

Structural mechanism for Bruton's tyrosine kinase activation at the cell membrane

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Bruton's tyrosine kinase (Btk) is critical for B cell proliferation and activation, and the development of Btk inhibitors is a vigorously pursued strategy for the treatment of various B cell malignancies. A detailed mechanistic understanding of Btk activation has, however, been lacking. Here, inspired by a previous suggestion that Btk activation might depend on dimerization of its lipid-binding PH–TH module on the cell membrane, we performed long-timescale molecular dynamics simulations of membrane-bound PH–TH modules and observed that they dimerized into a single predominant conformation. We found that the phospholipid PIP₃ stabilized the dimer allosterically by binding at multiple sites, and that the effects of PH–TH mutations on dimer stability were consistent with their known effects on Btk activity. Taken together, our simulation results strongly suggest that PIP₃-mediated dimerization of Btk at the cell membrane is a critical step in Btk activation.

Bruton's tyrosine kinase | dimerization | allosteric activation | PIP_3 | enhanced sampling

Bruton's tyrosine kinase (Btk), a Tec-family tyrosine kinase, is a peripheral membrane-binding protein present in all blood cells except for T cells and natural killer cells (1, 2). Btk participates in a number of receptor-mediated signaling pathways (2-6). In B cells, where Btk's biological functions are best understood, Btk activates phospholipase C-y2, which produces second messengers that are essential for B cell activation and proliferation. The up-regulation of Btk is associated with various B cell malignancies, including chronic lymphocytic leukemia and mantle cell lymphoma. Btk down-regulation causes X-linked agammaglobulinemia (XLA), a severe disease of primary immunodeficiency (7). Btk is also implicated in maintaining a number of autoimmune diseases, such as rheumatoid arthritis, systemic lupus erythematosus, and multiple sclerosis (8, 9). A structural characterization of Btk's activation mechanism would significantly advance our understanding of Btk's role in cell signaling and could facilitate the development of Btk-specific inhibitors (10), but such a characterization has thus far remained elusive.

Btk is autoinhibited in the cytoplasm, where it adopts a compact conformation similar to the inactive forms of the kinases c-Src and c-Abl (11-14), in which the SH2 and SH3 domains stabilize an inactive conformation of the kinase domain. In the case of Btk, its lipid-interaction module [the PH-TH module, composed of a pleckstrin homology (PH) domain and a Tec homology (TH) domain] acts in conjunction with the SH2 and SH3 domains to stabilize the inactive conformation of the Btk kinase domain (11, 15). It has been shown that Btk activation is primarily regulated by phosphorylation upon membrane recruitment (recruitment triggers transautophosphorylation or transphosphorylation by membranebound Src-family kinases) (16-22). Membrane recruitment occurs by way of interactions between the canonical lipid-binding site of the PH-TH module and PIP₃ membrane lipids, but the structural mechanism by which membrane recruitment leads to Btk activation is unknown.

Saraste and Hyvönen (23), after identifying a crystallographic dimer of the PH–TH module (often referred to as the "Saraste dimer"), suggested that dimerization of the PH–TH modules of membrane-bound Btk molecules may promote the transautophosphorylation of their kinase domains, leading to activation. Dimers of the PH–TH module have never been directly detected in solution by biophysical methods, however, making their functional relevance uncertain. Membrane binding might be critical for PH–TH dimerization due to the increased local Btk concentration (24, 25), but it is difficult to obtain atomic structural information about the protein on the membrane by traditional experimental means. Molecular dynamics (MD) simulations have enabled the observation of spontaneous binding events between proteins (26–32), providing an alternative, computational route for obtaining atomic structural information about protein oligomerization on membranes.

In this study, we use long-timescale MD simulations to examine whether and, if so, how the PH–TH module of Btk dimerizes on a membrane. In our simulations, which started from crystal structures of individual PH–TH modules and used no other prior structural information, the PH–TH modules spontaneously dimerized on the membrane into a single predominant conformation closely resembling the Saraste dimer. We observed that PIP₃ allosterically stabilized the dimer conformation, binding not only at the canonical PIP₃-binding site but at the peripheral IP₆binding site, which has been shown to be important for Btk activation in solution (11). At higher PIP₃ concentrations, an additional PIP₃ molecule bound at a site between the two PH–TH modules, interacting with both modules simultaneously, and there is evidence that this further stabilized the interface.

Our simulations also provide an explanation for the effects of multiple mutations in the PH–TH module that are known to be

Significance

Bruton's tyrosine kinase (Btk) activation on the cell membrane is critical for B cell proliferation and development, and Btk inhibition is a promising treatment for several hematologic cancers and autoimmune diseases. Here, we examine Btk activation using the results of long-timescale molecular dynamics simulations. In our simulations, Btk lipid-binding modules dimerized on the membrane in a single predominant conformation. We observed that the phospholipid PIP₃—in addition to its expected role of recruiting Btk to the membrane—allosterically mediated dimer formation and stability by binding at two novel sites. Our results provide strong evidence that PIP₃-mediated dimerization of Btk at the cell membrane is a critical step in Btk activation and suggest a potential approach to allosteric Btk inhibitor development.

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associated with Btk dysfunction: In our simulations, mutations that lead to down-regulation of Btk and cause immunodeficiency diseases destabilized the PH–TH dimer, and a mutation that leads to the up-regulation of Btk and causes cell overproliferation stabilized the dimer interface. These observations provide compelling evidence that the Saraste dimer is important for the regulation of Btk. Taking this together with our simulation results showing that PIP₃ can allosterically mediate dimer formation, we further propose that PIP₃ plays a physiological role as an allosteric activator of Btk by stimulating dimerization of the PH–TH modules and thus promoting transautophosphorylation of the kinase domains.

