## **RESEARCH PAPER**

OPEN ACCESS

Taylor & Francis

Taylor & Francis Group

# Seeking new approach for therapeutic treatment of cholera disease via inhibition of bacterial carbonic anhydrases: experimental and theoretical studies for sixteen benzenesulfonamide derivatives

Rosaria Gitto<sup>a</sup>\* (b), Laura De Luca<sup>a</sup>, Francesca Mancuso<sup>a</sup>, Sonia Del Prete<sup>b</sup>, Daniela Vullo<sup>c</sup>, Claudiu T. Supuran<sup>c</sup> (b) and Clemente Capasso<sup>b</sup>\* (b)

<sup>a</sup>Department of Chemical, Biological, Pharmaceutical and Environmental Sciences (CHIBIOFARAM), University of Messina, Messina, Italy; <sup>b</sup>Department of Biology, Agriculture and Food Sciences, Institute of Biosciences and Bioresources- CNR, Napoli, Italy; <sup>c</sup>NUROFARBA Department, University of Florence, Sesto Fiorentino, Italy

### ABSTRACT

A series of sixteen benzenesulfonamide derivatives has been synthesised and tested as inhibitors of *Vibrio* cholerae carbonic anhydrase (CA) enzymes, belonging to  $\alpha$ -CA,  $\beta$ -CA, and  $\gamma$ -CA classes (VchCA $\alpha$ , VchCA $\beta$ , and VchCA $\gamma$ ). The determined  $K_i$  values were compared to those of selected human CA isoforms (hCA I and hCA II). Structure-affinity relationship analysis highlighted that all tested compounds proved to be active inhibitors of VchCA $\alpha$  at nanomolar concentration. The VchCA $\beta$  activity was lower to respect inhibitory efficacy toward VchCA $\alpha$ , whereas, these benzenesulfonamide derivatives failed to inhibit VchCA $\gamma$ . Interestingly, compound **7e** combined the best activity toward VchCA $\alpha$  and VchCA $\beta$ . In order to obtain a model for binding mode of our inhibitors toward bacterial CAs, we carried out docking simulations by using the available crystal structures of VchCA $\beta$ .

### **ARTICLE HISTORY**

Received 4 April 2019 Revised 6 May 2019 Accepted 8 May 2019

### **KEYWORDS**

Carbonic anhydrase inhibitors (CAIs); benzenesulfonamides; *Vibrio cholera*; molecular docking

#### **GRAPHICAL ABSTRACT**



#### 1. Introduction

The gram-negative bacterium *Vibrio cholerae* is responsible for the secretory diarrheal disease cholera that infects millions of people worldwide and might cause high mortality if left untreated. While the rehydration therapy is an effective treatment approach as well as the use of antibiotics in severe infections, there is a pressing need for agents that prophylactically could prevent the injurious effects of a cholera infection.

There is evidence that protein ToxT is one of the transcription factors activating the *V. cholerae* virulence gene expression; therefore, ToxT is a target for the development of cholera therapeutics<sup>1,2</sup>. ToxT activity is (a) negatively controlled by the unsaturated fatty acids (UFAs) of the bile and (b) enhanced by the presence of bicarbonate in the upper small intestine where

*V. cholerae* preferentially colonises. Therefore, bicarbonate is likely an important chemical signal during the cholera disease<sup>3,4</sup>. Generally, bacteria increase cytosolic bicarbonate levels through bicarbonate transporter proteins and carbonic anhydrases (CAs, EC 4.2.1.)<sup>2</sup>. It has been demonstrated that *V. cholerae* can increase the cytosolic bicarbonate levels through the hydration of CO<sub>2</sub> by the action of CAs since *V. cholerae* lacks bicarbonate transporter proteins in its genome<sup>3</sup>. Thus, the CAs represent an attractive molecular target for the development of innovative anti-infective agents<sup>3</sup>. The metalloenzyme CAs are grouped into seven genetically distinct families, named  $\alpha$ -,  $\beta$ -,  $\gamma$ -,  $\delta$ -,  $\zeta$ -,  $\eta$ -, and  $\theta$ -CAs, with different structures and active site architecture as well as metal ions in the catalytic region (zinc, iron, cobalt, and cadmium)<sup>5</sup>. *Vibrio cholerae* encodes carbonic anhydrase

CONTACT Rosaria Gitto rosaria.gitto@unime.it Department of Chemical, Biological, Pharmaceutical and Environmental Sciences (CHIBIOFARAM), University of Messina, Viale Palatucci 13, I-98168 Messina, Italy; Clemente Capasso clemente.capasso@ibbr.cnr.it Department of Biology, Agriculture and Food Sciences, Institute of Biosciences and Bioresources-CNR, Via P. Castellino 111, I-80131, Napoli, Italy \*These authors equally contributed to the work.

© 2019 The Author(s). Published by Informa UK Limited, trading as Taylor & Francis Group.

