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Crystal structure-guided design of berberine-based novel chitinase inhibitors

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ABSTRACT

Glycoside hydrolase family 18 (GH18) chitinases play an important role in various organisms ranging from bacteria to mammals. Chitinase inhibitors have potential applications as pesticides, fungicides, and antiasthmatics. Berberine, a plant-derived isoquinoline alkaloid, was previously reported to inhibit against various GH18 chitinases with only moderate K_i values ranging between 20 and 70 μ M. In this report, we present for the first time the berberine-complexed crystal structure of SmChiB, a model GH18 chitinase from the bacterium Serratia marcescens. Based on the berberine-binding mode, a hydrophobic cavity-based optimisation strategy was developed to increase their inhibitory activity. A series of berberine derivatives were designed and synthesised, and their inhibitory activities against GH18 chitinases were evaluated. The compound $\mathbf{4c}$ showed 80-fold-elevated inhibitory activity against SmChiB and the human chitinase hAMCase with K_i values at the sub-micromolar level. The mechanism of improved inhibitory activities was proposed. This work provides a new strategy for developing novel chitinase inhibitors.

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KEYWORDS

Berberine; chitinase; inhibitor; structural optimisation

1. Introduction

Glycoside hydrolase family 18 (GH18) chitinases (EC 3.2.1.14) catalyse the degradation of chitin, a homopolymer of β -(1,4)-linked N-acetylglucosamine, which play important roles in various life processes. For instance, the well-studied bacterium *Serratia marcescens* produces several GH18 chitinases to efficiently degrade chitin for nutrition¹⁻⁴. Chitinases from parasites causing nematodosis⁵ and malaria⁶ are important for the development and pathogenesis of these organisms. Insect chitinases are required to digest chitin for growth and development^{7,8}. Human chitinases have been reported to be associated with asthma⁹, allergic response¹⁰, and other immunological disorders^{11,12}. Because of the extensive roles of GH18 chitinases, inhibitors targeting these enzymes have potential applications as therapeutic agents and agrochemicals.

Over the past century, natural products have served as a source and inspiration for a large fraction of the commercial pharmaceuticals for humans, animals, and crops^{13–15}. The taxonomic, functional, and chemical diversities of natural products offer inherent advantages for driving pharmaceutical discovery¹³. Many natural products have been reported to inhibit GH18 chitinases, such as allosamidin¹⁶, argifin¹⁷, argadin¹⁸, psammaplin A¹⁹, styloguanidine²⁰, phlegmacin B₁²¹, cyclo-(L-Arg-D-Pro)²² and methylxanthines derivatives²³, and some of these compounds have shown practical applications.

Berberine is a plant derived isoquinoline alkaloid distributed widely in plants of the Berberidaceae, Ranunculaceae, and Papaveraceae families^{24,25}. It has been used for thousands of years

in traditional Chinese and Ayurvedic medicine for antimicrobial and antiprotozoal activities 24,25 . Several studies have revealed that berberine shows enormous potential in treating various diseases, including cancer, diabetes, depression, cardiovascular and hypertension $^{26-29}$. Moreover, berberine has also been reported to have potential applications in agriculture for its antifungal, insecticidal and herbicidal activities $^{30-32}$. Recently, it was reported to be a competitive inhibitor of GH18 chitinases, including those from the human (HsCht and hAMCase) and insect pest Ostrinia furnacalis (OfChtl). Berberine showed moderate inhibitory activity towards HsCht, hAMCase and OfChtl, with K_i values of 19, 65 and 23 μ M, respectively 33 . The modest inhibitory activity of berberine against chitinases remain unsatisfactory for meeting pharmaceutical needs.

In this study, we used *Sm*ChiB, a model GH18 chitinase from the bacterium *S. marcescens*, to improve inhibitory activity of berberine. Based on the solved crystal structure of *Sm*ChiB-berberine complex, a variety of berberine analogues with much improved inhibitory activities were designed and synthesised. This work provides a new perspective for exploiting berberine as GH18 chitinase inhibitors.

