

(*E*)-6-(Furan-2-ylmethylidene)-6,7,8,9-tetrahydro-pyrido[2,1-*b*]quinazoline-11-thione

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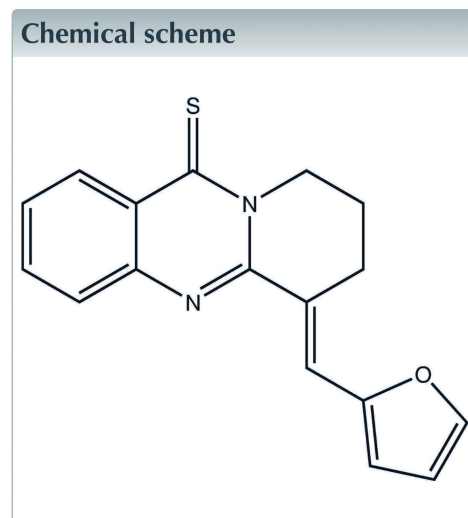
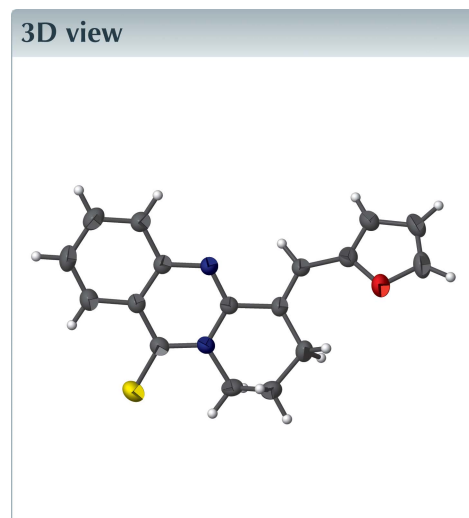
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Keywords: crystal structure; quinazolinthione; *E*-configuration.

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Structural data: full structural data are available from iucrdata.iucr.org

A quinazolinthione, C₁₇H₁₄N₂OS, was synthesized by the condensation reaction of 6,7,8,9-tetrahydro-11*H*-pyrido[2,1-*b*]quinazolin-11-thione with furfural. The molecule crystallizes in the monoclinic system (*Cc* space group) and has an *E* configuration with respect to the exocyclic C=C bond. In the crystal, molecules are linked through C—H··· π (furan) interactions, forming zigzag chains propagating along the [001] direction.



Structure description

Quinazoline derivatives are biologically active heterocyclic compounds (Shakhidoyatov, 1988; Elmuradov & Shakhidoyatov, 2006), used as drugs, such as cardiovascular agents (Volzhina & Yakhontov, 1982), herbicides (Chupp, 1974; Dayan, 2019), fungicides (Vicentini *et al.*, 2002; Sun *et al.*, 2011), *etc.* Among them, quinazoline and its homologues exhibit plural reactivity while maintaining several functional groups. The study of their reaction properties is of theoretical interest (Shakhidoyatov & Elmuradov, 2014). Alkylation and condensation reactions have been previously studied to produce tricyclic derivatives of quinazolinthione (Nasrullayev *et al.*, 2012; Nasrullaev *et al.*, 2015, 2016, 2017). In the present work, we report the crystal structure of a new quinazolinthione derivative.

The title compound (Fig. 1), consist of 6,7,8,9-tetrahydropyrido[2,1-*b*]quinazoline and furan-2-ylmethylene groups linked through the C6=C12 double bond [1.348 (5) Å]. The molecule adopts an *E* configuration relative to this bond. The quinazoline moiety is almost planar with an r.m.s. deviation of 0.0234 Å. Atoms C7 and C8 deviate from the plane through atoms C6, C5A, N10, C9 (r.m.s. deviation of 0.0053 Å) of the six-

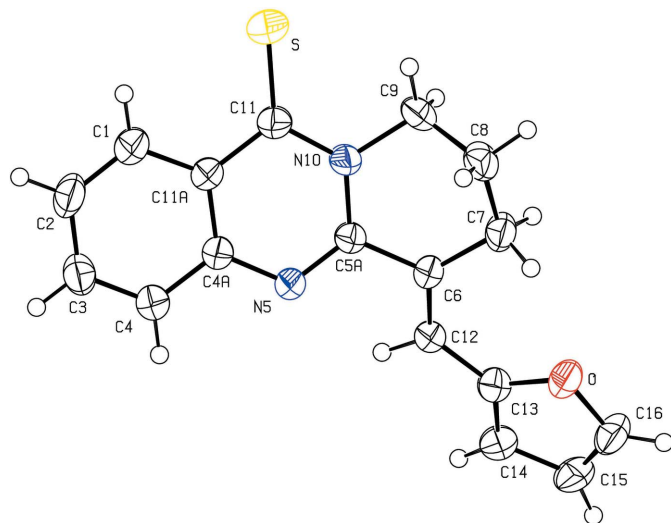


Figure 1
The molecular structure of title compound with the atom labelling. Displacement ellipsoids are drawn at the 50% probability level.