Results

The PH–TH Module Bound Multiple PIP₃ Lipids on the Membrane. To establish the utility of our MD simulations for studying the behavior of the PH–TH module, we first showed that our simulations could reproduce essential experimentally observed interactions between the PH–TH module and soluble inositol phosphates (*SI Appendix*, Texts T1 and T2). We then studied the binding of the PH–TH module of Btk onto a membrane containing 94% POPC and 6% PIP₃. We started the simulations by placing a PH–TH module in solution, about 10 Å away from the membrane surface, and positioned the module so that the canonical PIP₃-binding site did not face the PIP₃-containing leaflet of the membrane (Fig. 1*A*).

Within 2 μ s of simulated time, the PH–TH module was recruited onto the membrane, with a PIP₃ molecule bound to the canonical site (Fig. 1*B*). The binding pose of PIP₃ at the canonical site resembles that of IP₄ bound to the PH–TH module, as seen in crystal structures (33). Residues Asn-24, Tyr-39, Arg-28, and Lys-12 interacted with phosphate groups on the 2, 3, and 4 positions of the *myo*-inositol ring. In addition, Ser-21, Ser-14, and Lys-18 were also in close contact with the PIP₃ molecule and formed transient interactions with PIP₃ in the simulations. The bound PIP₃ stayed at the canonical binding site until the end of simulations.

Once the PH-TH module was recruited onto the membrane through PIP₃ binding to the canonical site, the peripheral IP₆binding site of the PH-TH module repeatedly bound a second PIP₃ lipid (Fig. 1*C*). Multiple PIP₃ binding and unbinding events were observed at the peripheral site in individual trajectories, and the overall occupancy rate of the peripheral site was ~96%; in further simulations at a lower (1.5%) PIP₃ concentration, the peripheral site occupancy was about 80%. The less stable binding of PIP_3 in the peripheral site is to be expected, since the peripheral site is relatively flat, whereas the canonical site has a well-defined groove where PIP₃ binds. PIP₃ assumed multiple binding poses at the peripheral site, with residues Lys-52, Tyr-40, and Arg-49 most frequently coordinating PIP₃ binding; these are also the key residues experimentally observed to coordinate IP_6 binding (11) (Fig. 1C). Residues Lys-36 and Ser-55, which are outside the area known to bind IP₆, also interacted with PIP₃ at the peripheral site.

Although MD simulation studies have shown that several PH domains bind anionic lipids at noncanonical binding sites (34–37), the peripheral site identified in our study has not been previously reported as a lipid-binding site for Btk. In our simulations, we also observed PIP₃ repeatedly occupying a third binding site, located on the other side of the IP₄-binding loop (*SI Appendix*, Fig. S1). We have not studied PIP₃ binding in this pocket in detail, however, due to its low occupancy rate (~30%) in our simulations.

Our simulations do not rule out the possibility that other types of anionic lipids, such as PIP_2 or PS, might also bind at these regions of the PH–TH module. In this work, we chose to exclusively study PIP_3 due to the critical signaling role it is known to play in mediating B cell activation, and its known function as a regulator of Btk activity. The finding that the PH–TH module can simultaneously bind multiple PIP_3 lipids on the membrane has, to our knowledge, not been noted in prior studies and has important implications for Btk activation on membranes, as we will show in the results below. **The PH-TH Modules Spontaneously Dimerized on the Membrane with a Single Predominant Interface.** PIP₃-bound PH-TH modules diffused freely on membranes in our simulations, and we were thus able to study the encounter process of two membrane-bound PH-TH modules. Here, we show that PH-TH modules can indeed spontaneously dimerize on membranes, with a single dimer interface predominating.

We began by performing conventional MD simulations of two separate, non-membrane-bound PH-TH modules and a piece of membrane containing 6% PIP₃ (Fig. 24). In 20 independent 20-µs trajectories, the individual modules always bound with the membrane, and a variety of PH-TH dimers subsequently formed, in each case within the first few microseconds of simulation time (*SI Appendix*, Fig. S2). Some dimer interfaces, despite having small interface areas (~500 Å²) and lacking hydrophobic packing interactions and hydrogen bonds, stayed bound through the end of the 20-µs simulations (*SI Appendix*, Fig. S2). The fact that such dimer interfaces did not dissociate in the simulations is not unexpected: Observing dissociation of protein-protein complexes is a well-known challenge for conventional MD simulations, since experimentally observed dissociation events often occur on a longer timescale than is accessible to MD simulations (32).

To sample PH–TH module conformations more extensively than would be possible using conventional MD simulations of reasonably achievable length, we employed an enhanced sampling method called "tempered binding." (32) This technique intermittently weakens the interactions between the two proteins (and thus all potential PH–TH dimer interfaces), thus increasing the frequency of dimer dissociation and accelerating the exploration of different protein–protein interfaces. The method nonetheless rigorously preserves the property of Boltzmann sampling at each interaction strength, with unscaled portions of the trajectory sampled from the same distribution as a conventional MD simulation with an unmodified Hamiltonian (see *SI Appendix, Methods* for details).

We performed 24 tempered binding simulations, each based on three replicas, with a total of ~6 ms of simulated time for all simulations combined. All tempered binding simulations started from the same configuration—in which two PH–TH modules were bound to the membrane but were not in contact with each other—but varied in membrane PIP₃ concentration, tempering strength, and protein backbone-correction strength.