2. Materials and methods

2.1. Chemicals and instruments

The uncorrected melting point (mp) was obtained with a Büchi B540 apparatus (Büchi Labortechnik AG, Switzerland). ¹H NMR, ¹³C

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(b) Supplemental data for this article can be accessed <u>here.</u>

NMR and ¹⁹F NMR were recorded on a Bruker AM-400 (¹H at 400 MHz, 13 C at 100 MHz and 19 F NMR at 376 MHz) spectrometer at 25 °C with samples prepared in dimethylsulphoxide (DMSO) and with tetramethylsilane (TMS) used as the internal standard. High-resolution electrospray ionisation mass spectra (HR-ESI-MS) were collected in an XEVO G2 TOF mass spectrometer (Waters, Milford, MA). The chromatographic columns for protein purification were purchased from GE Life Sciences (Beijing, China). The BCA protein assay kit was purchased from TaKaRa (Dalian, China). The yeast strain Pichia pastoris GS115, and the expression vectors pPIC9 and pPIC9K were purchased from Invitrogen (Beijing, 4-methylumbelliferyl-β-D-N,N'-diacetylchitobiose β -(GlcNAc)₂) and berberine were purchased from Sigma (Shanghai, China). All other chemicals of the highest purity were purchased from commercial sources.

2.2. General procedure for the synthesis of the compounds

The synthetic route for compounds is shown in Scheme 1. For the synthesis of the precursor berberrubine (2), a suspension of 1 g berberine hydrochloride (1) and 25 ml dimethylformamide (DMF) were stirred at 100 W for 1 h, and the solvent was removed under vacuum. The residue was purified by column chromatography (silica gel, DCM:MeOH = 3:1) to obtain the desired product berberrubine (2). For the synthesis of the precursor 3c, a suspension of 141 mg (1.0 mmol) of 6-fluoronicotinic acid in 1 ml of oxalyl chloride was stirred for 30 min at room temperature. After evaporation of oxalyl chloride in vacuo, compound 3c was obtained and used for the next reaction without any purification.

For the synthesis of the target compounds **4a-4c**, compounds **3a-3c** were added into a magnetically stirred solution of berberrubine (2) (0.20 mol) with 40 ml acetonitrile and 4 ml pyridine and stirred for 1-2 h at room temperature. The reaction was monitored by thin-layer chromatography (TLC). The resulting solid was filtered at room temperature and recrystallised twice from methyl alcohol to give the refined product.

Preparation of compound 4a: berberrubine (2) was treated with methyl succinyl chloride according to the general procedure to give the desired compound 4a as a yellow solid, yield 54.3%; mp: 192 °C (decomp). 1 H NMR (400 MHz, DMSO-d₆) δ 9.97 (s, 1H), 9.06 (s, 1H), 8.28 (d, $J = 9.2 \,\text{Hz}$, 1H), 8.22 (d, $J = 9.2 \,\text{Hz}$, 1H), 7.81 (s, 1H), 7.10 (s, 1H), 6.18 (s, 2H), 4.96 (t, J = 6.0 Hz, 2H), 4.03 (s, 3H), 3.68 (s, 3H), 3.21 (t, $J = 6.0 \,\text{Hz}$, 2H), 3.18 (t, $J = 6.8 \,\text{Hz}$, 2H), 2.81 (t, $J = 6.8 \,\text{Hz}$, 2H). ¹³ C NMR (101 MHz, DMSO-d₆) δ 172.20, 169.84, 150.45, 149.97, 147.69, 144.30, 138.12, 133.22, 132.93, 130.81, 126.82, 126.00, 121.06, 120.59, 120.31, 108.41, 105.51, 102.11, 57.24, 55.31, 51.68, 28.61, 28.52, 26.14. HR-ESI-MS calculated for $C_{24}H_{22}NO_7^+$ [M – Cl] +: 436.1396, found: 436.1395.

