## Research Article

# Thermostable Alkaline Phytase from *Alcaligenes* sp. in Improving Bioavailability of Phosphorus in Animal Feed: *In Vitro* Analysis

## Ponnuswamy Vijayaraghavan,<sup>1</sup> R. Raja Primiya,<sup>2</sup> and Samuel Gnana Prakash Vincent<sup>3</sup>

<sup>1</sup> Centre for Marine Science and Technology, Manonmaniam Sundaranar University, Rajakkamangalam,

Kanyakumari, Tamil Nadu 629 502, India

<sup>2</sup> P. G. Department of Microbiology, Ayya Nadar Janaki Ammal College, Sivakasi, Tamil Nadu 626 124, India

<sup>3</sup> International Centre for Nanobiotechnology, Centre for Marine Science and Technology, Manonmaniam Sundaranar University, Rajakkamangalam, Kanyakumari District, Tamil Nadu 629 502, India

Correspondence should be addressed to Ponnuswamy Vijayaraghavan; venzymes@gmail.com

Received 31 December 2012; Accepted 17 January 2013

Academic Editors: H. S. Garcia, G. Ladics, C. D. Murphy, S. Pan, H. Stamatis, and A. Trincone

Copyright © 2013 Ponnuswamy Vijayaraghavan et al. This is an open access article distributed under the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

A bacterial isolate, *Alcaligenes* sp. secreting phytase (EC 3.1.3.8), was isolated and characterized. The optimum conditions for the production of phytase included a fermentation period of 96 h, pH 8.0, and the addition of 1% (w/v) maltose and 1% (w/v) beef extract to the culture medium. This enzyme was purified to homogeneity and had an apparent molecular mass of 41 kDa. The optimum pH range and temperature for the activity of phytase were found to be 7.0-8.0 and 60°C, respectively. This enzyme was strongly inhibited by 0.005 M of  $Mn^{2+}$ ,  $Mg^{2+}$ , and  $Zn^{2+}$ . *In vitro* studies revealed that the phytase from *Alcaligenes* sp. released inorganic phosphate from plant phytates. Phytase released 1930 ± 28, 1740 ± 13, 1050 ± 31, 845 ± 7, 1935 ± 32, and 1655 ± 21 mg inorganic phosphate/kg plant phytates, namely, chick pea, corn, green pea, groundnut, pearl pea, and chick feed, respectively.

### 1. Introduction

Phytate-degrading enzyme (phytase, EC 3.1.3.8) preparations have a wide range of applications in animal and human nutrition. Besides that, these enzymes have also attracted considerable attention from both scientists and entrepreneurs in the areas of environmental protection and biotechnology. Phytases are of great interest in biotechnological applications, in particular for the reduction of phytate content in feed and food [1]. Phytases are capable of initiating the stepwise release of myoinositol and phosphoric acid, leading to the formation of myoinositol phosphate intermediates from phytate [2]. Most plant-origin foods have from 50% to 80% of their total phosphorus as phytate [3]. Phytate chelates essential minerals, binds to amino acids and proteins, inhibits digestive enzymes, and decreases the nutritive value of food [4]. Monogastric animals poorly utilized phytate-bound phosphorus, due to insufficient phytate-degrading activity in the gut [5]. Therefore, hydrolysis of phytate is desirable for releasing valuable nutrients for beneficial utilization. The addition of phytate-degrading enzymes can improve the nutritional value of plant-based foods by enhancing protein digestibility and mineral availability through phytate hydrolysis during digestion in the stomach or during food processing [6], thus reducing the phosphorus excretion of animals [7].

