Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

Diethyl 1,8-bis(4-methylphenyl)-11-oxatricyclo[6.2.1.$0^{2,7}$]undeca-2,4,6-triene-9,10-dicarboxylate

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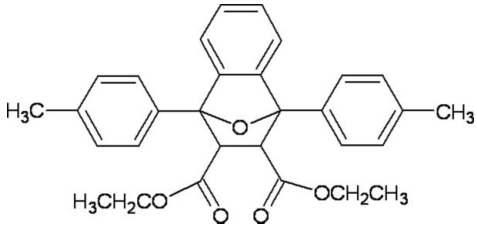
Received 30 January 2013; accepted 24 February 2013

Key indicators: single-crystal X-ray study; T = 293 K; mean σ(C–C) = 0.002 Å; disorder in main residue; R factor = 0.038; wR factor = 0.112; data-to-parameter ratio = 12.5.

The title compound, $C_{30}H_{30}O_5$, is the Diels–Alder adduct from 1,3-diphenylbenzo[*c*]furan and diethyl maleate. The molecule comprises a fused tricyclic system containing two five-membered rings, which are in envelope conformations with the O atom at the flap, and a six-membered ring adopting a boat conformation. The dihedral angle between the 4-methylphenyl substituents in the 1- and 8-positions is 62.1 (1)°. The ethyl group of one ester group and the ethoxy group of the other ester group are disordered over two sets of sites, with occupancy ratios of 0.43 (2):0.57 (2) and 0.804 (7):0.196 (7), respectively. In the crystal, inversion dimers are formed through pairs of C—H···O interactions.

Related literature

For background to Diels–Alder reactions, see: Akio & Toshiki (2010). For related structures, see: Bailey *et al.* (1995); Takahashi *et al.* (2003); Simpson *et al.* (2004); Toze *et al.* (2010). For puckering and asymmetry parameters, see: Cremer & Pople (1975); Nardelli (1983).



Experimental

Crystal data

$C_{30}H_{30}O_5$
M_r = 470.54
Triclinic, $P\overline{1}$
a = 9.8722 (3) Å
b = 10.7413 (3) Å
c = 13.3081 (3) Å
α = 109.319 (1)°
β = 105.045 (1)°
γ = 90.374 (1)°
V = 1279.45 (6) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 0.08 mm^{-1}
T = 293 K
0.30 × 0.20 × 0.20 mm

Data collection

Bruker Kappa APEXII CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2004)
T_{min} = 0.951, T_{max} = 0.953
21475 measured reflections
4505 independent reflections
3754 reflections with $I > 2\sigma(I)$
R_{int} = 0.027

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.038
$wR(F^2)$ = 0.112
S = 1.03
4505 reflections
360 parameters
84 restraints
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.23 e Å^{-3}
$\Delta\rho_{min}$ = −0.18 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C5—H5···O4^{i}	0.93	2.66	3.558 (2)	164
C11—H11···O2^{ii}	0.93	2.66	3.433 (2)	141

Symmetry codes: (i) $-x, -y+2, -z$; (ii) $x, y-1, z$.

Data collection: *APEX2* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012) and *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*, *PLATON* and *publCIF* (Westrip, 2010).

The authors thank Dr Babu Varghese, SAIF, IIT-Madras, India, for his help with the data collection.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IM2419).

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supplementary materials

Acta Cryst. (2013). E**69**, o500 [doi:10.1107/S1600536813005291]

Diethyl 1,8-bis(4-methylphenyl)-11-oxatricyclo[6.2.1.$0^{2,7}$]undeca-2,4,6-triene-9,10-dicarboxylate

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Comment

Diels-Alder adducts are valuable synthetic intermediates and their use for the synthesis of natural products and of polycyclic aromatic hydrocarbons is well documented (Akio & Toshiki, 2010). The title compound, $C_{30}H_{30}O_5$, comprises a fused tricyclic system and two 4-methylphenyl rings attached to this system (Fig.1). The tricyclic system consists of two 5-membered rings and one aromatic ring. In addition, two ethyl carboxylate units are attached to the tricyclic system. Geometrical parameters agree well with reported structures (Bailey *et al.* 1995; Takahashi *et al.* 2003; Simpson *et al.*, 2004; Toze *et al.*, 2010). The 5-membered ring $C_1\backslash C_2\backslash C_7\backslash C_8\backslash O_1$ adopts an envelope conformation with atom O_1 displaced by -(0.757) Å from the mean plane of the other ring atoms $C_1\backslash C_2\backslash C_7\backslash C_8$. The puckering parameters (Cremer & Pople, 1975) and asymmetry parameters (Nardelli, 1983) are q_2 = 0.515 (1) Å, φ = 143.2 (2)°, $\Delta_S(O_1)$ = 0.007 (1)° and $\Delta_2(O_1)$ = 0.312 (1)°.The second 5-membered ring $C_1\backslash C_{23}\backslash C_{27}\backslash C_8\backslash O_1$ also adopts an envelope conformation with O_1 displaced by -(0.835) Å from the mean plane of the other ring atoms $C_1\backslash C_{23}\backslash C_{27}\backslash C_8$. The puckering parameters (Cremer & Pople, 1975) and asymmetry parameters (Nardelli, 1983) are q_2 = 0.593 (1) Å, φ = -37.5 (1)°, $\Delta_S(O_1)$ = 0.012 (1)° and $\Delta_2(O_1)$ = 0.355 (1)°. The six membered ring $C_1/C_2/C_7/C_8/C_{27}/C_{23}$ adopts a boat conformation with puckering parameter q_2 = 0.951 (1) Å, θ=89.6 (9)° and φ = 359.9 (9)°.

