BRIEF REPORT

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CO₂ permeability of the rat erythrocyte membrane and its inhibition

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ABSTRACT

We have studied the CO₂ permeability of the erythrocyte membrane of the rat using a mass spectrometric method that employs ¹⁸O-labelled CO₂. The method yields, in addition, the intraerythrocytic carbonic anhydrase activity and the membrane HCO₃⁻⁻ permeability. For normal rat erythrocytes, we find at 37 °C a CO₂ permeability of 0.078 ± 0.015 cm/s, an intracellular carbonic anhydrase activity of 64,100, and a bicarbonate permeability of 2.1×10^{-3} cm/s. We studied whether the rat erythrocyte membrane possesses protein CO₂ channels similar to the human red cell membrane by applying the potential CO₂ channel inhibitors pCMBS, Dibac, phloretin, and DIDS. Phloretin and DIDS were able to reduce the CO₂ permeability by up to 50%. Since these effects cannot be attributed to the lipid part of the membrane, we conclude that the rat erythrocyte membrane is equipped with protein CO₂ channels that are responsible for at least 50% of its CO₂ permeability.

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Introduction

In 1998, Nakhoul et al. and Cooper and Boron^{1,2} have shown that the Xenopus oocyte membrane exhibits an increase in its permeability for CO₂ upon the incorporation of aquaporin-1 (AQP1), and Forster et al.³ demonstrated that the CO₂ permeability of the human red cell membrane decreases drastically under the exposure to 4,4'-diisothiocyanato-stilbene-2.2'-disulfonate (DIDS), which they interpreted as a highly effective inhibitory action of DIDS upon an as yet unidentified protein CO₂ channel in the erythrocyte membrane. Thereafter, Endeward and co-workers used human red cells deficient in AQP1 or in the Rhesus-associated glycoprotein (RhAG), to show that about 90% of the CO₂ permeability of the human red cell membrane is due to these two proteins acting as CO₂ channels⁴⁻⁶. They concluded that AQP1 and RhAG contribute about equally to this channel-mediated CO₂ pathway. Analogous studies have not been reported so far for any other mammalian species.

The aim of the present study was to investigate whether red blood cells of the rat are similarly equipped with a gas channelmediated CO_2 pathway. We chose the rat because this animal is relatively easily accessible to physiological studies assessing the exchange of CO_2 *in vivo*. We used the previously described mass spectrometric method to determine the CO_2 permeability of rat red cells in suspension^{7,8}. The red cells were exposed to several potential gas channel inhibitors, especially DIDS, but also Bis(1,3dibutylbarbituric acid)pentamethine oxonol, Dibac, to phloretin, and to Para-(chloromercuri)-benzenesulfonate, pCMBS. Two of these substances proved to be highly efficient inhibitors of the protein-mediated CO_2 pathway across the red cell membrane of the rat. In addition, we report here the first estimates of the intraerythrocytic carbonic anhydrase activity under physiological conditions of $37\,^\circ$ C, pH 7.2, and a CO₂ partial pressure of 40 mmHg.

Methods

Animals and blood samples

Blood samples were taken from 3 months old Lewis rats from the Central Animal Facility of the Hannover Medical School by tail venipuncture in accordance with local regulations for animal experimentation⁹. The blood was spun down at 5000 g for 20 min, plasma removed and cells washed three times in 0.9% NaCl. Haematocrit, cell count, and haemoglobin concentration were determined by standard techniques. Mean corpuscular volume (MCV) was \sim 63 fl, which is in agreement with previous reports^{10,11}. Rat erythrocyte surface area, which was needed in addition to mean corpuscular volume for calculation of P_{CO2} and P_{HCO3}⁻, was estimated from an established relation between red cell area and volume¹² to be 100 μ m². This may be compared to the published red cell surface areas published for mice and humans (90 μ m² or 147 μ m², respectively¹³). Neither of the transport inhibitors specified below and acting on membrane CO₂ permeability, namely phloretin and DIDS, had a significant effect on MCV after an exposure period of \sim 5 min; all MCV values varied between 62 and 65 fl. No spherocytes were observed either in controls or with inhibitors, all red blood cells exhibited the regular biconcave shape.

