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On the Structural and Chemical Characteristics of Co/Al₂O₃/graphene Interfaces for Graphene Spintronic Devices

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Bárbara Canto¹, Cristol P. Gouvea², Bráulio S. Archanjo², João E. Schmidt¹ & Daniel L. Baptista¹

We report a detailed investigation of the structural and chemical characteristics of thin evaporated Al₂O₃ tunnel barriers of variable thickness grown onto single-layer graphene sheets. Advanced electron microscopy and spectrum-imaging techniques were used to investigate the Co/Al₂O₃/graphene/SiO₂ interfaces. Direct observation of pinhole contacts was achieved using FIB cross-sectional lamellas. Spatially resolved EDX spectrum profiles confirmed the presence of direct point contacts between the Co layer and the graphene. The high surface diffusion properties of graphene led to cluster-like Al₂O₃ film growth, limiting the minimal possible thickness for complete barrier coverage onto graphene surfaces using standard Al evaporation methods. The results indicate a minimum thickness of nominally 3 nm Al₂O₃, resulting in a 0.6 nm rms rough film with a maximum thickness reaching 5 nm.

Graphene is a potential material for spintronic applications because of the combination of its expected long spin lifetime and high electron mobility. The spin diffusion distances observed in graphene are very long; i.e., few micrometers at room temperature^{1–6}. Experimentally, graphene spin-injection devices can be obtained by fabricating ferromagnetic metal contacts on graphene; these assemblies act as spin-current filters. However, previous studies showed that electrical spin injection from such ferromagnetic electrodes in direct contact (transparent contact) with graphene is not effective because of the conductance mismatch^{1,7}. Instead, the use of a thin insulating layer acting as a tunnel barrier (tunneling contact) between the graphene layer and the metal electrodes has proven to be an effective solution^{3,4,8–10}. Han *et al.* observed an increase of the injection efficiency from 1 to 26–30% by using tunneling contacts. Concomitantly, the spin relaxation time was also enhanced by more than ninefold, reaching 771 ps at room temperature⁴. The effect of direct metal contacts on spin lifetime measurements in graphene was investigated by Maassen *et al.*¹¹ An important discussion on the effect of low resistance contact-induced spin relaxation on Hanle precession curves is afforded. Recently, a theoretical closed-form expression for Hanle spin precession in different regimes was also provided, clearly demonstrating the influence of metal contacts on the spin relaxation mechanisms and also the importance of using tunneling contacts¹². Nevertheless, complete control of standard tunneling barrier fabrication on graphene sheets is still distant^{1,4,13–18}. Barrier structural and chemical non-uniformities seem to play a crucial role in the experimental spin relaxation time values; these are much shorter than expected (*ca.* microsecond)^{1,3,4,19} from the low intrinsic spin-orbit couplings of graphene²⁰. Barrier pinholes (pinhole contacts) are one of the barrier defects that may lower the metal/barrier/graphene interface quality, directly affecting spin injection and relaxation through the graphene^{1,3,19}.

¹Instituto de Física, PPGFis, PPGMicro, Universidade Federal do Rio Grande do Sul, Porto Alegre, 91501-970, RS, Brazil. ²Divisão de Metrologia de Materiais, INMETRO, Duque de Caxias, 25250-020, RJ, Brazil. Correspondence and requests for materials should be addressed to D.L.B. (email: dbaptista@gmail.com)