Only a limited number of bacterial phytases have been reported and studied [8]. Phytase has been isolated from bacteria such as *Escherichia coli* [9], *Pseudomonas* sp. [10], anaerobic rumen bacteria, particularly *Selemonas ruminantium, Megasphaera elsdenii, Prevotella* sp., *Mitsuokella multiacidus* [11], and *Raoultella* sp. [12]. Numerous studies have shown the effectiveness of supplemental microbial phytases in improving the utilization of phosphate from phytate [13]. Naturally occurring phytases having the required level of thermostability for application in animal feeds have not been found [14]. Hence, the main objective of the present study was to characterize alkaline- thermostable phytase from *Alcaligenes* sp., and possible application in *in vitro* digestion of plant phytate was evaluated. A study of this kind will improve our knowledge on the biotechnological application of phytase in feed industry.

#### 2. Materials and Methods

2.1. Bacterial Culture and Screening. The bacterial isolate, Alcaligenes sp., was obtained from the Microbiology Laboratory, Centre for Marine Science and Technology, M. S. University, Rajakkamangalam, Tamil Nadu, India. To screen phytase-producing microorganism effectively, the plate agar method was followed [15]. Sodium phytate (2%, w/v) was added as substrate in plate screening media. The Alcaligenes sp. was grown on this solid medium and incubated at 37°C for 48 h. Phytase plate screening was carried out by washing the colonies with distilled water from the agar surface and flooding the plate with 2% (w/v) cobalt chloride solution and incubated for 30 min at room temperature. Then replaced the cobalt chloride solution with a freshly prepared solution containing equal volume of 6.25% (w/v) ammonium molybdate and 0.42% (w/v) ammonium vanadate solution. Following further 30 min incubation the solution removed and the plates were examined for zones of clearing which indicates phytase activity.

2.2. Phytase Production. Phytase-secreting Alcaligenes sp. was subjected for phytase production in the production medium (containing 1 litre: sodium phytate, 10 g; (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, 1g; MgSO<sub>4</sub>·7H<sub>2</sub>O, 0.1g; CaCl<sub>2</sub>·2H<sub>2</sub>O, 0.1g; trace element solution, 1.0 mL; KCl, 0.7 g; glucose, 1 g; D-mannose, 1 g). The pH of the medium was adjusted to 7.0 by using 1N HCl/1N NaOH. Sodium phytate was filter-sterilized separately, and added after autoclaving, to the production medium. A loopfull of bacterial culture was transferred to 150 mL of phytase production medium in a 250-mL Erlenmeyer flask, and the culture was incubated at 37°C with shaking at 150 rpm. For the determination of the growth curve, 5 mL of the culture medium was withdrawn at regular intervals of 12 h and the cell density determined at 600 nm up to 120 h. To study the optimum incubation time for phytase production, the culture was withdrawn every 12h and centrifuged at  $10,000 \times g$  for 15 min. The culture supernatant was assayed for determining enzyme activity. Effect of pH on enzyme production was studied by adjusting the culture medium pH to 6.0, 7.0, 8.0, 9.0, and 10.0 by the addition of 1 N HCl/NaOH prior to sterilization. To study the effect of carbon and nitrogen sources on enzyme production, the organism was grown in the production medium containing additional 1% (w/v) carbon (arabinose, dextrose, maltose, sucrose, trehalose, mannose, and starch) and nitrogen sources (urea, ammonium sulphate, peptone, ammonium chloride, and beef extract). All experiments were conducted in triplicates and average values were reported.

2.3. Crude Enzyme Preparation and Phytase Assay. Bacterial culture was prepared for assay as suggested by Yanke et al. [11]. A cell-free extract was prepared from 96h-old culture by centrifuging it at 10,000 ×g for 15 min. The culture supernatant was assayed for determining phytase activity, which was based on the estimation of inorganic phosphate released on hydrolysis of phytic acid, at 37 °C. One unit of enzyme activity was defined as the amount of enzyme that liberates 1  $\mu$ mol of inorganic phosphate per minute under standard assay conditions. The liberated inorganic phosphate was measured according to the ammonium molybdate method [16]. The total protein content of the sample was estimated as described by Lowry et al. [17], and bovine serum albumin was used as the standard.

