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Five-year study of the effects of simulated nitrogen deposition levels and forms on soil nitrous oxide emissions from a temperate forest in northern China

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Abstract

Few studies have quantified the effects of different levels and forms of nitrogen (N) deposition on soil nitrous oxide (N₂O) emissions from temperate forest soils. A 5-year field experiment was conducted to investigate the effects of multiple forms and levels of N additions on soil N₂O emissions, by using the static closed chamber method at Xi Mountain Experimental Forest Station in northern China. The experiment included a control (no N added), and additions of NH_4NO_3 , $NaNO_3$, and $(NH_4)_2SO_4$ that each had two levels: 50 kg N ha⁻¹ yr⁻¹ and 150 kg N ha⁻¹ yr⁻¹. All plots were treated to simulate increased N deposition on a monthly schedule during the annual growing season (March to October) and soil N₂O emissions were measured monthly from March 2011 to February 2016. Simultaneously, the temperature, moisture, and inorganic N contents of soil were also measured to explore how the main factors may have affected soil N2O emission. The results showed that the types and levels of N addition significantly increased soil inorganic N contents, and the accumulation of soil NO₃⁻-N was significantly higher than that of soil NH₄⁺-N due to N addition. The three N forms significantly increased the average N₂O emissions (P < 0.05) in the order of NH₄NO₃ > (NH₄)₂SO₄ > NaNO₃ by 355.95%, 266.35%, and 187.71%, respectively, compared with control. The promotion of N₂O emission via the NH₄⁺–N addition was significantly more than that via the NO_3^--N addition, while N addition at a high level exerted a stronger effect than at the low-level. N addition exerted significantly stronger effects on cumulative N₂O emissions in the initial years, especially the third year when the increased cumulative N₂O emission reached their maximum. In the later years, the increases persisted but were weakened. Increasing inorganic N concentration could change soil from being N-limited to N-rich, and then N-saturated, and so the promotion on soil available N effect increased and then decreased. Moreover, the soil NH_4^+ –N, NO_3^- -N, temperature, and water-filled pore space were all positively correlated with soil N2O emissions. These findings suggest that atmospheric N deposition can significantly promote soil N₂O emission, and that exogenous NH_4^+ –N and NO_3^- -N inputs into temperate forests can have synergic effects on soil N₂O



role in the study design, data collection and analysis, decision to publish, or preparation of the manuscript. The specific roles of these authors are articulated in the 'author contributions' section.and analysis, decision to publish, or preparation of the manuscript.

Competing interests: The affiliation with Beijing Solid Waste Treatment Co., Ltd, This does not alter our adherence to PLOS ONE policies on sharing data and materials. emission. In future research, both aspects should be better distinguished in the N cycle and balance of terrestrial ecosystems by using ¹⁵N tracer methods.

Introduction

Nitrous oxide (N₂O) is not only a potent greenhouse gas whose global warming potential is 298- and 21-fold that of CO_2 and CH_4 , but it also contributes to stratospheric ozone depletion [1]. Emissions of N₂O from soil have been identified as the primary source (57%) of total global N₂O emissions [2].

Nitrification and denitrification are the two main processes that produce N_2O in soils and both can occur simultaneously (Fig 1). N_2O is produced by denitrifying bacteria during the reduction of NO_3^- or NO_2^- to N_2O and N_2 , or released as an intermediate product when nitrifying bacteria oxidize NH_4^+ -N to NO_3^- and NO_2^- [3]. These two processes may be affected by soil water content, temperature, N availability and pH, as well as other particular biotic or abiotic properties [4–6]. Inorganic N is a key factor regulating soil N₂O emission [4–5, 7–8] (Fig 1). In general, increasing available mineral N in soils leads to enhanced N₂O formation and emission via increased nitrification and denitrification rates [9]. Soil N₂O emission is also driven by soil temperature and water content [10]. Some previous studies indicated soil N₂O emissions were increased under conditions of higher soil water content and soil temperature [10]. The latter may regulate soil N₂O emission by influencing N₂O-producing microorganisms, such as nitrifying and denitrifying bacteria [11]. Furthermore, low soil moisture can reduce the temperature sensitivity of soil microbes, so that the diffusion of extracellular enzymes in the substrate are lowered [12].