This is an Open Access article distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0/), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

B Supplemental data for this article can be accessed here.

belonging to  $\alpha$ -CA,  $\beta$ -CA, and  $\gamma$ -CA classes (VchCA $\alpha$ , VchCA $\beta$ , and VchCA $\gamma$ )<sup>6</sup>, that could represent potential targets for the development of anti-infectives targeting V. cholerae colonisation<sup>7–9</sup>. The  $\alpha$ - and  $\beta$ -CAs are Zn(II) metalloenzymes, whereas  $\gamma$ -CAs use Fe(II) as metal ion; however they are also active with Zn(II) or Co(II). The metal ion is coordinated by three His residues in the  $\alpha$ - and  $\gamma$ -classes, and one His and two Cys residues in the  $\beta$ -class. Several classes of CA inhibitors (CAIs)<sup>10,11</sup> have been identified; among them, sulfonamide derivatives have been shown to anchor the metal-coordinated water molecule/hydroxide ion within catalytic site thus inducing CA inhibition. The prototype sulfonamide inhibitor acetazolamide (Figure 1, AAZ, 1) is clinically used as diuretic and anticonvulsant agent in humans; additionally, AAZ displays inhibitory effects toward bacterial isozymes. Moreover, as confirmation of the role of bicarbonate in virulence and colonisation processes, the carbonic anhydrase inhibitor ethoxzolamide (EZA, 2) proved to decrease the virulence gene expression and to reduce the growth of pathogen<sup>7</sup>.

The principal drawback using CAIs as antiinfectives agents is the lack of selectivity towards the pathogenic versus human isoforms<sup>7</sup>. For this reason, many research groups are continually involved in the synthesis of new CAIs or in the modification/ optimisation of the existing inhibitors, which are commonly tested on CAs from mammalian and pathogenic organisms<sup>10,11</sup>. Most of these inhibitors represent interesting leads towards the optimisation of new antibiotic agents showing excellent inhibitory efficiency and selectivity for the target CAs over the human (h) offtarget isoform hCA 1<sup>10,11</sup>. In the course of our efforts to identify novel selective CAIs, we have synthesised and tested a series of quinolone/isoquinoline-arylsulfonamides showing high affinity toward human CAs<sup>12-22</sup>. Specifically, we have reported the discovery of a set of heteroaryl-N-carbonylbenzenesulfonamides as a class of potent inhibitors of druggable CA isoforms (hCA II, hCA VII, hCA IX, and hCA XIV)<sup>15,17,18,20-22</sup>.

Recently, we have studied a new small library of CAIs containing azepine/piperidine/piperazine nucleus linked to benzenesulfonamide fragment and disclosed several compounds that demonstrated inhibitory effects in low nanomolar range toward hCAs<sup>23</sup>. This series of compounds has been rationally designed by using the cycloalkylamine nucleus as the core for binding contact in the middle area of catalytic site (Figure 1). The visual inspection of their binding mode in co-crystal adducts with hCA II and hCA VII confirmed that the sulfonamide portion is crucial for CA recognition process<sup>23</sup>. By considering the high inhibitory effects of this series of potent sulfonamide compounds, we



**Figure 1.** Chemical structures of well-known CAIs acetazolamide (AAZ, 1) and ethoxzolamide (EZA, 2) and designed 4-(cycloalkyl)-1-carbonylbenzenesulfonamides.

decided to further study their CA inhibitory profile thus exploiting their capability to act as potential anti-infective agents. Therefore, the primary purpose of our work is to provide new information about structure-affinity relationship relative to the inhibition of the three classes from *V. cholerae*, VchCA $\alpha$ , VchCA $\beta$ , and VchCA $\gamma$ . Particularly, our idea was to study in depth the role of hydrophobic interactions within VchCA cavity.

### 2. Materials and methods

### 2.1. Chemistry

All reagents were used without further purification and bought from common commercial suppliers. Microwave-assisted reactions were carried out in a Focussed Microwave TM Synthesis System, Model Discover (CEM Technology Ltd Buckingham, UK). Melting points were determined on a Buchi B-545 apparatus (BUCHI Labortechnik AG Flawil, Switzerland) and are uncorrected. By combustion analysis (C, H, N) carried out on a Carlo Erba Model 1106-Elemental Analyser we determined the purity of synthesised compounds; the results confirmed a > 95% purity. Merck Silica Gel 60 F254 plates were used for analytical TLC (Merck KGaA, Darmstadt, Germany). For detection, iodine vapour and UV light (254 nm) were used. Flash Chromatography (FC) was carried out on a Biotage SP1 EXP (Biotage AB Uppsala, Sweden). <sup>1</sup>H NMR spectra were measured in dimethylsulfoxided6 (DMSO-d<sub>6</sub>) and CDCl<sub>3</sub> with a Varian Gemini 300 spectrometer (Varian Inc. Palo Alto, California, USA); chemical shifts are expressed in  $\delta$  (ppm) and coupling constants (J) in hertz. All exchangeable protons were confirmed by addition of D<sub>2</sub>O. R<sub>f</sub> values were determined on TLC plates using a mixture of DCM/ MeOH (96/4) as eluent. All compounds have been re-synthesised following the previously reported procedure, that is briefly described below<sup>23</sup>.

