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Distal Ionic Substrate–Catalyst Interactions Enable Long-Range Stereocontrol: Access to Remote Quaternary Stereocenters through a Desymmetrizing Suzuki–Miyaura Reaction

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ABSTRACT: Spatial distancing of a substrate's reactive group and nonreactive catalyst-binding group from its pro-stereogenic element presents substantial hurdles in asymmetric catalysis. In this context, we report a desymmetrizing Suzuki–Miyaura reaction that establishes chirality at a remote quaternary carbon. The anionic, chiral catalyst exerts stereocontrol through electrostatic steering of substrate's reactive group and charged catalyst-binding group become increasingly distanced. This study demonstrates that precise long-range stereocontrol is achievable by engaging ionic substrate–ligand interactions at a distal position.

The remarkable ability of enzymes to utilize attractive noncovalent interactions with distant, nonreactive groups of substrates to accelerate reactions and modulate selectivity has been regarded as a fundamental distinction from smallmolecule catalysts.¹ In recent years, substantial advances, particularly by Phipps and co-workers,^{2,3} have been achieved in harnessing distal ionic substrate-ligand interactions to control regio- and site-selectivity of transition-metal-catalyzed transformations (Scheme 1a, top).⁴ By contrast, integrating distal ionic interactions represents a compelling, yet undeveloped enantiocontrol strategy in transition-metal catalysis (Scheme 1a, bottom).⁵ In a prominent work, Miller and co-workers accomplished remote desymmetrization⁶ through asymmetric Ullmann coupling (Scheme 1b).⁷ Mechanistic investigation revealed an exquisite preorganization through proximal trifluoroacetamide anion-Cu binding and a distal Cs⁺ bridge between the substrate's nonreacting enantiotopic arene and the peptide ligand's terminal carboxylate.^{7c} Notwithstanding, chiral ligand scaffolds bearing nonligating charged groups (mostly tethered chirotopic ionic groups to date⁸) are uncommon, and the general effects of ion-ion interaction's low directionality on long-range enantioinduction have not been studied.⁹ The broad potential of asymmetric transition-metal catalysis directed by distal ionic interactions has remained underexploited.

In pursue of such an enantiocontrol strategy, we targeted an untapped class of stereocenters through a transformation that allows us to rigorously test its viability. To date, remote desymmetrization to trisubstituted stereocenters has been made possible by only a handful of ingenious catalysts,^{6,7,10} and creation of remote quaternary carbon stereocenters has remained elusive.¹¹ Quaternary stereocenters embedded in fluorenes¹² and xanthenes¹³ possess distinctive ability to project chirality to distant loci of three-dimensional dispositions, an appealing feature for functional materials and pharmaceuticals¹⁴ (Scheme 1c). However, these enantioenriched molecules are accessible only through chiral chromatography.¹⁵ We envisaged that Pd-catalyzed desymmetrizing Suzuki–Miyaura reaction¹⁶ of bis(chloroaryl)methane derivatives could furnish this class of core quaternary stereocenters (Scheme 1d).

Establishing quaternary stereocenters bearing sterically similar geminal substituents poses a major obstacle in catalytic desymmetrization,¹⁷ and distant reactive groups may conceivably exacerbate the challenge. We were drawn to the design principle by Phipps and co-workers using cation bridges between anionic substrates and sulfonated dialkylbiaryl phosphines¹⁸ in site-selective cross-coupling of dichloroarenes.³ We surmised that a novel anionic dialkylbiphenyl phosphine-Pd catalyst could interact with the charged substituent $(Z^{-}M^{+})$ of the substrate preferentially (Scheme 1d). Furthermore, we reasoned that integrating the catalyst's axial chirality^{19,20}—spatial arrangement of Pd and phosphonate-into stereocontrol relay from ionic group Z to C-Cl bonds could be a viable approach to long-range asymmetric induction. As such, the effects of spatial distancing of ionic group Z and C-Cl bonds could be elucidated through judicious variation of the substrates. Here, we report that catalyst-controlled electrostatic steering of substrates led to realization of an enantioselective desymmetrizing Suzuki-Miyaura reaction that establishes chirality at remote quaternary stereocenters.

