

# Preparation and Characterization of a Polyclonal Antibody against Brominated Protein

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**Summary** (Di)bromotyrosine is formed by the specific reaction of eosinophil peroxidase and can be used as an eosinophil activation marker. In the present study, an antibody for (di)bromotyrosine in proteins was prepared to investigate the pathogenesis of eosinophil-related diseases such as allergic responses. A rabbit polyclonal antibody was raised against brominated keyhole limpet hemocyanin. The specificity of the antiserum was investigated with an enzyme-linked immunosorbent assay (ELISA). The antiserum recognized brominated bovine serum albumin (BSA) and dibromotyrosine-conjugated BSA. The antiserum also reacted with chlorinated BSA and di-iodotyrosine-conjugated BSA. Moreover, the specificity of the antiserum was investigated using competitive ELISA. Dibromotyrosine and di-iodotyrosine inhibited the recognition of brominated BSA by the antiserum. However, the recognition of brominated BSA by the antiserum was not inhibited by bromotyrosine, chlorotyrosine, iodotyrosine, nitrotyrosine, aminotyrosine, phosphotyrosine, or tyrosine. These results suggested that the epitope of the antiserum is dihalogenated tyrosine. Immunohistochemically, the antiserum stained brominated rat eosinophils but not chlorinated or nitrated eosinophils. In conclusion, an antiserum for dihalogenated protein was prepared. It is expected that the antiserum will be useful for the analysis of the pathogenesis of allergic diseases such as asthma and atopic dermatitis.

**Key Words:** (di)bromotyrosine, eosinophil activation marker, polyclonal antibody, allergic disease, oxidative stress

## Introduction

Eosinophils play a pivotal role in host defense, such as in killing invasive parasites [1]. Eosinophils are also involved in the pathophysiology of allergic diseases [1–5]. When

eosinophils are activated, eosinophil-specific granule proteins (eosinophil peroxidase (EPO), major basic protein, eosinophil cationic protein, and eosinophil-derived neurotoxin) are released and proceed to induce tissue damage.

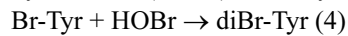
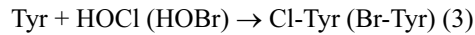
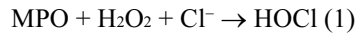
Halogenated tyrosine is formed by the specific reaction of myeloperoxidase (MPO), a neutrophil specific enzyme [6–8], and EPO [9]. MPO prefers chloride (Cl<sup>-</sup>) to bromide (Br<sup>-</sup>) as a substrate and produces hypochlorous acid (HOCl) (equation 1) [2, 10–14]. In contrast, EPO prefers Br<sup>-</sup> to Cl<sup>-</sup> as a substrate and produces hypobromous acid (HOBr) (equation

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2) [2, 9, 13, 15]. HOCl and HOBr halogenate tyrosine (Tyr) residues and produce chlorotyrosine (Cl-Tyr) and bromotyrosine (Br-Tyr), respectively (equation 3). Dibromotyrosine (diBr-Tyr) is also formed by the bromination of Br-Tyr by HOBr (equation 4). When EPO is activated, both Br-Tyr and diBr-Tyr are formed.



Several reports indicate that both EPO and MPO produce HOCl and HOBr *in vitro* [16–19]. However, a study using EPO-knockout mice and MPO-knockout mice showed clearly that the source of Br-Tyr is EPO, and that of Cl-Tyr is MPO *in vivo* [20]. Therefore, (di)Br-Tyr is expected to be useable as a marker for eosinophil activation *in vivo* [9]. In fact, high levels of Br-Tyr in the urine, endotracheal/bronchial aspirate, and sputum of asthma patients have been reported [2, 21, 22]. In these reports, bromotyrosine was measured by gas chromatography/mass spectrometry. If an antibody for (di)Br-Tyr is obtained, it will be useful for investigation of the pathogenesis of eosinophil-related allergic diseases such as asthma and atopic dermatitis. Thus, this study attempted to prepare a specific polyclonal antibody for (di)Br-Tyr.

