

Regular Article

Synthesis and biological evaluation of burnnettiene A derivatives enabling discovery of novel fungicide candidates

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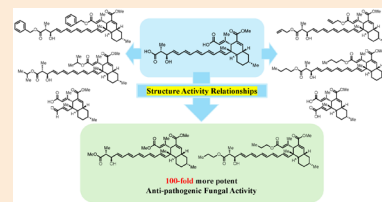
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Supplementary material

An antifungal polyene-decalin polyketide natural product, burnnettiene A (**1**) has been re-discovered from the culture broth of *Lecanicillium primulinum* (current name: *Flavocillium primulinum*) FKI-6715 strain utilizing our original multidrug-sensitive yeast system. This polyene-decalin polyketide natural product was originally isolated from *Aspergillus burnnettii*. The antifungal activity of **1** against *Candida albicans* has been reported. However, only one fungal species for the antifungal activity of **1** has been revealed, and details of the antifungal activity against other pathogenic fungus remain unknown. After extensive screening for antifungal activity, we found that **1** exhibits broad antifungal activity against pathogenic plant fungi, including *Colletotrichum gloeosporioides*, *Botrytis cinerea*, *Pyricularia oryzae*, *Leptosphaeria maculans*, and *Rhizoctonia solani*. Furthermore, we synthesized 12 derivatives from **1** and evaluated their antifungal activity to reveal the detailed structure–activity relationship. The methyl ester derivative showed antifungal activity against *Saccharomyces cerevisiae* 12geneΔ0HSR-iERG6 100-fold more potent than that of **1**. Our research indicates that **1** would be a promising natural product as a new fungicidal candidate and the methyl ester derivative especially has great potential.



Keywords: multidrug-sensitive budding yeast, natural product, derivatization, pathogenic plant fungi.

Introduction

We have been searching for new fungicide candidates from secondary metabolites produced by microorganisms using multidrug-sensitive yeast, *Saccharomyces cerevisiae* 12geneΔ0HSR-iERG6, as our test model.^{1,2} Using this system, we have been able to identify several new natural products with anti-microbial activities.^{3–7} Among them, we have recently dis-

covered a novel fungicide (sakurafusariene) against rice blast caused by *Pyricularia oryzae*.⁶ Sakurafusariene displayed extensive antifungal activity after screening of overlooked natural products utilizing the multidrug-sensitive yeast system. In this manner, we have displayed the utility of our strategy based on the screening method using the multidrug-sensitive yeast and extensive biological evaluation to discover the hidden antifungal activity. As a result, we re-discovered burnnettiene A (**1**)⁸ (Fig. 1), which was originally isolated as an antifungal compound.⁸ However, only one fungal species for the antifungal activity of **1** has been revealed and the detailed antifungal activity against other pathogenic fungus remains unknown. Pathogenic plant fungi threaten global food security for people all over the world.⁹ Additionally, pathogenic plant fungal diseases cause tremendous damage to the crop products comparable to enough food annually for 600 million people and there is growing resistance to current fungicides.¹⁰ Furthermore, food demand has been increasing due to population growth and economic devel-

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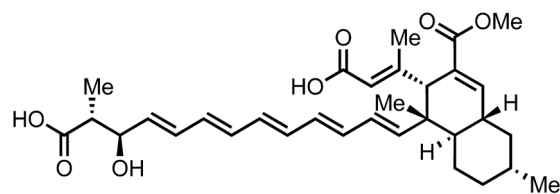


Fig. 1. Structure of burnnetiene A (1).

opment.¹¹) Therefore, there is need for novel fungicides to stabilize and increase crop supply. In this study, we re-discovered great potential of **1** as a new anti-fungicide candidate and synthesized 12 derivatives. Furthermore, we evaluated antifungal activity of derivatives, enabling us to verify the structure–activity relationship and discover a great fungicide candidate.

Materials and methods

1. General experimental procedures

High- and low-resolution mass spectra were obtained using an AB Sciex QSTAR Hybrid LC/MS/MS Systems (AB Sciex, Framingham, MA, USA) and JEOL JMS-T100LP (JEOL, Tokyo, Japan). NMR spectra were measured using a Varian XL-400 spectrometer (Agilent Technologies, CA, USA) with ¹H NMR and ¹³C NMR obtained at 400 MHz and 100 MHz, JEOL JNM-ECA-500 (JEOL, Tokyo, Japan) with ¹H NMR and ¹³C NMR obtained at 500 MHz and 125 MHz, and Bruker AV ANCE III HD600 (Bruker, Massachusetts, USA) with ¹H NMR and ¹³C NMR obtained at 600 MHz and 150 MHz in DMSO-*d*₆ and CDCl₃. The chemical shifts are reported in ppm and referenced to DMSO-*d*₆ (2.50 ppm) in the ¹H NMR spectra and DMSO-*d*₆ (39.52 ppm) in the ¹³C NMR, and CDCl₃ (7.26 ppm) in the ¹H NMR spectra and CDCl₃ (77.16 ppm) in the ¹³C NMR.

2. Antifungal activity evaluation

S. cerevisiae 12geneΔ0HSR-iERG6, *P. oryzae* APU15-60A (Quinone outside inhibitors (QoI)-sensitive strain),¹² *P. oryzae* APU15-63A (QoI-resistant strain),¹² *C. gloeosporioides* MAFF-237219,¹³ *L. maculans* MAFF-726728,¹⁴ *B. cinerea* MAFF-306820,¹⁵ and *R. solani* MAFF-237699¹⁶ were used as test organisms. Antifungal activity was evaluated by disc diffusion method.

Results and discussion

1. Antifungal activity screening

We screened compound **1** for antifungal activity against several pathogenic fungi. We found that **1** exhibits broad antifungal activity against plant pathogenic fungi including *C. gloeosporioides*, *B. cinerea*, *P. oryzae*, *L. maculans*, and *R. solani* (Table 1). These plant pathogenic fungi cause serious damage to wide variety of crops such as rice, vegetables, and fruits. To our delight, **1** showed potent antifungal activity against *C. gloeosporioides* and *B. cinerea* at a level comparable to the potent and broad-spectrum antifungal compound, amphotericin B. Quinone outside inhibitors (QoI) such as kresoxim-methyl are common fungicides for *P. oryzae*. However, widespread distribution of QoI-resistance *P. oryzae* is serious problem in rice-growing areas. Compound **1** are effective against both QoI-sensitive and resistant *P. oryzae* strains, showing great potential to become a new lead fungicide, which prompted us to explore **1** as a new antifungal candidate. Therefore, we decided to synthesize derivatives from **1** to verify the structure activity relationship and create more potent analogs.

