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# Synthesis of Both Enantiomers of Chiral Phenylalanine Derivatives Catalyzed by Cinchona Alkaloid Quaternary Ammonium Salts as Asymmetric Phase Transfer Catalysts 

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Abstract: A practical synthesis of both enantiomers of unnatural phenylalanine derivatives by using two pseudoenantiomeric phase transfer catalysts is described. Through asymmetric $\alpha$-alkylation of glycine Schiff base with substituted benzyl bromides and 1-(bromomethyl)naphthalene under the catalysis of $O$-allyl- $N$-(9-anthracenmethyl) cinchoninium bromide (1f) and $O$-allyl- $N$-(9-anthracenylmethyl)cinchonidium bromide (1i), respectively, a series of both ( $R$ )- and $(S)$-enantiomers of unnatural $\alpha$-amino acid derivatives were obtained in excellent yields and enantioselectivity. The synthetic method is simple and scalable, and the stereochemistry of the products is fully predictable and controlled: the cinchonine-type phase transfer catalyst $\mathbf{1 f}$ resulted in ( $R$ )- $\alpha$-amino acid derivatives, and the cinchonidine-type phase transfer catalyst $\mathbf{1 i}$ afforded (S)- $\alpha$-amino acid derivatives.

Keywords: unnatural phenylalanine derivatives; phase transfer catalysts; asymmetric $\alpha$-alkylation; glycine Schiff base

## 1. Introduction

Unnatural $\alpha$-amino acids are important building blocks for synthesis of peptides, pharmaceutical molecules and natural products. In particular, unnatural $\alpha$-phenylalanine derivatives have been the subject of numerous investigations for their extensive distribution in biological active compounds. For example, CPD-15A5, which is a small-molecule negative allosteric modulator (antagonist) for the $\beta_{2}$-adrenergic receptor ( $\beta_{2} A R$ ) [1], contains a (S)-3,5-dibromophenylalanine subunit (Figure 1). Levothyroxine, used for the treatment of hypothyroidism [2-4], has a (S)-3,5-diiodophenylalanine backbone. ADEP 4, which shows potent antibacterial activity against multidrug-resistant pathogens [5,6], has a (S)-3,5-difluorophenylalanine sidechain. In addition, LY355703, a potent and broad spectrum antitumor agent [7,8], is partially composed of $(R)$-(3-chloro-5-methoxy)phenylalanine. We are particularly interested in unnatural $\alpha$-phenylalanine derivatives [9] because they are the key building blocks for synthesizing a series of dipeptides as allosteric antagonists of the $\beta_{2}$-adrenergic receptor $\left(\beta_{2} A R\right)$ [1] in one of our ongoing research projects. We need both the $(R)$ - and (S)-enantiomers of $\alpha$-phenylalanine derivatives for structure-activity relationship studies.



ADEP 4


Levothyroxine


Figure 1. Chemical structures of some biologically active molecules containing $\alpha$-phenylalanine subunits.
Although many different methods for the synthesis of $\alpha$-phenylalanine derivatives have been reported in the literature [10-12], these methods have significant drawbacks, such as the use of very costly catalysts, low yields, and/or poor enantioselectivity for some derivatives. Asymmetric phase-transfer catalysis has been widely used for the synthesis of chiral $\alpha$-amino acids because of its operational simplicity, mild reaction conditions, and reduced environmental impact [13-17]. Quaternary cinchona alkaloid catalysts, discovered by O'Donnell et al. [18] and further improved by Lygo [19] and Corey [20], have been the most useful and practical chiral phase-transfer catalysts for the synthesis of $\alpha$-amino acids. On the other hand, only a few examples have been reported for asymmetric synthesis of disubstituted $\alpha$-phenylalanine derivatives by using quaternary cinchona alkaloid catalysts. Furthermore, some of the reported procedures are not suitable for wide range of substrates. For example, McAlister et al. prepared a series of substituted 2-nitrophenylalanine derivatives through asymmetric alkylation of N -(dibenzylidene)glycine tert-butyl ester with substituted 2-nitrobenzyl bromides using a cinchonidine phase transfter catalyst. 5-Methyl-2-nitrobenzyl bromide gave 5-methyl-2-nitrophenylalanine derivative with $100 \%$ ee; however, when the methyl group was replaced with a trifluoromethyl group, the ee value decreased to $90 \%$, and with a chloro group replacement, the corresponding product has only $75 \%$ ee [21].

Our group recently synthesized some biologically important compounds via asymmetric phase transfer catalysis [1,22]. We predicted that both enantiomers of the disubstituted $\alpha$-phenylalanine derivatives could be obtained by using two pseudoenantiomeric quaternary cinchona alkaloids as the phase transfer catalysts. Our objective was to develop a straightforward preparative-scale method for synthesizing the unnatural $\alpha$-phenylalanine derivatives with high chemical and optical purities in sufficient quantities to permit rapid preparation of the dipeptides for laboratory bioassays and animal studies. Herein we report a convenient synthesis of both the (R)- and (S)-enantiomers of $\alpha$-phenylalanine derivatives, including several disubstituted unnatural $\alpha$-phenylalanine derivatives which have not been reported in literature, with excellent yield and excellent enantioselectivity through asymmetric phase-transfer catalysis. The described procedure is simple, mild and scalable, and its usefulness has been demonstrated with the synthesis of a dipeptide derivative of the $\beta_{2} A R$ allosteric antagonist CPD-15A5 by using an enantiomer-enriched 3-chlorophenylalanine derivative.

## 2. Results and Discussion

### 2.1. Condition Screening

Chiral unnatural $\alpha$-phenylalanine derivatives were synthesized through the asymmetric $\alpha$-alkylation reaction of N -(dibenzylidene)glycine tert-butyl ester (2) [23,24] with substituted benzyl bromides catalyzed by a phase transfer catalyst. Compounds $\mathbf{1 a} \mathbf{- 1 h}$ (Figure 2) were chosen as phase transfer catalysts, and they were obtained from cinchonine according to the reported procedures [25-30].


1a


1d


1g


1b

$1 e$


1h


1c


1f

$1 i$

Figure 2. Catalysts 1a-1i.
Initially, we selected 3,5-dichlorophenyl bromide (3a) to react with 2 to optimize the asymmetric alkylation conditions similar to our previously described method [22]. Five equivalents of 3a were reacted with 2 in the presence of 0.1 equivalents of the catalyst and by using $50 \%$ aqueous KOH solution as base and toluene $/ \mathrm{CHCl}_{3}$ as solvent. A similar procedure was reported by Nájera et al. for preparing (S)-tert-butyl $N$-(diphenylmethylene)phenylalaninate [31]. The alkylation proceeded at room temperature for 24 h and afforded ( $R$ )-tert-butyl $N$-(diphenylmethylene)-(3,5-dichloro-phenyl)alaninate (4a). The results are summarized in Table 1. Catalyst $\mathbf{1 h}$ gave the best yield ( $99 \%$ ) and with moderate enantioselectivity ( $81 \%$ ee; Table 1, entry 8) for the desired product, but catalyst 1 f gave the highest enantioselectivity ( $94 \%$ ) (Table 1, entries 1 to 8 ). Clearly, catalyst $1 f$ was better than the other screened catalysts for the asymmetric $\alpha$-alkylation of 2 with 3a. For achieving better enantioselectivity, the reaction temperature was lowered to $-20^{\circ} \mathrm{C}$, and the yield was slightly improved (from $94 \%$ to $97 \%$ ) while the enantioselectivity also increased slightly (from $94 \%$ to $96 \%$ ) when the reaction time was increased to 48 h . When the reaction temperature was lowered to $-40^{\circ} \mathrm{C}$, the ee values improved a little bit (from $96 \%$ to $97 \%$ ), but the yield decreased (from $97 \%$ to $95 \%$ ) even after 72 h (Table 1, entry 10). After the reaction temperature was further lowered to $-60^{\circ} \mathrm{C}$, both the ee value and yield
decreased (Table 1, entry 9). In addition, increasing or decreasing the catalyst's amount didn't improve the enantioselectivity (no better than $96 \%$ ee). However, less amount of catalyst ( $5 \mathrm{~mol} \%$ ) resulted in lower yield ( $89 \%$, Table 1, entry 12) whereas $20 \mathrm{~mol} \%$ of catalyst only increased the yield a little bit (from $95 \%$ to $99 \%$. Table 1, entry 13) with slightly shorter reaction time ( 60 h ). Considering the reaction efficiency and enantioselectivity, the optimized conditions for asymmetric $\alpha$-alkylation of $\mathbf{2}$ with 3a were determined to be: $1 \mathrm{f}(10 \mathrm{~mol} \%)$, toluene $/ \mathrm{CHCl}_{3}$, and $50 \% \mathrm{KOH}$ ( 5 equivalents) at $-40^{\circ} \mathrm{C}$.