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China is now ranked third behind Europe and North America in terms of the scale of anthropogenic reactive N emissions, and has been experiencing a dramatic increase in anthropogenic reactive N due to its rapid economic development [10]. The average N deposition in our study area was 13.2 kg N ha⁻¹ yr⁻¹ in the 1980s and 21.1 kg N ha⁻¹ yr⁻¹ in the 2000s [13]. Alongside increases in N deposition there have been decreases in the ratio of NH_4^+ –N to NO_3^- -N deposition, from approximately 5 to 2, from the 1980s to the 2000s, although NH_4^+ –N remains the dominant form of N deposition [13]. Nationally, N deposition is a more serious issue in the north compared with the other regions of China [14].

Increasing N deposition could influence the production and emission of N₂O by disturbing the balance between microbial N mineralization and immobilization, with the consequences for the relative availability of soil NH₄⁺-N and NO₃⁻-N [15] (Fig 1). Most studies report that raising N addition levels could linearly stimulate soil N₂O emissions [4-5, 16-17]. A metaanalysis of global N addition experiments showed that N additions increased soil N₂O emissions by an average of 134% in terrestrial ecosystems [18]. Some plausible mechanisms have been proposed to clarify the promotion effect of N addition for soil N_2O emissions: (1) Without additional N, the N retention in soil is mainly used by plants and microorganisms to maintain biomass and growth, so less N becomes lost as gaseous N [19]; (2) The amount of additional N greatly exceeds the atmospheric N deposition, thus leading to N accumulation in forest soil, which can benefit nitrifying and denitrifying bacteria [20] which would stimulate the nitrification rate and N₂O emissions [21]. However, some studies indicated that N addition has no significant effect on soil N2O emission, which might be attributed to particular N addition threshold level for increased N_2O emissions [7, 22]. Thornton and Valente [23] found that the increased rate of soil N₂O emissions was low at high N-addition levels; this may have occurred because the high level N addition to soil drove other limitations, such as carbon availability, thereby decreasing the C/N ratios that regulate the status of N saturation, which likely had a strong influence on N₂O emission [24]. Furthermore, some studies have shown denitrification to be the main source of soil N_2O emissions [25–26], whereas other studies reported that nitrification were primarily responsible for soil N_2O emissions [7, 27–28]. Clearly then, how soil N₂O emissions respond to additional N appears to be inconsistent.

The NH_4^+ - N/NO_3^- -N ratio showed a decreasing trend in our study area [29], and so clarifying the response of soil N₂O emission to different forms and levels of N addition now is necessary. However, several previous studies that stimulated N deposition only considered NH_4NO_3 [2, 4, 6], while others that did examine N deposition in varied N forms only reported their short-term effects on N₂O emission [30]. In addition, some studies have focused on soil core incubations in the laboratory [31], which are conditions that differ greatly from those in the field. Therefore, from both a scientific and management perspective, further examination of the characteristics of different levels and forms of N addition is critically important for better understanding how N deposition affects soil N₂O emissions in temperate forest soils.

In our study, we report the results of continuous measurements of soil N₂O emissions over a 5-year period from a temperate forest in northern China. Based on the above analysis, we hypothesized that (1) N addition could increase soil N₂O emission and that this promotion effect likely increased with the N addition level; (2) Applying NO_3^- and NH_4^+ -N in combination could promote soil N₂O emission more than would their respective single applications.

Materials and methods

Study area

The study was conducted in a temperate forest of the Xi Mountain Experimental Forest Station (31°54′32″ N, 110°68′08″ E, 133 m a.s.l.) in Beijing, northern China. The station belongs to

Beijing Forestry University. The study area is characterized by a temperate continental monsoon climate with a maximum air temperature of 31° C in July and a minimum of -9° C in January. Mean annual temperature is 11.6°C and the average annual precipitation is 630 mm. During the 5-year experimental period, the yearly maximum and minimum temperatures were, respectively, 31, 31, 32, 33, 31° C and -9, -8, -5, -5° C, while the total precipitation received annually was 721, 759, 508, 500, and 459 mm. At this research station, *Quercus liaotungensis* is the zonal vegetation with an average age of 62 years. The diameter at breast height, canopy closure, average height, and density were 9.7 cm, 69%, 8.4 m, and 2963 trees ha⁻¹. The soil here is classified as Chromic Luvisols (WRB Soil Classification) composed of 51% sand, 40% silt, and 9% clay. The thickness of the soil humus horizon (A horizon) is approximately 3–5 cm, and the O horizon thickness < 3 cm. Before starting the experiment, soil samples from the upper 10 cm of soil in each plot (with three replicates) were collected by using corers in March 2011. Initial soil properties were measured and showed no significant differences among the plots (Table 1).