## Materials and Methods

### Reagents

Bovine serum albumin (BSA), Tyr, 3,5-diiodo-L-tyrosine (diI-Tyr) dihydrate, 3-nitro-L-tyrosine (NO<sub>2</sub>-Tyr), Cl-Tyr hydrochloride, and *O*-phospho-L-tyrosine (P-Tyr) were purchased from Sigma Chemical Co. (St. Louis, MO). 3-Amino-L-tyrosine (NH<sub>2</sub>-Tyr) dihydrochloride, iodotyrosine (I-Tyr), and 3-chloro-4-hydroxybenzoic acid (Cl-HBA) were from Aldrich Chemical Company, Inc. (Milwaukee, WI). 3,5-Dibromo-L-tyrosine (diBr-Tyr), 3,5-dibromo-4-hydroxybenzoic acid (diBr-HBA), 3,5-dichloro-4-hydroxybenzoic acid (diCl-HBA), 3-nitro-4-hydroxybenzoic acid (NO<sub>2</sub>-HBA), and 3-amino-4-hydroxybenzoic acid (NH<sub>2</sub>-HBA) were from Tokyo Kasei Kogyo Co. Ltd. (Tokyo, Japan). 3-Bromo-4-hydroxybenzoic acid (Br-HBA), sodium hypochlorite, hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), and normal goat serum were obtained from Wako Pure Chemical Industries, Ltd. (Osaka, Japan). 3,5-Dinitro-4-hydroxybenzoic acid (diNO<sub>2</sub>-HBA) was from AVOCADO Research Chemicals, Ltd. (Lancashire, UK). Peroxynitrite, diethylenetriamine-*N,N,N',N'',N'''*-pentaacetic acid (DTPA), and *N*-(6-maleimidocaproyloxy)-succinimide (EMCS) were from Dojindo Laboratories (Kumamoto, Japan). Imject<sup>®</sup> mariculture keyhole limpet hemocyanin (KLH), Freund's complete adjuvant, Freund's incomplete adjuvant, *N*-succinimidyl 3-(2-pyridyldithio)propionate (SPDP), and 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (EDC)

were from PIERCE Biotechnology, Inc. (Rockford, IL, USA). Sodium hypobromite was from Kanto Chemical Co., Ltd. (Tokyo, Japan). Solvents and other reagents were of the highest grade commercially available. The absorbance of hypochlorite and hypobromite were measured spectrophotometrically and their concentrations were calculated from their molar extinction coefficients ( $\epsilon_{292\text{nm}} = 350 \text{ M}^{-1}\text{cm}^{-1}$  [23] and  $\epsilon_{331\text{nm}} = 315 \text{ M}^{-1}\text{cm}^{-1}$  [24], respectively). Br-Tyr was prepared by the bromination of Tyr using sodium hypobromite and purified by high performance liquid chromatography (HPLC).

### Animals

Japanese white rabbits were purchased from Kiwa Laboratory Animals Co., Ltd. (Wakayama, Japan). Wistar rats were purchased from Shiroishi Lab. Animal Reproduction Inc. (Tokyo, Japan). The care and handling of the animals were in accordance with the Guidelines for the Care and Use of Laboratory Animals of Kanazawa University.

### Preparation of modified protein

One millimolar sodium hypobromite or 5 mM sodium hypochlorite was added to 5 mg/ml BSA or KLH at room temperature. After dialysis with 10 mM phosphate-buffered saline (pH 7.2), a part of the modified protein was digested with pronase (Roche Diagnostics, Basel, Switzerland). Br-Tyr and Cl-Tyr were analyzed by HPLC to confirm the bromination and chlorination, respectively, of tyrosine in the proteins. Water/methanol/acetic acid (= 97/3/0.1, v/v/v) and a CAPCELLPAK C18 column (4.6 × 250 mm, 5 μm, Shiseido Co., Ltd., Tokyo, Japan) were used as the mobile phase and column, respectively. DiBr-Tyr was analyzed by HPLC to confirm the dibromination of tyrosine in the proteins. Water/methanol/acetic acid (= 80/20/0.1, v/v/v) and a CAPCELLPAK C18 column (4.6 × 250 mm, 5 μm, Shiseido Co., Ltd., Tokyo, Japan) were used as the mobile phase and column, respectively. Modified tyrosine was detected at 274 nm. Nitrated BSA was prepared by the reaction of 5 mg/ml BSA with 1 mM peroxyxynitrite. Tyr or modified Tyr was conjugated with BSA using SPDP and EMCS as reported previously [25]. Modified hydroxybenzoic acid (HBA) was conjugated with BSA using EDC as reported previously [26]. After dialysis, the concentrations of the modified BSAs were determined by the Bradford protein quantitation assay (Bio-Rad Laboratories, Hercules, CA).

### Immunization

Brominated KLH (500 μg) as an immunogen in 2 ml of 10 mM phosphate-buffered saline (pH 7.2) was mixed with 4 ml of Freund's complete adjuvant (first injection) or Freund's incomplete adjuvant (booster injections). Arterial blood was drawn from the ear of the rabbit before the administration of the immunogen subcutaneously at 2 week

intervals. The titer of the antiserum was examined using an enzyme-linked immunosorbent assay (ELISA).

#### ELISA

Five hundred nanograms of antigen dissolved in 60 mM carbonate buffer (pH 9.6) was added to each well of a 96 well microplate and incubated at 4°C overnight. Three percent skim milk in 10 mM Tris-buffered saline was used for the blocking of non-specific binding. The rabbit antiserum (1:3,000) was used as a primary antibody and horseradish peroxidase-conjugated goat anti-rabbit-immunoglobulin (1:2,000, Dako Cytomation, Glostrup, Denmark) as a secondary antibody. Visualization was performed using a Colorimetric kit T (Sumitomo Bakelite Co., Ltd., Tokyo, Japan). The absorbance at 450 nm was measured using a microplate reader. When competitive ELISA was performed, the antiserum was preincubated with competitor for 2 h. The pH of each competitor solution was adjusted to 7.4 before addition to the antiserum.

