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Crystal structure of bis(bis{(E)-[(6-{(E)-[(4-fluorobenzyl)imino]methyl}pyridin-2vl)methylidene](4-fluorophenyl)amine}nickel(II)) tetrabromide nonahvdrate

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In the title complex, $[Ni(C_{21}H_{17}F_2N_3)_2]_2Br_4 \cdot 9H_2O$, there are two independent metal complexes per asymmetric unit and two ligands per metal complex. The structural features (bond lengths and angles) of the two complexes are almost identical. In each complex, the nickel(II) ion is coordinated in an octahedral environment by six N atoms from two chelating (9*E*)-*N*-({6-[(*E*)-(4-fluorobenzylimino)methyl]pyridin-2-yl}methylene)(4-fluorophenyl)methanammine ligands. The Ni-N bond lengths range from 1.973 (2) to 2.169 (2) Å, while the chelate N-Ni-N angles range from 77.01 (10) to $105.89 (9)^{\circ}$. Additionally, there are four bromide anions and nine solvent water molecules within the asymmetric unit. The water molecules form a hydrogen-bonded network, displaying C- $H \cdots O, C - H \cdots Br, O - H \cdots Br, O - H \cdots O$ and $O - H \cdots F$ interactions into layers parallel to (111). In each unit, the fluorophenyl rings of one ligand are stacked with the central ring of the other ligand via π - π interactions, with the closest centroid-to-plane distances being 3.445 (5), 3.636 (5), 3.397 (5) and 3.396 (5) Å.

Keywords: crystal structure; nickel(II) complex; octahedral geometry; Schiff base; $\pi - \pi$ interactions; pyridine derivatives.

CCDC reference: 1436891

1. Related literature

For general background to coordination complexes with Schiff bases, see: Vigato & Tamburini (2004); Gupta & Sutar (2008). For applications and bioactivity of metal complexes, see: Skyrianou et al. (2010). For related structures, see: You et al. (2014). For the preparation, see: Işıklan et al., 2011).



2. Experimental

2.1. Crystal data

 $[Ni(C_{21}H_{17}F_2N_3)_2]_2Br_4 \cdot 9H_2O$ $M_r = 1996.70$ Triclinic, $P\overline{1}$ a = 12.1000 (16) Åb = 12.5686 (16) Å c = 29.149 (4) Å $\alpha = 87.377(2)^{\circ}$ $\beta = 86.955 (2)^{\circ}$

2.2. Data collection

Bruker APEX CCD diffractometer Absorption correction: multi-scan (SADABS; Krause et al., 2015) $T_{\min} = 0.571, T_{\max} = 0.760$

2.3. Refinement

Table 1

$R[F^2 > 2\sigma(F^2)] = 0.042$	H atoms treated
$wR(F^2) = 0.106$	independent an
S = 1.01	refinement
19210 reflections	$\Delta \rho_{\rm max} = 0.84 \ {\rm e} \ {\rm \AA}$
1126 parameters	$\Delta \rho_{\rm min} = -0.77 \ {\rm e}$
27 restraints	

19210 independent reflections 15090 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.058$

94344 measured reflections

 $\gamma = 72.339 \ (2)^{\circ}$

Z = 2

 $V = 4216.0 (10) \text{ Å}^3$

Mo $K\alpha$ radiation

 $0.26 \times 0.20 \times 0.12 \text{ mm}$

 $\mu = 2.42 \text{ mm}^{-1}$

T = 100 K

H atoms treated by a mixture of
independent and constrained
refinement
$\Delta \rho_{\rm max} = 0.84 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.77 \ {\rm e} \ {\rm \AA}^{-3}$

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Selecte	he	hond	lengths	(Δ)
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Ni1-N4	1.984 (2)	Ni2-N10	1.973 (2)
Ni1-N1	1.984 (2)	Ni2-N7	1.981 (2)
Ni1-N6	2.141 (2)	Ni2-N11	2.125 (2)
Ni1-N5	2.151 (2)	Ni2-N9	2.158 (2)
Ni1-N2	2.158 (2)	Ni2-N12	2.161 (2)
Ni1-N3	2.165 (2)	Ni2-N8	2.169 (2)

Table 2		
Hydrogen-bond geometry	(Å,	°).

$D - H \cdots A$	<i>D</i> -H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
C4-H4···O6	0.95	2.60	3.505 (4)	160
$C6-H6\cdots F7^{i}$	0.95	2.35	3.238 (3)	155
$C18-H18\cdots F4^{i}$	0.95	2.45	3.363 (4)	162
C23-H23···O2	0.95	2.56	3.504 (4)	172
C25-H25···F2 ⁱⁱ	0.95	2.43	3.303 (4)	154
C28−H282···Br3 ⁱⁱⁱ	0.99	2.80	3.784 (3)	170
C31-H31···O5 ⁱⁱⁱ	0.95	2.66	3.557 (5)	157
C36-H362···Br2 ^{iv}	0.99	2.97	3.933 (3)	164
C39−H39···O3 ^v	0.95	2.44	3.373 (4)	166
$C44-H44\cdots F8^{vi}$	0.95	2.48	3.341 (4)	150
C46−H46···Br3 ⁱⁱ	0.95	2.79	3.708 (3)	164
C52−H52···O9 ^{iv}	0.95	2.53	3.292 (6)	138
C60−H60···F8 ^{vii}	0.95	2.55	3.329 (4)	139
C62−H62···O9 ^{viii}	0.95	2.48	3.365 (6)	156
C65-H65···Br4 ^{viii}	0.95	2.91	3.653 (3)	136
$C67 - H67 \cdots O2^{iv}$	0.95	2.58	3.503 (4)	165
C69−H69···Br1 ^{iv}	0.95	3.07	3.750 (3)	130
C73−H73···F1 ⁱⁱ	0.95	2.60	3.312 (4)	132
$C81 - H81 \cdots F5^{ix}$	0.95	2.50	3.436 (4)	169
$C83 - H83 \cdots O1^{x}$	0.95	2.52	3.263 (4)	135
O1−H111···Br4 ^{xi}	0.89(2)	2.57 (2)	3.449 (3)	167 (3)
O1−H112···Br4	0.88(2)	2.61 (2)	3.476 (3)	170 (3)
O2−H211···Br3	0.89(2)	2.50(2)	3.384 (3)	168 (3)
O2−H212···Br1	0.87 (2)	2.44 (2)	3.317 (2)	178 (3)
O3−H311···Br2	0.90(2)	2.39 (2)	3.274 (2)	169 (3)
O3−H312···Br1	0.89(2)	2.44 (2)	3.307 (2)	164 (3)
O4−H411···O5	0.95 (2)	2.12 (2)	3.033 (5)	161 (4)
O4−H412···Br1	0.93 (2)	2.32 (2)	3.250 (3)	178 (4)
O5−H511···O3	0.88(2)	1.93 (2)	2.809 (4)	172 (5)
O5−H512···O7 ^{xii}	0.90(2)	2.11 (3)	2.906 (4)	147 (4)
O6−H611···Br2 ^{iv}	0.86(2)	2.57 (2)	3.407 (2)	165 (3)
O6−H612···Br2 ⁱⁱⁱ	0.87(2)	2.44 (2)	3.306 (3)	173 (4)
O7−H711···O6	0.90 (2)	2.03 (2)	2.925 (4)	171 (4)
O8−H811···O9	0.94 (2)	2.45 (3)	3.358 (6)	163 (4)
O8−H812···Br4	0.94 (2)	2.71 (3)	3.579 (3)	154 (4)

 Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2014*/7 (Sheldrick, 2015); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL2014*/7.

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Acta Cryst. (2015). E71, m226-m227 [doi:10.1107/S2056989015021519]

Crystal structure of bis(bis{(*E*)-[(6-{(*E*)-[(4-fluorobenzyl)imino]methyl}pyridin-2yl)methylidene](4-fluorophenyl)amine}nickel(II)) tetrabromide nonahydrate

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S1. Chemical context

Schiff bases are the condensation product of an amine and an active carbonyl compound which are capable of forming coordination complexes with transition metal ions (Vigato *et al.*, 2004; Gupta *et al.*, 2008). In particular, metal complexes derived from Schiff bases with multiple binding sites are effective in many biochemical and antimicrobial applications (Skyrianou *et al.*, 2010).

S2. Structural commentary

The reported crystal structure of Ni(II) complex with 2,6-diformylpyridine based Schiff base, contains two metal complexes per asymmetric unit. Each metal complex is formed with two ligands, providing almost identical structure to each other (Fig. 1). In each unit, the nickel(II) ion is octahedrally coordinated with two chelating ligands. The Ni—N bond lengths vary from 1.973 (2) to 2.169 (2) Å, while the N—Ni—N angles are in the range from 77.01 (10) to 105.89 (9) ° (Table 1). Similar coordination pattern of of nickel(II) is reported in the literature (You *et al.* 2014). External bromides and water molecules are liked with the ligands via hydrogen bonding interactions with C—H···O, C—H···Br, O —H···Br, O —H···O and O—H···F bonds given in Fig. 2 and Table 2. Weak C—H···F interactions are also observed (3.238 (3) to 3.436 (4) Å). In each unit, two fluorophenyl rings of one ligand are stacked with the central pyridine ring of the other ligand *via* π - π interactions. In one unit, the distances from the plane of central ring (N1 C1 C2 C3 C4 C5) to the centroids of fluorophenyl rings (C29 C30 C31 C32 C33 C34 and C37 C38 C39 C40 C41 C42) are 3.445 (5) and 3.636 (5) Å; while in the other unit, the corresponding distances from the plane of central ring (N7 C43 C44 C45 C46 C47) to the centroids of fluorophenyl rings (C71 C72 C73 C74 C75 C76 and C79 C80 C81 C82 C83 C84) are 3.397 (5) and 3.396 (5) Å (Fig. 3).

S3. Supramolecular features

Intermolecular interactions form various C—H···O, C—H···Br, O—H···Br, O—H···O and O—H···F bonds (Table 1 and Figure 2). In each unit, there are also π - π interactions between the fluorophenyl rings of one ligand and the central ring of the other ligand.

S4. Synthesis and crystallization

2,6-diformylpyridine (0.500 g, 3.70 mmol) and 4-fluorobenzylamine (845.8 μ L, 7.40 mmol) were separately dissolved in 10 mL methanol. The two solutions were slowly mixed in 20 mL methanol with constant stirring over 10 minutes at room temperature following the similar method as described earlier (Işıklan *et al.*, 2011). The mixture was further stirred overnight and left at room temperature for 24 hours. White crystalline powder thus formed was filtered and dried at room temperature. Yield: 95 %. Melting point is 89°C.

