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Introduction

Low-dimensional perovskite materials have considerable potential for applications in optical devices due to the advantages associated with the spatial confinement of the excitons. One way to achieve such confinement is to reduce the particle size to several nanometres, commonly referred to as quantum dots.1-3 The main preparation methods of perovskite quantum dots are hot injection or ligand-assisted anti-solvent precipitation.⁴⁻⁶ The band gap and optical properties of these quantum dots depend on their size. Perovskite quantum dots exhibit high photoluminescence (PL) quantum efficiency (up to $\sim 90\%$ for the red and >50% in the violet spectral regions), which is advantageous for potential luminescence applications such as thin film displays.^{7,8} However, in solid-state thin-films, the PL efficiency decreases dramatically owing to efficient nonradiative recombination channels introduced by particle aggregation and surface trap states. An alternative way to achieve quantum confinement in perovskite materials is to reduce the structural dimension.9-12 Zero-dimensional (0D) perovskites

Highly efficient radiative recombination in intrinsically zero-dimensional perovskite micro-crystals prepared by thermally-assisted solution-phase synthesis[†]

Wei-Long Xu,^{ab} Siobhan J. Bradley, ^[D]^a Yang Xu,^a Fei Zheng, ^[D]^a Christopher R. Hall, ^[D]^a Kenneth P. Ghiggino ^[D]^a and Trevor A. Smith ^[D]^{*a}

Zero-dimensional (0D) quantum confinement can be achieved in perovskite materials by the confinement of electron and hole states to single PbX_6^{4-} perovskite octahedra. In this work, 0D perovskite (Cs₄PbBr₆) micro-crystals were prepared by a simple thermally-assisted solution method and thoroughly characterized. The micro-crystals show a high level of crystallinity and a high photoluminescence quantum yield of 45%. The radiative recombination coefficient of the 0D perovskite micro-crystals, $1.5 \times 10^{-8} \text{ s}^{-1} \text{ cm}^3$, is two orders of magnitude higher than that of typical three-dimensional perovskite materials. Temperature dependent luminescence measurements provide insight into the role of thermally-activated trap states. Spatially resolved measurements on single 0D perovskite micro-crystals reveal uniform photoluminescence intensity and emission decay behaviour suggesting the solution-based fabrication method yields a high-quality and homogenous single-crystal material. Such uniform emission reflects the intrinsic 0D nature of the material, which may be beneficial to device applications.

can be achieved by introducing structural barriers between the octahedral units of PbBr₆. The 0D structure of these perovskites can extend over large (mm) areas, similar to bulk materials.¹⁰ Such structures have been shown to exhibit high PL efficiency (up to 45%).¹² However, the source of the luminescence in these materials is still hotly debated. It has been suggested the emission arises from a low density of CsPbBr₃ impurities, while others suggest that the luminescence originates from carriers confined to 0D unit-cells with confinement imposed by the intrinsic 0D structure.¹³⁻¹⁷

In this work, we have synthesized 0D perovskite microcrystals by a simple thermally-assisted solution synthesis method. The PL quantum efficiency of the 0D perovskite microcrystals reaches 45%. The radiative recombination coefficient of the 0D perovskite material is almost two orders of magnitude higher than that of the corresponding three-dimensional perovskite. The increased exciton-phonon interactions originating from strong lattice vibrations at high temperature make the PL spectrum broaden and blue shift, and the emission lifetime increase. The high PL intensity at low temperature is attributed to an increasing dominance of radiative recombination over non-radiative deactivation. Spatial and temporal PL mapping of single 0D perovskite microcrystals reveals this synthesis method yields high-quality microcrystals with uniform and intense emission from intrinsic zero-dimensional perovskites.

[&]quot;ARC Centre of Excellence in Exciton Science, School of Chemistry, The University of Melbourne, Parkville, Victoria 3010, Australia. E-mail: trevoras@unimelb.edu.au

^bSchool of Photoelectric Engineering, Changzhou Institute of Technology, Changzhou, Jiangsu 213002, China

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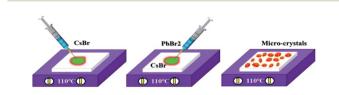
Results and discussion

The fabrication procedures of zero-dimensional perovskite micro-crystals are shown in Scheme 1. CsBr solution was added to a glass substrate on a hot plate at 110 $^{\circ}$ C. When the solution had evaporated to the saturated state, PbBr₂ solution was injected. The reaction was allowed to continue until the solvent had evaporated completely. Detailed information about the sample fabrication is provided in the ESI.[†]

