# Protein Thermodynamics Can Be Predicted Directly from Biological Growth Rates



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# Abstract

Life on Earth is capable of growing from temperatures well below freezing to above the boiling point of water, with some organisms preferring cooler and others hotter conditions. The growth rate of each organism ultimately depends on its intracellular chemical reactions. Here we show that a thermodynamic model based on a single, rate-limiting, enzyme-catalysed reaction accurately describes population growth rates in 230 diverse strains of unicellular and multicellular organisms. Collectively these represent all three domains of life, ranging from psychrophilic to hyperthermophilic, and including the highest temperature so far observed for growth (122°C). The results provide credible estimates of thermodynamic properties of proteins and obtain, purely from organism intrinsic growth rate data, relationships between parameters previously identified experimentally, thus bridging a gap between biochemistry and whole organism biology. We find that growth rates of both unicellular and multicellular life forms can be described by the same temperature dependence model. The model results provide strong support for a single highly-conserved reaction present in the last universal common ancestor (LUCA). This is remarkable in that it means that the growth rate dependence on temperature of unicellular and multicellular life forms can be explained by the same model.

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## Introduction

Temperature governs the rate of chemical reactions including those enzymic processes controlling the development of life on Earth from individual cells to complex populations and spanning temperatures from well below freezing to above the boiling point of water [1]. The growth rates of unicellular and multicellular organisms depend on numerous processes and steps, but all are in principle limited by enzymic reactions [2]. This realization provides a link that bridges the gap between biochemistry and whole organism biology. By using the assumption of a single ratelimiting reaction step we show that we can describe the growth rate of diverse poikilothermic life forms. The temperaturedependent growth curves of poikilothermic organisms across their biokinetic ranges have a characteristic shape that may appear superficially to be U-shaped, but attentive examination shows them to be more complex. The history of previous approaches to describing these curves is extensive [3-6]. We use a model to describe the effect of temperature on biological systems that assumes a single, rate-limiting, enzyme-catalyzed reaction using an Arrhenius form that also allows for protein denaturation. The relative success of microbial strains within populations has been shown to be critically dependent on protein denaturation [7]. Earlier we presented such a model and fitted it to 95 strains of microbes [8]. In this work in addition to data on microorganisms, we also include data on the intrinsic growth rates for insects and acari obtained from life table analysis and find that these multicellular strains are also well described by the model. In total, we model 230 datasets (called strains herein) that cover a temperature range of 124°C. Notable amongst the modeled strains is the inclusion of hyperthermophiles active at the highest temperatures so far known for biological growth (121°C [9], 122°C [10]). The lowest temperature modeled was  $-2^{\circ}$ C, below which growth rates cannot be reliably compared due to ice formation and the zone of thermal arrest. In this paper we address biological implications and results arising from examination of much more extensive data than previously used [8] and by grouping strains by their thermal optima rather than by taxonomy.

In essence, we model the growth rates of strains by assuming each strain is rate-limited by a single common enzyme which becomes denatured both at sufficiently high and at sufficiently low temperatures. The model uses growth rate data directly rather than modeling protein function. The model structure and definitions of the parameters are described in detail in the Materials and Methods. Briefly, we model the intrinsic growth rates for each strain (r) by using a function (equation 1) that describes a single, rate-limiting, enzyme-catalyzed reaction. The numerator of equation 1 has an Arrhenius form [11,12], and the denominator describes the temperature-dependent denaturation of that enzyme. It requires eight parameters, four of which are assumed common to all life and are therefore held fixed (viz. the change in enthalpy and entropy for protein unfolding  $\Delta H^*$ ,  $\Delta S^*$ , with associated convergence temperatures  $T_H^{\star}$ ,  $T_S^{\star}$ , respectively), and four additional parameters for each strain that are associated with a rate-limiting enzyme (viz. scaling constant c; enthalpy of activation  $\Delta H_A^{\ddagger}$ ; heat capacity change on denaturation  $\Delta C_P$ ; number of amino acid residues n). The model is fitted using a

Table 1. Posterior universal parameter estimates.		
Parameter	Mean	99% HPDI
Enthalpy change (J/mol amino acid residue), $\Delta H^{\star}$	4874	(4846, 4913)
Entropy change (J/K), $\Delta S^{\star}$	17.0	(16.9, 17.1)
Convergence temperature for enthalpy (K), $T_{H}^{\star}$	375.5	(375.1, 376.1)
Convergence temperature for entropy (K), $T_S^{\star}$	390.9	(390.3, 391.7)

Shown are the posterior means with 99% HPDI in parentheses. doi:10.1371/journal.pone.0096100.t001

Bayesian hierarchical modeling approach that allows all data to be simultaneously considered and estimates obtained in a single run.

## **Results and Discussion**

We examined several alternative model structures that classified strains either: I) with all strains in a single group; II) into taxonomically defined groups that correspond to the three domains of life [13]: Bacteria, Archaea, or Eukarya; III) taxonomically, but allowing for multicellularity: Bacteria, Archaea, unicellular Eukarya, or multicellular Eukarya; IV) into thermal groups: psychrophiles, mesophiles, thermophiles, or hyperthermophiles; V) into thermal groups, except for fungi: psychrophiles, mesophiles, fungal mesophiles, thermophiles, or hyperthermophiles. Using a Bayes factor [14] approach we determined that the best performing model grouped the strains by thermal group, except for fungi, which were put into a separate group (model V). This model performed better than model IV, which combined the unicellular mesophilic fungal (Ascomycota) strains with the multicellular mesophilic taxa that included insects and acari.

Parameter estimates for the universal and thermal group parameters are given in Tables 1 and 2, respectively. Detailed parameter estimates for all strains are given in Table S1. The estimates obtained here extend those provided by earlier analyzes [8] in their breadth and especially in their improved precision due to the much larger data set. In particular, the two convergence temperatures (universal parameters) are now estimated to within 1.0 and 1.4 degrees, respectively.

#### Model fit

The fits for all 230 strains are shown in Figure 1–7 and are excellent for almost all strains even including those with few data,

and across the large temperature range spanned by the data sets. For example, strains 12 and 13 grew at temperatures as low as 280K while strains 17 and 18 grew at temperatures in excess of 390K.

## Thermodynamic relationships

The probability of the native (catalytically active) state for the thermal groups is shown in Figure 8A; we refer to the latter as native state curves [15] since they represent the proportion of the rate-controlling enzyme that is in the native conformation. The curves for the probability of the native state have lower peaks for psychrophiles, mesophiles, and Ascomycota, and the curves are taller and progressively flattened for thermophiles and hyperthermophiles. The higher and flatter peaks for the thermophiles and hyperthermophiles suggests protein stability over an increasingly extended temperature range. The lower peak levels for the lower temperature groups might be interpreted as reduced stability for psychrophile [16] and Ascomycota proteins [17]. The psychrophile native state curve is also shifted to the left of the other groups, which are all approximately aligned at the same lower temperature ( $\approx 275$ K). The deviation of the psychrophiles below the other groups suggests that a mechanistic difference has evolved separating psychrophiles from the other groups.