#### Immunohistochemistry

Horse serum was injected intraperitoneally into a rat every 3 days [27]. At 24 h after the third injection, peritoneal exudate cells were collected by washing the abdominal cavity with Hanks' balanced salt solution. On a cytospin slide ( $2 \times 10^5$  cells), rat eosinophils were stained with Luxol fast blue. The cytospin slides (with eosinophil-containing cells) were incubated with 100  $\mu$ M sodium bromide, 140 mM sodium chloride, or 50  $\mu$ M sodium nitrite in the presence of 100  $\mu$ M DTPA plus 100  $\mu$ M H<sub>2</sub>O<sub>2</sub> in 50 mM potassium phosphate buffer (pH 7.4) for 5 min at 37°C. Control specimens were incubated with 100  $\mu$ M DTPA or 100  $\mu$ M DTPA plus 100  $\mu$ M H<sub>2</sub>O<sub>2</sub>. Cells were fixed with 10% formalin. Endogenous peroxidase activity was blocked with 0.3% H<sub>2</sub>O<sub>2</sub>-containing methanol. Five percent normal goat serum was used to block non-specific binding. The rabbit antiserum (1:4,000) was incubated with specimens at 4°C overnight. As a secondary antibody, anti-rabbit immunoglobulin antibody with horseradish peroxidase-labeled dextran polymer (1:16, Dako Cytomation) was used. Staining was performed by the reaction with 3,3'-diaminobenzidine tetrahydrochloride and H<sub>2</sub>O<sub>2</sub>. The cells were counterstained with hematoxylin.

#### Statistical analysis

Statistical analysis of the differences in the mean values was performed by two-way analysis of variance (ANOVA) and one-way ANOVA, followed by the Tukey's post hoc test using SPSS 11.5 software (SPSS, Chicago, IL).

## Results

#### Synthesis of the immunogen and antigens

KLH was brominated using HOBr (Fig. 1a). An analytical method for various modified Tyr residues using HPLC was developed to confirm the bromination of Tyr in brominated KLH. The retention times of Tyr, Cl-Tyr, Br-Tyr, and NO<sub>2</sub>-Tyr were 5.9, 14.9, 20.1, and 20.9 min, respectively (Fig. 2A). Another analytical method for diBr-Tyr (retention time: 11.1 min) was also developed (Fig. 2C). Using these methods, the pronase hydrolysate of brominated KLH was analyzed (Fig. 2B and D). Br-Tyr and diBr-Tyr were formed in KLH but Cl-Tyr and NO<sub>2</sub>-Tyr were not observed in the hydrolysate. Therefore, brominated KLH was used as the immunogen. Modified BSA, modified Tyr-conjugated BSA and modified HBA-conjugated BSA as antigens for ELISA were synthesized as described in the "Materials and Methods" section (Fig. 1a-c).

#### Characteristics of the antiserum: ELISA

Rabbit antiserum raised against brominated KLH was obtained, and the reactivity of the antiserum was investigated using ELISA (Fig. 3A). The recognition of brominated BSA by the antiserum (1:3,000) was proportional to the antigen concentrations and reached a plateau at 500 ng/well. In contrast, the antiserum did not recognize BSA. The titer of the antiserum was examined (Fig. 3B). The recognition of brominated BSA by the antiserum was inversely proportional to the dilution ratio of the antiserum. When a high concentration of antiserum was used, it recognized BSA very weakly. However, when 3,000-fold-diluted antiserum was used, the recognition of BSA was negligible. Thus, the optimum concentrations of the antigen and antiserum for ELISA were determined to be 500 ng/well and 1:3,000, respectively.

The antigen specificity of the antiserum was investigated using various modified BSAs as antigens for ELISA (Fig. 4). The antiserum recognized brominated BSA and diBr-Tyr-conjugated BSA. However, it did not recognize diBr-HBA-conjugated BSA. The antiserum reacted with diI-Tyr-conjugated BSA and chlorinated BSA. Weak recognition of Br-Tyr-conjugated BSA and Cl-Tyr-conjugated BSA by the antiserum was observed, but its level was the same as that of the recognition of Tyr-conjugated BSA by the antiserum. NO<sub>2</sub>-Tyr can be formed *in vivo* via peroxidase-catalyzed nitration of Tyr. NH<sub>2</sub>-Tyr was also formed *in vivo* via the reduction of NO<sub>2</sub>-Tyr. The antiserum weakly recognized NO<sub>2</sub>-Tyr-conjugated BSA and NH<sub>2</sub>-Tyr-conjugated BSA, but it did not react with nitrated BSA, Br-HBA-conjugated BSA, diCl-HBA-conjugate BSA, Cl-HBA-conjugated BSA, NO<sub>2</sub>-HBA-conjugated BSA, NH<sub>2</sub>-HBA-conjugated BSA, or BSA. These results suggest that the antiserum recognized dihalogenated tyrosine.

