

Bidirectional Iterative Approach to Sequence-Defined Unsaturated Oligoesters

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Cite This: *JACS Au* 2023, 3, 252–260

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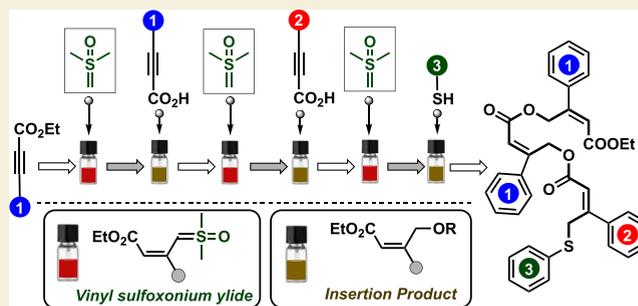
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ABSTRACT: Herein, we describe the development of a new strategy for the synthesis of unsaturated oligoesters *via* sequential metal- and reagent-free insertion of vinyl sulfoxonium ylides into the O–H bond of carboxylic acid. Like two directional coupling of amino acids (N- to C-terminal and C- to N-terminal) in peptide synthesis, the present approach offers a strategy in both directions to synthesize oligoesters. The sequential addition of the vinyl sulfoxonium ylide to the carboxylic acids (acid iteration sequence) in one direction and the sequential addition of the carboxylic acids to the vinyl sulfoxonium ylide (ylide iteration sequence) in another direction yield (Z)-configured unsaturated oligoesters. To perform this iteration, we have developed a highly regioselective insertion of vinyl sulfoxonium ylide into the X–H (X = O, N, C, S, halogen) bond of acids, thiols, phenols, amines, indoles, and halogen acids under metal-free reaction conditions. The insertion reaction is applied to a broad range of substrates (>50 examples, up to 99% yield) and eight iterative sequences. Mechanistic studies suggest that the rate-limiting step depends on the type of X–H insertion.

KEYWORDS: *sequence-defined polymerization, metal-free insertion, sulfur ylide, oligoester, iterative synthesis*



INTRODUCTION

In nature, sequence-defined polymers, such as peptides, are produced in a highly defined manner. Their precise composition provides unique chemical and biological properties to the materials.¹ However, unlike peptides and nucleotides, synthetic methods for sequence-defined polymers are rare.^{1,2} The discrete units of monomers in the sequence-specific synthetic polymer can produce more diversity than the biological polymer.^{3,4} Therefore, it is of great utility to achieve precise control of monomer sequence.⁵ The total number of combinations of sequence-controlled polymer depends on the number of monomeric units; as a result, each combination produces a distinct synthetic polymer and possesses unique properties (Figure 1a). This approach opens new opportunities for controlling the structural and macroscopic properties of the materials.^{1,6–9} Appropriate monomeric templates and a strategy for successively assembling these units are essential components of sequence-controlled polymerization.

The ribosome machinery in a living organism produces sequence-defined polymers *via* the complex biological mechanism. By utilizing this technology, the Suga group developed the mRNA-directed synthesis of polyesters (up to 4 monomeric units), composed of different α -hydroxy acids, whose length and sequence are controlled through reprogramming of the genetic code using flexizyme,¹⁰ and the wPURE (protein synthesis using recombinant elements) system (Scheme 1a).^{11–13} In 2013, the Junkers group reported the

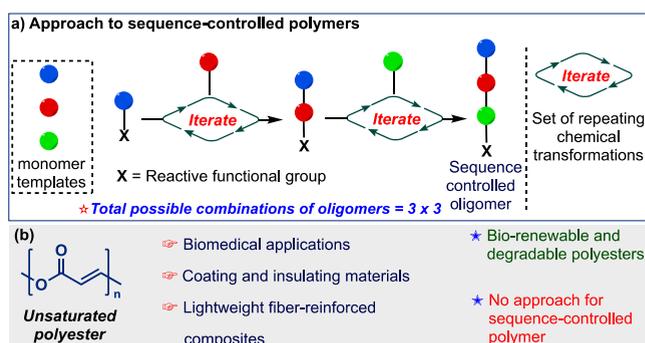


Figure 1. (a) Concept for the sequence-controlled oligomers and (b) importance and applications of unsaturated polyesters.

synthesis of sequence-controlled polyesters (up to 4 monomers) by sequential addition of monomers using the reversible addition–fragmentation radical transfer technique.¹⁴ Despite the few reports on sequence-defined polyesters synthesis, to the best of our knowledge, there is no report for unsaturated