Preparation of compound 4b: berberrubine (2) was treated with benzoyl chloride according to the general procedure to give the desired compound 4b as a yellow solid, yield 75.7%; mp: 214.5–216.2 °C. 1 H NMR (400 MHz, DMSO-d₆) δ 10.02 (s, 1H), 9.13 (s, 1H), 8.35 (d, $J = 9.2 \,\text{Hz}$, 1H), 8.31 - 8.26 (m, 3H), 7.89 - 7.82 (m, 2H), 7.71 (t, J = 7.8 Hz, 2H), 7.09 (s, 1H), 6.19 (s, 2H), 4.92 (t, J = 6.2 Hz, 2H), 4.03 (s, 3H), 3.20 (t, J = 6.2 Hz, 2H). ¹³C NMR (101 MHz, DMSO-d₆) δ 163.39, 150.39, 149.96, 147.68, 144.47, 138.12, 134.58, 133.53, 132.96, 130.84, 130.39, 129.08, 127.95, 126.96, 125.85, 121.20, 120.65, 120.32, 108.38, 105.55, 102.11, 57.27, 55.20, 26.12. HR-ESI-MS calculated for $C_{26}H_{20}NO_5^+$ [M – CI] +: 426.1341, found: 426.1342.

Preparation of compound 4c: berberrubine (2) was treated with 6-fluoronicotinoyl chloride according to the general procedure to give the desired compound 4c as a yellow solid, yield 63.7%; mp:216 °C (decomp). ¹H NMR (400 MHz, DMSO-d₆) δ 10.06 (s, 1H), 9.16 (s, 1H), 9.10 (s, 1H), 8.89 – 8.73 (m, 1H), 8.36 (d, $J = 9.0 \,\mathrm{Hz}$, 1H), 8.29 (d, $J = 9.0 \,\mathrm{Hz}$, 1H), 7.84 (s, 1H), 7.56 (d, J = 8.2 Hz, 1H), 7.10 (s, 1H), 6.19 (s, 2H), 4.90 (t, J = 4.1 Hz, 2H), 4.05 (s, 3H), 3.21 (t, J = 0.5 Hz, 2H). ¹³C NMR (101 MHz, DMSO-d₆) δ 166.30 (d, $J = 244.4 \,\text{Hz}$), 161.78, 151.38 (d, $J = 17.1 \,\text{Hz}$), 150.89, 150.53, 148.22, 144.98, 144.85 (d, J = 9.5 Hz), 138.79, 133.49, 133.31, 131.37, 127.79, 126.42, 123.59 (d, $J = 4.4 \,\mathrm{Hz}$), 121.50, 121.12, 120.80, 111.08 (d, $J = 37.9 \,\text{Hz}$), 108.91, 106.04, 102.63, 57.85, 55.80, 26.63. 19 F NMR (376 MHz, DMSO-d₆) δ -60.58 (d, J = 7.7 Hz, 1 F). HR-ESI-MS calculated for $C_{25}H_{18}FN_2O_5^+$ [M – CI] $^+$: 445.1200, found: 445.1198.

2.3. Enzyme preparation

ChiB from S. marcescens was expressed in Escherichia coli BL21 (DE3). The other GH18 chitinases including the catalytic domains of OfChtl from O. furnacalis, human HsCht and human hAMCase were expressed in P. pastoris GS115. All the proteins were purified by immobilised metal affinity chromatography (IMAC) as described previously³⁴. The purities of the target proteins were analysed by SDS-PAGE followed by Coomassie Brilliant Blue R-250 staining.

2.4. Inhibitory activity assay

The activity of GH18 chitinases were determined using MU-(GlcNAc)₂ as a substrate. The reaction mixtures used for determining the inhibitor activity of compounds consisted of $100 \,\mu\text{L}$ of 20 nM enzyme, 4 μ M MU-(GlcNAc)₂, 1 or 10 μ M inhibitors and 2% DMSO in the buffer (20 mM sodium phosphate, pH 6.0, for SmChiB, OfChtl and HsCht; 20 mM sodium citrate, pH 5.2, for hAMCase). The reaction in the absence of compounds was used as a positive control. After incubating at 30 °C for 25 min, an equal volume of 0.5 M sodium carbonate was added to the reaction

Scheme 1. Synthetic route for the preparation of compounds 4a-4c and precursors. Reagents and conditions: (i) microwave, DMF, stirred at 100 W for 1 h; (ii) acetonitrile, pyridine, stirred for 1-2 h, room temperature; (iii) oxalyl chloride, stirred for 30 min, room temperature.