The native state peak of each curve occurs at  $T_{\rm mes}$  which is functionally dependent on  $\Delta C_P$  (Table 3). Also in Table 3,  $T_{\rm opt}$ , the temperature of maximal growth rate, tracks very closely the upper end of the native state curve so that the temperature difference between  $T_{\rm opt}$  and the upper temperature of 50% stability ( $T_U$ ) is very small for all groups, ranging from 2.5° for mesophiles to 4.2° for fungal mesophiles. In contrast is the difference between  $T_{\rm opt}$  and the lower limit of the native state ( $T_L$ )

Thermal group	$\Delta H_A^{\ddagger a}$		$\Delta C_P{}^{\boldsymbol{b}}$		n <sup>c</sup>	
Psychrophiles	48.6	(29.3, 59.8)	49.7	(46.5, 52.5)	388	(267, 531)
Mesophiles	75.3	(72.6, 79.1)	59.9	(59.6, 60.2)	422	(388, 457)
Ascomycota	39.7	(37.2, 42.0)	61.7	(61.5, 62.0)	340	(323, 356)
Thermophiles	71.3	(65.9, 77.6)	71.4	(70.0, 72.7)	180	(156, 205)
Hyperthermophiles	96.0	(79.7, 123.8)	96.9	(92.1, 102.8)	101	(66, 144)

Table 2. Posterior estimates of thermal group parameters.

<sup>a</sup>Enthalpy of activation (J/mol).

<sup>b</sup>Heat capacity change (J/K mol-amino acid-residue).

<sup>c</sup>Number of amino acid residues.

Shown are the posterior means with  $99\%\ \text{HPDI}$  in parentheses.

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Figure 1. Fitted curves for strains 1–36. doi:10.1371/journal.pone.0096100.g001

which increases from a modest  $23^{\circ}$ C for psychrophiles but reaches as high as  $83^{\circ}$ C for hyperthermophiles. Last, the difference of  $T_{opt} - T_{mes}$  is virtually a constant for psychrophiles, mesophiles, fungal mesophiles ( $10^{\circ}-11^{\circ}$ ), but dramatically increases for thermophiles ( $23^{\circ}$ ) and hyperthermophiles ( $44^{\circ}$ ; Figure 8B). These observations suggest that as the enzymes adapted to withstand higher and higher temperatures, their optimal thermal activity did not lag far behind, and they lost little of their ability to function at lower temperatures.

We show in Figure 9A that the enthalpy of activation  $(\Delta H_A^{\ddagger})$  and in Figure 9B the heat capacity change  $(\Delta C_P)$  both generally increase with optimal temperature  $(T_{opt})$ . We can consider  $\Delta H_A^{\ddagger}$  as relating to enzyme activity and  $\Delta C_P$  as relating to enzyme stability [18] as well as hydrophobicity of the putative rate-controlling enzyme [19]. The  $\Delta H_A^{\ddagger}$  is smallest for Ascomycota followed by an increasing trend: psychrophiles, mesophiles/thermophiles, and then hyperthermophiles. The Metabolic Theory of Ecology [20], which describes metabolism and other biological processes in terms of an Arrhenius temperature dependence, explicitly assumes a constant enthalpy of activation (where it is called 'activation energy'), although other work implies that it may not be invariant [21]. Our results indicate that for the majority of strains in our data, which are mesophiles and thermophiles, the enthalpy of activation is roughly constant with only a minimal increasing trend in these groups with increasing  $T_{\text{opt}}$ , but for a broader range of strains the spread in the enthalpy of activation is much larger.

In the case of the Ascomycota, all strains considered were mesophilic and were consistent with some [17,22–24], but not all [25], experimental data. As a check we calculated a separate analysis of data for another Ascomycota species. We fitted the thermodynamic model (equation 1) to growth rate data not used in the Bayesian model for the Ascomycota species *Aspergillus candidus* [26] using PROC NLIN from the SAS System, version 9.2. This was the same method used previously [15] and required several parameters to be held constant to achieve convergence. We fixed  $\Delta S^* = 17.0, T_H^* = 375.5, T_S^* = 390.9$  (these being the best estimates that we now have from the Bayesian runs). We obtained the following estimates for the remaining five free parameters: numerator constant c = 5.087, enthalpy of activation  $\Delta H_A^* = 28,627$ , unfolding heat capacity change  $\Delta C_P = 62.19$ ,



Figure 2. Fitted curves for strains 37–72. doi:10.1371/journal.pone.0096100.g002

enthalpy change at the convergence temperature  $\Delta H^* = 4,872$ and number of amino acid residues n = 617.6. We note that the enthalpy of activation is very low, even lower than the values we have been getting for yeasts. The enthalpy change at the convergence temperature (4,872) is very close to the mean value estimated from the Bayesian run for that parameter, viz. 4,874. The *n* value of 617.6 is higher than the mean value obtained for that parameter from the Bayesian run for psychrophiles (388) and for yeasts (340), but we expect the value to be higher at the low temperature adaptation end of the temperature scale than at the thermophilic end of the adaptation scale, and that is the case. The heat capacity change for folding/unfolding of 62.2 is very close to that obtained for yeasts in this study.

The fungal proteins associated with the particular strains used in the Bayesian model may have low enthalpies of activation and, due to an inherent instability of yeast prion-type proteins, like psychrophilic proteins, are assisted by chaperones and chaperonins. Interestingly, their instability led to some workers suggesting that they are potentiators and facilitators of evolution [27]. In the case of the psychrophiles and hyperthermophiles, the apparent deviation of enthalpy of activation ( $\Delta H_A^{\dagger}$ ; Figure 9A) below and above the mesophiles and thermophiles suggests the possibility that the rate-limiting reaction has been subject to adaptation for their respective environments.

In Figure 9C we predict that the number of amino acid residues (n) declines with the optimal temperature for growth  $(T_{opt})$ . A negative correlation of protein length and optimal growth temperature has been reported [28,29]. In Figure 9D the average number of non-polar residues per amino acid residue  $(N_{ch})$  is predicted by the model to increase with optimal temperature  $(T_{opt})$ , as has been observed experimentally for psychrophilic Archaea [16]. This is consistent with the observation that the more thermophilic proteins of Archaea have a greater hydrophobicity compared to mesophilic homologues [30,31].