In these tempered binding simulations, the two PH-TH modules diffused freely on the membrane, and multiple association/dissociation events between the two modules were observed. We then analyzed all dimer conformations assumed by the PH-TH modules in these simulations: The two modules were in contact in 4 million of the 6 million total frames extracted from the 24 simulations, and we began by clustering the instantaneous structures in these 4 million frames according to their structural similarity (Fig. 2B and SI Appendix, Fig. S3). The largest cluster of these PH-TH dimer structures contained 61% of total dimer structures, while the nextlargest cluster contained less than 1%; given the small size of all but the largest, all further structural analysis was based only on the most populated cluster (Fig. 2B). Notably, clustering only those portions of the trajectory in which the protein-protein interactions were untempered led to the same predominant cluster, supporting the relevance of this dimer cluster under untempered conditions.

Dimers in the predominant cluster formed repeatedly in all 24 simulations (across all tempering parameters and PIP₃ concentrations). The dissociation time for these dimers varied from simulation to simulation, ranging from at least 15 μ s of simulation time to up to 500 μ s, which was the length of the longest simulations. We also observed multiple reversible dimerization events for these dimers in individual trajectories under certain tempering conditions (Fig. 2*C*). Dimer structures in other clusters, on the other hand, were short-lived, and most of them dissociated within a few microseconds of simulation time.

The Predominant PH–TH Dimer Interface Resembles the Saraste Dimer Interface. The most populated PH–TH dimer interface in our simulations resembles the Saraste dimer observed in crystal structures (Fig. 2*B*): The root-mean-square deviations (rmsds) of the PH–TH dimer interfaces in the largest cluster range from

2.5 to 4.5 Å compared with the Saraste dimer. The Saraste-like dimer interfaces that formed in our simulations had two distinct conformations (Fig. 3), with the primary structural difference between them occurring at the β 3– β 4 hairpin (Fig. 3*B*). In the conformation closest to the Saraste crystal structure, with an rmsd



Fig. 1. Spontaneous binding of the PH–TH module to the membrane. (*A*) Cartoon illustration of a membrane-binding simulation setup (see *SI Appendix, Methods* for details). (*B, Right*) Minimum distance between atoms in PIP₃ lipids and in the canonical binding site for three independent membrane-binding simulations. (*B, Left*) A structure taken from one of these simulations ($t = 10 \mu s$) showing a PIP₃ molecule bound at the canonical site. (*C, Right*) Minimum distance between atoms in PIP₃ lipids and in the peripheral binding site for three independent membrane-binding simulations. (*C, Left*) A structure taken from one of these simulations at the peripheral binding site for three independent membrane-binding simulations. (*C, Left*) A structure taken from one of these simulations at the peripheral site.



Fig. 2. Dimerization of the PH–TH module on the membrane. (A) Cartoon illustration of an encounter simulation setup (see *SI Appendix, Methods* for details). (B) Conformation clusters of PH–TH dimers formed in the simulations (see *SI Appendix, Methods* for details). A representative structure of the most populated dimer conformation is shown (see *Inset*). (The PIP₃ lipids bound in the canonical binding sites are also shown.) (C) Reversible dimerization of two membrane-bound PH–TH modules on a membrane that contains 1.5% PIP₃ and 98.5% POPC in tempered binding simulations. The rmsd at the Saraste interface is calculated for all interface residues (residues 9, 11, 42, 44, 92, and 95) of the two modules with respect to the Saraste dimer interface conformation seen in a crystal structure of the PH–TH module (PDB ID code 1BTK). (This same interface rmsd calculation is used for the rmsd plots in all other figures, unless stated otherwise.)

at the dimer interface of only ~2.5 Å (Fig. 3*A*), the interface is tightly packed. In the other conformation, the dimer interface is loosely packed, and the interface regions of the individual PH–TH modules are partially closed (Fig. 3*B*): The Phe-44 of each module adopts an alternative rotamer conformation (to that seen in crystal structures) and remains packed against its own module's Ile-92 and Ile-95 residues, as is also the case in simulations of a single PH–TH module. The backbone of the β 3– β 4 hairpin (residues 42–50) moves ~3.2 Å toward the center of the interface. The rmsd of the second conformation at the dimer interface is ~4.0 Å compared with the Saraste dimer crystal structure.

This loosely packed conformation of the Saraste interface has not been observed in crystal structures of the PH–TH dimer. Nevertheless, we find this conformation very informative: It provides evidence for a structural mechanism by which the tightly packed Saraste dimer interface can form while the interface regions of the individual PH–TH modules remain partially closed (*SI Appendix*, Text T1). In our simulations of an individual PH– TH module, residues at the Saraste interface region rarely adopted the open conformation, which is the conformation compatible with directly forming the tightly packed Saraste interface with another module. We thus did not expect the direct formation of the Saraste dimer interface to occur frequently in our simulations, since it would require the spontaneous opening of the interface regions of both PH–TH modules simultaneously during the encounter process. Indeed, none of the tightly packed Saraste dimer interfaces we observed in our tempered binding simulations formed through the direct encounter of the two modules with open Saraste interface regions: All Saraste dimer interfaces formed by way of conformational transitions through the loosely packed interface. We thus call the loosely packed interface the "pre-Saraste" interface. Furthermore, in one of our conventional MD simulations of two initially separated PH–TH modules on a membrane, we also observed the PH–TH modules spontaneously form the pre-Saraste dimer, strongly suggesting that the pre-Saraste dimer is an obligatory intermediate state between the unbound modules and the tightly packed Saraste dimer.

