Expanding the Repertoire of Amyloid Polymorphs by Co-polymerization of Related Protein Precursors*

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Background: Amyloid fibrils *in vivo* are rarely composed of a single protein, yet the consequences of co-polymerization of different proteins are relatively poorly understood.

Results: Fibrils formed by co-polymerizing two variants of β_2 -microglobulin were characterized alongside their homopolymer equivalents.

Conclusion: The three fibril types have different structural and thermodynamic properties.

Significance: Co-polymerization of protein precursors enhances the structural and thermodynamic diversity of amyloid fibrils.

Amyloid fibrils can be generated from proteins with diverse sequences and folds. Although amyloid fibrils assembled in vitro commonly involve a single protein precursor, fibrils formed in vivo can contain more than one protein sequence. How fibril structure and stability differ in fibrils composed of single proteins (homopolymeric fibrils) from those generated by co-polymerization of more than one protein sequence (heteropolymeric fibrils) is poorly understood. Here we compare the structure and stability of homo and heteropolymeric fibrils formed from human β_2 -microglobulin and its truncated variant Δ N6. We use an array of approaches (limited proteolysis, magic angle spinning NMR, Fourier transform infrared spectroscopy, and fluorescence) combined with measurements of thermodynamic stability to characterize the different fibril types. The results reveal fibrils with different structural properties, different side-chain packing, and strikingly different stabilities. These findings demonstrate how co-polymerization of related precursor sequences can expand the repertoire of structural and thermodynamic polymorphism in amyloid fibrils to an extent that is greater than that obtained by polymerization of a single precursor alone.

Amyloid fibrils are formed by the self-assembly of natively unfolded proteins and peptides such as $A\beta_{40/42}$ in Alzheimer disease (1), α -synuclein in Parkinson disease (2), and islet amyloid polypeptide in type II diabetes mellitus (3). In addition, self-assembly of folded proteins with all- α , all- β , or mixed α/β

of fibrils in the disorders dialysis-related amyloidosis (4) and hereditary systemic amyloidosis (5).
Despite the different conformational properties of amyloidogenic precursors, the fibrils that they form share common and the statement of the statemen

structures are all involved in human amyloidosis. These classes

of proteins include β_2 -microglobulin (β_2 m), the all- β precursor

genic precursors, the fibrils that they form share common structural characteristics: typically a long, straight, unbranched morphology and a cross- β architecture (6). Recent analyses of amyloid fibrils using MAS² NMR (7-10) and x-ray diffraction of crystals formed from short (4-7 residue) amyloidogenic peptides have revealed an array of structural architectures that conform to the cross- β fold (11). For some proteins/peptides the same amino acid sequence can form conformationally distinct amyloid structures by varying the growth conditions, revealing the polymorphism possible for an identical protein sequence (reviewed in Ref. 12). In other cases structural variations of the cross- β fold occur as metastable species during fibril assembly (9). Further complexity could arise by the conformational properties of the monomeric precursor (whether folded, partially folded, or disordered) influencing the fibril structure formed (13) or by the co-polymerization of related sequences into heteropolymeric fibrils (14–16).

The clinically important protein, human β_2 -microglobulin (h β_2 m), and its truncated variant, Δ N6, offer an opportunity to investigate the role of sequence and precursor conformation in amyloid polymorphism. h β_2 m is a 99-residue protein that has a seven β -stranded immunoglobulin fold (17). In the absence of additives such as Cu²⁺, detergents, trifluoroethanol, lipids, collagen, or glycosaminoglycans, h β_2 m is not able to form amyloid fibrils *in vitro* at neutral pH (for review, see Ref. 18). Instead, the amyloid potential of h β_2 m is unfurled only by unfolding the



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² The abbreviations used are: MAS, magic angle spinning; hβ₂m, human β₂-microglobulin; ANS, 8-anilino naphthalene sulfonate; HFIP, hexafluoroisopropanol; RFDR, radio frequency-driven recoupling; ZF TEDOR, Z-filtered transferred-echo double resonance; ThT, thioflavin T; TAMRA, 5(6)-carboxytetramethylrhodamine succinimidyl ester; GuHCl, guanidinium chloride; ESl, electrospray ionization.

protein, for example by acidification to pH 2 (19, 20). The fibrils formed under these conditions have been characterized in detail using MAS NMR (10), EPR (21), FTIR (22), limited proteolysis (23), and cryo-electron microscopy (EM) (24). These results have revealed that the fibrils formed from h β_2 m at pH 2 are composed of parallel, in-register β -strands that involve 90 of the 99 residues in the fibril core, the nine N-terminal residues retaining a dynamic conformation that is not integral to the fibril structure (10).

By contrast with the intransigence of $h\beta_2m$ to form amyloidlike fibrils at neutral pH, a natural variant of $h\beta_2m$ that is truncated by six residues at its N terminus (Δ N6) is able to form amyloid-like fibrils at pH 6–7 *in vitro* in the absence of additives (25). This truncation is the major modification of $h\beta_2m$ found in *ex vivo* fibrils (26). Despite truncation of the N-terminal six residues, Δ N6 displays only minor structural differences compared with $h\beta_2m$ in the native form (25). Although the structural properties of Δ N6 cannot explain its enhanced ability to form amyloid fibrils at neutral pH, increased conformational dynamics evidenced by NMR relaxation times (T_2 values) (25), hydrogen exchange protection (27–29), molecular dynamics simulations (30), and denaturation with guanidinium chloride (GuHCl) (31) have been linked to its ability to form fibrils at this pH.