Fig. 1a shows the crystalline structures for threedimensional (CsPbBr₃) and zero-dimensional (PbBr₆) perovskite materials. PbBr₆ comprises isolated octahedral units that do not share common halide atoms, serving to confine bandedge excitons within these small units. Fig. 1b shows results of X-ray diffraction (XRD) measurements, with the main specific crystallization peaks marked, revealing the rhombohedral phase of Cs₄PbBr₆ consistent with earlier reports (JCPDS no. 73-2478).^{15,18,19} The scanning electron microscopy (SEM) image (Fig. 1c) shows that the perovskite micro-crystals maintain a regular cuboidal structure over several micrometres. The energy-dispersive X-ray spectroscopy (EDS) mapping image (Fig. 1d) shows that Cs, Pb and Br are distributed in the perovskite crystals with a ratio of 23.8: 5.9: 35.2 (Fig. S1[†]), which is in good agreement with the stoichiometry of Cs₄PbBr₆. This result further verifies the perovskite micro-crystals are Cs₄PbBr₆. The high-resolution transmission electron microscopy (HRTEM) image (Fig. 1e) shows clear periodic lattice fringes. The measured *d*-spacing of 0.69 nm matches well with the (110) facets of Cs₄PbBr₆.^{14,17} The selected-area electron diffraction (SAED) image (Fig. 1f) verifies the single crystalline rhombohedral nature of the perovskite micro-crystals.

The photoluminescence spectra collected at room temperature are shown in Fig. S2.† The PL peak of the perovskite microcrystals is at 2.38 eV. Intensity dependent measurements reveal the PL intensity increases in a roughly logarithmic (negative exponential) manner (inset, Fig. S2†) over several hundreds of μ W in excitation power with no spectral shift. The PL quantum yield (QY) of the perovskite micro-crystals, measured in an integrating sphere, is 45%. The QY measurement results are shown in Fig. S3.†

Fig. 2 shows the time-resolved photoluminescence (TRPL) profiles of zero-dimensional perovskite micro-crystals at room temperature. The emission is very long-lived, spanning 6 μ s at low excitation powers. As the excitation power is increased from 0.1 nJ cm⁻² to 10 nJ cm⁻² the timescale decreases significantly, with the signal decaying within 2 μ s. The emission decays non-exponentially, as is usual for perovskite materials.^{12,17} In this



Scheme 1 The fabrication procedure of zero-dimensional perovskite micro-crystals.

work the emission decay profiles were analysed with a function comprising a simple sum of exponential terms:

$$I(t) = \sum_{i=1}^{n} A_i \mathrm{e}^{-t/\tau_i}$$
⁽¹⁾

The intensity weighted average lifetime²⁰ is calculated using the equation:

$$\tau_{\rm m} = \frac{\sum\limits_{i=1}^{N} A_i {\tau_i}^2}{\sum\limits_{i=1}^{N} A_i {\tau_i}}$$
(2)

where A_i is the amplitude coefficient for each decay time, τ_i .

The average emission decay time can be assumed to be proportional to the inverse of the sum of the rate constants for radiative, $k_{\rm r}$, and non-radiative, $k_{\rm nr}$, deactivation of the excited state:

$$\frac{1}{\tau_{\rm m}} = k_{\rm r} + k_{\rm nr} \tag{3}$$

The radiative relaxation rate is assumed to be constant for a given molecule, so the decreased average PL lifetime at higher excitation intensity may be related to an increase in multi-body interactions such as exciton–exciton annihilation.^{21,22} The high density of excitons generated at the higher excitation intensity can result in increased annihilation and non-radiative recombination rates. The relationship between PL lifetime and carrier density can be expressed as:^{23,24}

$$\tau_{\rm PL} = (A + B_{\rm rad} n_0)^{-1} \tag{4}$$

where *A* is the monomolecular trapping rate, representative of non-radiative recombination processes, and $B_{\rm rad}$ is the radiative recombination coefficient. The *A* (6.7 × 10⁶ s⁻¹) and $B_{\rm rad}$ (=(1.5 \pm 0.1) × 10⁻⁸ s⁻¹ cm³) values were calculated by fitting eqn (4) to the experimental data, as shown in the inset of Fig. 2. The radiative recombination coefficient *B* for zero-dimensional perovskite is significantly larger than that of comparable three-dimensional perovskite materials.^{23,24} Such a large coefficient is mainly attributed to the confinement of excitons within the isolated PbBr₆ units. At the same time, the monomolecular trapping rate of 0D perovskite is lower than that of three-dimensional perovskites.²³ These factors contribute to the high PLQY exhibited by 0D perovskite materials.