# 2.1.1. Synthetic procedures for benzenesulfonamides 5a-e, 6a, 7a-f, 8a-d

**Pathway i:** To a solution of 4-(aminosulfonyl)benzoic acid (4) (6 mmol) in THF (15 ml) the carbonylimidazole (6 mmol) (CDI) at 0 °C was added. The obtained mixture was stirred at room temperature for 3 h and then the appropriate amine derivative (15 mmol) in DMF (5 ml) was added dropwise. The reaction mixture was stirred at room temperature for 2 h. The solvent was removed *in vacuo*; by adding of aqueous solution of NaHCO<sub>3</sub> (5 ml) we obtained compounds **5a** and **5d** as crude products, which were purified through crystallization by a mixture of Et<sub>2</sub>O and EtOH (1:1)

**Pathway ii:** A mixture of 4-(aminosulfonyl)benzoic acid (4) (2 mmol) and N,N,N,N-tetramethyl-O-(1H-benzotriazol-1-yl)uranium hexafluorophosphate (HBTU) (2 mmol) in DMF(2 ml) was stirred at room temperature for 1 h. Then, a solution of the appropriate amine derivative (2 mmol) in TEA (2 mmol) was added dropwise. The reaction mixture was left overnight, then quenched with water (10 ml) and extracted with EtOAc ( $3 \times 10$  ml). The organic phase was dried with Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed *in vacuo*. The residue was purified by flash chromatography (DCM/ MeOH 96:4), crystallized by treatment with a mixture of Et<sub>2</sub>O and EtOH (1:1) to give the desired final compounds **5b**, **5c**, **6a**, **7a**-**f**, and **8a**-**d** as white powders; the chemical characterisation of all re-synthesised compounds was in good agreement with literature (see Supporting Material).

## 2.2. Preparation of the bacterial CAs

The three bacterial enzymes were prepared accordingly to the procedure reported by our groups<sup>24</sup>. Briefly, the GeneArt Company (Invitrogen), specialised in gene synthesis, designed the genes encoding for the bacterial  $\alpha$ ,  $\beta$ , and  $\gamma$ - CAs. The BL21 DE3 competent cells (Agilent) were transformed with the expression vector pET15-b containing the gene encoding for one of the three CA-classes. Subsequently, bacterial cells were induced with 1 mM IPTG and, after 30 min, treated with 0.1 M ZnCl<sub>2</sub>. After 4h, cells were harvested and disrupted by sonication at 4 °C. After centrifugation at 12,000×g for 45 min, the supernatant was incubated with His Select HF nickel affinity gel resin (Sigma) equilibrated in lysis buffer for 30 min. The protein was eluted with the wash buffer containing 200 mM imidazole. Collected fractions were dialysed against 50 mM Tris/HCl, pH 8. At this stage of purification, the protein was at least 95% pure.

### 2.3. Carbonic anhydrase inhibition assay

An Applied Photophysics stopped-flow instrument (Leatherhead, Surrey (UK)) has been used for assaying the CA catalysed CO<sub>2</sub> hydration activity<sup>25</sup>. Phenol red (at a concentration of 0.2 mM) has been used as an indicator, working at the absorbance maximum of 557 nm, with 20 mM TRIS (pH 8.3) as buffer, and 20 mM NaClO<sub>4</sub> (for maintaining constant the ionic strength), following the initial rates of the CA-catalysed CO<sub>2</sub> hydration reaction for a period of 10-100 s. The CO<sub>2</sub> concentrations ranged from 1.7 to 17 mM for the determination of the kinetic parameters (by Lineweaver-Burk plots) and inhibition constants. For each inhibitor, at least six traces of the initial 5-10% of the reaction have been used for determining the initial velocity. The un-catalysed rates were determined in the same manner and subtracted from the total observed rates. Stock solutions of inhibitor (10-100 mM) were prepared in distilled-deionized water, and dilutions up to 0.01 mM were done after that with the assay buffer. Inhibitor and enzyme solutions were preincubated together for 15 min at room temperature before assay, to allow for the formation of the E-I complex or the eventual active site-mediated hydrolysis of the inhibitor. The inhibition constants were obtained by non-linear least-squares methods using PRISM 3 and the Cheng-Prusoff equation, as reported earlier<sup>26-28</sup>, and represent the mean from at least three different determinations. The human CA isoforms were recombinant ones obtained in-house. All salts/small molecules were of the highest purity available, from Sigma-Aldrich (Milan, Italy).