We commenced our study by synthesizing 3'-phosphonate dialkylbiphenyl phosphines (Scheme 2). Racemic L1, readily prepared from RuPhos,²¹ was converted to L2 as separable atropo-diastereomers in three steps. Upon desulfinylation, the axial chirality of L2 was preserved in the resulting individual

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Scheme 1. Desymmetrization Strategy for Remote Quaternary Stereocenters Directed by Distal Ionic Interactions





Scheme 2. Synthesis of Anionic, Axially Chiral Ligands



enantiomers of L1 by a methyl "atropo-tag". Subsequent phosphonylation and hydrolysis afforded enantioenriched L4. Besides, L5 (depicted in Table 1) was prepared following an analogous synthetic route starting from SPhos.²²

The nature of substrate's catalyst-binding group is anticipated to influence the stereochemical outcome of desymmetrization if distal substrate-ligand interactions are operating. Therefore, we evaluated a range of Brønsted acidic groups³ (Table 1). Each group is separated from fluorene C9 by four rotatable bonds. This way, differences between their steric effects imposed on the pro-stereogenic center are

Table 1. Effect of Pendent Group on Pd-Catalyzed Desymmetrizing Suzuki–Miyaura Reaction^a



^{*a*}Reaction conditions: substrate (0.1 mmol), aryl boronic acid (0.1 mmol), [Pd(η^3 -C₃H₅)Cl]₂ (1.0 mol%), (S)-L4 (2.2 mol%), K₃PO₄ (3 equiv), THF (9.5 mL/mmol), H₂O (0.5 mL/mmol), 60 °C, 18 h. Isolated yields reported. Enantiomeric ratios (er) were determined by chiral high-performance liquid chromatography. ^{*b*}Isolated as ethyl ester. Tf = trifluoromethanesulfonyl.

minimized. Using (S)-L4 as ligand, the substrate bearing a distal triflamide underwent the desymmetrizing Suzuki-Miyaura reaction, affording the product in an encouraging 44% yield with 73:27 er (1). Replacing the triflamide with sulfo group (2) and carboxyl group (3) led to markedly improved results. By contrast, pendent hydrogen bond donors (4-6) resulted in comparably low enantioselectivity.

Subsequently, we focused our efforts on reaction optimization (Tables S1–S5 in the Supporting Information (SI)). The model reaction gave merely 56:44 er using SPhos-derived (S)-L5 as ligand (Table 1, 3). Investigating solvent effect using (S)-L4, we found that the enantioselectivity diminished in DMF (66:34 er). This observation is consistent with a participating cation bridge, which is disrupted by strong solvation of cations in polar aprotic solvents.^{7c} To probe the effects of cations, we surveyed alkali-metal hydroxides and carbonates as exogenous base. Similar results were observed using Na, K, and Cs bases irrespective of the counteranions (96:4–97:3 er), while Li bases were inferior. The reaction remained enantioselective using Bu₄NOH as base (91:9 er), suggesting that stereocontrol is attainable in the organic phase. Finally, a 2-MeTHF-aqueous K₃PO₄ system was identified as the optimal reaction media.

We next studied the effects of distancing the ionic pendent group (Table 2a, 3 and 7-12). Initially, we anticipated a steep drop in enantioselectivity once the distance between C-Cl bond and the distal carboxylate exceeds the span of catalyst. The entropic penalty incurred could obliterate the energetic differentiation of desymmetrization. Surprisingly, the catalyst system adapted well to changes in length of $(CH_2)_n$ (n = 1-7)linking the carboxyl group (32–67% yield, 82.5:17.5–96:4 er). Notably, desymmetrization was achieved even when the carboxylate was placed eight C-C bonds away from the quaternary carbon (12, 86.5:13.5 er). The results also substantiate the attractive nature of substrate-ligand interactions involving the distal carboxylate. Repulsive forces unlikely play the dominant role, because they can be easily avoided by shifting the carboxylate away without affecting the catalysis at the Pd center.