membered tetramethylene ring by 0.418 (8) and 0.912 (9) Å, respectively. These values are similar to those found for related compounds, for example 6,7,8,9-tetrahydro-11*H*-pyrido[2,1-*b*]quinazolin-11-thione and 6,7,8,9,10,12-hexahydroazepino[2,1-*b*]quinazolin-12-thione (Nasrullayev *et al.*, 2016).

In the crystal, molecules are linked by C—H... π (furan) interactions between molecules related by the *c* glide plane of

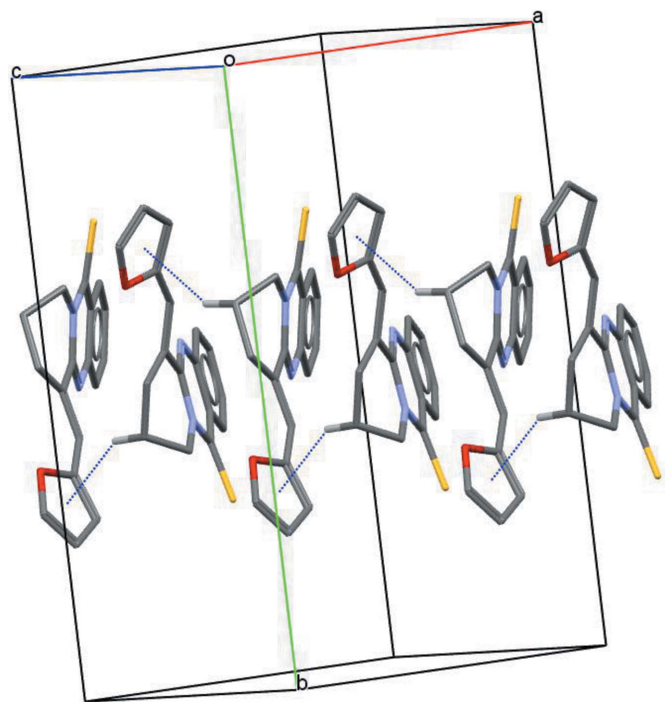


Figure 2
Chain of molecules of the title compound linked by C—H... π interactions. For clarity, H atoms not involved in these interactions have been omitted, and only atom H8B has been included.

Table 1
Hydrogen-bond geometry (Å, °).

*C*_g is the centroid of the furan ring (O/C2–C16).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C8—H8B... <i>C</i> _g ⁱ	0.97	2.76	3.596 (5)	145

Symmetry code: (i) *x*, −*y* + 1, *z* + ½

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₁₇ H ₁₄ N ₂ OS
<i>M</i> _r	294.36
Crystal system, space group	Monoclinic, <i>Cc</i>
Temperature (K)	295
<i>a</i> , <i>b</i> , <i>c</i> (Å)	9.4340 (19), 17.134 (4), 8.8260 (18)
β (°)	105.01 (4)
<i>V</i> (Å ³)	1378.0 (6)
<i>Z</i>	4
Radiation type	Cu <i>K</i> α
μ (mm ^{−1})	2.08
Crystal size (mm)	0.50 × 0.20 × 0.20
Data collection	
Diffractometer	Oxford Diffraction Xcalibur, Ruby
Absorption correction	Multi-scan (<i>CrysAlis PRO</i> ; Oxford Diffraction, 2009)
<i>T</i> _{min} , <i>T</i> _{max}	0.371, 1.000
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	2724, 1969, 1717
<i>R</i> _{int}	0.028
(<i>sin</i> θ / λ) _{max} (Å ^{−1})	0.628
Refinement	
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.045, 0.123, 1.04
No. of reflections	1969
No. of parameters	190
No. of restraints	2
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{max}$, $\Delta\rho_{min}$ (e Å ^{−3})	0.22, −0.25
Absolute structure	Flack <i>x</i> determined using 451 quotients [(<i>I</i> ⁺) − (<i>I</i> [−])]/[(<i>I</i> ⁺) + (<i>I</i> [−])] (Parsons <i>et al.</i> , 2013)
Absolute structure parameter	0.12 (3)

Computer programs: *CrysAlis PRO* (Oxford Diffraction, 2009), *SHELXS97* (Sheldrick, 2008), *SHELXL2014* (Sheldrick, 2015), *PLATON* (Spek, 2020) and *pubCIF* (Westrip, 2010).

space group *Cc*, forming zigzag chains propagating along the [001] direction (Table 1, Fig. 2).