Tombros *et al.*¹ noted that evaporated alumina (aluminum evaporation followed by oxidation) is commonly used to construct tunneling barriers. However, the possible existence of pinholes remains an important issue for the development of standard fabrication procedures for 1–5-nm-thick barriers^{1,13}. Han *et al.* also evaluated the influence of barrier roughness on the spin relaxation mechanisms by using molecular beam epitaxy MgO and TiO₂ seed layers as an alternative for the fabrication of smoother barrier layers on graphene¹⁴. Dublak *et al.* demonstrated the use of sputtering to deposit continuous 1-nm-thick Al₂O₃ onto graphene, but this technique damaged the graphene structure and hence reduces the applicability of such an approach^{15,16}. Fluorinated-graphene²¹ and h-BN^{22,23} monolayers were also successfully used as tunneling barriers; however, sensitive chemical processes and/or critical layer transfer steps are added. One such alternative approach enhanced important spin transport metrics, but the results have still not reached much longer spin relaxation times as expected²⁰. Recently, hydrogenated-graphene barrier was also proposed; the lower spin polarization was justified by the authors due to the possible presence of magnetic moments acting as spin scatterers in such tunnel barrier²⁴. Thus, a complete control and understanding of the structural and chemical nature of tunneling barriers on graphene and the role of graphene and barrier defects on spin injection and relaxation is still an experimental challenge. A detailed investigation concerning the metal/barrier/graphene interface nature using direct electron microscopy visualization and nanometer-resolved spectrum profiles is currently absent from the literature.

In this work, we report a detailed investigation of the structural and chemical characteristics of traditional thin evaporated Al₂O₃ barriers with variable thickness grown onto single-layer graphene sheets. Advanced electron microscopy and spectrum-imaging techniques were used to investigate the Co/Al₂O₃/graphene/SiO₂ and Co/graphene/SiO₂ interfaces. A direct cross-sectional observation of barrier pinholes is reported as well as results concerning the minimal barrier thickness necessary for complete graphene coverage using standard Al evaporation. The results are compared with standard thin Al₂O₃ deposition onto SiO₂/Si substrates.

Results and Discussions

The Co-coated evaporated Al₂O₃ barrier samples were analyzed systematically by advanced (S)-TEM techniques. Figures 1 and 2 show TEM and STEM cross-sectional images, respectively, of samples with different Al₂O₃ barrier thicknesses. The analyses included two substrate regions: with and without graphene. It is important to notice that each Al₂O₃ barrier sample presents regions with and without graphene at the same substrate, leading to the same nominal barrier deposition for both graphene/SiO₂ and SiO₂ regions. Parts (a) and (c) of both Fig. 1 (TEM) and 2 (STEM) correspond to images of the 3- and 1-nm-thick barriers deposited directly onto the SiO₂ surface in a region without graphene. For both barrier thicknesses, it is possible to observe a homogeneous layer covering the whole substrate surface. The Co layer is also observed as well as the top granular platinum-capping region, purposely deposited during the FIB lamella preparation to protect the integrity of the sample interface layers. It is important to note that both TEM and STEM images present similar features but in reverse contrast. The HAADF STEM images (Fig. 1) contain a high-Z contrast dependence where high-Z regions are imaged through the annular detector as brighter zones. In such cases, the Co layer is imaged brighter than the Al₂O₃ one.

A different morphological behavior was observed when the Al₂O₃ was deposited onto the graphene sheets. Figures 1(b,d) and 2(b,d) show cross-sectional images of monolayer graphene samples with nominally 3- and 1-nm-thick Al₂O₃ barriers. The images clearly show that the barrier thickness is larger compared with the same barrier deposited directly onto the SiO₂. The maximum thickness reached *ca.* 4 and 5 nm for respectively the nominally 1- and 3-nm-thick Al₂O₃ layers. Both TEM and STEM analyses showed that the surface coverage was incomplete: a few nanometers-large pinholes are evident in the graphene samples coated with the nominally 1-nm-thick Al₂O₃ barrier. The pinholes are especially clear in the STEM images, where the Co filling inside the holes is also visible. As noted above, the Co appears brighter than the Al₂O₃ in HAADF STEM images. Thus, the Co layer clearly contacts the graphene directly in those pinhole regions.

In contrast, the graphene sample with a nominally 3-nm-thick Al₂O₃ layer did not contain pinholes. An Al₂O₃ layer having a maximum thickness of *ca.* 5 nm and completely covering the whole graphene surface was observed.