Received: November 23, 2022

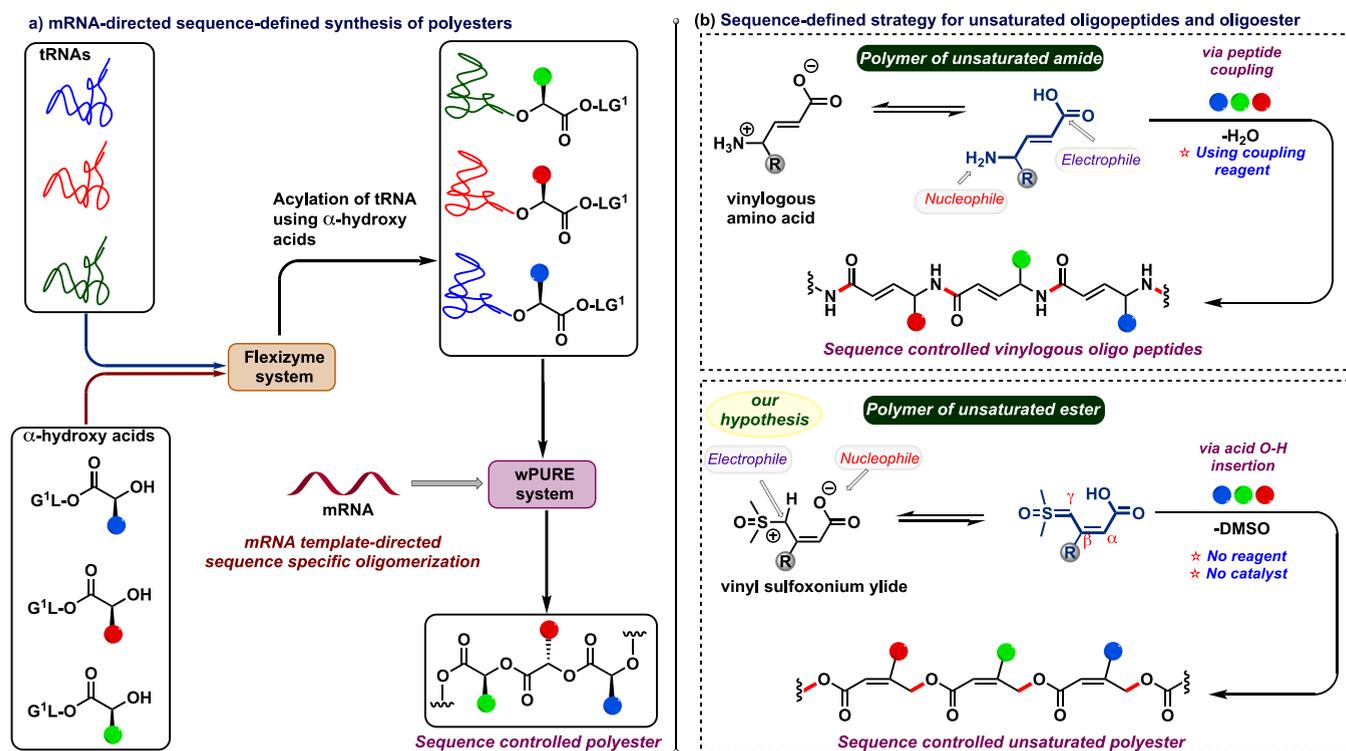
Revised: December 22, 2022

Accepted: December 22, 2022

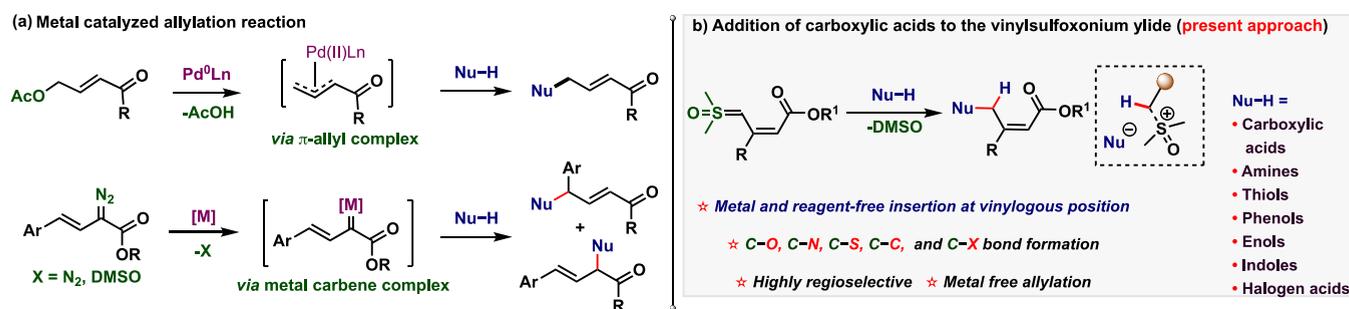
Published: January 3, 2023



Scheme 1. (a) Sequence-Defined Polyester Synthesis and (b) Design for the Sequence-Defined Unsaturated Polyesters from Vinyl Sulfoxonium Ylide



Scheme 2. (a) Vinylogous Addition of Nucleophiles and (b) Insertion of Diazo and Sulfoxonium Ylides into the Acids



polyesters which are the third-largest class of thermoset molding resins.² These polymers are well known for their biodegradability and biocompatibility^{15–17} and represent an interesting category due to their remarkable mechanical properties, chemical resistance such as corrosion, and lightness.¹⁸ Unsaturated polyesters are employed in many fields such as biomedical applications,^{19,20} coating and insulating materials,²¹ and lightweight fiber-reinforced composites for the automotive industry (Figure 1b).²² Moreover, the olefin group in the template can be chemically modified *via* Michael addition of various nucleophiles to the α,β -unsaturated esters.^{17,23,24} Therefore, developing efficient methods that can effectively generate sequence-defined unsaturated polymers is of great interest.^{1,6–9,25} Thus, we would like to develop a novel strategy for the template-based sequence-defined unsaturated oligoesters by the C–O coupling of α,β -unsaturated acids under mild reaction conditions (Scheme 1b, right). The conceptual design of the monomer template is crucial for a sequence-defined oligomer approach.^{26,27}

In the present approach, similar to a vinylogous amino acid which can be used to synthesize unsaturated oligopeptide

(Scheme 1b, left),²⁸ we visualized a vinyl sulfoxonium ylide monomeric template that contains a carboxylic acid and ylide groups may exist in a zwitterionic form, as shown in Scheme 1b, right. These ylide monomers can be coupled by the addition of a nucleophilic carboxylate ion of one monomer to the γ -(vinylogous) position of the other monomer ylide.