2. Preliminary structure activity relationship of 1

To verify the preliminary structure activity relationship of **1**, we first decided to functionalize the characteristic functional groups such as the polyene and carboxylic groups moieties in **1** (Scheme 1). Considering the instability of **1**, hydrogenation with H₂ and Pd/C was conducted to afford saturated compound **2**. We sought the structure–activity relationship about the decalin moiety and ozonolysis of **1** in the presence of pyridine, provided aldehyde derivative **3** and acid **4**. Next, our attention was shifted to derivatization of carboxylic groups in **1** and we tried amidation and esterification conditions using condensation agents. However, these conditions were unfruitful and even a mild methyl ester formation reagent, TMSCHN₂ was not successful. We encountered a similar problem in derivatization of a polyene natural product, sakurafusariene, whose low reactivity would attribute to its conjugation stabilization from the polyene moiety.⁹) Eventually, alkylation conditions using electrophiles and K₂CO₃ were chosen for the synthesis of sakurafusariene ester de-

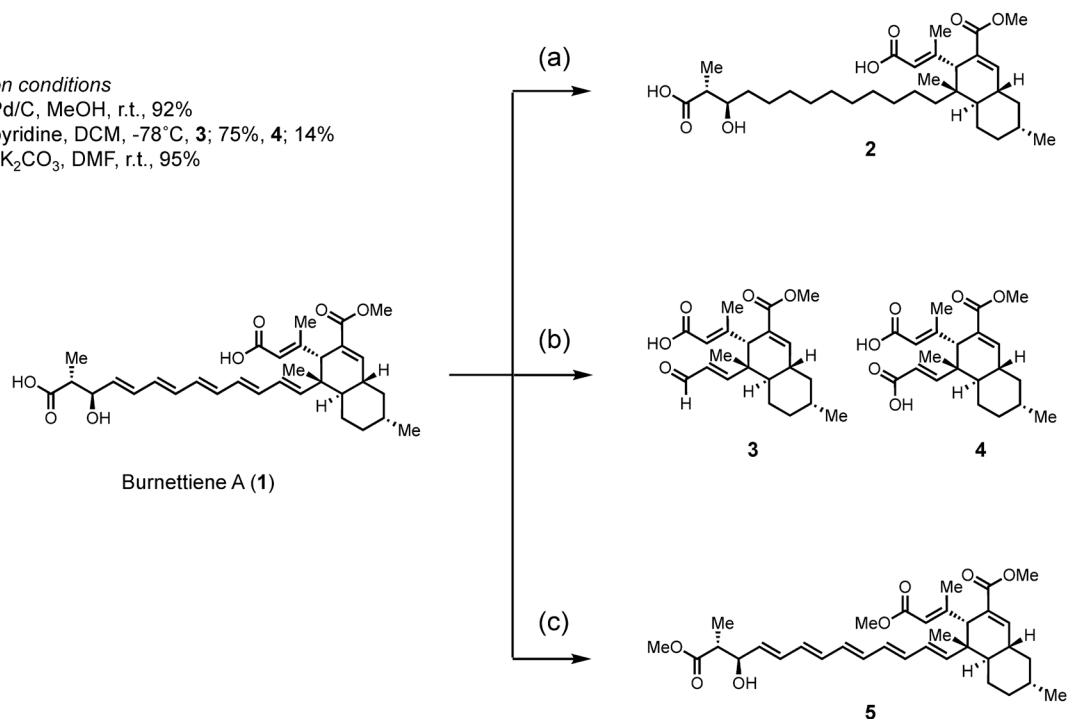
Table 1. Antifungal activity of **1** against phytopathogenic fungi.

Strain	Inhibition zone diameters (mm)							
	1 (μg/disk)						AMPH	KXM
	100	30	10	3	1	0.3	1	0.3
<i>Pyricularia oryzae</i> APU15-60A	12.7	11.5	8.5	—	—	N.T.	12.4	18.2
<i>Pyricularia oryzae</i> APU15-63A	13.4	13.1	10.3	—	—	N.T.	14.6	—
<i>Colletotrichum gloeosporioides</i> MAFF-237219	14.6/28.5	12.5/28.1	11.8/27.4	10.5/24.5	18.2	—	12.3	N.T.
<i>Leptosphaeria maculans</i> MAFF-726728	13.9	9.0	—	—	—	N.T.	13.0	N.T.
<i>Botrytis cinerea</i> MAFF-306820	15.1	13.3	12.9	12.2	11.6	—	8.4	N.T.
<i>Rhizoctonia solani</i> MAFF-237699	7.2	—	—	—	—	N.T.	8.2	N.T.

AMPH, Amphotericin B; KXM, Kresoxim-methyl, N.T., Not tested; —, No inhibition; Inner diameter/outer diameter, *Pyricularia oryzae* APU15-60A, Susceptible to QoI; *Pyricularia oryzae* APU15-63A, Resistance to QoI.

Reaction conditions

- a) H₂, Pd/C, MeOH, r.t., 92%
 b) O₃, pyridine, DCM, -78°C, **3**; 75%, **4**; 14%
 c) MeI, K₂CO₃, DMF, r.t., 95%



Scheme 1. Synthesis of Derivatives-1. a) H₂, Pd/C, MeOH, r.t., 92.0%, b) O₃, pyridine, DCM, -78.0°C, **3**; 75.0%, **4**; 14.0%, c) MeI, K₂CO₃, DMF, r.t., 95.0%.

rivatives, which prompted us to carry out alkylation of **1**. Methylation of the carboxylic groups in **1** utilizing MeI and K₂CO₃ proceeded to furnish methyl ester **5** as we expected. We evaluated the antifungal activity of those preliminary five derivatives against the multidrug-sensitive budding yeast (Table 2). The parent compound shows antifungal activity at 3 μg/disk, whereas saturated compound **2** did not show antifungal activity even at 100 μg/disk. Aldehyde derivative **3** and acid **4** retain antifungal activity, suggesting that the polyene moiety would not be an essential functional group, but the saturated side chain in **2** might negatively influence the antifungal activity due to its hydrophobicity or flexibility. To our delight, methyl ester derivative **5** exhibits a 100-fold increase in antifungal activity compared to **1**, driving us to synthesis more ester derivatives.