In this study we examine how the amyloid fibrils formed from folded $\Delta N6$ at pH 6.2 differ from those assembled by acid unfolded h β_2 m at pH 2. We characterize structural and thermodynamic differences between these two fibril types using MAS NMR, limited proteolysis with mass spectrometry, and spectroscopic measurements (FTIR, fluorescence, and ANS binding). Building on previous experiments which have shown that substoichiometric ratios of $\Delta N6$ are able to convert h $\beta_2 m$ into an amyloidogenic form at neutral pH (25), we examine how fibrils formed by co-incubation of these two proteins at pH 6.2 differ from those formed from each protein alone. The results reveal that the fibrils formed under each condition show different structural properties and side-chain packing and striking differences in their thermodynamic properties. The findings highlight the diversity of amyloid architectures that is possible for a given protein sequence and demonstrate how fibril polymorphism can be enhanced by the co-polymerization of proteins of related sequence.

EXPERIMENTAL PROCEDURES

Protein Preparation—h $β_2$ m and ΔN6 were produced as previously described (25). For NMR experiments ¹⁵N and ¹³C and ¹⁵N-labeled ΔN6 was prepared as described in Ref. 10.

Solution NMR Spectroscopy—Samples of 15 N-labeled protein (1 mg/ml) in either 50 mM MES, 120 mM NaCl, pH 6.2, or 10 mM sodium phosphate buffer, 50 mM NaCl, pH 2, 90% (v/v) H₂O, 10% (v/v) D₂O were used for solution NMR experiments. Spectra were recorded at 25 °C on a Varian Inova 750 MHz spectrometer.

Assembly of Amyloid Fibrils— Δ N6 fibrils and the mixed fibril sample were assembled in 50 mM MES buffer, 120 mM NaCl at pH 6.2. The mixed fibril sample was formed from a 1:1 molar ratio of h β_2 m: Δ N6 monomers. h β_2 m fibrils were formed in 10 mM sodium phosphate buffer containing 50 mM NaCl, pH 2.0. Assembly usually began with 1 mg/ml soluble protein. Fibril growth was performed in a BMG Fluostar Optima plate reader at 37 °C at 600 rpm. A final concentration of 10 μ M thioflavin T (ThT) was added where appropriate. Fibrils were left to assemble for ~5 days before analysis. Fibrillar h β_2 m for MAS NMR was formed at pH 2.5, as described in Ref. 10.

Detection of the Presence of an Intact Disulfide Bridge in $\Delta N6$ *Fibrils*— Δ N6 fibrils (60 µl of 80 µM) were centrifuged at 14,000 \times g for 20 min. The pellet was resuspended in hexafluoroisopropanol (HFIP), divided into three, and incubated overnight at 37 °C with gentle rotation (200 rpm), then air-dried. The first aliquot had no further treatment (control sample). 20 μ l of 20 mM iodoacetamide in 50 mM ammonium bicarbonate, pH 7, was added to sample two (alkylated sample). This sample was then incubated in the dark at room temperature for 30 min. The third aliquot (reduced alkylated sample) was resuspended in 20 μ l of 10 mM dithiothreitol in 50 mM ammonium bicarbonate, pH 7, and heated to 80 °C for 15 min. The sample was then cooled for 5 min at 4 °C and centrifuged at 14,000 \times *g* for 20 min, and 20 µl of 20 mM iodoacetamide added to the supernatant. This sample was then incubated in the dark at room temperature for 30 min. Samples were analyzed by Z-spray nanoelectrospray ionization mass spectrometry.

MAS NMR— $h\beta_2m$ and $\Delta N6$ -hydrated fibrils (35 and 45 mg, respectively) were collected by centrifugation (265,000 × g) and packed into 3.2-mm Bruker zirconia rotors. Solid-state NMR experiments were conducted at 277 K on a Bruker 900 MHz spectrometer and a custom designed 750 MHz spectrometer (courtesy of Dr. David J. Ruben, Francis Bitter Magnet Laboratory, Cambridge, MA).

Two kinds of MAS NMR techniques, RFDR and ZF TEDOR, were utilized to establish one-bond ¹³C-¹³C and ¹³C-¹⁵N correlations, respectively (32–34). RFDR spectra were acquired at 20-kHz MAS on a 900-MHz spectrometer. The ¹³C-¹³C dipolar coupling was recoupled in the rotor-synchronized RFDR mixing period during which 12.5- μ s π pulses and 83.3-kHz CW decoupling were applied on the ¹³C and ¹H channels, respectively. A total RFDR mixing time of 1.6 ms was used to realize one-bond ¹³C-¹³C correlations. One-bond ZF TEDOR experiments were conducted on a 750-MHz spectrometer and under 12.5-kHz sample spinning, with a total dipolar recoupling time of 1.6 ms and ¹H TPPM decoupling at 95 kHz during mixing and 83 kHz during acquisition.

Fluorescent Labeling and Confocal Imaging of $h\beta_2m$ and $\Delta N6$ Fibrils—A 10-fold molar excess of 5(6)-carboxytetramethylrhodamine succinimidyl ester (TAMRA) (Invitrogen) was titrated into monomeric $h\beta 2m$, and a 10-fold molar excess of fluorescein-5-isothiocyanate (FITC) (Molecular Probes) was titrated into monomeric $\Delta N6$. Labeling was allowed to continue for 45 min. Fluorescently labeled monomers of each protein were then purified (PD10 desalting column), and fibrils were formed by mixing these samples as described above at a 1:10 molar ratio of fluorescently labeled protein to each unlabeled protein (34). Confocal images were captured on a DeltaVision Deconvolution Microscope. Colocalization analysis was performed using Image J. At each pixel location the contributing intensity from both channels was assessed, and a scatter graph was plotted.

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FIGURE 1. Fibrils formed from h β_2 m and $\Delta N6.A$, shown is a ¹H,¹⁵N HSQC spectrum of $\Delta N6$ at pH 6.2. *B*, shown is fibril formation of 0.5 mg/ml $\Delta N6$ (black lines) and h β_2 m (gray lines, no growth) at pH 6.2 measured using thioflavin T fluorescence (relative fluorescence units (*rfu*)). Three replicates for each protein are shown. The *inset* shows an SDS-polyacrylamide gel of the supernatant of the $\Delta N6$ sample after an incubation time of 120 h (*lane i*) and before fibril growth (*lane ii*). *C*, shown are negative stain EM images of $\Delta N6$ fibrils. *Inset i* shows an expanded view, and *inset ii* shows the absence of h β_2 m fibrils under the same conditions (*scale bar* = 100 nm). *D*–*F* are as in *A*–*C*, but for h β_2 m at pH 2.0 in 10 mM sodium phosphate, 50 mM sodium chloride.