Temperature dependent PL spectra of perovskite microcrystals are shown in Fig. 3a. With increasing temperature, the PL peak position blue-shifts and its full width at half maximum (FWHM) gradually broadens (Fig. 3b). Meanwhile, the PL intensity decreases linearly, as shown in Fig. 3c. We describe the photo-physical processes within perovskite microcrystals at different temperatures with the aid of the schematics in Fig. 3d. Following photo-excitation, excitons relax to the ground state through radiative or non-radiative pathways. Phonon-assisted non-radiative recombination processes will be restricted at low temperature owing to weak lattice vibrations.

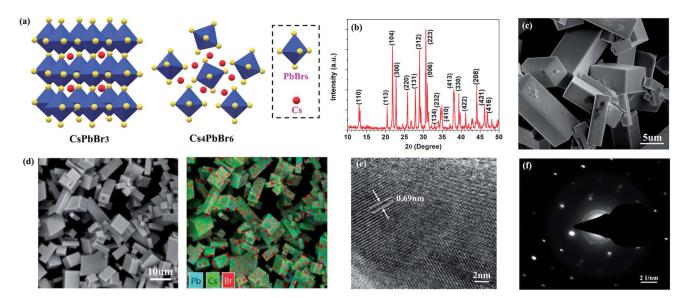


Fig. 1 (a) The crystalline structures of three-dimensional perovskite $CsPbBr_3$ and zero-dimensional perovskite Cs_4PbBr_6 . $PbBr_6$ units are isolated without sharing corners in Cs_4PbBr_6 ; (b) XRD; (c) SEM image; (d) EDS mapping image; (e) TEM image; and (f).

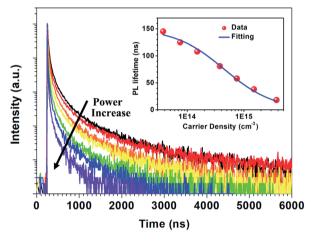


Fig. 2 PL decay dynamics under different excitation intensities. The inset shows the PL intensity average emission lifetime as a function of the excitation density for zero-dimensional perovskite micro-crystals.

The radiative recombination mechanism plays the dominant role, leading to the higher PL intensity, which is consistent with previous reports.^{11,25,26} As the temperature is increased, phonons are thermally activated and scatter the excitons, broadening the exciton linewidth. Both acoustic and optical photon scattering could cause the emission peak broadening.²⁹ The increase in temperature also leads to an expansion of the perovskite lattice, leading to a blue-shift in the exciton bandgap.^{27,28} Some defects exist in perovskite crystals that can capture excitons, leading to a decrease in PL intensity, which is more prevalent at higher temperatures. However, the linear variation of the FWHM with temperature indicates that the scattering from ionized impurities does not play a dominant role.²⁹⁻³¹

The temperature dependence of the TRPL of 0D perovskite micro-crystals is shown in Fig. 3e. Unusually, the timescale of the emission increases significantly with increasing temperature, whereas the PL intensity decreases with increasing temperature (Fig. 3c). Since $\phi_{\rm PL} = k_{\rm r}\tau$, these observations suggest that the radiative recombination rate, k_r , decreases while $k_{\rm nr}$ increases disproportionally, as the temperature increases. Fig. S4[†] shows that the TRPL is nearly independent of the excitation intensity at 77 K. This behaviour suggests that the excitons can radiatively recombine immediately without the impact of additional non-radiative recombination mechanisms such as exciton-exciton annihilation, even at high excitation intensity. The radiative recombination pathway plays the dominant role since the role of thermally activated trap-states is negligible and phonon scattering of excitons is weakened at low temperature, reducing the influence of exciton non-radiative recombination.25

