



Research article

Antioxidant, antidiabetic, and anti-inflammatory activities of flavonoid-rich fractions of *Solanum anguivi* Lam. fruit: *In vitro* and *ex vivo* studies

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ABSTRACT

Diabetes mellitus is a major, rapidly growing endocrine disorder in most countries. The high cost and side effects of conventional drugs for the management of this disease have shifted attention to medicinal plants. *Solanum anguivi* (*S. anguivi*) fruits has been reported to be a very good and rich source of polyphenols such as flavonoids, that can be exploited. Flavonoids are plant secondary metabolites widely found in vegetables, fruits and seeds and are known to be of medicinal significance in different range of diseases like diabetes. This study involved *in vitro* and *ex vivo* assays on the antioxidant, anti-inflammatory, and antidiabetic properties of flavonoid-rich fractions of *S. anguivi* fruits. Healthy male Wistar rats (n = 5) weighing 150–180 g were used for *ex vivo* antioxidant and antidiabetic studies, their liver was exercised for the experiment. The percentage yields of the three flavonoid-rich fractions (Fr. A, B, and C) of *S. anguivi* fruits obtained from the column chromatographic technique were 15.53 ± 0.75 , 11.53 ± 0.80 , and 10.17 ± 0.49 mg/g quercetin equivalents. The three fractions (A, B, and C) of *S. anguivi* fruits significantly scavenged both 2,2-diphenyl-1-picrylhydrazyl (DPPH) with fraction A having the lowest IC₅₀ value (26.14 ± 1.06 µg/ml) compared with fraction B (37.78 ± 5.12 µg/ml) and fraction C (38.24 ± 2.40 µg/ml) when compared with ascorbic acid with the least IC₅₀ value (15.27 ± 0.34 µg/ml). While fraction A (19.61 ± 1.19 µg/ml) scavenged nitric oxide (NO) radicals better than fraction B (22.97 ± 0.55 µg/ml) and fraction C (49.95 ± 6.18 µg/ml). Although ascorbic acid had better scavenging ability than the three fractions (17.23 ± 0.16 µg/ml). The flavonoid-rich fraction A shows better result in inhibiting α-glucosidase with IC₅₀ value of 16.24 µg/ml compared to fraction B (128.04 µg/ml) and fraction C (143.16 µg/ml). For α-amylase, flavonoid-rich fraction A had an IC₅₀ of 31.50 µg/ml compared to B (84.32 µg/ml) and C (145.40 µg/ml). The various controls also showed promising results with acarbose having IC₅₀ of 3.93 µg/mL and 15.66 µg/mL respectively for α-glucosidase and α-amylase. Our findings also showed that FeSO₄-induced tissue damage decreased the levels of GSH, SOD, and CAT activities while increasing the levels of MDA. In contrast, following treatment with the three flavonoid fractions of *S. anguivi* fruits helped to restore these parameters to near-normal levels, by significantly increasing the potential of GSH, SOD, CAT and reducing the levels of MDA which signifies that flavonoid-rich fractions of *S. anguivi* have great potential to address complications arising from oxidative stress. In addition, the three flavonoid-rich fractions A, B, and C of *S. anguivi* fruits exhibited *ex vivo* anti-inflammatory properties via reduced nitric oxide levels in iron-induced oxidative damage. Data

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obtained from this study shows that the flavonoid-rich fraction of *S. anguivi* possess anti-diabetic property via inhibition of α -glucosidase and α -amylase and antioxidant property via free radical scavenging. Also, comparing all the fractions, flavonoid-rich fraction A appears to be more potent compared to the fractions B and C. Further research will be needed in isolating and as well applying the fractions in real life situations in the management of diabetes.

1. Introduction

Medicinal plants are groups of plants with vital roles in alleviating human illness [1]. According to a survey done by the World Health Organization, medicinal plants have long been regarded for their medicinal potentials and nutritional importance, both of which aid in the treatment of various ailments including oxidative stress-related diseases and drug development [2]. Several experimental studies on medicinal plants have found significant differences in chemical composition (total flavonoid and total phenolic contents), mineral count, and biological activity between geographical regions and places around the world [3,4]. *Solanum anguivi* Lam. (SAG) is a rare ethnomedicinal herb belonging to the family Solanaceae. It is called African eggplant in English. The chronic administration and antioxidant activity of saponin extract of *S. anguivi* fruits in Wistar rats was reported by Ref. [5]. Their structure consists of 15 carbon skeletons and two aromatic rings (A and B) connected by three carbon chains. Flavonoids are further classified into 6 subclasses: flavonols, flavones, flavanones, isoflavones, flavanols, and anthocyanidins [6].