The nickel complex was obtained by mixing of the ligand (0.349 g, dissolved in 5 mL of methanol) and nickel(II) bromide hydrate (0.1092 g, dissolved in 5 mL of water) in water-methanol (50 mL) over 10 minutes under constant stirring at room temperature. After reducing the solvent to about 10 mL, diethyl ether was added dropwise. The precipitate thus obtained was filtered and washed with diethyl ether, providing a brownish product (0.3902 g, 92.7 % yield). Single crystal suitable for X-ray analysis was grown from the slow evaporation of the complex (107 mg) dissolved in water-methanol (10 mL, v/v, 1:1) after three weeks.

S5. Refinement details

The H atoms bonded to carbons were placed in calculated positions and treated as riding, C—H = 0.95 to 0.99 Å, with $U_{iso}(H) = 1.2U_{eq}(C)$. H atoms bonded to O were restrained to have similar O—H lengths and H—O—H angles.



Figure 1

Asymmetric unit of the title compound showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.



Figure 2

Crystal packing of the title compound viewed along the *a* axis showing hydrogen-bonding interactions as dashed lines.



Figure 3

Part of the crystal structure of compound (I) showing the π - π -stacking between fluorophenyl rings and the central pyridine ring.

Bis(bis{(*E*)-[(6-{(*E*)-[(4-fluorobenzyl)imino]methyl}pyridin-2-yl)methylidene](4-fluorophenyl)amine}nickel(II)) tetrabromide nonahydrate

Crystal data

Data collection

Bruker APEX CCD diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Krause *et al.*, 2015) $T_{\min} = 0.571$, $T_{\max} = 0.760$ 94344 measured reflections

Refinement

Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.042$ $wR(F^2) = 0.106$ S = 1.0119210 reflections Z = 2 F(000) = 2028 $D_x = 1.573 \text{ Mg m}^{-3}$ Mo K α radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 9955 reflections $\theta = 2.2-26.8^{\circ}$ $\mu = 2.42 \text{ mm}^{-1}$ T = 100 K Block, orange $0.26 \times 0.20 \times 0.12 \text{ mm}$

19210 independent reflections 15090 reflections with $I > 2\sigma(I)$ $R_{int} = 0.058$ $\theta_{max} = 27.4^\circ, \ \theta_{min} = 1.4^\circ$ $h = -15 \rightarrow 15$ $k = -16 \rightarrow 16$ $l = -37 \rightarrow 37$

1126 parameters27 restraintsPrimary atom site location: structure-invariant direct methodsSecondary atom site location: difference Fourier map

Hydrogen site location: difference Fourier map H atoms treated by a mixture of independent and constrained refinement
$$\begin{split} &w = 1/[\sigma^2(F_o^2) + (0.050P)^2 + 5.P] \\ &\text{where } P = (F_o^2 + 2F_c^2)/3 \\ &(\Delta/\sigma)_{\text{max}} = 0.001 \\ &\Delta\rho_{\text{max}} = 0.84 \text{ e } \text{\AA}^{-3} \\ &\Delta\rho_{\text{min}} = -0.77 \text{ e } \text{\AA}^{-3} \end{split}$$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å	2)
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	x	у	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
Ni1	0.76325 (3)	0.25409 (3)	0.36931 (2)	0.01149 (8)	
F1	0.58023 (17)	0.02363 (18)	0.19175 (7)	0.0338 (5)	
F2	0.3691 (2)	0.14418 (19)	0.51716 (9)	0.0509 (7)	
F3	0.5998 (2)	0.81422 (17)	0.36303 (9)	0.0558 (7)	
F4	1.32720 (16)	-0.10432 (16)	0.40078 (6)	0.0279 (4)	
N1	0.8897 (2)	0.32331 (19)	0.37240 (8)	0.0138 (5)	
N2	0.9073 (2)	0.13752 (19)	0.33378 (8)	0.0137 (5)	
N3	0.6810(2)	0.4043 (2)	0.40683 (8)	0.0160 (5)	
N4	0.6434 (2)	0.17671 (18)	0.36630 (8)	0.0123 (5)	
N5	0.6780 (2)	0.32319 (19)	0.30703 (8)	0.0138 (5)	
N6	0.7917 (2)	0.14722 (19)	0.42964 (8)	0.0137 (5)	
C1	0.9936 (2)	0.2743 (2)	0.35249 (9)	0.0141 (6)	
C2	1.0856 (3)	0.3181 (3)	0.35519 (10)	0.0204 (6)	
H2	1.1588	0.2832	0.3405	0.025*	
C3	1.0676 (3)	0.4144 (3)	0.37988 (11)	0.0240 (7)	
Н3	1.1291	0.4459	0.3825	0.029*	
C4	0.9591 (3)	0.4647 (3)	0.40086 (10)	0.0212 (7)	
H4	0.9457	0.5303	0.4180	0.025*	
C5	0.8714 (3)	0.4170 (2)	0.39614 (9)	0.0161 (6)	
C6	0.9988 (2)	0.1689 (2)	0.33110 (9)	0.0156 (6)	
H6	1.0679	0.1255	0.3158	0.019*	
C7	0.9170 (3)	0.0262 (2)	0.31722 (10)	0.0178 (6)	
H71	0.9099	-0.0234	0.3438	0.021*	
H72	0.9949	-0.0054	0.3023	0.021*	
C8	0.8260 (3)	0.0279 (2)	0.28343 (10)	0.0173 (6)	
C9	0.7620(3)	-0.0470 (2)	0.28993 (10)	0.0198 (6)	
H9	0.7743	-0.0964	0.3161	0.024*	
C10	0.6802 (3)	-0.0508 (3)	0.25858 (11)	0.0230 (7)	
H10	0.6382	-0.1036	0.2624	0.028*	
C11	0.6625 (3)	0.0246 (3)	0.22210 (11)	0.0234 (7)	
C12	0.7240 (3)	0.1001 (3)	0.21409 (11)	0.0235 (7)	
H12	0.7102	0.1502	0.1881	0.028*	
C13	0.8074 (3)	0.1005 (3)	0.24532 (10)	0.0194 (6)	
H13	0.8518	0.1509	0.2405	0.023*	

C14	0.7500 (3)	0.4593 (2)	0.41423 (10)	0.0171 (6)
H14	0.7245	0.5261	0.4310	0.021*
C15	0.5586 (3)	0.4513 (3)	0.42221 (12)	0.0236(7)
H151	0.5114	0.4798	0.3950	0.028*
H152	0.5512	0.5154	0.4419	0.028*
C16	0.5110 (3)	0.3684(2)	0.44844 (10)	0.0178 (6)
C17	0.5573 (3)	0.3180(3)	0.48937 (11)	0.0273(7)
H17	0.6217	0 3343	0 5013	0.033*
C18	0.5076(3)	0.2419(3)	0.51324 (11)	0.0317 (8)
H18	0.5376	0.2065	0 5414	0.038*
C19	0.4154(3)	0.2000	0.49476(13)	0.0321 (8)
C20	0.4154(3) 0.3667(3)	0.2200(3)	0.45514(12)	0.0321(0) 0.0278(7)
U20 Н20	0.3010	0.2544	0.439	0.0278(7)
C21	0.3010 0.4167(3)	0.2344 0.3431 (2)	0.43185 (11)	0.033 0.0212(7)
U21	0.3856	0.3771	0.4037	0.0212(7)
C^{22}	0.3830 0.5732(2)	0.3771 0.1085 (2)	0.4037 0.33003 (10)	0.025°
C22	0.3732(2) 0.4800(3)	0.1903(2) 0.1427(2)	0.33093(10) 0.32700(11)	0.0141(0)
C25	0.4899 (3)	0.1457 (2)	0.32709 (11)	0.0185 (0)
H23	0.4397	0.1399	0.3020	0.022*
C24	0.4827 (3)	0.0647 (2)	0.36099 (11)	0.0202 (6)
H24	0.4266	0.0262	0.3594	0.024*
025	0.5571 (3)	0.0417(2)	0.39727 (10)	0.01//(6)
H25	0.5531	-0.0130	0.4204	0.021*
C26	0.6372 (2)	0.1000 (2)	0.39906 (10)	0.0150 (6)
C27	0.5976 (2)	0.2817 (2)	0.29794 (10)	0.0150 (6)
H27	0.5551	0.3038	0.2708	0.018*
C28	0.7039 (3)	0.4057 (2)	0.27429 (10)	0.0181 (6)
H281	0.6571	0.4132	0.2468	0.022*
H282	0.7870	0.3799	0.2643	0.022*
C29	0.6768 (3)	0.5175 (2)	0.29632 (10)	0.0167 (6)
C30	0.7634 (3)	0.5661 (3)	0.30277 (12)	0.0264 (7)
H30	0.8406	0.5305	0.2918	0.032*
C31	0.7377 (3)	0.6665 (3)	0.32519 (14)	0.0367 (9)
H31	0.7965	0.7004	0.3296	0.044*
C32	0.6252 (4)	0.7157 (3)	0.34075 (13)	0.0342 (9)
C33	0.5371 (3)	0.6716 (3)	0.33478 (11)	0.0253 (7)
H33	0.4601	0.7085	0.3455	0.030*
C34	0.5638 (3)	0.5711 (2)	0.31246 (10)	0.0183 (6)
H34	0.5042	0.5383	0.3081	0.022*
C35	0.7228 (3)	0.0878 (2)	0.43438 (10)	0.0165 (6)
H35	0.7268	0.0373	0.4599	0.020*
C36	0.8771 (2)	0.1368 (2)	0.46492 (10)	0.0164 (6)
H361	0.8563	0.0967	0.4925	0.020*
H362	0.8755	0.2122	0.4741	0.020*
C37	0.9982 (2)	0.0740 (2)	0.44699 (9)	0.0138 (6)
C38	1.0194 (3)	-0.0315 (2)	0.42838 (9)	0.0152 (6)
H38	0.9572	-0.0624	0.4267	0.018*
C39	1.1298 (3)	-0.0916 (2)	0.41240 (9)	0.0168 (6)
H39	1.1438	-0.1625	0.3993	0.020*