Although the luminescence properties of zero-dimensional perovskites have been examined in several previous studies, the origin of the green emission from zero-dimensional Cs₄-PbBr₆ materials continues to be debated. Several papers suggest the green emission does not emanate from the zerodimensional material itself.³¹⁻³⁸ Reasons leading to this suggestion include; issues relating to the phase of the crystals (e.g. CsPbBr₃ \leftrightarrow Cs₄PbBr₆ phase transitions), trace-amounts of CsPbBr₃ as an impurity following synthesis, fast recombination of excitons and defect induced emission (such as Br vacancies or the presence of polybromides, self-trapped excitons, and interstitial hydroxyl groups). Recent reviews by Mohammed and co-workers^{38,39} discuss the various evidence, and the range of techniques used to gain it, for the possible sources of this green emission and suggests that it can be attributed to the existence of three different kinds of Cs₄PbBr₆ materials (pure, defective and hybrid).

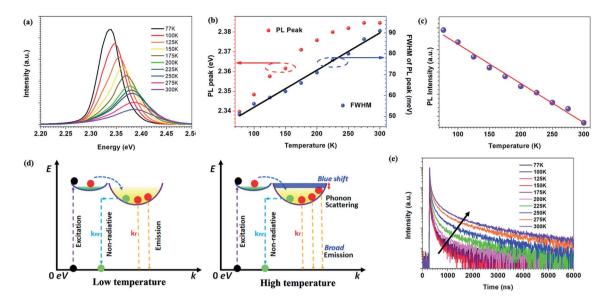


Fig. 3 (a) Temperature dependent PL spectra; (b) temperature dependent PL peak and FWHM; (c) the relationship between PL intensity and temperature; (d) photo-physical processes at low or high temperature; and (e) temperature dependent TRPL decay profiles of perovskite micro-crystals.

The majority of reported PL spectra of these materials are not spatially resolved. Similarly, the majority of the time-resolved measurements of these materials,11,12,40-45 as summarised by Akkerman et al.,32 report time-resolved PL data integrated over a whole sample, or in some cases from entire single crystals. Such large-area averaging would be insensitive to micrometre sized structural defects or states at crystal boundaries which may be present in these systems. Spatially and temporally resolved optical measurements can provide new information on the luminescence processes in micro-regions within these systems. We have performed time-resolved emission microscopy (TREM) measurements on the zero-dimensional perovskite micro-crystals under discussion. The confocal PL intensity image (Fig. 4a) demonstrates high crystalline uniformity within each crystal since the entire exposed area of the crystals is luminescent. This suggests that the emission originates from the zero-dimensional perovskite itself, rather than a small amount of localized CsPbBr3 impurities, at least on the spatial scales resolvable with current TREM instrumentation. The TREM image and the emission decay profiles corresponding to three typical regions in a single micro-crystal; the corner, the edge, and the body centre (marked A, B, and C, respectively), are shown in Fig. 4b and c. The decay profiles are far from exponential and were parameterized by fitting with sums of 3-4 exponential terms, in the absence of a better fit model. The TREM image (Fig. 4b) displays an amplitude average lifetime map calculated from a triple exponential model (limited by the SPCImage software used). The decay times, relative intensities, and amplitude- and intensity-average emission lifetimes, 20 τ_A and τ_{I} , respectively, of the three regions analysed as a sum of four exponential terms are listed in Table 1. The average PL lifetimes in the corner region (A) of this particular micro-crystal are slightly shorter than in the edge or body centre of the crystal (regions B and C), but this is more marked in the intensityaverage lifetimes. The variation in the average PL lifetimes from the corner to the rest of the crystal is accounted for by small contributions to the total PL of long-lived components found in the bulk of the micro-crystal (Table 1).

Our findings can be compared with previously reported TREM measurements of smaller ($\sim 1-2 \mu m$ diameter) phasepure zero-dimensional Cs₄PbBr₆ particles¹⁷ where the fluorescence was found to be emitted from the entire exposed area of the particles, concurring with our finding that the PL is not due

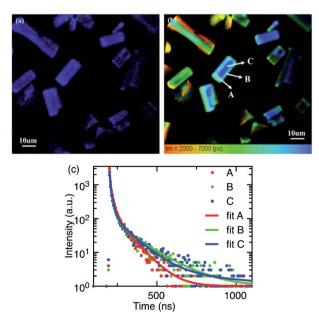


Fig. 4 (a) Confocal PL microscopy and (b) TREM images of perovskite micro-crystals (τ_m , amplitude-average lifetime, calculated from a triple exponential model); (c) PL decay profiles of a single perovskite micro-crystal recovered from the TREM image.