The dihedral angle between the rings $C_1/C_2/C_7/C_8/O_1$ and $C_1/C_{23}/C_{27}/C_8/O_1$ is 66.1 (1)°. The dihedral angle between the terminal 4-methylphenyl rings is 62.1 (1)°. One of the aromatic substituents (C9 - C15) is almost orthogonal to the plane formed by the six atoms C1, C2, C7, C8, C27 and C23 of the tricyclic ring, the dihedral angle being 84.8 (1)° (Nardelli, 1983). The atoms O5, C29 and C30 and C26 of ester groups are disordered over two sites with occupancy ratios of 0.804 (7): 0.196 (7) and 0.43 (2): 0.57 (2). The ester group is twisted from the mean plane of the tricyclic ring, with C30 towards C28 as evidenced by the torsion angle C30—C29—O5—C28 = 89.1 (5)°. The second ester group is co-planer with the attached trycyclic ring as evidenced by the torsion angle C24—O3—C25—C26 = -172 (9)°. Centrosymmetric dimers are formed by C—H···O interactions.

Experimental

1,3-Di-p-tolylisobenzofuran 4 (1.00 g, 3.36 mmole) was dissolved in toluene (25 ml) and treated with 2 equivalents of dimethyl maleate (1.16 g, 1.15 ml, 6.74 mmole). The reaction mixture was refluxed and the reaction was monitored by TLC. After 8 h, the mixture was cooled to room temperature. The solvent was removed and the residue was purified by column chromatography (Silica gel, 10%, EA/hexane) to give the adduct as a white solid. Yield: 1.41 g (89%) and m.p. 179° C. This adduct was crystallized from $CHCl_3/CH_3OH$ (3:1) by the slow evaporation method.

Refinement

All H atoms were positioned geometrically and allowed to ride on their parent atoms, with (C—H= 0.93–0.96 Å), and U_{iso}(H) = 1.5 U_{eq}(C) for methyl H atoms and 1.2U_{eq}(C) for other H atoms. In one ester group the ethyl group and in the other ester group the ethoxy group are disordered over two sites with occupancy ratio of 0.43 (2): 0.57 (2), 0.804 (7): 0.196 (7), respectively.

Computing details

Data collection: *APEX2* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT* (Bruker, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012) and *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008), *PLATON* (Spek, 2009) and *publCIF* (Westrip, 2010).

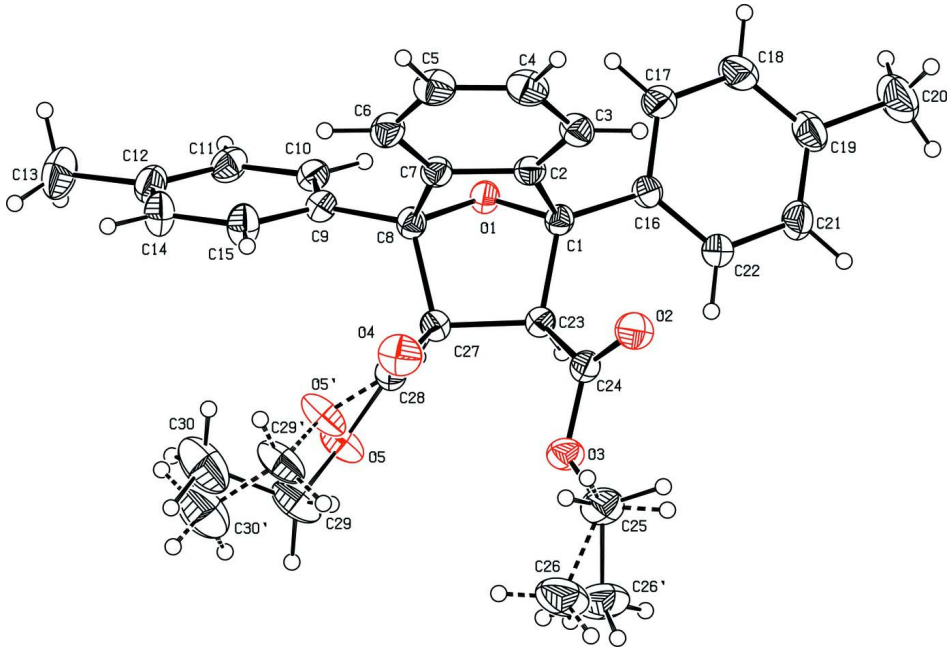

Figure 1

Molecular structure of the title compound, showing 30% probability displacement ellipsoids.