C40	1.2184 (3)	-0.0457 (3)	0.41611 (10)	0.0192 (6)
C41	1.2020 (3)	0.0561 (3)	0.43528 (11)	0.0228 (7)
H41	1.2656	0.0846	0.4380	0.027*
C42	1.0908 (3)	0.1167 (3)	0.45058 (10)	0.0195 (6)
H42	1.0780	0.1877	0.4635	0.023*
Ni2	0.77765 (3)	0.25297 (3)	0.87380 (2)	0.01204 (8)
F5	0.4807 (2)	0.79732 (17)	0.89071 (8)	0.0506 (7)
F6	1.3604 (2)	0.2490 (3)	0.88290 (9)	0.0621 (8)
F7	0.73668 (15)	-0.01445 (15)	0.68867 (6)	0.0238 (4)
F8	0.6187 (2)	-0.0641 (2)	1.03512 (8)	0.0489 (6)
N7	0.7323 (2)	0.1156 (2)	0.86789 (8)	0.0152 (5)
N8	0.6370 (2)	0.2721 (2)	0.92506 (8)	0.0157(5)
N9	0.8962(2)	0.1667(2)	0.82011 (8)	0.0179 (5)
N10	0.8291(2)	0.3870(2)	0.87726 (8)	0.0145(5)
N11	0.9127(2)	0.2000(2)	0.92119(8)	0.0153(5)
N12	0.6656(2)	0.3705 (2)	0.82711(8)	0.0148 (5)
C43	0.6403(3)	0.1026(2)	0.89251(10)	0.0169(6)
C44	0.6034(3)	0.0096 (3)	0.89291(10) 0.88746(11)	0.0242(7)
H44	0.5364	0.0018	0.9040	0.029*
C45	0.6672 (3)	-0.0719(3)	0.85746 (12)	0.0296 (8)
H45	0.6438	-0.1365	0.8536	0.036*
C46	0.7639(3)	-0.0596(3)	0.83341(12)	0.0278(7)
H46	0.8094	-0.1161	0.8137	0.033*
C47	0.7929 (3)	0.0383(2)	0.83883(10)	0.0189(6)
C48	0.7925(3)	0.0303(2) 0.1932(2)	0.03003(10) 0.92464(10)	0.0107(6)
H48	0.5258	0.1932 (2)	0.9445	0.021*
C49	0.5250	0.1552 0.3555 (2)	0.96102 (10)	0.021
H491	0.6590	0.3444	0.9832	0.0210(7)
H492	0.5282	0.3439	0.9779	0.026*
C50	0.5262	0.5439 0.4734 (2)	0.94172(10)	0.026
C51	0.3660(3) 0.4689(3)	0.1731(2) 0.5170(3)	0.91533(11)	0.0243(7)
U51 H51	0.4218	0.4715	0.9092	0.0245 (7)
C52	0.4210 0.4401 (3)	0.4715	0.9092 0.80788 (12)	0.029 0.0319(8)
U52 Н52	0.3744	0.6558	0.8796	0.0317 (0)
C53	0.5096 (3)	0.6896 (3)	0.8790 0.90792 (12)	0.033
C54	0.5090(3) 0.6044(3)	0.6514(3)	0.93415(13)	0.0317(0)
С54 Н54	0.6498	0.6982	0.9407	0.037*
C55	0.6325 (3)	0.5902	0.95092 (11)	0.037 0.0240(7)
Н55	0.6987	0.5135	0.9691	0.0246 (7)
C56	0.8857(3)	0.0708 (3)	0.81292 (10)	0.029 0.0208 (7)
U50 H56	0.0057 (5)	0.0222	0.7916	0.0258
C57	0.9302	0.0222 0.2055(3)	0.79100 (10)	0.023 0.0240(7)
H571	0.9394	0.2783	0.7762	0.029*
H572	1.0122	0.1515	0.7664	0.029*
C58	1 0792 (3)	0.2184(3)	0.81762 (11)	0.025
C59	1 1418 (3)	0.1330(3)	0.84669 (11)	0.0200(7) 0.0301(8)
Н59	1 1198	0.0667	0.8511	0.036*
C60	1 2355 (3)	0 1433 (4)	0.86923 (12)	0.0381 (0)
000	1.4000 (0)	(-)	0.00723 (12)	0.0301(7)

H60	1.2777	0.0854	0.8894	0.046*
C61	1.2655 (3)	0.2390 (4)	0.86174 (13)	0.0415 (10)
C62	1.2077 (3)	0.3257 (4)	0.83371 (14)	0.0426 (10)
H62	1.2314	0.3910	0.8294	0.051*
C63	1.1116 (3)	0.3148 (3)	0.81146 (12)	0.0342 (8)
H63	1.0686	0.3740	0.7921	0.041*
C64	0.9163 (3)	0.3837 (3)	0.90442 (9)	0.0166 (6)
C65	0.9589 (3)	0.4741 (3)	0.90627 (10)	0.0224 (7)
H65	1.0196	0.4721	0.9259	0.027*
C66	0.9101 (3)	0.5674 (3)	0.87855 (11)	0.0266 (7)
H66	0.9381	0.6303	0.8791	0.032*
C67	0.8210 (3)	0.5703 (3)	0.85002 (10)	0.0234 (7)
H67	0.7880	0.6339	0.8308	0.028*
C68	0.7815 (3)	0.4771 (2)	0.85045 (9)	0.0168 (6)
C69	0.6881 (3)	0.4634 (2)	0.82323 (10)	0.0172 (6)
H69	0.6457	0.5220	0.8034	0.021*
C70	0.5697 (3)	0.3566 (2)	0.80145 (10)	0.0179 (6)
H701	0.5356	0.4249	0.7824	0.022*
H702	0.5083	0.3456	0.8233	0.022*
C71	0.6133 (3)	0.2571 (2)	0.77086 (10)	0.0162 (6)
C72	0.5662 (3)	0.1685 (3)	0.77556 (10)	0.0189 (6)
H721	0.5052	0.1712	0.7978	0.023*
C73	0.6080 (3)	0.0763 (3)	0.74791 (10)	0.0194 (6)
H73	0.5772	0.0152	0.7512	0.023*
C74	0.6948 (3)	0.0763 (2)	0.71574 (10)	0.0173 (6)
C75	0.7429 (3)	0.1627 (3)	0.70934 (10)	0.0177 (6)
H75	0.8029	0.1599	0.6866	0.021*
C76	0.7001 (3)	0.2542 (3)	0.73748 (10)	0.0171 (6)
H76	0.7308	0.3153	0.7337	0.021*
C77	0.9603 (3)	0.2750 (3)	0.92889 (10)	0.0174 (6)
H77	1.0215	0.2612	0.9496	0.021*
C78	0.9627 (3)	0.0873 (2)	0.94129 (10)	0.0196 (6)
H781	1.0236	0.0882	0.9625	0.024*
H782	0.9996	0.0353	0.9165	0.024*
C79	0.8714 (3)	0.0458 (2)	0.96719 (10)	0.0181 (6)
C80	0.8033 (3)	0.1073 (3)	1.00239 (11)	0.0230 (7)
H80	0.8154	0.1750	1.0107	0.028*
C81	0.7179 (3)	0.0710 (3)	1.02553 (12)	0.0288 (8)
H81	0.6698	0.1135	1.0492	0.035*
C82	0.7048 (3)	-0.0294(3)	1.01308 (12)	0.0316 (8)
C83	0.7727(3)	-0.0940(3)	0.98003(12)	0.0315 (8)
H83	0.7628	-0.1635	0.9731	0.038*
C84	0.8569 (3)	-0.0558(3)	0.95670 (11)	0.0251 (7)
H84	0.9050	-0.0994	0.9333	0.030*
Br1	0 32248 (3)	0 43862 (2)	0 30040 (2)	0.02087(7)
Br2	0 16973 (3)	0 57593 (3)	0.30010(2) 0.47672(2)	0.02491(8)
Br3	0.01026 (3)	0 28738 (3)	0.22480(2)	0.02845(8)
Br4	0.87440(3)	0.36772 (3)	0.22400(2) 0.04185(2)	0.02045 (0)
	0.07	0.30772 (3)	0.07103 (2)	0.02705 (0)

01	0.8522 (2)	0.6364 (2)	-0.00019 (9)	0.0345 (6)
H111	0.923 (2)	0.625 (3)	-0.0133 (13)	0.041*
H112	0.848 (3)	0.571 (2)	0.0100 (13)	0.041*
O2	0.2986 (2)	0.2342 (2)	0.23730 (9)	0.0330 (6)
H211	0.2223 (17)	0.244 (3)	0.2383 (13)	0.040*
H212	0.303 (3)	0.289(3)	0.2538 (12)	0.040*
03	0.2289 (2)	0.6485 (2)	0.37117 (8)	0.0335 (6)
H311	0.220 (3)	0.619 (3)	0.3994 (7)	0.040*
H312	0.265 (3)	0.586 (2)	0.3565 (11)	0.040*
04	0.0575 (3)	0.5247 (3)	0.26652 (10)	0.0474 (7)
H411	0.038 (4)	0.595 (2)	0.2806 (14)	0.057*
H412	0.133 (2)	0.499 (3)	0.2770 (15)	0.057*
05	0.0106 (3)	0.7161 (3)	0.33174 (11)	0.0504 (8)
H511	0.082 (2)	0.698 (4)	0.3419 (15)	0.060*
H512	-0.026 (3)	0.769 (3)	0.3519 (13)	0.060*
O6	0.8831 (2)	0.6656 (2)	0.48592 (9)	0.0317 (6)
H611	0.860 (3)	0.613 (3)	0.4990 (13)	0.038*
H612	0.9584 (16)	0.644 (3)	0.4858 (13)	0.038*
O7	0.8465 (2)	0.8179 (2)	0.40539 (10)	0.0372 (6)
H711	0.850 (3)	0.773 (3)	0.4306 (10)	0.045*
H712	0.778 (2)	0.814 (3)	0.3994 (13)	0.045*
08	0.9354 (3)	0.1547 (3)	0.13081 (12)	0.0658 (10)
H811	0.893 (4)	0.221 (3)	0.1447 (15)	0.079*
H812	0.908 (4)	0.194 (4)	0.1034 (10)	0.079*
09	0.7736 (3)	0.4144 (4)	0.15719 (13)	0.0771 (11)
H911	0.843 (3)	0.379 (5)	0.1739 (15)	0.092*
H912	0.809 (4)	0.396 (5)	0.1269 (9)	0.092*

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ni1	0.01170 (18)	0.00862 (17)	0.01394 (17)	-0.00256 (14)	-0.00089 (13)	-0.00129 (13)
F1	0.0307 (11)	0.0424 (13)	0.0317 (11)	-0.0136 (10)	-0.0155 (9)	-0.0017 (9)
F2	0.0433 (14)	0.0392 (13)	0.0659 (16)	-0.0118 (11)	0.0065 (12)	0.0281 (12)
F3	0.0698 (17)	0.0147 (11)	0.0826 (18)	-0.0059 (11)	-0.0303 (14)	-0.0155 (11)
F4	0.0182 (9)	0.0332 (11)	0.0268 (10)	-0.0003 (8)	0.0022 (8)	0.0007 (8)
N1	0.0159 (12)	0.0123 (12)	0.0131 (11)	-0.0044 (10)	-0.0006 (9)	0.0007 (9)
N2	0.0167 (12)	0.0106 (12)	0.0119 (11)	-0.0005 (10)	-0.0024 (9)	-0.0021 (9)
N3	0.0165 (12)	0.0129 (12)	0.0176 (12)	-0.0037 (10)	0.0039 (10)	-0.0005 (10)
N4	0.0123 (12)	0.0072 (11)	0.0153 (11)	0.0003 (9)	-0.0001 (9)	-0.0022 (9)
N5	0.0134 (12)	0.0082 (11)	0.0174 (12)	0.0002 (9)	-0.0007 (9)	0.0001 (9)
N6	0.0155 (12)	0.0103 (11)	0.0127 (11)	0.0002 (10)	-0.0017 (9)	-0.0022 (9)
C1	0.0150 (14)	0.0158 (14)	0.0113 (13)	-0.0044 (12)	-0.0014 (11)	0.0012 (11)
C2	0.0153 (15)	0.0273 (17)	0.0203 (15)	-0.0093 (13)	-0.0007 (12)	0.0031 (13)
C3	0.0258 (17)	0.0297 (18)	0.0240 (16)	-0.0198 (15)	-0.0032 (13)	0.0020 (13)
C4	0.0311 (18)	0.0187 (16)	0.0185 (15)	-0.0141 (14)	-0.0031 (13)	-0.0020 (12)
C5	0.0224 (16)	0.0141 (14)	0.0141 (13)	-0.0087 (12)	-0.0028 (11)	0.0011 (11)
C6	0.0127 (14)	0.0183 (15)	0.0131 (13)	-0.0008 (12)	0.0001 (11)	-0.0009 (11)