The addition of electrophiles at the vinylogous position, such as vinylogous aldol/Michael/Mannich reactions, has been extensively explored using metal and metal-free catalysis.^{29–32} However, introducing nucleophiles to the γ -position under catalytic conditions is challenging and it can be obtained *via* either electron-deficient π -allylmetal complexes^{33–35} or *in situ*-generated diazo-derived metal vinyl carbenes (Scheme 2a).^{36–38}

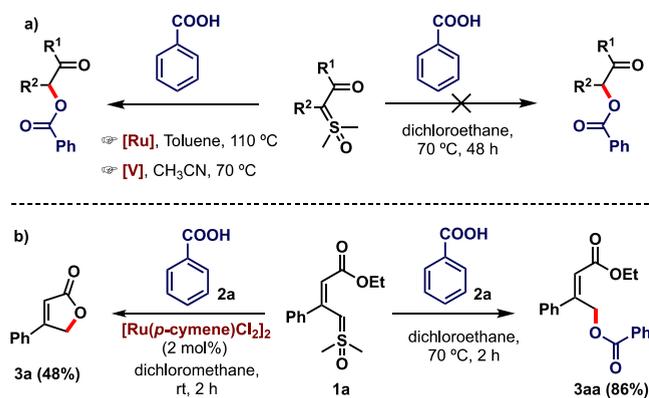
The addition reactions with carboxylic acids as a nucleophile are quite challenging because the oxygen atom of acid is less nucleophilic and also, these reactions are difficult using the Tsuji–Trost method due to the leaving ability of these groups during the formation of the π -allylpalladium complex. In fact, metal carbenes are highly reactive electrophilic intermediates. The Davies group developed an elegant method for silver-

catalyzed insertion of vinyl diazo compounds into the carboxylic acid.^{36,39} However, the stability of diazo compounds and regioselectivity for the nucleophilic addition are major concerns in this approach (Scheme 2a). Sulfoxonium ylides have proven to be alternative surrogates to diazo compounds for the insertion and carbene-mediated reactions.^{40–42} Recently, α -keto sulfoxonium ylides have been widely utilized for various X–H (X = N, O, S, C) insertion reactions under metal- and metal-free conditions.^{42–51} Despite the progress on the insertion reactions of α -keto sulfoxonium ylides, the chemistry of vinyl sulfoxonium ylides has been underexplored. Remarkably, our previous studies have demonstrated that the reactivity and stability of α -keto sulfoxonium ylides differ from vinyl sulfoxonium ylides.⁵² In this regard, our ongoing interest in the chemistry of sulfoxonium ylides,^{52–55} and the importance of regioselective allylation reactions in various applications,^{56–59} prompted us to develop a new allylation strategy using vinyl sulfoxonium ylides with variety of nucleophiles (Scheme 2b). Herein, we describe the insertion reactions of vinyl sulfoxonium ylides into the X–H (X = N, O, S, C, halogen) bond and apply this method to the synthesis of various sequence-defined unsaturated oligoesters.

RESULTS AND DISCUSSION

At the outset, we investigated vinyl sulfoxonium ylide insertion into the O–H bond of carboxylic acids. Recently, Mu and Sivasankar groups have reported the metal-carbene mediated insertion of α -keto sulfoxonium ylides into the carboxylic acids in good yields. No product formation was observed in the absence of the catalyst (Scheme 3a).^{60,61} For the present study,

Scheme 3. Insertion of Sulfoxonium Ylide into the O–H Bond of Acid under Metal and Metal-Free Conditions



the reaction between 0.5 mmol of vinyl sulfoxonium ylide **1a** (easily prepared from phenyl propiolate and dimethylsulfoxonium methylide) and 0.6 mmol of benzoic acid in the presence of the Ru-catalyst at room temperature did not yield the desired O–H insertion product; instead, afforded furanone **3a** in 48% yield (Scheme 3b, left).⁶² The optimization of the acid insertion was explored with various Brønsted and Lewis acid catalytic conditions and catalyst-free conditions (see the Supporting Information). Gratifyingly, in contrast to the reported literature,^{60,61} the insertion of ylide at 70 °C without any catalyst afforded the corresponding acid insertion product **3aa** in 86% isolated yield (Scheme 3b, right).