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Crystal data

$C_{30}H_{30}O_5$
M_r = 470.54
Triclinic, $P\overline{1}$
Hall symbol: -P 1
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c = 13.3081 (3) Å
α = 109.319 (1)°
β = 105.045 (1)°
γ = 90.374 (1)°
V = 1279.45 (6) Å^3

Z = 2
F(000) = 500
D_x = 1.221 Mg m^{-3}
Mo $K\alpha$ radiation, λ = 0.71073 Å
Cell parameters from 7905 reflections
θ = 2.0–25.0°
μ = 0.08 mm^{-1}
T = 293 K
Block, colourless
0.30 × 0.20 × 0.20 mm

Data collection

Bruker Kappa APEXII CCD diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
ω and φ scan
Absorption correction: multi-scan (*SADABS*; Bruker, 2004)
T_{min} = 0.951, T_{max} = 0.953
21475 measured reflections
4505 independent reflections
3754 reflections with $I > 2\sigma(I)$
R_{int} = 0.027
θ_{max} = 25.0°, θ_{min} = 2.0°
h = −11→11
k = −12→12
l = −15→15

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.038
$wR(F^2)$ = 0.112
S = 1.03
4505 reflections
360 parameters
84 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0574P)^2 + 0.3047P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max}$ = 0.23 e Å^{-3}
$\Delta\rho_{min}$ = −0.18 e Å^{-3}
Extinction correction: *SHELXL97* (Sheldrick, 2008), $Fc^*=kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Extinction coefficient: 0.028 (3)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	*y*	*z*	U_{iso}*/U_{eq}	Occ. (<1)
C1	0.16153 (14)	1.11798 (13)	0.42413 (11)	0.0342 (3)	
C2	0.18246 (14)	1.12120 (14)	0.31608 (11)	0.0364 (3)	
C3	0.25194 (16)	1.21282 (16)	0.29039 (13)	0.0452 (4)	
H3	0.2958	1.2926	0.3444	0.054*	
C4	0.25420 (18)	1.18202 (19)	0.18119 (15)	0.0555 (4)	
H4	0.3011	1.2418	0.1616	0.067*	
C5	0.18819 (18)	1.06438 (19)	0.10146 (14)	0.0554 (4)	
H5	0.1923	1.0454	0.0289	0.066*	
C6	0.11570 (17)	0.97368 (17)	0.12708 (13)	0.0482 (4)	
H6	0.0697	0.8949	0.0727	0.058*	
C7	0.11386 (14)	1.00369 (14)	0.23586 (11)	0.0376 (3)	
C8	0.05102 (14)	0.93225 (13)	0.29702 (11)	0.0362 (3)	
C9	0.02041 (15)	0.78454 (14)	0.24879 (12)	0.0393 (3)	
C10	0.08168 (15)	0.70361 (14)	0.30645 (12)	0.0404 (3)	
H10	0.1472	0.7408	0.3753	0.049*	