Diabetes mellitus (DM) is a chronic metabolic disease that is characterized by a deficiency in insulin production, insulin action or both [7]. Diabetes usually presents with polydipsia, polyuria, and microvascular problems involving the eyes, kidney and peripheral nerves along with cardiovascular challenges, including hypertension [7]. Despite efforts to curb the prevalence of diabetes mellitus, the disease is still on the rise. Hence, an effective therapeutic approach targeting diabetes precursor enzymes such as α -amylase and α -glucosidase for treating diabetes will be of significant benefit [8,9]. The α -amylase enzyme converts oligosaccharides to polysaccharides [10], while α -glucosidase catalyzes the final phase of carbohydrate hydrolysis, resulting in absorbable monosaccharides [11]. However, inhibition of the α -amylase enzyme reduces the rate of glucose absorption, especially after meals [12]. Oxidative stress is a main metabolic abnormality involved in the development of diabetic complications [13]. This results when there is an increased production of reactive oxygen species beyond the capacity of the antioxidant system. ROS production could increase as a result of abnormal metabolism of glucose, free fatty acids and other reactive metabolites in the diabetic mellitus state [14]. Chronic inflammation and oxidative stress are involved in the pathophysiology of diabetes mellitus. Proinflammatory cytokines can indirectly incite oxidative stress by activating macrophages, which play an essential role in removing the pathogen via the production of reactive oxygen species. Hence, diabetes mellitus, oxidative stress and inflammation act in a vicious cycle. Oxidative damage has been linked to the onset and progression of numerous liver diseases, including fibrosis, liver cancer, and cirrhosis [15]. *Solanum anguivi* has been reported as a natural source of antioxidant. Several studies have justified the testing of *S. anguivi* in an animal model. The administration of saponin from *S. anguivi* fruit showed marked hypoglycemic, antiperoxidative and antihyperlipidemic effects in alloxan-induced diabetic rats [16]. In addition, Elekofehinti et al. [16] reported that the methanolic extract of *S. anguivi* fruit exhibited antioxidant activity. Furthermore, the saponins extract from *S. anguivi* fruit was also reported to exhibit cholesterol lowering effect [17]. Despite the few pharmacological activities reported on the *S. anguivi* fruits, there is little information of its effects on metabolites and pathways involved in oxidative-mediated injury in liver tissues. Therefore, the aim of this study was to evaluate the antioxidant, antidiabetic, and anti-inflammatory activities of flavonoid-rich fractions of *Solanum anguivi* LAM. Fruit via *in vitro* and *ex vivo* studies.

2. Materials and methods

2.1. Collection of plant material

The fruits of *Solanum anguivi* were obtained from Oja Oba's Market in Ado Ekiti, Ekiti State, Nigeria. It was authenticated at the Herbarium Unit of the Department of Plant Science, Ekiti State University, Ado-Ekiti, with herbarium number UHAE 202/377. The fruits of *Solanum anguivi* were air-dried for four weeks, ground to a fine powder using a mechanical blender (Kenwood, Model BL490, China), and then stored in an airtight container for further investigation.

2.2. Preparation of methanolic and flavonoid-rich fractions of *S. anguivi* fruits

The extraction of flavonoids was carried out from the defatted *S. anguivi* fruits using petroleum ether [18]. The powdered plant was macerated in 80 % methanol to obtain the hydroalcoholic crude extract using an Erlenmeyer flask for 2 days at 25 °C. After 48 h, the filtrate was filtered using filter paper (Whatman No. 1). The flavonoid-rich lipids were further extracted using the method of [19]. The filtrate collected was concentrated to dryness using a rotary evaporator. Column chromatographic separation was carried out by preparing a homogenous slurry of the flavonoid-rich extract in a solvent system of chloroform and methanol (60:40). Next, the adsorption of the sample onto silica gel (20 g) was determined. The sample was then allowed to dry at room temperature to form powder. The powder was added from the top of the column with the aid of a funnel. The solvent system of chloroform and methanol (60:40) was carefully added from the top of the column by means of a funnel. Isocratic elution was further carried out to obtain the eluates. Eluates (20 mL each) were collected in test tubes marked fractions 1–19. The eluates with similar retention factor (Rf) values

were pooled together using a thin layer chromatographic procedure, which yielded three [3] subfractions, A (5.8 g), B (3.9 g) and C (3.4 g).

2.3. Determination of the total flavonoid content of flavonoid-rich fractions of *S. anguivi* fruits

Total flavonoids were analyzed using the aluminum chloride colorimetric method with slight modifications according to Ref. [20]. Quercetin was used to make the calibration curve. Ten [10] mg of quercetin was dissolved in 96 % ethanol and diluted to 2, 4, 6, 8 and 10 $\mu\text{g}/\text{mL}$. One [1] ml of each concentration of standard solutions as well as 1 ml of each sample solution were mixed with 3 mL 96 % ethanol, 0.2 mL 10 % aluminum chloride, 0.2 mL 1 M potassium acetate and 5.6 mL distilled water. The mixture was incubated at room temperature for 10 min with intermittent shaking. The absorbance of the resulting mixture was measured at 376 nm against a blank without aluminum chloride using a UV-Vis spectrophotometer. Total flavonoids were calculated as the mean \pm SD ($n = 3$) and expressed as the weight of quercetin equivalent (QE) in 100 mg extract.

2.4. Evaluation of in vitro antioxidant capacity

2.4.1. Assay of 2,2-diphenyl-1-picryl-hydrazyl radical scavenging ability

The method described by Ref. [21] was used to determine the 2,2-diphenyl-1-picrylhydrazyl scavenging effect of the flavonoid-rich fractions of *S. anguivi* fruit. One (1 ml) milliliter of flavonoid-rich fractions of *S. anguivi* fruit (15–240 $\mu\text{g}/\text{ml}$) or reference compound was added to a methanol solution of DPPH (0.4 mM). The mixture was left in the dark for 30 min before reading the absorbance at 517 nm. These tests were performed in triplicate using ascorbic acid as a reference. The radical scavenging ability was calculated as the percentage of DPPH discoloration as follows.

Calculation

$$\text{DPPH ability (\%)} = \frac{\text{Abs}(\text{sample}) - \text{Abs}(\text{blank})}{\text{Abs}(\text{blank})} \times 100$$

where A_{blank} is the absorbance with extract without DPPH.

A_{sample} is the absorbance of the reacting mixture with fractions and DPPH.