C7	0.0170 (15)	0.0112 (14)	0.0235 (15)	-0.0006 (12)	-0.0035 (12)	-0.0045 (12)
C8	0.0181 (15)	0.0122 (14)	0.0187 (14)	0.0009 (12)	-0.0018 (12)	-0.0073 (11)
C9	0.0206 (16)	0.0150 (15)	0.0216 (15)	-0.0020 (12)	0.0005 (12)	-0.0033 (12)
C10	0.0230 (17)	0.0188 (16)	0.0283 (17)	-0.0073 (13)	-0.0011 (13)	-0.0060 (13)
C11	0.0217 (17)	0.0265 (17)	0.0222 (16)	-0.0056 (14)	-0.0059 (13)	-0.0088 (13)
C12	0.0284 (18)	0.0209 (16)	0.0187 (15)	-0.0029 (14)	-0.0037 (13)	-0.0008 (12)
C13	0.0218 (16)	0.0161 (15)	0.0199 (15)	-0.0044 (13)	-0.0007 (12)	-0.0055 (12)
C14	0.0248 (16)	0.0099 (14)	0.0159 (14)	-0.0045 (12)	0.0031 (12)	-0.0022 (11)
C15	0.0173 (16)	0.0132 (15)	0.0367 (18)	-0.0009 (12)	0.0081 (13)	0.0005 (13)
C16	0.0192 (15)	0.0117 (14)	0.0185 (14)	-0.0001 (12)	0.0078 (12)	-0.0015 (11)
C17	0.0246 (17)	0.0287 (18)	0.0219 (16)	0.0018 (14)	0.0018 (13)	-0.0033 (14)
C18	0.036 (2)	0.0286 (19)	0.0198 (16)	0.0048 (16)	-0.0014 (14)	0.0073 (14)
C19	0.0271 (19)	0.0267 (19)	0.038 (2)	-0.0040 (15)	0.0097 (15)	0.0078 (15)
C20	0.0218 (17)	0.0196 (17)	0.040 (2)	-0.0043 (14)	0.0019 (15)	-0.0006 (14)
C21	0.0226 (16)	0.0136 (15)	0.0244 (16)	-0.0015 (13)	0.0033 (13)	-0.0009 (12)
C22	0.0127 (14)	0.0096 (13)	0.0176 (14)	0.0004 (11)	-0.0014 (11)	-0.0023 (11)
C23	0.0148 (15)	0.0140 (15)	0.0253 (16)	-0.0021 (12)	-0.0043 (12)	-0.0048 (12)
C24	0.0164 (15)	0.0137 (15)	0.0316 (17)	-0.0059 (12)	0.0002 (13)	-0.0033 (12)
C25	0.0186 (15)	0.0125 (14)	0.0211 (15)	-0.0038 (12)	0.0025 (12)	-0.0008 (11)
C26	0.0130 (14)	0.0119 (14)	0.0174 (14)	0.0001 (11)	0.0036 (11)	-0.0032 (11)
C27	0.0129 (14)	0.0124 (14)	0.0164 (14)	0.0016 (11)	-0.0013 (11)	-0.0016 (11)
C28	0.0209 (16)	0.0157 (15)	0.0167 (14)	-0.0048 (12)	0.0003 (12)	0.0039 (11)
C29	0.0210 (15)	0.0105 (14)	0.0188 (14)	-0.0051 (12)	-0.0048 (12)	0.0054 (11)
C30	0.0209 (17)	0.0217 (17)	0.0383 (19)	-0.0092 (14)	-0.0071 (14)	0.0092 (14)
C31	0.042 (2)	0.0210 (18)	0.056 (2)	-0.0219 (17)	-0.0220 (19)	0.0104 (17)
C32	0.048 (2)	0.0114 (16)	0.042 (2)	-0.0040 (16)	-0.0206 (18)	-0.0002 (14)
C33	0.0315 (19)	0.0120 (15)	0.0283 (17)	0.0008 (13)	-0.0112 (14)	0.0008 (13)
C34	0.0208 (16)	0.0129 (14)	0.0221 (15)	-0.0057 (12)	-0.0081 (12)	0.0045 (12)
C35	0.0186 (15)	0.0130 (14)	0.0163 (14)	-0.0031 (12)	0.0038 (11)	-0.0003 (11)
C36	0.0176 (15)	0.0143 (14)	0.0145 (14)	-0.0001 (12)	-0.0026 (11)	-0.0020 (11)
C37	0.0164 (14)	0.0140 (14)	0.0101 (12)	-0.0028 (12)	-0.0042 (11)	0.0018 (10)
C38	0.0179 (15)	0.0128 (14)	0.0158 (13)	-0.0057 (12)	-0.0040 (11)	0.0013 (11)
C39	0.0211 (15)	0.0138 (14)	0.0138 (13)	-0.0026 (12)	-0.0015 (11)	0.0000 (11)
C40	0.0172 (15)	0.0222 (16)	0.0148 (14)	-0.0018 (13)	0.0010 (11)	0.0037 (12)
C41	0.0211 (16)	0.0272 (17)	0.0245 (16)	-0.0139 (14)	-0.0046 (13)	0.0037 (13)
C42	0.0234 (16)	0.0152 (15)	0.0211 (15)	-0.0073 (13)	-0.0033 (12)	0.0000 (12)
Ni2	0.01326 (18)	0.01057 (17)	0.01248 (17)	-0.00355 (14)	-0.00116 (13)	-0.00184 (13)
F5	0.0703 (17)	0.0144 (10)	0.0599 (15)	-0.0068 (11)	0.0186 (13)	0.0064 (10)
F6	0.0318 (13)	0.113 (2)	0.0501 (15)	-0.0329 (15)	-0.0135 (11)	0.0024 (15)
F7	0.0231 (10)	0.0191 (9)	0.0266 (10)	-0.0016 (8)	-0.0026 (8)	-0.0071 (7)
F8	0.0427 (14)	0.0562 (15)	0.0556 (15)	-0.0294 (12)	-0.0054 (11)	0.0227 (12)
N7	0.0141 (12)	0.0123 (12)	0.0183 (12)	-0.0018 (10)	-0.0053 (10)	-0.0020 (9)
N8	0.0166 (12)	0.0126 (12)	0.0154 (12)	-0.0008 (10)	-0.0015 (10)	0.0001 (9)
N9	0.0142 (12)	0.0225 (14)	0.0146 (12)	-0.0017 (11)	-0.0017 (10)	-0.0032 (10)
N10	0.0178 (13)	0.0165 (12)	0.0101 (11)	-0.0068 (10)	0.0024 (9)	-0.0023 (9)
N11	0.0143 (12)	0.0162 (13)	0.0148 (12)	-0.0034 (10)	-0.0009 (9)	-0.0016 (9)
N12	0.0168 (13)	0.0139 (12)	0.0123 (11)	-0.0022 (10)	0.0001 (9)	-0.0031 (9)
C43	0.0166 (15)	0.0129 (14)	0.0218 (15)	-0.0046 (12)	-0.0073 (12)	0.0021 (11)