2.4. Properties of Alcaligenes Phytase. The optimum pH for the enzyme activity was determined by using the following buffers (0.1 M): citrate buffer (pH 4.0), succinate buffer (pH 5.0-6.0), tris-acetate buffer (pH 7.0), tris-HCl buffer (pH 8.0), and glycine-NaOH buffer (pH 9.0). The stability of the enzyme at various pH was examined by incubating the enzyme solution in buffers ranging in pH from 4 to 9 at 37°C for 30 min; enzyme activity was assayed as described earlier. The temperature profile of the enzyme was determined by performing the routine enzyme assay at varying incubation temperatures (30, 40, 50, 60, and 70°C). To determine the thermal stability of the enzyme, it was incubated for 30-70°C for 30 min before performing the routine enzyme assay. The effect of divalent ions and chemicals on enzyme activity was determined by incubating the enzyme with various divalent ions and chemicals (0.005 M), namely, Ca<sup>2+</sup>, Mn<sup>2+</sup>, Zn<sup>2+</sup>, Cu<sup>2+</sup>, Mg<sup>2+</sup>, ethylenediaminetetraacetic acid (EDTA), dithiothreitol, and  $\beta$ -mercaptoethanol for 30 min. The enzyme activity was assayed as described earlier.

2.5. Purification of Extracellular Alcaligenes Phytase. The crude enzyme preparation (140 mL) was fractionated with ammonium sulphate (60% to 80% saturation) as suggested by Fu et al. [18]. The protein precipitate was dissolved in a minimal volume of double-distilled water and the resulting enzyme was dialyzed against the buffer A (tris-HCl buffer, 0.025 M, pH 8.0) overnight at 4°C. The dialyzed sample was subjected to Diethyl-aminoethyl cellulose (DEAE cellulose) chromatography (Merck, Bangalore, India) using buffer A, and the bounded protein was eluted with a linear gradient of NaCl (0-1 M). The fractions having phytase activity were combined and concentrated using ammonium sulphate (80% saturation) and dialyzed against buffer B (tris-HCl buffer, 0.05 M, pH 8.0). The concentrated sample was loaded on a preequilibrated sephadex G-75 column (Amersham Biosciences, SE-751 84, Uppsala, Sweden) and eluted with buffer B.

2.6. Polyacrylamide Gel Electrophoresis. Sodium dodecyl sulphate-polyacrylamide gel electrophoresis (SDS-PAGE) was performed according to Laemmli [19]. The stacking gel was 5% (w/v) polyacrylamide, and the separating gel was 11% (w/v) polyacrylamide. The active fraction obtained from

the sephadex G-75 column was subjected to electrophoresis. After electrophoresis, the gel was treated and stained for phytase activity as described by Yanke et al. [11], with a cobalt chloride solution followed by an ammonium molyb-date/ammonium vanadate solution. The gel was documented using a gel documentation system (Syngene, Cambridge, CB4 1TF, UK) and the molecular weight of the purified enzyme determined.

2.7. In Vitro Hydrolysis of Plant Phytate. An in vitro experiment was performed to study the hydrolysing efficacy of *Alcaligenes* phytase on plant phytate [18]. Briefly, plant phytates from various sources, namely, chick pea, corn, green pea, groundnut, pearl pea, and chick feed named "Layer Mash" brought from the local market were used. All materials were ground into coarse powder and sieved (0.2-0.3 mm). Two grams of the above material were placed in a 50-mL conical flask. To this, 10 U crude enzyme and 5 mL of buffer B were added and the mixture was placed in an orbital shaker at 150 rpm for 30 min. A control experiment was carried out separately; for this, 10 mL of buffer B alone was applied to the raw materials. To the reaction mixture, 10 mL trichloroacetic acid (15%) was added to stop the reaction. The released inorganic phosphorus was determined [19].

