

Thermophilic biohydrogen production: how far are we?

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Abstract Apart from being applied as an energy carrier, hydrogen is in increasing demand as a commodity. Currently, the majority of hydrogen (H₂) is produced from fossil fuels, but from an environmental perspective, sustainable H₂ production should be considered. One of the possible ways of hydrogen production is through fermentation, in particular, at elevated temperature, i.e. thermophilic biohydrogen production. This short review recapitulates the current status in thermophilic biohydrogen production through fermentation of commercially viable substrates produced from readily available renewable resources, such as agricultural residues. The route to commercially viable biohydrogen production is a multidisciplinary enterprise. Microbiological studies have pointed out certain desirable physiological characteristics in H₂-producing microorganisms. More process-oriented research has identified best applicable reactor types and cultivation conditions. Techno-economic and life cycle analyses have identified key process bottlenecks with respect to economic feasibility and its environmental impact. The review has further identified current limitations and gaps in the knowledge, and also deliberates directions for future research and development of thermophilic biohydrogen production.

Keywords Thermophilic · Biohydrogen · Agricultural residues · Techno-economic analysis · LCA

Introduction

Our heavy dependency on fossil energy sources has created societal problems related to its inherent environmental pollution. It urges for alternative, clean and renewable sources, fuelling huge research interests among scientists, albeit without any

economic success so far. The utopian world of energy sufficiency, without any hazardous emissions, is thought to be plausible by many, if only renewable hydrogen (H₂) could replace fossil-based energy carriers (Schrope 2001). As early as 1874, Jules Verne fancied the concept of ‘Hydrogen economy’ in his book *The Mysterious Island* (Hoffmann 2001). Indeed, as a fuel, H₂ has many desirable properties, among others, rapid burning speed, higher energy yield, low minimum ignition point and very high octane number (Ingersoll 1996; Mu et al. 2006; Balat and Kirtay 2010; Luque et al. 2011). However, introducing H₂ into the society faces several key technical barriers including storage, delivery and its end-user applications (Balat and Kirtay 2010).

By far, petroleum refineries are the largest producers (non-merchant) and consumers of H₂ (Freedonia 2010). However, the increasing demand of low sulphur and clean-burning fossil fuels can be a driver for renewable (merchant) H₂ applied in petroleum refineries (Freedonia 2010). Moreover, other applications of H₂ as a non-fuel commodity in chemical manufacturing, glass making, heat treatment of metals and hydrogenation of processed foods will also contribute to higher near-future demands for renewable (merchant) H₂. In a recent study, it has been estimated that the cost price of H₂ as a non-fuel is about 1–2 €/kg H₂ as based on estimated oil prices in 2020 (Mansilla et al. 2012). Hence, it has become of the utmost importance to develop an efficient method to produce H₂ from renewable feedstocks. This review attempts to bring together the current status of thermophilic fermentative H₂ production as one of those methods. We also propose a list of preferred properties (A to I) that an ideal H₂-producing microorganism should possess. These properties are discussed throughout the review.

Microorganisms for thermophilic fermentative hydrogen production

Pure culture studies

Higher temperatures (≥ 60 °C) are energetically more favourable for biological H₂ production (Stams 1994), enabling thermophiles to reach higher H₂ yields than mesophiles

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(Schönheit and Schäfer 1995, property A). As a consequence, thermophiles produce fewer by-products, i.e. especially acetic acid as its generation is accompanied with formation of ATP (Kengen et al. 2009). Moreover, strictly anaerobic thermophilic conditions seem to restrict contamination by hydrogenotrophic methanogens. In general, thermophilic H₂ producers have also higher H₂ tolerances; however, the latter varies depending on the sugar(s) present in the feedstock (Willquist et al. 2011). Even so, a drawback of thermophiles is their relatively low volumetric productivity, as their tendency to grow in lower cell densities in suspension cultures than mesophiles (Chou et al. 2008). The highest biological H₂ productivities ever reported have been for mesophilic cultures (Das 2009), but their accompanying low H₂ yields remain a critical problem.

Thermophilic H₂ producers are found within both the bacterial and the archaeal domain, and several of them have been characterized with their genome annotated (see list in VanFossen et al. 2008). Most of them are able to hydrolyse various polysaccharides (Blumer-Schuetz et al. 2008) and can ferment the released hexoses and pentoses to H₂ with yields close to the theoretical maximum (or Thauer limit, 4 mol H₂/mol hexose; Thauer et al. 1977). Recent reviews list many of those microorganisms involved (Kengen et al. 2009; van Niel et al. 2011), so here, only a selection of the best performers is given (Tables 1 and 2). Most of these organisms were isolated from extremely hot and reducing conditions (Fiala and Stetter 1986; Huber et al. 1986; Rainey et al. 1994; Xue et al. 2001; Mäkinen et al. 2009), under which they produce reduced metabolic end products, including H₂, as an electron sink for reducing equivalents (Thauer et al. 1977).