Table 1. Optimization of the reaction conditions.


| Entry $^{\boldsymbol{a}}$ | 1 | Temp | Time/h | Yield $^{\boldsymbol{b}}$ | ee $^{\boldsymbol{c}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1a | rt | 24 | $81 \%$ | $88 \%$ |
| 2 | 1b | rt | 24 | $77 \%$ | $81 \%$ |
| 3 | 1c | rt | 24 | $93 \%$ | $77 \%$ |
| 4 | 1d | rt | 24 | $97 \%$ | $82 \%$ |
| 5 | 1e | rt | 24 | $72 \%$ | $54 \%$ |
| 6 | 1f | rt | 24 | $94 \%$ | $94 \%$ |
| 7 | 1g | rt | 24 | $74 \%$ | $78 \%$ |
| 8 | 1h | rt | 24 | $99 \%$ | $81 \%$ |
| 9 | 1f | $-20^{\circ} \mathrm{C}$ | 48 | $97 \%$ | $96 \%$ |
| 10 | 1f | $-40^{\circ} \mathrm{C}$ | 72 | $95 \%$ | $97 \%$ |
| 11 | 1f | $-60^{\circ} \mathrm{C}$ | 72 | $87 \%$ | $92 \%$ |
| $12^{d}$ | 1f | $-40^{\circ} \mathrm{C}$ | 72 | $89 \%$ | $96 \%$ |
| $13^{e}$ | 1f | $-40^{\circ} \mathrm{C}$ | 60 | $99 \%$ | $96 \%$ |

${ }^{a}$ Reactions were performed with $2(0.1 \mathrm{mmol})$, $\mathbf{3 a}(0.5 \mathrm{mmol})$, base $(0.5 \mathrm{mmol})$ and $\mathbf{1}(0.01 \mathrm{mmol})$ in toluene $/ \mathrm{CHCl}_{3}$ ${ }^{b}$ Isolated yield. ${ }^{c}$ Enantiomeric excess was determined by HPLC analysis using a chiral column with $n$-hexane-isopropanol as eluent. ${ }^{d}$ Reactions were performed with $2(0.1 \mathrm{mmol}), 3 \mathrm{a}(0.5 \mathrm{mmol})$, base $(0.5 \mathrm{mmol})$ and $\mathbf{1}(0.005 \mathrm{mmol}) .{ }^{e}$ Reactions were performed with $2(0.1 \mathrm{mmol}), 3 \mathbf{3 a}(0.5 \mathrm{mmol})$, base $(0.5 \mathrm{mmol})$ and $\mathbf{1}(0.02 \mathrm{mmol})$.

### 2.2. Substrate Expansion

With the optimal reaction conditions in hand, we investigated the scope and limitations of the asymmetric $\alpha$-alkylation of glycine Schiff base 2. A variety of disubstituted and monosubstituted benzyl bromides 3a-o as well as 1-(bromomethyl)naphthalene (30) were tested for the alkylation reaction with 2, and the results are outlined in Table 2. A variety of substituents, such as halo ( $\mathrm{F}, \mathrm{Cl}, \mathrm{Br}$ and I ), electron-withdrawing (nitro and difluoro groups), electron-donating (dimethoxy group) and $\alpha$-naphthyl groups, were well tolerated under the alkylation conditions, affording the desired products $\mathbf{4 a - 4 n}$. Eight disubstituted unnatural $\alpha$-phenylalanine derivatives $\mathbf{4 a} \mathbf{- 4 h}$ (Table 2, entries 1-8) were obtained with satisfactory yields and enantioselectivity. When the benzyl bromides containing strong electron-withdrawing groups were used, the corresponding $\alpha$-phenylalanine derivatives were prepared with excellent enantioselectivity. Under the same alkylation conditions, 1-(bromomethyl)naphthalene was reacted smoothly with 2, affording (R)-tert-butyl 2-((diphenylmethylene)amino)-3-(1-naphthyl)propanoate (4o) with 85\% yield and 97\% ee (Table 2, entry 15).

Table 2. Asymmetric alkylation of 2 with 3a-o under the catalysis of $\mathbf{1 f}$.

${ }^{a}$ Reactions were performed with $2(0.1 \mathrm{mmol})$, alkylation reagent $(0.5 \mathrm{mmol}), 50 \% \mathrm{KOH}(0.5 \mathrm{mmol})$ and 1f
$(0.01 \mathrm{mmol})$ in toluene $/ \mathrm{CHCl}_{3}$ at $-40{ }^{\circ} \mathrm{C} .{ }^{b}$ Isolated yield. ${ }^{c}$ Enantiomeric excess was determined by HPLC analysis using a chiral column with $n$-hexane-isopropanol as eluent.

After derivatives 4a-o were successfully obtained under the catalysis of cinchonine-type phase transfer catalyst 1f, we tried to synthesize the enantiomers of 4a-o by using a cinchonidine-type phase transfer catalyst, O -allyl- N -(9-anthracenemethyl) cinchonidium bromide (1i, Table 3), which is the pseudoenantiomer of $\mathbf{1 f}$ and was prepared according to the same procedure used for $\mathbf{1 f}$. To our satisfaction, all the enantiomers of $\mathbf{4 a - 4 o}$ were obtained with good to excellent yields ( $71 \%$ to $99 \%$ ) and excellent enantioselectivity ( $93 \%$ to $99 \%$ ee) (Table 3) under the identical conditions for 4a-4o except that catalyst $\mathbf{1 f}$ was replaced with 1i. Similar to the results in Table 2, alkylation of $\mathbf{2}$ with 2-chloro-6-fluorobenzyl bromide resulted in the highest enantioselectivity under the catalyst of $\mathbf{1 h}$ (99\% ee; Table 3, entry 5).

Table 3. Asymmetric alkylation of 2 with 3a-o under the catalysis of $\mathbf{1 i}$.


[^0]Finally, the absolute configuration of the newly synthesized $\alpha$-amino acid derivatives $4 \mathbf{a}-\mathbf{4 o}$ and $\mathbf{4 a ^ { \prime }}-\mathbf{4 \mathbf { o } ^ { \prime }}$ was established by comparison of their optical rotation values with those reported in the literature. For example, the (S)-configuration of $41^{\prime}$ was confirmed by comparing its optical rotation value $\left([\alpha]_{D}^{20}-105.4^{\circ}, c=1.09, \mathrm{CHCl}_{3}\right)$ with the reported result $\left([\alpha]_{D}^{20}-110.1^{\circ}, c=1.09, \mathrm{CHCl}_{3}\right)[32]$. Gratzer et al. synthesized the same compound through asymmetric $\alpha$-alkylation of glycine Schiff base catalyzed by $p$-biphenyl-containing pyrrolidinium ammonium bromide in $79 \%$ yield and $80 \%$ ee [32]. The $(R)$-configuration of $\mathbf{4 o}$ was established by comparison of its optical rotation value $\left([\alpha]_{D}^{20}\right.$ $331.6^{\circ}, c=1, \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) with the reported value $\left([\alpha]_{D}^{20} 343.7^{\circ}, c=1.19, \mathrm{CHCl}_{3}\right)$ [33]. After synthesizing $4 \mathbf{o}$ via asymmetric $\alpha$-alkylation of glycine Schiff base catalyzed by Maruoka catalyst [15], Ooi et al. cleaved the benzophenone imine and tert-butyl ester with 6 N HCl , protected the amino group with Boc, and then confirmed the ( $R$ )-configuration by comparing the HPLC retention time of the $N$-Boc protected amino acid with the literature value [33]. Therefore, $(R)$-configuration was assigned for $\mathbf{4 a} \mathbf{- 4 0}$, and (S)-configuration for $\mathbf{4 a}^{\mathbf{\prime}} \mathbf{- 4 \mathbf { o } ^ { \prime }}$.

