

Communication



# Deciphering the Disaggregation Mechanism of Amyloid Beta Aggregate by 4-(2-Hydroxyethyl)-1-Piperazinepropanesulfonic Acid Using Electrochemical Impedance Spectroscopy

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**Abstract:** Aggregation of amyloid- $\beta$  ( $\alpha\beta$ ) peptides into toxic oligomers, fibrils, and plaques is central in the molecular pathogenesis of Alzheimer's disease (AD) and is the primary focus of AD diagnostics. Disaggregation or elimination of toxic aß aggregates in patients is important for delaying the progression of neurodegenerative disorders in AD. Recently, 4-(2-hydroxyethyl)-1piperazinepropanesulfonic acid (EPPS) was introduced as a chemical agent that binds with toxic  $a\beta$  aggregates and transforms them into monomers to reduce the negative effects of  $a\beta$  aggregates in the brain. However, the mechanism of  $\alpha\beta$  disaggregation by EPPS has not yet been completely clarified. In this study, an electrochemical impedimetric immunosensor for  $a\beta$  diagnostics was developed by immobilizing a specific anti-amyloid- $\beta$  (a $\beta$ ) antibody onto a self-assembled monolayer functionalized with a new interdigitated chain-shaped electrode (anti- $a\beta$ /SAM/ICE). To investigate the ability of EPPS in recognizing AD by extricating  $\alpha\beta$  aggregation, commercially available  $\alpha\beta$ aggregates  $(a\beta_{agg})$  were used. Electrochemical impedance spectroscopy was used to probe the changes in charge transfer resistance (Rct) of the immunosensor after the specific binding of biosensor with a  $\beta_{agg}$ . The subsequent incubation of the a  $\beta_{agg}$  complex with a specific concentration of EPPS at different time intervals divulged AD progression. The decline in the R<sub>ct</sub> of the immunosensor started at 10 min of EPPS incubation and continued to decrease gradually from 20 min, indicating that the accumulation of  $a\beta_{agg}$  on the surface of the anti-a $\beta/SAM/ICE$  sensor has been extricated. Here, the kinetic disaggregation rate k value of  $a\beta_{agg}$  was found to be 0.038. This innovative study using electrochemical measurement to investigate the mechanism of  $a\beta_{agg}$  disaggregation by EPPS could provide a new perspective in monitoring the disaggregation periods of  $a\beta_{agg}$  from oligometric to monomeric form, and then support for the prediction and handling AD symptoms at different stages after treatment by a drug, EPPS.

**Keywords:** Alzheimer's disease; amyloid beta aggregate; 4-(2-hydroxyethyl)-1-piperazinepropanesulfonic acid; impedimetric immunosensor; electrochemical impedance spectroscopy; kinetic disaggregation of protein

## 1. Introduction

The deposition of the amyloid- $\beta$  ( $\alpha\beta$ ) protein in the cortex of the brain is reported as a leading cause of Alzheimer's disease (AD), the most common form of dementia [1]. According to the amyloid cascade hypothesis,  $\alpha\beta$  peptides tend to aggregate and coalesce into toxic insoluble oligomers and fibrils in plaques, which may cause decline in synaptic function and nerve cell damage [2,3].

Aggregation of  $\alpha\beta$  gradually increases throughout the brain, which leads to neurodegeneration, starting from mild to moderate symptoms in earlier Alzheimer's stages, and finally dementia [3,4]. Many efforts have been made to control and prevent Alzheimer's progression by inhibiting and reducing the aggregation of  $\alpha\beta$ . For instance, laser-induced



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**Copyright:** © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). destruction [5], ultrasonic treatment [6], temperature-induced dissociation [7], pulsed radio-frequency cold atmospheric plasma jet [8], and drug treatments [9,10] have all been used to eliminate the aggregation of a $\beta$ . Among these approaches, drug treatment is the most-studied means of reducing or eliminating a $\beta$  aggregation, as these agents provide the most practical strategy to target the a $\beta$  aggregates within the body. Some natural chemical compounds such as theaflavins from black tea, epigallocatechin 3-gallate from green tea, and components from coffee showed an inhibitory effect on the aggregation of a $\beta$  [11–13]. However, the precise mechanism of interaction between these agents and a $\beta$  aggregates remains uncertain, and none of these have been used to treat AD clinically.