As noted above, we observed a trend in increasing  $\Delta C_P$  from psychrophiles to mesophiles (including Ascomycota) to thermophiles to hyperthermophiles. Also, there appears to be a negative correlation between  $\Delta C_P$  (per amino acid residue) and n(Figure 9B, 9C)), illustrating that the relationship of these parameters can be complicated when examined with organismlevel data. In Figure 10 we show that  $\Delta C_P$  appears to decline as nincreases, but after partitioning the data into successive intervals of



Figure 3. Fitted curves for strains 73–108. doi:10.1371/journal.pone.0096100.g003

 $N_{ch}$  we see that within each interval they have a positive relationship. In Figure 10 we also show Graziano et al's predicted  $\Delta C_P = -46 +$ relationship [32] as a visual guide,  $30(1-1.54n^{-0.268})N_{ch}$ . The interpretation is that thermophilic proteins are more hydrophobic (larger  $N_{ch}$ ) and that as  $T_{opt}$ increases, the  $\Delta C_P$ , which is determined by the reorganization of water molecules around the polar and non-polar groups of the protein following denaturation, increases more rapidly as n, an index of the size of the protein, increases. This relationship is determined by the ratio of the buried and exposed surface of the proteins to avoid a close-packed core inaccessible to water molecules [32]. The total heat capacity change for the protein, given by  $n \times \Delta C_P$ , is shown in Figure 11 to decrease with  $T_{opt}$ . This is consistent with previously suggested mechanisms for stabilizing thermophilic proteins [33,34].

#### Stability-activity tradeoffs

Low temperature environments are constrained by low thermal energies and accordingly psychrophilic proteins have low enthalpies of activation, allowing biologically useful rates to be obtained at low temperatures. In the case of hyperthermophiles the environment is more demanding and therefore more stable proteins are predicted. These unfold more slowly [33] perhaps as a result of greater hydrophobicity [35] and an increased number of salt bridges [36], and also tend to be more highly expressed [37]. Many proteins also rely on assistance from molecular chaperones including the heat shock family of proteins, or the more complex structures known as chaperonins, to encourage correct protein folding and to rescue and repair misfolded proteins [38]. It is thought that proteins are maintained by evolution to be only as stable as needed for their environment [39,40], though their active centers are optimized to be maximally active at different temperatures [41].

Thermophilic proteins tend to be more stable against unfolding than their mesophilic equivalents [37]. Stability is achieved by an increase in enthalpic forces at higher temperatures while at lower temperatures proteins are more flexible becoming dependent on entropic forces [16,36,37]. At very low temperatures psychrophilic proteins are more flexible and less stable [18], also depending on chaperones, but to control cold denaturation [42]. It has been suggested that the balance of stability and activity arising from



Figure 4. Fitted curves for strains 109–144. doi:10.1371/journal.pone.0096100.g004

entropic and enthalpic forces is important for protein function [43], while in evolution, it is stability that is conserved [44]. Hyperthermophilic proteins are more slowly evolving than their mesophilic equivalents [31,37] presumably because mutations in thermophilic proteins would have more deleterious impacts [45] and would not be perpetuated.

Hyperthermophilic proteins can be less kinetically sensitive to temperature [46], an effect congruent to that described here. A notable example is serum albumin, which is promiscuously catalytic, stable up to 150°C, and is largely homologous within vertebrates [47]. In other words, the more robust enzymes in thermophiles and hyperthermophiles are stabilized over a broader temperature range than in mesophiles and psychrophiles. While we obtain this effect from modeling organism intrinsic growth data, it is found in protein denaturation curves of individual proteins. For example, denaturation curves of phosphoglycerate kinases from the thermophilic bacterium, *Thermus thermophilus*, have been found to be almost flat over a 60°C range whereas those from yeasts were strongly temperature-dependent [48]. The trimeric protein CutA1 from the hyperthermophile *Pyrococcus horikoshii* [49] is more stable at all temperatures above 0°C than its thermophilic and mesophilic equivalents. The CutA1 protein is universally distributed in bacteria, plants and animals [50]. We suggest that there may be many other hyperthermophilic proteins still to be found in organisms with lower temperature optima.

#### Unicellularity and multicellularity

The model fits unicellular specific growth rates [51] and intrinsic growth rates in the case of multicellular organisms derived by life table analysis [52]. The two rates are comparable since both describe the maximum growth rate after allowing for the mortality rate. We refer to them both as growth rates. A distinction between them is that the growth rate of multicellular organisms results from a more complicated sequence of events. However, the proportion of the time spent in particular developmental stages, such as pupa in insects and nymphs in mites, does not change with temperature since they depend equally on the temperature dependence of cell division [53]. In addition, within multicellular metazoan organisms there are control cells (thermosensory neurones) that are specialized in sensing heat shock and act to trigger an orchestrated hierarchical response to temperature change throughout the organism [54]. The remarkable implication of the excellent model



**Figure 5. Fitted curves for strains 145–180.** doi:10.1371/journal.pone.0096100.g005

fits is that the rate of biological growth at a given temperature, considered as a proportion of the maximum possible rate for a strain, whether in unicellular or multicellular organisms, is ultimately limited by the thermodynamics of enzyme reactions.

## The nature of the rate-limiting reaction

While the model performs excellently, both in terms of its general consistency with protein biochemistry and in the good fits obtained, some predictions do not fully agree with thermodynamic expectations and there exists the possibility that the underlying mechanism may be more complex than a single, rate-limiting, enzyme-catalyzed reaction. Nevertheless, the model underlines the importance of thermodynamics in biological processes especially those relating to the interaction between proteins and water molecules, which in turn may depend on the properties of water itself [55]. But if it does take the form of a single reaction then we can speculate on its nature. A mechanism by which cells control denaturation may be suggested by consideration of protein chaperones. Some examples are DnaK (Hsp70) and DnaJ (Hsp40) and the bacterial chaperonins GroEL and GroES [56]. Such systems act during *de novo* folding and to refold unfolded

substrate proteins [38]. They are triggered by the inflated exposure of hydrophobic groups in the unfolded proteins [38]. GroEL and GroES function together to create an Anfinsen hydrophilic cage containing charged residues that accumulate ordered water molecules, causing the substrate protein to bury its hydrophobic residues and refold into its native state [56,57]. The rate at which the GroEL and GroES function proceeds is controlled by ATP hydrolysis [58]. If heat shock proteins represent the rate-limiting step, the rate at which they function must be the critical factor. Those chaperones that are responsible for de novo folding and refolding are ATP-dependent [56]. Expression of important chaperones (GroEL, GroEL, GrpE, DnaK) seem to become silent as bacterial cells die from sudden thermal stress [59]. Therefore, we hypothesize that the rate-limiting may be linked to a process leading to or directly linked to protein folding. The modeled value of n varies 4-fold (Table 2) suggesting the reaction could take different forms in different strains linked to their temperature preference. Reactions potentially include a range of important enzymes either enacting or supporting protein folding with denaturation of the reaction leading to inhibition of the broad protein folding process. Possible examples include trigger factor



Figure 6. Fitted curves for strains 181–216. doi:10.1371/journal.pone.0096100.g006

[60], peptidylprolyl isomerases (the slow step in protein folding) [61], protein disaggregation [62] and maintenance of ATP availability to the folding system [63,64].