The chemical nature of the Co/Al₂O₃/graphene/SiO₂ interfaces at different regions was also probed by means of STEM-EDX spectrum profiling experiments. Figure 3(a) shows the normalized-intensity EDX profile along a section without pinhole for the nominally 1-nm-thick barrier sample (indicated by the arrow in Fig. 1(d)). As expected, the profile confirmed the presence of a well-defined Al₂O₃ layer between the graphene and the Co film. The thicknesses of the layers are consistent with those obtained by TEM and STEM although they appear larger because of the lower spatial resolution of the EDX technique. When the same profiling experiment was acquired for a region with pinholes (indicated by the arrow in Fig. 1(d)), the EDX Al signal from the Al₂O₃ barrier disappeared, confirming the formation of a direct interface between the graphene and the Co layer (Fig. 3(b)).

Such non-homogeneity of the Al₂O₃ barrier on the graphene is attributed to agglomeration during the Al deposition. The low-energy graphene surface induces high diffusion mobility of atomic species during Al deposition, which leads to cluster-like film (Volmer-Weber) growth²⁵. The coverage on graphene can be partial and the barrier thickness is greater than nominal. This clustering phenomenon limits the minimal thickness possible for complete barrier coverage on graphene surfaces using standard Al

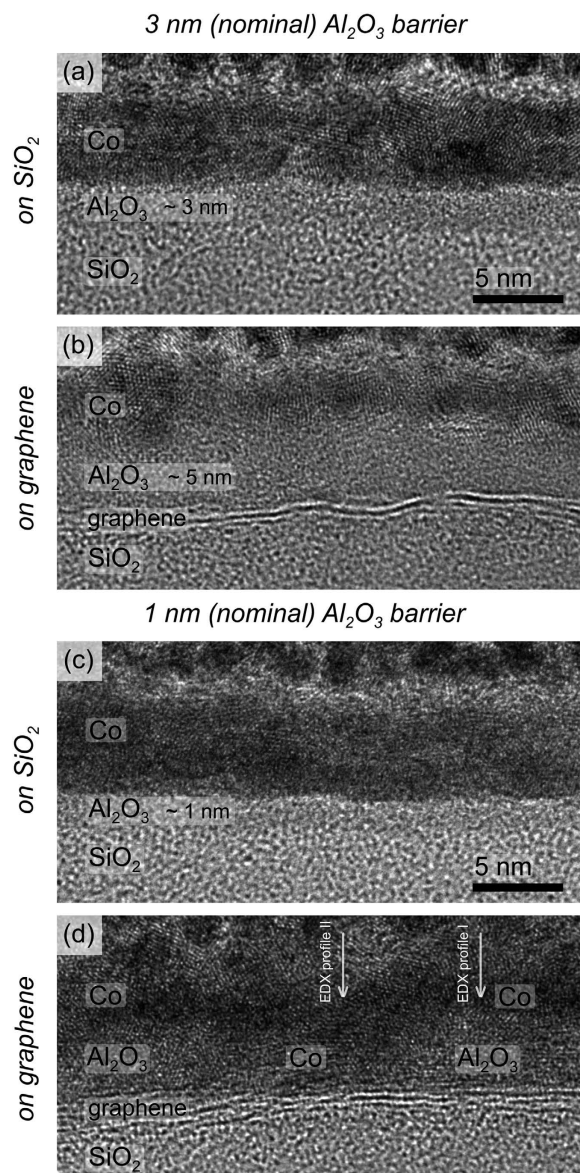


Figure 1. TEM cross-sectional images of samples with different Al_2O_3 barrier thicknesses, 3-nm-thick (nominal) barrier on (a) SiO_2 and (b) graphene, 1-nm-thick (nominal) barrier on (c) SiO_2 and (d) graphene.

evaporation methods. In such case, the results indicate an Al_2O_3 layer having a minimum thickness of nominally 3 nm, which means thickness reaching a maximum thickness of *ca.* 5 nm. Although the nominal (3 nm) thickness may lead to complete barrier coverage, AFM measurements indicated a relative high rms roughness of ~ 0.6 nm (see Supplementary Fig. S2 online).

The chemical nature of the barrier was also investigated using spatially resolved EELS during STEM-HAADF experiments. Figure 4 shows the Al L-edge spectrum of the 3-nm-thick (nominal) barrier sample. The analysis of the near-edge fine structure is consistent with the presence of Al from an Al_2O_3 phase, indicating complete oxidation of the evaporated Al^{26,27}.