Under stressful conditions of high H₂ partial pressures and/or high medium osmolality, organisms tend to shift their metabolism to other reduced end products such as lactate, ethanol and alanine (Table 2; Kengen et al. 2009; Willquist et al. 2009, 2011), which in turn affect the H₂ yield negatively. Although, when purified, these by-products have a reasonable market value for a variety of purposes, but their concentration in the effluent will be too low for an economically viable downstream process. Alternatively, microorganisms can be engineered to produce more valuable product under stress (property G). On the other hand, the by-products can be fed to a complementary process, which can further produce valuable products including H₂ and methane (Hallenbeck and Ghosh 2009).

Defined and undefined culture studies

As an alternative to pure cultures, enrichment cultures can be employed for H₂ production. Such enrichment cultures are usually obtained from methanogenic anaerobic digesters. In some cases, household or municipal waste is also used for enriching hydrogenogenic microorganisms (Table 3). Advantages of enriched consortia are their (1) higher robustness to fluctuations in the fermentation process, (2) proneness to form biofilms and (3) are able to tackle more different substrates, thus improving conversion efficiencies (Brenner et al. 2008). In addition, at industrial scale, sterilization of feedstocks is not cost-effective; so during the fermentation, consortia can offer resilience to any contamination. On the other hand, consequences are that methods are required to suppress methanogenic activities (O-Thong et al. 2008a, b), and all studies with enrichments showing best performances have H₂ yields hardly exceeding 2.5 mol H₂/mol hexose (first four rows in Table 3). The latter could be inherent to the history of the inoculum containing also non-hydrogen-producing microorganisms (Chaganti et al. 2012; Kargi et al. 2012). This might not be an issue for biowaste streams as feedstock, but it is of importance when considering more costly energy crops.

Besides pure culture and undefined cultures, an interesting third option is to design cocktails of H₂-producing strains that either possess complementary sugar preferences or display synergies, thereby increasing the H₂ yield and conversion efficiencies. In this respect, only few studies have been carried out so far for thermophilic H₂ production. Liu et al. (2008) investigated a natural co-culture of *Clostridium thermocellum* and *Thermoanaerobacterium thermosaccharolyticum* isolated from decomposing wheat straw. In batch fermentations on cellulose, the co-culture reached higher H₂ yields (1.8 mol H₂/mol glucose) than *Clostridium thermocellum* alone (0.8 mol H₂/mol glucose) mainly due to higher conversion efficiency. *C. thermocellum* hydrolyzed the cellulose but was not able to consume all glucose and cellobiose. *T. thermosaccharolyticum* thus fermented part of the sugar and possibly the lactate produced by the other partner (Liu et al. 2008). In another study, synergies were found between two *Caldicellulosiruptor* species each originating from a different habitat (Zeidan et al. 2010). The synergy could be based on the excretion of one or several compounds by one of the species, *Caldicellulosiruptor saccharolyticus* that stimulated both the growth rate and biomass yield of the second

Table 1 Overview of thermophilic hydrogen producing microorganisms (continued from Kengen et al. 2009)

Organism	Domain	T_{opt} (°C)	Cultivation	Substrate	Y_{H_2} mmol/mmol C6	References
<i>Thermobrachium celere</i>	Bacteria	67	Batch	Glucose	3.36	(Ciranna et al. 2011)
<i>Clostridium stercorarium</i> DSM 2910	Bacteria	58	Continuous	Lactose	1.57	(Collet et al. 2004)
<i>Thermovorax subterraneus</i>	Bacteria	70	Batch	Glucose	1.4	(Mäkinen et al. 2009)

Table 2 Metabolic features of thermophilic hydrogen producers (modified and continued from Chou et al. 2008)