Synthesis and crystallization

6,7,8,9-Tetrahydro-11*H*-pyrido[2,1-*b*]quinazolin-11-thione (1 mmol) was dissolved in 2–3 ml of glacial acetic acid and furfural (1 mmol) was added. The reaction mixture was refluxed for 5.5 h and cooled. Distilled water (10 ml) was added to the reaction mixture and the precipitate that formed was filtered off, washed with distilled water and dried. After recrystallization from cyclohexane solution, the title compound was recovered in good yield (68%), m.p. 170°C, *R*_f = 0.88. ¹H NMR, δ , p.p.m., *J* (Hz): 8.28 (1*H*, *d*, *J* = 8.2, H-1), 7.5 (1*H*, *t*, *J* = 8.2, H-2), 7.39 (1*H*, *d*, *J* = 1.7, H-5'), 7.30 (1*H*, *t*, *J* = 1.7, =CH), 7.23–7.29 (2*H*, *m*, H-3,4), 6.68 (1*H*, *d*, *J* = 3.4, H-3'),

6.3 (1H, dd, $J = 3.4, J = 1.7$, H-4'), 4.36 (2H, t, $J = 5.5$, δ -CH₂), 2.78 (2H, dt, $J = 6.8, J = 1.7$, β -CH₂), 1.85 (2H, m, γ -CH₂). IR spectrum: ν , cm⁻¹: 1569 (C=N), 1469 (C-N), 1272 (C=S). Light-orange prismatic single crystals suitable for X-ray diffraction analysis were obtained by were grown from acetone by slow evaporation of the solvent.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

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full crystallographic data

IUCrData (2020). 5 [https://doi.org/10.1107/S2414314620003569]

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Crystal data

C₁₇H₁₄N₂OS

M_r = 294.36

Monoclinic, *Cc*

a = 9.4340 (19) Å

b = 17.134 (4) Å

c = 8.8260 (18) Å

β = 105.01 (4)°

V = 1378.0 (6) Å³

Z = 4

F(000) = 616

D_x = 1.419 Mg m⁻³

Melting point: 443 K

Cu *K*α radiation, λ = 1.54184 Å

Cell parameters from 1371 reflections

θ = 5.2–75.6°

μ = 2.08 mm⁻¹

T = 295 K

Prism, light-orange

0.50 × 0.20 × 0.20 mm

Data collection

Oxford Diffraction Xcalibur, Ruby diffractometer

Radiation source: Enhance (Cu) X-ray Source

Detector resolution: 10.2576 pixels mm⁻¹

ω scans

Absorption correction: multi-scan

(CrysAlis PRO; Oxford Diffraction, 2009)

T_{min} = 0.371, *T_{max}* = 1.000

2724 measured reflections

1969 independent reflections

1717 reflections with *I* > 2σ(*I*)

R_{int} = 0.028

θ_{\max} = 75.7°, θ_{\min} = 5.2°

h = -11→11

k = -21→20

l = -10→8

Refinement

Refinement on *F*²

Least-squares matrix: full

R[*F*² > 2σ(*F*²)] = 0.045

wR(*F*²) = 0.123

S = 1.03

1969 reflections

190 parameters

2 restraints

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

w = 1/[σ²(*F_o*²) + (0.0857*P*)²]

where *P* = (*F_o*² + 2*F_c*²)/3

(Δ/σ)_{max} < 0.001

Δρ_{max} = 0.22 e Å⁻³

Δρ_{min} = -0.25 e Å⁻³

Absolute structure: Flack *x* determined using

451 quotients [(*I*⁺)-(*I*)]/[(*I*⁺)+(*I*)] (Parsons *et al.*, 2013)

Absolute structure parameter: 0.12 (3)

Special details

Refinement. All C-bound H atoms were positioned geometrically and refined using a riding model, with C—H = 0.93–0.97 Å and *U*_{iso}(H) = 1.2*U*_{eq}(carrier C).