Limited Proteolysis—Proteinases (chymotrypsin or aspergillopepsin I (Sigma)) were added at 1:100 (w/w) proteinase to protein ratios, and proteolysis was allowed to proceed for 30 min at 25 °C. Fibrillar samples were depolymerized after digestion in 100% (v/v) HFIP. Samples were air-dried then redissolved in 50:40:10 acetonitrile/water/acetic acid (v/v/v), and peptides were identified by infusing the sample into a Synapt HDMS (Micromass UK Ltd/Waters Corp., Manchester, UK) quadrupole-traveling wave IMS-oaTOF mass spectrometer.

Fourier Transform Infrared Spectroscopy—Monomeric proteins (2.5 mg/ml) were exchanged into D_2O . Fibrils were prepared as described above, except that the buffers were prepared using D_2O at the appropriate pD. Spectra were acquired on a Thermo-Nicolet 560 FTIR spectrometer.

Dot Blots—Dot blots using WO1 (35) and polyclonal anti- β_2 m antibodies (Dako) were performed according to Xue *et al.* (36).

Intrinsic Fluorescence and 8-Anilino Naphthalene Fluorescence Measurements—The fluorescence of 2.5 μ M monomer or fibrils was excited at 280 nm, and fluorescence emission was measured between 300 and 390 nm. The fluorescence of each sample was also measured in the presence of 250 μ M ANS to 1 μ M fibrils (monomer equivalent concentration). Excitation was at 389 nm. Fluorescence was measured using a Photon Technology International QM-1 spectrofluorimeter (PTI).

Determination of Fibril Stability—Fibrils (0.2 mg/ml) were diluted into different concentrations of GuHCl in the buffer in which each sample was prepared based on Shammas *et al.* (37). Solutions were incubated for 1.5 h at 25 °C then centrifuged in a Beckman ultracentrifuge at 313,000 × g for 45 min. The protein concentration of the supernatant was determined by the

absorbance at 280 nm using an extinction coefficient of 20065 $\rm cm^{-1}~\rm M^{-1}$ for both $\beta_2 m$ and $\Delta N6.$

RESULTS

Homopolymeric Assembly of $\Delta N6$ and Wild-type h $\beta_2 m$ into Amyloid-like Fibrils-Previous experiments have shown that the kinetics of Δ N6 fibrillation depend critically on the solution pH, with an enhanced rate of fibril formation occurring as the pH is lowered from pH 8.2 to pH 6.2 (25). To form fibrils from $\Delta N6$ under conditions in which the protein is initially folded but is able to assemble into amyloid-like fibrils rapidly, the conditions of fibril growth (pH, temperature, buffer ionic strength, and agitation rate) were varied. Here and throughout, ThT fluorescence was used to monitor the rate of fibril growth. Fibril yield and morphology were determined by estimation of the amount of unpolymerized monomer in the supernatants using SDS-PAGE and by negative stain transmission electron microscopy of the fibril samples. Having screened several different conditions, fibrils of Δ N6 were ultimately formed by incubation of 0.5 mg/ml protein in 50 mM MES, 120 mM NaCl (150 mM total ionic strength), pH 6.2, 37 °C, with agitation of 600 rpm in 96-well plates. Under these conditions $\Delta N6$ is natively folded as judged by NMR (Fig. 1A), and fibrils form within 48 h with a vield of >98% (Fig. 1B, black traces and inset) without visible formation of amorphous aggregates (Fig. 1C and inset i).

By contrast with the rapid formation of amyloid-like fibrils by $\Delta N6$ at pH 6.2, amyloid-like fibrils are not formed from native $h\beta_2m$ at pH 6.2, as judged by the same techniques (Fig. 1*B*, *solid gray line*, and *C*, *inset ii*). Acidification of $h\beta_2m$ to pH 2.0 results in a highly unfolded species (Fig. 1*D*) and renders the protein





FIGURE 2. **ESI-MS spectra reveal that the disulfide bond is intact in ΔN6 fibrils.** *A*, shown are monomers of ΔN6 released from fibrils formed at pH 6.2 by treatment with HFIP. *B* is as *A*, but the sample was treated with a 20-fold molar excess of iodoacetamide. *, results from derivatization of methionine (11, 194 Da) and a subsequent loss of the carboxyamido group (marked with \$) (62). *C* is as *B*, but the sample was treated with DTT followed by the addition of a ~20-fold molar excess (over the total thiol concentration) of iodoacetamide.

readily able to form amyloid-like fibrils with \sim 90% yield (38) (Fig. 1, *E* and *F*).

Previous results have shown that reduction of the single disulfide bond in h β_2 m enhances its fibrillogenic potential and that disulfide bond interchange can initiate $h\beta_2m$ fibril formation (39). To determine whether the disulfide bond linking residues 25–80 in the Δ N6 monomer is intact in the fibrils formed from Δ N6 at pH 6.2, the fibrils were disassembled by incubation with HFIP, and the status of the disulfide bond was determined using chemical modification with iodoacetamide, monitored using ESI-MS ("Experimental Procedures"). The results of these experiments (Fig. 2) showed that monomers released from the $\Delta N6$ fibrils in the absence or presence of iodoacetamide have a mass $11,137 \pm 1.14$ Da (Fig. 2, A and B), consistent with that expected for unmodified $\Delta N6$ (11,137 Da). $\Delta N6$ monomers released from fibrils treated with DTT and incubated with iodoacetamide resulted in a mixture of species (Fig. 2*C*): reduced, unalkylated protein (11,140 Da); alkylation of a single cysteine (11,196.9 Da); alkylation of both cysteines (11,253.8 Da). This demonstrates that the majority of monomers retain the disulfide linkage in the $\Delta N6$ homopolymeric fibrils.