C44	0.0253 (17)	0.0190 (16)	0.0309 (17)	-0.0101 (14)	-0.0088 (14)	0.0048 (13)
C45	0.037 (2)	0.0155 (16)	0.040 (2)	-0.0112 (15)	-0.0162 (16)	-0.0007 (14)
C46	0.0306 (19)	0.0171 (16)	0.0339 (18)	-0.0017 (14)	-0.0125 (15)	-0.0081 (14)
C47	0.0191 (15)	0.0132 (14)	0.0215 (15)	0.0012 (12)	-0.0078 (12)	-0.0047 (12)
C48	0.0153 (15)	0.0179 (15)	0.0196 (15)	-0.0051 (12)	-0.0013 (12)	0.0050 (12)
C49	0.0274 (17)	0.0157 (15)	0.0173 (15)	-0.0001 (13)	0.0019 (13)	-0.0029 (12)
C50	0.0178 (15)	0.0146 (14)	0.0158 (14)	-0.0028(12)	0.0043 (11)	-0.0040 (11)
C51	0.0260 (17)	0.0186 (16)	0.0302 (17)	-0.0087(14)	-0.0063 (14)	-0.0015 (13)
C52	0.033 (2)	0.0245 (18)	0.0323 (19)	0.0010 (15)	-0.0064(15)	0.0031 (15)
C53	0.044 (2)	0.0145 (16)	0.0324 (19)	-0.0056(15)	0.0144 (16)	0.0002 (14)
C54	0.0315(19)	0.0246 (18)	0.041 (2)	-0.0162(16)	0.0143 (16)	-0.0119(15)
C55	0.0178 (16)	0.0279(18)	0.0256(16)	-0.0051(14)	0.0042(13)	-0.0110(14)
C56	0.0169(15)	0.0271(16)	0.0187(15)	0 0034 (13)	-0.0047(12)	-0.0103(12)
C57	0.0216 (16)	0.0331(19)	0.0155(14)	-0.0058(14)	0.0030(12)	-0.0029(13)
C58	0.0190(16)	0.037(2)	0.0186 (15)	-0.0063(15)	0.0056(12)	-0.0068(14)
C59	0.0209(17)	0.037(2) 0.041(2)	0.0236(17)	-0.0030(15)	0.0039(13)	-0.0025(15)
C60	0.0209(17) 0.0192(18)	0.063(3)	0.0250(17) 0.0251(18)	-0.0037(18)	-0.0006(14)	0.0022(12)
C61	0.0192(18)	0.005(3)	0.0291(10) 0.0293(19)	-0.014(2)	-0.0049(15)	-0.0025(19)
C62	0.0200(10)	0.075(3)	0.0293(19)	-0.022(2)	0.0016(18)	-0.001(2)
C63	0.025(2)	0.037(3)	0.0321(19)	-0.0022(2)	-0.0002(15)	0.001(2)
C64	0.0233(15) 0.0181(15)	0.013(2) 0.0224(16)	0.0321(13)	-0.0100(13)	0.0002(13)	-0.0040(11)
C65	0.0266(17)	0.0221(10) 0.0274(17)	0.0120(15) 0.0193(15)	-0.0175(14)	0.0018 (13)	-0.0042(13)
C66	0.0200(17)	0.0271(17) 0.0232(17)	0.0193(12) 0.0261(17)	-0.0221(16)	0.0010(15)	-0.0023(13)
C67	0.035(2)	0.0202(17) 0.0207(16)	0.0201(17) 0.0173(15)	-0.0134(14)	0.0036(13)	0.0023(12)
C68	0.0239(16)	0.0207(10)	0.0175(13)	-0.0083(13)	0.0038(11)	-0.0013(11)
C69	0.0196(15)	0.0170(15)	0.0135(13)	-0.0005(12)	0.0008(11)	0.0004 (11)
C70	0.0170(15)	0.0120(15) 0.0176(15)	0.0168(14)	-0.0013(12)	-0.0052(12)	-0.0026(11)
C71	0.0160(14)	0.0174 (15)	0.0142(13)	-0.0030(12)	-0.0063(11)	0.0012 (11)
C72	0.0171 (15)	0.0241 (16)	0.0167(14)	-0.0077(13)	-0.0005(12)	-0.0019(12)
C73	0.0207(16)	0.0199(16)	0.0206 (15)	-0.0096(13)	-0.0053(12)	-0.0003(12)
C74	0.0169(15)	0.0150 (14)	0.0175(14)	0.0003(12)	-0.0063(11)	-0.0038(11)
C75	0.0169(15)	0.0215 (16)	0.0172(11) 0.0144(14)	-0.0045(12)	-0.0016(11)	0.0005(12)
C76	0.0184(15)	0.0212(10) 0.0182(15)	0.0154(14)	-0.0062(12)	-0.0055(11)	0.0000(12) 0.0040(11)
C77	0.0153(15)	0.0102(10)	0.0135(13)	-0.0060(13)	-0.0002(11)	-0.0034(12)
C78	0.0133(15) 0.0174(15)	0.0250(10) 0.0161(15)	0.0135(15) 0.0215(15)	0.0000(13)	-0.0049(12)	0.0000(12)
C79	0.0171(15) 0.0188(15)	0.0134(14)	0.0215(15) 0.0196(15)	-0.0008(12)	-0.0064(12)	0.0020(12) 0.0036(11)
C80	0.0180(15) 0.0282(18)	0.0154(14) 0.0151(15)	0.0190(15) 0.0242(16)	-0.0040(12)	-0.0067(13)	0.0030(11) 0.0043(12)
C81	0.0282(18)	0.0131(13) 0.0241(18)	0.0212(10) 0.0284(18)	-0.0008(15)	0.0007(13)	0.0013(12) 0.0072(14)
C82	0.0200(10) 0.0291(19)	0.0211(10) 0.035(2)	0.0201(10) 0.0340(19)	-0.0171(16)	-0.0023(11)	0.0072(11)
C83	0.0291(19)	0.022(2)	0.0310(19) 0.0353(19)	-0.0183(16)	-0.0173(17)	0.0092(15)
C84	0.0327(19)	0.0210(16)	0.0333(19) 0.0243(16)	-0.0042(14)	-0.0109(14)	0.00002(10)
Br1	0.0327(19) 0.01836(15)	0.0170(10) 0.01624(15)	0.0249(10) 0.02801(16)	-0.00542(12)	0.0109(14) 0.00252(12)	-0.0017(13)
Br?	0.01050(15) 0.03089(18)	0.01024(13) 0.02165(17)	0.02301(10) 0.02241(16)	-0.00779(14)	-0.00133(13)	-0.00336(12)
Br3	0.03009(10) 0.02149(17)	0.02109(17) 0.03430(19)	0.02241(10) 0.02683(17)	-0.00630(14)	0.00133(13) 0.00480(13)	0.00590(12)
Br4	0.02946(18)	0.02484(17)	0.02000(17) 0.03397(18)	-0.01384(14)	-0.00543(14)	-0.00575(14)
01	0.0240(15)	0.0261(13)	0.0441(15)	-0.0103(12)	-0.0014(12)	0.0001(11)
02	0.0323(14)	0.0226(13)	0.0427(15)	-0.0036(11)	-0.0120(12)	-0.0086(11)
03	0.0502(17)	0.0197(12)	0.0278(13)	-0.0075(12)	0.0080(12)	-0.0044(10)
00	0.00002(17)	0.0127 (12)	0.02,0(10)	0.00,0 (12)	5.0000 (1 <i>2)</i>	0.0011(10)

O4	0.0369 (16)	0.0457 (18)	0.0515 (18)	0.0006 (14)	-0.0119 (14)	0.0031 (14)
O5	0.0388 (17)	0.054 (2)	0.0511 (19)	-0.0050 (15)	0.0007 (14)	0.0132 (15)
06	0.0307 (14)	0.0213 (13)	0.0418 (15)	-0.0063 (11)	0.0031 (12)	-0.0035 (11)
O7	0.0322 (15)	0.0289 (14)	0.0532 (17)	-0.0109 (12)	-0.0159 (13)	-0.0025 (12)
08	0.088 (3)	0.065 (2)	0.0453 (19)	-0.026 (2)	0.0054 (18)	0.0024 (16)
O9	0.070 (3)	0.100 (3)	0.059 (2)	-0.022 (2)	-0.0060 (19)	-0.002 (2)

Geometric parameters (Å, °)

Ni1—N4	1.984 (2)	F8—C82	1.366 (4)
Ni1—N1	1.984 (2)	N7—C47	1.333 (4)
Ni1—N6	2.141 (2)	N7—C43	1.339 (4)
Ni1—N5	2.151 (2)	N8—C48	1.280 (4)
Ni1—N2	2.158 (2)	N8—C49	1.474 (4)
Ni1—N3	2.165 (2)	N9—C56	1.277 (4)
F1—C11	1.369 (4)	N9—C57	1.469 (4)
F2—C19	1.378 (4)	N10—C68	1.340 (4)
F3—C32	1.368 (4)	N10—C64	1.342 (4)
F4—C40	1.362 (3)	N11—C77	1.278 (4)
N1—C1	1.335 (4)	N11—C78	1.469 (4)
N1—C5	1.347 (4)	N12—C69	1.276 (4)
N2—C6	1.281 (4)	N12—C70	1.471 (4)
N2—C7	1.469 (4)	C43—C44	1.388 (4)
N3—C14	1.267 (4)	C43—C48	1.469 (4)
N3—C15	1.472 (4)	C44—C45	1.393 (5)
N4—C22	1.336 (4)	C44—H44	0.9500
N4—C26	1.341 (4)	C45—C46	1.377 (5)
N5—C27	1.278 (4)	C45—H45	0.9500
N5C28	1.469 (3)	C46—C47	1.396 (4)
N6—C35	1.275 (4)	C46—H46	0.9500
N6—C36	1.470 (4)	C47—C56	1.470 (4)
C1—C2	1.390 (4)	C48—H48	0.9500
C1—C6	1.473 (4)	C49—C50	1.503 (4)
C2—C3	1.391 (5)	C49—H491	0.9900
C2—H2	0.9500	C49—H492	0.9900
C3—C4	1.394 (5)	C50—C55	1.390 (4)
С3—Н3	0.9500	C50—C51	1.393 (4)
C4—C5	1.384 (4)	C51—C52	1.387 (5)
C4—H4	0.9500	C51—H51	0.9500
C5—C14	1.479 (4)	C52—C53	1.374 (5)
С6—Н6	0.9500	С52—Н52	0.9500
C7—C8	1.511 (4)	C53—C54	1.363 (5)
C7—H71	0.9900	C54—C55	1.383 (5)
С7—Н72	0.9900	С54—Н54	0.9500
C8—C13	1.388 (4)	С55—Н55	0.9500
C8—C9	1.389 (4)	С56—Н56	0.9500
C9—C10	1.395 (4)	C57—C58	1.514 (5)
С9—Н9	0.9500	С57—Н571	0.9900

C10—C11	1.372 (5)	С57—Н572	0.9900
С10—Н10	0.9500	C58—C63	1.384 (5)
C11—C12	1.378 (5)	C58—C59	1.391 (5)
C12—C13	1.395 (4)	C59—C60	1.382 (5)
С12—Н12	0.9500	С59—Н59	0.9500
С13—Н13	0.9500	C60—C61	1,363 (6)
C14—H14	0.9500	C60—H60	0.9500
C15—C16	1,499 (4)	C61—C62	1.364 (6)
C15—H151	0.9900	C62—C63	1.405 (5)
C15—H152	0.9900	C62—H62	0.9500
C16-C17	1 387 (4)	C63—H63	0.9500
C16-C21	1 389 (4)	C64—C65	1 387 (4)
C17 - C18	1 415 (5)	C64 - C77	1.367(1) 1 472(4)
C17—H17	0.9500	C65-C66	1.172(1) 1 386(5)
C18 - C19	1 364 (5)	C65—H65	0.9500
C18—H18	0.9500	C66—C67	1.387(5)
C19-C20	1 359 (5)	C66H66	0.9500
$C_{19}^{}C_{20}^{}C_{21}^{}C_{20}^{}C_{21}^{}C_{20}^{}C_{21}^{}C_{20}^{}C_{21}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{}C_{20}^{$	1.339(3) 1.383(4)	C67 C68	1.302(4)
$C_{20} = C_{21}$	0.9500	C67 H67	0.0500
C21 H21	0.9500	C68 - C60	0.9300
C_{21} C_{22} C_{23}	0.9300 1 303 (4)	C60 H60	1.471(4)
$C_{22} = C_{23}$	1.393(4)	C70 C71	0.9300
$C_{22} = C_{24}$	1.474(4) 1.287(4)	C70 + U701	1.313(4)
$C_{23} = C_{24}$	1.387 (4)	C70_H701	0.9900
C23—H23	0.9500	C70—H702	0.9900
C24—C25	1.388 (4)	C/1 - C/6	1.387 (4)
C24—H24	0.9500	C/1—C/2	1.395 (4)
C25—C26	1.387 (4)	C/2—C/3	1.391 (4)
C25—H25	0.9500	С/2—Н/21	0.9500
C26—C35	1.469 (4)	C/3—C/4	1.370 (4)
С27—Н27	0.9500	С/3—Н/3	0.9500
C28—C29	1.509 (4)	C74—C75	1.380 (4)
C28—H281	0.9900	С75—С76	1.394 (4)
C28—H282	0.9900	С75—Н75	0.9500
C29—C30	1.387 (4)	С76—Н76	0.9500
C29—C34	1.395 (4)	С77—Н77	0.9500
C30—C31	1.391 (5)	C78—C79	1.512 (4)
С30—Н30	0.9500	С78—Н781	0.9900
C31—C32	1.375 (6)	С78—Н782	0.9900
C31—H31	0.9500	C79—C80	1.388 (4)
C32—C33	1.364 (5)	C79—C84	1.390 (4)
C33—C34	1.389 (4)	C80—C81	1.384 (5)
С33—Н33	0.9500	C80—H80	0.9500
С34—Н34	0.9500	C81—C82	1.385 (5)
С35—Н35	0.9500	C81—H81	0.9500
C36—C37	1.516 (4)	C82—C83	1.358 (5)
С36—Н361	0.9900	C83—C84	1.387 (5)
С36—Н362	0.9900	С83—Н83	0.9500
C37—C42	1.393 (4)	C84—H84	0.9500