The Stability of the Saraste Interface Was Sensitive to Mutations on the PH–TH Module. The observation in our simulations that the PH– TH modules spontaneously dimerized with a single predominant interface on the membrane, where Btk performs its biological functions, suggests that this Saraste dimer interface may be biologically



Fig. 3. Structural analysis of Saraste dimers formed on membranes. (*A*) Structural comparison of Saraste dimers formed in the encounter simulations and in the Saraste dimer crystal structure (PDB ID code 1BTK). (*B*) An instantaneous structure of the pre-Saraste interface formed in the encounter simulations.

relevant. Additionally, in solution the transient dimerization of PH– TH modules at the Saraste interface in the presence of IP_6 can activate Btk molecules by promoting transautophosphorylation of their kinase domains, and mutations that destabilize the Saraste interface slow Btk activation in solution (11). Together, these findings suggest that the stability of the Saraste dimer may regulate activation of membrane-bound Btk. To test this hypothesis, we used MD simulations to study two mutations on the PH–TH module that are known to affect Btk activity, but for which a structural explanation is lacking. In our simulations, we examined whether and, if so, how these mutations influence the stability of the Saraste dimer on a membrane.

Initially, we explored using tempered binding simulations to study the thermodynamic equilibrium between a Saraste dimer and the unbound modules, but the results were inconclusive due to the small number of association and dissociation events that could be observed on our simulation timescale. Instead, we proceeded by qualitatively analyzing the potential structural consequences for Saraste dimer stability when different mutations were introduced. In this set of simulations, we used the tightly packed conformation of the Saraste dimer on a membrane as a starting structure and performed tempered binding simulations until the dimer dissociated (Fig. 4A).

We first studied the loss-of-function mutant F98V, which is located in the PH domain and has been identified in patients with the immunodeficiency disease XLA (38). The molecular basis for the loss of function produced by F98V is not clear: Phe-98 is involved neither in the hydrophobic-core packing interactions of the PH-TH module nor in directly mediating PIP₃binding interactions, suggesting that the loss-of-function effect of F98V is unlikely to be related to misfolding of the individual PH-TH module or to interference with its membrane recruitment. One effect we did observe in our simulations, however, was that substituting Phe with the less bulky Val loosened the packing between residues Leu-11 and Ile-9 (Fig. 4B). Ile-9 is part of the Saraste interface, suggesting that the loosened packing at the end of the $\alpha 1$ helix, allosterically produced by F98V, could affect the stability of the Saraste interface. Consistent with this notion, in our simulations, we observed that the F98V dimer was not stable in the tightly packed Saraste interface conformation, quickly relaxed into the pre-Saraste conformation, and then dissociated into separate PH-TH modules (Fig. 4B and SI Appendix, Fig. S4). In addition to F98V, we observed that the Saraste interface was less stable in another mutant known to cause XLA, F25S (SI Appendix, Fig. S4).

We next studied the effects of the mutation E41K, a gain-offunction mutation in the PH-TH module of Btk (39). Glu-41 is located on the β 3– β 4 hairpin, near the Saraste interface, and its mutation to lysine constitutively activates Btk and causes cell overproliferation (39). Previous studies have shown that the PIP₂-mediated membrane localization of the E41K mutant is more significant than that of the wild-type protein in resting cells (which have no PIP₃ lipids), pointing to an important role for PIP₂ binding in Btk activation in the absence of stimulation (40-43). Notably, the E41K mutant is also found to generate markedly enhanced BCR-induced Ca²⁺ signal in hematopoietic cells under stimulation conditions, leading to PIP₃ production (44, 45). This signaling enhancement effect of E41K in activated cells is likely not solely the result of augmented PIP₂-mediated membrane localization, however, since both the mutant and the wild-type protein are known to be recruited to the membranes through the well-established PIP₃ binding mechanism. The structural basis of the stronger membrane localization of the E41K mutant in activated cells, and the interactions between E41K dimers and membranes after the mutant is recruited onto the membrane, are not well understood. In our simulations, we observed Lys-41 repeatedly bind a third PIP₃ lipid at the top of the $\beta 3-\beta 4$ hairpin, confirming the notion, based on previous crystallographic studies, that the E41K mutant can bind an



Fig. 4. Stability analysis of Saraste dimer variants on a membrane. (A) Representative tempered binding simulation trajectories showing dissociation of the tightly packed Saraste conformation in two individual trajectories. The tempered binding simulations started from a Saraste dimer on a membrane containing 6% PIP₃ and 94% POPC. (B) Stability analysis of the F98V mutant, starting from the Saraste dimer conformation, on a membrane. An instantaneous structure from a simulation ($t = 20 \ \mu$ s) shows the local conformational rearrangement that occurred at the Saraste interface before the dimer dissociation of the F98V dimer in two individual trajectories. (C) Stability analysis of the E41K mutant, starting from the Saraste dimer conformation, on a membrane. An instantaneous structure from a simulation ($t = 100 \ \mu$ s) shows PIP₃ bound to residue Lys-41 at the bridging site before the dimer dissociated. Representative tempered binding simulation trajectories show the dissociation of the E41K dimer in two individual trajectories.

additional PIP₃ molecule (33). Although the stability of the E41K dimer was qualitatively comparable to that of the wild-type dimer in our simulations, the PIP₃ that bound at the top of the β 3- β 4 hairpin interacted simultaneously with both modules in the Saraste dimer (leading us to term this binding location the "bridging site"), revealing a structural mechanism that could further strengthen Saraste-interface stability (Fig. 4*C*). We thus speculate that the activating effect of E41K could be the result of additional PIP₃ molecules stabilizing the Saraste interface once the dimer has formed on the membrane, which could further promote the transautophosphorylation of Btk kinase domains.

Finally, we note that, as expected, Saraste dimers with mutations at the dimer interface (I9R/L29R, Y42R/F44R) either barely stayed in the tightly packed Saraste interface conformation or dissociated immediately in our simulations (*SI Appendix*, Fig. S4). These mutants have been previously shown to impair Btk activation by IP₆ in solution (11), and our simulation results strongly suggest that they are also likely to impede Btk activation on membranes.