Structural Analysis of Fibrils Formed from $\Delta N6$ and $h\beta_2m$ Using Solid State NMR—Our previous MAS NMR experiments have studied fibrils formed from $h\beta_2m$ at pH 2 (10). These studies identified a parallel-in-register intermolecular packing of the β -strands. The chemical shift analysis suggested that the β -strands within the fibril are distinct from those within native



FIGURE 3. MAS NMR spectra of uniformly ¹³C, ¹⁵N-labeled h β_2 m (*blue*) and **ΔN6** (*red*) fibrils. *A*, shown are one-bond ¹³C-¹³C correlations from a RFDR experiment. *B*, shown are backbone N α -C α correlations obtained with ZF TEDOR.

 $h\beta_2 m$ (25). Furthermore the MAS NMR experiments demonstrated that \sim 70% of the $h\beta_2 m$ protein sequence participates in β -strands within the rigid fibril core of the full-length protein.

To determine whether the fibrils formed from $\Delta N6$ and $h\beta_2m$ share structural homology at the residue-specific level, MAS NMR studies of the homopolymeric fibrils formed by Δ N6 at pH 6.2 were performed. Fig. 3A presents ¹³C-¹³C spectra of uniformly ${}^{13}C$, ${}^{15}N$ -labeled $\Delta N6$ fibrils formed at pH 6.2 (red) and uniformly ${}^{13}C$, ${}^{15}N$ -labeled h β_2 m fibrils formed at pH 2.5 (blue). Fig. 3B shows the two-dimensional ¹⁵N-¹³C correlation spectra of each sample recorded with ZF TEDOR mixing of 1.6 ms, showing mostly backbone N-C α correlations. The line width of cross-peaks in the Δ N6 spectra is 0.4–0.6 ppm for ¹³C and 0.8 - 1.2 ppm for ¹⁵N, comparable to peaks of h β_2 m fibrils in spectra acquired with the same acquisition parameters, suggesting similar structural homogeneity. No peak multiplicity was observed for either sample, with a single set of backbone ¹³C and ¹⁵N chemical shifts for all residues in these spectra, ruling out the possibility of polymorphism within each sample.

Compared with the spectra of $h\beta_2$ m fibrils, the spectra of $\Delta N6$ fibrils contain more backbone N α -C α and N α -CO crosspeaks for resolved amino acid types (Fig. 3*B*). For example,



three glycine cross-peaks were observed in spectra of the $\Delta N6$ fibrils compared with only one glycine cross-peak in spectra of the h β_2 m fibrils (*circled* in Fig. 3*B*). These data suggest that $\Delta N6$ fibrils are less dynamic than h β_2 m fibrils and hints that $\Delta N6$ fibrils may possess a more extensive β -sheet core than their wild-type counterparts. Consistent with this, the MAS INEPT spectrum of $\Delta N6$ fibrils contains only a few weak peaks (data not shown), suggesting that there are no regions that experience significant mobility in this truncated version of the protein. This is in contrast to fibrils formed from h β_2 m at pH 2.5 that showed significant dynamics for residues within the N-terminal 7 residues (10).

Further inspection of the spectra in Fig. 3 reveals that the majority of the cross-peaks are different and shifted from each other in the fibrils of $\Delta N6$ and $h\beta_2 m$, suggesting distinct secondary structures. Some of the differences might arise from the difference in pH (6.2 versus 2.5), especially for sites that participate in hydrogen bonding, such as protonated side chains. However, such effects cannot explain the global changes observed in the chemical shifts. Taking the TEDOR spectrum for example (Fig. 3*B*), the three glycine residues in the $\Delta N6$ spectrum (Gly-18, -29, and -43, circled in Fig. 3B) show clearly different chemical shifts to those of $h\beta_2m$ fibrils. Similarly, the differences in Ser and Thr C α -C β correlations (*enlarged* and circled in Fig. 3A) are on the order of 2.5-4.0 ppm, too large to be attributed to the effect of pH alone (40). These observations suggest that there are significant differences in the molecular conformations of the proteins in the fibrils formed from $h\beta_2 m$ and $\Delta N6$. Further analysis, including residue-specific assignment, will be needed to define these differences in more detail.

Formation of Mixed $\Delta N6:h\beta_2m$ Fibrils—Previous studies have shown that monomeric $\Delta N6$ is able to convert h $\beta_2 m$ into a conformation able to form amyloid fibrils at neutral pH. Quantitative incorporation of $h\beta_2 m$ monomers into amyloid fibrils occurred when mixed with equimolar $\Delta N6$ monomer at pH 6.2–7.2 (25). To further characterize the heteropolymeric fibrils formed by mixing monomeric $h\beta_2 m$ and $\Delta N6$, the two proteins were incubated separately or in an equimolar mixture at pH 6.2, and the formation of fibrils was monitored using ThT fluorescence (Fig. 4*A*). The results showed that $h\beta_2 m$ alone is not able to form fibrils at pH 6.2 under the conditions employed (60 and 120 μ M protein monomer, shown as *solid gray lines*), as confirmed by EM (Fig. 4*B*). In comparison, Δ N6 rapidly formed fibrils under these conditions (Fig. 4A, black solid and dashed lines). By contrast with previous results (25), under the conditions employed here, the rate of fibril growth decreases as the concentration of $\Delta N6$ is increased from 60 μ M to 120 μ M, suggestive of a complex assembly reaction, involving the formation of off-pathway oligomers (Fig. 4A). Interestingly the mixed sample, which contained 60 μ M concentrations of both Δ N6 and $h\beta_2 m$ monomers, formed fibrils at a rate similar to that of 120 μ M Δ N6 alone (Fig. 4*A*, gray dotted lines), consistent with co-polymerization of $\Delta N6$ and $h\beta_2 m$ during fibril assembly. The kinetics of fibril formation monitored using ThT fluorescence suggest that co-polymerization of h β_2 m and $\Delta N6$ does not arise from $\Delta N6$ seeding h β_2 m, as this would result in a lag phase similar (~20 h), if not shorter, than that of 60 μ M Δ N6 incubated alone. Instead, fibril formation is not observed in the

mixed sample until \sim 40 h of incubation. Transmission electron microscopy of the fibrils formed in the mixed sample (Fig. 4*C* and *inset*) confirmed the presence of fibrils, which have a long straight unbranched morphology.