Experimental design

The experiments were performed from March 2011 to February 2016. Seven $10 \text{ m} \times 10 \text{ m} \text{ N}$ addition plots, with three replicates each (n = 21 plots in total), were randomly established and distributed on a flat ground dominated by the Quercus liaotungensis community at the research station. To ensure plot independence, 1.5-m buffer strips were set up between adjacent plots. As deposition of NH_4^+ -N and NO_3^- -N showed great variation from month to month in the study area [30], three N-addition forms, namely NaNO₃, (NH₄)₂SO₄, and NH_4NO_3 , were used to simulate the effects of deposited NH_4^+-N , NO_3^--N , and their combination. According to the current level of atmospheric N deposition (30.6 kg N ha⁻¹ yr⁻¹) at the experimental site [30], two N-addition levels referred to as low N (L: 50 kg N ha⁻¹ yr⁻¹) and high N (H: 150 kg N ha⁻¹ yr⁻¹) were used to simulate a future increase in atmospheric N deposition by 1.5-fold and 5-fold. A control (0 kg N ha^{-1} yr⁻¹) was used to calculate the net effect of naturally occurring N addition to the soil. From 2011 to 2015, additional N was evenly sprayed on the soil surface in plots by using sprayers, with eight equal applications made from March to October (i.e., the growing season). If it rained, the scheduled N addition was postponed to 1 day after the rain day. To reduce the effect of additional water on the experiment, control plots received an equivalent deionized water treatment.

Gas sampling and measurement

Soil N₂O emission measurements were performed three times in the first week of each month, from March 2011 to February 2016. Soil N₂O emissions were measured using a static closed

			Treatment							
Variable	Control	NaNO ₃		(NI	H ₄) ₂ SO ₄	N	H₄NO₃			
		L	н	L	н	L	Н			
pН	7.18±0.28a	7.02±0.39a	7.16±0.49a	7.01±0.35a	7.20±0.28a	7.19±0.39a	7.12±0.21a			
Organic C (g kg ⁻¹)	29.97±0.88a	30.14±1.45a	29.45±1.73a	28.37±1.65a	29.15±0.83a	30.33±1.38a	31.06±1.67a			
Total N (g kg ⁻¹)	2.43±0.68a	2.35±0.84a	2.36±0.72a	2.40±0.68a	2.54±0.60a	2.39±0.66a	2.44±0.48a			
NH4 ⁺ -N (mg/kg)	2.82±0.20a	2.51±0.46a	2.37±0.43a	2.45±0.22a	2.41±0.36a	2.47±0.44a	2.55±0.36a			
NO ₃ ⁻ -N (mg/kg)	11.83±1.01a	12.80±1.02a	12.87±1.18a	13.05±1.09a	12.78±1.26a	13.06±1.22a	13.15±1.36a			

Table 1. Soil properties of the sampling area.

L: 50 kg N ha⁻¹ yr⁻¹; H: 150 kg N ha⁻¹ yr⁻¹.

Treatments with same letter mean no significant difference in the whole row parameters.

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opaque chamber and gas chromatography method [32]. The chamber was made of stainless steel and consisted of a fixed base and a removable top (without bottom, length \times width \times height = 50 cm \times 50 cm \times 50 cm). Before measurement, the base, which supported the sampling chamber, was installed into the soil at a depth of 20 cm for the entire experiment to avoid soil disturbance. Soil temperatures were measured in each plot at a depth of 5 cm nearby the chamber before and after collecting gas samples. And the average temperature value was used for emission calculation. The fixed base frame was free of vegetation. When collecting the gases, we inserted the removable top into the fixed base. The chamber was covered with thermal insulation cotton to reduce the impact of direct radiative heating in the chamber and a digital thermometer in the chamber was used to record its air temperature. Two fans were used to increase mixing and uniformity of air in the chamber.

Gas samples were collected three times, from a sampling outlet at the top of the chamber, from 09:00 to 11:00 AM. local time on the first, fourth, and seventh day after N addition in each month from March 2011 to February 2016. If unpredicted extreme weather occurred, such as heavy rain or snow, this gas sampling was rescheduled. Gas samples were taken using 100 mL plastic syringes at intervals of 0, 10, 20, and 30 minutes after closing the chamber and inserting polyethylene-coated aluminum bags for soil N₂O concentration analysis. Gas samples were analyzed within 6 h in a gas chromatograph (Agilent 7890A, Agilent Technologies Inc., Palo Alto, CA, USA) [33].