Raman spectroscopy was used to probe the effect of the Al_2O_3 barrier and the Co deposition on the graphene atomic structure quality; Fig. 5 shows the spectra acquired at each fabrication step. No apparent damage was observed for any barrier thickness. The diffusive character of the thermal evaporation of aluminum leads to non-energetic species deposition, which preserves the graphene integrity. Similar results have been reported in the literature for other tunnel barriers thermally deposited onto graphene sheets.

The use of Co sputtering deposition during the fabrication of ferromagnetic contacts may lead to moderate graphene damage. The Raman spectrum of graphene after direct (without barrier) Co deposition contained a prominent defects-related D band²⁸. This can be explained by momentum transference among energetic (a few eV) Co species and carbon atoms in graphene through knock-on collisions at the very beginning of the Co film deposition. The use of tunnel barriers between the graphene and

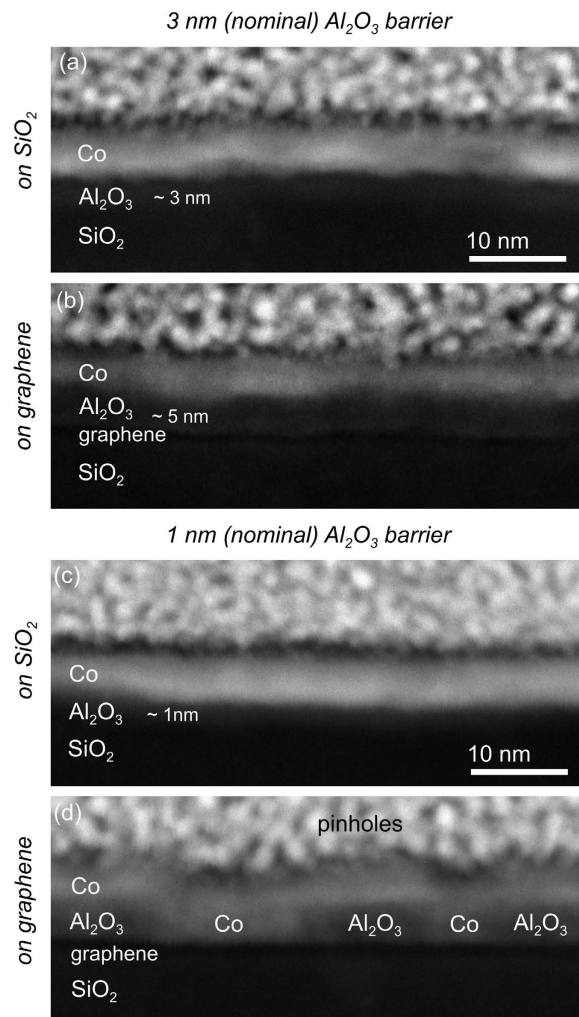


Figure 2. HAADF-STEM cross-sectional images of samples with different Al₂O₃ barrier thicknesses, 3-nm-thick (nominal) barrier on (a) SiO₂ and (b) graphene, 1-nm-thick (nominal) barrier on (c) SiO₂ and (d) graphene.

the Co contacts may prevent such damage. Stopping and range of ions in matter (SRIM) simulations²⁹ (not shown) indicated that even very thin (1 nm) Al₂O₃ barriers are very thick for few-eV sputtered Co species, and are sufficiently thick to prevent any damage to the graphene. However, the presence of pinholes in the barrier structure provides a direct path for Co-carbon collisions. Figure 5 shows the Raman spectra of graphenes after Co deposition having 1-, 2- and 3-nm-thick (nominal) Al₂O₃ barriers. The presence of the D band for the samples with the 1- and 2-nm-thick barriers clearly indicates the presence of inhomogeneous barriers containing many pinholes (see Supplementary Fig. S3 online). It is important to note that the I_D/I_G ratio is higher for the sample without a barrier, and that this ratio decreases as the thickness of the barrier increases. This indicates progressive coverage of the surface by the Al₂O₃ barrier. The D band is absent for the 3-nm-thick (nominal) barrier samples, confirming that this thickness is the approximate limit for complete coverage of the graphene by the Al₂O₃. The Raman analyses are in concordance with the electron microscopy results, indicating that Raman spectroscopy in such experiment is a useful technique for probing the existence of pinholes in tunnel barriers on graphene over large areas.