# 2.4. Modelled structure of the opening of the active site of $\beta$ -carbonic anhydrase from V. cholerae

The structure used to perform docking studies was built on different steps. First of all, the crystal structure of closed  $\beta$ -CA from *V. cholerae* was retrieved from the RCSB Protein Data Bank (PDB code 5CXK)<sup>29</sup>. The water molecules were discarded, then hydrogen atoms were added to the protein by using Discovery Studio 2.5.5 software<sup>30</sup>. In the second step, a superimposition of residues Cys42, Asp44, Arg46, His98, Cys101, Zn301 of  $\beta$ -CA from *V. cholerae* on corresponding residues Cys47, Asp49, Arg51, His103, Cys106, Zn228 of chain A of X-ray complex of AAZ with  $\beta$ -CA from the unicellular green alga *Coccomyxa* (*Co*-CA, PDB code 3UCJ)<sup>31</sup> was performed using PyMOL software (https://pymol.org).

The above-mentioned superimposition was useful to modify the orientation of the side-chains of the two amino acids Asp44 and Arg46 of  $\beta$ -CA from *V. cholerae* using as template the conformation of Asp49 and Arg51 of  $\beta$ -CA from *Coccomyxa*. The modified structure of  $\beta$ -CA from *V. cholerae* was minimised using the conjugate gradient algorithm of NAMD. Minimisation was performed for 500 steps keeping all atoms fixed except for the backbones of residue 43–47 and the side-chains of Asp44 and Arg46: in detail, to maintain the correct orientation of carboxylate moiety of Asp44 and guanidinium group of Arg46 also these groups were kept fixed.

The atom types were assigned using force field CHARMM v22 and the atomic charges according to Gasteiger–Marsili method by Vega<sup>32</sup>. Furthermore, in the obtained structure, assuming a similar binding mode of the AAZ in both of the  $\beta$ -CAs, we inserted in the catalytic site of  $\beta$ -CA from *V. cholerae* the X-ray conformation of the acetazolamide retrieved from  $\beta$ -CA from *Coccomyxa*. The side-chains of the residues around the inhibitor AAZ have been minimised by LigandScout<sup>33</sup>.

### 2.5. Docking studies

The complex obtained was used to perform docking studies by GOLD software<sup>34</sup> and the AAZ was discarded. The ligand structures were constructed by Vega 3.1.1 and the energy was minimised by using the Conjugate Gradient method (1000 steps). The minimised ligands were docked in their corresponding proteins using Gold Suite 5.0.1. The region of interest, which was used by the Gold programme, was defined as containing the residues within 10 Å of the original position of the acetazolamide in the model structure.

A scaffold constraint (penalty = 5.0) was used to restrict the solutions in which the sulfonamide moiety was able to coordinate the metal within the catalytic binding site. ChemPLP was chosen as the fitness function. The standard default settings were used in all the calculations and the ligands were submitted to 100 genetic algorithm runs. The "allow early termination" command was deactivated. Results differing by less than 0.75 Å in the ligand – all atom RMSD were clustered together. The conformations with the highest fitness values were chosen for both analysis and representation.

All obtained complexes were minimised keeping the Zn ion fixed and using the conjugated gradients algorithm by NAMD<sup>35</sup>. The results were displayed using the PyMOL software (https:// pymol.org).

### 3. Results and discussion

### 3.1. Chemistry

As displayed in Scheme 1, the studied 4-(cycloalkyl)-1-carbonylbenzenesulfonamides **5a-e**, **6a**, **7a-f**, **8a-d** have been resynthesised following a synthetic procedure previously reported by us<sup>23</sup>; experimental details can be found in "Materials and Methods" section as well as in Supporting Material.

### 3.2. Carbonic anhydrase inhibition

The inhibitory effects of this series of sixteen 4-(cycloalkyl-1-carbonyl)benzenesulfonamide derivatives **5a–e**, **6a**, **7a–f**, **8a–c**, and **8d** were evaluated against VchCA $\alpha$ , VchCA $\beta$ , and VchCA $\gamma$  by using the stopped-flow carbon dioxide hydrase assay. The obtained results are summarised in Table 1 and compared with the wellknown inhibitor AAZ (**1**). Moreover, the  $K_i$  values measured



Scheme 1. Reagents and conditions: (i) carbonyldiimidazole (DCI), THF, r.t., 3 h, then RR'NH, DMF r.t., 2 h; (ii) RR'NH, HBTU, DMF, TEA, r.t., overnight.

Table 1. Inhibition of hCA I, hCA II, VchCA $\alpha$ , VchCA $\beta$ , and VchCA $\gamma$  for compounds 5a–e, 6a, 7a–f, 8a–c and 8d and acetazolamide (1, AAZ) by a stopped flow CO<sub>2</sub> hydrase assay.