Notably, we find that the predicted temperature of maximum protein activity increases with optimal temperature but at a lesser rate (Table 3). The pattern implies that the range of thermal activity for the rate-controlling step in hyperthermophiles has a much larger potential range than in thermophiles, and these in turn larger than in mesophiles. We propose that the remarkable occurrence of thermophilic proteins such as serum albumin and CutA1 in non-thermophilic organisms may be examples of such a phenomenon. The model provides strong support for a single reaction system common to all life and, therefore, must have been strongly conserved since the time of the last universal common ancestor (LUCA). The question of a hyperthermophilic LUCA remains unresolved [65-70] and while we do not speculate on the LUCA's nature, the suggestion of a metabolic commonality in the form of a highly conserved rate-limiting reaction may prompt further considerations on this issue.

## Conclusions

- 1. Our focus has shifted away from domains, and towards thermal adaptation groups to which all life belongs, as it is adaptation to temperature, and not taxonomy, that is the factor of importance in explaining the variation among data sets.
- 2. Significantly, these results are obtained without any use of protein data, but only by growth rate data from unicellular and multicellular organisms, thereby bridging the gap between biochemistry and whole organism biology.
- 3. Using growth rate data that describe how quickly unicellular or multicellular populations grow under non-limiting conditions, we obtain estimates of thermodynamic parameters for protein denaturation consistent with the published literature on the physiology of organisms.
- 4. With this approach, we can now obtain relationships between these thermodynamic parameters that were previously identified from protein chemistry experiments.
- 5. As we now have a universal model that fits population growth data for organisms that can be prokaryotic or eukaryotic, as



**Figure 7. Fitted curves for strains 217–230.** doi:10.1371/journal.pone.0096100.g007

well as unicellular or multicellular, the organisms thermal adaptation position (i.e. whether it is a psychrophile, mesophile, thermophile or hyperthermophile) and, if a mesophile, whether it is single-celled or multi-celled, is sufficient to predict reliably its relative rate response to temperature.

- 6. We also advance the modeling approach by updating the universal parameters using adaptive direction sampling instead of Metropolis-coupled MCMC that we previously used [8], resulting in a greatly reduced run-time that will make further model development much more feasible.
- 7. We find it remarkable that unicellular and multicellular life forms that evolved over at least 3 billion years can be described by the same temperature dependence model.

# Methods

#### Data

The data summarized in Table S1 comprised 3,289 records of intrinsic growth rates (or rates of metabolism in some cases) of 230 strains from 31 Bacteria, 20 Archaea, and 77 Eukarya species. They covered a temperature range of 271.2-395.3K (-1.95-122.15°C). They included 10 psychrophiles (e.g. Gelidibacter sp.), 157 mesophiles (e.g. Escherichia coli), 43 mesophilic fungi (Ascomycota; e.g. Monascus ruber), 14 thermophiles (e.g. Acidianus brierleyi), and 6 hyperthermophiles (e.g. Methanopyrus kandleri). The thermal groups are defined below. Not all domains of life were represented in all thermal groups; Eukarya, in particular, is thought to have an upper limit of 60°C [71]. The organisms are very diverse and include acidophiles (e.g. Ferroplasma acidiphilum), halophiles (e.g. Haloarcula vallismortis), haloalkaliphiles (e.g. Natronococcus occultus), an alga (Chlorella pyrenoidosa), as well as multicellular organisms including insects (e.g. Clavigralla tomentosicollis), acari (e.g. Amblyseius womersleyi), and a collembola (Paronychiurus kimi).

## Model structure

Below, we refer to the observed growth rate as r and the modeled growth rate as F. The model shown in equation 1 below assumes that the growth rate is governed by a single, enzymecatalyzed reaction system that is limiting under all conditions. In the equation the quantity F is the predicted rate given the temperature and the values of the parameters. The numerator  $(T \exp(c - \Delta H_A^{\dagger}/RT))$  is essentially an Arrhenius model that describes the rate of the putative enzyme-catalyzed rate-controlling reaction (RCR) as a function of temperature while the denominator models the change in expected rate due to the effects of temperature on the conformation and, hence, catalytic activity of the putative enzyme catalyzing that reaction.



In equation 1: R is the gas constant (8.314 J/K mol); c is a scaling constant;  $\Delta H_A^{\ddagger}$  is the enthalpy of activation (J/mol); T is the temperature in degrees Kelvin;  $\Delta C_P$  is the heat capacity change (J/K mol-amino acid-residue) upon denaturation of the RCR; n is the number of amino acid residues;  $\Delta H^{\star}$  is the enthalpy change (J/mol amino acid residue) at  $T_H^{\star}$ , the convergence temperature for enthalpy (K) of protein unfolding;  $\Delta S^{\star}$  is the entropy change (J/K) at  $T_S^{\star}$ , the convergence temperature for entropy (K) of protein unfolding.

We derive several further quantities. One is the average number of non-polar hydrogen atoms per amino acid residue [32]:  $N_{ch} = (\Delta C_P + 46)/(30(1 - 1.54n^{-0.268}))$ . Another is  $T_{\text{mes}}$ , the temperature at which denaturation is minimized [15]. This temperature provides an index of temperature adaptation of the organism and was calculated as  $T_{\text{mes}} = T_H^* - \Delta H^* / \Delta C_P$ . Last, there is the optimal temperature for growth,  $T_{\text{opt}}$ , which was calculated numerically from the fitted growth rate curves.



**Figure 8.** Probability of native state curves and portions of strain fitted curves between  $T_{opt}$  and  $T_{mes}$ . **A**: probability of native state curves for the thermal groups showing the flat-topped curves for the hyperthermophiles, reduced peak maximas for psychrophiles and Ascomycota, common lower temperature limit for the mesophiles, thermophiles, and hyperthermophiles, and displacement of the psychrophile curve to lower temperatures than the other domains. **B**: the portions of the fitted growth curves for all strains between  $T_{opt}$  and  $T_{mes}$  showing a trend for a broader gap in the more thermophilic strains. The plot uses a logarithmic scale on the vertical axis and the reciprocal of the temperature on the bottom horizontal axis.