Recently, 4-(2-hydroxyethyl)-1-piperazinepropanesulfonic acid (EPPS) has been acknowledged as a promising drug candidate for dissociating and converting a $\beta$  aggregates into monomers. A study on a mouse model of Alzheimer's a $\beta$  plaques found that EPPS can contribute to the breakdown of the plaques, which can alleviate some of the AD symptoms in mice [14]. Despite its significance, the detailed mechanism of EPPS on the disaggregation of a $\beta$  aggregates has not been thoroughly elucidated.

Electrochemical immunosensors offer rapid response, high sensitivity, and selective detection of a $\beta$  at very low concentrations [15–18]. In this report, we created an impedimetric immunosensor for the diagnosis of a $\beta$  aggregates (a $\beta_{agg}$ ) and a $\beta$  peptide (a $\beta_{pep}$ ). It was produce by the immobilization of a specific anti-a $\beta$  antibody on a self-assembled monolayer functionalized with a new interdigitated chain-shaped electrode (anti-a $\beta$ /SAM/ICE). This was then used to monitor the effect of EPPS on the disaggregation a $\beta_{agg}$ . After the anti-a $\beta$ /SAM/ICE biosensor bound with a $\beta_{agg}$ , forming the a $\beta_{agg}$  complex, the a $\beta_{agg}$  complex was then incubated with a specific concentration of EPPS at different time intervals to divulge the disaggregation progress of a $\beta_{agg}$ . Electrochemical impedance spectroscopy (EIS) was used as a tool to confirm the changes in charge transfer resistance (R<sub>ct</sub>) of the immunosensor after forming the complex and EPPS treatment. We established a new method for studying the mechanism of EPPS-induced kinetic disaggregation of a $\beta_{agg}$  by analyzing the EIS results from this study, leading to new insights in the diagnostics and treatment of AD symptoms at various levels.

# 2. Materials and Methods

# 2.1. Materials

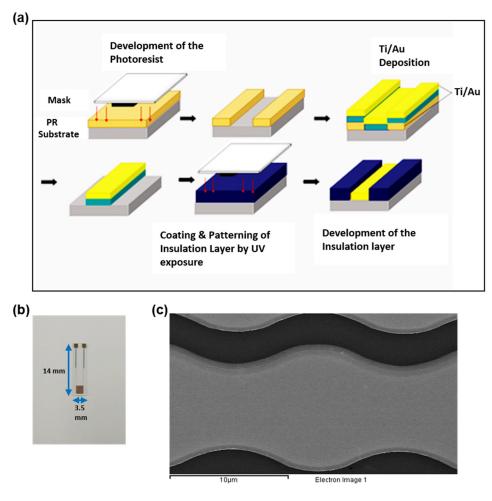
AggreSure<sup>TM</sup> Amyloid- $\beta$  1–42 aggregate ( $a\beta_{agg}$ ) is bought from AnaSpec (Fremont, CA 94555, USA). Anti- $a\beta$  antibody ab126649 (anti- $a\beta$ ), Amyloid beta peptide 1–42, human, ab120301 ( $a\beta_{pep}$ ) are bought from Abcam (Seoul, Korea).

De-ionized (DI) water is gained from Milli-Q system. Phosphate buffer saline (PBS, pH 7.4) is received from Tech-Innovation.

4-(2-hydroxyethyl)-1-piperazinepropanesulfonic acid (EPPS, BioXtra,  $\geq$ 99.5% [titration]), EDC hydrochloride (crystalline,  $\geq$ 98.0%), 6-mercaptohexanoic acid (MHA, 90%), 1-hydroxy-2,5-pyrrolidinedione (NHS, 98.0%), potassium ferrocyanide/ferricyanide ([Fe(CN)<sub>6</sub>]<sup>3-/4-</sup>), bovine serum albumin (BSA, 0.5% in 1 × PBS, pH 7.4) are obtained from Sigma Aldrich (Seoul, Korea).