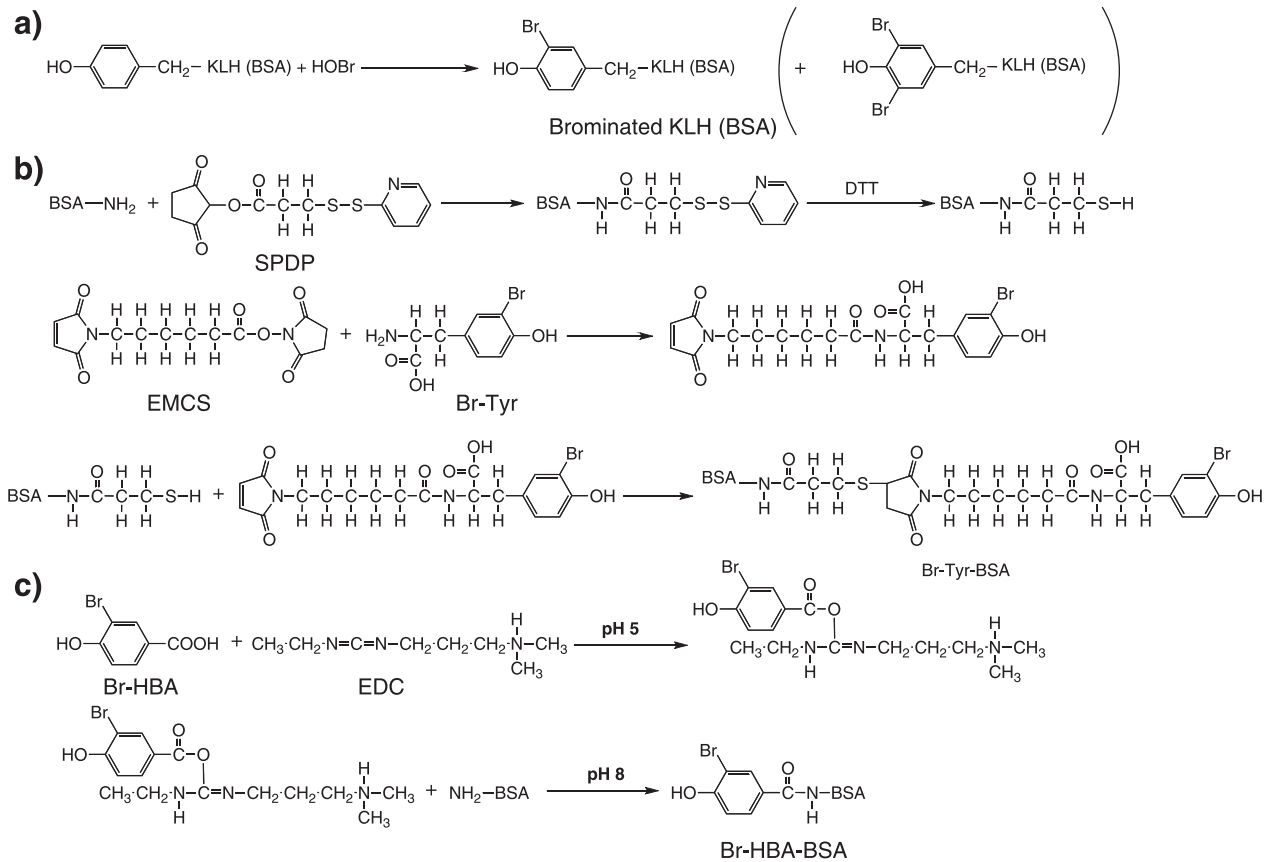


Fig. 1. Procedure for the synthesis of antigens. (A) Brominated KLH (BSA). (B) Modified Tyr-conjugated BSA. (C) Modified HBA-conjugated BSA.