mixtures to terminate the reaction, and the fluorescence produced by the released MU was quantified using a Varioskan Flash microplate reader (Thermo Fisher Scientific, Waltham, MA), with excitation and emission wavelengths of 360 and 450 nm, respectively. Experiments were performed in triplicate. For K_i value determination, three substrate concentrations (4, 8, and 12 μ M for SmChiB, HsCht and hAMCase, 1, 2, and 4 μ M for OfChtl) and varied inhibitor concentrations were used. The K_i values and types of inhibition were determined by linear fitting of the data in Dixon plots.

2.5. Fluorescence measurements

Tryptophan fluorescence (295 nm excitation) was measured at 25 °C from 300 to 400 nm with a Varioskan Flash microplate reader using excitation and emission band passes of 5 nm. Fluorescence quenching experiments were performed in a 200 μ L mixture containing $1\,\mu\mathrm{M}$ protein in the buffer (20 mM sodium phosphate, pH 6.0, for SmChiB, OfChtl and HsCht; 20 mM sodium citrate, pH 5.2, for hAMCase), and by the successive addition of aliquots of compounds stock solution. Since the crystal structure confirmed that berberine binding to SmChiB in a 1:1 ratio, the dissociation constant K_d were then determined using the modified Stern-Volmer equation^{35,36}.

$$F_0/(F_0-F) = 1/f_a + K_d/f_a[Q]$$

Where, F_0 and F are fluorescence intensities in the absence and presence of compounds, f_a is the fraction of the accessible fluorescence, [Q] is the concentration of the compounds and K_d is the dissociation equilibrium constant.

2.6. Protein crystallisation and structure determination

Crystallisation experiments were performed by the hanging drop-vapor diffusion method at 4°C. SmChiB was firstly desalted in the 20 mM Tris-HCl with 50 mM NaCl, pH 8.0, and concentrated to 10 mg/mL before crystallisation. The crystals of free SmChiB were obtained in 1.0-2.0 M ammonium sulphate, 10-20% glycerol, 100 mM HEPES, pH 7.0 and then soaked with berberine to a final concentration of 1 mM for 1 h to yield the SmChiB-berberine complex. The crystals were cryoprotected by immersion in mother liquor containing 25% glycerol and flashed-cooled liquid nitrogen.

X-ray diffraction data of the complex were collected at the National Centre for Protein Science, Shanghai (BL18U, Pilatus 3-6 M detector). The structure of SmChiB complexed with berberine was solved by molecular replacement with PHASER³⁷ using the structure of free SmChiB (PDB: le6n) as the search model. The PHENIX³⁸ was used for structure refinement. Coot³⁹ was used for manually building and extending the molecular models. PROCHECK⁴⁰ was used to check the stereochemical quality of the models. The coordinates of the SmChiB-berberine complex were deposited in the Protein Data Bank (PDB) as entries 7C34. All structure figures were generated using PyMOL (DeLano Scientific LLC).

2.7. Molecular docking

The PDB files of berberine and its analogs were prepared using PRODRG⁴¹. The crystal structures of *Sm*ChiB (PDB: 7C34), *Of*Chtl (PDB: 3WQW), HsCht (PDB: 1HKK) and hAMCase (PDB: 2YBT) were prepared by PyMOL as the templates for the molecular docking. MGLTools was used to generate the PDBQT files of the proteins and compounds. The active site boxes were set at $50 \times 50 \times 50$ \mathring{A}^3 , $70 \times 70 \times 60$ \mathring{A}^3 , $70 \times 70 \times 70$ \mathring{A}^3 and $60 \times 80 \times 60$ \mathring{A}^3 for SmChiB, HsCht, OfChtl and hAMCase using AutoGrid4⁴², respectively. Molecular Docking were performed by AutoDock4⁴² using the Lamarckian genetic algorithm with a population size of 100 individuals, 25000000 energy evaluations, and 27000 generations. Plausible docking models were selected from the abundant clusters [root-mean-square deviation (RMSD) = 2 Å] that had the lowest binding energies.