### 2.3. Application

The asymmetric $\alpha$-alkylation of glycine Schiff base with substituted benzyl bromides can be applied to the synthesis of new derivatives of CPD-15A5 as allosteric antagonists for the $\beta_{2} A R$, such as (2S)-3-(3-chlorophenyl)-2-((2S)-2-(2-cyclohexyl-2-phenylacetamido)-3-phenylpropanamido)N -methylpropanamide (13, Scheme 1). (S)-tert-Butyl N -(diphenylmethylene)-(3-chlorophenyl)alaninate ( $4 \mathbf{j}^{\prime}$ ) was hydrolyzed in refluxing hydrochloric acid to give (S)-3-chlorophenylalanine hydrochloride (5) in $92 \%$ yield, and then the amino group in 5 was protected with $\mathrm{Fmoc}-\mathrm{Cl}$, affording Fmoc-protected (S)-(3-chloro)phenylalanine (6). In the next step, the Fmoc-protected L-phenylalanine methylamide 7 was obtained by condensation of 6 with methylamine in the presence of O-benzo-triazol-1-yl- $N, N, N N^{\prime}, N^{\prime}$-tetramethyluronium hexafluorophosphate (HBTU) and hydroxybenzotriazole ( HOBt ). It should be pointed out that the acidic hydrolysis of $4 \mathbf{j}^{\prime}$ didn't racemize the amino acid, because the ee value of 7 is $96 \%$. After Fmoc deprotection of 7 (piperidine/DMF), the resulting L-phenylalanine methylamide 8 was coupled in the presence of $\mathrm{HBTU} / \mathrm{HOBt}$ with Fmoc-L-4-carbamoylphenylanine (9) to generate dipeptide 10 in 51\% yield. Upon treatment with piperidine in DMF, the Fmoc group in 10 was removed smoothly at room temperature, giving the corresponding amine $\mathbf{1 1}$ in $95 \%$ yield. In the final step, $\mathbf{1 1}$ was reacted with 2-cyclohexyl-2-phenyl acetic acid (12) to afford the desired product 13 in $63 \%$ yield.





Scheme 1. Synthetic route to 13.

## 3. Materials and Methods

### 3.1. Instruments and Reagents

Melting points were measured on SGW X-4B melting point apparatus (Shenguang, Shanghai, China). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectra were recorded on Avance $300(300 \mathrm{MHz})$ and $400(400 \mathrm{MHz})$ spectrometers (Bruker, Karlsruhe, Germany). Chemical shifts were reported in ppm from tetramethylsilane with the solvent resonance resulting from incomplete deuterium incorporation as the internal standard $\left(\mathrm{CDCl}_{3}: \delta 7.26 \mathrm{ppm}\right) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra were recorded on Bruker Avance $300(75 \mathrm{MHz})$ and $400(100 \mathrm{MHz})$ spectrometers with complete proton decoupling. Chemical shifts were reported in ppm from tetramethylsilane with the solvent resonance as the internal standard. High-resolution mass spectrometry was performed on a Thermo Orbitrap Elite, instrument (Agilent, Palo Alto, CA, USA). Optical rotations were measured on an Autopol IV ( $\mathrm{d}=589 \mathrm{~nm}, \mathrm{Hg}$ lamp, 50 mm cell) instrument (Rudolph, NJ, USA). The enantiomeric excess was determined by a 1260 infinity series HPLC (Agilent, Palo Alto, CA, USA) equipped with Chiralpak OD-H, AD-H and IA columns ( $4.6 \mathrm{~mm} \times 250 \mathrm{~mm}$, Daicel Chiral Technologies, Shanghai, China). Chemicals and solvents were purchased from Linfeng (Shanghai) and Annaiji (Shanghai) in China, and used as received. Purification of the products was carried out by flash column chromatography using silica gel (Yantai Jiangyou Company, Shandong, China, particle size $0.100-0.075 \mathrm{~mm}$ ).

### 3.2. General Methods

(R)-tert-Butyl N-(diphenylmethylene)-(3,5-dichlorophenyl)alaninate (4a; $\mathrm{R}=3,5-\mathrm{Cl}_{2}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): A 10 mL reaction tube was charged with $2(30 \mathrm{mg}, 0.1 \mathrm{mmol}), 3,5$-dichlorobenzyl bromide ( $119 \mathrm{mg}, 0.5 \mathrm{mmol}$,