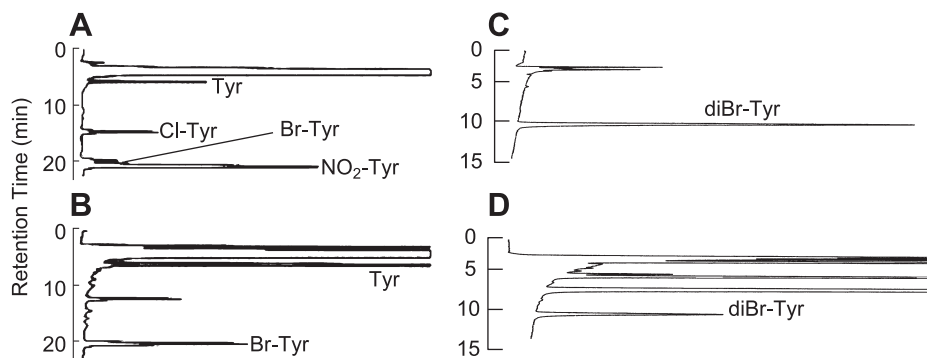


Fig. 2. HPLC analytical chromatograms of various modified Tyrs. (A, C) Standards. (B, D) Pronase hydrolysate of brominated KLH. (A, B) Analytical method for various modified Tyr residues. (C, D) Analytical method for diBr-Tyr.

#### Characteristics of the antiserum: competitive ELISA

To further confirm the antigen specificity of the antiserum, competitive ELISA was performed. When brominated BSA was used as an antigen, diBr-Tyr and diI-Tyr inhibited the recognition of the antigen by the antiserum (Fig. 5A). DiBr-HBA also inhibited the recognition of brominated BSA by the antiserum (Fig. 5C). Comparable results were obtained

when diBr-Tyr-conjugated BSA or chlorinated BSA was used as antigen (data not shown).

#### Characteristics of the antiserum: Immunohistochemistry

The antiserum was applied for immunohistochemical analysis. Part of a sample of rat peritoneal exudate cells was stained with Luxol fast blue, which selectively stains

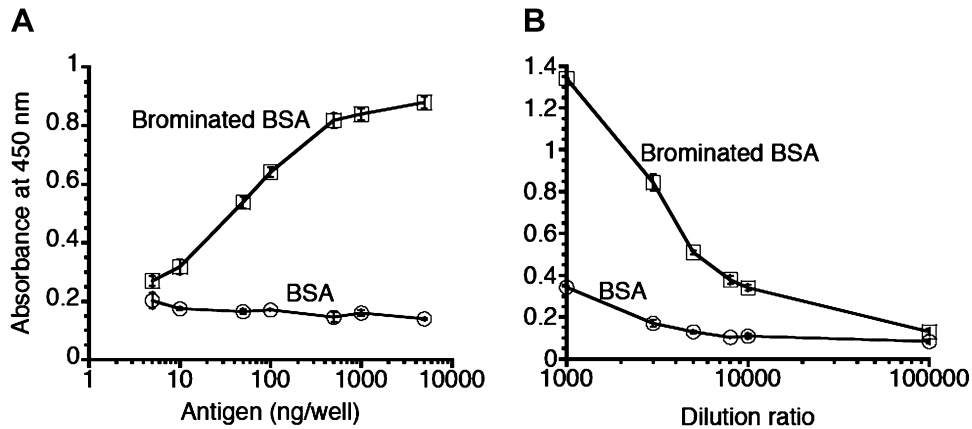


Fig. 3. Characteristics of the antiserum. (A) Reactivity of the antiserum. Each well was coated with 5 to 500 ng/well antigen (brominated BSA or BSA) and incubated with 3,000-fold-diluted antiserum. The results are expressed as mean  $\pm$  standard deviation ( $n = 3$ ). Statistical analysis of the differences was performed by two-way ANOVA. Significant main effects of the concentration of antigen ( $p < 0.001$ ) and the kind of antigen ( $p < 0.001$ ) and significant interactive effect between the concentration of antigen and the kind of antigen ( $p < 0.001$ ) are shown. (B) Titer of the antiserum. Each well was coated with 500 ng/well antigen (brominated BSA or BSA) and incubated with 1,000 to 100,000-fold-diluted antiserum. The results are expressed as mean  $\pm$  standard deviation ( $n = 3$ ). Statistical analysis of the differences was performed by two-way ANOVA. Significant main effects of the concentration of antigen ( $p < 0.001$ ) and the kind of antigen ( $p < 0.001$ ), and significant interactive effect between the concentration of antigen and the kind of antigen ( $p < 0.001$ ) are shown.

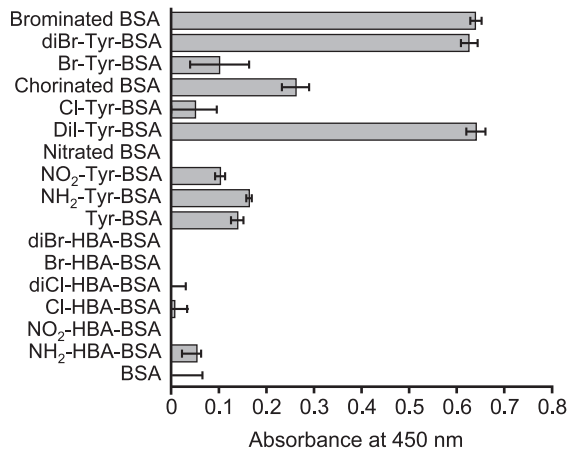


Fig. 4. The recognition of various modified tyrosines (Tyr) or hydroxybenzoic acids (HBA) in BSA by the antiserum. ELISA was performed to evaluate the specificity of the antiserum. Each well was coated with 500 ng/well of antigen and incubated with 3,000-fold-diluted antiserum. The results are expressed as mean  $\pm$  standard deviation ( $n = 3$ ). Statistical analysis of the differences was performed by one-way ANOVA ( $p < 0.001$ ).