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We allowed four parameters to have values specific to each strain:  $(c,\Delta H_A^{\dagger},n,\Delta C_P)$ . We assumed the strain parameters to be Gaussian distributed with means specific to their grouping within the model. We constructed alternative groupings of the strain

parameters, which we labeled: I, II, III, IV, and V. For model I we only used a single group to which all the strains belonged. In model II we allocated the strains to one of the taxonomic domains Bacteria, Archaea, or Eukarya. Model III was the same as model

Thermal group	$N_{ch}{}^{a}$	$T_{\rm mes}{}^{\boldsymbol{b}}$	$T_{\rm opt}{}^{\boldsymbol{c}}$	$T_L{}^d$	$T_U^{e}$	$T_U - T_{\rm opt}$	$T_{\rm opt} - T_L$	$T_{\rm opt} - T_{\rm mes}$
Psychrophiles	4.64	277	288	265	291	2.7	23	11
Mesophiles	5.08	294	305	283	307	2.5	23	11
Ascomycota	5.3	296	306	283	310	4.2	23	10
Thermophiles	6.35	307	330	285	332	2.5	45	23
Hyperthermophiles	8.64	325	369	286	372	2.4	83	44

## Table 3. Means of derived parameters.

<sup>a</sup>Average number of non-polar hydrogen atoms per amino acid residue.

<sup>b</sup>Temperature at which denaturation is minimized (K).

<sup>c</sup>Temperature at which growth is maximized (K).

<sup>d</sup>The lower temperature at which the putative rate-controlling enzyme is 50% denatured (K).

<sup>e</sup>The upper temperature at which the putative rate-controlling enzyme is 50% denatured (K).

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**Figure 9. Relationships between thermodynamic parameters and**  $T_{opt}$ **. A**: enthalpy of activation  $(\Delta H_A^{\ddagger})$  versus  $T_{opt}$ **. B**: heat capacity change  $(\Delta C_P)$  versus  $T_{opt}$ **. C**: number of amino acid residues (*n*) versus  $T_{opt}$ **. D**: average number of non-polar hydrogen atoms per amino acid residue ( $N_{ch}$ ) versus  $T_{opt}$ . **d**: average number of non-polar hydrogen atoms per amino acid residue ( $N_{ch}$ ) versus  $T_{opt}$ .

II except that we split Eukarya into unicellular and multicellular groups. Model IV grouped strains according to the four thermal groups given below, but ignored the taxonomic domains to which the strains belonged. Allocation to the thermal group followed an initial model fit from which we obtained estimates of  $T_{opt}$ . The strains were then allocated into the thermal groups as follows: psychrophile:  $T_{opt} \leq 17^{\circ}C$ ; mesophile:  $17^{\circ}C < T_{opt} \leq 45^{\circ}C$ ; thermophile:  $45^{\circ}C < T_{opt} \leq 80^{\circ}C$ ; hyperthermophile:  $T_{opt} > 80^{\circ}C$ . Model V was the same as model IV but included an additional group for the Ascomycota since exploratory work indicated they may differ from the other groups.

The remaining parameters  $(\Delta H^{\star}, T_{H}^{\star}, \Delta S^{\star}, T_{S}^{\star})$  described protein thermal stability limits [72–74] and were not expected to depend on the individual biochemistry of each strain. Indeed, our earlier study [8] and exploratory work supported this conclusion. Accordingly, in the model structure, these values were assumed common to all strains. We refer to these as universal parameters. To control the variance homogeneity we worked on the square root scale [75–77]. We assumed that the square root of the observed growth rate had a Gaussian distribution with a mean given by the square root of the modeled value,  $\sqrt{F}$ , and with an unknown precision (reciprocal variance),  $\sqrt{r} \sim N(\sqrt{F}, \tau)$ .

The data were standardized for each strain by dividing by the maximum rate for each strain so that all the standardized rates were in the range [0,1]. This ensured that the rates were not size-dependent. A subsequent standardization was conducted following an initial model fit by dividing the observed data for each strain by the fitted maximum rate for that strain. These model-scaled data were then used in subsequent analyzes. This procedure meant that the influence of the *c* parameter was effectively removed from the model.

# Implementation

We used a Bayesian approach to allow for uncertainty in measurement and parameters to be incorporated in a natural way



**Figure 10. Relationship between thermodynamic parameter values**  $\Delta C_P$ , *n* and  $N_{ch}$ . Shown is  $\Delta C_P$  versus *n* for all strains after partitioning the data into intervals based on  $N_{ch}$ . Each resulting set is indicated by different symbols and color shading, and for each Graziano *et al*'s predicted relationship [32] is plotted with the mean  $N_{ch}$  as labeled. Also shown is the  $T_{mes}$  (on the right-hand axis) corresponding to the  $\Delta C_P$  on the left-hand axis. doi:10.1371/journal.pone.0096100.g010

through the appropriate prior specification. We assigned normal priors to the strain parameters in which the means were specific to the taxonomic group for models I, II, and III, or thermal group for models IV and V:  $\theta_j \sim N(\theta_{d(j)}, \exp(\tau_{\theta}))$ , in which d(j) is the taxonomic or thermal group for strain *j*. The exp  $(\tau_{\theta})$  is the strain precision and models the variation between the strain parameters about the  $\theta_{d(j)}$  parameters. The taxonomic and thermal group means and the  $\tau_{\theta}$  were assigned uniform priors with limits informed by the biochemistry literature with the exception of cwhich was assigned a vague prior. The universal, thermal group and taxonomic group parameters were each assigned a uniform prior with limits informed by the biochemistry literature. Finally, the observational precision was assigned a gamma distribution,  $\tau \sim \Gamma(0.001, 0.001)$ . Prior specifications are documented in Table 4. Inference was obtained in the form of posterior means and variances using Markov Chain Monte Carlo (MCMC)

simulation [78]. We chose to update the parameters of each strain as a block using Haario updates [79]. We also used Haario updates for each set of taxonomic or thermal group mean parameters and the strain parameter precisions. For the universal parameters we used adaptive direction sampling [80] combined with a low probability stepping-stone proposal [81]. This resulted in a much reduced run-time compared to previous work [8]. The models were run for 1,000,000 iterations and the last 50% of iterations retained for further analysis. We compared the models using Bayes factors [14] obtained using a pseudo-prior approach [82]. There was a clear separation between the five models with model V being preferred over the other four models with Bayes factors of: 1.0e9, 7.0e7, 9.1e2, and 9.9e4. We therefore continued only with model V. We summarized parameters using posterior means, standard deviations, and 99% highest posterior density intervals (HPDI). A 99% HPDI is the shortest interval that contains a parameter with 99% probability.



**Figure 11. Total heat capacity change versus**  $T_{opt}$ . Shown is the total heat capacity change  $(n \times \Delta C_P)$  versus  $T_{opt}$ . Colors and symbols are: psychrophiles: dark blue circles; Ascomycota: green diamonds; mesophiles: light blue squares; thermophiles: orange triangles; hyperthermophiles: red inverted triangles. ۰<u>011</u> ~~~~ d

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Table 4. Priors for model parameters.