Furthermore, increasing the conformational rigidity by incorporating a double bond into the linker only led to

Table 2. Substrate Scope of Pd-Catalyzed Remote Desymmetrization to Quaternary Stereocenters^{a,b}



^aThe absolute configurations of products were assigned by analogy to 37. ^bStandard reaction conditions: substrate (0.25 mmol), aryl boronic acid (0.30 mmol), $Pd_2(dba)_3$ (1.0 mol%), (S)-L4 (2.2 mol%), K_3PO_4 (10 equiv), 2-MeTHF (20 mL/mmol), H_2O (1.6 mL/mmol), 60 °C, 18 h. Isolated yields reported. ^cIsolated as ethyl ester. dba = dibenzylideneacetone.

marginal decrease in enantioselectivity (Table 2a, 13, 91.5:8.5 er).

The ability to direct the catalyst is not unique to carboxylate, which presumably serves as a diffuse negative charge occupying the distal end of C9 substituent. Recently, fluorenes bearing pendent sulfonate have emerged as prominent components of conjugated polyelectrolytes.²³ Investigation of sulfo group in the desymmetrization reactions revealed that it functioned equally well, affording the products in up to 95:5 er (Table 2a, 2 and 14).

Besides the catalyst's effectiveness in long-range stereocontrol, we were excited by its ability to construct quaternary stereocenters bearing sterically similar geminal substituents. Gratifyingly, variations in the non-ionic C9 substituents including alkyl, benzyl, and phenyl groups, were well tolerated (Table 2b, 15–19, 61–71% yield, 89:11-94.5:5.5 er). Clearly, the size difference between the geminal substituents is not the main determinant of enantioselectivity.

The transformation is compatible with a broad spectrum of arylboronic acids (Table 2c). Substituents at the *para-* (20–22), *meta-* (23 and 24), and *ortho-* (25–27) positions, irrespective of electronic properties, had an insignificant influence on the enantioselectivity (57–73% yield, 93:7–97.5:2.5 er). Additionally, a wide range of polycyclic aromatics commonly employed in π -conjugated materials can be installed in 61–70% yield, 92.5:7.5–98:2 er (28–33).

The remote desymmetrization strategy is also applicable to accessing enantioenriched xanthenes (Table 2d). Specifically, dichloroxanthenes participated in the transformation with various electron-rich aryl (34-36), electron-deficient aryl (37-39), heteroaryl (40-42), and polycyclic aryl (43-46) boronic acids, affording the products in 42-70% yield, 93:7-97.5:2.5 er.

Intrigued by the catalyst's ability in exerting long-range stereocontrol, we further evaluated its adaptability to distancing the reactive group and to altering the catalystbinding substituent. First, we placed the C–Cl bonds farther apart (Scheme 3a). Despite the substantial structural change in the substrates, the catalyst remained capable of imparting asymmetric induction (47 and 48, up to 89.5:10.5 er). Next, we studied the stereochemical outcome of incorporating an oxygen atom adjacent to the pro-stereogenic carbon, which possibly provides additional interaction with the K⁺ bridge (Scheme 3b). Indeed, the remote desymmetrization reactions proceeded in up to 99:1 er (49 and 50).

Based on the results of control experiments, we concluded that K⁺, phosphonate of (S)-L4, and carboxylate of substrate contribute collectively to the ionic substrate—ligand interactions (Scheme 3c). Encapsulation of K⁺ by 18-crown-6 led to diminished enantioselectivity (61:39 er), and reduction in er paralleled the quantity of added 18-crown-6 (SI). The critical role of ligand's phosphonate was evidenced by the negligible enantioinduction by truncated ligand (*R*)-L1 (56:44 er). In comparison, the reaction using (S)-L3 gave 83:17 er. The ion dipole interaction between K⁺ and P=O of (S)-L3 is inferior to the ion—ion interaction between K⁺ and P $-O^-$ of (S)-L4 in asymmetric induction. In contrast to the preformed carboxylate salt (52), racemic product was obtained from corresponding ethyl ester (53), which lacks the key ion—ion interactions with K⁺.