eosinophil granules (Fig. 6A). These cells had ring-shaped nuclei, which are characteristic of rat eosinophils (Fig. 6A). The physiological concentrations of Br<sup>-</sup> and nitrite in the plasma of healthy humans are 16–101  $\mu$ M [28] and 0.1–20  $\mu$ M [29], respectively, and the concentration of Cl<sup>-</sup> in saline is 137 mM. In the inflammatory state, the concentra-

tion of nitrite in the plasma increases. Therefore, 100  $\mu$ M Br<sup>-</sup>, 140 mM Cl<sup>-</sup>, or 50  $\mu$ M nitrite was used to modify the rat eosinophils. Control and H<sub>2</sub>O<sub>2</sub>-treated eosinophils were not stained by the antiserum (Fig. 6B and C). This result indicates that the antiserum does not recognize the oxidative modification of tyrosine or other substrates. Brominated eosinophils were stained by the antiserum (Fig. 6D), whereas chlorinated cells and nitrated cells were not stained (Fig. 6E and F, respectively). The nitration of eosinophils on the slides by nitrite and H<sub>2</sub>O<sub>2</sub> was confirmed by staining with anti-NO<sub>2</sub>-Tyr polyclonal antibody (data not shown).

## Discussion

The rabbit antiserum obtained in the present study recognized diBr-Tyr in BSA. It reacted with diCl-Tyr in BSA. However, the antiserum specifically detected HOBr-modified protein of rat eosinophils in immunohistochemistry. This difference can be explained at least in part by the difference between the affinity of the antibody to diBr-Tyr and that to diCl-Tyr, the steric hindrance in the protein structure, the difference in protein species modified by the two chemicals (HOBr and HOCl), and/or the difference in the site of the tyrosine residues in a protein modified by the two chemicals. The site-directed modification of tyrosine (nitration, chlorination, bromination, oxidation) has been reported previously [30–33].

The present results from the competitive ELISA indicate that the affinity of the antiserum to diBr-Tyr and diI-Tyr was