3. Results and discussion

3.1. The binding mode of berberine in SmChiB

Berberine was reported as a competitive inhibitor towards several GH18 chitinases including OfChtl, HsCht, and hAMCase³³. In this study, we found berberine was also a competitive inhibitor against SmChiB, a well-studied GH18 chitinase, with a K_i value of 11.79 μ M (Table 1, Figure S1). To improve inhibitory activity, the binding mode of berberine in the SmChiB active pocket was analysed. The crystal structure of the SmChiB-berberine complex was obtained by soaking and was resolved to a resolution of 1.94 Å. The statistics of data collection and structure refinement are shown in Table 2. The coordinates of the SmChiB-berberine complex have been deposited in the PDB under accession number 7C34.

The crystal complex structure clearly revealed the location of berberine in the active pocket of SmChiB (Figure 1(B)). Berberine occupied the active pocket across the sugar-binding subsites +1 and +2 which was characterised by the aromatic Trp⁹⁷ and Trp²²⁰, respectively. The nomenclature for sugar-binding subsites was proposed by Davies et al^{43} , where subsite + n represents the reducing end and subsite -n represents the non-reducing end. The conjugate tetracycle plane (Figure 1(A)) of berberine was sandwiched by Trp^{97} and Trp^{220} to form π - π stacking interactions within distance of 4.2 and 4.3 Å respectively. This observation was consistent with our previous work that berberine inhibited GH18 chitinases mainly via π - π stacking interactions between the tetracycle plane and the aromatic residues in the binding pockets³³. This finding inspired us to retain the conjugate plane of berberine in the next design of berberine-based inhibitors. Moreover, an unoccupied hydrophobic cavity formed by residues Phe¹⁹⁰, Phe¹⁹¹, Leu²¹⁶, Phe²³⁹, and Leu²⁶⁵ was identified (Figure 1(B)). This space provides a favourable opportunity for the optimisation of berberine at its 9-O-position or 10-O-position. As the 9-O-position had been reported to be readily modified^{44–46}, it might be a candidate site for derivation of berberine for improving inhibitory activity.

3.2. Design, synthesis and biological activity of berberine analogs

Modification of berberine was performed by substituting a variety of groups at the 9-O-position. Three hydrophobic and bulky

Table 1. Inhibitory activities and binding affinities of the compounds towards different GH18 chitinases.

	μΜ							
	<i>Sm</i> ChiB		<i>Of</i> Chtl		<i>H</i> sCht		hAMCase	
Compounds	K _i	K_{d}	K_{i}	K_{d}	Ki	K_{d}	<i>K</i> _i	K _d
Berberine	11.79	11.98	23 ³³	20.61	19 ³³	21.89	65 ³³	58.03
4a	0.68	1.12	6.43	_	8.23	_	7.43	_
4b	2.36	2.99	7.28	_	11.39	_	4.62	_
4c	0.15	0.19	3.03	2.58	0.35	0.34	0.80	1.12

substituents including a methyl succinyl group, benzoyl group and 6-fluoronicotinoyl group were selected. The addition of a carbonyl group, fluorine atom and nitrogen atom in the substituents were expected to form hydrogen bonds with surrounding residues in the hydrophobic cavity. The compounds 4a-4c were obtained according to the synthetic route outlined in Scheme 1. Briefly, the key intermediate berberrubine (2) was obtained via microwave reaction⁴⁷. Berberrubine was then treated with excess acyl chloride 3a-3c in anhydrous acetonitrile (pyridine as catalyst) to produce the crude target compounds 4a-4c. The crude compounds were filtered at room temperature and then recrystallized twice from methyl alcohol to give the refined product. The reagents and materials that were used in the syntheses are easily commercially available, and the synthetic route resulted in high atom economy with good utilisation of reactant atoms in the end products.