2.5. Assay of nitric oxide radical scavenging ability

The nitric oxide-scavenging activity of flavonoid-rich fractions of *S. anguivi* fruit was determined according to the method described by Ref. [22]. Various concentrations (15–240 $\mu\text{g}/\text{ml}$) of flavonoid-rich fractions of *S. anguivi* fruit and standards were prepared. Sodium nitroprusside (2.5 ml, 10 mM) in phosphate buffered saline (PBS) pH 7.4 was added to 0.5 ml of different concentrations of flavonoid-rich fractions of *S. anguivi* fruit and standard. The reaction mixture was incubated at 25 °C for 150 min. After incubation, 0.5 ml of Griess reagent (1 % (w/v) sulphanilamide, 2 % (v/v) H_3PO_4 and 0.1 % (w/v) naphthylethylenediamine hydrochloride) was added. Sodium nitroprusside in PBS (2 ml) was used as a control. The nitric oxide radical scavenging activity of the flavonoid-rich fractions of *S. anguivi* fruit and ascorbic acid was calculated according to the following equation.

Calculation

The nitric oxide radical scavenging activity was calculated as follows:

$$\text{Nitric oxide (NO) scavenging ability (\%)} = \frac{\text{Abs}(\text{control}) - \text{Abs}(\text{sample})}{\text{Abs}(\text{control})} \times 100$$

2.6. Ferric reducing antioxidant ability

The reducing power of flavonoid-rich fractions of *S. anguivi* fruit was estimated according to the method of [23]. Various concentrations (15–240 $\mu\text{g}/\text{ml}$) of flavonoid-rich fractions of *S. anguivi* fruit were dissolved in 2.5 ml (200 mM) sodium phosphate buffer (pH 6.6) and 2.5 ml of 1 % potassium ferricyanide. The mixtures were incubated at 50 °C for 20 min, and then 2.5 ml of 10 % TCA was added to terminate the reaction, followed by centrifugation at 3000 \times g for 10 min. Then, 5 ml of the supernatant was mixed with an equal volume of water and 1 ml of 0.1 % ferric chloride. The same treatments were performed with standard ascorbic acid solution, and the absorbance was read at 700 nm. The reducing power was calculated and expressed as ascorbic acid equivalent.

2.7. Total antioxidant capacity

The total antioxidant capacity of flavonoid-rich fractions of *S. anguivi* fruit was estimated according to the method of [24]. One (1 ml) milliliter of the reagent (28 mM sodium phosphate, 4 mM ammonium molybdate and 0.6 M sulfuric acid) and 100 μl (0.1 ml) of the various concentrations (15–240 $\mu\text{g}/\text{ml}$) of flavonoid-rich fractions of *S. anguivi* fruit were placed in test tubes. The test tubes were then capped using foil paper and incubated in a water bath at 95 °C for approximately 90 min. The samples were then cooled, and the absorbance was read at 695 nm.

2.8. Determination of enzyme inhibitory activities of *Solanum anguivi* flavonoid-rich fraction

The flavonoid-rich fractions of *S. anguivi* fruit were evaluated for their antidiabetic activity using *in vitro* assays, such as α -amylase and α -glucosidase inhibitory activity [25,26].

2.9. Determination of alpha-amylase inhibitory activity

A volume of 250 μ l of the samples at different concentrations (15–240 μ g/ml) was incubated in a water bath with 500 μ l of porcine pancreatic amylase (2 U m/L) in 100 mmol/L phosphate buffer (pH 6.8) at 37 °C for 20 min. A known volume (250 μ l) of 1 % starch dissolved in 100 mmol/L phosphate buffer (pH 6.8) was added to the reaction mixture and incubated at 37 °C for 1 h. Then, one ml of dinitrosalicylic acid (1 % 3,5-dinitrosalicylic acid, 0.2 % phenol, 0.05 % Na₂SO₃ and 1 % sodium hydroxide) color reagent was added and heated at 100 °C for 10 min. After cooling to room temperature in a cold water bath, the absorbance of the resulting mixture was read at 540 nm. Acarbose was used as a reference. The experiment was carried out in triplicate.

Calculation

The percentage inhibition was calculated as follows:

$$\alpha\text{-amylase inhibition (\%)} = \frac{A_{\text{control}} - A_{\text{sample}}}{A_{\text{control}}} \times 100$$

where A_{sample} is the absorbance of the sample and A_{control} is the absorbance of the control.

2.10. Determination of alpha-glucosidase inhibitory activity

One (1 mg) milligram of α -glucosidase (*Saccharomyces cerevisiae*, Sigma–Aldrich, USA) was dissolved in 100 ml of phosphate buffer (pH 6.8) containing 200 mg of bovine serum albumin. The reaction mixture consisting of 10 μ l of sample at varying concentrations (15–240 μ g/ml) of flavonoid-rich fractions of *S. anguivi* fruit was premixed with 490 μ l of phosphate buffer pH 6.8 and 250 μ l of 5 mM *p*-nitrophenyl α -D-glucopyranoside. After preincubating at 37 °C for 15 min. The reaction was terminated by the addition of 2000 μ l of 200 mM Na₂CO₃. α -Glucosidase activity was determined spectrophotometrically at 400 nm by measuring the quantity of *p*-nitrophenol released from *p*-NPG. Acarbose was used as a positive control for the α -glucosidase inhibitor.

Calculation

The percentage inhibition was calculated as follows:

$$\alpha\text{-glucosidase inhibition (\%)} = \frac{Abs(\text{control}) - Abs(\text{sample})}{Abs(\text{control})} \times 100$$

where A_{sample} is the absorbance of the sample and A_{control} is the absorbance of the control.