The qualitative correlation between the stability of the mutated Saraste dimers and the functional consequences of these mutations suggests that the Saraste interface plays an important role in regulating Btk activation. **The Stability of the Saraste Dimer Interface Was also Sensitive to the Amount of PIP₃ in the Membrane.** In B cells, down-regulation of Btk activity causes immunodeficiency diseases, and up-regulation of Btk activity is correlated with autoimmune diseases and various blood cancers, suggesting that the activation of Btk must be tightly controlled to maintain normal B cell functions. The observation that Btk function is very sensitive to the stability of the Saraste dimer interface suggests that the stability of the Saraste dimer itself might be tightly controlled in cells. We now describe a set of simulation observations that collectively suggest that PIP₃, the anchor of Btk on membranes, is a potential allosteric regulator for Btk that may control the stability of the Saraste interface.

The first insight came from analyzing the tempered binding simulations discussed above, in which we found that Saraste dimers that formed on membranes with 6% PIP₃ lipid content did not dissociate as frequently as those that formed on membranes with 1.5% PIP₃ (as seen in Figs. 5A and 2C, respectively). Although we were unable to quantitatively compare the dissociation constants of the Saraste dimer at 1.5% and 6.0% PIP₃ membrane lipid concentrations due to the small number of association and dissociation events, we found that fluctuations at the pre-Saraste interface, which preceded dissociation in our simulations, were more pronounced in simulations with 1.5% membrane PIP₃ content than in those with 6% PIP₃ (Fig. 5B). Dimer conformations with rmsds greater than 4.5 Å from the Saraste dimer crystal structure at the interface were repeatedly visited in the 1.5% PIP₃ condition (Fig. 5B). These conformations resembled the partially broken Saraste dimer conformations seen in our solution-based conventional simulations (SI Appendix, Text T3 and Fig. S5). The large, relatively frequent conformational fluctuations at the dimer interface that repeatedly visit a structure on the dissociation pathway suggest that the dimers were weaker in the 1.5% PIP₃ condition. In simulations with 6% membrane PIP₃ content, the rmsd from the crystal structure at the Saraste interface consistently stayed below 4.5 Å. Overall, our observations provide strong evidence that the Saraste dimer was more stable in the 6.0% PIP₃ lipid condition than in the 1.5% PIP₃ lipid condition.



Fig. 5. Fluctuations at the pre-Saraste dimer interface. (*A*) Representative tempered binding simulation trajectories of two PH–TH modules on a membrane containing eight PIP₃ lipids. (*B*) Fluctuations at the pre-Saraste dimer interface that formed on membranes with two and eight PIP₃ lipids (1.5% and 6% membrane PIP₃ content, respectively).

PIP₃ Binding at the Canonical and Peripheral Sites of the PH-TH Modules Reduced Fluctuations at the Saraste Dimer Interface. Some polyphosphoinositides, such as PIP₂ and PIP₃, are known to spontaneously form clusters under certain conditions (46). One possible explanation for the apparently rare dissociation of the Saraste dimer in the 6% PIP₃ condition is that PIP₃ clusters spatially constrained the PH-TH dimer on the membrane, leading to infrequent dimer dissociation. We examined the localization pattern of PIP₃ in our simulations to investigate this possibility, but found that no stable PIP₃ clusters formed in any of our simulations. Indeed, the PIP₃ molecules were quite isolated from each other, as a result of binding to distinct sites on the PH-TH module. We found that this binding of PIP₃ at both the canonical site and the peripheral site of PH-TH modules had profound effects on the stability of the Saraste interface, suggesting that PIP₃ plays a role in regulating the PH-TH dimer allosterically.

To study more rigorously whether both the canonical and peripheral sites are involved in the stabilizing effect of PIP₃ binding described in the previous section, we performed another set of tempered binding simulations, varying only the number of PIP₃ lipids in the membrane (from one to eight), and analyzed the fluctuations at the dimer interface after the formation of the pre-Saraste interface. We first observed that the Saraste dimer interface was not stable in solution, and that PIP₃ binding significantly reduced fluctuations at the Saraste interface (SI Appendix, Fig. S5A). PIP₃'s stabilizing effect on the Saraste interface became evident when comparing the interface conformations when one or both of the two canonical sites in the Saraste dimer had bound PIP₃ ligands (Fig. 6 and *SI Appendix*, Fig. S5A): In simulations with only one PIP₃ molecule in the membrane, one PH-TH module in the Saraste dimer was attached to the membrane at the canonical site, while the other module remained in solution. We observed that conformations with large rmsds (4.5 Å or more) from the Saraste crystal structure at the dimer interface appeared frequently in simulations with only one PIP₃ molecule (Fig. 6). When two PIP₃ lipids were present, occupying both canonical binding sites in the Saraste dimer, the population of conformations with large rmsds at the Saraste dimer interface shrank from 16 to 8%, and the population of conformations with small rmsds (3.0 Å and below) at the Saraste dimer interface increased to 7% (Fig. 6).

We then observed that with four PIP₃ lipids in the membrane, the stability of the Saraste dimer improved further (Fig. 6). In this set of simulations, PIP₃ lipids typically occupied both the canonical and peripheral sites in the Saraste dimer: The occupancy rate at each peripheral binding site was ~80%. In these simulations, the population of conformations with large rmsds at the dimer interface fell further, to 3%. The population of conformations with small rmsds at the Saraste dimer interface increased from 7 to 15%, consistent with the notion that binding of PIP₃ at the peripheral site further stabilizes Saraste interface.

In the presence of membranes with eight PIP₃ molecules (6% membrane PIP₃ content), the canonical and peripheral sites of the Saraste dimer were almost continuously occupied in our simulations. We observed a further reduction in the population of conformations with large rmsds, from 3 to <1%. In addition, we found that the peak rmsd distribution was shifted from 3.7 Å to 3.2 Å, suggesting that the PIP₃ occupying the peripheral site of each PH–TH module stabilizes the Saraste interface.