C11	0.04615 (18)	0.56778 (15)	0.26241 (14)	0.0485 (4)	
H11	0.0881	0.5153	0.3027	0.058*	
C12	−0.04982 (19)	0.50845 (15)	0.16043 (14)	0.0524 (4)	
C13	−0.0889 (3)	0.36078 (18)	0.11335 (19)	0.0792 (6)	
H13A	−0.1722	0.3401	0.0520	0.119*	
H13B	−0.1065	0.3306	0.1695	0.119*	
H13C	−0.0128	0.3174	0.0885	0.119*	
C14	−0.1090 (2)	0.58990 (17)	0.10260 (15)	0.0589 (5)	
H14	−0.1727	0.5524	0.0329	0.071*	
C15	−0.07535 (18)	0.72575 (16)	0.14621 (14)	0.0534 (4)	
H15	−0.1177	0.7782	0.1060	0.064*	
C16	0.27490 (14)	1.19021 (14)	0.52804 (11)	0.0366 (3)	
C17	0.41180 (16)	1.15731 (18)	0.53553 (14)	0.0532 (4)	
H17	0.4309	1.0915	0.4764	0.064*	
C18	0.51974 (18)	1.2206 (2)	0.62912 (15)	0.0632 (5)	
H18	0.6105	1.1958	0.6328	0.076*	
C19	0.49591 (18)	1.32047 (19)	0.71785 (14)	0.0566 (4)	
C20	0.6155 (2)	1.3927 (3)	0.81909 (18)	0.0905 (8)	
H20A	0.6696	1.4543	0.8027	0.136*	
H20B	0.6750	1.3298	0.8398	0.136*	
H20C	0.5779	1.4399	0.8791	0.136*	
C21	0.35987 (18)	1.35224 (17)	0.71005 (13)	0.0525 (4)	
H21	0.3411	1.4185	0.7691	0.063*	
C22	0.25038 (16)	1.28856 (15)	0.61707 (12)	0.0434 (4)	
H22	0.1594	1.3121	0.6144	0.052*	
C23	0.00553 (14)	1.13855 (13)	0.42249 (11)	0.0337 (3)	
H23	−0.0045	1.1398	0.4942	0.040*	
C24	−0.04509 (15)	1.26460 (14)	0.40726 (11)	0.0373 (3)	
C25	−0.24375 (19)	1.38633 (18)	0.39529 (18)	0.0635 (5)	
H25A	−0.2326	1.3968	0.3281	0.076*	0.43 (2)
H25B	−0.2002	1.4658	0.4574	0.076*	0.43 (2)
H25C	−0.2742	1.3743	0.3170	0.076*	0.57 (2)
H25D	−0.1768	1.4649	0.4329	0.076*	0.57 (2)
C26	−0.3945 (9)	1.3647 (15)	0.3868 (18)	0.091 (3)	0.43 (2)
H26A	−0.4391	1.2929	0.3197	0.136*	0.43 (2)
H26B	−0.4385	1.4439	0.3859	0.136*	0.43 (2)
H26C	−0.4040	1.3432	0.4493	0.136*	0.43 (2)
C26′	−0.3651 (11)	1.4015 (12)	0.4408 (11)	0.087 (2)	0.57 (2)
H26D	−0.4239	1.3189	0.4107	0.130*	0.57 (2)
H26E	−0.4183	1.4687	0.4214	0.130*	0.57 (2)
H26F	−0.3328	1.4268	0.5200	0.130*	0.57 (2)
C27	−0.07443 (14)	1.00649 (13)	0.33396 (11)	0.0347 (3)	
H27	−0.1136	0.9575	0.3721	0.042*	
C28	−0.19370 (16)	1.02059 (15)	0.24236 (13)	0.0422 (4)	
O1	0.15563 (9)	0.97526 (9)	0.40324 (7)	0.0361 (2)	
O2	0.02359 (12)	1.34954 (11)	0.39779 (10)	0.0543 (3)	
O3	−0.17832 (11)	1.27117 (10)	0.41131 (9)	0.0479 (3)	
O4	−0.18455 (12)	1.07858 (13)	0.18168 (10)	0.0586 (3)	
O5	−0.3145 (3)	0.9610 (3)	0.2455 (3)	0.0586 (7)	0.804 (7)