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S	0.84058 (15)	0.72703 (6)	0.43163 (16)	0.0570 (4)
O	0.3377 (3)	0.35652 (18)	0.0790 (4)	0.0470 (7)
N5	0.7682 (4)	0.46996 (18)	0.4826 (4)	0.0375 (7)
C5A	0.6819 (4)	0.5126 (2)	0.3748 (4)	0.0335 (8)
N10	0.6960 (4)	0.59229 (18)	0.3632 (4)	0.0361 (7)
C11	0.8105 (4)	0.6337 (2)	0.4601 (5)	0.0371 (9)
C11A	0.9040 (4)	0.5873 (2)	0.5853 (5)	0.0348 (8)
C1	1.0187 (5)	0.6216 (2)	0.7004 (6)	0.0458 (10)
H1A	1.0354	0.6750	0.6982	0.055*
C2	1.1058 (5)	0.5765 (3)	0.8156 (6)	0.0523 (11)
H2A	1.1800	0.5998	0.8926	0.063*
C3	1.0842 (5)	0.4955 (3)	0.8188 (6)	0.0474 (10)
H3A	1.1453	0.4652	0.8963	0.057*
C4	0.9728 (4)	0.4610 (2)	0.7072 (5)	0.0417 (9)
H4A	0.9585	0.4073	0.7095	0.050*
C4A	0.8803 (4)	0.5065 (2)	0.5895 (5)	0.0352 (8)
C6	0.5610 (4)	0.4721 (2)	0.2591 (5)	0.0347 (8)
C7	0.4301 (5)	0.5176 (2)	0.1681 (5)	0.0429 (10)
H7A	0.3412	0.4939	0.1828	0.051*
H7B	0.4261	0.5153	0.0573	0.051*
C8	0.4371 (5)	0.6018 (2)	0.2199 (6)	0.0478 (11)
H8A	0.3663	0.6321	0.1432	0.057*
H8B	0.4115	0.6053	0.3193	0.057*
C9	0.5873 (5)	0.6349 (2)	0.2377 (6)	0.0493 (11)
H9A	0.6143	0.6303	0.1392	0.059*
H9B	0.5876	0.6898	0.2643	0.059*
C12	0.5812 (4)	0.3956 (2)	0.2360 (5)	0.0376 (9)
H12A	0.6717	0.3757	0.2907	0.045*
C13	0.4845 (5)	0.3402 (2)	0.1402 (5)	0.0391 (9)
C14	0.5114 (6)	0.2665 (2)	0.0973 (6)	0.0481 (11)
H14A	0.6018	0.2413	0.1226	0.058*
C15	0.3780 (6)	0.2352 (3)	0.0079 (6)	0.0539 (12)
H15A	0.3628	0.1856	-0.0360	0.065*
C16	0.2774 (6)	0.2914 (3)	-0.0014 (6)	0.0536 (12)
H16A	0.1791	0.2867	-0.0556	0.064*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S	0.0620 (7)	0.0336 (5)	0.0682 (8)	-0.0066 (5)	0.0039 (6)	0.0005 (5)
O	0.0388 (14)	0.0429 (16)	0.0531 (19)	-0.0076 (12)	0.0006 (14)	-0.0058 (14)
N5	0.0360 (15)	0.0295 (15)	0.0408 (19)	-0.0022 (13)	-0.0013 (14)	0.0014 (13)
C5A	0.0355 (18)	0.0289 (17)	0.033 (2)	0.0003 (15)	0.0039 (16)	0.0005 (15)
N10	0.0420 (17)	0.0263 (15)	0.0349 (18)	0.0003 (13)	0.0007 (15)	-0.0001 (13)
C11	0.038 (2)	0.0310 (17)	0.041 (2)	0.0002 (15)	0.0078 (17)	-0.0006 (15)