To determine whether co-incubation of $h\beta_2m$ and $\Delta N6$ resulted in fibrils containing both monomers, the fibrils were collected by centrifugation, resolubilized in 100% HFIP, and analyzed by ESI-MS (Fig. 4*D*). The resulting spectra contained peaks arising from $h\beta_2m$ and $\Delta N6$ (masses 11,859 \pm 1.19 and 11,136 \pm 1.13 Da, respectively) with approximately equal intensity, suggesting that the protein monomers co-polymerize into fibrils with equal probability.

Finally, to confirm that both monomers are present in the same fibril, $h\beta_2m$ was labeled with TAMRA and $\Delta N6$ with FITC under conditions that modify a single lysine on average. Fibril formation of each monomeric sample and the mixed sample was then allowed to proceed for 96 h at pH 6.2. The homo- and hetero-polymeric fibrils formed ("Experimental Procedures") were then compared using confocal fluorescence microscopy. The resulting images (15–20 per sample) and colocalization plots (Fig. 4, *E*–*H*) show that in the mixed sample both labeled monomers assemble into a single fibril containing approximately equal amounts of each protein precursor. These results provide further evidence that $\Delta N6$ is able to convert $h\beta_2m$ into a conformation able to co-assemble with $\Delta N6$ to form heteropolymeric fibrils.

Limited Proteolysis of Different Fibril Polymorphs-We next compared the fibril cores of the three different fibril types. Previous studies using limited proteolysis combined with mass spectrometry (23) have shown that the N-terminal 9 residues of $h\beta_2$ m fibrils formed at pH 2.5 are accessible to pepsin cleavage, implying the 90 remaining residues are part of the fibril core. A different fibril polymorph formed from the same protein at pH 3.6 (known as "worm-like" fibrils) possesses a less extensive core involving residues 40-74 (23). To determine the extent of the cores in Δ N6 homopolymeric fibrils and in heteropolymeric fibrils, cleavage with chymotrypsin or aspergillopepsin I was performed. The former enzyme cleaves predominantly at aromatic residues, with a reduced propensity to cleave at leucine and methionine. Its optimal activity occurs at pH 8 (41). Because $h\beta_2 m$ fibrils formed at pH 2 dissociate at this pH, incubation with aspergillopepsin I was used to cleave $h\beta_2 m$ fibrils at pH 2. Aspergillopepsin I has a propensity to cleave at basic amino acids and is catalytically active between pH 1 and 6 (42). As a consequence, this proteinase was also used to cleave $\Delta N6$ fibrils and the mixed fibrils.

The cleavage products detected after digestion of $\Delta N6$ fibrils with chymotrypsin or aspergillopepsin I are shown diagrammatically in Fig. 5. Cleavage of $\Delta N6$ fibrils with both proteinases occurred close to the termini (Gln-8, Tyr-10, Leu-87, and Trp-95) (numbering according to the h β_2 m sequence), resulting in peptides encompassing amino acids 9–99, 11–95, 11–99, 7–87, and 7–95. No cleavage was observed between residues 10 and 87 despite the presence of many potential cleavage sites (potential chymotrypsin cleavage sites depicted by the *gray bar* in Fig. 5). The results suggest that in $\Delta N6$ fibrils residues 12–86 form the core.





FIGURE 4. **Characterization of fibrils formed from mixtures of** Δ **N6 and** $h\beta_2$ **m at pH 6.2.** *A*, shown are ThT fluorescence traces of Δ N6 at 60 μ M (*solid black*), 120 μ M (*dashed black*), and a 60:60 μ M mixture of Δ N6 and $h\beta_2$ m (*dashed gray*). Note $h\beta_2$ m incubated alone (60 and 120 μ M) does not form fibrils under these conditions (*solid gray*). The kinetic traces of three different replicates are shown for each sample. Shown are negative stain EM images of the end point of incubation of $h\beta_2$ m alone (*B*) and the Δ N6: $h\beta_2$ m mixed fibril sample (the *inset* shows a single fibril; *C*) both at pH 6.2; *scale bars* are 100 nm. *rfu*, relative fluorescence units. *D*, shown is an ESI mass spectrum of depolymerized fibrils formed from a 1:1 (mol/mol) mixture of $h\beta_2$ m formed at pH 2. FITC-labeled Δ N6 fibrils formed at pH 6.2 (G) and fibrils formed from a 1:1 (mol/mol) mixture of TAMRA- $h\beta_2$ m formed at pH 2. FITC-labeled Δ N6 fibrils formed at pH 6.2 (G) and fibrils formed from a 1:1 (mol/mol) mixture of TAMRA- $h\beta_2$ m monomers and FITC- Δ N6 (*H*) are shown. The *yellow color* shows the superposition of red and green fluorescence. *Scale bar* = 5 μ m. The scattergraphs depict co-localization plots of the contribution from the green (FITC fluorescence, *x* axis) and red (TAMRA fluorescence, *y* axis) channels for each pixel location. The *y2* axis is the intensity of the signal. The images are 8 bit, thus the x and y axes are from 0–256 pixels.

For comparison, monomeric Δ N6 was also cleaved with chymotrypsin. Cleavage sites were observed at Tyr-26, Leu-40, Trp-60, Tyr-66, and Lys-75 consistent with the NMR structure of Δ N6 (25), which reveals these residues are located in surface-exposed loops. Accordingly, peptides 7–60, 27–60, 40–60, 61–99, 67–99, and 76–99 are identified using ESI-MS and ESI-MS/MS (Fig. 5).