3. Synthesis of ester derivatives

Methyl ester **5** was found to be a promising derivative and we synthesized several ester derivatives (Scheme 2). In terms of length of the alkyl group, ethyl, propionyl, and butyl esters were synthesized (**6–8**). To verify the influence of unsaturated and branched functional groups to the antifungal activity, allyl, benzyl, propargyl, isopropyl, and isobutyl esters were also derivatized from **1** (**9–13**).

4. Antifungal activity evaluation of derivatives

With 12 derivatives in hand, we evaluated the antifungal activity of them against QoI-sensitive *P. oryzae*, QoI-resistant *P. oryzae*, *C. gloeosporioides*, *L. maculans*, *B. cinerea*, and *R. solani*. Normal alkyl esters such as methyl, ethyl, and propyl exhibited potent antifungal activity against QoI-sensitive *P. oryzae* more

Table 2. Antifungal activity of **1–5** against *Saccharomyces cerevisiae* 12geneΔ0HSR-iERG6.

Compound	Inhibition zone diameter (mm)								
	μg/disk								
	100	30	10	3	1	0.3	0.1	0.03	0.01
1	19.3	12.2	11.8	9.2	—	—	—	N.T.	N.T.
2	—	—	—	—	—	—	—	N.T.	N.T.
3	9.3	8.2	—	—	—	—	—	N.T.	N.T.
4	10.0	10.3	—	—	—	—	—	N.T.	N.T.
5	16.0	15.6	12.3	11.0	10.9	9.1	7.7	6.5	—
AMPH	N.T.	N.T.	N.T.	N.T.	11.6	N.T.	N.T.	N.T.	N.T.

AMPH, Amphotericin B, N.T., Not tested; —, No inhibition.



Scheme 2. Synthesis of Ester Derivatives

- 6; R = Ethyl (45% yield)
 7; R = Propyl (26% yield)
 8; R = Butyl (21% yield)
 9; R = Allyl (22% yield)
 10; R = Benzyl (34% yield)
 11; R = Propargyl (32% yield)
 12; R = Isopropyl (28% yield)
 13; R = Isobutyl (7% yield)

Scheme 2. Synthesis of Ester Derivatives. 6; R=Ethyl (45.0% yield), 7; R=Propyl (26.0% yield), 8; R=Butyl (21.0% yield), 9; R=Allyl (22.0% yield), 10; R=Benzyl (34.0% yield), 11; R=Propargyl (32.0% yield), 12; R=Isopropyl (28.0% yield), 13; R=Isobutyl (7.0% yield).

than 10 to 30-folds compared to the parent compound (Table 3). Interestingly, aldehyde derivative 3, which did not show significant antifungal activity against the multidrug-sensitive budding yeast, displayed great potency against QoI-sensitive *P. oryzae*. Additionally, propargyl ester 11 also showed potent antifungal activity, which would provide a great opportunity to synthesize novel derivatives utilizing click chemistry.¹⁷⁾ Benzyl ester 10 and isobutyl 13 retained antifungal activity against QoI-resistance *P. oryzae*, suggesting that branch carbon chains attached to a methylene group would be a better functional group than the other groups. In the case of *C. gloeosporioides*, all derivatives showed weak antifungal activity compared to the parent compound, but the reduction product 2, which was not effective against the multidrug-sensitive budding yeast, exhibited slight better efficacy than the other derivatives. Furthermore, 2 was found to retain antifungal activity even against *L. maculans* comparable to 1, which would lead us to discover more stable derivatives based on 2. The antifungal activity of 5 against *B. cinerea* was comparably effective to that of the parent compound, and alkyl derivatives such as 6 and 12 maintained some effectiveness. Overall, esterification of 1 facilitated the discovery of novel antifungal derivatives. Methyl ester 5, especially displayed antifungal activity against QoI-sensitive *P. oryzae* and *B. cinerea*. Benzyl ester 10 exhibited good antifungal activity against QoI-resistance *P. oryzae*; introduction of substituents on the aromatic ring might open up possibility for more potent antifungal derivatives. Intriguingly, reduction derivative 2 is relatively active against *C. gloeosporioides* and *L. maculans*, representing new insight into more stable derivatives.

Conclusions

We have re-discovered a new antifungal natural product, burnettiene A (1) from the culture broth of *Flavocillium primulinum* FKI-6715 strain using a multidrug-sensitive budding yeast screening system.³⁾ After extensive antifungal activity screening to uncover hidden antifungal activity, 1 was found to exhibit broad antifungal activity against pathogenic plant fungi. This result prompted us to synthesize new derivatives and evaluated the antifungal activity of them to verify the structure activity rela-

tionship. As a first approach, we synthesized four derivatives focused on the characteristic functional groups of 1 and evaluated the antifungal activity of them against the multidrug-sensitive budding yeast, indicating the preliminary structure activity relationship and methyl ester 5 exhibited a 100-fold increase in antifungal activity compared to 1. Based on this finding, we derivatized the natural product to several ester derivatives. With 12 derivatives in hand, we evaluated antifungal activity against pathogenic plant fungi such as *C. gloeosporioides*, *B. cinerea*, *P. oryzae*, *L. maculans*, and *R. solani*. We obtained valuable knowledge about the structure activity relationship. Notably, methyl ester 5, showed more potent antifungal activity against QoI-sensitive *P. oryzae* than 1. Benzyl ester 10 and isobutyl 13 exhibited good antifungal activity against QoI-resistance *P. oryzae*. Moreover, reduction derivative 2, which does not show antifungal activity against the multidrug-sensitive budding yeast, was relatively active against *C. gloeosporioides* and *L. maculans*. Therefore, our study enabled us to discover appropriate seed compounds according to pathogenic plant fungi and showed the utility of our strategy based on the multidrug-sensitive yeast screening system and chemical synthesis. We are currently carrying out synthesis of new derivatives based on each seed compound and planning to evaluate the ant-fungal activity of them *in vivo* model.

Associated content

Preparation of the compounds

Burnettiene A (1)

Burnettiene A (1) used for derivatization was prepared from the cultured broth of *F. primulinum* FKI-6715 strain as shown in supporting information.