C29	−0.4441 (3)	0.9681 (3)	0.1669 (3)	0.0782 (10)	0.804 (7)
H29A	−0.4388	1.0503	0.1522	0.094*	0.804 (7)
H29B	−0.5227	0.9671	0.1980	0.094*	0.804 (7)
C30	−0.4678 (5)	0.8522 (5)	0.0606 (3)	0.1017 (14)	0.804 (7)
H30A	−0.3971	0.8603	0.0247	0.152*	0.804 (7)
H30B	−0.5594	0.8512	0.0126	0.152*	0.804 (7)
H30C	−0.4619	0.7712	0.0765	0.152*	0.804 (7)
O5′	−0.2954 (12)	0.9430 (15)	0.2035 (11)	0.064 (2)	0.196 (7)
C29′	−0.4114 (12)	0.9582 (17)	0.1155 (14)	0.077 (3)	0.196 (7)
H29C	−0.3818	0.9476	0.0494	0.093*	0.196 (7)
H29D	−0.4453	1.0448	0.1395	0.093*	0.196 (7)
C30′	−0.5239 (16)	0.8499 (19)	0.0939 (19)	0.104 (5)	0.196 (7)
H30D	−0.4892	0.7652	0.0683	0.156*	0.196 (7)
H30E	−0.6058	0.8556	0.0385	0.156*	0.196 (7)
H30F	−0.5486	0.8599	0.1612	0.156*	0.196 (7)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0362 (7)	0.0306 (7)	0.0349 (7)	0.0041 (5)	0.0088 (6)	0.0109 (6)
C2	0.0330 (7)	0.0402 (8)	0.0362 (7)	0.0066 (6)	0.0103 (6)	0.0125 (6)
C3	0.0409 (8)	0.0479 (9)	0.0481 (9)	0.0005 (7)	0.0134 (7)	0.0173 (7)
C4	0.0531 (10)	0.0686 (11)	0.0566 (10)	0.0035 (8)	0.0230 (8)	0.0306 (9)
C5	0.0562 (10)	0.0763 (12)	0.0407 (9)	0.0097 (9)	0.0200 (8)	0.0240 (9)
C6	0.0495 (9)	0.0536 (9)	0.0361 (8)	0.0078 (7)	0.0112 (7)	0.0091 (7)
C7	0.0360 (7)	0.0395 (8)	0.0359 (7)	0.0067 (6)	0.0090 (6)	0.0119 (6)
C8	0.0370 (7)	0.0348 (7)	0.0330 (7)	0.0049 (6)	0.0051 (6)	0.0104 (6)
C9	0.0407 (8)	0.0352 (8)	0.0405 (8)	0.0050 (6)	0.0124 (6)	0.0101 (6)
C10	0.0417 (8)	0.0385 (8)	0.0411 (8)	0.0076 (6)	0.0136 (6)	0.0119 (6)
C11	0.0582 (10)	0.0386 (8)	0.0545 (10)	0.0128 (7)	0.0228 (8)	0.0179 (7)
C12	0.0623 (10)	0.0361 (8)	0.0558 (10)	0.0021 (7)	0.0230 (8)	0.0071 (7)
C13	0.1085 (17)	0.0395 (10)	0.0787 (14)	−0.0028 (10)	0.0258 (13)	0.0067 (9)
C14	0.0661 (11)	0.0458 (9)	0.0473 (10)	−0.0056 (8)	0.0021 (8)	0.0040 (8)
C15	0.0610 (10)	0.0415 (9)	0.0464 (9)	0.0022 (7)	0.0000 (8)	0.0121 (7)
C16	0.0367 (7)	0.0369 (7)	0.0354 (7)	0.0023 (6)	0.0078 (6)	0.0132 (6)
C17	0.0410 (9)	0.0607 (10)	0.0460 (9)	0.0083 (7)	0.0083 (7)	0.0062 (8)
C18	0.0366 (9)	0.0819 (13)	0.0579 (11)	0.0053 (8)	0.0048 (8)	0.0136 (10)
C19	0.0463 (9)	0.0692 (11)	0.0438 (9)	−0.0116 (8)	0.0026 (7)	0.0137 (8)
C20	0.0609 (12)	0.123 (2)	0.0564 (12)	−0.0219 (13)	−0.0039 (10)	0.0063 (12)
C21	0.0569 (10)	0.0503 (9)	0.0393 (9)	−0.0051 (8)	0.0113 (7)	0.0030 (7)
C22	0.0411 (8)	0.0431 (8)	0.0417 (8)	0.0023 (6)	0.0109 (6)	0.0095 (7)
C23	0.0352 (7)	0.0345 (7)	0.0316 (7)	0.0046 (6)	0.0087 (6)	0.0119 (6)
C24	0.0393 (8)	0.0354 (7)	0.0349 (7)	0.0052 (6)	0.0090 (6)	0.0102 (6)
C25	0.0553 (10)	0.0506 (10)	0.0903 (14)	0.0237 (8)	0.0199 (10)	0.0313 (10)
C26	0.051 (4)	0.080 (6)	0.151 (9)	0.023 (3)	0.028 (5)	0.053 (6)
C26′	0.068 (4)	0.083 (5)	0.127 (6)	0.040 (3)	0.044 (4)	0.047 (5)
C27	0.0352 (7)	0.0334 (7)	0.0361 (7)	0.0027 (6)	0.0081 (6)	0.0141 (6)
C28	0.0403 (8)	0.0376 (8)	0.0431 (8)	0.0057 (6)	0.0047 (7)	0.0119 (7)
O1	0.0377 (5)	0.0324 (5)	0.0344 (5)	0.0049 (4)	0.0045 (4)	0.0106 (4)

O2	0.0534 (7)	0.0397 (6)	0.0768 (8)	0.0059 (5)	0.0227 (6)	0.0256 (6)
O3	0.0405 (6)	0.0424 (6)	0.0644 (7)	0.0132 (5)	0.0161 (5)	0.0219 (5)
O4	0.0597 (7)	0.0728 (8)	0.0491 (7)	0.0125 (6)	0.0080 (6)	0.0337 (6)
O5	0.0323 (9)	0.0749 (13)	0.0728 (16)	−0.0031 (8)	0.0011 (10)	0.0415 (13)
C29	0.0372 (14)	0.107 (2)	0.089 (2)	−0.0006 (13)	−0.0061 (14)	0.0509 (19)
C30	0.075 (3)	0.137 (3)	0.084 (3)	−0.027 (2)	−0.0103 (18)	0.053 (2)
O5′	0.037 (4)	0.087 (4)	0.073 (5)	−0.016 (3)	−0.001 (4)	0.046 (4)
C29′	0.037 (5)	0.110 (6)	0.079 (6)	−0.010 (4)	−0.006 (4)	0.042 (5)
C30′	0.063 (8)	0.126 (9)	0.110 (9)	−0.004 (7)	−0.010 (7)	0.048 (8)