Table 2 Details of data collection and structure refinement

Table 2. Details of data collection and structure refinement.				
	SmChiB-Berberine			
Space group	P 2 ₁ 2 ₁ 2 ₁			
Unit-cell parameters				
a, b, c (Å)	56.09, 103.77, 186.51			
α , β , γ (°)	90, 90, 90			
Wavelength (Å)	0.9778			
Temperature (K)	100			
Resolution (Å)	29.78-1.94 (2.008-1.94)			
Unique reflections	81619 (7997)			
Observed reflections	991829			
R _{merge}	0.05 (0.152)			
Average multiplicity	7.0 (6.8)			
Mean I/σI	8.55 (2.1)			
Completeness (%)	99.87 (99.27)			
R/R _{free}	0.163/0.195			
Protein residues	992			
Water molecules	740			
Other atoms	84			
Root mean square deviations				
Bond lengths (Å)	0.012			
Bond angles (°)	1.38			
Wilson B factor (Å ²)	27.44			
Average B factor (Å ²)	32.24			
Protein atoms	31.30			
Water molecular	33.40			
Ligand molecules	80.50			
Ramachandran plot (%)				
Most favoured	98			
Additionally allowed	2			
Outliers	0			
PDB code	7C34			

The newly synthesised compounds 4a-4c were then evaluated for inhibition activities against SmChiB. The results revealed that all three compounds were competitive inhibitors against SmChiB and showed improved inhibitory activities when compared with that of berberine. The K_i values of compound **4a-4c** against SmChiB were 0.68, 2.36, and 0.15 μ M, respectively (Table 1, Figure S1).

Berberine have been reported to be a broad-spectrum inhibitor of several GH18 chitinases, and therefore the inhibitory activities of compounds 4a-4c against other GH18 chitinases, including OfChtl, HsCht, and hAMCase, were also evaluated. As shown in Table 1 and Figure S2, compounds 4a-4c exhibited various degree of improvement in inhibitory activities against OfChtl, HsCht, and hAMCase when compared to berberine. In particular, compound 4c showed the highest inhibitory activity against all tested GH18 chitinases. The K_i values of compound **4c** towards OfChtl, HsCht, and hAMCase were 3.03, 0.35, and 0.80 μ M, respectively.

3.3. Inhibition mechanism of the compounds

According to our optimisation strategy, the modification of 9-Oposition may improve the inhibitory activity of berberine by enhancing its binding affinities towards chitinases. To confirm this hypothesis, K_d values were determined using tryptophan fluorescence quenching experiments. As revealed in Table 1 and Figure S3, berberine binds to SmChiB with a K_d of 11.98 μ M. The K_d values of compounds 4a-4c against SmChiB were 1.12, 2.99 and 0.19 μ M, respectively, showing good agreement with K_i values (Table 1 and Figure S1). The K_d values of berberine towards OfChtl, HsCht and hAMCase were 20.61, 21.89 and 58.03 μ M, respectively, which were also consistent with the K_i values. Compared with berberine, compound 4c showed a much lower K_d against OfChtl, HsCht and hAMCase with values of 2.58, 0.34 and 1.12 μ M (Table 1, Figure S4). These results showed that modification at the 9-O-position increased binding affinity.

To further understand the inhibitory mechanism of compounds 4a-4c, the structure-based molecular docking was performed to reveal the details responsible for the enhanced binding affinity. The binding modes of 4a-4c against SmChiB were nearly identical to that of berberine. The tetracycle moiety of the compounds occupied the substrate-binding cleft from subsites +1 to +2 via π - π stacking interactions with Trp⁹⁷ and Trp²²⁰. The 9-*O*-position substituents of the berberine analogues inserted into the extended hydrophobic pocket (Figure 2). As a result, the berberine

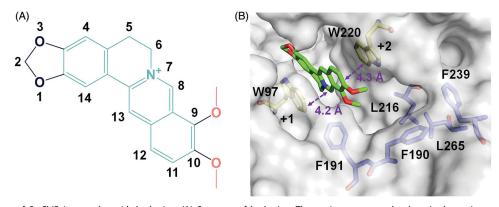


Figure 1. Crystal structure of SmChiB in complex with berberine. (A) Structure of berberine. The conjugate tetracycle plane is shown in green-cyan, the 9-O-methoxy and 10-O-methoxy moieties are shown in pink. (B) Binding mode of berberine in the active pocket of SmChiB. Berberine is shown in stick representation with carbon atoms in green. The aromatic residues that stack with berberine are labelled and shown in stick representation with carbon atoms in yellow. The amino residues formi ing the hydrophobic cavity extended near the ± 2 subsite are labelled and shown as stick representation with carbon atoms in blue. The numbers indicate the subsite to which the berberine is bound.