Parameter (with supporting literature references)	Priors
Scaling constant	$\begin{array}{ll} c_{j} \sim & \mathrm{N}(c_{d(j)}, \exp(\tau_{c})) \\ c_{d} \sim & \mathrm{Unif}(-70, 70) \\ \tau_{c} \sim & \mathrm{Unif}(-6.4537, -0.9085) \end{array}$
Enthalpy of activation [4,83–92]	$ \begin{array}{ll} \Delta H_{A_j}^{\ddagger} \sim & \operatorname{N}\left(\Delta H_{A_d(j)}^{\ddagger} , \exp\left(\tau_{\Delta H_A^{\ddagger}}\right)\right) \\ \Delta H_{A_d}^{\ddagger} \sim & \operatorname{Unif}(0.01, 200000) \\ \tau_{\Delta H_A^{\ddagger}} \sim & \operatorname{Unif}(-21.9926, -16.4474) \end{array} $
Heat capacity change [93,94]	$\begin{array}{ll} \Delta C_{P_j} \sim & \mathrm{N}(\Delta C_{P_{d(j)}} \ , \exp(\tau_{\Delta C_P})) \\ \Delta C_{P_d} \sim & \mathrm{Unif}(37, 118) \\ \tau_{\Delta C_P} \sim & \mathrm{Unif}(-6.8289, -1.2837) \end{array}$
Number of amino acid residues [95,96]	$ \begin{array}{ll} n_{j} \sim & \mathrm{N}(n_{d(j)}, \exp(\tau_{n})) \\ n_{d} \sim & \mathrm{Unif}(1, 20000) \\ \tau_{n} \sim & \mathrm{Unif}(-13.3692, -7.8240) \end{array} $
Enthalpy change at convergence temperature [97]	$\Delta H^{\star} \sim \mathrm{Unif}(3000,7000)$
Entropy change at convergence temperature [97]	$\Delta S^{\star} \sim \text{Unif}(10,30)$
Convergence temperature for enthalpy [94,97,98]	$T_{H}^{\star} \sim \mathrm{Unif}(320,\!420)$
Convergence temperature for entropy [97]	$T_S^{\star} \sim \text{Unif}(320,420)$

Shown are the prior distributions which are either Gaussian or uniform distributions. The parameters of the Gaussian distributions are their means and precisions (reciprocal variances). Strain level parameters are subscripted by j, taxonomic or thermal group parameters by d, and membership of strain j in group d by d(j). doi:10.1371/journal.pone.0096100.t004

### **Supporting Information**

Table S1Posterior strain parameter estimates showingmeans and standard deviations in square brackets.(PDF)

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## References

- 1. Rothschild LJ, Mancinelli RL (2001) Life in extreme environments. Nature 409: 1092–1101.
- Stegelmann C, Andreasen A, Campbell CT (2009) Degree of rate control: How much the energies of intermediates and transition states control rates. J Am Chem Soc 131: 8077–8082.
- Briere JF, Pracros P, Le Roux AY, Pierre JS (1999) A novel rate model of temperature-dependent development for arthropods. Environ Entomol 28: 22– 29.
- Johnson FH, Lewin I (1946) The growth rate of *E. coli* in relation to temperature, quinine and coenzyme. J Cell Compar Physl 28: 47–75.
- Schoolfield RM, Sharpe PJH, Magnuson CE (1981) Non-linear regression of biological temperaturedependent rate models based on absolute reaction-rate theory. J Theor Biol 88: 719–731.
- Sharpe PJH, DeMichele DW (1977) Reaction kinetics of poikilotherm development. J Theor Biol 64: 649–670.
- Peña MI, Davlieva M, Bennett MR, Olson JS, Shamoo Y (2010) Evolutionary fates within a microbial population highlight an essential role for protein folding during natural selection. Mol Syst Biol 6: 387.
- Corkrey R, Olley J, Ratkowsky D, McMeekin T, Ross T (2012) Universality of thermodynamic constants governing biological growth rates. PLoS ONE 7: e32003.
- Kashefi K, Lovley DR (2003) Extending the upper temperature limit for life. Science 301: 934.
- Takai K, Nakamura K, Toki T, Tsunogai U, Miyazaki M, et al. (2008) Cell proliferation at 122°C and isotopically heavy CH4 production by a hyperthermophilic methanogen under high-pressure cultivation. Proc Natl Acad Sci USA 105: 10949–10954.
- Laidler KJ (1997) A brief history of enzyme kinetics. In: Cornish-Bowden A, editor, New Beer in an Old Bottle: Eduard Buchner and the Growth of Biochemical Knowledge, Valencia, Spain: Universitat de Valencia. pp. 127–133.
- Laidler KJ, Bunting PS (1973) The chemical kinetics of enzyme action. Oxford: Clarendon Press, second edition, 471 pp.
- Woese CR, Kandler O, Wheelis ML (1990) Towards a natural system of organisms: Proposal for the domains Archaea, Bacteria, and Eucarya. Proc Natl Acad Sci USA 87: 4576–4579.
- 14. Kass RE, Raftery AE (1995) Bayes factors. J Am Stat Assoc 90: 773-795.
- Ratkowsky DA, Olley J, Ross T (2005) Unifying temperature effects on the growth rate of bacteria and the stability of globular proteins. J Theor Biol 233: 351–362.
- Feller G (2013) Psychrophilic enzymes: from folding to function and biotechnology. Scientifica 2013: Article ID 512840.
- Serra A, Strehaiano P, Taillandier P (2005) Inuence of temperature and pH on Saccharomyces bayanus var. uvarum growth; impact of a wine yeast interspecific hybridization on these parameters. Int J Food Microbiol 104: 257–265.
- D'Amico S, Marx JC, Gerday C, Feller G (2003) Activity-stability relationships in extremophilic enzymes. J Biol Chem 278: 7891–7896.
- Murphy KP, Privalov PL, Gill SJ (1990) Common features of protein unfolding and dissolution of hydrophobic compounds. Science 247: 559–561.
- Sibly RM, Brown JH, Kodric-Brown A (2012) Metabolic ecology: a scaling approach. Chichester: John Wiley & Sons, 375 pp.
- Dell AI, Pawar S, Savage VM (2011) Systematic variation in the temperature dependence of physiological and ecological traits. Proc Natl Acad Sci USA 108: 10591–10596.
- Davidson G, Phelps K, Sunderland KD, Pell JK, Ball BV, et al. (2003) Study of temperaturegrowth interactions of entomopathogenic fungi with potential for control of *Varroa destructor* (Acari: Mesostigmata) using a nonlinear model of poikilotherm development. J Appl Microbiol 94: 816–825.
- Humphrey AE (1979) Fermentation process modeling: An overview. Ann N Y Acad Sci 326: 17–33.
- Lee JH, Williamson D, Rogers PL (1980) The effect of temperature on the kinetics of ethanol production by Saccharomyces uvarum. Biotechnol Lett 2: 83–88.
- Urit T, Li M, Bley T, Löser C (2013) Growth of *Kluyveromyces marxianus* and formation of ethyl acetate depending on temperature. Appl Microbiol Biot 97: 10359–10371.
- Huchet V, Pavan S, Lochardet A, Divanac'h ML, Postollec F, et al. (2013) Development and application of a predictive model of Aspergillus candidus growth as a tool to improve shelf life of bakery products. Food Microbiol 36: 254–259.