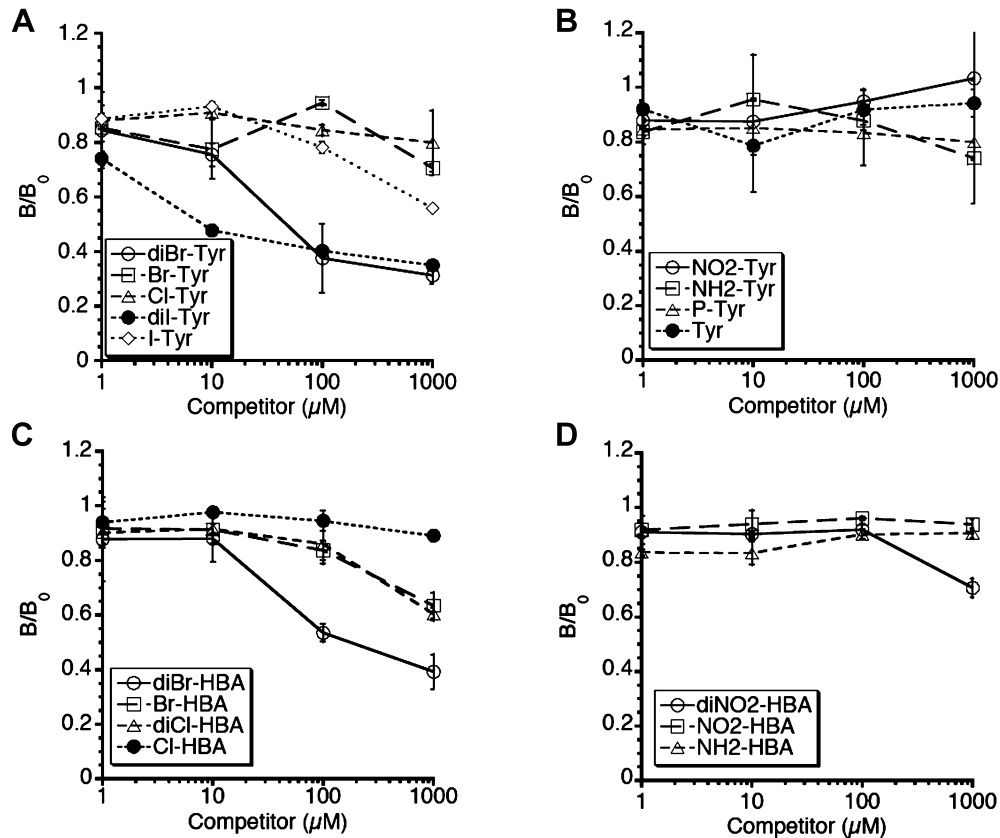


Fig. 5. Competitive ELISA using modified Tyrs and HBAs. The rabbit antiserum was preincubated with various concentrations of each modified Tyr or modified HBA. Each well was coated with 500 ng/well of brominated BSA and incubated with 3,000-fold-diluted antiserum. The results are expressed as the mean  $\pm$  standard deviation ( $n = 3$ ). Statistical analysis of the differences was performed by two-way ANOVA. Significant main effects of the concentration of competitor ( $p < 0.001$ ) and the kind of competitor ( $p < 0.001$ ) and interactive effect between the concentration of competitor and the kind of competitor ( $p < 0.001$ ) are shown.

strongest. The antiserum strongly recognized brominated BSA, diBr-Tyr-conjugated BSA, and diI-Tyr-conjugated BSA but only weakly recognized Br-Tyr-conjugated BSA and Cl-Tyr-conjugated BSA in the ELISA tests. DiNO<sub>2</sub>-Tyr could not be recognized by the antiserum since diNO<sub>2</sub>-Tyr was probably produced during nitration of BSA by peroxynitrite and the antiserum did not recognize nitrated BSA. Thus, the epitope recognized by the antiserum is likely to be dihalogenated Tyr.

The antiserum weakly recognized NO<sub>2</sub>-Tyr-conjugated BSA at a similar level to Tyr-conjugated BSA by ELISA. Nitrated BSA was not recognized by the antiserum. These results suggest that the antiserum could recognize the tyrosine structure conjugated with the cross-linker but not the intrinsic tyrosine structure in BSA. Probably, the antiserum recognized Br-Tyr-conjugated BSA, Cl-Tyr-conjugated BSA, and NH<sub>2</sub>-Tyr-conjugated BSA by recognition of the cross-linked Tyr structure. The recognition of brominated BSA and chlorinated BSA by the antiserum showed that the antiserum could recognize dihalogenation of intrinsic

tyrosine in protein.

The recognition of brominated BSA by the antiserum was inhibited by diBr-HBA, but the antiserum did not recognize diBr-HBA-conjugated BSA. The cause of this discrepancy may be steric hindrance due to binding of diBr-HBA to the cross-linker.

Several antibodies for halogenated tyrosine have been reported [8, 34–36]. Monoclonal anti-Cl-Tyr antibody (clone 2D10G9) raised against HOCl-modified low density lipoprotein [34, 36] also recognized HOBr-treated BSA by western blotting analysis [37]. Moreover, a monoclonal antibody (clone 3A5) raised against *N*-bromosuccinimide-modified KLH recognized diBr-Tyr-conjugated BSA and HOCl-modified BSA as well as *N*-bromosuccinimide-modified BSA by ELISA [35]. The recognition of diBr-Tyr-conjugated BSA by the antibody was inhibited by various (di)halogenated Tyrs and HBAs, such as diBr-Tyr, *N*-acetylchlorotyrosine, diI-Tyr, diBr-HBA, diCl-HBA, and Br-Tyr, using competitive ELISA [35]. The epitope of this antibody was assumed to be dihalogenated Tyr and HBA