Organism	Fermentability of feedstocks/polymers	CCR	Auxotrophy to amino acids	Electron carriers	Hydrogenase ^a	Reductant sink	References
<i>Clostridia</i> (<i>Cl. thermocellum</i>)	Starch, cellulose, lignocellulose	Yes	No	NADH, ferredoxin	Uptake, Fe-only, FNOR	Alcohol, organic acids, lactate	Johnson et al. (1981), Desvaux (2006)
<i>Thermococcales</i> (<i>Pyrococcus furiosus</i>)	Maltose, cellobiose, β -glucans, starch	No	Yes	Ferredoxin	MBH, NiFe-only, FNOR	Alanine, ethanol	Hoaki et al. (1994), Maeder et al. (1999), Silva et al. (2000), Robb et al. (2001)
<i>Thermotogales</i> (<i>T. maritima</i> / <i>T. neapolitana</i>)	Cellulose, xylan, starch, cellobiose, lignocellulose	Yes	No	NADH, ferredoxin	Fe-only, NMOR, FNOR	Lactate, alanine	Schönheit and Schäfer (1995), Vargas and Noll (1996), Rinker and Kelly (2000), Bonch-Osmolovskaya (2001)
<i>Caldicellulosiruptor</i> (<i>C. saccharolyticus</i>)	Cellulose (avicel, amorp.), xylan, pectin, α -glucan, β -glucan, lignocellulose, guar gum	No	No	NADH, ferredoxin	Fe-only, NiFe-only	Lactate, ethanol	Rainey et al. (1994), de Vrije et al. (2007), van de Werken et al. (2008), Ivanova et al. (2008), Willquist and van Niel (2012)
<i>Thermoanaerobacter</i> (<i>T. tengcongensis</i> MBA)	Starch, sucrose, glycerol	Yes	Yes	NADH, Ferredoxin	Fe-only, NiFe-only	Ethanol	Xue et al. (2001), Warner and Lolkema (2003), Soboh et al. (2004)

CCR carbon catabolite repression

^a Types of hydrogenases—uptake, NiFe type hydrogen uptake hydrogenase, FNOR (ferredoxin:NAD(P)H oxidoreductase), Fe-only, Fe-only evolution hydrogenase, NiFe-only, NiFe-only evolution hydrogenase, NMOR (NADH:methylviologen oxidoreductase) and MBH (membrane-bound hydrogenase)

species *Caldicellulosiruptor kristjanssonii*. This co-culture revealed a remarkable stability in continuous culture even when sharing one carbon and energy source in the medium. Moreover, this co-culture possessed better H₂ yields (3.7 mol H₂/mol glucose) than either species alone (3.5 mol H₂/mol glucose) under the same conditions (Zeidan et al. 2010). As has been reviewed recently (Ren et al. 2011), other defined co-culture studies have been carried out focusing on improved thermophilic cellulose hydrolysis, in which synergistic mechanisms play a major role.

Enzymes of thermophilic hydrogen production

Hydrogenases

Evolution of H₂ through reduction of a proton is carried out by metalloenzymes, i.e. *hydrogenases*, which differ with respect to their size, structure, electron donors, and metal ions present in their active site (Meyer 2007; Vignais and Billoud 2007). Hydrogenases are known to be very sensitive to oxygen (O₂) (Vignais and Billoud 2007); even 1 % of O₂ can completely inhibit their H₂-forming capacity but not H₂ oxidation (Lukey et al. 2011). Hallenbeck and Gosh (2009) and Hallenbeck et al. (2012) argue that, for a variety of reasons, a limited amount of aerobic respiration along with fermentation may help achieve H₂ yields near the absolute maximum (12 mol/mol of hexose) through complete conversion of glucose to CO₂ via the TCA cycle. A recent study revealed successful

engineering of an O₂-tolerant [NiFe] *hydrogenase*, through site-directed mutagenesis (Lukey et al. 2011). In addition, native O₂-tolerant hydrogenases have been found in *Ralstonia eutropha* H16 (Burgdorf et al. 2005) and *Aquifex aeolicus* (Guiral et al. 2006). Such O₂-tolerant hydrogenases could be instrumental for performing micro-aerobic fermentations (property I). However, further research is needed to assess the validity of this hypothesis.

Hydrogenases use NADH or reduced ferredoxin (Fd_{red}) as electron donors, which are formed in the catabolism of organic substrates (Kengen et al. 2009). Under standard conditions, the mid-point redox potential for redox couples, NAD⁺/NADH and oxidized ferredoxin (Fd_{ox})/Fd_{red} is −320 and −398 mV, respectively (Thauer et al. 1977), which clearly indicates that Fd-dependent H₂ production is thermodynamically more favourable (property C). Alternatively, other relatively uncommon hydrogenases, such as ferredoxin:NAD(P)H *oxidoreductase* (FNOR, or electron-bifurcating hydrogenases) and membrane-bound hydrogenases (MBH), can also be desired for an ideal H₂ producer. FNOR produces H₂ using both NADH and Fd_{red} simultaneously, by coupling unfavourable oxidation of NADH with exergonic oxidation of Fd_{red} (Schut and Adams 2009), whereas MBH conserves valuable energy by coupling H₂ evolution to ATP synthesis via proton translocation (Sapra et al. 2003).