The chymotrypsin or aspergillopepsin I cleavage patterns for β_2 m/ Δ N6 heteropolymeric fibrils (Fig. 5) revealed that the core of these fibrils resembles that of fibrils formed from Δ N6 alone. Cleavage sites were observed at residues Gln-8, Tyr-10, Leu-87, and Trp-95, resulting in peptides 9–99, 11–99, 0–87, and 0–95 respectively. The core of these heteropolymeric fibrils, thus, also involves residues 12–86. Cleavage of h β_2 m fibrils with





FIGURE 5. Limited proteolysis of $\Delta N6$ monomer, $h\beta_2m$, and $\Delta N6$ homopolymeric fibrils and heteropolymeric fibrils. Limited proteolysis was performed using aspergillopepsin I at pH 6 and chymotrypsin at pH 8 ($\Delta N6$ fibrils, mixed fibrils, and $\Delta N6$ monomer) or aspergillopepsin I only at pH 2 ($h\beta_2m$ fibrils) at a 1:100 (w/w) proteinase:protein ratio, mapped using ESI-MS and ESI-MS/MS. Potential chymotrypsin cleavage sites are found throughout the sequence of $h\beta_2m$ (*gray bar*). Cleavage sites in the fibrils and $\Delta N6$ monomer are shown using *vertical bars*. The *horizontal filled bars* represent the peptide fragments observed. Cleavage of $\Delta N6$ monomers with chymotrypsin at pH 8 is also shown (*pink*).

aspergillopepsin I at pH 2.0 showed cleavages at Met-0, Gln-2, Gln-8, and Asp-96 (resulting in the peptides 1–99, 3–99, 9–99, and 0–96 (Fig. 5)), consistent with previous results suggesting a more extensive fibril core (residues 10-95) (23).

Spectroscopic Analysis of Homopolymeric and Heteropolymeric Fibrils—Having demonstrated the ability of Δ N6 and h β_2 m to assemble alone (at different pH) or together (at pH 6.2) into homopolymeric or heteropolymeric fibrils with similar fibril cores, we next sought to characterize the conformational properties of the different fibrils formed using spectroscopic analyses. FTIR spectroscopy is able to distinguish between amyloid fibrils and other β -sheet-containing structures. The cross- β architecture of amyloid results in an absorbance band at ~1620 cm⁻¹, whereas β -sheet structures in globular proteins absorb typically at around 1640 cm⁻¹ (22).

To confirm that incubation of $\Delta N6$ monomers at pH 6.2 results in fibrils with the characteristic properties of amyloid and to compare the underlying structures of the amyloid fibrils formed from $h\beta_2$ m at pH 2.0, $\Delta N6$ at pH 6.2, and the 1:1 mixture of h β_2 m: Δ N6 at pH 6.2, each of the fibril samples was analyzed using FTIR (Fig. 6A). All three fibril types give rise to a maximum absorbance band at 1620 cm^{-1} , typical of amyloid. Indeed, the FTIR spectrum of the heteropolymeric fibril sample is indistinguishable from that of $\Delta N6$ fibrils, whereas the h β_2 m fibrils give rise to an additional band at \sim 1650 cm⁻¹ that has been observed previously for these fibrils (22). By contrast, monomeric $\Delta N6$ gives rise to an absorbance maximum at ~1640 cm⁻¹, typical of that expected for β -sheet structure within globular proteins, whereas $h\beta_2m$ monomers at pH 2.0 show an absorbance maximum at ~ 1650 cm⁻¹, typical of unfolded polypeptide chains (43).

The anti-fibril antibody (IgM) WO1 binds to an epitope found in many amyloid fibrils and is a useful tool for confirming that fibrils have an amyloid conformation (35). All three fibril types were dotted onto nitrocellulose membranes and incubated with the WO1 anti-fibrillar antibody using an anti- β_2 m antibody as a control. In all three fibril types a strong positive



FIGURE 6. Spectroscopic analysis of the fibrils formed from $\Delta N6$, $h\beta_2m$, and a mixture of the two monomers. *A*, shown are FTIR absorbance spectra of $h\beta_2m$ monomer at pH 2 (*blue*) and $\Delta N6$ (*red*) monomer at pH 6.2 (*dotted lines*) and fibrils formed from $h\beta_2m$ pH 2 (*blue solid line*), $\Delta N6$ at pH 6.2 (*red solid line*) and a 1:1 mixture of $\Delta N6$: $h\beta_2m$ at pH 6.2 (*green*). For clarity the spectrum of $\Delta N6$ fibrils has been vertically offset. Its spectrum is otherwise very similar to that of the heteropolymeric fibrils. *au*, normalized absorbence units. *B*, dot blots of different samples incubated with the antibody WO1 and a polyclonal anti- β_2m antibody are shown. *C*, shown are fluorescence emission spectra of ANS in the presence of fibrils formed from $\Delta N6$ (*red*), $h\beta_2m$ (*blue*), and the mixed fibrils (*green*). Spectra of monomeric $\Delta N6$ and $h\beta_2m$ are also shown at ~0 relative fluorescence units (*rfu*). *D*, shown are intrinsic fluorescence emission spectra of $\Delta N6$ monomer at pH 6.2 (*red dotted line*) and $h\beta_2m$ monomer at pH 2 (*blue dotted line*) and $h\beta_2m$ fibrils (*blue solid line*), $\Delta N6$ fibrils (*red solid line*) and the mixed fibrils (*green*).

reactivity resulted from incubation with WO1 (Fig. 6*B*), consistent with the presence of cross- β structures. As expected, no binding of WO1 was observed to Δ N6 monomers.