n <sub>i</sub> (mvi)					
Cmp	hCA I <sup>b</sup>	hCA II <sup>c</sup>	VchCAα	VchCAβ	VchCAγ
5a	9.2	5.7	843.2	>10,000	>10,000
5b	4.4	0.63	296.4	>10,000	>10,000
5c	187	6.1	528.8	>10,000	>10,000
5d	7.7	0.79	588.9	>10,000	>10,000
5e	75.5	1.5	241.9	3754	>10,000
ба	85.0	3.8	275.4	>10,000	>10,000
7a	1.7	0.5	43.3	3308	>10,000
7b	9.3	0.6	193.8	3714	>10,000
7c	6.6	0.6	45.1	3945	>10,000
7d	6.3	0.7	58.8	2776	>10,000
7e	6.5	0.6	89.9	806.4	>10,000
7f	7.7	0.6	72.1	6612	>10,000
8a	6.8	3.0	470.4	>10,000	>10,000
8b	0.69	0.5	95.7	5008	>10,000
8c	50.1	5.7	275.5	4816	>10,000
8d	91.1	2.0	47.0	4156	>10,000
AAZ (1)	250	12.1	6.8	451	473

<sup>a</sup>Mean from three different assays (errors were in the range of  $\pm 5-10\%$  of the reported values).

b,cData reported in reference<sup>23</sup>.

 $K_{\rm c}$  (nM)<sup>a</sup>

against bacterial CAs were compared with an affinity towards selected human CA isoforms hCA I and hCA  $\rm II^{23}.$ 

As reported in Table 1, all studied compounds proved to be inhibitors of the VchCA $\alpha$  and displayed  $K_{iS}$  falling in a wide range of 43.3 and 843.2 nM. In terms of structure-affinity relationship considerations, the data collected in Table 1 suggested that the unsubstituted 4-(piperidine-1-carbonyl)benzenesulfonamide (**5a**), the three methyl-substituted analogues **5b**, **5c**, **5d** as well as cyclohomologue compound **6a** were less active than corresponding 4-aryl-substituted compounds **7a**, **7b**, **7c**, **7d**, **7e**, and **7f**. Compound **7a** ( $K_i$  value of 43.3 nM) demonstrated the best activity having the phenyl ring linked to 4-position of piperidine nucleus. The introduction of an additional hydroxyl group, nitrile or acetyl functionalities at C-4 position of piperidine nucleus induces a weak decrease of affinity. The presence of 4'-chlorophenyl ring and hydroxyl group at 4-position of piperidine nucleus of compound **7c** afforded equi-active inhibitor of unsubstituted 4-phenyl analog 7a. Among the series of benzenesulfonamides 8a-d containing the piperazine core, the 4-(4-phenylpiperazine-1-carbonyl)benzenesulfonamide (8d) displayed the best inhibitory effect ( $K_i$ value of 47.0 nM), thus confirming that the presence of a bulky aromatic group anchored to the 4-position of cycloalkylamine nucleus generally improves the affinity against VchCAa. It is interesting to note the impact of the introduction of a methylene spacer between piperazine core and phenyl moiety: it was found a ten-fold reduction of activity of the compound 4-(4-benzylpiperazine-1-carbonyl)benzenesulfonamide (8a) ( $K_i = 470 \text{ nM}$ ) when compared to parent compound 8d. This evidence might be due to a different binding within the catalytic cavity. Moreover, the introduction of a fluorine atom in the para position of the phenyl ring of benzylpiperazine-sulfonamide 8b (K<sub>i</sub> value of 95.7 nM) improves the affinity toward VchCA $\alpha$ . On the contrary, the presence of a benzhydryl moiety (compound 8c,  $K_i = 275.5$  nM) seems well tolerated to respect the benzyl-fragment (compound 8a).

Concerning the activity on VchCA $\beta$ , several compounds proved to be inhibitors at low micromolar concentration showing inhibition data ranging from 806.4 nM to 6612 nM. Overall, the presence of a 4-aryl moiety resulted advantageous for inhibitory effects toward VchCA $\beta$ . The best affinity was found for 4-(4-cyano-4-phenyl-piperidine-1-carbonyl)benzenesulfonamide compound **7e** ( $K_i = 806.4$  nM). The presence of other small substituents, such as acetyl for compound **7f** or hydroxyl group for compounds **7 b**, **7c**, and **7d** did not significantly influence the inhibitory effects toward VchCA $\beta$ . Moreover, the replacement of piperidine nucleus with piperazine one resulted in divergent effects: (a) it was found a completely loss of inhibitory activity for unsubstituted compound **8a**, whereas compounds **8b**, **8c**, and **8d** shared similar potency as moderate inhibitors of VchCA $\beta$  class.

On the contrary, all studied compounds were totally ineffective inhibitors against VchCA $\gamma$  enzyme. Despite these compounds demonstrated interesting selectivity against VchCA $\alpha$  to respect  $\beta$ and  $\gamma$ -classes, they proved to be active inhibitors of ubiquitously expressed hCAI and hCA II at low nanomolar concentration, thus impairing their employment as therapeutics in humans.