C11A	0.0349 (17)	0.0320 (17)	0.036 (2)	-0.0028 (15)	0.0058 (17)	-0.0019 (15)
C1	0.044 (2)	0.040 (2)	0.048 (3)	-0.0052 (17)	0.002 (2)	-0.0046 (18)
C2	0.041 (2)	0.060 (3)	0.048 (3)	-0.011 (2)	-0.004 (2)	-0.009 (2)
C3	0.038 (2)	0.056 (3)	0.043 (2)	0.0028 (18)	0.0003 (19)	0.005 (2)
C4	0.038 (2)	0.0384 (19)	0.044 (3)	-0.0003 (16)	0.0030 (19)	0.0025 (17)
C4A	0.0320 (17)	0.0336 (17)	0.037 (2)	-0.0013 (14)	0.0040 (17)	0.0006 (15)
C6	0.0319 (17)	0.0336 (18)	0.035 (2)	-0.0034 (15)	0.0017 (16)	-0.0014 (14)
C7	0.0366 (19)	0.041 (2)	0.044 (2)	-0.0005 (16)	-0.0024 (18)	-0.0082 (17)
C8	0.047 (2)	0.037 (2)	0.051 (3)	0.0117 (17)	-0.003 (2)	-0.0025 (18)
C9	0.057 (3)	0.035 (2)	0.046 (3)	0.0042 (19)	-0.005 (2)	0.0103 (18)
C12	0.0372 (18)	0.0307 (18)	0.040 (2)	-0.0014 (15)	0.0001 (17)	0.0028 (16)
C13	0.042 (2)	0.037 (2)	0.035 (2)	-0.0020 (16)	0.0048 (18)	-0.0004 (16)
C14	0.054 (2)	0.033 (2)	0.051 (3)	-0.0027 (18)	0.003 (2)	-0.0029 (18)
C15	0.067 (3)	0.035 (2)	0.054 (3)	-0.014 (2)	0.004 (2)	-0.0083 (19)
C16	0.050 (2)	0.053 (3)	0.051 (3)	-0.019 (2)	0.001 (2)	-0.010 (2)

Geometric parameters (Å, °)

S—C11	1.655 (4)	C4—H4A	0.9300
O—C16	1.365 (5)	C6—C12	1.348 (5)
O—C13	1.378 (5)	C6—C7	1.504 (5)
N5—C5A	1.304 (5)	C7—C8	1.510 (5)
N5—C4A	1.373 (5)	C7—H7A	0.9700
C5A—N10	1.379 (4)	C7—H7B	0.9700
C5A—C6	1.491 (5)	C8—C9	1.497 (7)
N10—C11	1.387 (5)	C8—H8A	0.9700
N10—C9	1.491 (5)	C8—H8B	0.9700
C11—C11A	1.458 (6)	C9—H9A	0.9700
C11A—C4A	1.405 (5)	C9—H9B	0.9700
C11A—C1	1.406 (6)	C12—C13	1.431 (6)
C1—C2	1.368 (7)	C12—H12A	0.9300
C1—H1A	0.9300	C13—C14	1.361 (6)
C2—C3	1.404 (7)	C14—C15	1.407 (7)
C2—H2A	0.9300	C14—H14A	0.9300
C3—C4	1.374 (6)	C15—C16	1.340 (7)
C3—H3A	0.9300	C15—H15A	0.9300
C4—C4A	1.407 (5)	C16—H16A	0.9300
C16—O—C13	106.2 (4)	C6—C7—H7A	109.3
C5A—N5—C4A	118.1 (3)	C8—C7—H7A	109.3
N5—C5A—N10	123.7 (3)	C6—C7—H7B	109.3
N5—C5A—C6	117.5 (3)	C8—C7—H7B	109.3
N10—C5A—C6	118.8 (3)	H7A—C7—H7B	108.0
C5A—N10—C11	122.4 (3)	C9—C8—C7	111.1 (4)
C5A—N10—C9	118.7 (3)	C9—C8—H8A	109.4
C11—N10—C9	118.9 (3)	C7—C8—H8A	109.4
N10—C11—C11A	114.2 (3)	C9—C8—H8B	109.4
N10—C11—S	122.5 (3)	C7—C8—H8B	109.4

C11A—C11—S	123.2 (3)	H8A—C8—H8B	108.0
C4A—C11A—C1	119.3 (4)	N10—C9—C8	110.0 (4)
C4A—C11A—C11	119.2 (3)	N10—C9—H9A	109.7
C1—C11A—C11	121.4 (4)	C8—C9—H9A	109.7
C2—C1—C11A	120.2 (4)	N10—C9—H9B	109.7
C2—C1—H1A	119.9	C8—C9—H9B	109.7
C11A—C1—H1A	119.9	H9A—C9—H9B	108.2
C1—C2—C3	120.7 (4)	C6—C12—C13	129.8 (4)
C1—C2—H2A	119.7	C6—C12—H12A	115.1
C3—C2—H2A	119.7	C13—C12—H12A	115.1
C4—C3—C2	120.0 (4)	C14—C13—O	108.7 (4)
C4—C3—H3A	120.0	C14—C13—C12	130.0 (4)
C2—C3—H3A	120.0	O—C13—C12	121.3 (4)
C3—C4—C4A	120.2 (4)	C13—C14—C15	107.8 (4)
C3—C4—H4A	119.9	C13—C14—H14A	126.1
C4A—C4—H4A	119.9	C15—C14—H14A	126.1
N5—C4A—C11A	122.1 (3)	C16—C15—C14	106.1 (4)
N5—C4A—C4	118.3 (3)	C16—C15—H15A	127.0
C11A—C4A—C4	119.6 (4)	C14—C15—H15A	127.0
C12—C6—C5A	116.3 (3)	C15—C16—O	111.2 (4)
C12—C6—C7	123.5 (3)	C15—C16—H16A	124.4
C5A—C6—C7	120.1 (3)	O—C16—H16A	124.4
C6—C7—C8	111.6 (3)		