#### 3. Results and Discussion

*3.1. Screening of Phytase Secreting Alcaligenes sp.* The isolate, *Alcaligenes* sp. secreted alkaline phytase when it was grown on phytase screening medium. This organism produced a 2.0-mm zone around the colony after 48 h. This organism was used for further studies.

3.2. Growth and Phytase Production. The Alcaligenes sp. attained maximum growth after 72 h of fermentation, and the absorbance was 1.972  $\pm$  0.036 at 600 nm. The absorbance declined to  $1.728 \pm 0.027$ ,  $1.615 \pm 0.039$ ,  $1.619 \pm 0.781$ , and  $1.407 \pm 0.033$  at 84, 96, and 108 h of incubation, respectively. The isolate produced an interesting phytase when grown in minimal medium containing sodium phytate as the sole source of phosphorus. The enzyme secretion increased to its maximum of  $6.202 \pm 0.082 \text{ U/mL}$  at 96 h of fermentation. Enzyme production was found to be  $0.008 \pm 0.005$ ,  $1.811 \pm$  $0.221, 4.351 \pm 0.192$ , and  $3.610 \pm 0.191$  U/mL at 24, 48, 72, and 120 h of fermentation, respectively, (Figure 1). In this organism, phytase was expressed on the onset of late log phase. The reduction in enzyme yield after the optimum period was probably due to the depletion of nutrients that are available to microorganism. This result was in accordance with the observation made with Yersinia kristensenii [18].

3.3. Effect of pH on Enzyme Production. Effect of pH on enzyme production was studied by culturing the organism at various pH levels (6.0–10.0) using 1 N HCl/NaOH. Phytase production was  $1.5 \pm 0.15$ ,  $2.96 \pm 0.16$ ,  $2.0 \pm 0.23$ ,  $1.7 \pm 0.10$  U/mL at pH 6.0, 7.0, 9.0, and 10.0, respectively. Enzyme production was found to be high at pH 8.0 ( $4.9 \pm 0.18$  U/mL). At high pH level (9.0), phytase production decreased. At

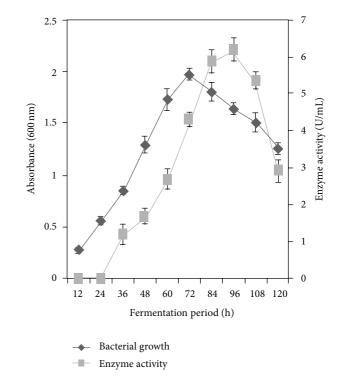
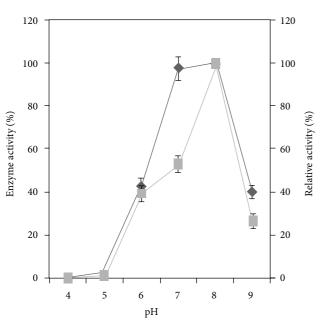


FIGURE 1: Growth and production of phytase from *Alcaligenes* sp. The isolate was inoculated and incubated for 120 h at  $37^{\circ}$ C. The culture was withdrawn every 12 h, and the growth enzyme activity was determined. The result was the mean of three different repeats. The error bar indicates standard deviation.

higher pH, the metabolic action of the bacterium may be suppressed and thus enzyme production decreased. This result was in accordance with the observations made with *Mitsuokella jalaludinii* by Lan et al. [20].

3.4. Effect of Carbon and Nitrogen Sources on Enzyme Production. Phytase production was found to be high (6.2  $\pm$ 0.018 U/mL) in the production medium supplemented with 1% (w/v) maltose. Enzyme production was  $5.0 \pm 0.032$ ,  $4.33 \pm$ 0.01, 4.8  $\pm$  0.05, 3.47  $\pm$  0.09, and 3.07  $\pm$  0.008 U/mL in the production medium containing starch, dextrose, trehalose, sucrose, and arabinose, respectively. Mannose (1%, w/v) repressed the production of phytase. Similar results were obtained with Bacillus sp. KHU-10 [21]. Phytase production was high  $(6.5 \pm 0.026 \text{ U/mL})$  in the production medium containing beef extract (1%, w/v) as the sole source of nitrogen. Enzyme production was  $3.35 \pm 0.05$ ,  $3.0 \pm 0.03$ ,  $4.6 \pm 0.13$ , and  $5.2 \pm 0.08$  U/mL in production medium containing urea, ammonium sulphate, peptone, and ammonium chloride, respectively. This result was in accordance with the observation made with Bacillus sp. KHU-10 [21].