2.11. *Ex vivo* antioxidant study

2.11.1. Experimental animals

Healthy male Wistar rats (n = 5) weighing 150–180 g were used for *ex vivo* studies. They were obtained from the Animal House of the College of Medicine, Ekiti State University, Ado-Ekiti, Ekiti State. The animals were housed in aluminum cages under standard laboratory conditions (12-h light/dark cycle, 25 \pm 2 °C). They were fed rat feed (Top Feeds, a product of Premier Feed Mills of Nigeria PLC, Ogorode, Sapele, Delta State, Nigeria) and tap water *ad libitum*. The animals were acclimatized for two weeks before the start of the experiment. This study was approved, and an ethical clearance certificate (ORDI/AD/EAC/2023/118) was issued by the University Ethical Committee through the Office of Research, Development, and Innovation (ORDI), Ekiti State University.

2.11.2. Induction of tissue oxidative stress and treatment

After being fasted overnight, the rats were anesthetized with halothane, and the livers were harvested. The harvested livers were homogenized and centrifuged at 15,000 rpm and 40 °C. The supernatant was separated into tubes for *ex vivo* investigations. Oxidative stress was induced in isolated hepatic tissue as described by Ref. [27]. A total of 30 μ l of 15 mM ferrous phosphate (FeSO₄) was incubated with 100 μ l of different concentrations of the column chromatographic fractions (CCF) and 100 μ l of hepatic tissue homogenate for 30 min at 37 °C. Thereafter, catalase, superoxide dismutase activities, reduced glutathione levels, and MDA levels were determined.

2.11.3. Measurement of lipid peroxidation

The level of lipid peroxidation was measured according to a previously reported method by Ref. [28]. Then, 200 μ l of 8.1 % SDS solution, 750 μ l of 20 % acetic acid, 2 ml of 0.25 % TBA solution and 850 μ l of water were added to 200 μ l of column chromatographic fractions (CCF) with different concentrations (15–240 μ g/ml) and boiled for 1 h. The absorbance was measured at 532 nm after cooling. The MDA concentrations of the samples were calculated from the MDA standard curve.

2.11.4. Determination of catalase activity

Catalase activity was determined using a spectrophotometric method as described by Ref. [29]. The sample (100 μ l) was mixed and incubated with 1000 μ l of 65 μ M H₂O₂ in 6.0 mM sodium phosphate buffer, pH 7.4 at 37 °C for 2 min. The reaction was stopped by adding 4000 μ l of 32.4 mM ammonium molybdate, and the absorbance was measured at 347 nm against the reagent blank. A standard tube contained all the reagents except the tissue sample, and the control tube contained all the reagents except H₂O₂. Catalase activity was calculated using the following formula:

$$\text{CAT (kU)} = \frac{2.303}{t * \left[\frac{\log S_0}{s-M} \right] * V_t / V_s}$$

t: time; S^o: absorbance of standard tube; S: absorbance of test tube; M: absorbance of control test (correction factor); V_t: total volume of reagents in test tube; V_s: volume of test sample.

2.11.5. Determination of superoxide dismutase (SOD) enzyme activity

This assay was carried out according to the method of [30]. Samples were treated as above. Then, 170 μ l of 0.1 mM diethylenetriaminepentaacetic acid (DETAPAC) and 15 μ l of the infusion were placed in a 96-well plate. Then, 15 μ l of 1.6 mM 6-hydroxydopamine (6-HD) was added, and the mixture was quickly mixed by gently tapping all four sides of the plates. The absorbance was recorded at 492 nm for 3 min at 1 min intervals. SOD enzyme activity was calculated using the following formula:

$$\text{Activity} = (A_1 - A_b) / \epsilon_{490} * RV * Df / S_v$$

ϵ_{490} = Molar absorptivity at 490 nm = 1.742/mM/cm; A₁ and A_b = Reaction rate for sample and blank, respectively; RV = Reaction volume; Df = Dilution factor; S_v = Sample volume.

2.11.6. Determination of nitric oxide level

The level of nitric oxide (NO) in the tissues was determined using Griess' method as described previously [31]. Then, 100 μ l of the samples or distilled water (blank) was incubated with 100 μ l of Griess reagent at 25 °C in the dark for 30 min. Absorbance was read at 548 nm.

2.12. Data analysis

Data were expressed as the means \pm SD of three determinations. GraphPad Prism 9 (GraphPad Software Inc., San Diego, California, USA) was used to conduct statistical analysis on the data using one-way Analysis of variance (ANOVA), as well as the calculation of IC₅₀. The data were considered statistically significant at $p < 0.05$.

3. Results

3.1. Total flavonoid content

The total flavonoid contents of the flavonoid-rich fractions (A, B, and C) of *S. anguivi* were 15.53 \pm 0.75, 11.53 \pm 0.80, and 10.17 \pm 0.49 mg/g quercetin equivalent, respectively.

3.2. In vitro antioxidant activity

The flavonoid-rich fractions of *S. anguivi* showed significant ($p < 0.05$) antioxidant activities compared to ascorbic acid measured by DPPH, NO, ferric reducing antioxidant power (FRAP), and total antioxidant capacity methods, as depicted in Figs. 1–4. The flavonoid-rich fractions (A, B, and C) of *S. anguivi* significantly ($p < 0.05$) scavenged free radicals with increasing concentrations. The

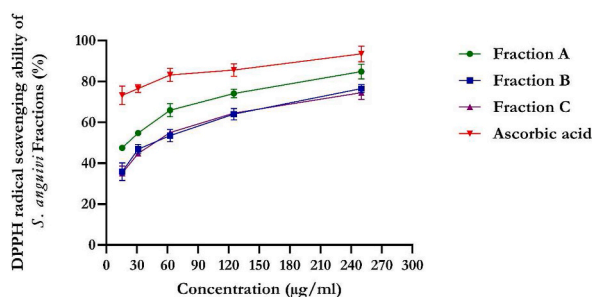


Fig. 1. DPPH scavenging activity of flavonoid-rich fractions of *S. anguivi* fruits. Values represent the mean \pm standard deviation (n = 3).