# 2.2. Development of the Immunosensor for Monitoring the Impact of EPPS on the Disaggregation of $a\beta_{agg}$

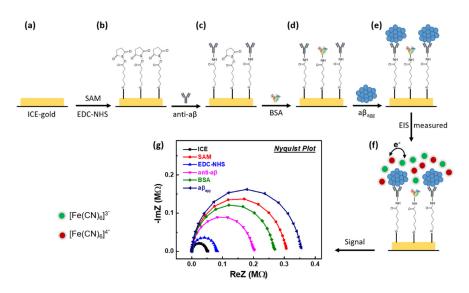
On a glass transparency substrate ( $13.5 \times 16.0 \text{ mm}$ ), an interdigitated chain-shaped electrode (ICE) is made. Ti and Au have been deposited in an electron beam-evaporator with electrodes 25 nm and 50 nm of thickness. The lift-off procedure then formed the coupled electrode finger consisting of 5 µm distance and width for working and reference electrodes. The ICE was then cleaned with 100% ethanol, DI water, and dried in nitrogen gas flow to prepare for the development of the immunosensor. Schematic of the fabrication and microscope image of ICE are displayed in Figure 1.



**Figure 1.** (a) Fabrication of ICE, (b) Photograph of ICE, and (c) Scanning electron microscope image of chain-shaped gold finger of ICE.

The development of the immunosensor for diagnosis of  $a\beta_{agg}$  is displayed in Figure 2. A self-assembled monolayer (SAM) has been mounted on the purified ICE gold surface in 100 mM MHA for 24 hours at room temperature. A solution of 75 mM EDC and 5 mM NHS was then incubated with the SAM-modified electrode to activate the carboxyl group for antibody binding. Then, ten anti-a $\beta$  antibody microliters (100 µg·mL<sup>-1</sup>) in PBS (pH ~ 7.4) are dripped off into a modified electrode and incubated for 1 hour in a moist compartment at 4 °C. Therefore, the EDC/NHS-activated SAM molecules on the modified electrodes were attached to the anti-a $\beta$  antibody through the amino group. After that, nonspecific adsorption was blocked by incubating with BSA (0.5% in 1 × PBS, pH 7.4) for 20 min at 4 °C to develop the anti-a $\beta$ /SAM/ICE biosensor. The developed anti-a $\beta$ /SAM/ICE biosensor was then incubated with a $\beta_{agg}$  (in PBS (pH ~ 7.4)) for 20 min at room temperature for the recognition of a $\beta_{agg}$ .

To monitor the effect of EPPS on the kinetics disaggregation of  $a\beta_{agg}$ , the antia $\beta$ /SAM/ICE biosensor was incubated with  $a\beta_{agg}$  (2.5 × 10<sup>-3</sup> mg·mL<sup>-1</sup>) for 20 min to enable a specific binding between the antibody and antigen, called the  $a\beta_{agg}$  complex. Subsequently, the  $a\beta_{agg}$  complex was incubated with 20 mM·L<sup>-1</sup> of EPPS for various time periods to study the disaggregation process of  $a\beta_{agg}$  by electrochemical impedance measurement.



**Figure 2.** (**a**–**e**) Development procedure the immunosensor anti- $a\beta/SAM/ICE$  for the recognition  $a\beta_{agg}$ , (**f**) impedimetric measurement mechanism in the recognition  $a\beta_{agg}$  by the developed biosensor anti- $a\beta/SAM/ICE$ , and (**g**) impedimetric results represent by Nyquist plot at each step of the development anti- $a\beta/SAM/ICE$  biosensor in the recognition  $a\beta_{agg}$ .

#### 2.3. EIS Measurements and Functional Group Characterization

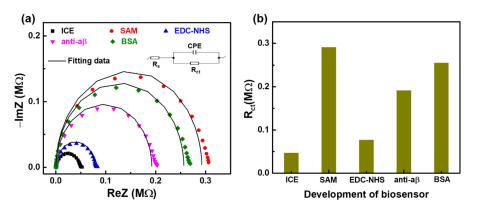
EIS was performed using EC-Lab (SP-200, Bio-Logic, France) in 5 mM  $[Fe(CN)_6]^{3-/4-}$ in PBS (pH 7.4) at each step of the anti-a $\beta$ /SAM/ICE biosensor development and various time intervals of the disaggregation process of a $\beta_{agg}$ . The EIS data was obtained by putting on a 10 mV of sinus amplitude which satisfy the linear properties of electrical response of the object between the working and reference electrode from 100 mHz to 1 MHz of range frequency. Z-Fit software (EC-Lab, Bio-Logic, France) was used to fit the obtained EIS data through by the Randle's equivalent circuit.

Fourier-transform infrared (FT-IR) spectra was characterized by using Jasco-4600 FTIR (USA) spectrometer to confirm the functional groups for the fabrication steps of the anti- $a\beta$ /SAM/ICE biosensor.