Saturated derivative 2

To a solution of 1 (6.0 mg, 11.1 μ mol) in MeOH (0.7 mL, 15.9 mM), 10% Pd/C (1.0 mg, 0.9 mmol) was added and stirred at room temperature under a H₂ atmosphere for 1 hr. The reaction mixture was filtered with celite and concentrated under reduced pressure. The residue was purified by prep. TLC to obtain 2 (5.6 mg, 10.2 μ mol, 92.0%). $[\alpha]_D^{23} -77.5$ (c 0.1, CHCl₃); UV (CHCl₃) λ_{max} (ϵ) 243 (6,964), 377 (1,535). ¹H NMR (400 MHz, DMSO-*d*₆) δ 6.75 (s, 1H), 5.42 (s, 1H), 3.78 (s, 1H), 3.60 (s, 3H),

Table 3. Antifungal activity of 1–13 against phytopathogenic fungi.

Strain	Compound	Inhibition zone diameter (mm)						Strain	Compound	Inhibition zone diameter (mm)				
		$\mu\text{g}/\text{disk}$								$\mu\text{g}/\text{disk}$				
		100	30	10	3	1	0.3			100	30	10	3	1
<i>Pyricularia oryzae</i> APU15-60A (QoIS)	1	11.6	10.9	8.9	—	—	—	<i>Pyricularia oryzae</i> APU15-63A (QoIR)	1	12.8	11.9	9.5	—	—
	2	11.1	8.2	7.1	—	—	—		2	8.2	—	—	—	—
	3	9.0	8.7	7.7	7.2	7.0	—		3	11.9	10.5	—	—	—
	4	13.2	10.9	10.5	—	—	N.T.		4	13.9	—	—	—	—
	5	13.7	11.4	12.2	10.4	8.4	7.1		5	14.8	9.3	—	—	—
	6	12.5	11.5	10.3	10.4	9.8	—		6	12.9	—	—	—	—
	7	14.0	12.4	10.5	9.4	10.3	7.8		7	11.7	—	—	—	—
	8	14.6	11.6	—	—	—	N.T.		8	14.4	—	—	—	—
	9	14.5	12.0	—	—	—	N.T.		9	16.9	—	—	—	—
	10	11.8	11.3	11.5	—	—	N.T.		10	9.0	8.2	6.3	—	—
	11	14.1	11.0	9.8	9.6	—	N.T.		11	13.4	—	—	—	—
	12	14.1	11.6	10.6	—	—	N.T.		12	12.9	—	—	—	—
	13	14.1	—	—	—	—	N.T.		13	13.9	11.5	10.9	—	—
	AMPH	N.T.	N.T.	N.T.	N.T.	10.4	N.T.	AMPH	N.T.	N.T.	N.T.	N.T.	10.3	
	KXM	N.T.	N.T.	N.T.	N.T.	N.T.	21.5	KXM	N.T.	N.T.	N.T.	N.T.	—	
<i>Colletotrichum gloeosporioides</i> MAFF-237219	1	19.9	18.3	16.9	13.0	12.5	N.T.	<i>Botrytis cinerea</i> MAFF-306820	1	15.0	13.3	11.5	10.2	9.1
	2	10.5	9.9	—	—	—	N.T.		2	11.6	—	—	—	—
	3	9.1	—	—	—	—	N.T.		3	9.9	—	—	—	—
	4	—	—	—	—	—	N.T.		4	9.6	—	—	—	—
	5	14.0	—	—	—	—	N.T.		5	11.1	10.1	9.1	9.0	—
	6	—	—	—	—	—	N.T.		6	10.1	9.5	9.4	—	—
	7	17.7	—	—	—	—	N.T.		7	—	—	—	—	—
	8	18.4	—	—	—	—	N.T.		8	9.8	9.3	—	—	—
	9	—	—	—	—	—	N.T.		9	7.7	—	—	—	—
	10	—	—	—	—	—	N.T.		10	9.9	8.6	—	—	—
	11	14.3	—	—	—	—	N.T.		11	9.8	8.5	—	—	—
	12	14.2	—	—	—	—	N.T.		12	10.6	9.9	9.9	—	—
	13	15.9	—	—	—	—	N.T.		13	10.7	9.3	—	—	—
	AMPH	N.T.	N.T.	N.T.	N.T.	13.3	N.T.	AMPH	N.T.	N.T.	N.T.	N.T.	10.2	
<i>Leptosphaeria maculans</i> MAFF-726728	1	12.7	10.5	8.9	—	—	N.T.	<i>Rizoctonia solani</i> MAFF-237699	1	—	—	—	—	—
	2	8.7	8.6	8.1	—	—	N.T.		2	—	—	—	—	—
	3	8.3	—	—	—	—	N.T.		3	—	—	—	—	—
	4	—	—	—	—	—	N.T.		4	—	—	—	—	—
	5	—	—	—	—	—	N.T.		5	—	—	—	—	—
	6	—	—	—	—	—	N.T.		6	—	—	—	—	—
	7	—	—	—	—	—	N.T.		7	—	—	—	—	—
	8	—	—	—	—	—	N.T.		8	—	—	—	—	—
	9	—	—	—	—	—	N.T.		9	—	—	—	—	—
	10	—	—	—	—	—	N.T.		10	—	—	—	—	—
	11	—	—	—	—	—	N.T.		11	—	—	—	—	—
	12	10.3	—	—	—	—	N.T.		12	—	—	—	—	—
	13	9.9	8.1	—	—	—	N.T.		13	—	—	—	—	—
	AMPH	N.T.	N.T.	N.T.	N.T.	12.8	N.T.	AMPH	N.T.	N.T.	N.T.	N.T.	8.5	

AMPH, Amphotericin B; KXM, Kresoxim-methyl, N.T., Not tested; —, No inhibition, *Pyricularia oryzae* APU15-60A, Susceptible to QoI; *Pyricularia oryzae* APU15-63A, Resistance to QoI.

2.80 (s, 1H), 2.32 (m, 1H), 2.13 (s, 3H), 1.96 (m, 1H), 1.93 (m, 1H), 1.75 (m, 1H), 1.50 (m, 1H), 1.40 (m, 2H), 1.37 (m, 1H), 1.19–1.27 (complex m, 18H), 1.23 (m, 1H), 0.96 (d, $J=7.0$ Hz, 3H), 0.90 (d, $J=6.6$ Hz, 3H), 0.81–1.11 (complex m, 3H), 0.75 (s, 3H); and ^{13}C NMR (125 MHz, DMSO- d_6) δ 176.55, 167.36, 166.54, 154.79, 142.83, 130.24, 120.30, 79.24, 71.66, 54.62, 51.63, 45.90, 40.47, 38.55, 38.30, 36.89, 35.39, 33.34, 32.83, 32.28, 30.22, 29.18, 29.10, 29.00, 28.64, 25.45, 25.21, 23.44, 22.32, 21.56, 18.43, 12.87. ESI-MS m/z 547.3648 $[\text{M}-\text{H}]^-$ (Calcd. for $\text{C}_{32}\text{H}_{51}\text{O}_7$, m/z 547.3635 $[\text{M}-\text{H}]^-$).