5 equivalent), catalyst $1 f\left(6 \mathrm{mg}, 0.01 \mathrm{mmol}, 0.1\right.$ equivalent) and toluene and $\mathrm{CHCl}_{3}(1.5 \mathrm{~mL}, 2: 1 \mathrm{v} / \mathrm{v})$, and the mixture was cooled to $-40^{\circ} \mathrm{C}$. After the mixture was stirred for $10 \mathrm{~min}, 50 \% \mathrm{aq}$. $\mathrm{KOH}(28$ $\mu \mathrm{L}, 0.1 \mathrm{mmol}, 5$ equivalent) was added, and the whole reaction mixture was stirred at $-40^{\circ} \mathrm{C}$ for 72 h before being allowed to warm to ambient temperature. The reaction was quenched by adding $\mathrm{H}_{2} \mathrm{O}(2 \mathrm{~mL})$, and the resulting mixture was extracted with EtOAc $(3 \times 10 \mathrm{~mL})$. The combined extracts were washed with brine ( 10 mL ) and dried (anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ), and the crude product was purified by flash column chromatography (eluting with hexane/EtOAc, 50:1) to afford $4 \mathbf{4 a}$ ( $43 \mathrm{mg}, 95 \%$ yield) as light yellow liquid. $97 \%$ ee; $[\alpha]_{D}^{20} 178.8^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.57$ (s, $1 \mathrm{H}), 7.55(\mathrm{~d}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.40-7.29(\mathrm{~m}, 6 \mathrm{H}), 7.16(\mathrm{t}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.76$ $(\mathrm{d}, J=6.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.12(\mathrm{dd}, J=8.9,4.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.19-3.08(\mathrm{~m}, 2 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta 171.1,170.2,141.7,139.2,136.1,134.4,130.3,128.7,128.6,128.3,128.3,128.0,127.6,126.4,81.6$, 66.9, 38.9, 28.0; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{Cl}_{2} \mathrm{NO}_{2}[\mathrm{M}+\mathrm{H}]^{+} 454.1335$, found: 454.1333. HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane $/$ isopropanol $=95: 5$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N -(diphenylmethylene)-(3,5-dichlorophenyl)alaninate ( $4 \mathbf{a}^{\prime} ; \mathrm{R}=3,5-\mathrm{Cl}_{2}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $\mathbf{4 a}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4 \mathbf { a } ^ { \prime }}$ was obtained as light yellow liquid. $98 \%$ yield; $97 \%$ ee; $[\alpha]_{D}^{29}-183.0^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
(R)-tert-Butyl N-(diphenylmethylene)-(3,5-difluorophenyl)alaninate ( $4 \mathbf{b} ; \mathrm{R}=3,5-\mathrm{F}_{2}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $\mathbf{4 a}$ except that 3,5-dichlorophenyl bromide was replaced with 3,5-difluorobenzyl bromide, $\mathbf{4 b}$ was obtained as white solid. M.p. $35-37{ }^{\circ} \mathrm{C}$; $98 \%$ yield; $94 \%$ ee; $[\alpha]_{D}^{29} 150.2^{\circ}(c=1.0$, $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.58(\mathrm{~s}, 1 \mathrm{H}), 7.56(\mathrm{~d}, J=1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.39-7.25(\mathrm{~m}, 6 \mathrm{H}), 6.78$ $(\mathrm{d}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.63-6.59(\mathrm{~m}, 3 \mathrm{H}), 4.13(\mathrm{dd}, J=8.9,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.23-3.10(\mathrm{~m}, 2 \mathrm{H}), 1.44(\mathrm{~s}, 9 \mathrm{H})$; ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 170.9,170.2,163.9,163.8,161.5,161.3,142.4,142.3,142.2,139.2,136.1$, $130.3,128.7,128.5,128.2,128.0,127.6,112.6,112.6,112.4,112.3,101.9,101.6,101.4,81.5,67.1,39.3,28.0$; HRMS (ESI, positive): calcd. for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{~F}_{2} \mathrm{NO}_{2}[\mathrm{M}+\mathrm{H}]^{+} 422.1926$, found: 422.1925. HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane $/$ isopropanol $=98: 2$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3,5-difluorophenyl)alaninate ( $4 \mathbf{b}^{\prime} ; \mathrm{R}=3,5-\mathrm{F}_{2}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $\mathbf{4 b}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4} \mathbf{b}^{\prime}$ was obtained as white solid. $99 \%$ yield; $94 \%$ ee; $[\alpha]_{D}^{29}-154.6^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
(R)-tert-Butyl $N$-(diphenylmethylene)-(3,5-dibromophenyl)alaninate ( $4 \mathrm{c} ; \mathrm{R}=3,5-\mathrm{Br}_{2}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $\mathbf{4 a}$ except that 3,5-dichlorophenyl bromide was replaced with 3,5-dibromobenzyl bromide, 4 c was obtained as colorless liquid. $95 \%$ yield; $93 \%$ ee; $[\alpha]_{D}^{29} 174.6^{\circ}$ (c = 1.0, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.56(\mathrm{~s}, 1 \mathrm{H}), 7.54(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{~s}, 1 \mathrm{H}), 7.39-7.25(\mathrm{~m}, 6 \mathrm{H})$, $7.15(\mathrm{~d}, J=1.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.76(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.10(\mathrm{dd}, J=8.8,4.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.18-3.07(\mathrm{~m}, 1 \mathrm{H}), 1.46$ (s, 1H); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ): $\delta 171.1,170.1,142.3,139.2,136.1,131.8,131.6,130.3,128.7$, 128.6, 128.3, 128.0, 127.6, 122.4, 81.6, 66.9, 38.8, 28.0; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{Br}_{2} \mathrm{NO}_{2}$ $[\mathrm{M}+\mathrm{H}]^{+} 542.0325,544.0304$, found: 542.0326, 544.0306. HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane/isopropanol $=95: 5$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3,5-dibromophenyl)alaninate ( $4 \mathrm{c}^{\prime} ; \mathrm{R}=3,5-\mathrm{Br}_{2}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $\mathbf{4 c}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4} \mathbf{c}^{\prime}$ was obtained as colorless liquid. $83 \%$ yield, $95 \%$ ee; $[\alpha]_{D}^{29}-86^{\circ}\left(c=1.3, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
(R)-tert-Butyl N-(diphenylmethylene)-(3-chloro-5-fluorophenyl)alaninate ( $4 \mathrm{~d} ; \mathrm{R}=3-\mathrm{Cl}-5-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $4 \mathbf{a}$ except that 3,5 -dichlorophenyl bromide was replaced with 3-chloro-5-fluorophenyl bromide, 4 d was obtained as colorless liquid. $93 \%$ yield; $97 \%$ ee; $[\alpha]_{D}^{20}$ $169.0^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.57(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.41-7.28(\mathrm{~m}, 6 \mathrm{H}), 6.89$ $(\mathrm{d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{~s}, 1 \mathrm{H}), 6.78(\mathrm{~d}, J=5.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.71(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.13(\mathrm{q}, J=4.4 \mathrm{~Hz}, 1 \mathrm{H})$, 3.31-3.10 (m, 2H), 1.45 ( $\mathrm{s}, 9 \mathrm{H}$ ); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 171.1,170.3,163.7,161.2,142.3,142.2$, $139.3,136.3,134.5,134.4,130.5,128.8,128.7,128.4,128.1,127.7,125.9,125.9,115.4,115.2,114.2,114.0,81.7$,
67.1,39.2, 28.1; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{25} \mathrm{ClFNaNO}_{2}[\mathrm{M}+\mathrm{Na}]^{+} 460.1450$, found: 460.1450 . HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane $/$ isopropanol $=95: 5$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3-chloro-5-fluorophenyl)alaninate ( $4 \mathrm{~d}^{\prime} ; \mathrm{R}=3-\mathrm{Cl}-5-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $\mathbf{4 d}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4 \mathbf { d } ^ { \prime }}$ was obtained as colorless liquid. $97 \%$ yield, $98 \%$ ee; $[\alpha]_{D}^{20}-168.2^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
(R)-tert-Butyl N-(diphenylmethylene)-(2-chloro-6-fluorophenyl)alaninate ( $4 \mathbf{e} ; \mathrm{R}=2-\mathrm{Cl}-6-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $4 \mathbf{a}$ except that 3,5-dichlorophenyl bromide was replaced with 2-chloro-6-fluorophenyl bromide, $4 \mathbf{e}$ was obtained as colorless liquid. $68 \%$ yield; $98 \%$ ee; $[\alpha]_{D}^{20} 274.2^{\circ}$ $\left(c=0.9, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.60-7.57(\mathrm{~m}, 2 \mathrm{H}), 7.36-7.26(\mathrm{~m}, 6 \mathrm{H}), 7.11-7.01(\mathrm{~m}$, $2 \mathrm{H}), 6.88-6.81(\mathrm{~m}, 1 \mathrm{H}), 6.68(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 4.39-4.34(\mathrm{~m}, 1 \mathrm{H}), 3.52-3.45(\mathrm{~m}, 1 \mathrm{H}), 3.36-3.29(\mathrm{~m}, 1 \mathrm{H})$, 1.45 (s, 9H); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 170.8,170.6,163.6,160.3,139.6,136.2,136.1,136.0,130.2$, 129.0, 128.5, 128.2, 128.1, 128.0, 127.8, 125.1, 125.0, 124.7, 124.5, 114.0, 113.7, 81.4, 64.6, 30.1, 28.1; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{25} \mathrm{ClFNO}_{2} \mathrm{Na}\left[\mathrm{M}+\mathrm{Na}{ }^{+} 460.1450\right.$, found: 460.1447. HPLC analysis: Daicel Chiralcel IC, $n$-hexane $/$ isopropanol $=98: 2$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(2-chloro-6-fluorophenyl)alaninate ( $4 \mathbf{e}^{\prime} ; \mathrm{R}=2-\mathrm{Cl}-6-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $\mathbf{4 e}$ except that catalyst $1 f$ was replaced with $\mathbf{1 i}$, enantiomer $4 \mathbf{e}^{\prime}$ was obtained as colorless liquid. $74 \%$ yield, $99 \%$ ee; $[\alpha]_{D}^{20}-231.0^{\circ}\left(c=0.8, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
(R)-tert-Butyl N -(diphenylmethylene)-(3-chloro-4-fluorophenyl)alaninate ( $4 \mathbf{f} ; \mathrm{R}=3-\mathrm{Cl}-4-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $4 \mathbf{a}$ except that 3,5 -dichlorophenyl bromide was replaced with 3-chloro-4-fluorophenyl bromide, $4 \mathbf{f}$ was obtained as colorless liquid. $86 \%$ yield; $97 \%$ ee; $[\alpha]_{D}^{20} 178.5^{\circ}$ (c $\left.=1.1, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.59-7.55(\mathrm{~m}, 2 \mathrm{H}), 7.41-7.27(\mathrm{~m}, 6 \mathrm{H}), 7.07(\mathrm{~d}, J=7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.97(\mathrm{~s}, 1 \mathrm{H}), 6.94(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.10(\mathrm{q}, J=4.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.20-3.07$ $(\mathrm{m}, 2 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 170.9,170.5,158.5,155.2,139.4,136.3,135.5,131.8$, 130.5, 129.7, 128.8, 128.6, 128.4, 128.1, 127.7, 120.5, 120.3, 116.2, 116.0, 81.6, 67.5, 38.6, 28.2; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{25} \mathrm{ClFNO}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 460.1450$, found: 460.1445 . HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane $/$ isopropanol $=95: 5$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3-chloro-4-fluorophenyl)alaninate (4f'; $\mathrm{R}=3-\mathrm{Cl}-4-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $4 f$ except that catalyst $1 f$ was replaced with $1 \mathbf{i}$, enantiomer $\mathbf{4 f}^{\prime}$ was obtained as colorless liquid. $88 \%$ yield, $97 \%$ ee; $[\alpha]_{D}^{20}-170.0^{\circ}\left(c=1.2, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
(R)-tert-Butyl N-(diphenylmethylene)-(3-bromo-5-fluorophenyl)alaninate ( $\mathbf{4} \mathbf{g} ; \mathrm{R}=3-\mathrm{Br}-5-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $4 \mathbf{a}$ except that 3,5 -dichlorophenyl bromide was replaced with 3-bromo-5-fluorophenyl bromide, 4 g was obtained as colorless liquid. $83 \%$ yield; $96 \%$ ee; $[\alpha]_{D}^{20}$ $151.7^{\circ}\left(c=1.3, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.57(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.41-7.29(\mathrm{~m}, 6 \mathrm{H}), 7.05$ $(\mathrm{d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.00(\mathrm{~s}, 1 \mathrm{H}), 6.75(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 3 \mathrm{H}), 4.12(\mathrm{q}, J=4.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.21-3.09(\mathrm{~m}, 2 \mathrm{H}), 1.45$ (s, 9H); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 171.1,170.3,163.4,161.2,142.7,142.6,139.3,136.3,130.5,130.2$, $128.9,128.8,128.8,128.7,128.4,128.1,127.7,122.1,122.0,117.1,116.8,116.0,115.8,81.7,67.1,39.1,28.1$; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{BrFNO}_{2}[\mathrm{M}+\mathrm{H}]^{+} 482.1125$, found: 482.1119. HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane $/$ isopropanol $=95: 5$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3-bromo-5-fluorophenyl)alaninate ( $4 \mathrm{~g}^{\prime} ; \mathrm{R}=3-\mathrm{Br}-5-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $\mathbf{4 g}$ except that catalyst $1 f$ was replaced with $\mathbf{1 i}$, enantiomer $4 \mathbf{g}^{\prime}$ was obtained as colorless liquid. $86 \%$ yield, $98 \%$ ee; $[\alpha]_{D}^{20}-154.4^{\circ}\left(c=1.3, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
(R)-tert-Butyl N-(diphenylmethylene)-(3,5-dimethoxyphenyl)alaninate ( $4 \mathbf{h} ; \mathrm{R}=3,5-(\mathrm{MeO})_{2}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $4 \mathbf{a}$ except that 3,5 -dichlorophenyl bromide was replaced with 3,5 -dimethoxybenzyl bromide, 4 h was obtained as colorless liquid. $85 \%$ yield; $96 \%$ ee; $[\alpha]_{D}^{29} 160.2^{\circ}$ $\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.60(\mathrm{~s}, 1 \mathrm{H}), 7.58(\mathrm{~d}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.38-7.26(\mathrm{~m}, 6 \mathrm{H})$, $6.65(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.27(\mathrm{q}, ~ J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.21(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.12(\mathrm{dd}, J=9.3,4.2 \mathrm{~Hz}, 1 \mathrm{H})$, $3.63(\mathrm{~s}, 6 \mathrm{H}), 3.19-3.08(\mathrm{~m}, 2 \mathrm{H}), 1.46(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 170.8,170.2,160.4,140.5$,