In conclusion, direct observation of pinhole contacts was achieved using FIB cross-sectional and advanced high-resolution TEM and STEM analyses. Spatially resolved EDX spectrum profiling showed the nature of direct point-contacts between the Co ferromagnetic layer and the graphene. Raman spectroscopy indicated that moderate damage occurred over large areas of the graphene during the Co sputtering deposition. The presence of pinholes in the barrier structure provided a direct path for Co-carbon collisions. Such pinholes were widely distributed on the samples having 1- and 2-nm-thick (nominal) Al₂O₃ barriers. Only the 3-nm-thick (nominal) barrier provided complete coverage of the graphene surface, thereby preserving the graphene integrity during the Co deposition. No pinholes were directly or indirectly observed. However, this thicker barrier had a relative high rms roughness of *ca.* 0.6 nm. The high surface diffusion properties of graphene led to cluster-like Al₂O₃ film growth. This phenomenon

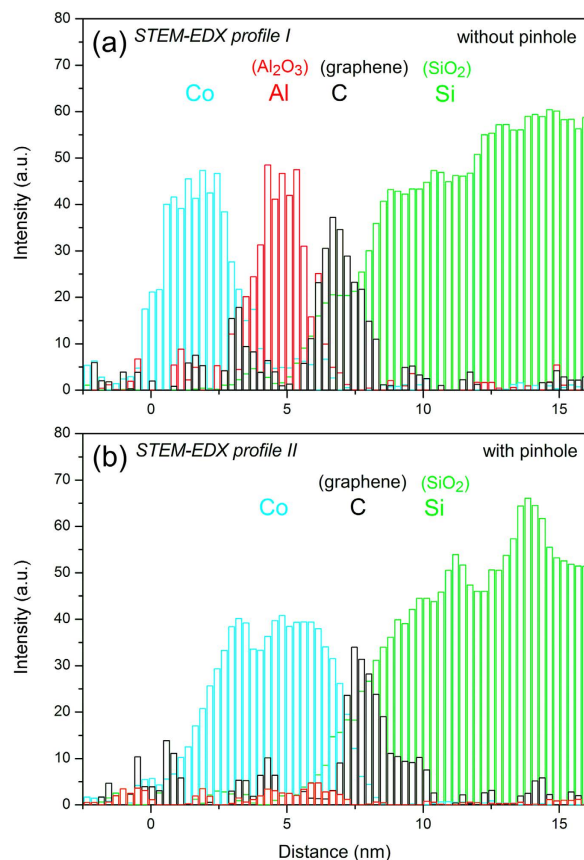


Figure 3. Normalized-intensity EDX profile along a section (a) without pinhole and (b) with pinhole for the nominally 1-nm-thick barrier sample.

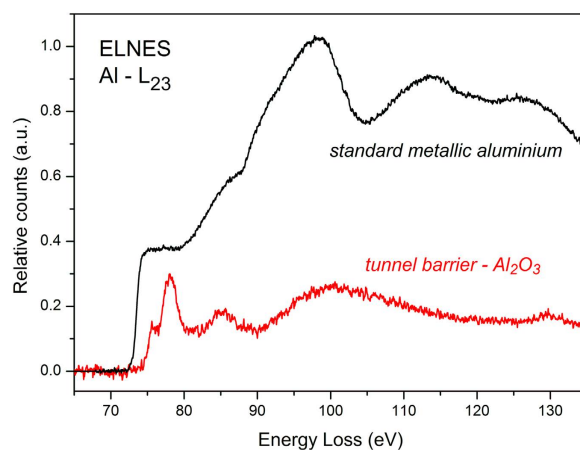


Figure 4. Spatially resolved EELS of the tunnel barrier showing the Al L-edge spectrum, the near-edge fine structure is consistent with the presence of Al from an Al₂O₃ phase, indicating complete oxidation of the evaporated Al layer; a spectrum of a standard metallic Al sample is also showed for comparison.

limits the minimal possible thickness for complete barrier coverage on graphene surfaces using standard Al evaporation methods. The minimum required Al₂O₃ layer thickness for complete coverage of nominally 3 nm becomes *ca.* 5 nm in practice because of this clustering.