Redox enzymes

The central carbon metabolism of thermophilic H₂ producers has diverse metabolic pathways to reduce electron carriers, i.e. Fd or NAD⁺. Bacterial H₂ producers oxidize glyceraldehyde-

Table 3 Selection of thermophilic H₂ production using mixed/pure culture in various reactor types and/or industrial media. First four are best cases using model substrates

Reactor type	Conditions	Feedstock/substrate		Organism/source of inoculum for mixed culture	Enrichment with complex substrate	H ₂ yield ml-H ₂ /gVS mol/C6 mol	Q _{H2} (mM/h)	References	
		Method of cultivation	HRT (h)						T (°C)
CSTR	Continuous	2.86	72	6.7 (C)	Glucose	ND	3.0	12.4	de Vrije et al. (2007)
CSTR+carrier	Continuous	3	58	6 (C)	Glucose	ND	1.54	45.80	Koskinen et al. (2008)
Gas lift fermentor	Continuous	5	85	6 (C)	Pyruvate	ND	2.18†	9.46	Kanai et al. (2005)
UASB	Continuous	0.75	60	5 (C)	Sucrose	ND	1.3	152	O-Thong et al. (2008a, b)
UASB	Continuous	24	70	5.1 (nC)	Wheat straw hydrolysate	89.00	ND	1.52	Kongjian et al. (2010a, b)
CSTR	Continuous	24	70	7 (nC)	Pig slurry	ND	ND	~4.6	Kotsopoulos et al. (2009)
CSTR	Continuous	24	55	5.25 (nC)	Rapeseed straw stillage from ethanol plant	40.00	ND	6.04	Luo et al. (2011)
CSTR	Continuous	12	60	6.8 (C)	Sugar factory waste water	ND	2.5	8.30	Ueno et al. (1996)
CSTR	Continuous	4	60	5.5 (C)	Tofu waste water+glucose	ND	2.3	20.70	Kim and Lee (2010)
biofilm	Continuous	3	55	5 (C)	Sucrose	ND	1.59	4.66	Keskin et al. (2011)
Anaerobic filter	Continuous	24	70	5.4 (nC)	Wheat straw hydrolysate	ND	ND	0.85	Kongjian et al. (2010a, b)
Membrane bioreactor	Continuous	4	60	5.5 (C)	Tofu waste water	ND	1.45	34.25	Kim et al. (2011a, b)
Upflow anaerobic	Continuous	2	55	5.5 (C)	Rice winery wastewater	ND	1.9	3.81	Yu et al. (2002)
Semi-continuous	Continuous	16	60	5.5 (nC)	Cassava stillage	56.70	ND	6.21	Luo et al. (2010)
UASB	Continuous	24	55	nd	De-sugared molasses	159.60	ND	7.76	Kongjian et al. (2011)
EGSB	Continuous	6	70	nd	Glucose, arabinose	ND	ND	4.66	Abreu et al. (2010)
ASBR	Batch	96	60	5.5 (C)	palm oil mill effluent (POME)	ND	2.60	1.08	O-Thong et al. (2008a, b)
ASBR	Batch	48	60	5.5 (C)	POME	ND	ND	16.90	Prasertsan et al. (2009)

EGSB expanded granular sludge blanket, C pH controlled, nC pH not controlled, ND not determined, † estimated

3-phosphate (GAP) via *GAP dehydrogenase* generating one ATP and one NADH in the reaction. However, re-oxidation of the latter to H_2 is inherent to a thermodynamic constraint and thus instead it might easily be oxidized to undesired electron sinks, such as lactate and/or ethanol (for details, see Bielen et al. 2013). In contrast, archaeal H_2 producers have a unique enzyme, *GAP oxidoreductase* (GAPOR), which oxidizes GAP generating one Fd_{red} but no ATP. Thus, introduction of GAPOR in bacterial H_2 producers may redirect more pyruvate flux towards acetate generating the required ATP and will consequently improve H_2 yields. In addition, a host of other redox enzymes may also be involved in oxidation of substrates other than conventional sugars such as glycerol or rhamnose.

Similarly, at the pyruvate node, most of the distinguished thermophilic H_2 producers possess *pyruvate:ferredoxin oxidoreductase* (PFOR), which oxidizes pyruvate to generate Fd_{red} (Carere et al. 2012). In contrast, most mesophilic H_2 producers possess *pyruvate:formate lyase* (PFL) which generates formate (Carere et al. 2012). Some mesophilic organisms containing PFL also possess FHL to oxidize formate to CO_2 and H_2 . Nevertheless, PFOR remains a better enzyme for oxidation of pyruvate, contributing to higher H_2 yields in thermophilic H_2 producers.