139.5, 136.3, 130.1, 128.7, 128.2, 127.9, 127.9, 127.7, 107.4, 99.1, 81.3, 67.7, 55.1, 39.8, 28.0; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{28} \mathrm{H}_{32} \mathrm{NO}_{4}[\mathrm{M}+\mathrm{H}]^{+} 446.2326$, found: 446.2327. HPLC analysis: Daicel Chiralcel IA, $n$-hexane $/$ isopropanol $=95: 5$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3,5-dimethoxyphenyl)alaninate ( $4 \mathbf{h}^{\prime} ; \mathrm{R}=3,5-(\mathrm{MeO})_{2}-\mathrm{C}_{6} \mathrm{H}_{3}$ ): Under the same reaction conditions for $\mathbf{4 h}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4} \mathbf{h}^{\mathbf{\prime}}$ was obtained as colorless liquid. $98 \%$ yield, $98 \%$ ee; $[\alpha]_{D}^{29}-211.4^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
(R)-tert-Butyl N-(diphenylmethylene)-(3-fluorophenyl)alaninate ( $4 \mathbf{i} ; \mathrm{R}=3-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4 a}$ except that 3,5-dichlorophenyl bromide was replaced with 3-fluorobenzyl bromide, $4 \mathbf{i}$ was obtained as colorless liquid. $77 \%$ yield; $95 \%$ ee (Lit $90 \%$ ee [34]); $[\alpha]_{D}^{29} 169.2^{\circ}\left(c=1.1, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ ); ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.57(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.38-7.24(\mathrm{~m}, 6 \mathrm{H}), 7.14(\mathrm{dd}, J=14.1,7.8 \mathrm{~Hz}, 1 \mathrm{H})$, $6.86-6.83(\mathrm{~m}, 2 \mathrm{H}), 6.75(\mathrm{~d}, J=9.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.12(\mathrm{dd}, J=9.0,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.25-3.13$ $(\mathrm{m}, 2 \mathrm{H}), 1.44(\mathrm{~s}, 9 \mathrm{H}){ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 170.6,170.5,163.8,161.4,140.9,140.8,139.3,136.2$, $130.2,129.4,129.4,128.7,128.3,128.1,127.9,127.6,125.5,125.5,116.6,116.4,113.1,112.9,81.3,67.5,39.2$, 28.0. HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{FNNaO}_{2}[\mathrm{M}+\mathrm{Na}]^{+} 426.1840$, found: 426.1846. HPLC analysis: Daicel Chiralcel AD-H, $n$-hexane $/$ isopropanol $=97: 3$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3-fluorophenyl)alaninate ( $4 \mathbf{i}^{\prime} ; \mathrm{R}=3-\mathrm{F}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4 i}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4 i} \mathbf{i}^{\prime}$ was obtained as colorless liquid. $82 \%$ yield; $95 \%$ ee (Lit $96 \%$ ee $[35]$ ); $[\alpha]_{D}^{29}-156.7^{\circ}\left(c=0.9, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ ).
(R)-tert-Butyl N-(diphenylmethylene)-(3-chlorophenyl)alaninate ( $4 \mathbf{j} ; \mathrm{R}=3-\mathrm{Cl}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4 a}$ except that 3,5-dichlorophenyl bromide was replaced with 3-chlorobenzyl bromide, $4 \mathbf{j}$ was obtained as light yellow liquid. $76 \%$ yield; $96 \%$ ee (Lit $95 \%$ ee [30]); $[\alpha]_{D}^{29} 227.4^{\circ}$ (c =1.0, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.56(\mathrm{~d}, \mathrm{~J}=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.38-7.25(\mathrm{~m}, 6 \mathrm{H}), 7.15-7.09(\mathrm{~m}, 2 \mathrm{H})$, $7.02(\mathrm{~s}, 1 \mathrm{H}), 6.97(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.67(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.11(\mathrm{dd}, J=9.0,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.22-3.11$ $(\mathrm{m}, 2 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 170.7,170.5,140.4,139.3,136.2,133.8,130.2$, 129.8, 129.3, 128.7, 128.4, 128.1, 128.1, 128.0, 127.6, 126.3, 81.3, 67.4, 39.1, 28.0; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{27} \mathrm{ClNO}_{2}[\mathrm{M}+\mathrm{H}]^{+} 420.1725$, found: 420.1724. HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane $/$ isopropanol $=95: 5$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3-chlorophenyl)alaninate ( $4 \mathrm{j}^{\prime} ; \mathrm{R}=3-\mathrm{Cl}^{2}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4} \mathbf{j}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4 j} \mathbf{j}^{\prime}(1.6 \mathrm{~g}, 92 \%$ yield) was obtained as light yellow oil, and used for synthesis of 13. 97\% ee (Lit 91\% ee [36], Lit 92\% ee $[37,38]) ;[\alpha]_{D}^{29}-223.4^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;\left[\operatorname{Lit}[\alpha]_{D}^{20}-16.3^{\circ}\left(c=0.2, \mathrm{CHCl}_{3}\right)[30]\right]$.
(R)-tert-Butyl N-(diphenylmethylene)-(3-bromophenyl)alaninate ( $4 \mathbf{k}$; $\mathrm{R}=3-\mathrm{Br}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4 a}$ except that 3,5-dichlorophenyl bromide was replaced with 3-bromobenzyl bromide, $4 \mathbf{k}$ was obtained as colorless liquid. $98 \%$ yield, $95 \%$ ee (Lit $92 \%$ ee [30]); $[\alpha]_{D}^{29} 185.4^{\circ}(c=1.0$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.56(\mathrm{~d}, \mathrm{~J}=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.36-7.30(\mathrm{~m}, 7 \mathrm{H}), 7.18(\mathrm{~s}, 1 \mathrm{H}), 7.07-7.03$ $(\mathrm{m}, 2 \mathrm{H}), 6.67(\mathrm{~s}, 2 \mathrm{H}), 4.12-4.09(\mathrm{~m}, 1 \mathrm{H}), 3.21-3.10(\mathrm{~m}, 2 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$ $170.7,170.4,140.7,139.3,136.2,132.7,130.2,129.6,129.2,128.7,128.6,128.4,128.2,128.0,127.5,122.1,81.4$, 67.4, 39.1, 18.0; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{BrNaNO}_{2}[\mathrm{M}+\mathrm{Na}]^{+} 486.1039$, found: 486.1040 . HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane $/$ isopropanol $=98: 2$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3-bromophenyl)alaninate ( $4 \mathbf{k}^{\prime} ; \mathrm{R}=3-\mathrm{Br}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4 k}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4} \mathbf{k}^{\prime}$ was obtained as colorless liquid. $93 \%$ yield, $93 \%$ ee; $[\alpha]_{D}^{29}-188.2^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
(R)-tert-Butyl N-(diphenylmethylene)-(4-bromophenyl)alaninate ( $41 ; \mathrm{R}=4-\mathrm{Br}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4 a}$ except that 3,5-dichlorophenyl bromide was replaced with 4-bromobenzyl bromide, 41 was obtained as colorless liquid. $95 \%$ yield; $95 \%$ ee; $[\alpha]_{D}^{29} 127.2^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta 7.58(\mathrm{~s}, 1 \mathrm{H}), 7.56(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.40-7.29(\mathrm{~m}, 8 \mathrm{H}), 6.93(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.67(\mathrm{~d}, J=6.4$ $\mathrm{Hz}, 2 \mathrm{H}), 4.09(\mathrm{dd}, J=9.0,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.19-3.08(\mathrm{~m}, 2 \mathrm{H}), 1.44(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}(300 \mathrm{MHz} \mathrm{CDCl} 3): \delta$
170.6, 170.5, 139.3, 137.4, 136.2, 131.6, 130.2, 128.7, 128.3, 128.1, 128.0, 127.6, 120.0, 81.3, 67.5, 38.9, 28.0; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{27} \mathrm{BrNO}_{2}[\mathrm{M}+\mathrm{H}]^{+} 464.1220$, found: 464.1220 . HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane $/$ isopropanol $=98: 2$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(4-bromophenyl)alaninate (41'; $\mathrm{R}=4-\mathrm{Br}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for 41 except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $4 \mathbf{1}^{\prime}$ was obtained as colorless liquid. $95 \%$ yield; $95 \%$ ee (Lit $80 \%$ ee [32]); $[\alpha]_{D}^{29}-134.4^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$; $\left[\mathrm{Lit}[\alpha]_{D}^{27}-110.1^{\circ}\right.$ ( $c=1.09, \mathrm{CHCl}_{3}$ ) [32]].
(R)-tert-Butyl N-(diphenylmethylene)-(3-iodophenyl)alaninate $\left(4 \mathbf{m} ; \mathrm{R}=3-\mathrm{I}-\mathrm{C}_{6} \mathrm{H}_{4}\right)$ : Under the same reaction conditions for $\mathbf{4 a}$ except that 3,5-dichlorophenyl bromide was replaced with 3-iodobenzyl bromide, $\mathbf{4 m}$ was obtained as colorless liquid. $82 \%$ yield, $96 \%$ ee; $[\alpha]_{D}^{29} 235.6^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta 7.56(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.49(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.38-7.25(\mathrm{~m}, 7 \mathrm{H}), 7.05(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H})$, $6.92(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.65(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.09(\mathrm{dd}, J=4.8,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.18-3.07(\mathrm{~m}, 2 \mathrm{H}), 1.45(\mathrm{~s}$, 9H); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 170.7,170.4,140.8,139.3,138.6,136.2,135.2,130.2,129.8,129.2$, 128.7, 128.3, 128.2, 127.9, 127.6, 94.1, 81.4, 67.4, 39.0, 28.0; HRMS: Calcd. for $\mathrm{C}_{26} \mathrm{H}_{27} \mathrm{INO}_{2}{ }^{+}[\mathrm{M}+\mathrm{H}]^{+}$ 512.1081, found: 512.1083. HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane/isopropanol $=95: 5$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N-(diphenylmethylene)-(3-iodophenyl)alaninate ( $4 \mathbf{m}^{\prime} ; \mathrm{R}=3-\mathrm{I}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4 m}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4 \mathbf { m } ^ { \prime }}$ was obtained as colorless liquid. $86 \%$ yield; $94 \%$ ee (Lit 95\% ee [39]); $[\alpha]_{D}^{29}-191.4^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ ).
(R)-tert-Butyl $N$-(diphenylmethylene)-(4-nitrophenyl)alaninate ( $4 \mathbf{n} ; \mathrm{R}=4-\mathrm{NO}_{2}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4 a}$ except that 3,5-dichlorophenyl bromide was replaced with 4-nitrobenzyl bromide, 4 n was obtained as light yellow solid. M.p. $129-131^{\circ} \mathrm{C} ; 99 \%$ yield; $95 \%$ ee (Lit $90 \%$ ee [30]); $[\alpha]_{D}^{29} 184.8^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 8.06(\mathrm{dd}, J=7.0,1.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.58(\mathrm{~s}, 1 \mathrm{H})$, $7.56(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.41-7.36(\mathrm{~m}, 2 \mathrm{H}), 7.34-7.29(\mathrm{~m}, 4 \mathrm{H}), 7.26(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{~s}, 1 \mathrm{H}), 6.71$ $(\mathrm{d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.18(\mathrm{dd}, J=8.1,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.35-3.25(\mathrm{~m}, 2 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): ~ \delta 170.9,170.0,146.5,146.4,139.0,135.9,130.6,130.4,128.7,128.6,128.3,128.0,127.4,123.2,81.6$, 66.9, 39.3, 28.0; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{26} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{4}[\mathrm{M}+\mathrm{H}]^{+} 431.1965$, found: 431.1962 . HPLC analysis: Daicel Chiralcel IA, $n$-hexane/isopropanol $=94: 6$, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl N -(diphenylmethylene)-(4-nitrophenyl)alaninate ( $4 \mathbf{n}$; $\mathrm{R}=4-\mathrm{NO}_{2}-\mathrm{C}_{6} \mathrm{H}_{4}$ ): Under the same reaction conditions for $\mathbf{4} \mathbf{n}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4} \mathbf{n}^{\prime}$ was obtained as light yellow solid. $96 \%$ yield; $94 \%$ ee (Lit 99\% ee [39]); $[\alpha]_{D}^{29}-165.4^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ ).
(R)-tert-Butyl 2-((diphenylmethylene)amino)-3-(1-naphthyl)propanoate (40; $\mathrm{R}=\alpha$-naphthyl): Under the same reaction conditions for $\mathbf{4 a}$ except that 3,5-dichlorophenyl bromide was replaced with 1-(bromomethyl)naphthalene, 40 was obtained as colorless liquid. $81 \%$ yield; $96 \%$ ee (Lit $96 \%$ ee [40], Lit 99\% ee [33]); [ $\alpha]_{D}^{29} 331.6^{\circ}\left(c=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ ); [Lit $\left.[\alpha]_{D}^{27} 343.7^{\circ}\left(c=1.19, \mathrm{CHCl}_{3}\right)[33]\right] ;{ }^{1} \mathrm{H}-\mathrm{NMR}(400$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.78(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.72-7.67(\mathrm{~m}, 2 \mathrm{H}), 7.51(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.38(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H})$, $7.33-7.22(\mathrm{~m}, 6 \mathrm{H}), 7.11(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.24(\mathrm{~s}, 2 \mathrm{H}), 4.32(\mathrm{dd}, J=9.5,4.0 \mathrm{~Hz}$, $1 \mathrm{H}), 3.80(\mathrm{dd}, J=13.7,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.50(\mathrm{dd}, J=13.7,9.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.46(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}(75 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ): $\delta 171.0,170.2,139.3,135.8,134.1,133.6,132.2,130.0,128.6,128.4,128.2,128.2,127.8,127.6,127.2$, $127.0,125.6,125.2,125.2,123.6,81.1,66.5,36.6,28.0$; HRMS (ESI, positive): Calcd. for $\mathrm{C}_{30} \mathrm{H}_{29} \mathrm{NNaO}_{2}{ }^{+}$ $[\mathrm{M}+\mathrm{Na}]^{+} 458.2091$, found: 458.2091. HPLC analysis: Daicel Chiralcel AD-H, $n$-hexane/isopropanol = 97:3, flow rate $=0.5 \mathrm{~mL} / \mathrm{min}$.
(S)-tert-Butyl 2-((diphenylmethylene)amino)-3-(1-naphthyl)propanoate ( $4 \mathbf{o}^{\prime} ; \mathrm{R}=\alpha$-naphthyl): Under the same reaction conditions for $\mathbf{4 0}$ except that catalyst $\mathbf{1 f}$ was replaced with $\mathbf{1 i}$, enantiomer $\mathbf{4 0} \mathbf{o}^{\prime}$ was obtained as colorless liquid. $85 \%$ yield; $97 \%$ ee (Lit $98 \%$ ee [41]); $[\alpha]_{D}^{29}-295.3^{\circ}\left(c=0.9, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ ).
(S)-3-Chloro phenylalanine hydrochloride (5): A mixture of $4 \mathbf{j}^{\prime}(1.2 \mathrm{~g}, 2.8 \mathrm{mmol})$ and $6 \mathrm{M} \mathrm{HCl}(6 \mathrm{~mL})$ was heated at $100^{\circ} \mathrm{C}$ for 3 h , and then cooled to ambient temperature, resulting in white precipitates.