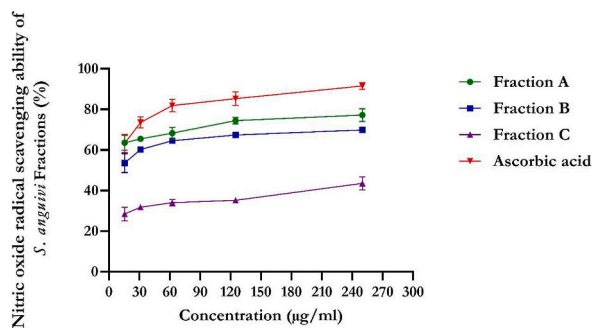


Fig. 2. NO scavenging activity of flavonoid-rich fractions of *S. anguivi* fruits. Values represent the mean \pm standard deviation (n = 3).

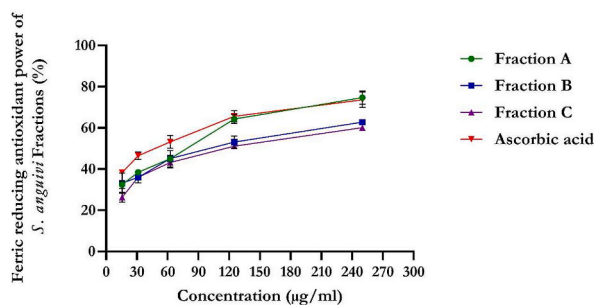


Fig. 3. Ferric reducing antioxidant power of flavonoid-rich fractions of *S. anguivi* fruits. Values represent the mean \pm standard deviation (n = 3).

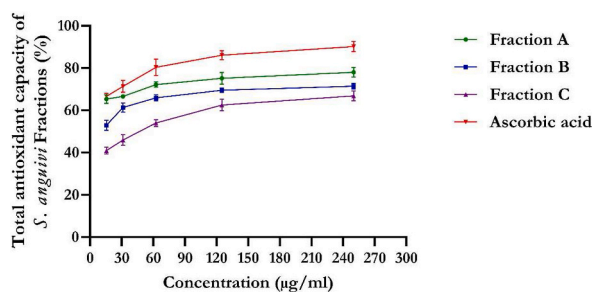


Fig. 4. Total antioxidant capacity of flavonoid-rich fractions of *S. anguivi* fruits. Values represent the mean \pm standard deviation (n = 3).

three fractions (A, B, and C) of *S. anguivi* fruits significantly scavenged both 2,2-diphenyl-1-picrylhydrazyl (DPPH) with fraction A having the lowest IC_{50} value ($26.14 \pm 1.06 \mu\text{g/ml}$) compared with fraction B ($37.78 \pm 5.12 \mu\text{g/ml}$) and fraction C ($38.24 \pm 2.40 \mu\text{g/ml}$) when compared with ascorbic acid with the least IC_{50} value ($15.27 \pm 0.34 \mu\text{g/ml}$). While fraction A ($19.61 \pm 1.19 \mu\text{g/ml}$) scavenged nitric oxide (NO) radicals better than fraction B ($22.97 \pm 0.55 \mu\text{g/ml}$) and fraction C ($49.95 \pm 6.18 \mu\text{g/ml}$). Although

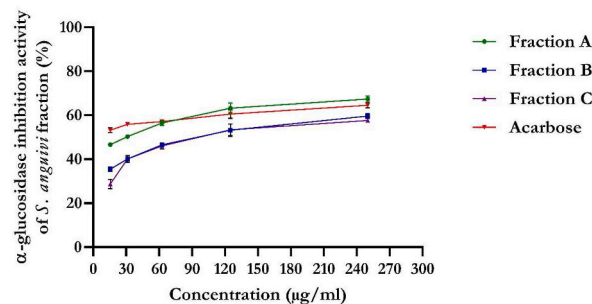


Fig. 5. α -Glucosidase activity of flavonoid-rich fractions of *S. anguivi* fruits. Values represent the mean \pm standard deviation (n = 3).

ascorbic acid had better scavenging ability than the three fractions ($17.23 \pm 0.16 \mu\text{g/ml}$). However, flavonoid-rich fraction A of *S. anguivi* showed the most potent antioxidant activity of all three fractions.

3.2.1. In vitro enzyme inhibitory activities

The flavonoid-rich fractions of *S. anguivi* fruits showed a remarkable inhibitory effect on α -glucosidase activity in a dose-dependent manner, as shown in Fig. 5. Flavonoid-rich fraction A of *S. anguivi* fruits showed better inhibitory activity ($\text{IC}_{50} = 16.24 \mu\text{g/ml}$) against α -glucosidase than fractions B ($\text{IC}_{50} = 128.04 \mu\text{g/ml}$) and C ($\text{IC}_{50} = 143.16 \mu\text{g/ml}$). The standard, acarbose, showed 64.57 % inhibition of α -glucosidase ($\text{IC}_{50} = 3.93 \mu\text{g/ml}$).