Soil N₂O emissions were calculated as follows [33]:

$$F_{N_2O} = D \times H \times (\Delta c / \Delta t)$$
⁽¹⁾

where, $F_{N_{2}O}$ refers to N_2O emission (µg m⁻² h⁻¹); D refers to the gas density of the chamber (mol m⁻³); D = WP/RT; W refers to the molar mass of N_2O (g mol⁻¹); P refers to air pressure (Pa); T refers to the air temperature inside the chamber (K); R refers to the gas constant (J mol⁻¹ K⁻¹); H refers to the height of the sampling chamber (m); and $\Delta c/\Delta t$ denotes the linear slope of the concentration change over the measurement period.

Soil cumulative N_2O emissions were calculated by interpolating the N_2O emissions measured between sampling periods [34]. Cumulative N_2O emissions were calculated spanning the time period from March to February next year as follows [35]:

Cumulative N₂O emissions =
$$\frac{\sum_{i=1}^{n} 0.5 \times (F_i + F_{i+1}) \times (t_{i+1} - t_i) \times 24}{100000}$$
 (2)

where, F is the N₂O emissions (μ g m⁻² h⁻¹); *i* is the sampling number, i.e., samples collected in March had a value of 1 and those collected next February had a value of 12; and t is the sampling time based on the Julian day.

Soil sampling and measurement

Considering that N₂O release mainly occurred in the mineral horizon, litter was first removed from the soil surface (O horizon < 3 cm) when sampling the soil. Soil samples at 0–10 cm depth were collected from near the static chambers monthly. Soil samples were passed through a 2-mm sieve to remove roots, gravel, and stones for soil analyses. Part of the fresh soil was used for soil NH_4^+ –N and NO_3^- –N content analyses, while the remaining portion was airdried for pH measurement. Soil NO_3^- –N and NH_4^+ –N concentrations were determined by the KCl extraction method [5]. Soil water content (WC) was measured using the standard oven-drying method at 105°C for 8 h. Bulk density (BD) was determined by the core method.

Water-filled pore space (WFPS) (%) was calculated based on the equation:

$$WFPS = (WC \times BD) \times \frac{100}{(1 - BD/2.65)}$$
(3)

where $2.65 \text{ (g cm}^{-3})$ refers to the assumed soil particle density.

Statistical analysis

All statistical analyses were conducted by SPSS v22.0 (IBM Corp., Armonk, USA) and the significance level for all statistical tests was set at P = 0.05. The differences in initial soil properties between different N-addition plots were examined using one-way analysis of variance (ANOVA) and least significant difference (LSD). Repeated-measures ANOVA was used to analyze the effects of N forms, N levels, experimental years, and their interactions on the temporal variation of soil N₂O emissions, annual cumulative N₂O emissions, ST, WFPS, and inorganic N concentrations. We examined the differences in annual N₂O emissions within each single year among the N additions by one-way ANOVA and LSD testing, and the differences within each N addition throughout the 5 years. Pearson's correlation analyses and linear regression analyses were used to examine the relationships between soil N₂O emissions and environmental variables. Means and standard deviations of N₂O emissions were calculated, and the plot values represented means (n = 3) ± standard error (SE).

Results

Soil N₂O emissions under N addition

During the 5-year experimental period, the temperate forest soil was a net source of N₂O. Soil N₂O emissions were higher between May and September, but the values were lower and leveled off in other times of each year. Meanwhile, the peak of soil N₂O emissions was concentrated in August of each year (Fig 2). Soil N₂O emissions were significantly influenced by N forms, N levels, and the sampling time (P < 0.01), but the interaction effect of N forms and levels, months and N levels or months, N forms and N levels, did not significantly influence the soil N₂O emissions (P > 0.05, Table 2).

Promotion effects of different N forms and levels

Different levels and forms of N addition and experimental time all significantly influenced the soil N₂O emissions (P < 0.01, Table 2). As for the two N-level addition treatments, the N-addition treatments significantly increased soil N₂O emissions, and this promotion effect was enhanced as the N-addition levels increased (Table 2, Fig 2). Soil N₂O emissions ranged from 1.30 µg m⁻² h⁻¹ to 34.44 µg m⁻² h⁻¹, with an average value of 11.55 µg m⁻² h⁻¹ in the control plots (Fig 2). Compared to the control, the average N₂O emissions in the low- and high-level N addition plots significantly increased by 186.02% and 353.98%, respectively. The maximal emissions were obtained in August 2013 for the low and high nitrogen addition serials, which were 163.23 and 276.33 µg m⁻² h⁻¹ in the L-NH₄NO₃ and H-NH₄NO₃ addition plots respectively, for all the three added nitrogen forms (Fig 2).