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Inhibitors

Any potential extracellular carbonic anhydrase activity resulting from red cell lysis that may occur during the mass spectrometric determination of P_{CO2} and P_{HCO3}^{-} was inhibited by the addition of the extracellular carbonic anhydrase inhibitor FC5-208A (2,4,6trimethyl-1-(4-sulfamoyl-phenyl)-pyridinium perchlorate salt)¹⁴ to the assay at a final concentration of 5×10^{-5} M. Thus, it was ensured that no extracellular carbonic activity was present during the mass spectrometric experiment with dilute red cell suspensions. Inhibition of channel-mediated membrane CO₂ permeability was attempted by the following chemicals: DIDS (4,4'-diisothiocyanato-stilbene-2.2'-disulfonate; Sigma-Aldrich, Seelze, Germany), which has previously been shown by us to be an efficient inhibitor of human red cell P_{CO2} as well as $P_{HCO3}{}^{3,4,5}\!;$ DiBAC (bis(1,3dibutylbarbituric acid)pentamethine oxonol; Invitrogen GmbH, Karlsruhe, Germany), which is an established inhibitor of the erythrocytic HCO_3^- -Cl⁻ exchanger¹⁵ but does not inhibit P_{CO_2} in human red cells⁴; pCMBS (para-(chloromercuri)-benzenesulfonate; Toronto Research Chemicals, North York, Canada; C367750), an established inhibitor of the aquaporin-1 water¹⁶ and CO₂^{2,5} channels; phloretin (Sigma-Aldrich, Merck KGaA, Darmstadt, Germany; P7912), which is known to inhibit red cell bicarbonate-chloride exchange besides the transport of several other substrates¹⁷.

Determination of CO₂ and HCO₃⁻ permeabilities

We have previously reported how the CO₂ permeability of plasma membranes can be determined for red cells or other cells in suspension using a mass spectrometric method^{4,5,7,8}. In principle, cells are exposed to a solution of $C^{18}O^{16}O/HC^{18}O^{16}O_2^{-}$ that is labelled with ¹⁸O to a degree of 1%. In this solution, C¹⁸O¹⁶O and $HC^{18}O_2^{16}O_2^{-}$ react with water or H^+ , thereby transferring by a defined probability the label ¹⁸O from the CO_2 -HCO₃⁻ pool into the much larger pool of water. This reaction is slow, but inside red cells due to their high carbonic anhydrase activity becomes much faster. The exchange of ${}^{18}O$ from CO_2 -HCO₃⁻ into the water pool causes a decay of the species C¹⁸O¹⁶O (mass 46), and we observe this decay vs. time after the start of the exposure of the cells to the solution. In a first rapid phase, the carbonic anhydrase-containing cells rapidly take up C¹⁸O¹⁶O. The kinetics of this process depends on the permeability of the membrane to CO₂ and on the speed of the intracellular conversion of CO₂ to HCO₃⁻, that is, on intracellular carbonic anhydrase activity. The rate of disappearance of C¹⁸O¹⁶O from the extracellular fluid is followed by a mass spectrometer equipped with a special inlet system for fluids as first described by Itada and Forster¹⁸.

Examples are shown in Figure 1. From the time course of the rapid first phase of the disappearance of $C^{18}O^{16}O$ (see Figure 1), the membrane permeability for CO_2 can be calculated, if the intracellular carbonic anhydrase activity has been determined independently⁷. After the first rapid phase of the mass spectrometric record, a slower phase follows (also seen in Figure 1), which is to a major extent determined by the transport $HC^{18}O^{16}O_2^{-}$ across the membrane. Thus, this second phase allows one to determine membrane HCO_3^{-} permeability⁷. For a complete review of the method see⁸.

To perform mass spectrometric measurements, a chamber with a volume of 2.2 ml was used that was attached to the high vacuum of the mass spectrometer via the previously published inlet system⁷. This chamber had a water jacket that kept the solutions at 37° C and contained a magnetic stirrer mixing the content continuously. The solution in the chamber consisted of 110 mM NaCl,



Figure 1. Time course of the decay of ¹⁸O in CO₂ vs. time for rat red cells in the presence and absence of DIDS. Y-axis is log ($10^{7} \cdot (\Delta[CO_2^*])$), where $\Delta[CO_2^*]$ is the concentration of ¹⁸O-labelled CO₂ minus its final value at isotope equilibrium, in the unit 10^{-7} M. The Y-axis gives the logarithm of this value after it has been multiplied by 10^{7} . The curve shows three phases: (1) a pre-phase representing the slow uncatalysed decay of ¹⁸O-labelled CO₂, (2) by adding, at the sharp bend in the curve, red cells into the measuring chamber the next phase is initiated, which we call the rapid first phase after addition of red cells and which is strongly dependent on P_{CO2}, (3) a second slower phase follows, which is dependent on P_{HCO3}⁻⁻. During the mass spectrometric measurement of red cells an extracellular pH of 7.4 and a CO₂ partial pressure of 40 mmHg prevail.