# 3. Results and Discussion

## 3.1. Investigating the Development of Impedimetric Immunosensor

As shown in Figure 3a, EIS measurements were carried out to determine the Nyquistplots with the imaginary impedance component (-Im(Z)) plotted with the actual impedance component (Re(Z)) to confirm the successful modification of substantial ingredients to the electrode surface at different stages of the development the anti- $a\beta$ /SAM/ICE immunosensor.



**Figure 3.** EIS results at each step of the development of biosensor anti-aβ/SAM/ICE display in (**a**) Nyquist plots and (**b**) charge-transfer resistance (R<sub>ct</sub>).

The equivalent circuit model of Randle insert into Figure 3a was used to fit the Nyquist plot data, showing three EIS parameters, including the C (capacitance for the electrode-solution interface element),  $R_{ct}$  (interfacial charge-transfer resistance that corresponds to the semicircular diameter in the Nyquist plot), and  $R_s$  (solution resistance).

The obtained fitting value of  $R_{ct}$  from Figure 3a was displayed in Figure 3b and Table 1. After SAM deposition that can be attributed to the  $-COO^-$  group and the carbon chain SAM which prevents the transfer of the negative charged  $[Fe(CN)_6]^{3-/4-}$  redox couple to the electrode surface, the value of  $R_{ct}$  has increased dramatically. Then,  $R_{ct}$  decreased after the EDC–NHS layer was deposited on the SAM-modified electrode, suggesting that the  $-COO^-$  groups of SAM were activated with the EDC–NHS carbodiimide coupler, the formation of succinimide ester on the modified electrode surface of SAM allows fast electron transfer between the electrode and the interface [19]. The two continuous immobilization steps of anti-a $\beta$  and BSA showed an increase in  $R_{ct}$  due to the binding of a specific material on the sensor surface which enhanced its surface resistance [20], demonstrating that the surface of the biosensor was covered by specific materials.

**Table 1.** EIS parameters of the bare and modified ICE extrapolated by fitting the measured spectra to the equivalent circuit model shown in Figure 3a.

Electrode	<b>R</b> <sub>s</sub> (Ω)	C (F)	$R_{ct}(\Omega)$
ICE	529.3	$55.37 imes10^{-9}$	47,294
SAM	522.8	$77.78  imes 10^{-9}$	291,956
EDC-NHS	524.9	$69.69 \times 10^{-9}$	77,691
anti-aβ	524	$75.88  imes 10^{-9}$	191,820
BSA	504.5	$76.49 \times 10^{-9}$	255,926

Fourier-transform infrared (FT-IR) spectra was also used to confirm the fabrication of the anti- $a\beta$ /SAM/ICE biosensor, and the results were displayed in Figure 4. The FT-IR result of the SAM-modified ICE showed the typical peaks near 1601 cm<sup>-1</sup> of carbonyl from the carboxylic acid moiety on the alkanethiols of SAM [21], indicating the immobilization of SAM on the ICE. FT-IR result of the immobilized anti- $a\beta$  on the EDC-NHS coupler activated SAM-modified ICE (blue curve in Figure 4) showed the typical peaks of protein at near 3374 cm<sup>-1</sup> of N-H stretching of amide II [22], 2925 and 2849 cm<sup>-1</sup> of C-H stretching of -CH<sub>2</sub> of protein [23,24], 920 cm<sup>-1</sup> of C-O stretching [25], and 602 cm<sup>-1</sup> of N-H deformation [22], which demonstrated the successful development of the anti- $a\beta$ /SAM/ICE biosensor.

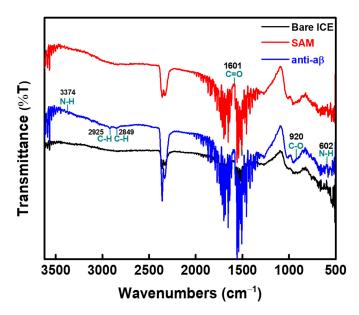
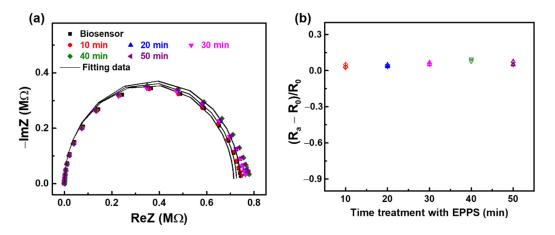


Figure 4. FT-IR results for the fabrication of the anti- $a\beta$ /SAM/ICE biosensor.