Further evidence for the importance of the PH–TH module's peripheral binding site in stabilizing the Saraste interface is seen in simulations we performed to reveal the thermodynamic equilibrium between the tightly packed Saraste conformation and the more loosely packed pre-Saraste conformation (*SI Appendix*, Fig. S64; see *SI Appendix*, *Methods* for details). In simulations with 6% PIP₃ content, the PH–TH dimer reversibly converted between the pre-Saraste conformation (45%) and the tightly packed Saraste conformation (55%) before dissociating



Fig. 6. Sensitivity of fluctuations at the pre-Saraste dimer interface to membrane PIP₃ concentration. Fluctuations at the pre-Saraste dimer interface are shown, measured under various membrane PIP₃ concentrations. The histograms of the rmsd of the interface residues are calculated from 100- μ s simulation trajectories initiated from Saraste dimers that originally formed in tempered binding simulations; only data from the portion of the trajectories where the interactions were unscaled are used.

(*SI Appendix*, Fig. S6*B*). The free energy difference of the transition between the two Saraste conformations was 0.05 ± 0.2 kcal·mol⁻¹. In the presence of a membrane with 1.5% PIP₃ content, we found that the fraction of the tightly packed Saraste dimer fell to 40%, and the free energy difference between the two Saraste conformations decreased to -0.4 ± 0.15 kcal·mol⁻¹, consistent with the notion that the tightly packed Saraste dimer conformation is less stable in the presence of a membrane with 1.5% PIP₃ content.

PIP₃ Binding at the Canonical Site and the Peripheral Site Rigidified the Saraste Dimer Interface in an Individual PH–TH Module. Our simulations point to important roles for the canonical site and the peripheral site in controlling Saraste dimer formation, but the precise mechanisms by which PIP₃ binding strengthens Saraste dimer stability are not clear. After the binding of PIP₃ at the

two sites, we observed no obvious structural change in the PH– TH module that might lead to a more stable dimer conformation. We thus investigated the possibility that the binding of two PIP₃ molecules could restrict the range of conformations that the Saraste interface region can adopt.

To investigate this possibility, we studied the conformational dynamics of an individual PH–TH module under various conditions: In solution-based simulations, the rmsd from the Saraste crystal structure at the dimer interface ranged from 2.5 to 6.5 Å, with a peak distribution at ~4.8 Å (Fig. 7*A*). When the PH–TH module was recruited onto a membrane with only one PIP₃ lipid occupying the canonical binding site, the fluctuations at the Saraste dimer interface decreased, with the distribution peak shifting from 4.8 to 4.2 Å rmsd (Fig. 7*A*). In the presence of two PIP₃ lipids, the peripheral site was also occupied, and the peak distribution decreased further, to 3.5 Å (Fig. 7*A*). These observations indicate that the conformational dynamics of the PH–TH module are more restrained when both the canonical site and the peripheral site are occupied by PIP₃.

At High Concentrations, PIP₃ Bound at a Bridging Site, Where It Interacted with both PH-TH Modules in a Saraste Dimer. In the presence of membranes with 6% PIP₃ content, the canonical and peripheral sites of the Saraste dimer were mostly occupied in our simulations. Notably, at this concentration, we found that the area between the tip region of the β 3– β 4 hairpin of one PH-TH module and the edge of the canonical binding site of the other module also bound a PIP₃ molecule (Fig. 7*B*). In an earlier section, we suggested that PIP₃ binding at this bridging site might explain the activating effect of the E41K mutant; in the presence of a membrane with 6% PIP₃ content, the bridging site was transiently occupied by a PIP₃ lipid even in the absence of this mutation. In one representative structure extracted from our simulations, the phosphates at positions 1 and 4 of the inositol



Fig. 7. Speculation on the mechanisms by which binding multiple PIP₃ molecules stabilizes the Saraste dimer on the membrane. (A) Fluctuations at the Saraste dimer interface region in an individual PH–TH module in conventional MD simulations in solution and on the membrane. (B) An instantaneous structure from a tempered binding simulation showing that a PIP₃ lipid can simultaneously interact with both modules in the Saraste dimer at the bridging site, which formed near the dimer interface.

ring of PIP₃ interacted simultaneously with the Arg-46 residue of one module and Lys-27 of the other, which we label modules A and B, respectively (Fig. 7*B*). The hydroxyl group at position 2 also occasionally formed a hydrogen bond with Glu-45 of module A. The average occupancy rate of the bridging site was low, at ~40% when the membrane had 6% PIP₃ content, suggesting that PIP₃ binding at this site is of low affinity. We speculate that the bridging site might contribute to Saraste dimer stability only in the presence of a large number of PIP₃ lipids.

The two mechanisms we have described-stabilization by PIP₃ of the individual PH-TH modules in the Saraste dimercompatible conformation and, at high concentrations, the interaction of PIP₃ with both modules simultaneously to stabilize the dimer once it has formed-together support the notion that PIP₃ can allosterically stabilize the Saraste dimer. Our simulations also showed that the phosphate group at the position 3' of the inositol ring contributes to the binding of PIP₃ at both the bridging site and the peripheral site (Figs. 1C and 4C), supporting the idea that PIP₃ is a stronger binder at these alternative sites than its precursor molecule PIP₂. (Consistent with this structural observation, the binding of PIP₂ in the alternative sites is less stable than that of PIP₃ in our simulations; SI Appendix, Text T6.) Because the concentration of PIP₂ in the inner leaflet of the plasma membrane is generally thought to be much higher than that of PIP₃, however, it is possible that PIP₂ may also bind at these alternative sites.