20 mM HEPES and 25 mM NaHCO₃ labelled with ¹⁸O to the extent of 1%. pH was adjusted to 7.4 and monitored during the entire measurement with a pH electrode. The reaction was started by adding $10\,\mu$ l of red cell suspension with a haematocrit of 5% into the reaction chamber. In all experiments 5×10^{-5} M FC5-208A was present. Red cell suspensions were incubated with this carbonic anhydrase inhibitor for 2-5 min before use in the mass spectrometer experiment. Likewise, if any of the above membrane transport inhibitors were used, they were incubated with red cells 2-5 min prior to the red cells' addition into the mass spectrometer's reaction chamber. For all inhibitors, it was ascertained that the same inhibitor concentration used for preincubation also existed in the bicarbonate buffer present in the reaction chamber. The same preincubation times for inhibitors have previously been used for human red cells^{4,5}. Carbonic anhydrase activity was measured in suitably diluted lysed red cells, which gives in a plot like that of Figure 1 a mono- rather than a biphasic response³. The inhibitor-free lysates had a pH of 7.2 and a Cl⁻ concentration of 63 mM, mimicking the intraerythrocytic conditions. The acceleration factor describing the increase in slope caused by the addition of the lysate was used to calculate the intraerythrocyic carbonic anhydrase activity from (acceleration factor - 1) \times dilution factor³.

Results

Intraerythrocytic carbonic anhydrase activity

Intraerythrocytic carbonic anhydrase activity (Ai) was determined at pH 7.2 from mass spectrometry of lysed red cells. Activity was determined for the blood sample from each animal studied, and used to calculate P_{CO2} and P_{HCO3}^{-} from the data for this same blood's red cells using the system of equations that describe the entire ¹⁸O-exchange process⁷. On average, Ai was 64,100±6200 (SD; n = 31). Ai was defined as (acceleration factor -1), where the



Figure 2. (a) CO₂ permeability of rat red cells at 37 °C and effect of four inhibitors. Number of blood samples *n* from left to right: 36, 9, 8, 8, 10, 10. *Indicates statistical significance of difference to control (one-way ANOVA followed by Dunnett's post-test). Inhibitor concentrations given are in mol/l. pCMBS was used at a concentration of 1 mM, phloretin at 300 µM, and Dibac at 5×10^{-7} M. (b) HCO₃⁻ permeability of rat red cells at 37 °C and effect of four inhibitors. n from left to right: 36, 8, 8, 10, 10. *Indicates statistical significance of difference to control (one-way ANOVA followed by Dunnett's post-test). pCMBS was used at a concentration of 1 mM, phloretin at 300 µM, and Dibac at 5×10^{-7} M.

acceleration factor is the factor by which the speed of CO_2 hydration in the undiluted red cell interior was accelerated over the uncatalysed rate.

Membrane CO₂ permeability

Figure 2(a) shows the results for P_{CO2} of the rat erythrocyte membrane at 37 °C. The control value in the absence of potential gas channel inhibitors is 0.078 ± 0.015 cm/s (SD; n = 36; left-hand column). The other columns of Figure 2(a) show the effect of various inhibitors on P_{CO2} . 300 µM phloretin and DIDS at concentrations both of 10^{-5} and 10^{-4} M significantly reduce CO₂ permeability. In the presence of 10^{-4} M DIDS, P_{CO2} amounts to 0.037 ± 0.006 cm/s (n = 10), that is, to about 1/2 of the control value. 1 mM pCMBS has no effect on P_{CO2} , and 5×10^{-7} M Dibac, a rather specific inhibitor of the erythrocytic Cl⁻-HCO₃⁻ exchanger AE1¹⁵, has a very minor, although statistically significant, effect on P_{CO2} (0.071 ± 0.006 cm/s; n = 8). In conclusion, two of the four inhibitors used reduce P_{CO2} to a major extent.

Membrane HCO₃⁻ permeability

Figure 2(b) shows the effects of the same group of inhibitors on P_{HCO3}^{-} . The control value of P_{HCO3}^{-} amounts to 0.0021 ± 0.0002 cm/s (n = 36; left-hand column). 1 mM pCMBS has again no significant effect, but all other inhibitors including Dibac have major and statistically significant effects. With $5 \cdot \times 10^{-7}$ M

Dibac, P_{HCO3}^{-} is reduced to 0.00091 ± 0.00011 cm/s (n = 8), that is, to less than 1/2 of the control value.