3.5. Effect of pH on Enzyme Activity and Stability. In Alcaligenes sp. phytase activity was found to be high (100% activity) at pH 8.0. The relative enzyme activity was  $0.9 \pm 0.13\%$ ,  $3.4 \pm 0.18\%$ ,  $44.5 \pm 3.4$ ,  $98 \pm 5.5\%$ , and  $40 \pm 2.2\%$  at pH 4.0, 5.0, 6.0, 7.0, and 9.0, respectively. This enzyme was also highly



Activity

Stability

FIGURE 2: Effect of pH on phytase activity and stability. The *Alcaligenes* sp. was inoculated in the minimal medium containing 1% (w/v) sodium phytate as the sole phosphorus source and incubated at 37°C for 144 h in an orbital shaker at 150 rpm. The result was the mean of three different repeats. *Error bar* standard deviation.

stable at pH 8.0 at which the relative enzyme activity was 100%. The relative enzyme activity was  $0.4 \pm 0.16$ ,  $40 \pm 0.21$ ,  $47 \pm 0.18$ , and  $27 \pm 0.24$ %, at pH 4.0, 5.0, 6.0, 7.0, and 9.0, respectively, (Figure 2). This property was in accordance with the previously reported results. Some bacterial phytases have broad pH optima but are shifted towards a more basic pH range [22, 23]. This enzyme was highly stable at pH 8.0 (100% relative activity).

3.6. Effect of Temperature on Enzyme Activity and Stability. The effect of phytase activity was studied by incubating the reaction mixture at various temperatures (30-70°C). The Alcaligenes phytase was highly active (100% activity) at 60°C (Figure 3). Optimal temperatures in which most phytases are active vary from 37 to 77°C [24]. Similar result was reported by Dvořáková [4] with Bacillus subtilis and with Pseudomonas syringae [25]. The relative enzyme activity was found to be high (100%) at 60°. The relative enzyme activity was 59  $\pm$ 1.6%, 68  $\pm$  2.7%, 92  $\pm$  3.4%, and 86  $\pm$  5.3% at 30, 40, 50, and 70°, respectively. Results indicated that the enzyme was stable up to 60 degrees of denaturing temperature. The relative enzyme activity was 64.7  $\pm$  4% at 70°C (Figure 3). This result shows that Alcaligenes phytase is relatively thermostable and is preferable for use in feed industry. This enzyme activity was comparable with Bacillus laevolacticus phytase [26].

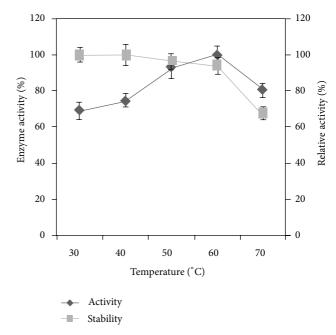


FIGURE 3: Effect of temperature on enzyme activity and stability. The enzyme sample was incubated with substrate for 30 min at temperatures ranging from 30 to 70°C. The result was the mean of three different repeats. *Error bar* standard deviation.