### 3.3. Modeling

In order to gain more information about the binding interaction of the most active compounds into the enzymatic catalytic site, we carried out a theoretical structural study for understanding and revealing the binding mode of inhibitors toward bacterial CA classes. The crystal structure of apoenzyme VchCA $\beta$  has been recently reported by Ferraroni et al. (PDB code 5CXK)<sup>29</sup>, whereas, the determined structures of VchCAa and VchCAy from V. cholerae are not currently available. The crystal structures of  $\beta$ -CA are reported for other species<sup>29,36-39</sup>; these structural data furnished information about the homology of their 3-D folds as well as the shape of catalytic site, that displays the zinc coordinated by two cysteines, one-histidine and one-aspartic amino acid residue (the so-called "closed active site"). Particularly, for VchCA $\beta$ s the catalytic zinc ion is coordinated by Cys42, Asp44, His98, and Cys101. (See schematic representation in Supporting Material). The catalytic site of VchCA $\beta$ s assumes the so-called "closed active site" conformation as an inactive state. The pH-tuned movement of the Asp 44 residue from metal ion converts the "closed active site" to the "open active site". Since the available 3D structure the  $\beta$ -CA from V. cholerae was solved only in the "closed active site" configuration<sup>29</sup>, the rational design of novel inhibitors through a classical structure-based approach is currently hampered. To overcome the lack of this crucial structural information, in the first step of our theoretical study, we performed a simulation for the opening of the active site of  $\beta$ -CA from V. cholerae. In more details, we superimposed the 3D X-ray structure of "closed active site" of  $\beta$ -CA from V. cholerae (PDB code 5CXK)<sup>29</sup> to "open active site" of available  $\beta$ -CA from *Coccomyxa* in complex with inhibitor AAZ (1) (PDB code 3UCJ)<sup>31</sup>. Based on the conformation of side chain of crucial residues Asp49 and Arg51 of  $\beta$ -CA from Coccomvxa, we successively modified the orientation of the corresponding side-chain of Asp44 and Arg46, that are considered responsible for proton shuttling and "closed state" of  $\beta$ -CA. By assuming that inhibitor 1 displays a similar binding mode in these two  $\beta$ -CAs, we extracted AAZ from  $\beta$ -CA of *Coccomyxa* and then docked this inhibitor in the built 3D structure of  $\beta$ -CA from V. cholerae; in the last step, we minimised all system. As a result, we obtained the hypothetical "open active site" of  $\beta$ -CA from V. cholerae as the "first in silico model", useful for mapping the inhibitor binding interactions; our idea was to translate these data for understanding the inhibitory effects of the above mentioned sixteen 4-(cycloalkyl-1-carbonyl)benzensulfonamide derivatives (5a-e, **6a**, **7a–f**, **8a–d**) toward VchCAβ.

Figure 2 (A) displays AAZ (1) bound directly to the "hypothetic open" active site of  $\beta$ -CA from *V. cholerae*. As expected, the sulfonamide moiety anchors the catalytic zinc ion; additionally, the nitrogen of the N-acetamido group forms a hydrogen bond to the oxygen atom of Gly102, the thiadiazole ring establishes  $\pi/\pi$  interaction with Tyr83 which is reinforced by further hydrogen bond contacts between heterocyclic nitrogen atoms and hydroxyphenyl substituent. This network of interactions might explain the affinity for VchCA $\beta$  at submicromolar concentration ( $K_i$  value of 451 nM, Table 1), thus supporting the reliability of our modelled VchCA $\beta$  catalytic cavity. Then we studied the docking poses into our



**Figure 2.** Docking poses into our "modelled" VchCA $\beta$  catalytic cavity for acetazolamide (A), sulfonamide derivatives **5a** (B) and **7e** (C). The interactions between the VchCA $\beta$  and inhibitors AAZ, **5a** and **7e** were examined using PyMOL and LIGPLUS<sup>40</sup> software; residues involved in hydrophobic interactions and hydrogen bonds (yellow dashed lines) and Zn ion coordination are shown.

"modelled" VchCA $\beta$  catalytic cavity for all 4-(cycloalkyl-1-carbonyl)benzensulfonamide derivatives (**5a–e**, **6a**, **7a–f**, **8a–d**) bearing the sulfonamide as a minimal structural requirement to anchor the zinc ion in the deep catalytic site of all CAs.

In Figure 2(B), we described the best pose of prototype unsubstituted compound **5a** showing the ability to create a

network of interactions with the deep area of the catalytic site near to zinc ion. As expected, the sulfonamide group is engaged in polar contacts with the CA cavity, whereas the remaining portion of molecule occupies the middle area establishing few hydrophobic interactions with residues Tyr83, Leu113, and Ala106. Therefore, we hypothesised that these interactions could be not sufficient to inhibit VchCA $\beta$  ( $K_i > 10,000 \text{ nM}$ ). This consideration might be applicable to justify the low efficacy demonstrated by sulfonamide derivatives 5 b-d and 6a structurally related to parent compound 5a. Interestingly, the best active inhibitor **7e** ( $K_i = 806.4 \text{ nM}$ , see Figure 2(C)) demonstrated the ability to form additional interactions with a cluster of hydrophobic residues Ala106, Pro111, and Leu 113 when compared with unsubstituted analog 5a, thus suggesting that the 4-phenyl substituent of compound 7e is a crucial fragment to make more favourable hydrophobic contacts in the top of the cavity. This result indicates that the ligands having a good affinity towards VchCA $\beta$  are expected to establish additional hydrogen bonding interactions or hydrophobic contacts in the middle or top region of the enzymatic cavity.