*Hydrogen-bond geometry (Å, °)*C_g is the centroid of the furan ring (O/C2—C16).

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C8—H8B \cdots C _g ⁱ	0.97	2.76	3.596 (5)	145

Symmetry code: (i) *x*, $-\gamma+1$, *z*+1/2.

full crystallographic data

IUCrData (2020). 5 [https://doi.org/10.1107/S2414314620003569]

(*E*)-6-(Furan-2-ylmethylidene)-6,7,8,9-tetrahydropyrido[2,1-*b*]quinazoline-11-thione

Akmal Tojiboev, Azizbek Nasrullaev, Kambarali Turgunov, Burkhan Elmuradov and Bakhodir Tashkhodjaev

(*E*)-6-(Furan-2-ylmethylidene)-6,7,8,9-tetrahydropyrido[2,1-*b*]quinazoline-11-thione

Crystal data

$C_{17}H_{14}N_2OS$

$M_r = 294.36$

Monoclinic, *Cc*

$a = 9.4340$ (19) Å

$b = 17.134$ (4) Å

$c = 8.8260$ (18) Å

$\beta = 105.01$ (4)°

$V = 1378.0$ (6) Å³

$Z = 4$

$F(000) = 616$

$D_x = 1.419$ Mg m⁻³

Melting point: 443 K

Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å

Cell parameters from 1371 reflections

$\theta = 5.2$ – 75.6 °

$\mu = 2.08$ mm⁻¹

$T = 295$ K

Prism, light-orange

$0.50 \times 0.20 \times 0.20$ mm

Data collection

Oxford Diffraction Xcalibur, Ruby diffractometer

Radiation source: Enhance (Cu) X-ray Source

Detector resolution: 10.2576 pixels mm⁻¹

ω scans

Absorption correction: multi-scan

(CrysAlis PRO; Oxford Diffraction, 2009)

$T_{\min} = 0.371$, $T_{\max} = 1.000$

2724 measured reflections

1969 independent reflections

1717 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.028$

$\theta_{\max} = 75.7$ °, $\theta_{\min} = 5.2$ °

$h = -11 \rightarrow 11$

$k = -21 \rightarrow 20$

$l = -10 \rightarrow 8$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.045$

$wR(F^2) = 0.123$

$S = 1.03$

1969 reflections

190 parameters

2 restraints

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0857P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.22$ e Å⁻³

$\Delta\rho_{\min} = -0.25$ e Å⁻³

Absolute structure: Flack x determined using

451 quotients $[(I^-)-(I)]/[(I^+)+(I)]$ (Parsons *et al.*, 2013)

Absolute structure parameter: 0.12 (3)