As for the N-addition treatments using the different forms of N, soil N₂O emissions were significantly increased by NH₄NO₃, (NH₄)₂SO₄, and NaNO₃ additions in the order of NH₄NO₃ > (NH₄)₂SO₄ > NaNO₃ > control for the same level of N addition (Fig 2, Table 2). Compared to the control, the average N₂O emissions in the NH₄NO₃, (NH₄)₂SO₄, and NaNO₃ addition plots significantly increased by 355.95%, 266.35%, and 187.71%, respectively (Fig 2).

There was no significant interaction between N form and N level on soil N₂O emissions (P > 0.05, Table 2).

Interannual soil cumulative N₂O emissions under N addition

Except for the interaction between N form and N level, year, N form and N level as well as all their interactions exerted significant effects on cumulative N₂O emissions (<u>Table 2</u>). In the



Fig 2. Variations of soil N₂O emissions applied with different forms and levels of N addition among five-year experimental period. L: 50 kg N ha⁻¹ yr⁻¹; H: 150 kg N ha⁻¹ yr⁻¹; H: 150 kg N ha⁻¹ yr⁻¹. Error bars indicate the standard error of the mean (n = 9).

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N	I₂O emissions		Cumulative N ₂ O emissions				
Subjects	d.f.	F	Subjects	d.f.	F		
Between subjects							
Form	2	8.367 **	Form	2	57.816 **		
Level	1	24.979 **	Level	1	172.416 **		
Form × Level	2	0.165 ns	Form × Level	2	1.138 ns		
Within subjects							
Month	59	28.630 **	Year	4	120.249 **		
Month × Form	118	3.774 **	Year × Form	8	19.198 **		
Month × Level	59	2.040 ns	Year × Level	4	10.061 **		
Month × Form × Level	118	0.624 ns	Year × Form × Level	8	2.571 *		

Table 2. Summary of repeated measures ANOVA results (*F* values) indicating the effects of different forms and levels of N addition and experimental time on temporal variation of soil N₂O emissions and annual cumulative N₂O emissions.

* *P* < 0.05

** P < 0.01 and ns P > 0.05.

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control plot, cumulative N₂O emissions showed no significant differences among the 5 years (Table 3). However, in the N addition plots, the promotion effect of additional N on soil N₂O emissions increased over time in the initial years, but then it decreased. As for the three N-form additions, ANOVA showed that the annual emissions were basically elevated by NH₄NO₃, (NH₄)₂SO₄, and NaNO₃ additions in the order of NH₄NO₃ > (NH₄)₂SO₄ > NaNO₃ > control (P < 0.05), but no significant differences were found between NaNO₃ and control plots in the first year (Table 3, P > 0.05).

Environmental variables and their correlation with N₂O emissions

During the 5-year period, air temperature had a clear seasonal pattern with higher temperatures in wet seasons (May to September) and lower in dry seasons (November to February). Soil temperature (ST) at the 5-cm depth fluctuated greatly, following changes in air temperature. The highest ST was 29.9°C and lowest was -7.2°C. WFPS ranged from 10.20% to 69.84% and fluctuated greatly (Fig 3). There were no significant differences among different N-addition plots on ST and WFPS (P > 0.05, Table 4).

Table 3. Cumulative N₂O emission (kg N ha⁻¹ yr⁻¹) from different N addition treatments plots.

Treatments		Cumula	tive N ₂ O emissions (kg N	ha ⁻¹ yr ⁻¹)	
	2011	2012	2013	2014	2015
Control	0.98±0.02 _{a(a)}	1.01±0.09 _{a(a)}	1.08±0.08 _{a(a)}	1.04±0.15 _{a(a)}	0.97±0.14 _{a(a)}
L-NH ₄ NO ₃	2.52±0.03 _{d(a)}	4.28±0.34 _{d(d)}	5.32±0.43 _{d(e)}	3.18±0.10 _{c(c)}	2.88±0.04 _{b(b)}
H-NH₄NO ₃	4.73±0.04 _{f(a)}	5.82±0.75 _{d(b)}	7.52±0.77 _{f(c)}	5.22±0.46 _{e(b)}	4.59±0.74 _{d(a)}
L-(NH ₄) ₂ SO ₄	1.69±0.15 _{b(a)}	3.16±0.16 _{c(b)}	3.84±0.54 _{c(c)}	2.75±0.36 _{bc(b)}	2.86±0.23 _{b(b)}
H-(NH ₄) ₂ SO ₄	3.41±0.00 _{e(a)}	5.25±0.24 _{e(c)}	6.21±0.25 _{e(d)}	4.13±0.48 _{d(b)}	3.71±0.28 _{c(b)}
L-NaNO ₃	1.15±0.18 _{a(a)}	1.89±0.20 _{b(b)}	2.57±0.23 _{b(c)}	2.65±0.25 _{b(c)}	2.63±0.32 _{b(c)}
H-NaNO ₃	2.31±0.17 _{c(a)}	3.30±0.05 _{c(b)}	4.97±0.73 _{cd(c)}	3.95±0.28 _{d(c)}	3.67±0.69 _{c(b)}

L: 50 kg N ha⁻¹ yr⁻¹; H: 150 kg N ha⁻¹ yr⁻¹.