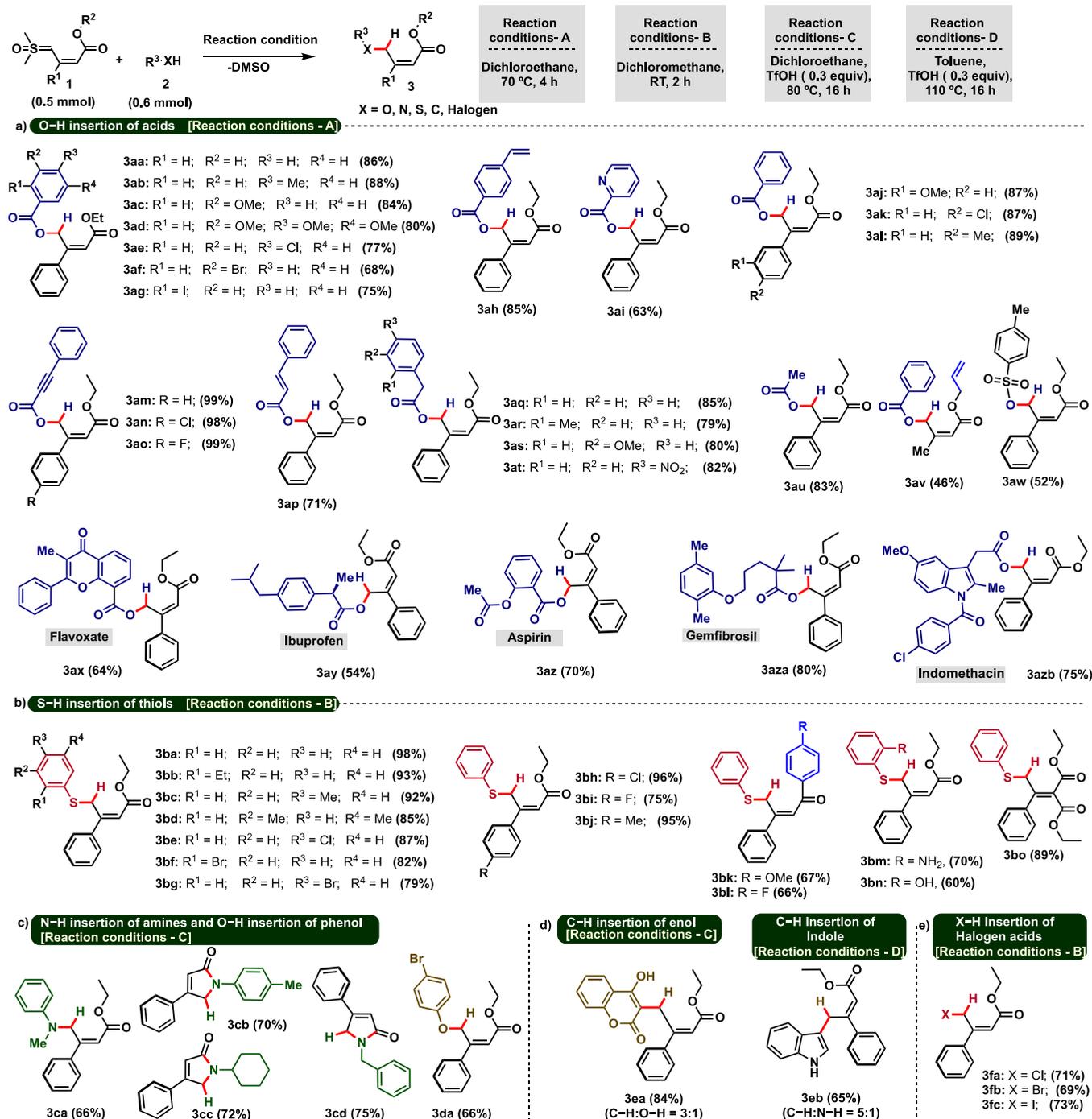
We then explored the effect of various substituents on the aryl groups of the acid and vinyl sulfoxonium ylide, as summarized in Scheme 4a. Benzoic acid with the *p*-methyl

group gave insertion product **3ab** in 88% yield. The presence of electron-donating groups in benzoic acid did not improve the reaction yield, giving insertion products **3ac** and **3ad** in 84 and 80% yields, respectively. A slight decrease in the yields was observed with halogen substituents at para, meta, and ortho positions on the benzene ring, affording **3ae**, **3af**, and **3ag** in 77, 68, and 75% yields, respectively. When using 4-vinylbenzoic acid, the corresponding insertion product **3ah** was obtained in 85% yield. Lower yields were obtained when the heteroaromatic groups, such as pyridine, were incorporated (**3ai**, 63%), likely due to the reduced protonation of the ylide carbon. The substituents on the benzene ring of the vinyl sulfoxonium moiety did not affect the reaction yields and afforded the corresponding γ -insertion products **3aj**–**3al** in 87–89% yields. The insertion of vinyl sulfoxonium ylide into alkynoic and alkenoic acids was also investigated. The reaction with propargylic acids afforded insertion products **3am**–**3ao** in excellent yields (98–99%). Due to the presence of electron-deficient alkyne and alkene units in the template of **3am**–**3ao**, they can be useful in various metal- and radical-mediated reactions.^{63–65} Cinnamic acid also gave its corresponding insertion product **3ap** but in a decreased yield (71%). Alkyl acids, such as various arene-substituted 2-arylacetic acids, gave their corresponding insertion products **3aq**–**3at** in good yields. The reaction is also compatible with acetic acid and afforded corresponding insertion product **3au** in 83% yield. An alkyl-substituted vinyl sulfoxonium ylide gave insertion product **3av** in low yield (46%), which was possibly due to the low stability of its corresponding ylide. Interestingly, strong acids, such as sulfonic acids, which produce weak conjugate tosylate nucleophile, also underwent insertion reactions and afforded corresponding sulfonates **3aw** in 52% yield. It is noteworthy that the reaction could be readily scaled up to a 5.0 mmol scale and obtained 1.25 g of the product **3aa**. Moreover, the robustness of the insertion reaction was illustrated by the efficient derivatization of several marketed drugs, such as flavoxate (**3ax**, 64%), ibuprofen (**3ay**, 54%), aspirin (**3az**, 70%), gemfibrozil (**3aza**, 80%), and indomethacin (**3azb**, 75%) indicating the potential of this protocol for the generation of lead compounds in drug discovery.

Next, we would like to explore the present method with various nucleophiles such as thiols, amines, phenols, enols, indoles, and halogen acids (Scheme 4b–e). The reaction of the ylide with the benzenethiol gave the insertion product **3ba** in 98% yield (Scheme 4b). Interestingly, the reaction did not require heating and proceeded smoothly at room temperature. Alkyl groups at the ortho, meta, and para positions and halogen groups at the ortho and para positions of the benzene ring of aryl thiol gave good yields of S–H inserted products **3bb**–**3bg**. The S–H insertion was also tolerated with keto-stabilized vinyl sulfoxonium ylides and obtained their corresponding γ -insertion products **3bk** and **3bl** in 67 and 66% yields, respectively. The reaction is highly chemoselective toward S–H insertion in the presence of other nucleophilic groups, such as –NH₂ or –OH, and afforded the insertion products **3bm** and **3bn** in 70 and 60% yields, respectively. Moreover, the reaction was also investigated using α,β -disubstituted vinyl sulfoxonium ylide with benzene thiol, affording a corresponding S–H insertion product **3bo** in 89% yield.