Special details

Refinement. All C-bound H atoms were positioned geometrically and refined using a riding model, with C—H = 0.93–0.97 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier C})$.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S	0.84058 (15)	0.72703 (6)	0.43163 (16)	0.0570 (4)
O	0.3377 (3)	0.35652 (18)	0.0790 (4)	0.0470 (7)
N5	0.7682 (4)	0.46996 (18)	0.4826 (4)	0.0375 (7)
C5A	0.6819 (4)	0.5126 (2)	0.3748 (4)	0.0335 (8)
N10	0.6960 (4)	0.59229 (18)	0.3632 (4)	0.0361 (7)
C11	0.8105 (4)	0.6337 (2)	0.4601 (5)	0.0371 (9)
C11A	0.9040 (4)	0.5873 (2)	0.5853 (5)	0.0348 (8)
C1	1.0187 (5)	0.6216 (2)	0.7004 (6)	0.0458 (10)
H1A	1.0354	0.6750	0.6982	0.055*
C2	1.1058 (5)	0.5765 (3)	0.8156 (6)	0.0523 (11)
H2A	1.1800	0.5998	0.8926	0.063*
C3	1.0842 (5)	0.4955 (3)	0.8188 (6)	0.0474 (10)
H3A	1.1453	0.4652	0.8963	0.057*
C4	0.9728 (4)	0.4610 (2)	0.7072 (5)	0.0417 (9)
H4A	0.9585	0.4073	0.7095	0.050*
C4A	0.8803 (4)	0.5065 (2)	0.5895 (5)	0.0352 (8)
C6	0.5610 (4)	0.4721 (2)	0.2591 (5)	0.0347 (8)
C7	0.4301 (5)	0.5176 (2)	0.1681 (5)	0.0429 (10)
H7A	0.3412	0.4939	0.1828	0.051*
H7B	0.4261	0.5153	0.0573	0.051*
C8	0.4371 (5)	0.6018 (2)	0.2199 (6)	0.0478 (11)
H8A	0.3663	0.6321	0.1432	0.057*
H8B	0.4115	0.6053	0.3193	0.057*
C9	0.5873 (5)	0.6349 (2)	0.2377 (6)	0.0493 (11)
H9A	0.6143	0.6303	0.1392	0.059*
H9B	0.5876	0.6898	0.2643	0.059*
C12	0.5812 (4)	0.3956 (2)	0.2360 (5)	0.0376 (9)
H12A	0.6717	0.3757	0.2907	0.045*
C13	0.4845 (5)	0.3402 (2)	0.1402 (5)	0.0391 (9)
C14	0.5114 (6)	0.2665 (2)	0.0973 (6)	0.0481 (11)
H14A	0.6018	0.2413	0.1226	0.058*
C15	0.3780 (6)	0.2352 (3)	0.0079 (6)	0.0539 (12)
H15A	0.3628	0.1856	-0.0360	0.065*
C16	0.2774 (6)	0.2914 (3)	-0.0014 (6)	0.0536 (12)
H16A	0.1791	0.2867	-0.0556	0.064*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S	0.0620 (7)	0.0336 (5)	0.0682 (8)	-0.0066 (5)	0.0039 (6)	0.0005 (5)
O	0.0388 (14)	0.0429 (16)	0.0531 (19)	-0.0076 (12)	0.0006 (14)	-0.0058 (14)
N5	0.0360 (15)	0.0295 (15)	0.0408 (19)	-0.0022 (13)	-0.0013 (14)	0.0014 (13)
C5A	0.0355 (18)	0.0289 (17)	0.033 (2)	0.0003 (15)	0.0039 (16)	0.0005 (15)
N10	0.0420 (17)	0.0263 (15)	0.0349 (18)	0.0003 (13)	0.0007 (15)	-0.0001 (13)
C11	0.038 (2)	0.0310 (17)	0.041 (2)	0.0002 (15)	0.0078 (17)	-0.0006 (15)

C11A	0.0349 (17)	0.0320 (17)	0.036 (2)	-0.0028 (15)	0.0058 (17)	-0.0019 (15)
C1	0.044 (2)	0.040 (2)	0.048 (3)	-0.0052 (17)	0.002 (2)	-0.0046 (18)
C2	0.041 (2)	0.060 (3)	0.048 (3)	-0.011 (2)	-0.004 (2)	-0.009 (2)
C3	0.038 (2)	0.056 (3)	0.043 (2)	0.0028 (18)	0.0003 (19)	0.005 (2)
C4	0.038 (2)	0.0384 (19)	0.044 (3)	-0.0003 (16)	0.0030 (19)	0.0025 (17)
C4A	0.0320 (17)	0.0336 (17)	0.037 (2)	-0.0013 (14)	0.0040 (17)	0.0006 (15)
C6	0.0319 (17)	0.0336 (18)	0.035 (2)	-0.0034 (15)	0.0017 (16)	-0.0014 (14)
C7	0.0366 (19)	0.041 (2)	0.044 (2)	-0.0005 (16)	-0.0024 (18)	-0.0082 (17)
C8	0.047 (2)	0.037 (2)	0.051 (3)	0.0117 (17)	-0.003 (2)	-0.0025 (18)
C9	0.057 (3)	0.035 (2)	0.046 (3)	0.0042 (19)	-0.005 (2)	0.0103 (18)
C12	0.0372 (18)	0.0307 (18)	0.040 (2)	-0.0014 (15)	0.0001 (17)	0.0028 (16)
C13	0.042 (2)	0.037 (2)	0.035 (2)	-0.0020 (16)	0.0048 (18)	-0.0004 (16)
C14	0.054 (2)	0.033 (2)	0.051 (3)	-0.0027 (18)	0.003 (2)	-0.0029 (18)
C15	0.067 (3)	0.035 (2)	0.054 (3)	-0.014 (2)	0.004 (2)	-0.0083 (19)
C16	0.050 (2)	0.053 (3)	0.051 (3)	-0.019 (2)	0.001 (2)	-0.010 (2)