Geometric parameters (Å, º)

C1—O1	1.4626 (16)	C20—H20C	0.9600
C1—C16	1.4973 (19)	C21—C22	1.380 (2)
C1—C2	1.5161 (19)	C21—H21	0.9300
C1—C23	1.5528 (18)	C22—H22	0.9300
C2—C3	1.375 (2)	C23—C24	1.5039 (19)
C2—C7	1.383 (2)	C23—C27	1.5557 (19)
C3—C4	1.386 (2)	C23—H23	0.9800
C3—H3	0.9300	C24—O2	1.1918 (17)
C4—C5	1.375 (3)	C24—O3	1.3317 (17)
C4—H4	0.9300	C25—O3	1.4513 (19)
C5—C6	1.384 (2)	C25—C26′	1.464 (6)
C5—H5	0.9300	C25—C26	1.474 (8)
C6—C7	1.380 (2)	C25—H25A	0.9700
C6—H6	0.9300	C25—H25B	0.9700
C7—C8	1.518 (2)	C25—H25C	0.9700
C8—O1	1.4429 (16)	C25—H25D	0.9700
C8—C9	1.4967 (19)	C26—H26A	0.9600
C8—C27	1.5750 (19)	C26—H26B	0.9600
C9—C15	1.381 (2)	C26—H26C	0.9600
C9—C10	1.385 (2)	C26′—H26D	0.9600
C10—C11	1.384 (2)	C26′—H26E	0.9600
C10—H10	0.9300	C26′—H26F	0.9600
C11—C12	1.378 (2)	C27—C28	1.5082 (19)
C11—H11	0.9300	C27—H27	0.9800
C12—C14	1.385 (2)	C28—O5′	1.185 (12)
C12—C13	1.505 (2)	C28—O4	1.1908 (18)
C13—H13A	0.9600	C28—O5	1.367 (3)
C13—H13B	0.9600	O5—C29	1.448 (3)
C13—H13C	0.9600	C29—C30	1.506 (5)
C14—C15	1.382 (2)	C29—H29A	0.9700
C14—H14	0.9300	C29—H29B	0.9700
C15—H15	0.9300	C30—H30A	0.9600
C16—C22	1.379 (2)	C30—H30B	0.9600
C16—C17	1.386 (2)	C30—H30C	0.9600
C17—C18	1.375 (2)	O5′—C29′	1.464 (9)
C17—H17	0.9300	C29′—C30′	1.505 (10)
C18—C19	1.382 (3)	C29′—H29C	0.9700
C18—H18	0.9300	C29′—H29D	0.9700