C37—C38	1.400 (4)	O1—H111	0.893 (18)
C38—C39	1.388 (4)	O1—H112	0.878 (18)
C38—H38	0.9500	O2—H211	0.893 (18)
C39—C40	1.373 (4)	O2—H212	0.873 (18)
С39—Н39	0.9500	O3—H311	0.899 (18)
C40—C41	1.375 (5)	O3—H312	0.892 (18)
C41—C42	1.391 (4)	O4—H411	0.949(19)
C41—H41	0.9500	O4—H412	0.933 (19)
C42—H42	0.9500	05—H511	0.884 (19)
Ni2—N10	1 973 (2)	05—H512	0.896 (19)
Ni2—N7	1.973(2) 1.981(2)	06—H611	0.853(18)
Ni2—N11	2 125 (2)	O6—H612	0.868 (18)
Ni2N9	2.123(2) 2.158(2)	07—H711	0.899 (18)
Ni2—N12	2.150(2) 2.161(2)	07—H712	0.876 (18)
Ni2N8	2.161(2) 2.169(2)	0, -11, 12 08-H811	0.936 (19)
F5C53	1.369(4)	08—H812	0.930(19) 0.943(19)
F6_C61	1.309(4) 1.373(4)	00 H012	0.943 (19)
F7 C74	1.373(4) 1.367(3)	O_{9} H912	0.971(19)
r/—C/4	1.507 (5)	09—11912	0.907 (19)
N4Ni1N1	176 85 (10)	N11_Nj2_N9	89 68 (9)
N4—Ni1—N6	77 31 (9)	N10 - Ni2 - N12	77 22 (10)
N1—Ni1—N6	100.97 (9)	$N7_Ni2_N12$	102 95 (9)
N4—Ni1—N5	77.05.(9)	N11Ni2N12	154.80(9)
N1—Ni1—N5	104 62 (9)	N9 - Ni2 - N12	94 39 (9)
N6—Ni1—N5	154.36(9)	N10—Nj2—N8	105 89 (9)
N4—Ni1—N2	104.30(9) 100 20 (9)	N7—Ni2—N8	77 01 (10)
N1N1N2	77.09.(9)	N11Ni2N8	95 48 (9)
N6—Ni1—N2	90 34 (9)	N9—Ni2—N8	154 21 (10)
N5—Ni1—N2	93 75 (9)	N12—Ni2—N8	91 60 (9)
N4—Ni1—N3	105 70 (9)	C47 - N7 - C43	1211(3)
N1—Ni1—N3	76 99 (9)	C47—N7—Ni2	1193(2)
N6—Ni1—N3	94 22 (9)	C43 - N7 - Ni2	119.3(2) 119.7(2)
N5—Ni1—N3	93.06(9)	C48 - N8 - C49	119.7(2) 118.3(3)
N2N1N3	154 08 (9)	C48—N8—Ni2	110.5(3) 1124(2)
C1 - N1 - C5	1205(2)	C49—N8—Ni2	112.1(2) 129.1(2)
C1 - N1 - N1	119 77 (19)	$C_{56} N_{9} C_{57}$	129.1(2) 1184(3)
C_{5} N1 Ni1	119.69 (19)	$C_{56} N_{9} N_{12}$	110.1(3)
C6-N2-C7	117.6 (2)	C57 - N9 - Ni2	112.0(2) 128.9(2)
C6-N2-Ni1	112 71 (19)	C68 - N10 - C64	120.9(2) 121.1(3)
C7 - N2 - Ni1	129 34 (19)	C68—N10—Ni2	1196(2)
$C_{14} N_{3} C_{15}$	129.31(19) 117.8(3)	C64 - N10 - Ni2	119.0 (2)
C14—N3—Nil	117.8(3) 112.9(2)	C77 - N11 - C78	119.2(2) 118.5(3)
C15 N3 Ni1	112.9(2) 129.1(2)	C77—N11—Ni2	113.6(2)
C_{22} N4— C_{26}	129.1(2) 121.1(2)	C78—N11—Ni2	127 60 (19)
C22N4Ni1	119 76 (19)	C69 - N12 - C70	118 7 (2)
C26—N4—Ni1	119.12 (19)	C69 - N12 - Ni2	110.7(2) 112.6(2)
$C_{27} N_{5} C_{28}$	118.4 (2)	C70 - N12 - Ni2	128 62 (18)
C27—N5—Ni1	113.13 (19)	N7—C43—C44	120.9 (3)

C28—N5—Ni1	128.45 (19)	N7—C43—C48	112.5 (2)
C35—N6—C36	118.6 (2)	C44—C43—C48	126.6 (3)
C35—N6—Ni1	113.24 (19)	C43—C44—C45	118.2 (3)
C36—N6—Ni1	128.19 (18)	C43—C44—H44	120.9
N1—C1—C2	121.5 (3)	C45—C44—H44	120.9
N1—C1—C6	112.4 (2)	C46—C45—C44	120.5 (3)
C2—C1—C6	126.0 (3)	C46—C45—H45	119.7
C1 - C2 - C3	118 4 (3)	C44—C45—H45	119 7
C1 - C2 - H2	120.8	C45-C46-C47	1180(3)
$C_3 = C_2 = H_2$	120.8	$C_{45} - C_{46} - H_{46}$	121.0
$C_2 C_3 C_4$	110.8 (3)	C_{47} C_{46} H_{46}	121.0
$C_2 = C_3 = C_4$	120.1	N7 C47 C46	121.0 121.3(3)
$C_2 = C_3 = H_3$	120.1	N7 C47 C56	121.5(3)
$C_{4} = C_{3} = 115$	120.1	11 - 04 - 050	112.9(3)
$C_5 = C_4 = C_5$	110.3 (3)	10 - 10 - 10 - 10 - 10 - 10 - 10 - 10 -	123.7(3)
$C_3 = C_4 = H_4$	120.8	$N_{0} = C_{40} = C_{40}$	118.1 (3)
C3—C4—H4	120.8	N8-C48-H48	121.0
NI-C5-C4	121.4 (3)	C43—C48—H48	121.0
N1—C5—C14	111.8 (2)	N8—C49—C50	112.4 (2)
C4—C5—C14	126.8 (3)	N8—C49—H491	109.1
N2—C6—C1	118.0 (3)	C50—C49—H491	109.1
N2—C6—H6	121.0	N8—C49—H492	109.1
C1—C6—H6	121.0	С50—С49—Н492	109.1
N2—C7—C8	113.1 (2)	H491—C49—H492	107.8
N2—C7—H71	109.0	C55—C50—C51	118.6 (3)
С8—С7—Н71	109.0	C55—C50—C49	120.4 (3)
N2—C7—H72	109.0	C51—C50—C49	121.0 (3)
С8—С7—Н72	109.0	C52—C51—C50	120.9 (3)
H71—C7—H72	107.8	C52—C51—H51	119.6
C13—C8—C9	119.4 (3)	С50—С51—Н51	119.6
C13—C8—C7	121.3 (3)	C53—C52—C51	117.8 (3)
C9—C8—C7	119.3 (3)	С53—С52—Н52	121.1
C8-C9-C10	1210(3)	С51—С52—Н52	121.1
C8—C9—H9	119 5	C54 - C53 - F5	121.1 118.8(3)
C10-C9-H9	119.5	C_{54} C_{53} C_{52}	123.5(3)
$C_{11} - C_{10} - C_{9}$	117.6 (3)	F_{5} C_{53} C_{52}	123.3(3) 117.7(3)
$C_{11} C_{10} H_{10}$	121.2	C_{53}^{53} C_{54}^{55} C_{55}^{55}	117.7(3)
C_{10} C_{10} H_{10}	121.2	$C_{55} = C_{54} = C_{55}$	121.0
$E_{2} = C_{10} = 110$	121.2 119 1 (2)	$C_{55} = C_{54} = H_{54}$	121.0
F1 = C11 = C10	110.1(3) 1184(2)	$C_{55} - C_{54} - H_{54}$	121.0
FI = CII = CI2	110.4(3)	$C_{54} = C_{55} = C_{50}$	121.5 (5)
	123.5 (3)	С54—С55—Н55	119.3
C11—C12—C13	117.8 (3)	С50—С55—Н55	119.3
C11—C12—H12	121.1	N9-C56-C47	117.7 (3)
C13—C12—H12	121.1	N9—C56—H56	121.1
C8—C13—C12	120.7 (3)	C47—C56—H56	121.1
С8—С13—Н13	119.7	N9—C57—C58	112.8 (2)
C12—C13—H13	119.7	N9—C57—H571	109.0
N3—C14—C5	118.4 (3)	C58—C57—H571	109.0
N3—C14—H14	120.8	N9—C57—H572	109.0