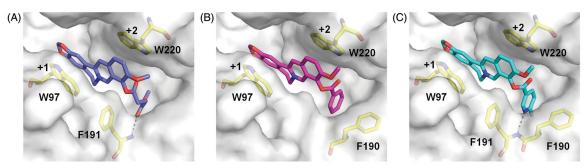


Figure 2. Modelled complex structures of compounds 4a-4c in SmChiB. Details of the interaction of compound 4a (A), 4b (B) and 4c (C) with SmChiB. Compound 4a is shown in blue, compound 4b is shown in magenta and compound 4c is shown in cyan. Residues that participate in binding are shown in yellow. Hydrogen bonds are shown as dashed black lines. The numbers indicate the subsite to which the compounds are bound.

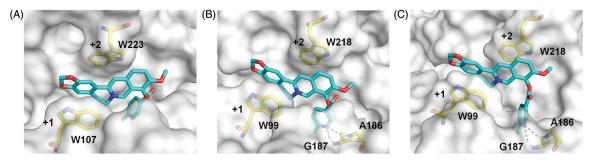


Figure 3. Modelled structures of compound 4c in OfChtl (A), HsCht (B), and hAMCase (C). Compound 4c is shown in stick representation with carbon atoms in cyan. Amino acids that interact with compound 4c are labelled and shown in stick representation with carbon atoms in yellow. Hydrogen bonds are shown as dashed black lines. The numbers indicate the subsite to which compound 4c is bound.

analogues were able to form additional interactions such as hydrogen bonds and stacking interactions with surrounding residues of SmChiB, which may account for their increased inhibitory activities. For example, compound 4a formed a hydrogen bond with Phe¹⁹¹ via the carbonyl group of the methyl succinyl substituent. Compound 4b formed a T-shaped stacking interaction with Phe¹⁹⁰ via its benzene ring. For compound **4c**, the 6-fluoronicotinoyl substituent group formed both hydrogen bond and Tshaped stacking interactions with Phe¹⁹¹ and Phe¹⁹⁰, respectively, which may account for its optimal inhibitory activity against SmChiB.

As compound 4c showed the highest inhibitory activity against all GH18 chitinases studied, its binding mechanism towards OfChtl, HsCht and hAMCase were also studied by molecular docking (Figure 3). The molecular docking results revealed that compound 4c inhibited these enzymes via a similar mechanism. Compound 4c bound in the active pocket across the +1 and +2subsites of the enzymes. The conjugate tetracycle plane formed π - π stacking interactions with conserved tryptophan residues (Trp¹⁰⁷ and Trp²²³ in OfChtl, Trp⁹⁹ and Trp²¹⁸ in HsCht and hAMCase). The 6-fluoronicotinoyl group inserted into the extended hydrophobic pockets of enzymes that results in hydrogen bonds formation between the nitrogen atom of the heterocyclic aromatic ring and the hydrogen atom of surrounding residues (Gly¹⁸⁷ and Ala¹⁸⁶ in *Hs*Cht and hAMCase). These hydrogen bonds appear to be important for its inhibitory activity, because the K_i values of compound 4c against HsCht and hAMCase were 54 and 81-fold higher than that of berberine, whereas the failure formation of hydrogen bond of compound 4c with OfChtl and the inhibitory activity only increased 7.5-fold higher than that of berberine (Table 1, Figure S2).

4. Conclusions

In the current study, we successfully developed a structure-optimisation strategy of berberine to improve the inhibitory activity against GH18 chitinases by taking advantage of an unoccupied hydrophobic cavity of the target enzyme. This work firstly reveals the binding mode of the natural product berberine in GH18 chitinase and provides a new path for exploring berberine-based molecules as promising GH18 chitinase inhibitors.

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Disclosure statement

All authors declare no competing financial interest.

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