Table 1 PL decay times, relative pre-exponential factors, and amplitude- (τ_A) and intensity- (τ_I) average PL lifetimes resulting from four-exponential fitting of three typical regions in a single micro-crystal

Decay term	A (corner)	B (edge)	C (centre)
τ_1	3.6 ns (22%)	3.5 ns (26%)	3.9 ns (33%)
$ au_2$	4.3 ns (21%)	12.6 ns (63%)	14.3 ns (57%)
τ_3	17.4 ns (49%)	50.0 ns (6%)	96.9 ns (9%)
$ au_4$	83.1 ns (9%)	150 ns (5%)	400 ns (0.3%)
$\tau_{\rm A}$	17.4 ns	19.6 ns	19.7 ns
$\tau_{\rm I}$	43.3 ns	72.8 ns	74.6 ns

to small quantities of impurity localized in some specific places of the particles. Unlike the uniform PL intensity observed over our micro-crystals (Fig. 4a) the spatial distribution of PL intensity at the central region was found to be significantly higher relative to the edges. Amplitude-average PL lifetimes, very similar to those of the larger microcrystals discussed here, were reported despite the shorter timescale of the measurements, and similarly slightly longer-lived emission was observed towards the centre of the microdisks than the edges. A short decay component with a very similar lifetime to the present case (\sim 3.4 ns) was observed across the whole microdisk and attributed to trapped exciton recombination inside the crystal, while the long-lived emission (\sim 19 ns) was ascribed to surface energy state mediated recombination.

There is some indication of slightly longer-lived emission emanating from the centre of the crystals compared to the edges in some of the other microcrystals in the present samples (Fig. 4b), but in comparison with the microdisk work¹⁷, the steady-state and time-resolved emission microscopy behaviour of the large zero-dimensional Cs_4PbBr_6 micro-crystals under investigation here indicates that these crystals are highly uniform in composition and less susceptible to the effects of the surface trap states. The general uniformity of the time-resolved emission behaviour within these crystals suggests that any deep-trap-mediated non-radiative recombinational deactivation is not dominant in these large crystals and the green PL is ascribed to enhanced trap-assisted radiative recombination that is quite uniform over the entire crystal.

Conclusions

In summary, zero-dimensional Cs_4PbBr_6 micro-crystals of high crystalline quality were prepared by a thermally-assisted solution method and exhibited a high PL quantum yield of 45%. Spatial confinement of the exciton can be achieved within these zero-dimensional perovskite cells despite the overall size of the material spanning several micrometres. The radiative recombination coefficient of the 0D perovskite micro-crystals, $1.5 \times 10^{-8} \text{ s}^{-1} \text{ cm}^3$, is two orders of magnitude higher than that of the equivalent three-dimensional perovskite, which contributes to the high PL efficiency. At low temperatures, the radiative recombination rate is further increased due to the weak phonon scattering effect. We conclude that the PL properties of zero-dimensional Cs_4PbBr_6 micro-crystals originate from their

intrinsic structural characteristics rather than a small amount of CsPbBr₃ impurities. Our investigation provides additional insights into the photophysical dynamics of zero-dimensional perovskite micro-crystals which have potential applications in light-emitting devices and photonics.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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References