The flavonoid-rich fractions of *S. anguivi* fruits also led to a significant inhibitory effect ($\text{IC}_{50} = 31.50 \mu\text{g/ml}$) against α -amylase activity in a dose-dependent manner (Fig. 6) compared to flavonoid-rich fractions B ($\text{IC}_{50} = 84.32 \mu\text{g/ml}$) and C ($\text{IC}_{50} = 145.40 \mu\text{g/ml}$). The standard, acarbose, showed 70.88 % inhibition of α -amylase ($\text{IC}_{50} = 15.66 \mu\text{g/ml}$).

3.3. Ex vivo antioxidative studies

3.3.1. Lipid peroxidation

The inhibitory potentials of the flavonoid-rich fractions of *S. anguivi* fruits on lipid peroxidation are shown in Fig. 7. Upon induction of oxidative injury in hepatic tissue, there was an increase in MDA levels. However, treatment with the flavonoid-rich fractions of *S. anguivi* fruits led to a significant and dose-dependent reduction in MDA levels. Those treated with flavonoid-rich fractions A possesses a greater reduction in the inhibitory activity of the plant compared to the B and C fractions, which also show a reduction but not up to the control.

3.3.2. Superoxide dismutase enzyme activity

The reduced SOD activities in hepatic tissues further indicate an occurrence of oxidative injury upon incubation with FeSO_4 , as shown in Fig. 8. SOD enzyme activity was significantly increased in the flavonoid-rich fractions of *S. anguivi* fruit-treated tissues compared with the untreated tissue. The group treated with flavonoid-rich fraction A possesses a significant ($p < 0.05$) increase in the SOD activity of the liver compared to the fractions B and C.

3.3.3. Catalase activity

The reduced CAT activities in hepatic tissues further indicate an occurrence of oxidative injury upon incubation with FeSO_4 , as shown in Fig. 9. Flavonoid-rich fractions of *S. anguivi* fruits significantly reduced the amount of H_2O_2 remaining in Fe^{2+} -induced oxidative stress, as shown in Fig. 9, indicating enhanced catalase activity as it converted H_2O_2 to H_2O . The group treated with flavonoid-rich fraction A possesses a significant ($p < 0.05$) increase in the CAT activity of the liver compared to the fractions B and C.

3.3.4. Glutathione reduced (GSH) level

Incubation of hepatic tissues with FeSO_4 caused a significant ($p < 0.05$) reduction in GSH levels, indicating an incidence of oxidative injury, as shown in Fig. 10. Treatment with flavonoid-rich fractions of *S. anguivi* fruits significantly ($p < 0.05$) elevated the levels compared to ascorbic acid, with the highest concentration of flavonoid-rich fractions of *S. anguivi* fruits showing an increase that portrays potent activity. The flavonoid-rich fractions A extract treated group, possess a significant ($p < 0.05$) increase in the GSH activity of the liver compared to the fractions B and C.

3.4. Ex vivo inflammatory studies

3.4.1. Nitric oxide (NO) level

Induction of hepatotoxicity significantly ($p < 0.05$) elevated hepatic NO levels, as shown in Fig. 11. The dose-dependent NO depletion upon treatment with flavonoid-rich fractions of *S. anguivi* fruits may therefore indicate a pro-inflammatory-protective effect on iron-induced hepatotoxicity.

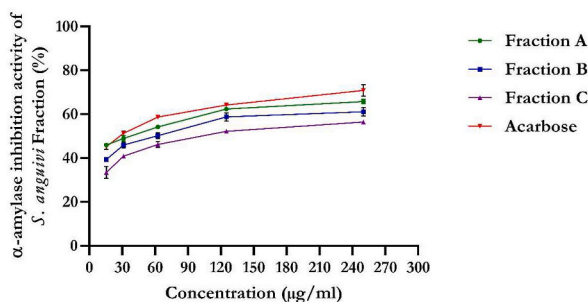


Fig. 6. α -amylase activity of flavonoid-rich fractions of *S. anguivi* fruits. Values represent the mean \pm standard deviation ($n = 3$).

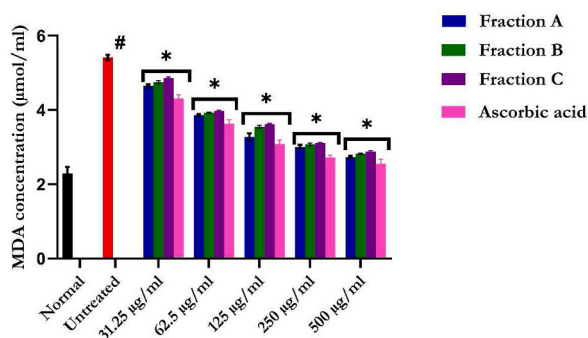


Fig. 7. Lipid peroxidation inhibitory activity of flavonoid-rich fractions of *S. anguivi* fruits. Values represent the mean \pm standard deviation ($n = 3$). *Statistically significant ($p < 0.05$) compared with untreated tissues; #statistically significant ($p < 0.05$) compared with normal tissues. Normal: liver tissues not treated with FeSO_4 and/or flavonoid-rich fractions of *S. anguivi* fruits. Untreated: liver tissues treated with FeSO_4 only.

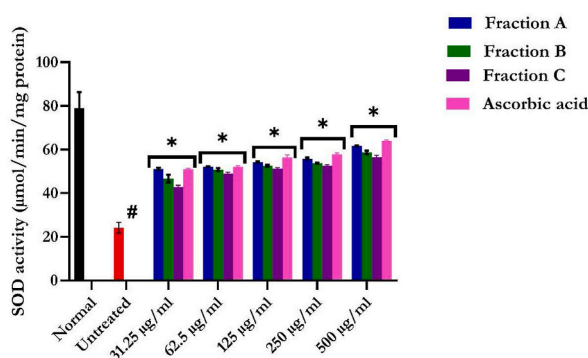


Fig. 8. Superoxide dismutase activity of flavonoid-rich fractions of *S. anguivi* fruits. Values represent the mean \pm standard deviation ($n = 3$). *Statistically significant ($p < 0.05$) compared with untreated tissues; #statistically significant ($p < 0.05$) compared with normal tissues. Normal: liver tissues not treated with FeSO_4 and/or flavonoid-rich fractions of *S. anguivi* fruits. Untreated: liver tissues treated with FeSO_4 only.