With the success of acid and thiol insertion, we endeavored to extend the method to the –NH and –OH insertion of amines and alcohols, respectively (Scheme 4c). The optimized reaction conditions utilized for acid and thiols did not afford

Scheme 4. X–H Insertion of Vinyl Sulfoxonium Ylides into Acids, Thiols, Amines, Phenol, Enols, Indoles, and Halogen Acids



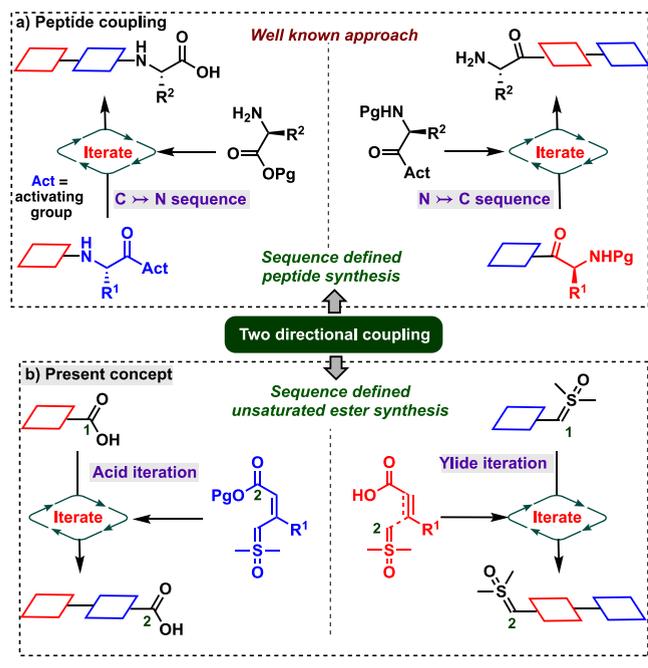
the N–H insertion of amines. The reaction requires an additional acid source to promote the salt formation with sulfoxonium ylide. In the presence of TfOH (30 mol %), the reaction of *N*-methyl aniline with vinyl sulfoxonium ylide afforded the N–H insertion product 3ca in a 66% yield (Scheme 4c). Using primary amines, the reaction yielded lactams 3cb–3cd in 70–75% yields. Similarly, the reaction of vinyl sulfoxonium ylide with 4-bromo phenol also afforded the corresponding O–H insertion product 3da in 66% yield. We then investigated the reactivity of compounds containing an enol ether, such as 4-hydroxycoumarin, with the ylide that gave a mixture of regioisomers resulting from C–H and O–H

insertion products 3ea in 84% yield with 3:1 rr (C–H/O–H) (Scheme 4d). Moreover, we observed regioselective C–H insertion of unprotected indole with the ylide that gave 3eb in 61% yield and 5:1 rr (C–H/N–H). We next explored the γ -insertion of the ylide into the halogen acids (Scheme 4e). The reaction of vinyl sulfoxonium ylides with HCl, HBr, and HI afforded corresponding X–H inserted products 3fa (71%), 3fb (69%), and 3fc (73%), respectively (Scheme 4e).

With the success of these vinylogous insertion reactions, we would like to utilize this method for the iterative synthesis of sequence-defined unsaturated oligoesters in either direction. Generally, peptides are obtained by the coupling of protected

amino acids in either N-to C-terminal direction (Scheme 5a, right) or C- to N-terminal direction (Scheme 5a, left) *via* solid-phase peptide synthesis or conventional solution-mediated synthesis.⁶⁶

Scheme 5. Concept for the Bidirectional Iterative Process



Although peptide coupling has been well known over the past 13 decades, coupling of an amide bond without a coupling reagent is challenging in both the solution and solid phases.^{67,68} Inspired by the coupling of amino acids in either direction (Scheme 5a), we conceptualized reagent-free synthesis of the sequence-defined unsaturated ester by C–O coupling in the solution phase (Scheme 5b). Based on our developed method, the addition of an acid to the vinyl sulfoxonium ylide can lead to the O–H insertion at the γ -position (Scheme 5b, left). Selective saponification of the terminal ester in the insertion product may lead to its corresponding acid, which can be further utilized in the acid iteration sequence. In the other direction, as shown in Scheme 5b (right), the addition of a propargylic acid to the vinyl sulfoxonium ylide gives the insertion products such as **3am–3ao**, as shown in Scheme 4a. To this ynoate, adding dimethylsulfoxonium methylide at the propargylic ester part may lead to new vinyl sulfoxonium ylide, which can be utilized for further iteration.⁶⁹ To the best of our knowledge, no studies have reported the use of a bidirectional iterative coupling process for the synthesis of organic molecules, with the exception of peptide synthesis.^{1,2,70}

At the outset, the iteration was carried out by the sequential addition of an acid to the vinyl sulfoxonium ylide (Scheme 6). The basic strategy for all these sequences is shown in Scheme 6a. Initially, the addition of acid to the trimethyl silyl ethanol-protected vinyl sulfoxonium ylide **4** leads to O–H insertion to give **5**, which contains two ester linkages. Treating insertion product **5** with tetra-*n*-butylammonium fluoride in tetrahydrofuran afforded mono-iterated acid **6**. This acid can be further used in the iteration sequence. Using this strategy (Scheme 6a), seven different iterative sequences were designed to obtain various unsaturated oligoesters, as shown in Scheme