Different superscripts of lowercase letters outside the parentheses indicate the significant differences at the level of P < 0.05 between the treatments within the same column, inside the parentheses indicate difference at the level of P < 0.05 between the experimental years within the same row.

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Fig 3. Water filled pore space (WFPS), soil temperature and air temperature in the observed period.

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Soil NH_4^+ -N and NO_3^- -N concentrations exhibited significant seasonal variation, with a single peak value in the N addition plots. The maximum value appeared between June and August, while the minimum was observed from November to March (Fig 4, Fig 5).

Soil NO₃⁻–N accumulated significantly in N addition plots and its concentration ranged from 15.24 to 53.33 mg kg⁻¹. N level had a significant promotion effect on soil NO₃⁻–N concentrations, with those under the and high level N addition had a significantly greater promotion on it compared with that of low level (P < 0.05, Table 4). The concentrations from lowand high-level N addition plots were, respectively, 142.14% and 172.90% greater than those from the control (12.03 mg kg⁻¹).

Soil NH_4^+ –N significantly accumulated in the N addition plots and its concentration ranged from 2.32 to 6.74 mg kg⁻¹. The accumulation of NH_4^+ –N caused by N addition was less than that of NO_3^- –N in soil. Soil NH_4^+ –N concentration was significantly influenced by the

Table 4. Summary of repeated measures ANOVA results (F values) indicating the effects of different forms and levels of N addition and experimen-
tal time on soil temperature at 5 cm soil depth (ST), water-filled pore space (WFPS), and the concentrations of soil inorganic N (NO ₃ ⁻ and NH ₄ ⁺).

	d.f.	ST		WFPS	WFPS		NO ₃		NH4 ⁺	
		F	P	F	P	F	P	F	P	
Between subjects										
Form	2	0.054	0.947	0.082	0.921	3.096	0.051	6.928	0.002	
Level	1	0.000	0.996	0.004	0.947	7.633	0.007	1.680	0.199	
Form×Level	2	0.020	0.981	0.011	0.989	0.008	0.992	0.273	0.762	
Within subjects										
Date	59	6.498	< 0.001	39.766	< 0001	10.772	< 0.001	4.253	0.002	
Date×Form	118	0.542	0.824	0.251	0.980	1.740	0.089	1.158	0.325	
Date×Level	59	0.422	0.793	0.249	0.910	1.289	0.274	0.630	0.641	
Date×Form×Level	118	0.358	0.942	0.609	0.770	0.916	0.504	0.259	0.978	

Significant effects (P < 0.05) are highlighted in bold.

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Fig 4. Variations of soil NH₄⁺–N concentrations applied with different forms and levels of N addition among five–year experimental period (a) NaNO₃ addition plots; (b) (NH₄)₂SO₄ addition plots; (c) NH₄NO₃ addition plots. L: 50 kg N ha⁻¹ yr⁻¹; H: 150 kg N ha⁻¹ yr⁻¹. Error bars indicate the standard error of the mean (n = 9).

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three N forms, which increased NH_4^+ –N in the order of $(NH_4)_2SO_4 > NH_4NO_3 > NaNO_3$ and by 57.40%, 36.27%, and 31.84% when compared with the control (2.98 mg kg⁻¹), respectively (Fig 4, Fig 5).



Fig 5. Variations of soil NO_3^-N concentrations applied with different forms and levels of N addition among five-year experimental period (a) NaNO₃ addition plots; (b) (NH₄)₂SO₄ addition plots; (c) NH₄NO₃ addition plots. L: 50 kg N ha⁻¹ yr⁻¹; H: 150 kg N ha⁻¹ yr⁻¹. Error bars indicate the standard error of the mean (n = 9).

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The correlation analysis showed that soil N₂O emissions were positively correlated with ST at 5 cm depth, WFPS at a 10-cm depth, and soil inorganic nitrogen concentration (Fig 6). In addition, a linear equation showed that soil N₂O emissions were extremely significantly (P < 0.01) correlated with ST, WFPS, and soil NH₄⁺-N and NO₃⁻-N (Fig 6).