3.7. Effect of Ions and Chemicals on Enzyme Activity. The effect of divalent ions on enzyme activity was investigated by allowing the divalent ions (0.005 M) to react with the phytase sample. Divalent ions have been found to modulate Alcaligenes sp. phytase activity. However, it is difficult to determine whether the inhibitory effect of various metals is due to the direct binding of the metal ions to the enzyme, or whether the metal ions form poorly soluble complexes with phytic acid, thereby decreasing the active substrate concentration. Among divalent, Ca<sup>2+</sup> enhanced the relative enzyme activity (103  $\pm$  2.89%) when compared with the control (100%).  $Mn^{2+}$ ,  $Zn^{2+}$ ,  $Cu^{2+}$ , and  $Mg^{2+}$  inhibited the enzyme activity and the relative activity was 46  $\pm$  3.8%,  $39 \pm 2.92\%$ ,  $73 \pm 1.3\%$ , and  $21 \pm 0.8\%$ , respectively. EDTA also inhibited phytase activity (49  $\pm$  1.3%). Similar results were obtained with Enterobacter sp. [23] and Yersinia kristeensenii [18].

3.8. Purification of Alcaligenes Phytase. Phytase was purified to homogeneity by sequential ammonium sulphate precipitation, anionic exchange chromatography, and gel filtration chromatography. The DEAE cellulose chromatography was still the major technique for purification of *Alcaligenes* phytase, because majority of contaminating proteins were removed at this step. In this step 3.57-fold purification was achieved with 25.3% yield. The recovery and purification were 7.07%- and 4.75-fold, respectively, after sephadex G-75 gel filtration chromatography. The purification procedure of phytase is summarized in Table 1. The molecular weight of the enzyme was found to be 41 kDa (Figure 4). A twostep chromatographic process, similar to that employed with

Purification step	Total activity (U)	Total protein (mg)	Specific activity (U/mg)	Purification (fold)	Yield (%)
Crude enzyme	980	182	5.38	1.0	100
$80\% (NH_4)_2 SO_4$	562	84.6	6.64	1.2	57.3
DEAE cellulose	248	12.9	19.2	3.57	25.3
Sephadex G-75	69.3	2.8	25.7	4.75	7.07

TABLE 1: Summary of the purification of the phytase from Alcaligenes sp.

TABLE 2: *In vitro* hydrolysis of plant phytate by alkaline-stable phytase from *Alcaligenes* sp.

Phytate source	Released inorganic phosphorus (mg/kg raw material)		
Chick pea	$1930 \pm 23$		
Corn	$1740 \pm 86$		
Green pea	$1050 \pm 61$		
Ground nut	$845 \pm 35$		
Pearl pea	$1935 \pm 42$		
Chick feed	$1655 \pm 25$		

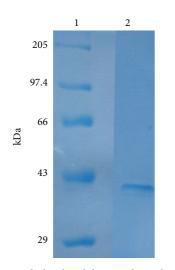


FIGURE 4: Sodium dodecyl sulphate polyacrylamide gel electrophoresis (11%) of the purified phytase from *Alcaligenes* sp. Lane 1: molecular mass standards; Lane 2: purified sample from sephadex G-75.

Aspergillus ficuum [27], was followed. Homogeneity of the purified enzyme was revealed by SDS-PAGE which showed a single band with an apparent molecular mass of 41 kDa (Figure 4). This result similar with a molecular mass was reported in bacteria [28].

3.9. Invitro Hydrolysis of Plant Phytate. In the present study various sources of plant phytates were used to investigate the effect of phytate, and the released inorganic phosphorus was measured. Plant phytates, namely, chick pea, corn, green pea, groundnut, pearl pea, and chick feed named "Layer Mash" were chosen as the substrate for *in vitro* hydrolysis. The maximal inorganic phosphorus release was obtained with pearl pea. The released inorganic phosphorus from these

sources was significantly high (Table 2) than other reported results. A similar finding was reported by Fu et al. [18] with *Yersinia kristeensenii*. This property of the *Alcaligenes* phytase makes this a potential supplement in the animal feed industry. The plant phytate sources mentioned above are widely used as raw materials in the animal feed industry. Phytase cannot be fully used by agastric fish like carps, in which the pH of the digestive tract is about 6.5–8.4. Interestingly, *Alcaligenes* phytase was active and stable at this range of pH. Thus, this phytase could be used effectively in the feed of monogastric animals and other organisms.