Reactors and culture conditions applied for thermophilic biohydrogen production

Conventional and advanced bioreactors

A majority of the research on thermophilic H_2 production is directed to determining physiological characteristics of the microorganisms involved (Kengen et al. 2009). This requires well-controlled laboratory conditions; hence, the continuous stirred tank reactor (CSTR) combined with sparging gas (usually N_2) is the obvious choice. However, as a system cultivating cells in suspension, the CSTR does not allow biomass retention thus restricting the extent of substrate conversion and also limiting the hydraulic retention time (HRT), making low productivities inherent to this system. In recent years, there have been several investigations of advanced bioreactor systems that allow biomass retention and low HRT, of which a selection based on best performance is presented in Table 3 (for a more extensive list of reactor studies, see Ren et al. 2011). The CSTR clearly has an upper productivity limit of about 20 mmol $H_2/L/h$, which can be further improved to about threefold if cells are immobilized on a carrier (Koskinen et al. 2008; Table 3). Yet, to make the process economically feasible, the productivity should be at least an order of magnitude higher. Interestingly, a comparative study between an upflow anaerobic sludge blanket reactor (UASB) and CSTR revealed that in the former biomass retention of a pure culture

of a thermophile can increase productivity by nearly 15-folds (O-Thong et al. 2008a, b; Table 3). However, it must be noted that, since H_2 production is a growth-dependent phenomenon for most of the thermophiles (Schröder et al. 1994; Schönheit and Schäfer 1995; van Niel et al. 2003), the strategy of biomass retention usually results in lower H_2 yields (Table 3) compared to suspension cultures in CSTR (Table 1; Kengen et al. 2009). Nevertheless, biomass retention is required to achieve high conversion efficiencies. Although not studied in depth yet, other promising reactor configurations are based on the trickle bed reactor (van Groenestijn et al. 2002; Oh et al. 2004) and membrane bioreactor (Kim et al. 2011a, b). Like with the UASB, these reactor systems can operate without a sparging gas, which will significantly simplify the process and reduce operation costs. Instead, back-mixing is a promising alternative to sparging gas, for which the recycle ratio will be a crucial parameter to optimize H_2 production (Fontes Lima and Zaiat 2012).

Culture conditions

So far, the best H_2 production performances have been observed at neutral to slightly acidic pH, but in the presence of pH control (Table 3). Generation of volatile fatty acids as by-products decreases the pH of the fermentation medium, which can be corrected with an alkaline agent. However, the latter causes significant environmental impact (Ochs et al. 2010) and also restricts water recirculation due to accumulation of salts. Moreover, addition of caustic agents like sodium hydroxide incurs significant costs (Ljunggren et al. 2011a, b). When using expensive feedstocks such as lignocellulosic hydrolysates, it might require pH control to keep up better H_2 yields and productivities. In that case, it will remain a challenge to find cheap caustic agents and how to deal with the environmental burden of the fermentation effluent. Most of the cultures performed without any pH control have been achieved with undefined consortia, wherein thermophilic acetoclastic methanogens may have helped in maintaining the pH in a suitable range by consuming acetic acid. In some cases, the feed were supplemented with cheap caustic agents, such as sodium bicarbonate or urea (O-Thong et al. 2008a, b; Kongjan and Angelidaki 2010; Kongjan et al. 2010a, b; Abreu et al. 2012). Alternatively, performing fermentations at slightly acidic pH (~pH 6) may also ensure that lesser amounts of alkaline agent are added to the medium. To increase the productivity further, the substrate concentration in the reactor can be increased. However, this will require a H_2 producer that can withstand osmotic pressure exerted by high substrate/product concentrations (property H). In addition, various different modes of reactor operation—batch, continuous, semi-continuous and anaerobic sequential batch reactor (ASBR)—have been evaluated (Table 3) to increase H_2 productivities, of which the continuous mode of operation in particular has been the preferred choice (Table 3).

Complex media

Complex substrates, such as yeast extract or peptone, are regularly used as nutritional supplements to aid the growth of microorganisms at lab scale. Apart from providing amino acids, these media supplements provide buffering capacity, reducing agents and chelators for metal ions. So far, most of the physiological studies on thermophilic H₂ producers have been performed containing such complex substrates (Kengen et al. 2009; van Niel et al. 2011). Moreover, some of the applied studies for more practical evaluation of biohydrogen production have also been performed using such substrates (Table 3). However, use of yeast extract and/or peptone can incur significant production cost in any industrial process (Ljunggren and Zacchi 2010). Hence, an organism with the ability to synthesize all the amino acids will allow omission of complex substrates from the medium and thus help reduce the costs (property D).

Appropriate feedstock

Over the years, a variety of readily available feedstocks including industrial and municipal waste streams, glycerol from biodiesel production and various lignocellulosic materials have been evaluated for H₂ production (Table 3) with reasonable success. Lignocellulosic materials generally consist of a range of crop residues, dedicated energy crops, saw dust, forest residues and solid animal waste. For a more extensive list of industrial substrates used for biological H₂ production, see van Niel et al. (2011). It is often difficult to estimate the extent of future usage of these feedstocks due to their heterogeneous nature, uncertainties in their availability and sustainable recoverability, and their competing traditional applications (Gregg and Smith 2010; Rosillo-Calle and Woods 2012). Nevertheless, crop residues are estimated to be about 10¹⁰ tons/year globally (Lal 2005) and hence are increasingly considered as a potential feedstock for biological H₂ production.