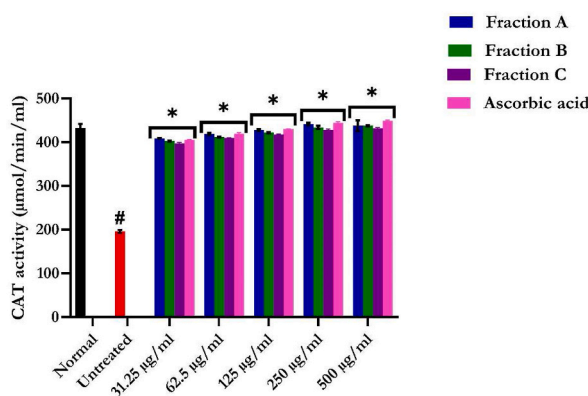


Fig. 9. Catalase activity of flavonoid-rich fractions of *S. anguivi* fruits in oxidative hepatic injury. Data are presented as the mean \pm SD ($n = 3$). *Statistically significant ($p < 0.05$) compared with untreated tissues; #statistically significant ($p < 0.05$) compared with normal tissues. Normal: liver tissues not treated with FeSO_4 and/or flavonoid-rich fractions of *S. anguivi* fruits. Untreated: liver tissues treated with FeSO_4 only.

4. Discussion

Flavonoids help regulate cellular activity and fight off radicals that causes oxidative damage on the body. Methanolic extract is a frequent choice for isolating flavonoids from *S. anguivi* fruits because compared to other solvents like water or ethanol, methanol can penetrate plant cell walls more efficiently, leading to better extraction of intracellular flavonoids [32]. In a study published by Ref. [33] found that a methanolic extract of *S. anguivi* fruits had the highest total phenolic and flavonoid content compared to other

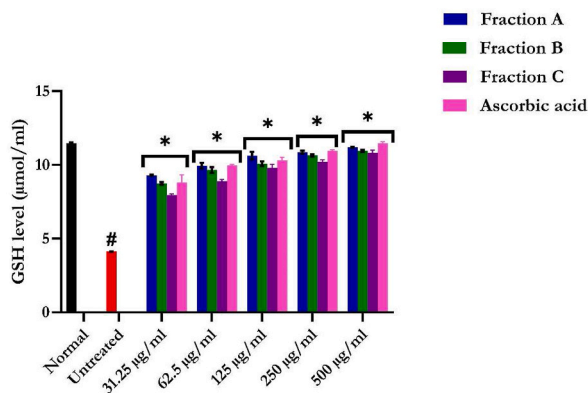


Fig. 10. Reduced glutathione (GSH) level of flavonoid-rich fractions of *S. anguivi* fruits in oxidative hepatic injury. Data are presented as the mean \pm SD (n = 3). *Statistically significant ($p < 0.05$) compared with untreated tissues; #statistically significant ($p < 0.05$) compared with normal tissues. Normal: liver tissues not treated with FeSO_4 and/or flavonoid-rich fractions of *S. anguivi* fruits. Untreated: liver tissues treated with FeSO_4 only.

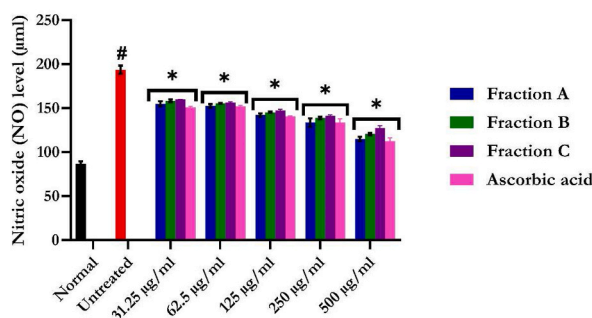


Fig. 11. Nitric oxide (NO) level of flavonoid-rich fractions of *S. anguivi* fruits in oxidative hepatic injury. Data are presented as the mean \pm SD; n = 3. *Statistically significant ($p < 0.05$) compared with untreated tissues; #statistically significant ($p < 0.05$) compared with normal tissues. Normal: liver tissues not treated with FeSO_4 and/or flavonoid-rich fractions of *S. anguivi* fruits. Untreated: liver tissues treated with FeSO_4 only.

solvents like water and ethanol. In addition, another study by Ref. [34] demonstrated the antioxidant and anti-inflammatory activity of a methanolic extract from *S. anguivi* fruits, attributed to its flavonoid content. Hence the reason why we selected the solvent methanol in extraction our flavonoid-rich extract of *S. anguivi* fruits for this study.

Several techniques have been used to determine the antioxidant activity *in vitro* to allow rapid screening of substances since substances that have low antioxidant activity *in vitro* will probably show little activity *in vivo* [35]. Free radicals are known to play a definite role in a wide variety of pathological manifestations. Antioxidants fight against free radicals and protect us from various diseases. They exert their action either by scavenging reactive oxygen species or protecting antioxidant defense mechanisms [36]. The electron donation ability of natural products can be measured by 2,2-diphenyl-1-picrylhydrazyl radical (DPPH) purple-colored solution bleaching [35]. A large decrease in the absorbance of the reaction mixture indicates significant free radical scavenging activity of the compound under test [37]. In the present study, among all the fractions tested, the flavonoid-rich fraction A of *S. anguivi* fruits showed a markedly higher inhibition percentage and was positively correlated with the total flavonoid content. This result also correlates with several other studies that carried out DPPH *in-vitro* analysis on flavonoid rich extracts, and they all show remarkable promising results [38–41]. The results of this study suggest that the flavonoid-rich fractions of *S. anguivi* have a hydrogen donating ability that stabilizes free radicals and scavenges potential damage.