Aldehyde derivative 3 and acid derivative 4

To a solution of **1** (10.0 mg, 18.6 μmol) in pyridine (7.4 μL , 92.9 μmol) and dichloromethane (740.0 μL , 25.0 mM) at -78.0°C , O_3 was bubbled through the solution for few minutes. The reaction mixture was allowed to warm to room temperature and then 1N HCl aq. was added. The resulting mixture was extracted with CHCl_3 , then the organic phase was dried over Na_2SO_4 and concentrated under vacuum. The residue was purified by prep. TLC ($\text{CHCl}_3/\text{MeOH}=5:1$) to obtain **3** (5.0 mg, 13.9 μmol , 75.0%) and **4** (1.0 mg, 2.7 μmol , 14.0%). Physicochemical properties of **3**: $[\alpha]_D^{23} -146.6$ (c 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 242 (12,463), 279 (8,104), 323 (3,242); ^1H NMR (400 MHz, CDCl_3) δ 9.56 (d, $J=11.4$ Hz, 1H), 6.95 (s, 1H), 6.49 (d, $J=24.0$ Hz, 1H), 6.20 (dd, $J=24.0, 11.4$ Hz, 1H), 5.59 (s, 1H), 3.70 (s, 3H), 3.07 (s, 1H), 2.23 (d, $J=1.36$ Hz, 3H), 1.99 (m, 1H), 1.95 (m, 1H), 1.78 (m, 1H), 1.53 (m, 1H), 1.39 (m, 1H), 1.23 (m, 1H), 1.10 (s, 3H), 0.96 (d, $J=6.6$ Hz, 3H), 0.85–1.07 (complex m, 3H); and ^{13}C NMR (125 MHz, CDCl_3) δ 193.86, 170.73, 166.81, 164.77, 152.12, 143.76, 132.16, 129.31, 118.76, 56.40, 52.09, 42.89, 40.70, 40.14, 38.62, 35.32, 33.48, 29.84, 26.87, 22.31, 18.19. $[\text{M}+\text{NH}_4]^+$ (Calcd. for $\text{C}_{21}\text{H}_{32}\text{NO}_5$, m/z 378.2280 $[\text{M}+\text{NH}_4]^+$). Physicochemical properties of **4**: $[\alpha]_D^{23} -149.3$ (c 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 242 (11,022), 260 (7,862), 310 (6,169); ^1H NMR (600 MHz, CDCl_3) δ 6.93 (s, 1H), 6.72 (d, $J=16.7$ Hz, 1H), 5.86 (d, $J=16.7$ Hz, 1H), 5.55 (s, 1H), 3.69 (s, 3H), 3.03 (s, 1H), 2.17 (s, 3H), 1.96 (m, 1H), 1.93 (m, 1H), 1.75 (m, 1H), 1.50 (m, 1H), 1.37 (m, 1H), 1.23 (m, 1H), 1.07 (s, 3H), 0.94 (d, $J=6.5$ Hz, 3H), 0.79–1.03 (complex m, 3H); and ^{13}C NMR (150 MHz, CDCl_3) δ 207.14, 172.11, 166.92, 162.67, 158.13, 143.74, 129.47, 120.16, 118.84, 56.41, 52.02, 42.40, 40.73, 39.94, 38.75, 35.31, 33.55, 31.06, 26.88, 22.34, 18.00. ESI-MS m/z 394.2227 $[\text{M}+\text{NH}_4]^+$ (Calcd. for $\text{C}_{21}\text{H}_{32}\text{NO}_6$, m/z 394.2230 $[\text{M}+\text{NH}_4]^+$).

General method for preparation of alkyl derivatives 5 and 11

To a solution of **1** (10.0 mg, 18.6 μmol) in DMF (740.0 μL , 25.0 mM) was added alkyl halide (92.9 μmol) and K_2CO_3 (12.7 mg, 92.9 μmol) at room temperature and stirred for 1 hr. The reaction mixture was quenched with brine and diluted with EtOAc. The organic phase was dried over Na_2SO_4 and concentrated under reduced pressure. The residue was purified by prep. TLC to obtain alkyl derivatives.

Methyl ester 5

Burnettiene A methyl ester **5** was prepared by added iodoethane (5.7 μL , 92.9 μmol) according to the general method

for preparation of alkyl derivatives. The residue was purified by prep. TLC (n -hexane/EtOAc=1:1) to obtain **5** (10.0 mg, 17.7 μmol , 95.0%). $[\alpha]_D^{23} -235.1$ (c 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 241 (10,987), 310 (21,690), 324 (39,643), 339 (56,915), 357 (55,669). ^1H NMR (400 MHz, CDCl_3) δ 6.88 (s, 1H), 6.10–6.35 (complex m, 8H), 5.65 (dd, $J=14.9, 7.3$ Hz, 1H), 5.48 (d, $J=1.5$ Hz, 1H), 5.34 (d, $J=15.4$ Hz, 1H), 4.26 (t, $J=10.1$ Hz, 1H), 3.71 (s, 3H), 3.68 (s, 3H), 3.67 (s, 3H), 2.96 (s, 1H), 2.58 (m, 1H), 2.22 (d, $J=1.6$ Hz, 3H), 1.86–1.97 (complex m, 2H), 1.73 (m, 1H), 1.44 (m, 1H), 1.23–1.28 (complex m, 2H), 1.02 (s, 3H), 1.16 (d, $J=7.2, 3\text{H}$), 0.93 (d, $J=6.6$ Hz, 3H), 0.85–1.00 (complex m, 3H); and ^{13}C NMR (100 MHz, CDCl_3) δ 176.04, 167.27, 167.18, 161.82, 143.59, 142.47, 134.19, 133.99, 133.75, 132.92, 132.75, 132.43, 131.72, 131.62, 130.09, 129.25, 118.09, 74.60, 56.87, 52.01, 51.87, 51.03, 45.73, 41.82, 40.93, 40.76, 39.01, 35.60, 33.60, 29.83, 26.78, 22.41, 18.40, 14.24. ESI-MS m/z 584.3582 $[\text{M}+\text{NH}_4]^+$ (Calcd. for $\text{C}_{34}\text{H}_{50}\text{NO}_7$, m/z 584.3587 $[\text{M}+\text{NH}_4]^+$).