Lignocellulosic feedstocks largely contain lignin, hemicellulose and cellulose, albeit in diverse fractions depending on the nature of the feedstock (Sun and Cheng 2002). Various physico-chemical methods are available to separate lignin from hemicellulose and cellulose. Solubilized polymers of hemicellulose and cellulose are further hydrolysed to mono- or disaccharides depending on the enzymes used for hydrolysis (Sun and Cheng 2002). Application of thermophilic, hydrolytic enzymes will allow integration of hydrolysis and fermentation together in a single step, i.e. simultaneous saccharification and fermentation (SSF). *Caldicellulosiruptor saccharolyticus*, *Clostridium thermocellum* and *Thermotogales* are known to secrete hydrolytic enzymes required for hydrolysis of pretreated lignocellulosic materials into monosaccharides (Table 2; property E), which, therefore, become ideal candidates for such a consolidated process. Alternatively, hydrolytic enzymes produced by

these organisms during fermentations can be separated from the effluent and used for the hydrolysis, minimizing the cost for enzymes. However, the cost of separation and re-usability of 'spent' effluent remains to be studied to conclude its feasibility.

A diverse fraction of monosaccharides, such as glucose, xylose, arabinose, mannose, galactose and uronic acid, are obtained upon hydrolysis of pretreated lignocellulosic materials (Maris et al. 2006). Organisms having a diverse catabolic range for sugars will strengthen the robustness of the process and allows flexibility in the choice of feedstock. This will be of particular importance considering the seasonal and unpredictable availability of agricultural residues. Moreover, co-utilization of sugars present in the hydrolysate is very important for economically viable process. Thus, organisms having a natural ability to co-utilize the sugars will ideally be preferred (property F) over organisms unable to do so owing to 'carbon catabolite repression'.

LCA/economical feasibility studies inherent on process development

So far, hardly any attempts have been made to evaluate the potential of any existing thermophilic biohydrogen production technology on a scale beyond that of laboratory studies. Nevertheless, a few techno-economic and life cycle analyses (LCA) have been performed using available literature to identify potential bottlenecks from environmental as well as techno-economical perspectives and steer the research towards pre-emptive measures.

Life cycle analysis

LCA involves assessment of environmental impact of different stages of a product's life cycle typically from cradle-to-grave. Ochs et al. (2010) performed a LCA evaluation (cradle-to-gate) of a proposed plant for thermophilic production of biohydrogen using potato steam peels under the assumption of a complete substrate oxidation to produce only CO₂ and sewage as by-products.¹ The study revealed that, during thermophilic fermentation, process inputs such as phosphates and alkali produced using fossil fuels are the most potential contributors to high environmental impact.² Moreover, as discussed earlier, the presence of excessive salts in the growth medium can restrict the recirculation of process water, which can add to the environmental impact. Hence, measures are needed to be taken to minimize the usage of phosphate buffers in the growth medium as well as evaluating strategies for

¹ Authors assumed a complementary step of photo-fermentation for further oxidation of DF by-products.

² Environmental impact for pretreatment of a biomass will vary depending on the nature of biomass and the method of pretreatment used. Hence, the pretreatment phase has been omitted from the discussion.

minimizing addition of alkali agents during fermentations. In addition, to minimize environmental impact, a complementary process capable of converting the generated by-products present in the effluent, thereby reducing the chemical oxygen demand, is an absolute requirement. A recent study reports about 93 % reduction in COD after converting the effluent to methane via anaerobic digestion (Willquist et al. 2012).

Techno-economic evaluation

Techno-economic analysis assesses the technical feasibility of the different parts involved in the process and also the effect of different parameters on the cost of production with the help of computer programs such as Aspen Plus (Aspen Technology, Burlington, USA). Recent technological advancements allow heat recovery in the fermentation step. Such being the case, when compared to mesophilic fermentation, additional heat demand required in thermophilic fermentation did not incur significantly higher costs (Ljunggren and Zacchi 2010). On the other hand, the production cost is largely influenced by (1) the cost of media ingredients and (2) low substrate (sugar) concentrations (Ljunggren and Zacchi 2010). As discussed above, yeast extract is the most expensive component of the medium and is not needed by the H₂ producers having the ability to synthesize most of the growth factors present in the yeast extract. Secondly, low substrate concentrations in the medium will require larger reactors along with larger facility and consequently will demand more water and energy. Increasing the substrate concentration may not be a quick and easy solution, as it increases the osmolality of the medium causing undesirable effects on microbial biomass and H₂ yields (Ljunggren et al. 2011a, b).