Fig 6. Relationships between soil N₂O emissions and soil NH₄⁺–N concentration (a), soil NO₃⁻–N concentration (b), soil temperature (5 cm depth) (c), WFPS (10 cm depth) (d).

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Discussion

Promotion effects of N addition on soil N₂O emissions

Our results showed that the temperature plantation in northern China was a source of atmospheric N₂O under natural conditions. The mean N₂O emissions value in the control was 11.55 μ g N₂O–N m⁻² h⁻¹; this rate is comparable to that reported by Butterbach-Bahl et al. [16] who found that the N₂O emissions in soils of spruce forests in Germany and Ireland ranged from 3.5 to 16.4 μ g N m⁻² h⁻¹. In our study, NaNO₃, (NH₄)₂SO₄, and NH₄NO₃ addition at levels of 50 and 150 kg N ha⁻¹ yr⁻¹ significantly increased soil N₂O emissions by an average of 115.26% to 260.15%, 182.92% to 349.77%, 259.89% to 452.02%, respectively. The rate of increase was lower than that for a subtropical forest of the Qianyanzhou Ecological Station, where it was increased by 403% to 762% [5]. Except for L-NaNO₃ addition, the increase in soil N₂O emission was higher than the global average (134%) [18]. On one hand, these results indicate that the temperature plantation had high turnover rates of soil N and responded to the increased N deposition. On the other hand, to measure the peak N₂O emissions in our study, the gas samples were collected in the first, fourth and seventh days after N was added; hence, the cumulative N_2O emissions might have been overestimated since the N_2O emissions should have been measured weekly during growing season (to properly reflect an average impact over time). In our previous work, N addition significantly increased the amount of soil microbes and changed the soil microbial community structure in our study area [36]. Soil urease activities were significantly increased by N additions, which promoted soil N_2O emission [37]. Therefore, the weakened N limitation brought about by a higher litter decomposition rate and greater microbial activity could explain the increased N_2O emissions we found here [38].

N₂O emissions under different N addition forms and levels

Based on our observations over 5 years, the results supported our hypothesis that soil annual cumulative N_2O emissions increased under elevated N-addition levels. Positive correlations between N-addition levels and soil N_2O emissions have been found in many previous studies [4, 6, 39–40]. However, at our site, NaNO₃ addition at a rate of 50 kg N ha⁻¹ yr⁻¹ did not stimulate a significant increase in the cumulative N_2O emissions in the first year. Perhaps this is because of a threshold response of soil N_2O emissions to the N additions [4, 7, 41]. Specifically, such a response is determined by the competition between plants and soil microbes for available N, and thus emissions will not significantly increase until the plant N demands have been satisfied [4, 7, 42].

Considering the addition of different N forms, both NH₄⁺-N and NO₃⁻-N significantly promoted soil N₂O emission and exogenous NH_4^+ -N and NO_3^- -N inputs into our temperate forest had synergic effects on soil N₂O emission; this result supports our hypothesis and is also consistent with the finding elsewhere that exogenous NH₄⁺-N and NO₃⁻-N additions into boreal forest soil can have a synergic effect on its N₂O emissions [43]. The promotion of NH_4^+ – N (NH₄NO₃ and (NH₄)₂SO₄) additions for N₂O emission exceeded that provided by the NO_3^- -N addition. This result is consistent with other studies finding higher N_2O emissions from ammonium sources than from nitrate sources [7, 26]. Two potential mechanisms may be responsible for this phenomenon: (1) high immobilization of NO₃⁻-N and nitrification rates, coupled to a low denitrification potential, led more NO_3^--N to accumulate in soil [44]; (2) poor mobility of NH_4^+ created depletion zones around the plant roots, leaving more N input exposed to microorganisms in soils. However, most research to date suggests that denitrification is the main process driving N_2O production [25, 45]. Yet when WFPS is in the range of 30 to 70%, nitrification can become the main process driving N_2O production, as denitrification rates increase rapidly when WFPS exceeds 60% [7]. WFPS in our research plots was at a low level for most of the 5-yr monitoring period, only exceeding 60% for a few months, which likely provided less than optimal conditions for the denitrification process [46]. Given this trend in WFPS, we indirectly conclude that NH_4^+ -N had higher conversion efficiency to N₂O than NO₃⁻-N at our forest site.