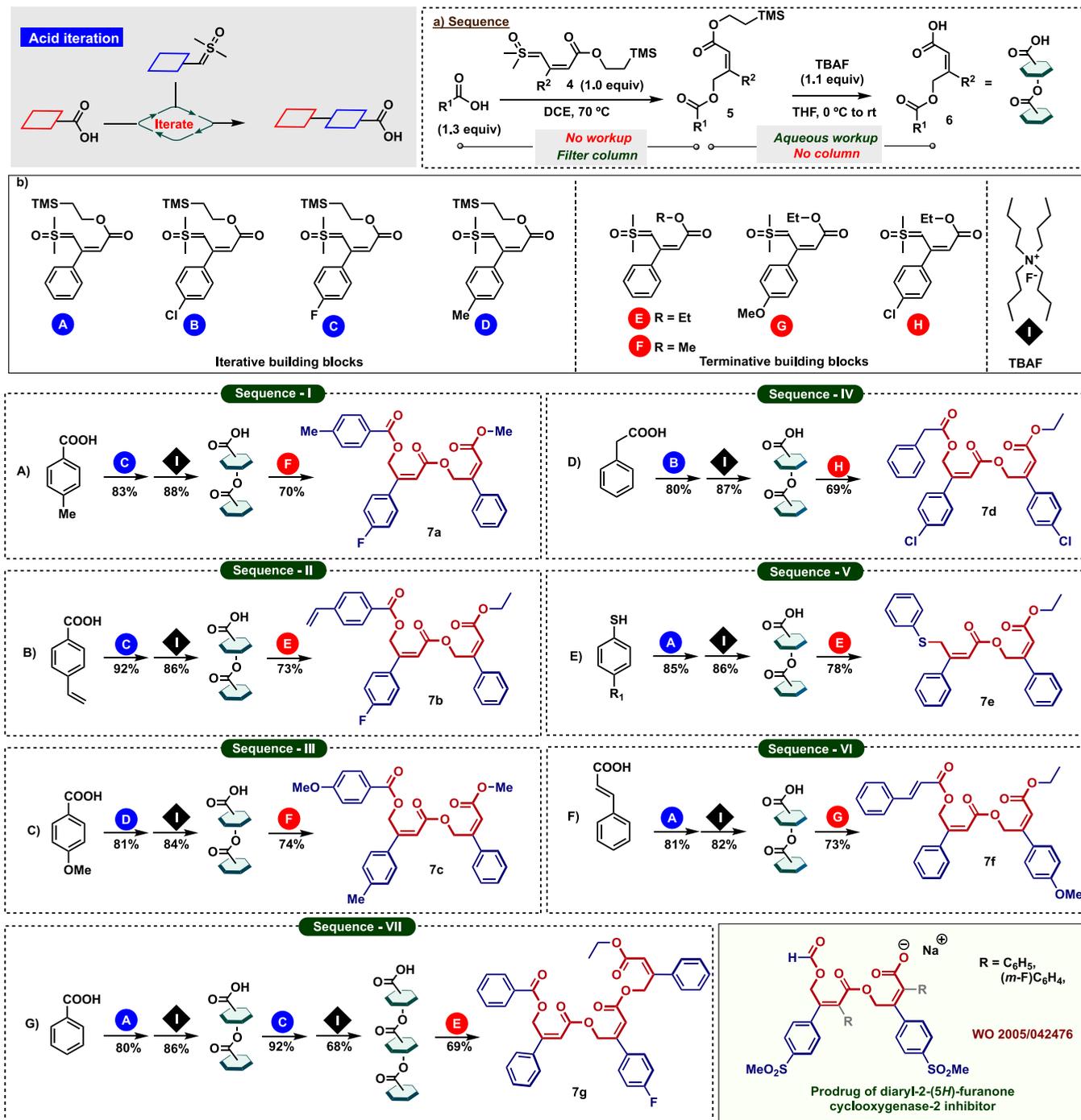
6b. These sequences were initiated using aryl, alkyl, and alkenyl carboxylic acids and aryl thiols. This oligomerization was propagated using trimethylsilyl ethanol-protected vinyl sulfoxonium ylides (A–D) as iterative building blocks. Finally, the sequence ends with ethyl or methyl ester-substituted vinyl sulfoxonium ylides (E–H) as the terminating building blocks and afforded unsaturated oligoesters **7a–7g**. After the first insertion, the reaction mixture was subjected to deprotection of trimethyl silyl ethanol part to obtain **6**, which does not require chromatographic purification for further insertion. However, our attempt to one-pot multi-iteration *via* sequential addition of reagents to obtain the unsaturated oligoesters was unsuccessful. Sequences I–IV and VI are two unit sequences containing three ester linkages. Remarkably, we applied this iteration approach to the four-unit oligomer (sequence-VII). The generated Z-configured unsaturated oligoesters are the basic scaffolds of biologically important compounds. In 2005, researchers from Merck patented that unsaturated oligoesters containing molecules show the ability of prodrugs that convert *in vivo* to diaryl-2-(SH)-furanones, useful in the treatment of cyclooxygenase-2 mediated diseases.⁷¹

With the success of the acid iteration, we next turned our attention to performing the ylide iteration (Scheme 7). Treating phenyl propiolate with dimethylsulfoxonium methylide (obtained from trimethyl sulfoxonium iodide with base) gave vinyl sulfoxonium ylide **1a** in 70% yield. The reaction of phenyl propiolic acid with the vinyl sulfoxonium ylide followed by the addition of dimethylsulfoxonium methylide gave iterated sulfoxonium ylide **8** in 45% yield, and its structure was unambiguously confirmed using X-ray crystallography. A similar sequence was applied using 4-chlorophenylpropionic acid to obtain extended sulfoxonium ylide (**9**, 48%), which can be further utilized in either similar unsaturated oligoesters synthesis or can be used for the carbene-mediated transformation.⁵² It is noteworthy that the iterative generation of carbene precursor was not reported to date. Finally, this sequence was terminated with thiophenol to obtain compound **10** in 72% yield.