#### 4. Conclusions

Phytate-rich plant feed restricts the bioavailability of phosphorus along with other minerals. It is clear that supplemental phytase can enhance the bioavailability of phosphorus and other minerals in plant-based feed. The *Alcaligenes* phytase had an optimum pH at alkaline range (7.0-8.0) and, fortunately, the fish gut pH fell at this range. Hence, this phytase could be used effectively to increase the bioavailability of phosphorus in feed. Further investigation about phytase application in fish feed is needed to study in *in vivo*. The progress in biotechnology of phytase is remarkable and attracting worldwide attention. Overproduction of phytase can be achieved by physical, chemical methods of mutagenesis and through recombinant DNA technology.

#### References

- A. Vohra and T. Satyanarayana, "Phytases: microbial sources, production, purification, and potential biotechnological applications," *Critical Reviews in Biotechnology*, vol. 23, no. 1, pp. 29– 60, 2003.
- [2] U. Konietzny and R. Greiner, "Molecular and catalytic properties of phytate-degrading enzymes (phytases)," *International Journal of Food Science and Technology*, vol. 37, no. 7, pp. 791– 812, 2002.
- [3] B. F. Harland and E. R. Morris, "Phytate: a good or a bad food component?" *Nutrition Research*, vol. 15, no. 5, pp. 733–754, 1995.
- [4] J. Dvořáková, "Phytase: sources, preparation and exploitation," Folia Microbiologica, vol. 43, no. 4, pp. 323–338, 1998.
- [5] H. Fandrejewski, S. Raj, D. Weremko et al., "Apparent digestibility in experiment feeds and the effect of commercial phytase," *Asian-Australasian Journal of Animal Sciences*, vol. 10, no. 6, pp. 665–670, 1997.
- [6] A. S. Sandberg and T. Andlid, "Phytogenic and microbial phytases in human nutrition," *International Journal of Food Science and Technology*, vol. 37, no. 7, pp. 823–833, 2002.