### 4. Conclusions

In conclusions, a small series of benzenesulfonamides has been screened as inhibitors of *V. cholerae* carbonic anhydrases. Compound **7e** demonstrated interesting affinity against VchCA $\alpha$  and VchCA $\beta$  with  $K_i$  values of 89.9 and 806.4 nM, respectively. The predicted binding mode of compound **7e** in the modelled catalytic site of VchCA $\beta$  suggests that the introduction of an extra aromatic ring might improve the contacts in the top area of the catalytic cavity, thus furnishing suggestions for the rational design of new compounds targeting *V. cholerae* CAs.

### **Disclosure statement**

No potential conflict of interest was reported by the authors.

### Funding

This work was financially supported by Fondo di Ateneo per la Ricerca [PRA grant number ORME09SPNC – Università degli Studi di Messina].

# ORCID

Rosaria Gitto D http://orcid.org/0000-0003-0002-2253 Claudiu T. Supuran D http://orcid.org/0000-0003-4262-0323 Clemente Capasso D http://orcid.org/0000-0003-3314-2411

# References

- 1. Peterson KM, Gellings PS. Multiple intraintestinal signals coordinate the regulation of Vibrio cholerae virulence determinants. Pathog Dis 2018;76.
- 2. Muanprasat C, Chatsudthipong V. Cholera: pathophysiology and emerging therapeutic targets. Future Med Chem 2013;5: 781–98.
- Thomson JJ, Withey JH. Bicarbonate increases binding affinity of Vibrio cholerae ToxT to virulence gene promoters. J Bacteriol 2014;196:3872–80.

- Cobaxin M, Martinez H, Ayala G, et al. Cholera toxin expression by El Tor *Vibrio cholerae* in shallow culture growth conditions. Microb Pathog 2014;66:5–13.
- Lomelino CL, Andring JT, McKenna R. Crystallography and its impact on carbonic anhydrase research. Int J Med Chem 2018;2018:9419521.
- 6. Capasso C, Supuran CT. Bacterial, fungal and protozoan carbonic anhydrases as drug targets. Expert Opin Ther Targets 2015;19:1689–704.
- 7. Modak JK, Liu YC, Machuca MA, et al. Structural basis for the inhibition of *Helicobacter pylori* alpha-carbonic anhydrase by sulfonamides. PLoS One 2015;10:e0127149.
- Capasso C, Supuran CT. Anti-infective carbonic anhydrase inhibitors: a patent and literature review. Expert Opin Ther Pat 2013;23:693–704.
- 9. Capasso C, Supuran CT. Inhibition of bacterial carbonic anhydrases as a novel approach to escape drug resistance. Curr Top Med Chem 2017;17:1237–48.
- Alterio V, Di Fiore A, D'Ambrosio K, et al. Multiple binding modes of inhibitors to carbonic anhydrases: how to design specific drugs targeting 15 different isoforms? Chem Rev 2012;112:4421–68.
- 11. Supuran CT. Carbonic anhydrase inhibitors and their potential in a range of therapeutic areas. Expert Opin Ther Pat 2018;28:709–12.
- 12. De Luca L, Mancuso F, Ferro S, et al. Inhibitory effects and structural insights for a novel series of coumarin-based compounds that selectively target human CA IX and CA XII carbonic anhydrases. Eur J Med Chem 2018;143:276–82.
- 13. Bruno E, Buemi MR, Di Fiore A, et al. Probing molecular interactions between Human Carbonic Anhydrases (hCAs) and a novel class of benzenesulfonamides. J Med Chem 2017;60:4316–26.
- Bruno E, Buemi MR, De Luca L, et al. In vivo evaluation of selective carbonic anhydrase inhibitors as potential anticonvulsant agents. ChemMedChem 2016;11:1812–8.
- Buemi MR, De Luca L, Ferro S, et al. Carbonic anhydrase inhibitors: design, synthesis and structural characterization of new heteroaryl-N-carbonylbenzenesulfonamides targeting druggable human carbonic anhydrase isoforms. Eur J Med Chem 2015;102:223–32.
- De Luca L, Ferro S, Damiano FM, et al. Structure-based screening for the discovery of new carbonic anhydrase VII inhibitors. Eur J Med Chem 2014;71:105–11.
- 17. Gitto R, Damiano FM, Mader P, et al. Synthesis, structureactivity relationship studies, and X-ray crystallographic analysis of arylsulfonamides as potent carbonic anhydrase inhibitors. J Med Chem 2012;55:3891–9.
- Gitto R, Damiano FM, De Luca L, et al. Synthesis and biological profile of new 1,2,3,4-tetrahydroisoquinolines as selective carbonic anhydrase inhibitors. Bioorg Med Chem 2011;19:7003–7.
- 19. Mader P, Brynda J, Gitto R, et al. Structural basis for the interaction between carbonic anhydrase and 1,2,3,4-tetrahydroisoquinolin-2-ylsulfonamides. J Med Chem 2011;54:2522–6.
- Gitto R, Agnello S, Ferro S, et al. Identification of potent and selective human carbonic anhydrase VII (hCA VII) inhibitors. ChemMedChem 2010;5:823.
- Gitto R, Agnello S, Ferro S, et al. Identification of 3,4-Dihydroisoquinoline-2(1H)-sulfonamides as potent carbonic anhydrase inhibitors: synthesis, biological evaluation, and enzyme-ligand X-ray studies. J Med Chem 2010;53:2401–8.