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#### **Author Contributions**

Conceived and designed the experiments: RC. Performed the experiments: RC. Analyzed the data: RC. Contributed reagents/materials/analysis tools: RC DAR. Wrote the paper: RC TAM JPB DAR JO TR.

- True HL, Lindquist SL (2000) A yeast prion provides a mechanism for genetic variation and phenotypic diversity. Nature 407: 477–483.
- Brocchieri L, Karlin S (2005) Protein length in eukaryotic and prokaryotic proteomes. Nucleic Acids Res 33: 3390–3400.
- Das R, Gerstein M (2000) The stability of thermophilic proteins: a study based on comprehensive genome comparison. Funct Integr Genomic 1: 76–88.
- Berezovsky IN, Shakhnovich EI (2005) Physics and evolution of thermophilic adaptation. Proc Natl Acad Sci USA 102: 12742–12747.
- Takano K, Aoi A, Koga Y, Kanaya S (2013) Evolvability of thermophilic proteins from archaea and bacteria. Biochemistry 52: 4774–4780.
- Graziano G, Catanzano F, Barone G (1998) Prediction of the heat capacity change on thermal denaturation of globular proteins. Thermochim Acta 321: 23-31.
- Luke KA, Higgins CL, Wittung-Stafshede P (2007) Thermodynamic stability and folding of proteins from hyperthermophilic organisms. FEBS J 274: 4023– 4033.
- Razvi A, Scholtz JM (2006) Lessons in stability from thermophilic proteins. Protein Sci 15: 1569–1578.
- Mukaiyama A, Takano K (2009) Slow unfolding of monomeric proteins from hyperthermophiles with reversible unfolding. Int J Mol Sci 10: 1369–1385.
- Kumar S, Nussinov R (2001) How do thermophilic proteins deal with heat? Cell Mol Life Sci 58: 1216–1233.
- Cherry JL (2010) Highly expressed and slowly evolving proteins share compositional properties with thermophilic proteins. Mol Biol Evol 27: 735–741.
- Richter K, Haslbeck M, Buchner J (2010) The heat shock response: life on the verge of death. Mol Cell 40: 253–266.
- Bloom JD, Labthavikul ST, Otey CR, Arnold FH (2006) Protein stability promotes evolvability. Proc Natl Acad Sci USA 103: 5869–5874.
- Wang X, Minasov G, Shoichet BK (2002) Evolution of an antibiotic resistance enzyme constrained by stability and activity trade-offs. J Mol Biol 320: 85–95.
- Svingor A, Kardos J, Hajdú I, Németh A, Závodszky P (2001) A better enzyme to cope with cold: Comparative exibility studies on psychrotrophic, mesophilic, and thermophilic IPMDHs. J Biol Chem 276: 28121–28125.
- Ferrer M, Chernikova TN, Timmis KN, Golyshin PN (2004) Expression of a temperature-sensitive esterase in a novel chaperone-based *Escherichia coli* strain. Appl Environ Microbiol 70: 4499–4504.
- Hollien J, Marqusee S (1999) A thermodynamic comparison of mesophilic and thermophilic ribonucleases H. Biochemistry 38: 3831–3836.
- Ashenberg O, Gong LI, Bloom JD (2013) Mutational effects on stability are largely conserved during protein evolution. Proc Natl Acad Sci USA: 201314781.
- Drake JW (2009) Avoiding dangerous missense: thermophiles display especially low mutation rates. PLoS Genet 5: e1000520.
- Collins T, Meuwis MA, Gerday C, Feller G (2003) Activity, stability and exibility in glycosidases adapted to extreme thermal environments. J Mol Biol 328: 419– 428.
- Córdova J, Ryan JD, Boonyaratanakornkit BB, Clark DS (2008) Esterase activity of bovine serum albumin up to 160 C: a new benchmark for biocatalysis. Enzyme Microb Tech 42: 278–283.
- Nojima H, Ikai A, Oshima T, Noda H (1977) Reversible thermal unfolding of thermostable phosphoglycerate kinase. Thermostability associated with mean zero enthalpy change. J Mol Biol 116: 429–442.
- 49. Sawano M, Yamamoto H, Ogasahara K, Kidokoro Si, Katoh S, et al. (2008) Thermodynamic basis for the stabilities of three CutA1s from Pyrococcus horikoshii, Thermus thermophilus, and Oryza sativa, with unusually high denaturation temperatures. Biochemistry 47: 721–730.
- Hirata A, Sato A, Tadokoro T, Koga Y, Kanaya S, et al. (2012) A stable protein - CutA1. In: Faraggi DE, editor, Protein Structure, InTech. pp. 249–263. Available: http://www.intechopen.com/books/protein-structure/a-stableprotein-cuta1.
- McMeekin TA, Olley JN, Ross T, Ratkowsky DA (1993) Predictive Microbiology: Theory and Application. Taunton, Somerset, England: Research Studies Press Ltd.
- Birch LC (1948) The intrinsic rate of natural increase of an insect population. J Anim Ecol: 15–26.
- Jarosík V, Kratochvíl L, Honék A, Dixon AFG (2004) A general rule for the dependence of developmental rate on temperature in ectothermic animals. Proc R Soc London, Ser B 271: S219–S221.