Although NH_4^+-N was always the major N form in local actual N deposition [47], since 1980 its NH_4^+-N / NO_3^--N ratio has decreased [48]. Considering the stronger promotion of N₂O emission by NH_4^+ , and the decreasing proportion of NH_4^+ in N deposition, we expect that the increased soil N₂O emission stimulated by N deposition at our site will not persist into the future.

Interannual soil N₂O emissions under N addition

Considering the time scale, we found a sharp increase in the annual N_2O emissions in the first three years, but after this point the rate of increase diminished. Soil reaches N saturation when the N input exceeds the N demanded by plants and microorganisms [49]. Early successional

forests are always defined as N-limited, because of the limited N availability for vigorous plant growth and the lack of N-fixing plants or bacteria, whereas mature tropical forests and oldgrowth subtropical forests are typically grouped as being N-saturated [50]. Being N-limited is relative to being N-rich, and this necessarily depends on the soil N availability and the response of vegetation to any N addition [50]. In our study area, N was clearly a limiting factor in the initial years based on amount and stimulating effect of N addition upon tree biomass. Continuing the N addition could shift the soil from being N-limited to N-rich, and then becoming N-saturated, such that soil N₂O emissions may appear to reach a steady state at high N levels [23]. In addition, Liu and Song [51] found that soil microbial activities may be limited by carbon availability when N is abundant. The suppression of soil N₂O emissions by longterm N additions was possibly due to a lack of readily available organic carbon [52] and/or adverse effects on mineralization of organic carbon under conditions of high N addition [53]. Therefore, our field experiment highlights the importance carrying out long-term studies to avoid possibly overestimating the N addition effects on N₂O emissions from short-term observations.

Relationships between soil N2O emissions and soil properties

In our study area, the soil concentration of NO_3^--N was higher than that of NH_4^+-N , and the accumulation of NO_3^--N caused by N addition was more than soil NH_4^+-N concentration. On the one hand, although the soil NO_3^--N concentration was directly increased by NO_3^--N addition, the NH_4^+-N addition could have enhanced the activity of soil nitrifiers and led to the NO_3^--N accumulation in soil we found. This finding and interpretation is consistent with some previous studies carried out in tropical and subtropical forests [54–55]. On the other hand, several studies using the ¹⁵N tracing method suggest that plants in temperate forest at our site preferred NH_4^+-N , which led to more NH_4^+-N becoming assimilated, such that the accumulation of NH_4^+-N in the soil was relatively little and brief [56].

We found that the soil N₂O emissions were significantly correlated with concentrations of soil NH₄⁺-N and NO₃⁻-N, suggesting soil N₂O emission was dominated by both nitrification and denitrification processes. Since atmospheric N deposition can significantly promote soil N₂O emission, and exogenous NH₄⁺-N and NO₃⁻-N inputs into temperate forests may have synergic effects on soil N2O emission, in the future both of these aspects ought to be distinguished in the dynamics of the N cycle and balance in terrestrial ecosystems by using ¹⁵N tracer methods. High ST, together with a relatively high WFPS, tend to promote both nitrification and denitrification processes [57] and consequently, high N₂O emissions, an interpretation that is consistent with many previous findings [58–59]. In particular, high WFPS may promote microbial movement and the expansion of the soil anaerobic microbial community [43]. Warm temperatures benefit soil nitrifying and denitrifying bacteria activities [11], which may explain the seasonal variation in the relatively high N₂O emissions that occurred from May to September that we observed in this study. Many other complex factors may have a played a role in determining our results, such as soil pH, soil C availability, and the microbial community structure, since they jointly influence the two key processes of nitrification and denitrification that are involved in soil N_2O production [36, 60].

Conclusions

This study emphasizes the effects of different N forms and levels on N_2O emissions from a temperate forest over 5-year experimental period. We found that the accumulation of soil NO_3^- -N was significantly higher than that of soil NH_4^+ -N due to N addition. N addition initially promoted soil N_2O emission yet this promoting effect, although it existed, weakened in

the following years. High level N addition had a stronger promotion effect upon soil N₂O emission than did the low level N addition. Meanwhile, the combined application of NH_4^+ -N and NO_3^- -N promotes N₂O emissions more than their single applications, and NH_4^+ -N addition had a stronger promotion effect for soil N₂O emission than did the NO_3^- -N addition. In addition, WFPS, ST, soil NH_4^+ -N, and NO_3^- -N were all positively related to the N₂O emissions. In the future, the long-term observation of soil N₂O emissions, and the measurement of microbial functional groups using ¹⁵N tracer methods, will be necessary to clarify the mechanisms responsible for the soil N₂O emissions.

Supporting information

S1 File. Data set underlying the findings. (XLSX)

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