5 ( $525 \mathrm{mg}, 92 \%$ yield) was obtained by suction filtration as white solid, and was used for the next step without further purification. M.p. $264-266{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 3.18(\mathrm{~d}, \mathrm{~J}=6.2 \mathrm{~Hz}$, $2 \mathrm{H}), 4.17(\mathrm{~s}, 1 \mathrm{H}), 7.26-7.39(\mathrm{~m}, 4 \mathrm{H}), 8.62(\mathrm{~s}, 3 \mathrm{H}), 13.88(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta$ $35.1,52.9,127.2,128.5,129.5,130.4,133.1,137.7,170.2$; HRMS (ESI, positive): calcd. for $\mathrm{C}_{9} \mathrm{H}_{11} \mathrm{ClNO}_{2}$ $[\mathrm{M}+\mathrm{H}]^{+}$200.0473, found: 200.0474 .
(9H-Fluoren-9-yl)methyl (S)-(3-chloro)phenylalanine (6): To an ice-cold solution of 5 ( $525 \mathrm{mg}, 2.6 \mathrm{mmol}$ ), dioxane ( 3 mL ) and $10 \% \mathrm{Na}_{2} \mathrm{CO}_{3}$ aqueous solution ( 6 mL ) was added dropwise a solution of $\mathrm{Fmoc}-\mathrm{Cl}$ ( $673 \mathrm{mg}, 2.6 \mathrm{mmol}$ ) in dioxane ( 3 mL ). The mixture was stirred at $0^{\circ} \mathrm{C}$ for 4 h , and then warmed to ambient temperature with the stirring continued for an additional 18 h . The reaction was quenched by adding $2 \mathrm{M} \mathrm{HCl}(5 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$. The resulting mixture was extracted with $\mathrm{EtOAc}(2 \times 60 \mathrm{~mL})$, and the combined extracts were washed with brine $(2 \times 30 \mathrm{~mL})$ and dried. The crude product was purified by flash column chromatography (eluting with hexane/EtOAc, 20:1) to give 6 ( $644 \mathrm{mg}, 59 \%$ yield) as white solid. M.p. $123-125{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 2.88(\mathrm{q}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H})$, $3.17(\mathrm{~s}, 1 \mathrm{H}), 4.09-4.17(\mathrm{~m}, 3 \mathrm{H}), 4.25(\mathrm{q}, J=4.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.20-7.39(\mathrm{~m}, 9 \mathrm{H}), 7.61(\mathrm{~d}, \mathrm{~J}=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.87$ (d, $J=7.5 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 36.5,46.6,55.9,65.6,120.2,125.2,125.4,126.3$, 127.1, 127.7, 128.0, 129.2, 129.9, 132.7, 140.6, 140.7, 141.2, 143.8, 143.9, 155.9, 173.9; HRMS (ESI, positive): calcd. for $\mathrm{C}_{24} \mathrm{H}_{20} \mathrm{ClNO}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 444.0973$, found: 444.0967.
(9H-Fluoren-9-yl)methyl(S)-(3-(3-chlorophenyl)-1-(methylamino)-1-oxopropan-2-yl)carbamate (7): Methyl-amine hydrochloride ( $202 \mathrm{mg}, 3 \mathrm{mmol}$ ) and $\mathrm{N}, \mathrm{N}$-diisopropylethylamine (DIEA, $388 \mathrm{mg}, 3 \mathrm{mmol}$ ) was added successively to an ice-cold stirred solution of the substituted 6 ( $624 \mathrm{mg}, 1.5 \mathrm{mmol}$ ), $\mathrm{HOBt}(405 \mathrm{mg}, 3 \mathrm{mmol})$ and $\operatorname{HBTU}(1.1 \mathrm{~g}, 3 \mathrm{mmol})$ in DMF $(6 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. The reaction mixture was stirred for 30 min at the same temperature, and then allowed to warm to ambient temperature while the stirring continued for an additional 12 h . The solvents and volatiles were removed under reduced pressure, and the residue was dissolved in $\mathrm{EtOAc}(150 \mathrm{~mL})$ and then washed with saturated $\mathrm{NaHCO}_{3}$ solution $(50 \mathrm{~mL})$ and brine $\left(2 \times 50 \mathrm{~mL}\right.$ ) and finally dried (anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ). After the solvent was concentrated, the crude product was crystallized from EtOAc to give 7 ( $504 \mathrm{mg}, 77 \%$ yield) as white solid. M.p. $179-181^{\circ} \mathrm{C} ; 96 \%$ ee; $[\alpha]_{D}^{21}-0.6^{\circ}\left(c=0.7, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 2.57(\mathrm{~d}, J=4.5 \mathrm{~Hz}$, $3 \mathrm{H}), 2.76-2.98(\mathrm{~m}, 4 \mathrm{H}), 3.16(\mathrm{~s}, 1 \mathrm{H}), 4.46(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.14-7.29(\mathrm{~m}, 5 \mathrm{H}), 7.38(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 2 \mathrm{H})$, $7.76(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.91(\mathrm{~d}, J=4.3 \mathrm{~Hz}, 2 \mathrm{H}), 8.12(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right):$ § 26.0, 37.6, 47.0, 56.6, 66.1, 120.6, 125.7, 125.8, 126.8, 127.5, 128.1, 128.4, 129.6, 130.3, 133.1, 141.0, 141.1, 141.4, 144.2, 144.2, 156.3, 172.1; HRMS (ESI, positive): calcd. for $\mathrm{C}_{25} \mathrm{H}_{24} \mathrm{ClN}_{2} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+} 435.1470$, found: 435.1470. HPLC analysis: Daicel Chiralcel OD-H, $n$-hexane/isopropanol $=85: 15$, flow rate $=$ $1.0 \mathrm{~mL} / \mathrm{min}$.
(S)-2-Amino-3-chloro-N-methylpropanamides (8): To a stirred solution of 7 ( $480 \mathrm{mg}, 1.1 \mathrm{mmol}$ ) in DMF $(4 \mathrm{~mL})$ was added piperidine $(2 \mathrm{~mL})$ at room temperature. The reaction mixture was stirred at ambient temperature and under nitrogen atmosphere for 2 h . The solvent and volatiles were removed under reduced pressure, and the residue was purified by flash column chromatography (eluting with $\mathrm{DCM} / \mathrm{MeOH}, 20: 1)$ to afford $8\left(210 \mathrm{mg}, 90 \%\right.$ yield) as light yellow liquid; ${ }^{1} \mathrm{H}-\mathrm{NMR}(300 \mathrm{MHz}$, DMSO- $d_{6}$ ): $\delta 7.91(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.32-7.23(\mathrm{~m}, 3 \mathrm{H}), 7.16(\mathrm{~m}, 1 \mathrm{H}), 3.42(\mathrm{dd}, J=8.1,5.2 \mathrm{~Hz}, 1 \mathrm{H})$, $2.92(\mathrm{dd}, J=13.4,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.77(\mathrm{~s}, 2 \mathrm{H}), 2.65(\mathrm{dd}, J=13.4,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.58(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}\right.$, DMSO- $d_{6}$ ): $\delta 174.3,141.4,132.9,130.0,129.3,128.2,126.3,56.1,40.5,25.6$; HRMS (ESI, positive): calcd. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{ClN}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}]^{+}$213.0789, found: 213.0789.
(9H-Fluoren-9-yl)methyl((S)-1-(((S)-3-(3-chlorophenyl)-1-(methylamino)-1-oxopropan-2-yl)amino)-3
-(4-carbamoylphenyl)-1-oxopropan-2-yl)carbamate (10): HOBt ( $126 \mathrm{mg}, 0.9 \mathrm{mmol}$ ) and HBTU ( 354 mg , $0.9 \mathrm{mmol})$ were added to a stirred solution of $9(0.52 \mathrm{mmol})$ in DMF $(6 \mathrm{~mL})$ at rt. After the mixture was cooled to $0{ }^{\circ} \mathrm{C}, 8(165 \mathrm{mg}, 0.8 \mathrm{mmol})$ and DIEA ( 1 mmol ) were introduced. After the whole reaction mixture was stirred at rt for 12 h , the solvents and volatiles were removed under reduced pressure. The solid residue was crystallized from dichloromethane to give $\mathbf{1 0}$ ( $350 \mathrm{mg}, 51 \%$ yield) as white solid.
M.p. $238-240{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 2.57(\mathrm{~d}, J=4.3 \mathrm{~Hz}, 3 \mathrm{H}), 2.68-2.88(\mathrm{~m}, 2 \mathrm{H}), 2.95-3.01$ $(\mathrm{m}, 2 \mathrm{H}), 4.00-4.31(\mathrm{~m}, 4 \mathrm{H}), 4.44-4.51(\mathrm{~m}, 1 \mathrm{H}), 7.16-7.42(\mathrm{~m}, 12 \mathrm{H}), 7.60(\mathrm{t}, J=7.57 \mathrm{~Hz}, 3 \mathrm{H}), 7.78-7.93(\mathrm{~m}$, $6 \mathrm{H}), 8.24(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 25.6,37.4,37.7,46.6,53.9,56.0,65.8$, $120.2,125.3,125.4,126.5,127.2,127.4,127.7,128.1,129.1,130.0,132.4,132.8,140.3,140.8,141.6,143.8$, 143.9, 155.8, 167.9, 171.0, 171.3; HRMS (ESI, positive): calcd. for $\mathrm{C}_{35} \mathrm{H}_{33} \mathrm{ClN}_{4} \mathrm{O}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 647.2032$, found: 647.2035.