# 3.2. Monitoring the Effect of EPPS on a BDisaggregation

EPPS was used to target the disaggregation of  $a\beta_{agg}$  into its monomeric form. To confirm whether EPPS had an effect on the disaggregation of  $a\beta_{agg}$ , EIS was used to explore the change in  $R_{ct}$  of the specific anti- $a\beta/SAM/ICE$  biosensor for recognizing  $a\beta$  during incubation with EPPS at different time intervals.

Firstly, the impact of EPPS on the anti-a $\beta$  antibody was studied. EIS measurements of the anti-a $\beta$ /SAM/ICE biosensor after incubation with 20 mM·L<sup>-1</sup> EPPS at different time intervals are displayed in Figure 5. The Nyquist plots in Figure 5a did not change significantly after treatment with EPPS. Using Randle's equivalent circuit to fit the EIS data of Nyquist plots in Figure 5a, the R<sub>ct</sub> value was generated that serve to normalize the response of the anti-a $\beta$ /SAM/ICE biosensor to the EPPS treatment at 10, 20, 30, 40, 50 min in Figure 5b, where R<sub>0</sub> represents the R<sub>ct</sub> value of the biosensor with the immobilization of anti-a $\beta$  antibody, and R<sub>a</sub> represents the R<sub>ct</sub> value of the anti-a $\beta$ /SAM/ICE biosensor within the treatment of EPPS. The normalized value (R<sub>a</sub> – R<sub>0</sub>)/R<sub>0</sub> of the anti-a $\beta$ /SAM/ICE biosensor in Figure 5b did not change significantly after treatment with EPPS at various time periods, indicating that EPPS did not affect the anti-a $\beta$  antibody. The fitting EIS data were shown in Table 2.



**Figure 5.** EIS results of the biosensor anti- $a\beta/SAM/ICE$  with the treatment in EPPS at different time intervals display in (a) Nyquist plots and (b) Normalization of R<sub>ct</sub>. (*n* = 3; three data points shown).

486.3

50 min

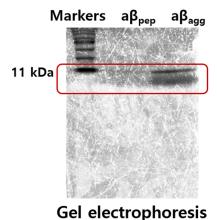
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Electrode	R <sub>s</sub> (Ω)	C (F)	R <sub>ct</sub> (Ω)	
Biosensor	490.9	$55.55 \times 10^{-9}$	715,084	
10 min	491.9	$56.72 \times 10^{-9}$	724,551	
20 min	487.5	$57.66 \times 10^{-9}$	726,538	
30 min	488.7	$57.93 \times 10^{-9}$	724,904	
40 min	488.5	$58.36 \times 10^{-9}$	743 262	

 $58.83 \times 10^{-9}$ 

730,896

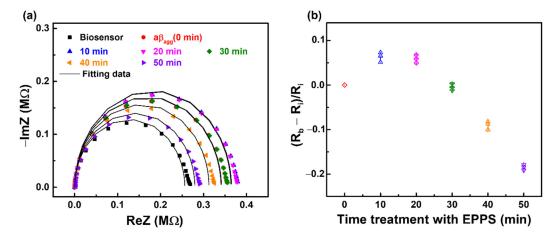
**Table 2.** EIS parameters of the anti- $a\beta$ /SAM/ICE biosensor after treatment with EPPS at different time intervals, extrapolated by fitting the measured spectra in Figure 5a to the equivalent circuit model of Randle.

Gel electrophoresis method [26] was used to confirm the aggregation state of  $a\beta$  in this manuscript. As seen in Figure 6, the gel electrophoresis result of  $a\beta_{agg}$  showed the molecular-weight ranging of 11 kDa for oligomer form, which corresponds to a smear band containing monomers and low-molecular-weight oligomers ranging between 4 and 18 kDa in size of  $a\beta$  oligomers [27], demonstrating the oligomeric form of  $a\beta_{agg}$  in this research.