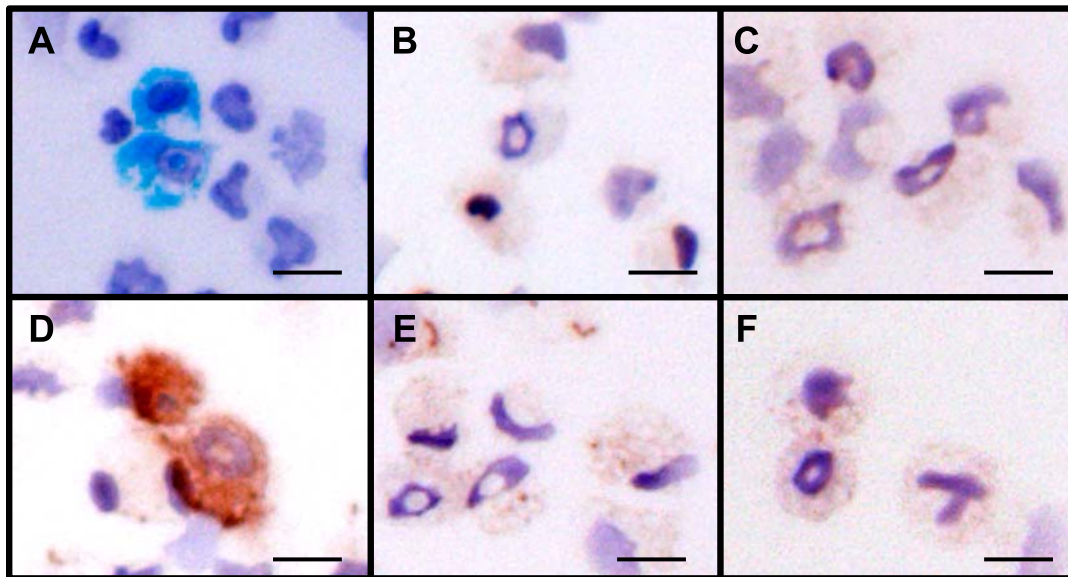


Fig. 6. Immunohistochemical staining of brominated protein in rat eosinophils. (A) Rat eosinophils were detected using Luxol fast blue staining. (B-E) Chemically treated specimens (eosinophil-containing cells) were incubated with 4,000-fold-diluted rabbit antiserum. (B) Control (100  $\mu$ M DTPA). (C) B + 100  $\mu$ M  $H_2O_2$ . (D) C + 100  $\mu$ M  $Br^-$ . (E) C + 140 mM  $Cl^-$ . (F) C + 50  $\mu$ M nitrite. Scale bars represent 25  $\mu$ m.

[35]. Thus, the discrimination of each of the (di)halogenated proteins by antibody may be difficult. Therefore, polyclonal anti-Cl-Tyr antibody raised against Cl-HBA-conjugated KLH [8] may also recognize various halogenated proteins, such as brominated protein. Our antiserum raised against HOBr-modified KLH also recognized various (di)halogenated Tyr (but not halogenated HBA) using ELISA. However, the antiserum specifically stained brominated rat eosinophils using an immunohistochemical method. The possibility that chlorination of rat eosinophils did not occur could not be denied in the present study because confirmation of the chlorination of rat eosinophils on a slide is difficult. Although the antibodies reported for halogenated tyrosine may recognize specific antigens immunohistochemically, this point has not been investigated.

It may be difficult to produce a specific antibody for (di)Br-Tyr in proteins because only one or two residues are different between (di)Br-Tyr and other modified tyrosine residues. In particular, the nature of one halogen is similar to other halogens.

This study attempted to make a specific anti-Br-Tyr antibody by immunization of Br-Tyr-conjugated KLH or Br-HBA-conjugated KLH into a rabbit, but this proved not to be achievable. An antiserum that recognized diBr-Tyr was successfully recovered when brominated KLH was used as the immunogen. However, the antiserum did not recognize Br-Tyr. A previously reported monoclonal antibody also recognized dihalogenated BSA, but not Br-Tyr-conjugated BSA [35]. Therefore, it may also be difficult to make an anti-

Br-Tyr antibody because of the probable weak characteristics of Br-Tyr as an epitope.

The content of Br-Tyr and Cl-Tyr on proteins in bronchoalveolar lavage fluid recovered from lung segments of asthmatic subjects was increased by allergen-challenge [38]. The levels of Br-Tyr and Cl-Tyr in the urine [22], endotracheal/bronchial aspirate [2], and sputum [27] in asthma patients were high. In these cases, diBr-Tyr and diCl-Tyr could also be formed. In allergic diseases, activated eosinophils could produce (di)Br-Tyr. Inflammation could also occur in allergic diseases, neutrophils could be activated, and (di)Cl-Tyr may be formed. Therefore, the current antiserum can be used for the analysis of the pathogenesis in allergy diseases, although it was not a diBr-Tyr-specific antiserum.

#### Acknowledgement

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#### Abbreviations

Br-HBA, bromohydroxybenzoic acid; Br-Tyr, bromotyrosine; Cl-HBA, chlorohydroxybenzoic acid; Cl-Tyr, chlorotyrosine; diBr-HBA, dibromohydroxybenzoic acid; diBr-Tyr, dibromotyrosine; diCl-HBA, dichlorohydroxybenzoic acid; diBr-Tyr, dichlorotyrosine; DTPA, diethylenetriamine-



*N,N,N',N''*-pentaacetic acid; diNO<sub>2</sub>-HBA, dinitrohydroxybenzoic acid; diI-Tyr, di-iodotyrosine; EDC, 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide-HCl; EMCS, *N*-(6-maleimidocaproyloxy)-succinimide; EPO, eosinophil peroxidase; I-Tyr, iodotyrosine; HBA, hydroxybenzoic acid; KLH, keyhole limpet hemocyanin; MPO, myeloperoxidase; NH<sub>2</sub>-HBA, aminohydroxybenzoic acid; NH<sub>2</sub>-Tyr, aminotyrosine; NO<sub>2</sub>-HBA, nitrohydroxybenzoic acid; NO<sub>2</sub>-Tyr, nitrotyrosine; P-Tyr, phosphotyrosine; SPDP, *N*-succinimidyl 3-(2-pyridyldithio)propionate; Tyr, tyrosine.

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