Propargyl ester 11

Burnettiene A propargyl ester **11** was prepared by added propargyl bromide (7.0 μL , 92.9 μmol) according to the general method for preparation of alkyl derivatives. The residue was purified by prep. TLC (n -hexane/EtOAc=2:1) to obtain **11** (3.6 mg, 5.9 μmol , 32.0%). $[\alpha]_D^{23} -303.9$ (c 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 241 (12,717), 310 (23,713), 324 (40,607), 339 (53,323), 358 (53,323). ^1H NMR (400 MHz, CDCl_3) δ 6.89 (s, 1H), 6.10–6.36 (complex m, 8H), 5.66 (dd, $J=14.9, 7.1$ Hz, 1H), 5.51 (s, 1H), 5.33 (d, $J=15.2$ Hz, 1H), 4.69–4.72 (complex m, 4H), 4.30 (t, $J=14.3$ Hz, 1H), 3.67 (s, 3H), 2.97 (s, 1H), 2.63 (m, 1H), 2.49 (d, $J=2.4$ Hz, 1H), 2.47 (d, $J=2.4$ Hz, 1H), 2.23 (d, $J=1.4$ Hz, 3H), 1.86–1.97 (complex m, 2H), 1.73 (m, 1H), 1.40–1.50 (complex m, 2H), 1.22 (m, 1H), 1.18 (d, $J=7.2$ Hz, 3H), 1.03 (s, 3H), 0.94 (d, $J=6.5$ Hz, 3H), 0.81–1.00 (complex m, 3H); and ^{13}C NMR (125 MHz, CDCl_3) δ 177.01, 174.66, 167.20, 165.70, 143.85, 142.34, 139.84, 134.35, 134.07, 133.75, 133.15, 132.45, 132.40, 131.78, 131.55, 129.92, 129.33, 78.33, 77.58, 75.20, 74.75, 74.55, 57.01, 52.29, 51.91, 51.42, 45.74, 41.91, 40.85, 40.76, 38.99, 35.53, 33.60, 29.84, 26.76, 22.40, 18.44, 14.03. ESI-MS m/z 632.3589 $[\text{M}+\text{NH}_4]^+$ (Calcd. for $\text{C}_{38}\text{H}_{50}\text{NO}_7$, m/z 632.3587 $[\text{M}+\text{NH}_4]^+$).

General method for preparation of alkyl derivatives 6–10 and 11–13

To a solution of **1** (20.0 mg, 37.2 μmol) in DMF (740.0 μL , 50.0 mM) was added alkyl halide (186.0 μmol) and K_2CO_3 (25.7 mg, 186.0 μmol) at room temperature. The mixture was stirred for 1 hr, then the reaction mixture was quenched with brine and diluted with EtOAc. The organic phase was dried over Na_2SO_4 and concentrated under reduced pressure. The residue was purified by prep. TLC to obtain alkyl derivatives.

Ethyl ester 6

Burnettiene A ethyl ester **6** was prepared by added iodoethane (14.8 μL , 186.0 μmol) according to the general method for preparation of alkyl derivatives. The residue was purified by prep. TLC (n -hexane/EtOAc=2:1) to obtain **6** (10.0 mg, 16.8 μmol , 45.0%). $[\alpha]_D^{23} -196.3$ (c 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 241 (16,391), 296 (17,106), 311 (24,065), 324 (34,537), 339 (41,626), 357

(40,326), 386 (8,130), 411 (6,439). ^1H NMR (400 MHz, CDCl_3) δ 6.89 (s, 1H), 6.12–6.32 (complex m, 8H), 5.65 (dd, $J=14.8$, 6.5 Hz, 1H), 5.45 (s, 1H), 5.35 (d, $J=15.2$ Hz, 1H), 4.26 (m, 1H), 4.09–4.19 (complex m, 4H), 3.67 (s, 3H), 2.95 (s, 1H), 2.56 (m, 1H), 2.20 (s, 3H), 1.90–1.97 (complex m, 2H), 1.73 (m, 1H), 1.47 (m, 1H), 1.23–1.34 (complex m, 8H), 1.02 (s, 3H), 1.16 (d, $J=7.1$ Hz, 3H), 0.95 (d, $J=6.8$ Hz, 3H), 0.79–1.02 (complex m, 3H); and ^{13}C NMR (100 MHz, CDCl_3) δ 175.64, 167.33, 166.88, 143.60, 142.53, 134.11, 133.95, 133.75, 132.87, 132.78, 132.45, 131.70, 131.66, 130.11, 129.22, 118.57, 74.56, 60.86, 59.84, 56.89, 51.88, 45.76, 41.84, 40.94, 40.74, 39.04, 35.60, 33.62, 29.84, 26.79, 22.83, 22.42, 18.42, 14.45, 14.34, 14.25. ESI-MS m/z 612.3903 $[\text{M}+\text{NH}_4]^+$ (Calcd. for $\text{C}_{36}\text{H}_{54}\text{NO}_7$, m/z 612.3900 $[\text{M}+\text{NH}_4]^+$).

Propyl ester 7

Burnnettene A propyl ester **7** was prepared by added 1-iodopropane (21.0 μL , 186.0 μmol) according to the general method for preparation of alkyl derivatives. The residue was purified by prep. TLC (*n*-hexane/EtOAc=2:1) to obtain **7** (6.0 mg, 9.6 μmol , 26.0%). $[\alpha]_{\text{D}}^{23}$ -241.1 (*c* 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 241 (16,058), 296 (16,680), 311 (24,460), 324 (36,908), 339 (46,741), 357 (45,745), 390 (9,958), 411 (7,967). ^1H NMR (400 MHz, CDCl_3) δ 6.89 (s, 1H), 6.10–6.35 (complex m, 8H), 5.66 (dd, $J=14.9$, 7.0 Hz, 1H), 5.46 (s, 1H), 5.35 (d, $J=15.2$ Hz, 1H), 4.25 (t, $J=13.7$ Hz, 1H), 4.00–4.09 (complex m, 4H), 3.67 (s, 3H), 2.95 (s, 1H), 2.57 (m, 1H), 2.20 (s, 3H), 1.87–1.96 (complex m, 2H), 1.63–1.72 (complex m, 5H), 1.39–1.49 (complex m, 2H), 1.21 (m, 1H), 1.02 (s, 3H), 1.16 (d, $J=7.2$ Hz, 3H), 0.96 (d, $J=3.2$ Hz, 3H), 0.94 (d, $J=2.8$ Hz, 3H), 0.92 (d, $J=2.0$ Hz, 3H), 0.78–0.99 (complex m, 3H); and ^{13}C NMR (100 MHz, CDCl_3) δ 175.69, 167.32, 167.31, 166.98, 143.60, 142.52, 134.07, 133.92, 133.73, 132.91, 132.74, 132.45, 131.69, 130.09, 129.20, 118.67, 74.51, 66.43, 65.57, 56.89, 51.86, 45.82, 41.82, 41.03, 39.04, 35.62, 33.60, 32.05, 31.81, 29.82, 26.78, 23.92, 22.42, 22.15, 18.41, 14.26, 10.63, 10.49. ESI-MS m/z 640.4220 $[\text{M}+\text{NH}_4]^+$ (Calcd. for $\text{C}_{38}\text{H}_{58}\text{NO}_7$, m/z 640.4213 $[\text{M}+\text{NH}_4]^+$).