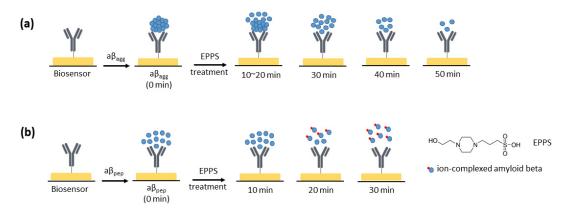


**Figure 6.** Gel electrophoresis results of  $a\beta_{pep}$  and  $a\beta_{agg}$ .

The process of disaggregation of  $a\beta_{agg}$  by EPPS at different time intervals is described in Section 2.2. Initially, the anti- $a\beta$ /SAM/ICE biosensor was incubated with  $a\beta_{agg}$  $(2.5 \times 10^{-3} \text{ mg} \cdot \text{mL}^{-1})$  for 20 min in the absence of EPPS and measured via EIS to examine the  $a\beta$  recognition of the biosensor. EIS measurements showed an increase in the diameter and height of the semicircle in the Nyquist plot, corresponding to the enhanced R<sub>ct</sub> after the anti- $a\beta$ /SAM/ICE biosensor was incubated with  $a\beta_{agg}$ , which is marked by 0 min in Figure 7a. This enhancement of R<sub>ct</sub> is ascribed to the hindrance of electron transfer between the  $[Fe(CN)_6]^{3-/4-}$  redox couple and the electrode surface due to successful specific binding between anti-a  $\beta$  antibody and  $a\beta_{agg},$  called  $a\beta_{agg}$  complex on the surface of the sensor. The biosensor with the  $a\beta_{agg}$  complex was subsequently incubated for different time periods (10, 20, 30, 40, and 50 min) with 20 mM·L<sup>-1</sup> of EPPS and was used for EIS measurements. As shown in Figure 7a, after 10 min of incubation in EPPS, the R<sub>ct</sub> of the biosensor with the  $a\beta_{agg}$  complex in the Nyquist plot showed an increase in the height and diameter of the semicircle, and the Nyquist plot remained the same for the next 20 min of incubation with EPPS, which indicated that the  $[Fe(CN)_6]^{3-/4-}$  ion transport towards the electrode surface was blocked due to excessive  $a\beta$  on the electrode surface. These results implied that the disaggregation of  $a\beta_{agg}$  started after 10 min of incubation with EPPS and was continuously maintained after 20 min of incubation. This disaggregation process converted the  $a\beta_{agg}$  into small  $a\beta$  monomers, and still retained a certain amount of  $a\beta_{agg}$ , creating an abundant compound of  $a\beta$  monomers and  $a\beta_{agg}$  on the biosensor surface, as delineated in Figure 8a, which hindered the transportation of  $[Fe(CN)_6]^{3-/4-}$  redox couple to the electrode surface, leading to an increase in  $R_{ct}$ . Figure 7a showed the decline in Rct after treatment with EPPS at 30 min, which manifested the disaggregation of the remaining  $a\beta_{agg}$  into several monomeric  $a\beta$  molecules on the sensor surface, suggesting that the movement of ion  $[Fe(CN)_6]^{3-/4-}$  toward the electrode surface was facilitated; the impeded ion  $[Fe(CN)_6]^{3-/4-}$  movement was caused by the presence of a dense  $a\beta_{agg}$ that was reduced by converting the  $a\beta_{agg}$  molecules into  $a\beta$  monomers, leading to the expansion of spacing between the specific aß molecules on the electrode surface, resulting in facilitating transport of redox couple  $[Fe(CN)_6]^{3-/4-}$ . This decline of R<sub>ct</sub> continued to occur at 40 and 50 min during EPPS treatment, which indicated the continuous disaggregation of  $a\beta_{agg}$ . A graphical description of the disaggregation process of  $a\beta_{agg}$  by EPPS at various time intervals is shown in Figure 8a. The response of  $a\beta_{agg}$  to EPPS treatment at time periods was evaluated by normalizing the R<sub>ct</sub> value of Figure 7a and plotted against time, as  $(R_b - R_i)/R_i$  in Figure 7b, where  $R_i$  represents the  $R_{ct}$  of the biosensor with the immobilization of  $a\beta_{agg}$ , and  $R_b$  represents the  $R_{ct}$  of the biosensor with the immobilization of  $a\beta_{agg}$  within the EPPS treatment. The normalized value  $(R_b - R_i)/R_i$  in Figure 7b increased after 10 min of EPPS treatment, remained stable after 20 min, and gradually decreased over the 30, 40, and 50 min points, which indicated the disaggregation process of  $a\beta_{agg}$  at different time intervals starting at 10 min and confirmed the effect of EPPS on  $a\beta_{agg}$  disaggregation, corresponding to the results and explanation in Figure 7a. A detailed graphical description of the disaggregation process of  $a\beta_{agg}$  by EPPS at various time intervals is provided in Figure 8a. The fitting EIS data were shown in Table 3.