Other aspects

Given that H₂ needs a unique and costly distribution infrastructure, a decentralized model of production can be imagined for a biomass-dependent thermophilic H₂ process. A decentralized production will also benefit from a locally available market. Alternatively, thermophilic biohydrogen can be produced in an add-on plant to another industrial process. For example, by-products and waste heat of a sugar factory can be used for the production of biohydrogen (Markowski et al. 2010).

Challenges and outlook

Up till now, the physiology of thermophilic H₂ producers has been studied to a reasonable depth for a few thermophilic H₂ producers only (Verhaart et al. 2010; Willquist et al. 2010; van Niel et al. 2011). Insight into the preferences and the physiological boundaries of H₂ producers will facilitate reactor

design and optimal operation conditions. Genetic engineering is an important tool to obtain required knowledge in the most convenient way. In spite of the availability of vectors for genetic modification, no major breakthrough has been reported for the genetic modification of distinguished thermophilic H₂ producers (Desai et al. 2004; Tyurin et al. 2004; Waage et al. 2010; Han et al. 2012; Chung et al. 2013). Indeed, challenges for performing proper modifications are related mainly to the practical hurdles inherent to these organisms, such as strict anaerobic nature limiting their ability to grow on solid media, scarcity of selection markers and unique defence mechanisms restricting transformation with foreign DNA (Noll and Vargas 1997; Thomas and Nielsen 2005; Chung et al. 2012). Knowledge obtained here gives feedback to (1) the essential composition of the feedstocks, (2) under what conditions the reactor should be operating and (3) what kind of by-products can be expected. Next to wet experiments, *in silico* experiments using genome-scale metabolic models will facilitate obtaining this knowledge and, moreover, identifying new metabolic engineering strategies. These metabolic models are now becoming available for several of the thermophilic H₂ producers (Zhang et al. 2009; Roberts et al. 2010; Munro et al. 2011; Zeidan 2011; Nogales et al. 2012).

One such new strategy could be to engineer H₂-producing pathways in thermophiles to increase H₂ yields beyond the current limit of 4 to up to 8 mol H₂/mole hexose, such as the oxidative pentose phosphate pathway (OPPP) (Hallenbeck and Benemann 2002; de Vrije et al. 2007). In simulations with a metabolic model of *Thermotoga maritima* this is possible, provided that the NADPH produced in the OPPP is either reoxidized with an introduced NADPH-NADH *transhydrogenase* or a NADPH-Fd *reductase* (Nogales et al. 2012). The outcome of significantly improved H₂ yields in the models was based on optimizing for H₂ production. However, it is expected that in reality most organisms will naturally choose for optimizing their growth rate, which then results in a marginal improvement of H₂ yields. Indeed, introducing non-native redox pathways hardly improved H₂ yields (Kim et al. 2011a, b). This is mainly attributed to thermodynamic barriers, which is not covered by the current genome-wide metabolic models, showing, as yet, the limited power of these models.

The majority of H₂ producers have been isolated from environments low in free carbohydrates and usually possess variety of hydrolases to breakdown polysaccharides. Mono- and disaccharides are released slowly, being often the rate-determining step of growth. As a consequence, the organisms are exposed only to low sugar concentrations and thus relatively sugar-sensitive strains are selected (Willquist et al. 2010). However, when applied in bioreactors, growth of these organisms is influenced strongly by osmotic pressure exerted by high sugar concentrations (Ljunggren et al. 2011a, b). For industrial application, this is a negative characteristic, for which solutions have to be found. Among several options,

SSF is probably the most practical one. Indeed, a positive effect of SSF for H₂ production has been described recently (Quéméneur et al. 2012), but more studies are necessary to evaluate whether it is applicable for thermophilic H₂ production. Alternatively, osmotolerant strains can be used (property H), which can be obtained by either genetic transformation or evolutionary adaptation. This will allow high substrate loading rates, thus lower the water demand, and it might keep any contamination further at bay.