The insertion reaction of vinyl sulfoxonium ylide with benzoic acid was monitored using React IR (Scheme 8, left). The reaction of the benzoic acid with the sulfoxonium ylide gave a salt at room temperature, which was indicated by the continuous decay of the ylide concentration and a gradual increase in salt formation. Upon heating the reaction at 70 °C, the resulting salt was converted into the insertion product **3aa**, which indicates that step 2 is the rate-determining step. The similar experiment was performed using thiophenol (Scheme 8, right). We did not notice any salt formation during the reaction at room temperature. This indicates that the second step may be very fast after protonation in the case of thiol addition. This observation strongly suggests that protonation was the rate-limiting step for S–H insertion.

CONCLUSIONS

In conclusion, we have developed a novel bidirectional X–H insertion for the synthesis of unsaturated oligoesters from aryl-substituted vinyl sulfoxonium ylide under metal- and reagent-free conditions. Inspired by vinylogous oligopeptides, we envisioned that vinyl sulfoxonium ylide containing acid derivative could serve as a template to synthesize the Z-configured unsaturated oligoesters *via* sequential acid iteration or sequential ylide iteration. To perform this assembly line synthesis, we have developed a method for the insertion of

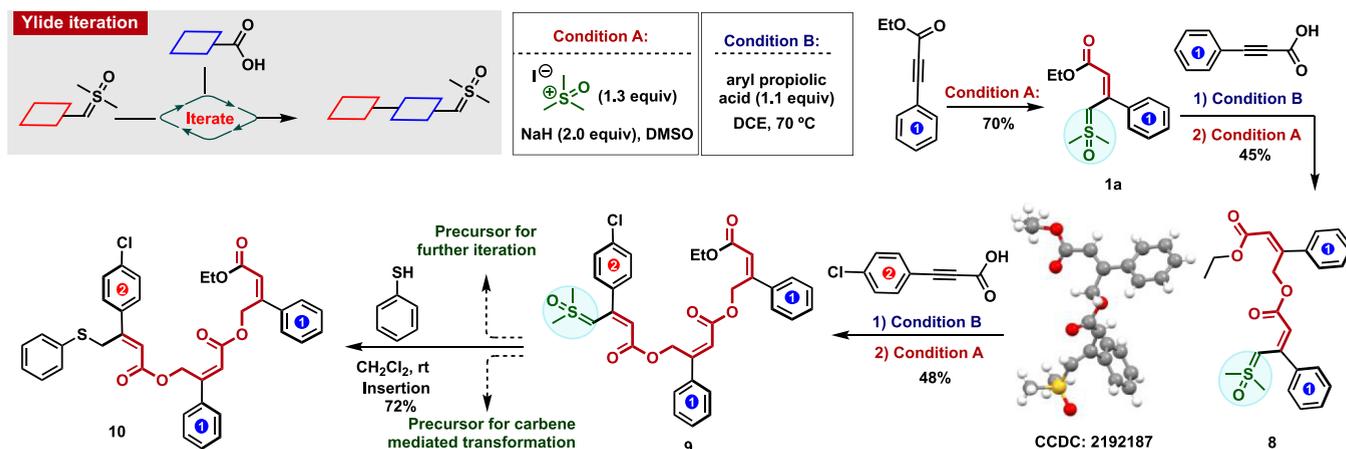
Scheme 6. Unsaturated Oligoester Synthesis *via* Acid Iteration

vinyl sulfoxonium ylide into X–H (X = O, N, S, C, halogen) bonds of acids, thiols, amines, phenols, enols, indoles, and halogen acids under mild reaction conditions. Considering the importance and applications of allylation reactions and metal- and reagent-free coupling of monomers for oligomerization, we believe this simple and convenient strategy will be useful in the pharmaceutical and medicinal chemistry.