The organization of side chains in the three fibril types was then probed using binding of the dye ANS as an indication of surface-exposed hydrophobicity (Fig. 6*C*) and the fluorescence



emission of the tryptophan residues to indicate differences in the environment of the two tryptophan residues in the three fibril types (Fig. 6*D*). Interestingly, incubation of each fibril type with ANS resulted in different fluorescence emission spectra, suggesting differences in surface hydrophobicity. The fluorescence emission λ_{max} values for ANS were 513, 485, and 474 nm for heteropolymeric fibrils, Δ N6 fibrils, and h β_2 m fibrils, respectively, compared with 544 nm for free ANS. Note that the λ_{max} of ANS does not change between pH 2 and 6.2, although the intensity of the emission is pH-dependent (data not shown).

 $h\beta_2 m$ and $\Delta N6$ contain two tryptophan residues. Trp-60 is solvent-exposed, whereas Trp-95 is buried from solvent in both folded proteins (25). By contrast, Trp-95 is solvent-exposed in the fibrils formed from $h\beta_2$ m at pH 2 (44). At pH 6.2 the fluorescence emission spectrum of monomeric $\Delta N6$ has a λ_{max} 335 nm, similar to that of $h\beta_2$ m at neutral pH (44), suggesting that the environments for the two tryptophan residues are similar to those of native $h\beta_2 m$ (25). By contrast, the spectrum of monomeric $h\beta_2m$ at pH 2.0 has a λ_{max} at 345 nm, consistent with unfolding of the protein at this pH. The tryptophan fluorescence emission spectra of the proteins in the three fibril types differ significantly; although a blue shift in the fluorescence maximum was observed for all three fibril samples compared with their monomeric precursors, the magnitude of this shift differs significantly for the different samples ($\Delta N6$ fibrils $\lambda_{\rm max}$ = 330 nm; h β_2 m fibrils at pH 2 $\lambda_{\rm max}$ = 340 nm; heteropolymeric fibril sample $\lambda_{max} = 336$ nm). These data indicate that the packing of the indole rings of Trp-60 and/or Trp-95 differs in the three fibril types, consistent with the results obtained using ANS fluorescence described above.

Fibril Polymorphs Have Different Stability-The studies described above have shown that the heteropolymeric fibrils composed of h β_2 m and $\Delta N6$ form a unique polymorph with properties distinct from both of their homopolymeric counterparts. To determine how the structural differences observed for the three fibril types influence their stability, each sample was titrated with GuHCl, and the extent of denaturation was determined by quantifying the amount of soluble material released after incubation of each sample for 1.5 h at each concentration of denaturant ("Experimental Procedures"). Stability was determined at the pH at which the fibrils were initially formed at (pH 2 for h β_2 m fibrils and pH 6.2 for Δ N6 and the heteropolymeric fibrils). The results of these experiments (Fig. 7) show that the $\Delta N6$ fibrils are significantly less stable than the fibrils formed from $h\beta_2$ m, with an apparent denaturation midpoint of 2.2 M GuHCl compared with 4.2 M for h β_2 m fibrils. The heteropolymeric fibril sample is less stable than both of its homopolymeric counterparts, with an apparent midpoint for denaturation of 1.5 M GuHCl. Even in the absence of GuHCl, significant soluble material was present in the supernatant of the mixed fibrils after ultracentrifugation, suggesting that the critical concentration for polymerization is increased for this combination of monomer precursors compared with h β_2 m or $\Delta N6$ assembly alone.

DISCUSSION

Here we have investigated the effects of a naturally occurring N-terminal truncation of $\beta_2 m$ on the thermodynamic and



FIGURE 7. Thermodynamic stability of fibrils formed from $\Delta N6$ (*red*), $h\beta_2m$ (*blue*), and heteropolymeric fibrils (*green*). The release of soluble material was measured using absorbance at 280 nm after incubation in GuHCI for 1.5 h. The results are presented as the proportion of soluble material by dividing the concentration of the soluble monomer by the total starting monomer concentration. The fit is to guide the eye.

structural properties of amyloid fibrils formed from this variant alone or from a 1:1 mixture of h β_2 m and Δ N6 monomers. Despite subtle differences in the structures of Δ N6 and h β_2 m monomers at pH 6.2, these two proteins possess fundamentally different abilities to form amyloid fibrils at this pH (25). We show here that the two proteins are able to co-polymerize to form amyloid fibrils that have unique structural and thermodynamic properties.

Fig. 8 depicts three possible schemes for how co-polymerization of $h\beta_2m$ and $\Delta N6$ may occur. The central path begins with a collision between monomeric $h\beta_2m$ and $\Delta N6$, whereupon $h\beta_2m$ undergoes a conformational conversion to an amyloidcompetent state (25). This is thought to occur by the displacement of the A-strand from the native β -sandwich structure of $h\beta_2m$ (25), leading to isomerization of cis Pro-32 to trans, and further partial unfolding of $h\beta_2m$. The equal incorporation of $h\beta_2m$ and $\Delta N6$ monomers into heteropolymers, as shown here by mass spectrometry and confocal microscopy, are consistent with such a scheme.

Another possibility, shown in the top scheme in Fig. 8 is that $\Delta N6$ forms a homopolymeric oligomer followed by an interaction with h β_2 m, from which the heteropolymeric fibrils form. These heteropolymeric oligomers may also form from an initial $\Delta N6:h\beta_2m$ dimer, with the two pathways in a dynamic equilibrium. The final pathway, depicted as the lower scheme in Fig. 8, is that $\Delta N6$ forms homopolymeric fibrils first, which then seed elongation with monomeric $h\beta_2$ m. A seeding mechanism for the system described here, although possible (45), is unlikely for two reasons. First, the ThT kinetics show that the presence of $h\beta_2m$ extends the lag phase of $\Delta N6$ fibril formation compared with the same concentration of $\Delta N6$ incubated alone, suggesting that an interaction occurs between $h\beta_2m$ and $\Delta N6$ before fibrils are formed. Second, the confocal images of the fibrils formed from mixing $\Delta N6$ and $h\beta_2 m$ show no evidence of a seeded-elongation reaction such as that observed for extension of h β_2 m at pH 2 (46) and in other systems (47). Overall, therefore, heteropolymerization is most likely to occur through monomer-monomer or monomer-oligomer interactions of





FIGURE 8. Co-polymerization of h β_2 m and Δ N6 can occur by a variety of different possible mechanisms, involving oligomer formation, initial heterodimer formation, or cross-seeding. See Discussion for details.