**Figure 7.** EIS results monitoring the disaggregation  $a\beta_{agg}$  with the treatment in EPPS at different time intervals display in (a) Nyquist plots and (b) Normalization of R<sub>ct</sub>. (*n* = 3; three data points shown).



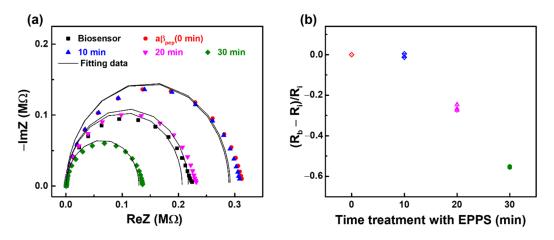
**Figure 8.** (a) Disaggregation mechanism of  $a\beta_{agg}$  by EPPS, and (b) the impact of EPPS on  $a\beta_{pep}$ .

**Table 3.** EIS parameters of the anti- $a\beta/SAM/ICE$  biosensor after incubating with  $a\beta_{agg}$ , and then conduct to treatment with EPPS at different time intervals, extrapolated by fitting the measured spectra in Figure 7a to the equivalent circuit model of Randle.

Electrode	R <sub>s</sub> (Ω)	C (F)	$R_{ct}(\Omega)$
Biosensor	504.8	$76.10  imes 10^{-9}$	228,331
$a\beta_{agg}$ (0 min)	519.2	$76.29  imes 10^{-9}$	331,631
10 min	517.4	$75.98  imes 10^{-9}$	366,841
20 min	517.1	$75.89  imes 10^{-9}$	363,145
30 min	519.1	$75.08  imes 10^{-9}$	334,918
40 min	519.5	$74.81  imes 10^{-9}$	307,364
50 min	517.6	$74.71 imes10^{-9}$	274,678

The effect of EPPS on the a $\beta$  1–42 peptide (a $\beta_{pep}$ ), a long peptide of 1–42 amino acids, was also studied for comparison with the obtained results of a $\beta_{agg}$ , which are mentioned above. The experiment of a $\beta_{pep}$  treatment in EPPS was carried out through similar processes as described in Section 2 with a $\beta_{agg}$  replaced by a $\beta_{pep}$ . Nyquist plot of EIS result in Figure 9a show an increase in R<sub>ct</sub> of the biosensor anti-a $\beta$ /SAM/ICE after 20 min of incubation with a $\beta_{pep}$ . This increment of R<sub>ct</sub> is expected due to the repulsion

between the negative charge of immobilized  $a\beta_{pep}$  antigen on the anti- $a\beta/SAM/ICE$ biosensor with redox couple, that could prevent the redox couple from reaching the SAMmodified electrode, confirming the successful binding between anti- $a\beta$  and  $a\beta_{pep}$  [28]. The  $R_{ct}$  of the biosensor with the immobilized  $a\beta_{pep}$  remained the same after 10 min of treatment in EPPS, indicating that  $a\beta_{pep}$  was not affected by EPPS for the initial 10 min treatment. R<sub>ct</sub> significantly decreased after 20 and 30 min of subsequent EPPS treatment, indicating the facilitating electron transfer at the electrode interface. This decrease of R<sub>ct</sub> may be attributed to the fragmentation of  $a\beta_{pep}$  by EPPS, leading to the formation of ion-complexed amyloid beta with  $[Fe(CN)_6]^{3-/4-}$  [29,30], which results in the decreasing of Rct and enhancing of electrochemical property of the sensor at 20 min treatment, and at more treatment time of 30 min [16]. The normalized result  $(R_{\rm b} - R_{\rm i})/R_{\rm i}$  is displayed in Figure 9b, where  $R_i$  represents the  $R_{ct}$  of the biosensor with the immobilization of  $a\beta_{pep}$ , and  $R_b$  represents the  $R_{ct}$  of the biosensor with the immobilization of  $a\beta_{pep}$ within the EPPS treatment, which corresponds with the data of Nyquist plots in Figure 9a. The  $(R_b - R_i)/R_i$  was stable after 10 min, and rapidly decreased over the 20 and 30 min points following EPPS treatment. Figure 8b shows the effect of EPPS treatment on  $a\beta_{pep}$ , and EIS results are shown in Figure 9. The fitting EIS data were shown in Table 4.