Butyl ester 8

Burnnettene A butyl ester **8** was prepared by added 1-iodobutane (21.0 μL , 186.0 μmol) according to the general method for preparation of alkyl derivatives. The residue was purified by prep. TLC to obtain **8** (5.1 mg, 7.8 μmol , 21.0%). $[\alpha]_{\text{D}}^{23}$ -243.6 (*c* 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 241 (14,569), 296 (15,545), 310 (22,504), 324 (32,781), 339 (39,936), 357 (40,001), 390 (11,773), 411 (9,691). ^1H NMR (400 MHz, CDCl_3) δ 6.89 (s, 1H), 6.12–6.33 (complex m, 8H), 5.66 (dd, $J=14.7$, 7.1 Hz, 1H), 5.46 (s, 1H), 5.35 (d, $J=15.4$ Hz, 1H), 4.25 (t, $J=13.8$ Hz, 1H), 4.03–4.17 (complex m, 4H), 3.67 (s, 3H), 2.96 (s, 1H), 2.57 (m, 1H), 2.20 (s, 3H), 1.88–1.97 (complex m, 2H), 1.73 (m, 1H), 1.59–1.66 (complex m, 4H), 1.34–1.49 (complex m, 6H), 1.23 (m, 1H), 1.17 (d, $J=7.2$, 3H), 1.02 (s, 3H), 0.91–0.98 (complex m, 9H), 0.79–0.95 (complex m, 3H); and ^{13}C NMR (100 MHz, CDCl_3) δ 175.71, 168.85, 167.33, 167.01, 143.61, 142.54, 134.10, 133.93, 133.74, 132.89, 132.74, 132.45, 131.70, 131.68, 130.09, 129.21, 118.67, 74.52, 64.74, 63.91, 56.89, 51.87, 45.79, 41.83, 40.96,

40.77, 39.05, 35.63, 33.61, 30.86, 30.77, 29.84, 26.80, 22.83, 22.43, 19.42, 19.25, 18.42, 13.94, 13.83. ESI-MS m/z 668.4529 $[\text{M}+\text{NH}_4]^+$ (Calcd. for $\text{C}_{40}\text{H}_{62}\text{NO}_7$, m/z 668.4526 $[\text{M}+\text{NH}_4]^+$).

Allyl ester 9

Burnnettene A allyl ester **9** was prepared by added allyl bromide (15.9 μL , 186.0 μmol) according to the general method for preparation of alkyl derivatives. The residue was purified by prep. TLC (*n*-hexane/EtOAc=2:1) to obtain **9** (5.0 mg, 8.1 μmol , 22.0%). $[\alpha]_{\text{D}}^{23}$ -285.7 (*c* 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 242 (13,666), 310 (26,404), 323 (49,468), 339 (74,141), 357 (70,864). ^1H NMR (500 MHz, CDCl_3) δ 6.89 (s, 1H), 6.11–6.35 (complex m, 8H), 5.88–5.99 (complex m, 2H), 5.66 (dd, $J=14.9$, 7.2 Hz, 1H), 5.50 (s, 1H), 5.35 (d, $J=15.3$ Hz, 1H), 5.34 (brs, 1H), 5.31 (brs, 1H), 5.25 (s, 1H), 5.23 (s, 1H), 4.55–4.64 (complex m, 4H), 4.28 (t, $J=13.9$ Hz, 1H), 3.66 (s, 3H), 2.96 (s, 1H), 2.61 (m, 1H), 2.22 (s, 3H), 1.87–1.96 (complex m, 2H), 1.73 (m, 1H), 1.41–1.51 (complex m, 2H), 1.24 (m, 1H), 1.18 (d, $J=7.2$ Hz, 3H), 1.02 (s, 3H), 0.93 (d, $J=6.6$ Hz, 3H), 0.84–0.99 (complex m, 3H); and ^{13}C NMR (125 MHz, CDCl_3) δ 175.26, 167.28, 166.40, 162.21, 143.70, 142.45, 134.19, 133.98, 133.74, 132.93, 132.80, 132.72, 132.44, 132.06, 131.73, 131.62, 130.01, 129.25, 118.56, 118.19, 118.09, 74.57, 65.46, 64.76, 56.90, 51.89, 45.82, 41.84, 40.89, 40.74, 38.99, 35.56, 32.06, 26.81, 22.83, 22.41, 18.40, 14.25. ESI-MS m/z 636.3903 $[\text{M}+\text{NH}_4]^+$ (Calcd. for $\text{C}_{38}\text{H}_{54}\text{NO}_7$, m/z 636.3900 $[\text{M}+\text{NH}_4]^+$).