So far, reactor design for thermophilic H₂ production is still in the lab-scale phase. The trials show low to high volumetric productivities, but usually are combined with low H₂ yields (Oh et al. 2004). Owing to the trade-off between H₂ productivity and yield, a choice should be made between higher productivity and higher yield to make the process most profitable. Generally, for a process using inexpensive raw materials, H₂ productivity can be given more importance than H₂ yields, and vice versa for processes utilizing rather expensive raw materials. The level of productivity normally depends on the use of pure cultures or undefined consortia, type of reactor (configuration), efficiency of H₂ removal and type of feedstock. For economic reasons, volumetric productivities need to be high, making the choice of bioreactor type crucial. At best, the reactor should allow for short HRT, high biomass

retention times and fast removal of H₂, hence trickling filters and UASB-type reactors are among the best choices (van Niel et al. 2011). These reactors are designed for long biomass retention times and H₂ can be removed effectively through optimal back mixing or sparging gas. However, the latter complicates gas handling and downstream processing and thus adds to extra cost. To ensure high hydrogen yields, thermophiles possessing an acceptable high hydrogen tolerance should be applied in these reactors, e.g. up to 60 kPa (Willquist et al. 2011). A reactor coupled to a selective membrane allowing for in situ removal of H₂ is another promising concept as it has seen to increase volumetric productivities (Lee et al. 2007). H₂ can be separated effectively with dense ceramic membranes (Lu et al. 2007) and substantial experience how to overcome limitations with these kinds of membranes, such as bio-fouling and energy demand, has been gained in wastewater treatment (Judd 2008). Therefore, ceramic membrane-based bioreactors for thermophilic H₂ production could have promising potential.

Conclusions

The last few decades of research on thermophilic biohydrogen producers have given plentiful insights into their physiology and intricate metabolism. While microbial physiologists will continue to explore the unknowns and/or continue to modify the known organisms in search of ideal H₂-producing microorganism, we propose a combination of features that such an ideal H₂ producer may possess: (A) thermophilic, (B) has specific vectors/tools designed for genetic modification(s), (C) possesses Fd-dependent hydrogenases, (D) is not auxotrophic to any amino acids, (E) has ability to degrade a wide range of biomass, (F) can metabolize multiple sugars simultaneously (absence of carbon catabolite repression), (G) when under stress shifts metabolism to useful by-products, (H) is tolerant to high osmotic stress exerted by high substrate/by-product concentrations and (I) is oxygen-tolerant. Amongst the distinguished H₂ producers, organisms belonging to the genera *Caldicellulosiruptor* and *Thermotoga* come closest to being ideal H₂ producers (Fig. 1). Alternatively, a consortium of known microorganisms might be designed possessing together all the features listed above.

Process engineers will keep on testing more types of feedstocks and reactor configurations to enhance productivities and yields, whereas systems analysts and economists continue scrutinizing new available published data to merit processes on their environmental impact and cost-effectiveness. However, the majority of the current challenges can only be overcome through intensive cross-disciplinary collaboration, thus ensuring essential synergies for the development of a commercial thermophilic H₂ production process. And probably the economic feasibility outcomes may very well indicate that

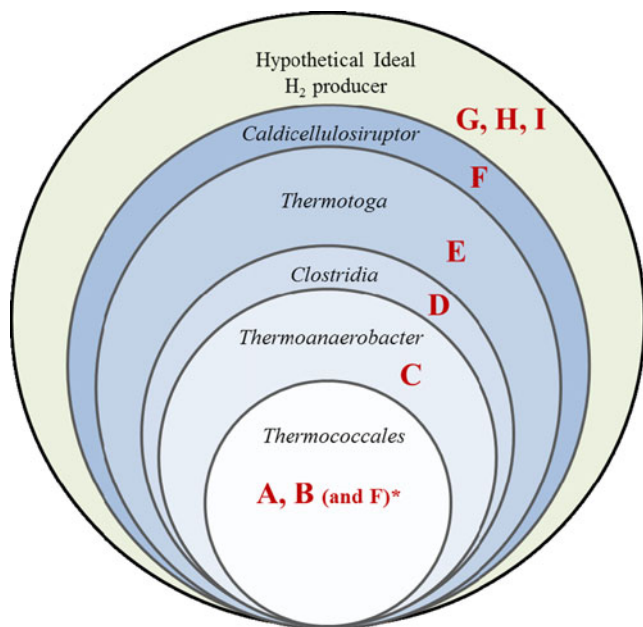


Fig. 1 A Venn diagram displaying comparison between distinguished H₂ producers with respect to desirable properties an ideal H₂ producer may possess. A, thermophilic; B, has specific vectors/tools designed for genetic modification(s); C, possesses Fd-dependent hydrogenases; D, is not auxotrophic to any amino acids; E, has ability to degrade a wide range of biomass; F, can metabolize multiple sugars simultaneously (absence of carbon catabolite repression); G, when under stress shifts metabolism to useful by-products; H, is tolerant to high osmotic stress exerted by high substrate/by-product concentrations and I, is oxygen-tolerant. (Asterisk, note: property F is also present in *Thermococcales* and is indeed absent from other genera as depicted)

thermophilic biological H₂ production best fits into a biorefinery process (Willquist et al. 2012).

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Conflict of interest The authors declare that they have no conflict of interest.

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