- [7] V. Ravindran, P. H. Selle, G. Ravindran, P. C. H. Morel, A. K. Kies, and W. L. Bryden, "Microbial phytase improves performance, apparent metabolizable energy, and ileal amino acid digestibility of broilers fed a lysine-deficient diet," *Poultry Science*, vol. 80, no. 3, pp. 338–344, 2001.
- [8] M. Jorquera, O. Martínez, F. Maruyama, P. Marschner, and M. D. L. L. Mora, "Current and future biotechnological applications of bacterial phytases and phytase-producing bacteria," *Microbes and Environments*, vol. 23, no. 3, pp. 182–191, 2008.
- [9] R. Greiner, U. Konietzny, and K. D. Jany, "Purification and characterization of two phytases from *Escherichia coli*," *Archives* of *Biochemistry and Biophysics*, vol. 303, no. 1, pp. 107–113, 1993.
- [10] A. E. Richardson and P. A. Hadobas, "Soil isolates of *Pseudomonas* spp. that utilize inositol phosphates," *Canadian Journal of Microbiology*, vol. 43, no. 6, pp. 509–516, 1997.
- [11] L. J. Yanke, H. D. Bae, L. B. Selinger, and K. J. Cheng, "Phytase activity of anaerobic ruminal bacteria," *Microbiology*, vol. 144, no. 6, pp. 1565–1573, 1998.
- [12] A. Sajidan, A. Farouk, R. Greiner, P. Jungblut, E. C. Müller, and R. Borriss, "Molecular and physiological characterisation of a 3-phytase from soil bacterium *Klebsiella* sp. ASR1," *Applied Microbiology and Biotechnology*, vol. 65, no. 1, pp. 110–118, 2004.
- [13] O. P. Walz and J. Pallauf, "Microbial phytase combined with amino acid supplementation reduces P and N excretion of growing and finishing pigs without loss of performance," *International Journal of Food Science and Technology*, vol. 37, no. 7, pp. 835–848, 2002.
- [14] X. G. Lei and C. H. Stahl, "Biotechnological development of effective phytases for mineral nutrition and environmental protection," *Applied Microbiology and Biotechnology*, vol. 57, no. 4, pp. 474–481, 2001.
- [15] S. J. Howson and R. P. Davis, "Production of phytatehydrolysing enzyme by some fungi," *Enzyme and Microbial Technology*, vol. 5, no. 5, pp. 377–382, 1983.
- [16] H. H. Taussky and E. Skorr, "A microcolorimetric method for the determination of inorganic phosphorus," *The Journal of Biological Chemistry*, vol. 202, pp. 675–685, 1953.
- [17] O. H. Lowry, N. J. Rosebrough, A. L. Farr, and R. J. Randall, "Protein measurement with the folin phenol reagent," *The Journal of Biological Chemistry*, vol. 193, no. 1, pp. 265–275, 1951.
- [18] D. Fu, H. Huang, H. Luo et al., "A highly pH-stable phytase from Yersinia kristeensenii: cloning, expression, and characterization," *Enzyme and Microbial Technology*, vol. 42, no. 6, pp. 499–505, 2008.
- [19] U. K. Laemmli, "Cleavage of structural proteins during the assembly of the head of bacteriophage T4," *Nature*, vol. 227, no. 5259, pp. 680–685, 1970.
- [20] G. Q. Lan, N. Abdullah, S. Jalaludin, and Y. W. Ho, "Optimization of carbon and nitrogen sources for phytase production by *Mitsuokella jalaludinii*, a new rumen bacterial species," *Letters in Applied Microbiology*, vol. 35, no. 2, pp. 157–161, 2002.
- [21] Y. M. Choi, H. J. Suh, and J. M. Kim, "Purification and properties of extracellular phytase from bacillus sp. KHU-10," *Protein Journal*, vol. 20, no. 4, pp. 287–292, 2001.
- [22] M. Shimizu, "Purification and characterization of phytase from Bacillus subtilis (natto) N-77," Bioscience, Biotechnology, and Biochemistry, vol. 56, pp. 1266–1269, 1992.
- [23] S. J. Yoon, Y. J. Choi, H. K. Min et al., "Isolation and identification of phytase-producing bacterium, *Enterobacter* sp. 4, and enzymatic properties of phytase enzyme," *Enzyme and Microbial Technology*, vol. 18, no. 6, pp. 449–454, 1996.

- [24] A. Hara, S. Ebina, A. Kondo, and T. Funaguma, "A new type of phytase from pollen of *Typha latifolia L.*," *Agricultural and Biological Chemistry*, vol. 49, pp. 3539–3544, 1985.
- [25] J. S. Cho, C. W. Lee, S. H. Kang et al., "Purification and characterization of a phytase from *Pseudomonas syringae* MOKI," *Current Microbiology*, vol. 47, no. 4, pp. 290–294, 2003.
- [26] H. K. Gulati, B. S. Chadha, and H. S. Saini, "Production of feed enzymes (phytase and plant cell wall hydrolyzing enzymes) by *Mucor indicus* MTCC 6333: purification and characterization of phytase," *Folia Microbiologica*, vol. 52, no. 5, pp. 491–497, 2007.
- [27] A. H. Ullah and D. M. Gibson, "Extracellular phytase (E.C. 3.1.3.8) from Aspergillus ficuum NRRL 3135: purification and characterization," *Preparative Biochemistry*, vol. 17, no. 1, pp. 63– 91, 1987.
- [28] R. Greiner and A. E. Farouk, "Purification and characterization of a bacterial phytase whose properties make it exceptionally useful as a feed supplement," *Protein Journal*, vol. 26, no. 7, pp. 467–474, 2007.