4-((S)-2-Amino-3-(((S)-3-(3-chlorophenyl)-1-(methylamino)-1-oxopropan-2-yl)amino)-3-oxopropyl)benzamide (11): Piperidine ( 2 mL ) was added to a stirred solution of $10(330 \mathrm{mg}, 0.5 \mathrm{mmol})$ dissolved in DMF $(4 \mathrm{~mL})$, and the mixture was stirred at rt for 2 h . After the completion of the reaction, the solvent and volatiles were removed under reduced pressure, and the solid residue was crystallized from EtOAc to give 11 as light yellow solid ( $200 \mathrm{mg}, 95 \%$ yield). M.p. $237-239{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta$ $2.54(\mathrm{t}, \mathrm{J}=10.5 \mathrm{~Hz}, 3 \mathrm{H}), 2.79-3.39(\mathrm{~m}, 3 \mathrm{H}), 4.46(\mathrm{~s}, 1 \mathrm{H}), 7.12-7.31(\mathrm{~m}, 6 \mathrm{H}), 7.77(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.99$ $(\mathrm{t}, J=9.1 \mathrm{~Hz}, 2 \mathrm{H}), 8.20(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 25.6,37.6,53.3,56.2,126.4$, 127.4, 128.1, 129.1, 129.2, 129.9, 132.2, 132.7, 140.4, 142.2, 167.8, 171.0, 174.0; HRMS (ESI, positive): calcd. for $\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{ClN}_{4} \mathrm{O}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$245.1351, found: 245.1350.
(2S)-3-(3-Chlorophenyl)-2-((2S)-2-(2-cyclohexyl-2-phenylacetamido)-3-phenylpropan-amido)-N-methyl -propanamide (13): HOBt ( $114 \mathrm{mg}, 0.8 \mathrm{mmol}$ ) and HBTU ( $320 \mathrm{mg}, 0.8 \mathrm{mmol}$ ) were added to a stirred solution of $\mathbf{1 2}(0.8 \mathrm{mmol})$ in DMF $(4 \mathrm{~mL})$ at rt. After the mixture was cooled to $0^{\circ} \mathrm{C}, \mathbf{1 1}(170 \mathrm{mg}$, 0.4 mmol ) was added, followed by addition of DIEA ( 1.2 mmol ). The reaction mixture was stirred at ambient temperature for 12 h , the solvent and volatiles were evaporated under reduced pressure, and then the solid residue was crystallized from EtOAc to generate $13(160 \mathrm{mg}, 63 \%$ yield) as white solid. M.p. $243-245{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 0.56-0.64(\mathrm{~m}, 3 \mathrm{H}), 0.81-0.93(\mathrm{~m}, 6 \mathrm{H})$, $1.06-1.59(\mathrm{~m}, 3 \mathrm{H}), 2.55(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 3 \mathrm{H}), 2.68-3.06(\mathrm{~m}, 3 \mathrm{H}), 3.19(\mathrm{~d}, J=10.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.39-4.44(\mathrm{~m}, 2 \mathrm{H})$, $6.96(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.18-7.33(\mathrm{~m}, 7 \mathrm{H}), 7.52(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.77-7.94(\mathrm{~m}, 3 \mathrm{H}), 8.21-8.25(\mathrm{~m}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}\right.$, DMSO- $d_{6}$ ): $\delta 24.9,25.2,25.9,26.0,30.8,37.7,39.1,42.7,54.1,57.3,121.8,126.8,127.5$, 128.6, 128.7, 129.3, 129.7, 130.6, 132.2, 132.5, 132.6, 140.8, 141.3, 141.7, 168.1, 171.1, 171.2, 173.0; HRMS (ESI, positive): calcd. for $\mathrm{C}_{34} \mathrm{H}_{40} \mathrm{ClN}_{4} \mathrm{O}_{4}[\mathrm{M}+\mathrm{H}]^{+}$603.2733, found: 603.2730.