C5—C14—H14	120.8	С58—С57—Н572	109.0
N3—C15—C16	113.2 (2)	Н571—С57—Н572	107.8
N3—C15—H151	108.9	C63—C58—C59	119.2 (3)
C16—C15—H151	108.9	C63—C58—C57	119.5 (3)
N3—C15—H152	108.9	C59—C58—C57	121.2 (3)
C16—C15—H152	108.9	C60—C59—C58	120.9 (4)
H151—C15—H152	107.7	С60—С59—Н59	119.6
C17—C16—C21	119.2 (3)	С58—С59—Н59	119.6
C17—C16—C15	121.7 (3)	C61—C60—C59	118.1 (4)
C21—C16—C15	119.2 (3)	С61—С60—Н60	121.0
C16—C17—C18	119.4 (3)	С59—С60—Н60	121.0
С16—С17—Н17	120.3	C60—C61—C62	123.9 (4)
С18—С17—Н17	120.3	C60—C61—F6	118.7 (4)
C19—C18—C17	118.2 (3)	C62—C61—F6	117.5 (4)
C19—C18—H18	120.9	C61—C62—C63	117.5 (4)
C17—C18—H18	120.9	C61—C62—H62	121.2
C20-C19-C18	124.1 (3)	С63—С62—Н62	121.2
C_{20} C_{19} F_{2}	117.6(3)	C_{58} — C_{63} — C_{62}	120.5(4)
C18 - C19 - F2	118 3 (3)	$C_{58} - C_{63} - H_{63}$	119.8
C19 - C20 - C21	1172(3)	C62 - C63 - H63	119.8
C19 - C20 - H20	121.4	N10-C64-C65	121.1(3)
C_{21} C_{20} H_{20}	121.1	N10-C64-C77	121.1(3) 1123(3)
C_{20} C_{21} C_{20} C_{120} C_{16}	121.4 122.0(3)	C_{65} C_{64} C_{77}	112.5(3) 126 5(3)
C_{20} C_{21} C_{10} C_{21} H_{21}	119.0	C66-C65-C64	120.5(3)
$C_{20} = C_{21} = H_{21}$	110.0	$C_{00} = C_{00} = C_{01}$	121.0
$N_{10} = C_{21} = M_{21}$	119.0	C64 C65 H65	121.0
N4 C22 C27	121.5(3) 112.3(2)	$C_{04} = C_{05} = 1105$	121.0 120.8(3)
$C_{22} = C_{27} = C_{27}$	112.3(2) 126.4(3)	C65 C66 H66	120.8 (3)
$C_{23} = C_{22} = C_{27}$	120.4(3)	C67 C66 H66	119.0
$C_{24} = C_{23} = C_{22}$	117.3 (3)	C66 C67 C68	119.0
$C_{24} = C_{23} = H_{23}$	121.1	$C_{00} = C_{07} = C_{08}$	110.0 (5)
$C_{22} = C_{23} = H_{23}$	121.1 120.2(2)	$C_{00} = C_{07} = H_{07}$	121.0
$C_{23} = C_{24} = C_{23}$	120.5 (5)	1000000000000000000000000000000000000	121.0 120.0(3)
$C_{23} = C_{24} = H_{24}$	119.8	N10 - C68 - C67	120.9(3)
$C_{23} = C_{24} = H_{24}$	119.8	N10-C08-C09	112.3(2)
$C_{20} = C_{23} = C_{24}$	118.0 (5)	12 - 60 - 609	120.0(3)
$C_{20} = C_{23} = H_{23}$	120.7	N12 - C69 - C68	117.9(3)
C24—C25—H25	120.7	N12—C69—H69	121.0
N4-C26-C25	120.7 (3)	C68—C69—H69	121.0
N4-C26-C35	112.4 (2)	N12-C70-C71	110.2 (2)
$C_{25} = C_{26} = C_{35}$	126.8 (3)	N12—C/0—H/01	109.6
N5-C27-C22	117.8 (3)	C/I_C/0_H/01	109.6
N5—C27—H27	121.1	N12—C/0—H/02	109.6
C22—C27—H27	121.1	C/1—C/0—H/02	109.6
N5—C28—C29	110.3 (2)	H/01—C/0—H/02	108.1
N5—C28—H281	109.6	C7/6—C71—C72	119.5 (3)
C29—C28—H281	109.6	C ⁷ /6—C71—C70	119.8 (3)
N5—C28—H282	109.6	C72—C71—C70	120.7 (3)
C29—C28—H282	109.6	C73—C72—C71	120.4 (3)

H281—C28—H282	108.1	С73—С72—Н721	119.8
C30—C29—C34	119.1 (3)	С71—С72—Н721	119.8
C30—C29—C28	121.2 (3)	C74—C73—C72	118.1 (3)
C34—C29—C28	119.7 (3)	С74—С73—Н73	120.9
C29—C30—C31	120.4 (3)	С72—С73—Н73	120.9
С29—С30—Н30	119.8	F7—C74—C73	118.3 (3)
C31—C30—H30	119.8	F7—C74—C75	118.1 (3)
$C_{32} - C_{31} - C_{30}$	1184(3)	C73 - C74 - C75	123.5(3)
C_{32} C_{31} H_{31}	120.8	C74 - C75 - C76	123.5(3) 117.5(3)
$C_{32} = C_{31} = H_{31}$	120.8	C74 $C75$ $H75$	121.3
C_{22} C_{22} E_{2}	120.0	$C_{14} = C_{15} = 1175$	121.3
$C_{33} = C_{32} = C_{31}$	110.1(3)	C70-C73-H73	121.5
C_{33} C_{32} C_{31}	123.3 (3)	C/I_C/6_C/5	120.9 (3)
F3-C32-C31	118.6 (3)	C/1_C/6_H/6	119.6
C32—C33—C34	117.8 (3)	C/5—C/6—H/6	119.6
С32—С33—Н33	121.1	N11—C77—C64	117.3 (3)
С34—С33—Н33	121.1	N11—C77—H77	121.3
C33—C34—C29	121.1 (3)	С64—С77—Н77	121.3
С33—С34—Н34	119.5	N11—C78—C79	111.6 (2)
С29—С34—Н34	119.5	N11—C78—H781	109.3
N6-C35-C26	117.9 (3)	С79—С78—Н781	109.3
N6—C35—H35	121.1	N11—C78—H782	109.3
С26—С35—Н35	121.1	С79—С78—Н782	109.3
N6—C36—C37	110.8 (2)	H781—C78—H782	108.0
N6—C36—H361	109.5	C80—C79—C84	119.3 (3)
C37—C36—H361	109.5	C80 - C79 - C78	1204(3)
N6-C36-H362	109.5	C84 - C79 - C78	120.1(3) 120.2(3)
C_{37} C_{36} H_{362}	109.5	C81 - C80 - C79	120.2(3)
$H_{361} = C_{36} = H_{362}$	109.5	$C_{81} C_{80} H_{80}$	110.7 (3)
$C_{12}^{(1)} = C_{22}^{(1)} = C_{2$	110.1	C_{20} C_{20} U_{20}	119.7
C42 - C37 - C38	110.0(3)	$C_{79} = C_{80} = H_{80}$	119.7
(42-(3)-(36))	121.2 (3)	$C_{80} = C_{81} = C_{82}$	117.9(3)
$C_{38} = C_{37} = C_{36}$	119.9 (3)		121.1
C39—C38—C37	121.1 (3)	C82—C81—H81	121.1
С39—С38—Н38	119.5	C83—C82—F8	118.9 (3)
С37—С38—Н38	119.5	C83—C82—C81	123.1 (3)
C40—C39—C38	118.1 (3)	F8—C82—C81	117.9 (3)
С40—С39—Н39	120.9	C82—C83—C84	118.3 (3)
С38—С39—Н39	120.9	С82—С83—Н83	120.8
F4C40C39	118.6 (3)	С84—С83—Н83	120.8
F4C40C41	118.7 (3)	C83—C84—C79	120.6 (3)
C39—C40—C41	122.7 (3)	C83—C84—H84	119.7
C40—C41—C42	118.9 (3)	С79—С84—Н84	119.7
C40—C41—H41	120.6	H111—O1—H112	106 (3)
C42—C41—H41	120.6	H211—O2—H212	102 (3)
C41—C42—C37	120.3 (3)	H311—O3—H312	100 (3)
C41 - C42 - H42	119.8	H411—04—H412	95 (3)
C_{37} C_{42} H_{42}	119.8	H511_05_H512	100 (3)
N10_Ni2_N7	177 10 (10)	H61106H612	108(3)
$\frac{1}{10} - \frac{1}{10} - \frac{1}{10}$	77 58 (10)	H711 07 H712	02(3)
1110-1112-1111	11.30(10)	11/11-0/-11/12	14 (3)

N7—Ni2—N11	102.20 (9)	H811—O8—H812	84 (3)
N10—Ni2—N9	99.90 (10)	Н911—О9—Н912	96 (3)
N7—Ni2—N9	77.20 (10)		
C5—N1—C1—C2	-0.5 (4)	C47—N7—C43—C44	-1.9 (4)
Ni1—N1—C1—C2	-177.8 (2)	Ni2—N7—C43—C44	177.0 (2)
C5—N1—C1—C6	175.8 (2)	C47—N7—C43—C48	176.1 (3)
Ni1—N1—C1—C6	-1.5 (3)	Ni2—N7—C43—C48	-4.9 (3)
N1-C1-C2-C3	0.9 (4)	N7—C43—C44—C45	2.5 (4)
C6-C1-C2-C3	-174.9 (3)	C48—C43—C44—C45	-175.3 (3)
C1—C2—C3—C4	-0.4 (4)	C43—C44—C45—C46	-0.3(5)
C2—C3—C4—C5	-0.4(4)	C44—C45—C46—C47	-2.3(5)
C1—N1—C5—C4	-0.4(4)	C43—N7—C47—C46	-0.8(4)
Ni1—N1—C5—C4	177.0 (2)	Ni2—N7—C47—C46	-179.8(2)
C1—N1—C5—C14	178.4 (2)	C43—N7—C47—C56	176.4 (3)
Ni1—N1—C5—C14	-4.3 (3)	Ni2—N7—C47—C56	-2.6(3)
C_{3} C_{4} C_{5} N_{1}	0.8(4)	C45-C46-C47-N7	2.9(5)
C_{3} C_{4} C_{5} C_{14}	-177.7(3)	C45 - C46 - C47 - C56	-173.9(3)
C7 - N2 - C6 - C1	-173.0(2)	C49 - N8 - C48 - C43	-172.5(2)
$N_{1} N_{2} C_{6} C_{1}$	1/5.0(2)	$N_{12} N_{12} N_{12} C_{13} $	27(3)
N1 C1 C6 N2	1.1(5) 0.1(4)	N12 N03 C48 N18	2.7(3)
$C_2 = C_1 = C_0 = N_2$	176.2(3)	$C_{44} C_{43} C_{46} N_{8}$	1.1(4)
$C_2 - C_1 - C_0 - N_2$	-128.3(3)	$C_{44} = C_{43} = C_{40} = 100$	-1207(3)
$C_0 = N_2 = C_7 = C_8$	-128.5(3)	$N_{2}^{2} N_{2}^{2} C_{40}^{2} C_{50}^{2}$	-129.7(3)
NII - N2 - C7 - C8	58.7(5)	N12 - N6 - C49 - C30	30.0(3)
$N_2 - C_7 - C_8 - C_{13}$	51.5 (4)	N8 - C49 - C30 - C53	-112.5(3)
$N_2 - C_1 - C_8 - C_9$	-130.6(3)	N8—C49—C50—C51	69.2 (4)
C13—C8—C9—C10	0.4 (4)	C55—C50—C51—C52	1.1 (5)
C/C8C9C10	-177.7 (3)	C49—C50—C51—C52	179.5 (3)
C8—C9—C10—C11	-2.2 (5)	C50—C51—C52—C53	-0.7 (5)
C9—C10—C11—F1	-178.4 (3)	C51—C52—C53—C54	-0.3 (5)
C9—C10—C11—C12	2.5 (5)	C51—C52—C53—F5	-179.9 (3)
F1—C11—C12—C13	179.8 (3)	F5—C53—C54—C55	-179.5 (3)
C10—C11—C12—C13	-1.0(5)	C52—C53—C54—C55	0.9 (5)
C9—C8—C13—C12	1.1 (4)	C53—C54—C55—C50	-0.5 (5)
C7—C8—C13—C12	179.2 (3)	C51—C50—C55—C54	-0.5 (5)
C11—C12—C13—C8	-0.8(5)	C49—C50—C55—C54	-178.9 (3)
C15—N3—C14—C5	-175.9 (2)	C57—N9—C56—C47	-173.5 (3)
Ni1—N3—C14—C5	1.4 (3)	Ni2—N9—C56—C47	4.1 (3)
N1—C5—C14—N3	1.7 (4)	N7—C47—C56—N9	-1.3 (4)
C4—C5—C14—N3	-179.7 (3)	C46—C47—C56—N9	175.7 (3)
C14—N3—C15—C16	-131.9 (3)	C56—N9—C57—C58	-114.9 (3)
Ni1—N3—C15—C16	51.4 (4)	Ni2—N9—C57—C58	67.9 (3)
N3-C15-C16-C17	61.3 (4)	N9—C57—C58—C63	-132.6 (3)
N3-C15-C16-C21	-120.1 (3)	N9—C57—C58—C59	50.2 (4)
C21—C16—C17—C18	0.1 (5)	C63—C58—C59—C60	-0.2 (5)
C15—C16—C17—C18	178.7 (3)	C57—C58—C59—C60	177.0 (3)
C16—C17—C18—C19	0.2 (5)	C58—C59—C60—C61	-0.7 (5)
C17—C18—C19—C20	-1.2 (6)	C59—C60—C61—C62	0.8 (6)
	× /		· · /