- 1 H. Wu, S. Wang, F. Cao, J. Zhou, Q. Wu, H. Wang, X. Li, L. Yin and X. Yang, *Chem. Mater.*, 2019, **31**, 1936–1940.
- 2 Y. Tong, E. Bladt, M. F. Aygüler, A. Manzi, K. Z. Milowska, V. A. Hintermayr, P. Docampo, S. Bals, A. S. Urban and L. Polavarapu, *Angew. Chem., Int. Ed.*, 2016, 55, 13887–13892.
- 3 A. Swarnkar, R. Chulliyil, V. K. Ravi, M. Irfanullah, A. Chowdhury and A. Nag, *Angew. Chem., Int. Ed.*, 2015, 54, 15424–15428.
- 4 J. Song, J. Li, L. Xu, J. Li, F. Zhang, B. Han, Q. Shan and H. Zeng, *Adv. Mater.*, 2018, **30**, 1800764.
- L. Protesescu, S. Yakunin, M. I. Bodnarchuk, F. Krieg, R. Caputo, C. H. Hendon, R. X. Yang, A. Walsh and M. V. Kovalenko, *Nano Lett.*, 2015, 15, 3692–3696.
- 6 B. A. Koscher, J. K. Swabeck, N. D. Bronstein and A. P. Alivisatos, J. Am. Chem. Soc., 2017, **139**, 6566–6569.
- 7 S. Zou, C. Liu, R. Li, F. Jiang, X. Chen, Y. Liu and M. Hong, *Adv. Mater.*, 2019, 1900606.
- 8 X. Zhang, X. Bai, H. Wu, X. Zhang, C. Sun, Y. Zhang,
 W. Zhang, W. Zheng, W. W. Yu and A. L. Rogach, *Angew. Chem., Int. Ed.*, 2018, 57, 3337–3342.
- 9 Z. Bao, H.-C. Wang, Z.-F. Jiang, R.-J. Chung and R.-S. Liu, *Inorg. Chem.*, 2018, **57**, 13071–13074.
- 10 X. Chen, F. Zhang, Y. Ge, L. Shi, S. Huang, J. Tang, Z. Lv, L. Zhang, B. Zou and H. Zhong, *Adv. Funct. Mater.*, 2018, 28, 1706567.
- 11 M. De Bastiani, I. Dursun, Y. Zhang, B. A. Alshankiti, X.-H. Miao, J. Yin, E. Yengel, E. Alarousu, B. Turedi, J. M. Almutlaq, M. I. Saidaminov, S. Mitra, I. Gereige, A. AlSaggaf, Y. Zhu, Y. Han, I. S. Roqan, J.-L. Bredas, O. F. Mohammed and O. M. Bakr, *Chem. Mater.*, 2017, 29, 7108–7113.
- M. I. Saidaminov, J. Almutlaq, S. Sarmah, I. Dursun,
 A. A. Zhumekenov, R. Begum, J. Pan, N. Cho,
 O. F. Mohammed and O. M. Bakr, *ACS Energy Lett.*, 2016, 1, 840–845.

- 13 Z. Zhang, Y. Zhu, W. Wang, W. Zheng, R. Lin, X. Li,
 H. Zhang, D. Zhong and F. Huang, *Cryst. Growth Des.*, 2018, 18, 6393–6398.
- 14 Y. Zhang, L. Sinatra, E. Alarousu, J. Yin, A. M. El-Zohry,
 O. M. Bakr and O. F. Mohammed, *J. Phys. Chem. C*, 2018,
 122, 6493–6498.
- 15 W. Zhai, J. Lin, Q. Li, K. Zheng, Y. Huang, Y. Yao, X. He, L. Li, C. Yu, C. Liu, Y. Fang, Z. Liu and C. Tang, *Chem. Mater.*, 2018, 30, 3714–3721.
- 16 S. Seth and A. Samanta, J. Phys. Chem. Lett., 2017, 9, 176–183.
- 17 S. Seth and A. Samanta, J. Phys. Chem. Lett., 2017, 8, 4461–4467.
- Y. Zhang, M. I. Saidaminov, I. Dursun, H. Yang, B. Murali,
 E. Alarousu, E. Yengel, B. A. Alshankiti, O. M. Bakr and
 O. F. Mohammed, *J. Phys. Chem. Lett.*, 2017, 8, 961–965.
- 19 P. Uthirakumar, J.-H. Yun, M. Devendiran, W. W. Lee and I.-H. Lee, *J. Lumin.*, 2019, **209**, 163–169.
- 20 A. Sillen and Y. Engelborghs, *Photochem. Photobiol.*, 1998, **67**, 475–486.
- 21 X. Wen, W. Chen, J. Yang, Q. Ou, T. Yang, C. Zhou, H. Lin, Z. Wang, Y. Zhang and G. Conibeer, ACS Appl. Mater. Interfaces, 2018, 10, 31586–31593.
- 22 T. Zhang, Z. Chen, Y. Shi and Q.-H. Xu, *Nanoscale*, 2019, **11**, 3186–3192.
- 23 V. D'Innocenzo, A. R. Srimath Kandada, M. De Bastiani, M. Gandini and A. Petrozza, J. Am. Chem. Soc., 2014, 136, 17730–17733.
- 24 Y. Yamada, T. Nakamura, M. Endo, A. Wakamiya and Y. Kanemitsu, *J. Am. Chem. Soc.*, 2014, **136**, 11610–11613.
- 25 S. D. Stranks, V. M. Burlakov, T. Leijtens, J. M. Ball, A. Goriely and H. J. Snaith, *Phys. Rev. Appl.*, 2014, 2, 034007.
- 26 K. Wu, A. Bera, C. Ma, Y. Du, Y. Yang, L. Li and T. Wu, *Phys. Chem. Chem. Phys.*, 2014, **16**, 22476–22481.
- 27 K. Thirumal, W. K. Chong, W. Xie, R. Ganguly, S. K. Muduli, M. Sherburne, M. Asta, S. Mhaisalkar, T. C. Sum and H. S. Soo, *Chem. Mater.*, 2017, 29, 3947–3953.
- 28 K. Miyata, T. L. Atallah and X.-Y. Zhu, Sci. Adv., 2017, 3, e1701469.
- 29 Z. Guo, X. Wu, T. Zhu, X. Zhu and L. Huang, *ACS Nano*, 2016, **10**, 9992–9998.