Methods

Graphene flakes were obtained by micromechanical cleavage of single-crystal graphite^{30,31} (see Supplementary Fig. S1 online). The flakes were placed onto 90-nm-thick SiO₂ films that had been thermally grown on silicon substrates. Monolayer graphenes were initially localized using optical microscopy.

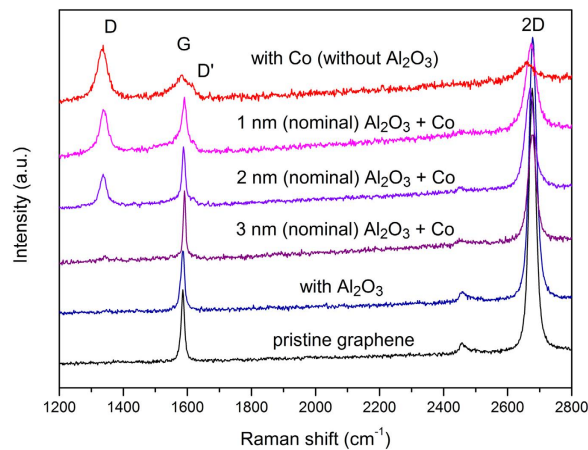


Figure 5. Raman spectra of graphene at each fabrication step, the pristine exfoliated graphene presents a typical spectrum for a high quality structure with absent D peak and high 2D one, the effect of the Al_2O_3 barrier and the Co deposition on the graphene atomic structure quality is showed.

The crystalline quality and the number of layers of each flake were probed using Raman spectroscopy (Renishaw inVia confocal microscope operated with a 532-nm solid-state laser). Al_2O_3 barriers were then deposited by thermal evaporation of aluminum *in vacuo* (base pressure of 10^{-7} Torr) with posterior ambient oxidation. Different nominal Al_2O_3 barrier thicknesses (expansion factor of 1.28 from Al thickness) were evaluated; i.e., 1, 2 and 3 nm. The samples were then coated with a Co layer, which represented the ferromagnetic electrodes of a spintronic device. A 4-nm-thick Co layer was deposited by DC-magnetron sputtering at 3 mTorr of Ar, 130 mA (*ca.* 160 V) and a base pressure of 10^{-7} Torr. The samples were monitored by Raman spectroscopy between each fabrication step. A Cs-corrected FEI Titan 80/300 S/TEM microscope was used to obtain scanning and conventional transmission electron micrographs of the Co/ Al_2O_3 /graphene/ SiO_2 interfaces. High-angle annular dark field (HAADF) images were acquired using the STEM mode. Space-resolved elemental analyses were performed *via* energy dispersive X-ray (EDX) spectroscopy to map the layer interfaces. The chemical nature of the Al_2O_3 barriers was also analyzed using electron energy loss spectroscopy (EELS). The cross-sectional (S)-TEM samples were previously prepared using focused ion beam (FIB) protocols for lamella preparation (FEI Helios NanoLab DualBeam). Topographical aspects of the Al_2O_3 barriers were also acquired through atomic force microscopy measurements using a Bruker Multimode 8 AFM operated in the intermittent mode at a drive amplitude of 120 mV with a Si tip having $k = 5 \text{ N/m}$.

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Author Contributions

B.C., D.L.B and J.E.S. conceived the experiments. B.C. and D.L.B. conducted the samples fabrication, Raman and AFM measurements and all sets of data analyses. B.C. and D.L.B wrote the main manuscript text. D.L.B. performed the (S)TEM experiments. C.P.G. and B.S.A. worked on the FIB lamella preparation. All authors reviewed the manuscript. All authors have given approval to the final version of the manuscript.

Additional Information

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