C19—C21	1.374 (2)	C30′—H30D	0.9600
C19—C20	1.511 (2)	C30′—H30E	0.9600
C20—H20A	0.9600	C30′—H30F	0.9600
C20—H20B	0.9600		

O1—C1—C16	110.12 (10)	C22—C21—H21	119.1
O1—C1—C2	100.29 (10)	C16—C22—C21	120.49 (14)
C16—C1—C2	117.12 (11)	C16—C22—H22	119.8
O1—C1—C23	99.13 (10)	C21—C22—H22	119.8
C16—C1—C23	118.59 (11)	C24—C23—C1	116.13 (11)
C2—C1—C23	108.50 (11)	C24—C23—C27	116.58 (11)
C3—C2—C7	121.60 (13)	C1—C23—C27	102.32 (10)
C3—C2—C1	132.64 (13)	C24—C23—H23	107.1
C7—C2—C1	105.76 (12)	C1—C23—H23	107.1
C2—C3—C4	117.65 (15)	C27—C23—H23	107.1
C2—C3—H3	121.2	O2—C24—O3	124.22 (13)
C4—C3—H3	121.2	O2—C24—C23	126.09 (13)
C5—C4—C3	120.99 (15)	O3—C24—C23	109.61 (12)
C5—C4—H4	119.5	O3—C25—C26′	108.0 (4)
C3—C4—H4	119.5	O3—C25—C26	107.3 (6)
C4—C5—C6	121.24 (15)	C26′—C25—C26	26.7 (5)
C4—C5—H5	119.4	O3—C25—H25A	110.2
C6—C5—H5	119.4	C26′—C25—H25A	130.6
C7—C6—C5	117.93 (15)	C26—C25—H25A	110.2
C7—C6—H6	121.0	O3—C25—H25B	110.2
C5—C6—H6	121.0	C26′—C25—H25B	85.8
C6—C7—C2	120.57 (14)	C26—C25—H25B	110.2
C6—C7—C8	134.20 (14)	H25A—C25—H25B	108.5
C2—C7—C8	105.22 (12)	O3—C25—H25C	110.1
O1—C8—C9	112.25 (11)	C26′—C25—H25C	110.1
O1—C8—C7	100.69 (10)	C26—C25—H25C	86.1
C9—C8—C7	118.93 (12)	H25A—C25—H25C	26.2
O1—C8—C27	98.67 (10)	H25B—C25—H25C	128.7
C9—C8—C27	113.99 (11)	O3—C25—H25D	110.1
C7—C8—C27	109.66 (11)	C26′—C25—H25D	110.2
C15—C9—C10	118.17 (14)	C26—C25—H25D	131.3
C15—C9—C8	120.12 (13)	H25A—C25—H25D	84.6
C10—C9—C8	121.66 (13)	H25B—C25—H25D	26.4
C11—C10—C9	120.54 (15)	H25C—C25—H25D	108.4
C11—C10—H10	119.7	C25—C26—H26A	109.5
C9—C10—H10	119.7	C25—C26—H26B	109.5
C12—C11—C10	121.65 (15)	C25—C26—H26C	109.5
C12—C11—H11	119.2	C25—C26′—H26D	109.5
C10—C11—H11	119.2	C25—C26′—H26E	109.5
C11—C12—C14	117.43 (15)	C25—C26′—H26F	109.5
C11—C12—C13	121.41 (17)	C28—C27—C23	115.71 (11)
C14—C12—C13	121.17 (17)	C28—C27—C8	116.03 (11)
C12—C13—H13A	109.5	C23—C27—C8	101.22 (10)
C12—C13—H13B	109.5	C28—C27—H27	107.8

H13A—C13—H13B	109.5	C23—C27—H27	107.8
C12—C13—H13C	109.5	C8—C27—H27	107.8
H13A—C13—H13C	109.5	O5′—C28—O4	110.8 (6)
H13B—C13—H13C	109.5	O5′—C28—O5	26.3 (6)
C15—C14—C12	121.43 (16)	O4—C28—O5	126.40 (18)
C15—C14—H14	119.3	O5′—C28—C27	120.1 (6)
C12—C14—H14	119.3	O4—C28—C27	125.79 (14)
C9—C15—C14	120.77 (16)	O5—C28—C27	107.73 (16)
C9—C15—H15	119.6	C8—O1—C1	98.05 (9)
C14—C15—H15	119.6	C24—O3—C25	116.27 (12)
C22—C16—C17	118.01 (14)	C28—O5—C29	116.9 (2)
C22—C16—C1	123.37 (13)	O5—C29—C30	110.0 (3)
C17—C16—C1	118.62 (13)	O5—C29—H29A	109.7
C18—C17—C16	120.91 (15)	C30—C29—H29A	109.7
C18—C17—H17	119.5	O5—C29—H29B	109.7
C16—C17—H17	119.5	C30—C29—H29B	109.7
C17—C18—C19	121.23 (16)	H29A—C29—H29B	108.2
C17—C18—H18	119.4	C28—O5′—C29′	119.8 (10)
C19—C18—H18	119.4	O5′—C29′—C30′	104.5 (11)
C21—C19—C18	117.54 (15)	O5′—C29′—H29C	110.8
C21—C19—C20	121.38 (17)	C30′—C29′—H29C	110.8
C18—C19—C20	121.07 (17)	O5′—C29′—H29D	110.8
C19—C20—H20A	109.5	C30′—C29′—H29D	110.8
C19—C20—H20B	109.5	H29C—C29′—H29D	108.9
H20A—C20—H20B	109.5	C29′—C30′—H30D	109.5
C19—C20—H20C	109.5	C29′—C30′—H30E	109.5
H20A—C20—H20C	109.5	H30D—C30′—H30E	109.5
H20B—C20—H20C	109.5	C29′—C30′—H30F	109.5
C19—C21—C22	121.80 (15)	H30D—C30′—H30F	109.5
C19—C21—H21	119.1	H30E—C30′—H30F	109.5