- 30 J. Yin, Y. Zhang, A. Bruno, C. Soci, O. M. Bakr, J.-L. Brédas and O. F. Mohammed, *ACS Energy Lett.*, 2017, **2**, 2805–2811.
- 31 A. D. Wright, C. Verdi, R. L. Milot, G. E. Eperon, M. A. Pérez-Osorio, H. J. Snaith, F. Giustino, M. B. Johnston and L. M. Herz, *Nat. Commun.*, 2016, 7, 11755.
- 32 Q. A. Akkerman, A. L. Abdelhady and L. Manna, J. Phys. Chem. Lett., 2018, 9, 2326–2337.
- 33 J. Bao and V. G. Hadjiev, Nano-Micro Lett., 2019, 11, 26.
- 34 D. Han, H. Shi, W. Ming, C. Zhou, B. Ma, B. Saparov, Y.-Z. Ma, S. Chen and M.-H. Du, *J. Mater. Chem. C*, 2018, 6, 6398–6405.
- 35 J. Yin, H. Yang, K. Song, A. M. El-Zohry, Y. Han, O. M. Bakr,
 J. L. Bredas and O. F. Mohammed, *J. Phys. Chem. Lett.*, 2018,
 9, 5490–5495.
- 36 X. Li, F. Cao, D. Yu, J. Chen, Z. Sun, Y. Shen, Y. Zhu, L. Wang,
 Y. Wei and Y. Wu, *Small*, 2017, 13, 1603996.
- 37 Q. A. Akkerman, S. Park, E. Radicchi, F. Nunzi, E. Mosconi,
 F. De Angelis, R. Brescia, P. Rastogi, M. Prato and
 L. Manna, *Nano Lett.*, 2017, 17, 1924–1930.
- 38 O. F. Mohammed, J. Phys. Chem. Lett., 2019, 10, 5886-5888.
- 39 L. Wang, H. Liu, Y. Zhang and O. F. Mohammed, *ACS Energy Lett.*, 2019, **5**, 87–99.
- 40 J. H. Cha, J. H. Han, W. Yin, C. Park, Y. Park, T. K. Ahn, J. H. Cho and D. Y. Jung, *J. Phys. Chem. Lett.*, 2017, 8, 565– 570.
- 41 D. Chen, Z. Wan, X. Chen, Y. Yuan and J. Zhong, *J. Mater. Chem. C*, 2016, 4, 10646–10653.
- 42 Y. Ling, L. Tan, X. Wang, Y. Zhou, Y. Xin, B. Ma, K. Hanson and H. Gao, *J. Phys. Chem. Lett.*, 2017, **8**, 3266–3271.
- 43 L. N. Quan, R. Quintero-Bermudez, O. Voznyy, G. Walters,
 A. Jain, J. Z. Fan, X. Zheng, Z. Yang and E. H. Sargent, *Adv. Mater.*, 2017, 29, 1605941–1605946.
- 44 J. Xu, W. Huang, P. Li, D. R. Onken, C. Dun, Y. Guo,
 K. B. Ucer, C. Lu, H. Wang, S. M. Geyer, R. T. Williams and
 D. L. Carroll, *Adv. Mater.*, 2017, 29, 1703701–1703710.
- 45 Z. Gan, F. Zheng, W. Mao, C. Zhou, W. Chen, U. Bach,
 P. Tapping, T. W. Kee, J. A. Davis, B. Jia and X. Wen, *Nanoscale*, 2019, 11, 14676–14683.