The B cell plasma membrane is composed of many lipids-such as cholesterols, PE, PS, and sphingolipids-in addition to the PIP₃ and PC lipids studied in this work. These differences in membrane composition mean that our results concerning binding at different PIP₃ concentrations should be viewed in a qualitative manner. Little is known about the actual local concentration of PIP₃ near B cell receptors, where Btk primarily functions. Although the overall concentration of PIP₃ is found to be <1% in B cells, it is also known that PIP₃ can form mobile clusters or be found in small lipid domains. PIP₂ has been found in protein microdomains with a local concentration as high as 82% (47). PIP₃ has also been found in protein microdomains, with a local concentration up to 16% (48). In our simulations, the concentration of PIP₃ was 1-6%, falling within the range of potential PIP₃ concentration in cells. It is thus reasonable to expect that both mechanisms observed in our simulations may occur in nature.

Discussion

A Structural Model of Btk Activation on Membranes. Based on our atomic-level MD simulations, we have presented two important findings, which together provide a potential mechanism for Btk activation (*SI Appendix*, Fig. S7). First, we observed that Btk PH–TH modules spontaneously dimerized on membranes into a single predominant conformation, one which resembles the Saraste dimer. This finding provides a structural explanation for the effects of mutations in the PH–TH module known to cause severe Btk dysfunction, as these mutations affected the stability of the dimer.

The other major finding we have presented in this work is that PIP₃ can allosterically stabilize the Saraste dimer interface, and may thus play a regulatory role in Btk activation. In prior models of Btk regulation, PIP₃ has been considered a membrane anchor for the kinase, but no further regulatory role for PIP₃ has previously been proposed. Our simulations provide evidence that PIP₃ controls Saraste dimer stability, which may in turn have a profound impact on Btk activation: In order for Btk to be activated by transautophosphorylation, the two kinases need to come close together and form a transient enzyme-substrate complex. At a single-molecule level, the lifetime of the enzymesubstrate complex ought to be longer than the time required for transferring the phosphate from ATP to the substrate tyrosine. Increasing the lifetime of the Saraste dimer would thus make the transautophosphorylation reaction more efficient. [Although for this work we have not studied in detail whether PIP₃ binding affects the on rate of PH–TH dimer formation, which would also promote Btk activation, there is some evidence from our simulations that tends to support this proposition (*SI Appendix*, Texts T4 and T5).]

PH domains, which act as membrane-binding modules and protein-protein interaction modules, are frequently found in the human genome (49, 50). Dimerization of membrane-bound PH domains has received little attention in the past, but PH domain dimerization may play an important role in enzyme regulation. Recently, the PH domain of phosphoinositide-dependent kinase-1 (PDK1), a serine/threonine kinase that is important for cell differentiation and proliferation pathways, was shown to dimerize on a supported bilayer system (51). The structural basis for this dimerization is not clear, but it has been speculated that the PDK1 dimer represents an inhibitory state of the kinase that is not capable of substrate binding (52). This notion, together with the activating role of the PH-TH module of Btk proposed in this paper, leads us to speculate that PH domain dimerization might be a common regulatory mechanism of PH domain-containing enzymes that function on membranes.

Some receptor tyrosine kinases, such as epidermal growth factor receptor (EGFR), self-associate into high-order oligomers as part of the activation process (53, 54). We have not studied whether dimers of the PH–TH module might form high-order oligomers on membranes, but we note that such high-order oligomers, if they exist, could further strengthen the stability of the Saraste dimer interface and contribute to the activation of Btk.

The Saraste Dimer Interface and the Peripheral Binding Site in Btk Are Less Conserved in Early Stages of B Cell Evolution. Neither the Saraste dimer interface nor the peripheral binding site in Btk is conserved in the PH–TH modules of Tec and Itk, the other two members of the Tec family of kinases. This raises the question of when the PIP₃-dependent dimerization of the PH–TH module of Btk originated.

We examined the conservation of those residues that constitute the Saraste dimer interface and the peripheral binding site, comparing the sequence of human Btk with that of other mammals (105 species), birds (47 species), and fish (41 species). We found that these residues are for the most part well conserved in mammals and birds, and much less extensively conserved in fish (SI Appendix, Fig. S8). In contrast, PI3K (the enzyme responsible for generating PIP₃) and residues in the canonical binding site of Btk are conserved in all eukaryotes. This analysis suggests that membrane binding is a more ancient function than allosteric binding for PIP₃ in its interactions with the PH domain. The origins of the adaptive immune system, in particular B cells, are thought to be linked to the emergence of jawed fish, such as Chondrichthyes and Teleost fish (55, 56). B cells in fish have antigen-specific IgM responses similar to those in mammals, but the time required for fish B cells to generate a significant antigen-specific response is generally much longer than that required in mammals (56), suggesting that the signal transduction pathways that control the antigenspecific response, in which Btk plays an important regulatory role, are much less sensitive in fish than in mammals. One possibility is that the utilization of a PIP₃-dependent dimerization mechanism for Btk activation in mammals is a consequence of natural selection for high sensitivity of the B cell response.

Alternative Strategies for Developing Btk Inhibitors That Are Selective and Can Inhibit the Ibrutinib-Resistant C4815 Mutant. A number of ATP-competitive Btk inhibitors have been identified in recent years (57, 58). The most clinically tested of these is ibrutinib (59), which has been approved by the Food and Drug Administration for the treatment of mantle cell lymphoma (in 2013), chronic lymphocytic leukemia (in 2014), and a particular form of non-Hodgkin lymphoma (in 2015). Despite ibrutinib's high efficacy in the treatment of multiple B cell cancers, ibrutinib-resistant mutations—in particular, C481S—have emerged in a substantial fraction of all chronic lymphocytic leukemia patients treated with ibrutinib (60). In addition, ibrutinib has off-target effects on EGFR, ITK, and Tec-family kinases, which may result in various adverse effects in some patients.