O1—C1—C2—C3	−148.50 (15)	C18—C19—C21—C22	−0.7 (3)
C16—C1—C2—C3	−29.4 (2)	C20—C19—C21—C22	178.76 (18)
C23—C1—C2—C3	108.15 (17)	C17—C16—C22—C21	0.4 (2)
O1—C1—C2—C7	31.14 (13)	C1—C16—C22—C21	−179.34 (14)
C16—C1—C2—C7	150.21 (12)	C19—C21—C22—C16	−0.2 (3)
C23—C1—C2—C7	−72.21 (13)	O1—C1—C23—C24	−162.37 (11)
C7—C2—C3—C4	−1.6 (2)	C16—C1—C23—C24	78.66 (15)
C1—C2—C3—C4	177.97 (15)	C2—C1—C23—C24	−58.21 (15)
C2—C3—C4—C5	0.5 (2)	O1—C1—C23—C27	−34.26 (12)
C3—C4—C5—C6	0.9 (3)	C16—C1—C23—C27	−153.23 (11)
C4—C5—C6—C7	−1.2 (2)	C2—C1—C23—C27	69.90 (12)
C5—C6—C7—C2	0.1 (2)	C1—C23—C24—O2	−0.9 (2)
C5—C6—C7—C8	−178.61 (15)	C27—C23—C24—O2	−121.67 (16)
C3—C2—C7—C6	1.3 (2)	C1—C23—C24—O3	−177.87 (11)
C1—C2—C7—C6	−178.37 (13)	C27—C23—C24—O3	61.39 (15)
C3—C2—C7—C8	−179.61 (13)	C24—C23—C27—C28	0.28 (17)
C1—C2—C7—C8	0.70 (14)	C1—C23—C27—C28	−127.54 (12)
C6—C7—C8—O1	146.03 (16)	C24—C23—C27—C8	126.58 (12)

C2—C7—C8—O1	−32.84 (13)	C1—C23—C27—C8	−1.25 (12)
C6—C7—C8—C9	23.0 (2)	O1—C8—C27—C28	162.98 (11)
C2—C7—C8—C9	−155.84 (12)	C9—C8—C27—C28	−77.88 (15)
C6—C7—C8—C27	−110.65 (17)	C7—C8—C27—C28	58.28 (15)
C2—C7—C8—C27	70.48 (13)	O1—C8—C27—C23	36.89 (11)
O1—C8—C9—C15	−178.75 (13)	C9—C8—C27—C23	156.03 (11)
C7—C8—C9—C15	−61.68 (19)	C7—C8—C27—C23	−67.81 (13)
C27—C8—C9—C15	70.13 (18)	C23—C27—C28—O5′	−145.8 (9)
O1—C8—C9—C10	3.92 (18)	C8—C27—C28—O5′	95.8 (9)
C7—C8—C9—C10	120.99 (15)	C23—C27—C28—O4	56.7 (2)
C27—C8—C9—C10	−107.20 (15)	C8—C27—C28—O4	−61.7 (2)
C15—C9—C10—C11	−0.7 (2)	C23—C27—C28—O5	−120.24 (18)
C8—C9—C10—C11	176.67 (13)	C8—C27—C28—O5	121.37 (18)
C9—C10—C11—C12	0.4 (2)	C9—C8—O1—C1	178.96 (11)
C10—C11—C12—C14	0.6 (2)	C7—C8—O1—C1	51.43 (11)
C10—C11—C12—C13	−179.54 (17)	C27—C8—O1—C1	−60.60 (11)
C11—C12—C14—C15	−1.3 (3)	C16—C1—O1—C8	−174.89 (10)
C13—C12—C14—C15	178.88 (18)	C2—C1—O1—C8	−50.84 (11)
C10—C9—C15—C14	0.1 (2)	C23—C1—O1—C8	60.00 (11)
C8—C9—C15—C14	−177.36 (15)	O2—C24—O3—C25	4.4 (2)
C12—C14—C15—C9	0.9 (3)	C23—C24—O3—C25	−178.61 (13)
O1—C1—C16—C22	−122.79 (14)	C26′—C25—O3—C24	−160.0 (6)
C2—C1—C16—C22	123.55 (15)	C26—C25—O3—C24	172.0 (9)
C23—C1—C16—C22	−9.7 (2)	O5′—C28—O5—C29	−60.0 (15)
O1—C1—C16—C17	57.42 (17)	O4—C28—O5—C29	0.5 (4)
C2—C1—C16—C17	−56.24 (18)	C27—C28—O5—C29	177.4 (2)
C23—C1—C16—C17	170.51 (13)	C28—O5—C29—C30	89.0 (5)
C22—C16—C17—C18	0.2 (3)	O4—C28—O5′—C29′	−20.1 (14)
C1—C16—C17—C18	180.00 (16)	O5—C28—O5′—C29′	111 (2)
C16—C17—C18—C19	−1.1 (3)	C27—C28—O5′—C29′	179.3 (9)
C17—C18—C19—C21	1.3 (3)	C28—O5′—C29′—C30′	−176 (2)
C17—C18—C19—C20	−178.1 (2)		

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C5—H5···O4^{i}	0.93	2.66	3.558 (2)	164
C11—H11···O2^{ii}	0.93	2.66	3.433 (2)	141

Symmetry codes: (i) −*x*, −*y*+2, −*z*; (ii) *x*, *y*−1, *z*.