Benzyl ester 10

Burnnettene A benzyl ester **10** was prepared by added benzyl bromide (21.9 μL , 186.0 μmol) according to the general method for preparation of alkyl derivatives. The residue was purified by prep. TLC (*n*-hexane/EtOAc=2:1) to obtain **10** (9.2 mg, 12.8 μmol , 34.0%). $[\alpha]_{\text{D}}^{23}$ -259.2 (*c* 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 241 (19,468), 311 (29,945), 324 (47,198), 340 (65,661), 358 (65,158), 392 (13,290), 414 (10,848). ^1H NMR (400 MHz, CDCl_3) δ 7.32–7.40 (complex m, 10H), 6.89 (s, 1H), 6.11–6.34 (complex m, 8H), 5.65 (dd, $J=14.8$, 7.1 Hz, 1H), 5.53 (s, 1H), 5.35 (d, $J=15.4$ Hz, 1H), 5.10–5.18 (complex m, 4H), 4.29 (t, $J=13.0$ Hz, 1H), 3.66 (s, 3H), 2.98 (s, 1H), 2.65 (m, 1H), 2.24 (d, $J=1.6$ Hz, 3H), 1.88–1.94 (complex m, 2H), 1.73 (m, 1H), 1.41–1.47 (complex m, 2H), 1.22 (m, 1H), 1.19 (d, $J=7.2$ Hz, 3H), 1.03 (s, 3H), 0.93 (d, $J=6.5$ Hz, 3H), 0.80–0.98 (complex m, 3H); and ^{13}C NMR (100 MHz, CDCl_3) δ 175.34, 167.27, 166.58, 162.26, 143.73, 142.46, 136.45, 135.88, 134.18, 133.97, 133.73, 132.85, 132.72, 132.45, 131.73, 131.62, 129.99, 129.25, 128.72 (2C), 128.65 (2C), 128.47 (2C), 128.40, 128.27 (2C), 128.23, 118.14, 74.52, 66.58, 65.77, 56.97, 51.87, 45.85, 41.88, 40.88, 40.77, 39.02, 35.56, 33.59, 28.83, 26.76, 22.39, 18.43, 14.16. ESI-MS m/z 736.4221 $[\text{M}+\text{NH}_4]^+$ (Calcd. for $\text{C}_{46}\text{H}_{58}\text{NO}_7$, m/z 736.4213 $[\text{M}+\text{NH}_4]^+$).

Isopropyl ester 12

Burnnettene A isopropyl ester **12** was prepared by added isopropyl iodide (18.6 μL , 186.0 μmol) according to the general method for preparation of alkyl derivatives. The residue was purified by prep. TLC to obtain **12** (6.4 mg, 10.3 μmol , 28.0%). $[\alpha]_{\text{D}}^{23}$ -288.2 (*c* 0.1, CHCl_3); UV (CHCl_3) λ_{max} (ϵ) 241 (24,833), 311 (41,078),

324 (65,786), 339 (87,072), 357 (86,201), 390 (19,481), 412 (16,200). ¹H NMR (400 MHz, CDCl₃) δ 6.88 (s, 1H), 6.09–6.35 (complex m, 8H), 5.66 (dd, *J*=14.7, 7.0 Hz, 1H), 5.41 (s, 1H), 5.36 (d, *J*=15.4 Hz, 1H), 4.97–5.08 (complex m, 2H), 4.25 (t, *J*=13.0 Hz, 1H), 3.67 (s, 3H), 2.94 (s, 1H), 2.52 (m, 1H), 2.19 (s, 3H), 1.86–1.96 (complex m, 2H), 1.72 (m, 1H), 1.40–1.49 (complex m, 2H), 1.22–1.29 (complex m, 13H), 1.15 (d, *J*=7.2 Hz, 3H), 1.01 (s, 3H), 0.94 (d, *J*=6.4 Hz, 3H), 0.79–0.99 (complex m, 3H); and ¹³C NMR (100 MHz, CDCl₃) δ 175.17, 168.07, 167.37, 166.48, 143.60, 142.57, 139.43, 134.55, 133.90, 132.98, 132.51, 131.67, 130.10, 129.18, 128.01, 127.45, 119.15, 74.50, 68.26, 67.89, 67.09, 56.88, 51.88, 45.80, 41.84, 40.93, 40.71, 39.05, 35.61, 33.62, 29.83, 26.79, 22.43, 22.14, 22.08, 18.42, 14.20, 12.91. ESI-MS *m/z* 640.4207 [M+NH₄]⁺ (Calcd. for C₃₈H₅₈NO₇, *m/z* 640.4213 [M+NH₄]⁺).

Isobutyl ester 13

Burnettiene A isobutyl ester **13** was prepared by added 1-iodo-2-methylpropane (21.4 μL, 186.0 μmol) according to the general method for preparation of alkyl derivatives. The residue was purified by prep. TLC to obtain **13** (1.6 mg, 2.5 μmol, 7.0%). [α]_D²³ –306.1 (*c* 0.1, CHCl₃); UV (CHCl₃) λ_{max} (ε) 242 (20,423), 310 (40,261), 324 (73,822), 339 (108,034), 357 (105,238). ¹H NMR (600 MHz, CDCl₃) δ 6.90 (s, 1H), 6.11–6.34 (complex m, 8H), 5.67 (dd, *J*=15.1, 7.1 Hz, 1H), 5.36 (d, *J*=15.4 Hz, 1H), 5.47 (s, 1H), 4.27 (t, *J*=14.2 Hz, 1H), 3.83–3.90 (complex m, 4H), 3.67 (s, 3H), 2.96 (s, 1H), 2.59 (m, 1H), 2.20 (s, 3H), 1.89–1.98 (complex m, 4H), 1.74 (m, 1H), 1.40–1.50 (complex m, 2H), 1.23 (m, 1H), 1.18 (d, *J*=7.1 Hz, 3H), 1.02 (s, 3H), 0.92–0.95 (complex m, 15H), 0.79–1.00 (complex m, 3H); and ¹³C NMR (150 MHz, CDCl₃) δ 168.61, 167.33, 167.02, 160.81, 143.62, 143.19, 139.63, 138.31, 136.57, 135.32, 134.58, 132.53, 132.48, 131.71, 130.07, 129.23, 127.93, 127.05, 118.79, 70.86, 70.22, 56.92, 51.88, 41.91, 41.00, 40.81, 39.07, 35.67, 33.61, 29.84 (2C), 28.01, 27.86, 26.82, 22.45, 19.39(2C), 19.34, 18.43, 12.95. ESI-MS *m/z* 668.4523 [M+NH₄]⁺ (Calcd. for C₄₀H₆₂NO₇, *m/z* 668.4526 [M+NH₄]⁺).

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Author Contributions

A. K., A. N., and I. T. contributed equally. Conceptualization: A. K.; Funding acquisition: A. Y.; Investigation: A. K., K. S., M. H., K. N., T. T., H. K., and K. N.; Project administration: Y. A.; Resources: S. F., T. C. and T. U.; Supervision: Y. A. and T. U.; Writing – original draft: A. K.; Writing – review & editing: all authors.

Electronic supplementary materials

The online version of this article contains supplementary materials, which are available at <https://www.jstage.jst.go.jp/browse/jpestics/>.

Supporting Information

¹H-NMR and ¹³C-NMR spectrum of burnettiene A derivatives **2–13**. Preparation method for burnettiene A (**1**).

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