Geometric parameters (Å, °)

S—C11	1.655 (4)	C4—H4A	0.9300
O—C16	1.365 (5)	C6—C12	1.348 (5)
O—C13	1.378 (5)	C6—C7	1.504 (5)
N5—C5A	1.304 (5)	C7—C8	1.510 (5)
N5—C4A	1.373 (5)	C7—H7A	0.9700
C5A—N10	1.379 (4)	C7—H7B	0.9700
C5A—C6	1.491 (5)	C8—C9	1.497 (7)
N10—C11	1.387 (5)	C8—H8A	0.9700
N10—C9	1.491 (5)	C8—H8B	0.9700
C11—C11A	1.458 (6)	C9—H9A	0.9700
C11A—C4A	1.405 (5)	C9—H9B	0.9700
C11A—C1	1.406 (6)	C12—C13	1.431 (6)
C1—C2	1.368 (7)	C12—H12A	0.9300
C1—H1A	0.9300	C13—C14	1.361 (6)
C2—C3	1.404 (7)	C14—C15	1.407 (7)
C2—H2A	0.9300	C14—H14A	0.9300
C3—C4	1.374 (6)	C15—C16	1.340 (7)
C3—H3A	0.9300	C15—H15A	0.9300
C4—C4A	1.407 (5)	C16—H16A	0.9300
C16—O—C13	106.2 (4)	C6—C7—H7A	109.3
C5A—N5—C4A	118.1 (3)	C8—C7—H7A	109.3
N5—C5A—N10	123.7 (3)	C6—C7—H7B	109.3
N5—C5A—C6	117.5 (3)	C8—C7—H7B	109.3
N10—C5A—C6	118.8 (3)	H7A—C7—H7B	108.0
C5A—N10—C11	122.4 (3)	C9—C8—C7	111.1 (4)
C5A—N10—C9	118.7 (3)	C9—C8—H8A	109.4
C11—N10—C9	118.9 (3)	C7—C8—H8A	109.4
N10—C11—C11A	114.2 (3)	C9—C8—H8B	109.4
N10—C11—S	122.5 (3)	C7—C8—H8B	109.4

C11A—C11—S	123.2 (3)	H8A—C8—H8B	108.0
C4A—C11A—C1	119.3 (4)	N10—C9—C8	110.0 (4)
C4A—C11A—C11	119.2 (3)	N10—C9—H9A	109.7
C1—C11A—C11	121.4 (4)	C8—C9—H9A	109.7
C2—C1—C11A	120.2 (4)	N10—C9—H9B	109.7
C2—C1—H1A	119.9	C8—C9—H9B	109.7
C11A—C1—H1A	119.9	H9A—C9—H9B	108.2
C1—C2—C3	120.7 (4)	C6—C12—C13	129.8 (4)
C1—C2—H2A	119.7	C6—C12—H12A	115.1
C3—C2—H2A	119.7	C13—C12—H12A	115.1
C4—C3—C2	120.0 (4)	C14—C13—O	108.7 (4)
C4—C3—H3A	120.0	C14—C13—C12	130.0 (4)
C2—C3—H3A	120.0	O—C13—C12	121.3 (4)
C3—C4—C4A	120.2 (4)	C13—C14—C15	107.8 (4)
C3—C4—H4A	119.9	C13—C14—H14A	126.1
C4A—C4—H4A	119.9	C15—C14—H14A	126.1
N5—C4A—C11A	122.1 (3)	C16—C15—C14	106.1 (4)
N5—C4A—C4	118.3 (3)	C16—C15—H15A	127.0
C11A—C4A—C4	119.6 (4)	C14—C15—H15A	127.0
C12—C6—C5A	116.3 (3)	C15—C16—O	111.2 (4)
C12—C6—C7	123.5 (3)	C15—C16—H16A	124.4
C5A—C6—C7	120.1 (3)	O—C16—H16A	124.4
C6—C7—C8	111.6 (3)		

*Hydrogen-bond geometry (Å, °)*C_g is the centroid of the furan ring (O/C2—C16).

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C8—H8B \cdots C _g ⁱ	0.97	2.76	3.596 (5)	145

Symmetry code: (i) *x*, $-\gamma+1$, *z*+1/2.