The emerging resistance to and off-target side effects of ibrutinib have led to the active development of second-generation and more specific Btk inhibitors. Our finding that PIP₃-dependent dimerization may be a feature of Btk activation suggests a potential alternative approach to the development of such inhibitors in which small molecules disrupt the dimerization of PH–TH modules. Such molecules could potentially bind either at the Saraste dimer interface or in other regions on the PH–TH module that could allosterically impair Saraste dimer formation. Inhibitors targeting the PH–TH module should act on both wild-type Btk and the C481S mutant, and should not interfere with functions of other kinases, thus addressing both of these ibrutinib-related shortcomings at the same time.

Methods

Detailed methods can be found in *SI Appendix*. A brief summary is provided here.

General Simulation Details. All production simulations were run on Anton (61), a specialized machine for MD simulations, using the Amber ff99SB-ILDN force field for proteins (62–64), the CHARMM TIP3P model for water (65), and the CHARMM36 force field for POPC lipids (66). The initial parameters of soluble inositol phosphates and PIP₃ lipids were generated with the generalized Amber force field (67), with the atomic charges refit by restrained electrostatic potential fitting. The system was neutralized and salted with NaCl, with a final concentration of 0.15 M. The system was equilibrated in the NPT ensemble for 100 ns using gDesmond (68) on a commodity GPU cluster. Production runs were subsequently performed in the NVT ensemble from the final frame of the NPT relaxation simulation (69) by coupling the system to a variant (70) of the Nosé–Hoover thermostat (71) at 310 K with a relaxation time of 1 ps. A RESPA (72) integrator was used, with an inner time step of 2.5 fs. The long-range electrostatic forces were calculated in *k*-space using a grid-based method with Gaussian spreading to the grid every 7.5 fs (73).

Tempered Binding Simulations. To enhance the sampling in our PH–TH module encounter simulations, we used an approach called "tempered binding," (32) in which the system's Hamiltonian (energy function) is tempered as opposed to its temperature, as is done in conventional tempering methods. Tempered binding dynamically scales various atomic interactions during an MD simulation by a factor, λ , that is updated among a ladder of discrete values, λ_i . The atomic interactions are unscaled at the lowest rung of the ladder (rung 0). The structures sampled at rung 0 are sampled from the same Boltzmann distribution as a conventional MD simulation. Our tempered binding simulations linearly scaled the near electrostatic interactions between the positively charged atoms on one protein and the negatively charged atoms on the other protein, and vice versa (interactions within a protein, and between protein and water, were not scaled). We varied the maximum tempering strength (i.e., the scale factor reached at the highest rung; the scale factor at the lowest rung is 1) from 0.991 to 0.997, and found

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that the efficiency of dimer formation was not sensitive to the maximum tempering strength in this range.

Clustering of Dimers Formed on Membranes. All frames from all replicas of 24 runs (for a total of 6.55 ms of simulation time) were extracted. Protein structures from these frames were clustered using hierarchical density-based spatial clustering of applications with noise (74). In total, there were 5,571,555 such structures. Frames in which there was no dimer structure were removed. The filtered dataset contained n = 4,379,672 structures. A total of five clusters were identified, with 1,132,602 (26%) of the points analyzed categorized as noise. The primary cluster contained 2,679,781 points (61%). The structures for each cluster's best approximate representation vector were identified and visualized.

Calculation of Interface Rmsd. The rmsd values at the Saraste interface were calculated using all of the nonhydrogen atoms in residues at the dimer interface (IIe-9, Leu-29, Tyr-42, Phe-44, IIe-92, IIe-95). The positions of these residues in the crystal structure of the PH–TH dimer [Protein Data Bank (PDB) ID code 1BTK] were used as the reference.

Equilibrium Simulations Between the Saraste Conformation and the Pre-Saraste Conformation. The population fractions of the tightly packed Saraste conformation and of the pre-Saraste conformation were calculated as follows: First, the rmsd values for all frames at rung 0 were calculated with respect to the crystal structure, as in the other analyses. Only frames with rmsd <7 Å were considered. We classified frames with rmsd <3.1 Å as being in the tightly packed conformation, and frames with rmsd >3.1 Å (and <7 Å) as being in the pre-Saraste conformation. The percentage of the population for each conformation was then calculated as the fraction of the total frames associated with that conformation. Simulation trajectories with more than 10 transition events between the Saraste and pre-Saraste conformation were used for the free energy calculation. The final free energy value reported was averaged from five (for the 8-PIP₃ condition) and three (for the 2-PIP₃ condition) independent simulations. The total numbers of transition events for the 8-PIP₃ and 2-PIP₃ conditions were 86 and 81, respectively.

Bioinformatics Analysis. The Btk sequences used for evolution analysis were compiled using a series of protein–protein BLAST searches in the National Center for Biotechnology Information nonredundant protein database, using the human Btk PH–TH module (residues 1–172) as the reference (75, 76). The initial search generated 772 sequences with a minimal sequence identity of 48%. Btk sequences of mammals, birds, and fish were then extracted from the initial search results based on taxonomy IDs. The sequence pool was then filtered based on the following rules: (*i*) duplicate entries were removed; (*iii*) sequences labeled as "partial" or "synthetic" were removed; (*iii*) sequences used for multiple-sequence alignment was 334. The multiple-sequence alignment of the 334 sequence logos were generated using the online tool WebLogo3 (79, 80). The final figures were manually adjusted from the WebLogo3 result to show only the residues at the sites of interest.

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