- Prahlad V, Cornelius T, Morimoto RI (2008) Regulation of the cellular heat shock response in Caenorhabditis elegans by thermosensory neurons. Science 320: 811–814.
- 55. Wiggins P (2008) Life depends upon two kinds of water. PLoS ONE 3: e1406.
- Kim YE, Hipp M, Bracher A, Hayer-Hartl M, Hartl FU (2013) Molecular chaperone functions in protein folding and proteostasis. Annu Rev Biochem 82: 323–355.
- Ellis RJ (2003) Protein folding: importance of the Anfinsen cage. Curr Biol 13: R881–R883.
- Ye X, Lorimer GH (2013) Substrate protein switches GroE chaperonins from asymmetric to symmetric cycling by catalyzing nucleotide exchange. Proc Natl Acad Sci USA 110: E4289–E4297.
- Kort R, Keijser BJ, Caspers MP, Schuren FH, Montijn R (2008) Transcriptional activity around bacterial cell death reveals molecular biomarkers for cell viability. BMC Genomics 9: 590.
- Hoffmann A, Bukau B, Kramer G (2010) Structure and function of the molecular chaperone Trigger Factor. Biochim Biophys Acta 1803: 650–661.
- Fischer G, Schmid FX (1990) The mechanism of protein folding. Implications of in vitro refolding models for de novo protein folding and translocation in the cell. Biochemistry 29: 2205–2212.
- Rosenzweig R, Moradi S, Zarrine-Afsar A, Glover JR, Kay LE (2013) Unraveling the mechanism of protein disaggregation through a ClpB-DnaK interaction. Science 339: 1080–1083.
- Rothman JE, Schekman R (2011) Molecular mechanism of protein folding in the cell. Cell 146: 851–854.
- 64. Okajima T, Kitaguchi D, Fujii K, Matsuoka H, Goto S, et al. (2002) Novel trimeric adenylate kinase from an extremely thermoacidophilic archaeon, *Sulfolobus solfataricus*: Molecular cloning, nucleotide sequencing, Expression in *Escherichia coli*, and characterization of the recombinant enzyme. Biosci Biotech Bioch 66: 2112–2124.
- Boussau B, Blanquart S, Necsulea A, Lartillot N, Gouy M (2008) Parallel adaptations to high temperatures in the Archaean con. Nature 456: 942–945.
- Forterre P (1996) A hot topic: the origin of hyperthermophiles. Cell 85: 789–792.
  Glansdorff N (2000) About the last common ancestor, the universal life-tree and
- lateral gene transfer: a reappraisal. Mol Microbiol 38: 177–185.
- Groussin M, Boussau B, Charles S, Blanquart S, Gouy M (2013) The molecular signal for the adaptation to cold temperature during early life on Earth. Biol Letters 9: 20130608.
- Glansdorff N, Xu Y, Labedan B (2008) The Last Universal Common Ancestor: emergence, constitution and genetic legacy of an elusive forerunner. Biol Direct 3: 29.
- Becerra A, Delaye L, Islas S, Lazcano A (2007) The very early stages of biological evolution and the nature of the last common ancestor of the three major cell domains. Annu Rev Ecol Evol Syst 38: 361–379.
- Tansey MR, Brock TD (1972) The upper temperature limit for eukaryotic organisms. Proc Natl Acad Sci USA 69: 2426–2428.
- Makhatadze GI, Privalov PL (1993) Contribution of hydration to protein-folding thermodynamics: I. The enthalpy of hydration. J Mol Biol 232: 639–659.
- Privalov PL, Gill SJ (1988) Stability of protein structure and hydrophobic interaction. Adv Protein Chem 39: 191–234.
- Privalov PL, Makhatadze GI (1993) Contribution of hydration to protein-folding thermodynamics: II. The entropy and Gibbs energy of hydration. J Mol Biol 232: 660–679.
- Alber SA, Schaffner DW (1992) Evaluation of data transformations used with the square root and Schoolfield models for predicting bacterial growth rate. Appl Environ Microbiol 58: 3337–3342.

- Ratkowsky DA, Ross T, Macario N, Dommett TW, Kamperman L (1996) Choosing probability distributions for modelling generation time variability. J Appl Microbiol 80: 131–137.
- Ng TM, Schaffner DW (1997) Mathematical models for the effects of pH, temperature, and sodium chloride on the growth of Bacillus stearothermophilus in salty carrots. Appl Environ Microbiol 63: 1237–1243.
- Brooks SP (1998) Markov chain Monte Carlo method and its application. J Roy Stat Soc D-Sta 47: 69–100.
- Haario H, Saksman E, Tamminen J (2001) An adaptive Metropolis algorithm. Bernoulli 7: 223–242.
- Gilks WR, Roberts GO, George EI (1994) Adaptive direction sampling. J Roy Stat Soc D-Sta 43: 179–189.
- Gilks WR, Roberts GO (1996) Strategies for improving MCMC. In: Gilks W, Richardson S, Spiegelhalter D, editors, Markov chain Monte Carlo in practice, Boca Raton: Chapman & Hall/CRC. pp. 89–114.
- Carlin BP, Chib S (1995) Bayesian model choice via Markov chain Monte Carlo methods. J Roy Stat Soc B Met: 473–484.
- Billing E (1974) The effect of temperature on the growth of the fireblight pathogen, *Envinia amylovora*. J Appl Bacteriol 37: 643–648.
- 84. Ingraham JL (1958) Growth of psychrophilic bacteria. J Bacteriol 76: 75-80.
- Coultate TP, Sundaram TK (1975) Energetics of *Bacillus stearothermophilus* growth: molar growth yield and temperature effects on growth efficiency. J Bacteriol 121: 55–64.
- Hanus FJ, Morita RY (1968) Significance of the temperature characteristic of growth. J Bacteriol 95: 736–737.
- Mennett RH, Nakayama TOM (1971) Inuence of temperature on substrate and energy conversion in Pseudomonas uorescens. Appl Environ Microbiol 22: 772– 776.
- Ng H, Ingraham JL, Marr AG (1962) Damage and derepression in *Escherichia coli* resulting from growth at low temperatures. J Bacteriol 84: 331–339.
- Price PB, Sowers T (2004) Temperature dependence of metabolic rates for microbial growth, maintenance, and survival. Proc Natl Acad Sci USA 101: 4631–4636.
- Raison JK (1973) The inuence of temperature-induced phase changes on the kinetics of respiratory and other membrane-associated enzyme systems. J Bioenerg Biomembr 4: 285–309.
- Raison JK (1973) Temperature-induced phase changes in membrane lipids and their inuence in metabolic regulation. Symp Soc Exp Biol 27: 485–512.
- Shaw MK (1967) Effect of abrupt temperature shift on the growth of mesophilic and psychrophilic yeasts. J Bacteriol 93: 1332–1336.
- McCrary BS, Edmondson SP, Shriver JW (1996) Hyperthermophile protein folding thermodynamics: differential scanning calorimetry and chemical denaturation of Sac7d. J Mol Biol 264: 784–805.
- Ragone R (2004) Phenomenological similarities between protein denaturation and small molecule dissolution: Insights into the mechanism driving the thermal resistance of globular proteins. Proteins: Struct, Funct, Bioinf 54: 323–332.
- Franks F (1988) Characterization of Proteins. Clifton, New Jersey: The Humana Press Inc, 561 pp.
- Honda S, Yamasaki K, Sawada Y, Morii H (2004) 10 residue folded peptide designed by segment statistics. Structure 12: 1507–1518.
- Liu L, Yang C, Guo QX (2000) A study on the enthalpy-entropy compensation in protein unfolding. Biophys Chem 84: 239–251.
- Jiang X, Farid RS, Pistor E, Farid H (2000) A new approach to the design of uniquely folded thermally stable proteins. Protein Sci 9: 403–416.