- 22. Gitto R, Ferro S, Agnello S, et al. Synthesis and evaluation of pharmacological profile of 1-aryl-6,7-dimethoxy-3,4-dihydroi-soquinoline-2(1H)-sulfonamides. Bioorg Med Chem 2009;17: 3659–64.
- 23. Buemi MR, Di Fiore A, De Luca L, et al. Exploring structural properties of potent human carbonic anhydrase inhibitors bearing a 4-(cycloalkylamino-1-carbonyl)benzenesulfonamide moiety. Eur J Med Chem 2019;163:443–52.
- 24. De Luca V, Del Prete S, Supuran CT, et al. Protonography, a new technique for the analysis of carbonic anhydrase activity. J Enzyme Inhib Med Chem 2015;30:277–82.
- 25. Khalifah RG. The carbon dioxide hydration activity of carbonic anhydrase. I. Stop-flow kinetic studies on the native human isoenzymes B and C. J Biol Chem 1971;246:2561–73.
- 26. Del Prete S, Vullo D, De Luca V, et al. Anion inhibition profiles of alpha-, beta- and gamma-carbonic anhydrases from the pathogenic bacterium *Vibrio cholerae*. Bioorg Med Chem 2016;24:3413–7.
- 27. Del Prete S, Vullo D, De Luca V, et al. Anion inhibition profiles of the complete domain of the eta-carbonic anhydrase from *Plasmodium falciparum*. Bioorg Med Chem 2016;24: 4410–4.
- 28. De Luca V, Vullo D, Del Prete S, et al. Cloning, characterization and anion inhibition studies of a gamma-carbonic anhydrase from the Antarctic bacterium *Colwellia psychrerythraea*. Bioorg Med Chem 2016;24:835–40.
- 29. Ferraroni M, Del Prete S, Vullo D, et al. Crystal structure and kinetic studies of a tetrameric type II  $\beta$ -carbonic anhydrase from the pathogenic bacterium *Vibrio cholerae*. Acta Crystallogr D Biol Crystallogr 2015;71:2449–56.
- Discovery Studio 2.5.5 Accelrys. Biovia, San Diego, CA; 2009. [cited 2019 Jun 13]. Available from: http://www.accelrys.com.

- 31. Huang S, Hainzl T, Grundstrom C, et al. Structural studies of beta-carbonic anhydrase from the green alga *Coccomyxa*: inhibitor complexes with anions and acetazolamide. PLoS One 2011;6:e28458.
- Pedretti A, Villa L, Vistoli G. VEGA An open platform to develop chemo-bio-informatics applications, using plug-in architecture and script programming. J Comput-Aided Mol Des 2004;18:167–73.
- 33. Wolber G, Langer T. LigandScout: 3-D pharmacophores derived from protein-bound ligands and their use as virtual screening filters. J Chem Inf Model 2005;45:160–9.
- 34. Jones G, Willett P, Glen RC, et al. Development and validation of a genetic algorithm for flexible docking. J Mol Biol 1997;267:727–48.
- 35. Phillips JC, Braun R, Wang W, et al. Scalable molecular dynamics with NAMD. J Comput Chem 2005;26:1781–802.
- 36. Cronk JD, O'Neill JW, Cronk MR, et al. Cloning, crystallization and preliminary characterization of a beta-carbonic anhydrase from *Escherichia coli*. Acta Crystallogr D Biol Crystallogr 2000;56:1176.
- 37. Rowlett RS, Tu C, Lee J, et al. Allosteric site variants of *Haemophilus influenzae* beta-carbonic anhydrase. Biochemistry 2009;48:6146–56.
- 38. Covarrubias AS, Bergfors T, Jones TA, et al. Structural mechanics of the pH-dependent activity of beta-carbonic anhydrase from *Mycobacterium tuberculosis*. J Biol Chem 2006;281:4993–9.
- 39. Nishimori I, Minakuchi T, Vullo D, et al. Inhibition studies of the beta-carbonic anhydrases from the bacterial pathogen *Salmonella enterica* serovar *Typhimurium* with sulfonamides and sulfamates. Bioorg Med Chem 2011;19:5023–30.
- 40. Laskowski RA, Swindells MB. LigPlot+: multiple ligandprotein interaction diagrams for drug discovery. J Chem Inf Model 2011;51:2778–86.