 2020;35(1):1060–8.