Excessive nitric oxide concentration has been associated with diabetes mellitus. Flavonoid-rich fractions of *S. anguivi* fruits exhibited effectiveness in scavenging NO, which could be attributed to their antioxidant properties. The bioactive agents of *S. anguivi* compete with oxygen to react with nitric oxide, thereby suppressing nitrite generation. The results align with previously reported studies [42].

In the reducing power assay, the yellow color of the test solution changes to green depending on the reducing power of the test specimen. The presence of the reductants in the solution causes the reduction of the Fe^{3+} /ferricyanide complex to the ferrous form. Therefore, Fe^{2+} can be monitored by absorbance measurements at 700 nm. Previous reports have suggested that reducing power exerts antioxidant action by donating a hydrogen atom to break the free radical chain [40]. Increasing absorbance at 700 nm indicates an increase in reducing ability. The antioxidants present in fractions A, B, and C of *S. anguivi* caused their reduction of the Fe^{3+} /ferricyanide complex to the ferrous form and thus proved the reducing power. The present study demonstrated that the

flavonoid-rich fraction A of *S. anguivi* exhibited the highest total antioxidant capacity. Recent studies have shown that many flavonoids and related polyphenols contribute significantly to the antioxidant capacity of medicinal plants [43,44]. The scavenging abilities of the flavonoid-rich fractions of *S. anguivi* could serve in managing diabetes mellitus, as free radicals are involved in the pathogenesis of diabetes mellitus. This results correlates with previously reported studies on antioxidant activities of flavonoid-rich extract of *Syzygium cumini* and *Dalbergiella welwitschii* [44,45].

Inhibitors of the α -glucosidase and α -amylase enzymes involved in carbohydrate metabolism are a recognized technique in the management of diabetes mellitus for decreasing postprandial hyperglycemia [8]. Efficient control of the glycemic index in diabetes requires moderate levels of α -amylase inhibitors and potent levels of α -glucosidase inhibitors, as they aid in effectively regulating the available dietary sugar needed for absorption in the small intestine [46]. The inhibitory effect of the flavonoid-rich fractions of *S. anguivi* on the activity of α -glucosidase and α -amylase was shown to be concentration dependent and could serve as an indicator of the potential antidiabetic effect of the fractions.

The liver, as the main detoxifying organ, helps to maintain metabolic homeostasis. The organ metabolizes various compounds that produce ROS [47], causing liver parenchymal cells to be more vulnerable to oxidative stress. Disequilibrium in the levels of pro-oxidant and antioxidant causes oxidative stress, which has been linked to the pathogenesis and development of hepatic toxicity [48,49]. In this study, the effect of FeSO₄ on the liver caused oxidative injury and significantly reduced CAT and SOD activities, as well as GSH levels, with an accompanying increase in MDA levels, showing the development of oxidative stress and inflammation. When administered with flavonoid-rich fractions of *S. anguivi*, significant reversion of these alterations caused by FeSO₄ was observed, confirming the extract's resolving effect on hepatic oxidative stress. The most prominent hepatoprotective effect was evident in fraction A. This finding correlates with earlier studies on the flavonoid-rich fraction of *Dalbergiella welwitschii* leaf [45].

Proinflammation has also been related to the advancement of oxidative hepatic toxicity, with ROS generation identified as a major mechanistic link. The increased production of O₂•⁻ interacts with NO, resulting in the formation of peroxynitrite (ONOO⁻), a powerful free radical, as documented by Erukainure et al. [50]. The increased levels of NO observed in FeSO₄-exposed liver tissues show a pro-inflammatory effect. Hence, these results support the anti-inflammatory effect of flavonoid-rich fractions of *S. anguivi* in oxidative hepatic injury. Our findings are in agreement with earlier study published by Refs. [51,52].

5. Conclusion

The results emphasize the importance of investigating natural sources, such as flavonoid-rich fractions, for their potential therapeutic advantages in diabetes treatment. The study investigated different fractions of the flavonoid-rich fraction (A, B, and C) and also compared them with one another and standard drug. They all show promising effects, but the most potent one according to this study, is fraction A when it was compared to both B and C. The fractions were able to provide anti-inflammatory, anti-diabetic and antioxidant potentials to the studied organ. In addition, *S. anguivi* flavonoid-rich fractions could be used to treat oxidative hepatic injury in a variety of ways, all of which point to the therapeutic and protective potential of *S. anguivi* flavonoid-rich fractions for oxidative hepatic injury. More research into the therapeutic properties of *S. anguivi* and their flavonoid capacity should be studied further especially in the management of and treatments of diabetes.

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Data availability

The datasets used and/or analyzed during the current study are available from the corresponding author upon reasonable request.

CRedit authorship contribution statement

Adebola Busola Ojo: Writing – review & editing, Writing – original draft, Validation, Resources, Project administration, Methodology, Investigation, Funding acquisition, Data curation, Conceptualization. **Issac Gbadura Adanlawo:** Writing – review & editing, Writing – original draft, Supervision, Resources, Project administration, Methodology, Investigation, Funding acquisition, Conceptualization.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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