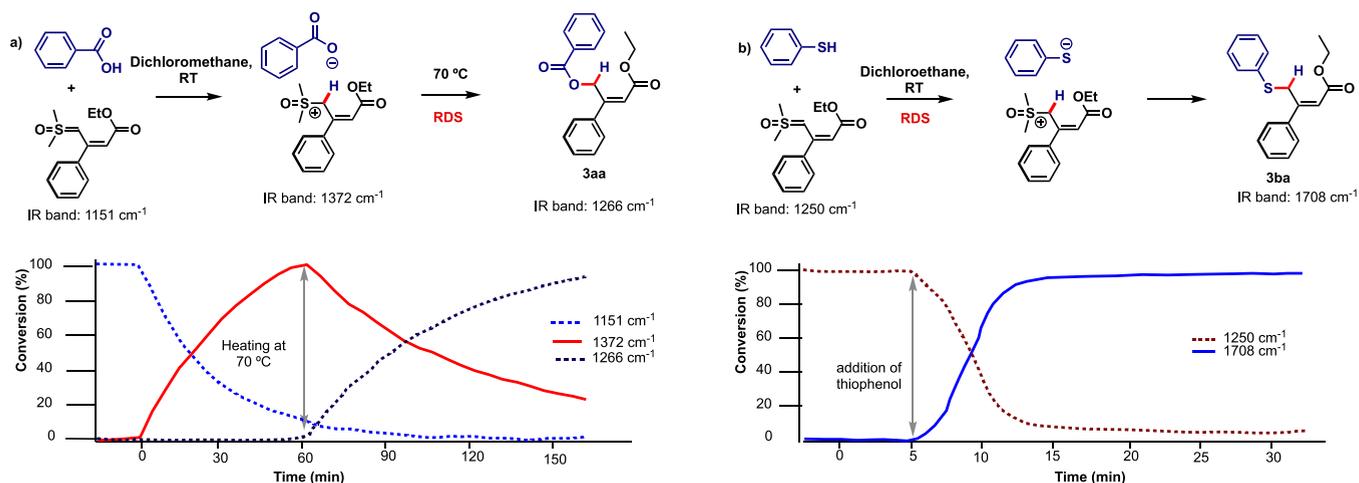
METHODS

General Procedure for Ylide Insertion into the O–H Bond of the Acids (Scheme 4a)

To a reaction tube, crushed 4 Å molecular sieves were added and then the sieves were flame-dried under an argon atmosphere. Vinyl sulfoxonium ylide (0.5 mmol) in dichloroethane (2 mL) was added to the cooled reaction tube. Then, to the stirring solution, aryl or alkyl acid (0.6 mmol) in dichloroethane (3 mL) was added at a time. Then, the reaction was kept on heating at 70 °C for 3–4 h. The reaction mixture was filtered through a pad of Celite (and eluting with dichloromethane) and the filtrate was concentrated *in vacuo*. The residue was purified using flash column chromatography.

Scheme 7. Unsaturated Oligoester Synthesis *via* Ylide Iteration

Scheme 8. React-IR Studies for the Insertion of Vinyl Sulfoxonium Ylide into the X–H Bond of Benzoic Acid and Thiophenol



General Procedure for Thiol S–H and Halogen Acid Insertion Reactions (Scheme 4b,e)

To a reaction tube, crushed 4 Å molecular sieves were added and then the sieves were flame-dried under an argon atmosphere. Vinyl sulfoxonium ylide (0.5 mmol) in dichloromethane (2 mL) was added to the cooled reaction tube. Then, to the stirring solution, benzenethiols or halogen acids (0.6 mmol) in dichloromethane (3 mL) were added at a time. Then, the reaction was kept on stirring at room temperature for 2 h. The reaction mixture was filtered through a pad of Celite by eluting with dichloromethane and the filtrate was concentrated *in vacuo*. The residue was purified using flash column chromatography.

General Procedure for the X–H Insertion of Amines and Phenols (Scheme 4c)

To a reaction tube, crushed 4 Å molecular sieves were added and then the sieves were flame-dried under an argon atmosphere. Vinyl sulfoxonium ylide (0.5 mmol) in dichloroethane (2 mL) was added to the cooled reaction tube. Then, to the stirring solution, primary and secondary amines and phenols (0.6 mmol) in dichloroethane (3 mL) and TfOH (0.3 equiv) were added at a time. Then, the reaction was kept on heating at 80 °C for 16 h. The reaction mixture was filtered through a pad of Celite by eluting with dichloromethane and the filtrate was concentrated *in vacuo*. The residue was purified using flash column chromatography.

General Procedure for Enol and Indole C–H Insertion Reactions (Scheme 4d)

To a reaction tube, crushed 4 Å molecular sieves were added and then the sieves were flame-dried under an argon atmosphere. Vinyl sulfoxonium ylide (0.5 mmol) in dichloroethane (2 mL) was added to the cooled reaction tube. Then, to the stirring solution, indole (0.6 mmol) in toluene (3 mL) and TfOH (0.3 equiv) was added at a time. Then, the reaction was kept on heating at 110 °C for 8–10 h. The reaction mixture was filtered through a pad of Celite by eluting with dichloromethane and the filtrate was concentrated *in vacuo*. The residue was purified using flash column chromatography.

General Procedure for the Acid Iteration (Scheme 6a)

To a reaction tube, crushed 4 Å molecular sieves were added and then the sieves were flame-dried under an argon atmosphere. Vinyl sulfoxonium ylide (0.5 mmol) in dichloroethane (2 mL) was added to the cooled reaction tube. Then, to the stirring solution, carboxylic acid (0.6 mmol) in dichloroethane (3 mL) was added at a time. Then, the reaction was kept on heating at 70 °C for 3–4 h. Then, the concentrated reaction mixture was filtered through a pad of silica using dichloromethane (20 mL). The eluted sulfoxonium ylide was concentrated and dissolved in tetrahydrofuran (0.20 M). To this reaction mixture, tetrabutylammonium fluoride (1.1 equiv) was added dropwise at 0 °C and stirred at room temperature for 2–3 h. After the disappearance of the starting material, the reaction was quenched using water and extracted with diethyl ether and dried over anhydrous Na₂SO₄. The solvent was evaporated and concentrated under high *vacuo*.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/jacsau.2c00641>.

Representative experimental procedures, characterization data, and spectra for all new compounds (PDF)

Crystal data and structure refinement for **8** (CIF)

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<https://pubs.acs.org/doi/10.1021/jacsau.2c00641>

Author Contributions

J.V. supervised this study. The manuscript was written through the contributions of all authors. All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

J.V. acknowledges the financial support from SERB, New Delhi (grant no. SRG/2020/000275). S.B. thanks the University Grants Commission of India for the UGC fellowship. D.K.G. thanks MHRD (GoI) for the Prime Minister's Research Fellowship. We thank Central Research Facility (IIT Delhi) for providing instrumentation facility.

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