 $h\beta_2m$ and $\Delta N6$. As a consequence, sequence truncation not only results in the ability of $h\beta_2m$ to form amyloid fibrils at neutral pH but also results in the formation of a heteropolymeric fibril with unique properties.

Amyloid Polymorphism Revealed through the Co-polymeriza*tion of* $\beta_2 m$ —Different packing of side chains in the h $\beta_2 m$, $\Delta N6$, and the heteropolymeric fibrils, indicated by their MAS NMR spectra, ANS binding, and tryptophan fluorescence spectra, results in a pronounced difference in the stability of the fibrils formed. Polymorphism has been previously categorized based on structure (48); however, here we portray an additional form of polymorphism, termed here "stability polymorphism," in which co-polymerization of related fibril precursors leads to fibrils with unique structural and thermodynamic signatures. Whether stability polymorphism affects the biological response to fibrils requires further study. Given that amyloid plaques in *vivo* have been shown to be reservoirs of toxic oligomers (49), differences in amyloid stability and, therefore, the rate of depolymerization into harmful species may indeed result in differential effects of fibrils on cell toxicity.

Polymorphism and co-polymerization of proteins are intimately linked, with polypeptide heterogeneity giving rise to an array of potential changes in amyloid structure and/or stability. Fibrils composed of multiple species can arise through co-polymerization of two pools of monomer as shown here as well as through cross-seeding, in which existing fibrils (seeds) of one species catalyze fibril formation of monomers of a different sequence. This "dock and lock mechanism" occurs when a fully solvated monomer weakly binds to the peptides in the fibril and adopts their conformation (50, 51). When seeds are present, they can also have the effect of templating their structure onto the monomer pool, resulting in a structurally different seeded fibril to *de novo* fibrils formed by their unseeded counterparts (52). Some amyloid fibrils are also capable of accommodating peptides with mismatched sequences, enabling conformational switching during the cross-seeding reaction that results in fibrils of a new structure (53). However, there are limits to cross-seeding; as the sequence identity between the seed and the monomer decreases, the efficiency of the seeding reaction is reduced (14). Such events give rise to the species barrier in which a protein from one species is unable to seed the same protein from a different species, such as observed for prions (54), h β_2 m, and murine β_2 m (25) as well as other protein species (55).

Co-polymerization; a Common Feature of Amyloid Assembly-Co-polymerization of different protein precursors may be a common phenomenon in amyloid disease. In vivo, many amyloid deposits are heterogeneous in composition, containing monomers with variations in protein length (truncations), sequence (mutations), composition (*e.g.* the ratio of $A\beta 40$: A β 42), post-translational modifications, and the presence of amyloid-associated co-factors (for review, see Ref. 48). In the system described here we demonstrated the co-polymerization of β_2 m and its truncated counterpart Δ N6. This has relevance to the disease dialysis-related amyloidosis, as \sim 30% of the protein found in amyloid plaques is $\Delta N6$, with the remainder being predominantly $h\beta_2 m$ (26). Whether co-polymerization of these proteins occurs during assembly or post-assembly by proteolysis of the $h\beta_2 m$ homopolymer is not clear. Likewise in Alzheimer disease N-terminally truncated, pyroglutamated forms of amyloid- β -peptide co-polymerize with A β 42 at levels as low as 5% mol/mol, resulting in oligomers that are more toxic than either protein oligomerizing alone (56). Additionally the ratio of A β 40:42 has been shown to be critical in determining toxicity and the area of amyloid deposition in Alzheimer disease (for review, see Ref. 48). Although A β 42 was thought to be



the predominant toxic species in Alzheimer disease, there is now evidence that A β 43 can accelerate amyloid- β pathology, as A β 43 has a higher propensity to aggregate and is more neurotoxic than A β 42 (57). Such species are capable of co-polymerization, which presumably will result in an array of different oligomeric and fibrillar species with unique structural, thermodynamic, kinetic, and functional properties.

Inclusions of tau and α -synuclein are present in individuals with sporadic neurodegenerative disorders, and a two-step mechanism of initiation followed by propagation has been proposed to explain how these two proteins interact (58). Similarly an elegant study using immunogold labeling of transthyretinderived peptides showed that various guest peptides can be randomly inserted into the growing fibril (59). Moreover the same study used insulin fibrils doped with transthyretin peptides and found that the kinetics of fibril formation of both species must be relatively evenly matched for co-polymerization to occur (59). Co-incubation of proteins can also result in suppression of fibril formation. In yeast the interactions between different prions through cross-seeding can promote or inhibit prion propagation (60). A conformationally constrained analog of (islet amyloid polypeptide), designed to be a mimic of the non-amyloidogenic IAPP conformation, has also been shown to be able to bind prefibrillar A β and heteroassociate to block and reverse A β self-assembly (61).

The structural and thermodynamic studies described here demonstrate that combining β_2 m and Δ N6 monomers does not prevent fibril formation but in fact can enhance the ability of $h\beta_2$ m to form fibrils and extend the repertoire of polymorphs formed. We reveal here that the heteropolymers formed by copolymerization of Δ N6 and $h\beta_2$ m have unique structural properties and a unique thermodynamic signature compared with their homopolymeric forms. How this is encoded by differences in structure will require further high resolution information, so that the thermodynamic differences can be rationalized in structural terms. Understanding this process further may also shed light on the fundamental molecular mechanisms of fibril formation and how the presence of heteropolymeric assemblies can affect the extent, rate, and biological consequences of amyloid deposition.

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