C17—C18—C19—F2	178.0 (3)	C59—C60—C61—F6	-178.2 (3)
C18—C19—C20—C21	1.8 (5)	C60—C61—C62—C63	0.0 (6)
F2-C19-C20-C21	-177.3 (3)	F6—C61—C62—C63	179.0 (3)
C19—C20—C21—C16	-1.5 (5)	C59—C58—C63—C62	1.0 (5)
C17—C16—C21—C20	0.6 (5)	C57—C58—C63—C62	-176.2(3)
C15—C16—C21—C20	-178.1(3)	C61—C62—C63—C58	-1.0(6)
$C_{26} - N_{4} - C_{22} - C_{23}$	1.0 (4)	C68—N10—C64—C65	1.1 (4)
Ni1 - N4 - C22 - C23	178 4 (2)	Ni2—N10—C64—C65	1770(2)
$C_{26} N_{4} C_{22} C_{27}$	-1775(2)	C68 - N10 - C64 - C77	-1767(2)
Ni1 $-$ N4 $-$ C22 $-$ C27	-0.1(3)	Ni2 $-$ N10 $-$ C64 $-$ C77	-0.8(3)
$N4-C^{2}-C^{2}-C^{2}$	-0.6(4)	N10-C64-C65-C66	-1.2(4)
C_{27} C_{22} C_{23} C_{24}	177.7(3)	C77 - C64 - C65 - C66	1.2(4)
$C_{22}^{22} = C_{23}^{22} = C_{24}^{22} = C_{25}^{22}$	-0.2(4)	C64 - C65 - C66 - C67	170.2(5)
$C_{22} = C_{23} = C_{24} = C_{25}$	0.2(4)	C65 C66 C67 C68	0.5(5)
$C_{23} = C_{24} = C_{23} = C_{26}$	-0.5(4)	C64 N10 C68 C67	-0.1(4)
$N_{22} = N_{4} = C_{20} = C_{23}$	-177.0(2)	$N_{12} = N_{10} = C_{03} = C_{07}$	-1760(2)
NII - N4 - C20 - C25	-1/7.9(2)	N12 - N10 - C08 - C07	-170.0(2)
122 - 104 - 220 - 235	1/9.2(2)	10-100 - 1	1/9.0(2)
N11 - N4 - C26 - C35	1.8(3)	N12 - N10 - C08 - C09	3.8(3)
C_{24} C_{25} C_{26} N_{4}	-0.4(4)	C66-C67-C68-N10	-0.7(4)
$C_{24} = C_{25} = C_{26} = C_{35}$	-180.0(3)	C66-C6/-C68-C69	1/9.6 (3)
$C_{28} = N_{5} = C_{27} = C_{22}$	1/9.5 (2)	C/0 = N12 = C69 = C68	1//.8(2)
N11—N5—C27—C22	1.9 (3)	N12—N12—C69—C68	-0.9 (3)
N4—C22—C27—N5	-1.3(4)	N10—C68—C69—N12	-1.7 (4)
C23—C22—C27—N5	-179.7 (3)	C67—C68—C69—N12	178.1 (3)
C27—N5—C28—C29	115.7 (3)	C69—N12—C70—C71	123.2 (3)
Ni1—N5—C28—C29	-67.1 (3)	Ni2—N12—C70—C71	-58.3 (3)
N5-C28-C29-C30	116.8 (3)	N12—C70—C71—C76	-59.2 (3)
N5—C28—C29—C34	-60.3 (3)	N12—C70—C71—C72	121.5 (3)
C34—C29—C30—C31	0.3 (5)	C76—C71—C72—C73	1.8 (4)
C28—C29—C30—C31	-176.8 (3)	C70—C71—C72—C73	-178.9 (3)
C29—C30—C31—C32	0.2 (5)	C71—C72—C73—C74	-1.0 (4)
C30—C31—C32—C33	-0.9 (6)	C72—C73—C74—F7	179.6 (3)
C30—C31—C32—F3	179.6 (3)	C72—C73—C74—C75	0.0 (5)
F3—C32—C33—C34	-179.5 (3)	F7—C74—C75—C76	-179.4 (2)
C31—C32—C33—C34	1.1 (5)	C73—C74—C75—C76	0.2 (4)
C32—C33—C34—C29	-0.6 (5)	C72—C71—C76—C75	-1.6 (4)
C30—C29—C34—C33	-0.1 (4)	C70—C71—C76—C75	179.0 (3)
C28—C29—C34—C33	177.0 (3)	C74—C75—C76—C71	0.6 (4)
C36—N6—C35—C26	179.4 (2)	C78—N11—C77—C64	173.7 (2)
Ni1—N6—C35—C26	0.3 (3)	Ni2—N11—C77—C64	-0.2 (3)
N4-C26-C35-N6	-1.3 (4)	N10-C64-C77-N11	0.7 (4)
C25-C26-C35-N6	178.3 (3)	C65—C64—C77—N11	-177.0 (3)
C35—N6—C36—C37	106.9 (3)	C77—N11—C78—C79	128.7 (3)
Ni1—N6—C36—C37	-74.2 (3)	Ni2—N11—C78—C79	-58.3 (3)
N6-C36-C37-C42	129.2 (3)	N11—C78—C79—C80	-55.9 (4)
N6-C36-C37-C38	-53.9 (3)	N11—C78—C79—C84	125.7 (3)
C42—C37—C38—C39	-2.0 (4)	C84—C79—C80—C81	-3.0(5)
C36—C37—C38—C39	-178.9 (2)	C78—C79—C80—C81	178.6 (3)

C37—C38—C39—C40	1.2 (4)	C79—C80—C81—C82	1.4 (5)
C38—C39—C40—F4	179.6 (2)	C80—C81—C82—C83	1.2 (5)
C38—C39—C40—C41	0.6 (4)	C80—C81—C82—F8	-178.4 (3)
F4—C40—C41—C42	179.4 (3)	F8—C82—C83—C84	177.6 (3)
C39—C40—C41—C42	-1.6 (5)	C81—C82—C83—C84	-2.0 (5)
C40—C41—C42—C37	0.8 (4)	C82—C83—C84—C79	0.3 (5)
C38—C37—C42—C41	0.9 (4)	C80—C79—C84—C83	2.1 (5)
C36—C37—C42—C41	177.9 (3)	C78—C79—C84—C83	-179.5 (3)

Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	D—H···A
C4—H4…O6	0.95	2.60	3.505 (4)	160
C6—H6…F7 ⁱ	0.95	2.35	3.238 (3)	155
C18—H18…F4 ⁱ	0.95	2.45	3.363 (4)	162
С23—Н23…О2	0.95	2.56	3.504 (4)	172
C25—H25…F2 ⁱⁱ	0.95	2.43	3.303 (4)	154
C28—H282···Br3 ⁱⁱⁱ	0.99	2.80	3.784 (3)	170
С31—Н31…О5 ^{ііі}	0.95	2.66	3.557 (5)	157
C36—H362…Br2 ^{iv}	0.99	2.97	3.933 (3)	164
C39—H39····O3 ^v	0.95	2.44	3.373 (4)	166
C44—H44····F8 ^{vi}	0.95	2.48	3.341 (4)	150
C46—H46····Br3 ⁱⁱ	0.95	2.79	3.708 (3)	164
C52—H52····O9 ^{iv}	0.95	2.53	3.292 (6)	138
C60—H60····F8 ^{vii}	0.95	2.55	3.329 (4)	139
С62—Н62…О9 ^{viii}	0.95	2.48	3.365 (6)	156
C65—H65····Br4 ^{viii}	0.95	2.91	3.653 (3)	136
C67—H67····O2 ^{iv}	0.95	2.58	3.503 (4)	165
C69—H69····Br1 ^{iv}	0.95	3.07	3.750 (3)	130
C73—H73…F1 ⁱⁱ	0.95	2.60	3.312 (4)	132
C81—H81····F5 ^{ix}	0.95	2.50	3.436 (4)	169
C83—H83…O1 ^x	0.95	2.52	3.263 (4)	135
O1—H1111····Br4 ^{xi}	0.89 (2)	2.57 (2)	3.449 (3)	167 (3)
O1—H112…Br4	0.88 (2)	2.61 (2)	3.476 (3)	170 (3)
O2—H211…Br3	0.89 (2)	2.50 (2)	3.384 (3)	168 (3)
O2—H212…Br1	0.87 (2)	2.44 (2)	3.317 (2)	178 (3)
O3—H311…Br2	0.90 (2)	2.39 (2)	3.274 (2)	169 (3)
O3—H312…Br1	0.89 (2)	2.44 (2)	3.307 (2)	164 (3)
O4—H411…O5	0.95 (2)	2.12 (2)	3.033 (5)	161 (4)
O4—H412…Br1	0.93 (2)	2.32 (2)	3.250 (3)	178 (4)
O5—H511…O3	0.88 (2)	1.93 (2)	2.809 (4)	172 (5)
O5—H512…O7 ^{xii}	0.90 (2)	2.11 (3)	2.906 (4)	147 (4)
O6—H611…Br2 ^{iv}	0.86 (2)	2.57 (2)	3.407 (2)	165 (3)
O6—H612···Br2 ⁱⁱⁱ	0.87 (2)	2.44 (2)	3.306 (3)	173 (4)
O7—H711…O6	0.90 (2)	2.03 (2)	2.925 (4)	171 (4)

O8—H811…O9	0.94 (2)	2.45 (3)	3.358 (6)	163 (4)
O8—H812…Br4	0.94 (2)	2.71 (3)	3.579 (3)	154 (4)

Symmetry codes: (i) -x+2, -y, -z+1; (ii) -x+1, -y, -z+1; (iii) x+1, y, z; (iv) -x+1, -y+1, -z+1; (v) x+1, y-1, z; (vi) -x+1, -y, -z+2; (vii) -x+2, -y, -z+2; (viii) -x+2, -z+2; (viii) -x+2; (vii) -x+2