**Figure 9.** EIS results monitoring the impact of EPPS on the  $a\beta_{pep}$  at different time intervals display in (**a**) Nyquist plots and (**b**) Normalization of R<sub>ct</sub>. (*n* = 3; three data points shown).

**Table 4.** EIS parameters of the anti- $a\beta$ /SAM/ICE biosensor after incubating with  $a\beta_{pep}$ , and then conduct to treatment with EPPS at different time intervals, extrapolated by fitting the measured spectra in Figure 9a to the equivalent circuit model of Randle.

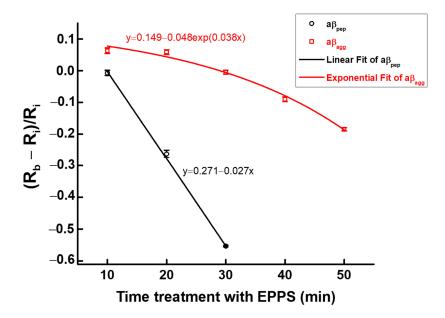
El a stras d a	<b>B</b> (O)	C (E)	P (O)
Electrode	$\mathbf{R}_{\mathbf{s}}(\mathbf{\Omega})$	C (F)	$R_{ct}(\Omega)$
Biosensor	633.5	$91.3  imes 10^{-9}$	205,978
aβ <sub>pep</sub> (0 min)	635	$92.96 \times 10^{-9}$	289,018
10 min	632.6	$92.17  imes 10^{-9}$	289,110
20 min	629.8	$90.69  imes 10^{-9}$	209,893
30 min	639.9	$88.06 \times 10^{-9}$	129,281

By examining the EIS results of  $a\beta_{agg}$  and  $a\beta_{pep}$  in the treatment with EPPS in this study, it is observed that EPPS affects the disaggregation of  $a\beta_{agg}$  gradually and leads to the establishment of a disaggregation mechanism, as shown in Figure 8, while EPPS causes rapid disruption of  $a\beta_{pep}$ .

## 3.3. Kinetic Disaggregation Rate k

The kinetic disaggregation rate *k* of protein is an important parameter in evaluating the functioning of neurodegenerative diseases [31]. However, there have not been report

on the *k* value for  $a\beta_{agg}$  at present. To estimate the disaggregation rate *k* value for  $a\beta_{agg}$ , the normalized data for the treatments of  $a\beta_{agg}$  and  $a\beta_{pep}$  in EPPS were plotted and fitted versus time, as shown in Figure 10. Data for the disaggregation experiment  $a\beta_{agg}$  represent the best fit of the data to a single-exponential function  $y = 0.149 - 0.048 \exp(0.038x)$ , whereas data for the treatment  $a\beta_{pep}$  in EPPS express a linear fit function y = 0.271 - 0.027x. According to the studies of kinetic disaggregation of proteins, the fitting data for protein disaggregation has been expressed by an exponential function  $F = y_0 + A \exp(-kt)$  [31,32], where *k* represents the disaggregation rate. Consequently, the disaggregation rate *k* of  $a\beta_{agg}$  was determined to be 0.038 by our method, providing a reference parameter for the disaggregation of  $a\beta_{agg}$  for future studies.



**Figure 10.** Fitting of two sets of disaggregation data for  $a\beta_{agg}$  and  $a\beta_{pep}$ , respectively; symbols and bars represent the average and standard deviation of the data (n = 3).

#### 4. Conclusions

A new concept of measuring  $a\beta_{agg}$  disaggregation following treatment with EPPS via EIS signaling was introduced in this report. An electrochemical impedimetric immunosensor was developed to monitor EIS signals at the disaggregation stages of  $a\beta_{agg}$  during treatment with EPPS at different time intervals. The obtained EIS results confirmed that the disaggregation of  $a\beta_{agg}$  by treatment in EPPS occurred after 10 min, and this disaggregation process continued gradually over the next 50 min. A single exponential function for  $a\beta_{agg}$  disaggregation was also found in this report, with a kinetic disaggregation rate (*k*) value of 0.038. The data indicated the successful application of EIS measurement in monitoring and establishing the mechanism for the impact of EPPS on the disaggregation of  $a\beta_{agg}$ , which may lead to new insights in the diagnosis and treatment of signs of AD at various phases.

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