The oxidative addition $step^{24}$ is plausibly selectivitydetermining, while other steps in the catalytic cycle could contribute to the enantioselectivity.²⁵ On the basis of the

Scheme 3. Substrate Scope Expansion and Control Experiments to Probe the Effect of Spatial Distancing and the Role of Distal Ionic Interactions



absolute configurations of (S)-L4 and 37, we hypothesized a model²⁶ to illustrate the putative distal ionic interactions (Scheme 3d). Unlike enzymes' large and deep binding clefts that confer substrate specificity, Pd-(S)-L4, which carries a diffuse negative charge at an unshielded phosphonate, preserves distal ionic interactions when it adapts to substrates' structural diversity in pendent groups and linkers, non-ionic substituents (R), and distanced C-Cl bonds.

Nature utilizes long-range electrostatic attractions to significantly accelerate biochemical processes that require precise orientations of biomolecules.²⁷ We postulated that the Pd-catalyzed remote desymmetrization follows the same principle of electrostatic steering of charged substrates.²⁸ To elucidate this phenomenon, we carried out competition experiments between carboxylate acid 51 and ethyl ester 53 (Scheme 3e). Under the standard conditions, 51 reacted predominantly regardless of the electronic property of aryl boronic acids (entries 1 and 2). Such selectivity is catalystcontrolled, as competition experiments using RuPhos slightly favored 53 (entries 3 and 4). The observations, coupled with the noticeable difference between their enantioselectivities (Scheme 3c), indicate that compared with 53, the ionic interactions arising from distal carboxylate of 51 lead to a preferential increase in the rate of selectivity-determining step at one of the enantiotopic reaction sites.

The desymmetrization strategy offers efficient access to core quaternary stereocenters that project substituents to widely spaced positions (Scheme 4). As an illustration, 3 underwent Pd-catalyzed C-B, C-C, and C-N bond formation reactions (Scheme 4a), furnishing combinations of functionalities at two distant sites (54–56). Moreover, the sequential desymmetriz-

Scheme 4. Synthetic Applications of Pd-Catalyzed Desymmetrizing Suzuki-Miyaura Reaction





ing cross-coupling is enantiodivergent (Scheme 4b). Starting from 57, the stereochemical outcome was precisely controlled by the choreography of heteroaryl and alkenyl boronic acids (58 and 59), where both enantiomers of 60 were synthesized using (S)-L4 as ligand. Subsequent transformations afforded spirocycle 61 as a β -secretase inhibitor¹³ analog.

As a practical feature, the remote desymmetrization can be readily adopted to construct chiral building blocks of fluorenebased materials without rerouting existing syntheses. For example, desymmetrization of **51** with 4-B(dan) phenylboronic pinacol ester (dan = naphthalene-1,8-diaminato) proceeded smoothly on a 1 mmol scale using 1 mol% Pd-(S)-L4, affording AB-type monomer^{12d} 63 in 97:3 er upon deprotection of coupling product 62. Notably, we also succeeded in synthesizing enantioenriched (99:1 er) AA-type monomer 64 in one step using 1,4-phenylenediboronic pinacol ester as bis-coupling partner (Scheme 4c). Additionally, the pendent carboxyl group can be readily converted to other

functionalities, such as ethylene glycol chain of a chiral precursor for polyimine dynamers^{12c} (Scheme 4d, 65).

In summary, we have realized a desymmetrizing Suzuki– Miyaura reaction that establishes chirality at a remote quaternary carbon. The anionic catalyst's ability to transmit asymmetry across large distances enables facile access to enantioenriched molecules that project chirality to widely spaced loci. We have demonstrated that by engaging distal ionic substrate–catalyst interactions, it is possible to surmount the hurdle in asymmetric catalysis arising from spatial distancing of substrate's reactive group and catalyst-binding group. We anticipate that pursuing this strategy could stimulate rational design of catalysts capable of long-range asymmetric induction to create chirality that would be difficult to construct using conventional methods.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/jacs.1c12345.

Experimental details and characterization data (PDF)

Accession Codes

CCDC 2054848 and 2054849 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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