Discussion

Red cell carbonic anhydrase activity in rats

We report here an intraerythrocytic carbonic anhydrase activity of 64,100, which may be compared with the value of 21,000 we obtained for human red cells with the same method under identical conditions⁴. Thus, rat red cells have a 3 times higher Ai compared to human red cells. While the absolute numbers of these activities cannot be compared with the results obtained by Larimer and Schmidt-Nielsen¹⁹ due to highly different methods and conditions, it may be noted that these authors similarly found a 2.5 times higher Ai in rats compared to humans. In general, these authors observed a tendency of Ai to increase with decreasing body weight in mammals. The functional significance of the higher Ai values in smaller animals is not clear. Larimer and Schmidt-Nielsen speculated that the high carbonic anhydrase activity in small animals with their higher specific rate of oxygen consumption may improve oxygen release from red cells by an accelerated production of H⁺ from CO₂, thus accelerating the contribution of the Bohr effect to O₂ release¹⁹. Their studies have been discussed and updated in a more recent review²⁰.

CO₂ permeability of the red cell membrane

P_{CO2} value

We report here a P_{CO2} of red cells of the rat of 0.078 cm/s at 37 °C. This value is half of the value reported previously for human red cells, 0.15 cm/s^{4,5}. It must be considered, on the other hand, that the surface-to-volume ratio of rat red cells is >10% greater than that of human red cells, and that the rat red cell thickness is estimated to be 10% less than that of human red cells, both factors accelerating the speed of uptake of gases by rat red cells. This may lead to a gas uptake kinetics that overall is similar in rat and human red cells. This corresponds to the similar lung capillary transit times that have been reported for both rats and humans^{21,22}.

Effects of inhibitors and role of CO2 channels

None of the potential gas channel inhibitors studied here is expected to have an effect on the CO₂ permeability of the pure phospholipid bilayer. It has been shown that DIDS has no effect on P_{CO2} of pure phospholipid vesicles and cell membranes without gas channels^{23,24}, but lowers P_{CO2} in both types of membranes when gas channels are present^{4,5,23,25}. Similarly, it has been shown that pCMBS does not affect the P_{CO2} of membranes in the absence of gas channels, but it does reduce P_{CO2} when the gas channel aquaporin 1 is present^{2,5}. No such direct evidence is available for Dibac and phloretin, but there is overwhelming evidence in the literature that these inhibitors suppress specifically the transport activity of the anion exchanger AE1 (Dibac¹⁵) or affect a wide range of transport proteins (phloretin¹⁷), respectively, but not the phospholipid membrane per se. We conclude that both DIDS and phloretin reduce membrane P_{CO2} by acting on a protein CO₂ channel present in the membrane of rat erythrocytes. The maximal inhibition of P_{CO2} achieved here amounts to 50% of the control P_{CO2} . This suggests that at least 50% of the P_{CO2} of rat red cells is due to a gas channel. This is a little less than has been

reported for normal human red cells, where P_{CO2} fell by DIDS to about 1/3 of the control value^{5,6}.

Our previous work has shown that Dibac does not affect P_{CO2} in human erythrocytes, although it drastically reduces P_{HCO3}^{4} . This is not unexpected as Dibac is a highly specific inhibitor of AE1¹⁵. Both findings are thus in line with the present observation of no major inhibition of P_{CO2} in rat erythrocytes by Dibac. However, we find here also no effect of pCMBS on the P_{CO2} of rat red cells, while previously we found a moderate suppressive effect of 1 mM pCMBS on P_{CO2} of human red cells. Importantly, this effect was substantially less pronounced than the effect of DIDS⁵. It may be noted that neither DIDS nor pCMBS permeate the red cell membrane and thus cannot affect intraerythrocytic carbonic anhydrase^{4,5}.

A reasonable hypothesis to explain these observations comes from the following considerations:

- A. In human red cells, there are two major CO_2 channels, AQP1 and RhAG, that contribute about equally to the CO_2 permeability of the membrane and together are responsible for about 90% of the total CO_2 permeability^{4,5}.
- B. It is clear that DIDS inhibits quite effectively both AQP1 and RhAG CO_2 channels, perhaps RhAG even a little more markedly than AQP1⁶.
- C. pCMBS inhibits P_{CO2} of human red cells clearly less than does DIDS, and this inhibitory effect seems to be exclusively due to AQP1, because in AQP1-deficient human red cells, whose only CO₂ channel is RhAG, an effect of pCMBS is not detectable⁵.
- D. According to molecular dynamics studies, the central pore of the aquaporin 1 tetramer is the major pathway of CO₂ across aquaporin 1, and the water channel of the aquaporin 1 monomer is a minor CO₂ pathway^{26,27}.
- E. Thus, it seems likely that the inhibitory effect of pCMBS we see on P_{CO2} of human red cells is mainly due to inhibition of the water (and CO₂) channel of monomeric AQP1. It is well known that pCMBS, by acting on this pathway, largely blocks its water as well as its CO₂ permeability via binding to a cysteine in position 189 in the water pore^{2,28,29}. pCMBS does not act on the RhAG CO₂ pathway and likely does not act on the central pore of the aquaporin tetramer. It is probable that the properties of human AQP1 discussed under C) to E) apply analogously to rat AQP1.
- F. In rat erythrocytes it is well established that AQP1 as well as RhAG are present in the membrane^{30–32}. Unfortunately, to our knowledge there is no quantitative information on the abundance of AQP1 and RhAG expression in rat red cells.

Overall we draw the following conclusions:

- 1. Rat, as well as human red cells, express CO_2 channels in their membrane, which constitute a sizable part of the membrane CO_2 permeability. These channels can very effectively be inhibited by DIDS and also, to a lesser extent, by phloretin. The effect of DIDS is qualitatively compatible with results obtained for human red cells, suggesting that AQP1 and RhAG represent the CO_2 channels of rat erythrocytes. Likewise, the effect of phloretin reported here is similar to that observed in human red cells (own unpublished observations).
- 2. However, the fact that the effect of DIDS on P_{CO2} is smaller in comparison to human red cells, suggests that the role of CO₂

channels in the rat may be somewhat less pronounced than it is in human red cells.

- 3. This would be in agreement with the present observation of a lack of an effect of pCMBS on P_{CO2} , which might indicate that it is AQP1, which is expressed to a lesser degree in rat erythrocytes. However, conclusive evidence on this last point is lacking. An alternative explanation might a lower cholesterol content of the rat erythrocyte membrane, which would increase the membrane's intrinsic permeability to CO_2^{23} and thus render the effect of CO_2 channels less visible. However, the cholesterol content of the rat erythrocyte membrane has not been reported, and also, the CO_2 permeabilities seen here for rat red cells are lower than those for human red cells.
- 4. Instead, the lower P_{CO2} of rat red cells might at least partially – be caused by the reduced role of CO_2 channels in rats compared to human erythrocytes.

HCO₃⁻ permeability of the red cell membrane

Applying the present mass spectrometric technique, we have previously reported the bicarbonate permeability of normal human red blood cells to be $1.2-1.3 \times 10^{-3}$ cm/s at $37 \circ C^{4-6}$. The present control value for rat red cells is 2.1×10^{-3} cm/s at the same temperature. This may be surprising in the view of the P_{CO2} that is lower for rat red cells than it is for human red cells (see above). It may be considered that in the case of HCO₃⁻ uptake or release, the membrane constitutes the major diffusion resistance since P_{HCO3}^{-} is almost 50 times lower than P_{CO2} , and since the intracellular diffusion of HCO₃⁻ in comparison is much faster (albeit almost equally fast as the intracellular diffusion of CO₂). If then the product of P x cellular surface area is decisive for the kinetics of HCO_3^{-} , we would, with the numbers given by¹³ and above, obtain products of 1.25×10^{-3} cm/s \times 146 μ m² for humans and of 2.1×10^{-3} cm/s $\times 100 \ \mu\text{m}^2$ for rats, respectively. In other words, these products agree within about 15%, and, as in the case of CO₂ uptake kinetics, we thus expect similar kinetics of HCO₃⁻ exchange in both rat and human red cells. The inhibitory profile of P_{HCO3}⁻ seen in Figure 2(b) with strong effects of Dibac, phloretin and DIDS is as expected from the literature, as detailed in the Methods section.

Outlook

We present clear evidence that the red cell membrane of the rat possesses protein channels for CO₂, which can effectively be inhibited by DIDS and by phloretin. In studies on the role of gas channels for red cell CO₂ exchange, the difficulty may arise that both these inhibitors inhibit CO₂ permeation as well as HCO₃⁻ permeation. Both these processes contribute to overall CO₂ exchange, although with quite different kinetics. This problem can, at least *in vitro*, be solved by complementing studies with DIDS or phloretin with experiments using Dibac, which very efficiently inhibits P_{HCO3}⁻ without affecting P_{CO2}. This should allow one to dissect the effects from inhibition of P_{CO2} from those resulting from inhibition of P_{HCO3}⁻.

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Disclosure statement

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