The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$-NMR spectra of the compounds are available in supplementary materials.

## 4. Conclusions

In summary, we have developed a practical method for synthesizing both enantiomers of unnatural $\alpha$-amino acid derivatives by asymmetric $\alpha$-alkylation of $N$-(dibenzylidene)glycine tert-butyl ester (2) with substituted benzyl bromides and 1-(bromomethyl)naphthalene under the catalysis of O -allyl- N -(9-anthracenmethyl) cinchodium bromide (1f) and O -allyl- N -(9-anthracenmethyl) bromide (1i), respectively. A series of both $(R)$ - and (S)-enantiomers of unnatural $\alpha$-amino acid derivatives were obtained in good to excellent yields and with excellent enantioselectivity, and the procedure is simple, mild and scalable. Furthermore, the stereochemistry of the products is fully predictable and controlled: the cinchonine-type phase transfer catalyst 1 f resulted in all $(R)$ - $\alpha$-amino acid derivatives, whereas the cinchonidine-type phase transfer catalyst $\mathbf{1 i}$ afforded the $(S)-\alpha$-amino acid derivatives. Both resulting enantiomers of the substituted $\alpha$-phenylalanine derivatives have been used for synthesizing new allosteric antagonists for $\beta_{2}$ AR.

Supplementary Materials: The supplementary materials containing NMR spectra and HPLC chromatograms can be accessed online.

Author Contributions: X.C. conceived and designed the experiments; L.J. and S.Z. carried out the synthesis and characterization of all compounds; All authors discussed the contents of the manuscript.
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Sample Availability: Samples of the compounds $\mathbf{4 a - 4 o}, \mathbf{4 a ^ { \prime }} \mathbf{- 4 \mathbf { o } ^ { \prime }}$ and 5-13 are available from the authors.
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[^0]:    ${ }^{a}$ Reactions were performed with $2(0.1 \mathrm{mmol})$, alkylation reagent ( 0.5 mmol$), 50 \% \mathrm{KOH}(0.5 \mathrm{mmol})$ and $\mathbf{1 h}$ ( 0.01 mmol ) in toluene $/ \mathrm{CHCl}_{3}$ at $-40{ }^{\circ} \mathrm{C}$. ${ }^{b}$ Isolated yield. ${ }^{c}$ Enantiomeric excess was determined by HPLC analysis using a chiral column with $n$-hexane-isopropanol as eluent.

