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## Highly Stereoselective Synthesis of Cyclopentanes bearing Four Stereocenters by a Rhodium Carbene–Initiated Domino Sequence

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### Abstract

Stereoselective synthesis of a cyclopentane nucleus by convergent annulations constitutes a significant challenge for synthetic chemists. Though a number of biologically relevant cyclopentane natural products are known, more often than not, the cyclopentane core is assembled in a stepwise fashion due to lack of efficient annulation strategies. Herein, we report the rhodium-catalyzed reactions of vinyl diazoacetates with (*E*)-1,3-disubstituted 2-butenols generate cyclopentanes, containing four new stereogenic centers with very high levels of stereoselectivity (99% ee, >97 : 3 dr). The reaction proceeds by a carbene-initiated domino sequence consisting of five distinct steps: rhodium-bound oxonium ylide formation, [2,3]-sigmatropic rearrangement, oxy-Cope rearrangement, enol-keto tautomerization, and finally an intramolecular carbonyl ene reaction. A systematic study is presented detailing how to control chirality transfer in each of the four stereo-defining steps of the cascade, culminating in the development of a highly stereoselective process.

### INTRODUCTION

Construction of cyclopentanes constitutes an area of significant interest in the organic community,<sup>1–3</sup> as evidenced by numerous campaigns toward landmark natural products, including the prostaglandins,<sup>4–16</sup> jatrophanes,<sup>17–25</sup> and pactamycin.<sup>26,27</sup> Synthetic cyclopentanes are a central motif common to prostaglandin antagonist antiglaucoma agents.<sup>28–32</sup> Indeed, the convergent, asymmetric synthesis of a cyclopentane nucleus bearing multiple chiral centers presents a distinct challenge for modern synthetic chemists. A general difficulty for a [3 + 2]-cycloaddition approach is developing a mild protocol to render a

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#### AUTHOR CONTRIBUTIONS

B.T.P. designed and performed experiments, analyzed data, and co-wrote the manuscript. H.M.L.D. supervised the project and co-wrote the manuscript.

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The X-ray crystallographic coordinates for structure **24a** have been deposited at the Cambridge Crystallographic Data Centre (CCDC), under deposition number CCDC 955223. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

#### FINANCIAL INTERESTS

The authors declare no competing financial interests.

transient carbon dipole. One of the classic strategies for chiral cyclopentene synthesis involves stereoselective cyclopropanation of a 1,3-diene, followed by vinylcyclopropane rearrangement.<sup>33–36</sup> Contemporary efforts have involved the coupling of a suitably activated olefinic motif with an all carbon (1,3)-dipole bound to a chiral metal center via a formal [3+2]-cycloaddition.<sup>37–40</sup> A significant advancement has been the enantioselective cycloaddition of palladated trimethylenemethane with electron-deficient alkenes developed by Trost and co-workers.<sup>41–43</sup> Alternatively, Fu<sup>44,45</sup> and Zhang<sup>46</sup> have disclosed enantioselective variants of the Lu cycloaddition<sup>47–53</sup> implementing chiral phosphine catalysts. Though a number of methods are available, the development of complementary strategies for the rapid construction of complex cyclopentanes is nonetheless desirable.

In this article, we describe a convergent, stereoselective synthesis of cyclopentanes with four stereogenic centers from an annulation cascade involving vinyl diazoacetates and chiral allyl alcohols. Moreover, a detailed analysis to account for correlations between starting material architecture and reaction efficacy is described.

## RESULTS

### 1<sup>st</sup> Generation Synthesis

In a 2011 report, we described our initial attempts to synthesize complex cyclopentanes **3** by means of a cascade sequence.<sup>54</sup> The strategy involved the coupling of racemic 3,3-dimethyl allyl alcohols (**2**) and donor/acceptor-substituted Rh(II) carbenes<sup>55,56</sup> derived from vinyl diazoacetates **1** in the presence of the dual-catalyst system, Rh<sub>2</sub>(DOSP)<sub>4</sub> and Sc(OTf)<sub>3</sub>, shown in Figure 1a. Implementing 1 mole % of chiral Rh catalyst enabled a high yielding preparation of a range of cyclopentanes **3**, containing four stereogenic centers, as single diastereomers. Though we were pleased with the general efficacy with which a high degree of structural complexity was introduced from readily available building blocks, the reaction was characterized by a significant shortcoming: the enantioselectivity for the synthesis of **3** was variable (68–92% ee). In this paper we describe our diagnosis of the stereochemical limitations in our initial study, and the logical design of a stereocontrolled cyclopentane **5** synthesis, as illustrated in Figure 1b.

Our previous synthesis of cyclopentanes **3** involved four discrete, stereo-defining reactions, as illustrated for the specific example in Figure 2.<sup>52</sup> The steps include: Rh-bound oxonium ylide formation (**6** and **7** → **I**), [2,3]-sigmatropic rearrangement (**I** → **8**),<sup>55</sup> oxy-Cope rearrangement (**8** → **II** → **9**), and intramolecular carbonyl ene reaction (**9** → **III** → **10**). Although the metal-bound ylide intermediate **I** is neither isolated nor observed in the tandem reaction, its intermediacy is well precedented in the literature.<sup>54–60</sup> From previous studies on the tandem ylide formation/[2,3]-sigmatropic rearrangement<sup>61</sup> as well as control studies,<sup>52</sup> we have analyzed the stereochemically relevant steps in the cascade sequence. We observed excellent enantiocontrol during the rhodium-catalyzed process for generating the 3-hydroxy-1,5-hexadiene **8** (98% ee). Intermediate **8** then underwent a [3,3]-sigmatropic rearrangement through a chair-like transition-state (**II**). Upon a subsequent enol–keto tautomerization, the product **9**, containing two new stereocenters, was furnished. The oxy-Cope rearrangement afforded a single diastereomer of ketoester **9**; an observation consistent

with chair-like transition states being solely operative.<sup>52,61–64</sup> Isolation and chiral HPLC analysis of **9**, however, indicated partial racemization during the course of this step of the cascade sequence. The terminating sequence of the cascade – intramolecular carbonyl ene reaction via **III** – proceeded smoothly to furnish cyclopentane **10** as a single diastereomer. No further degradation in enantiopurity was observed during the ene reaction. These studies identify the origins of the moderate overall enantioinduction: modest chirality transfer during the oxy-Cope rearrangement.

### Stereochemical Limitations

A rationale for the erosion in enantiopurity observed during the course of the oxy-Cope rearrangement is provided schematically in Figure 3. Based upon cyclohexane *A*-values, the equatorial versus axial positioning of carbomethoxy (Figure 3a) and hydroxy (Figure 3b) substituents would be favored by 1.25 and 0.55 kcal mol<sup>-1</sup>, respectively.<sup>65</sup> Equilibration between two possible chair-like intermediates for the [3,3]-sigmatropic rearrangement, shown in Figure 3c, is more complicated. Both geminal disubstitution<sup>66</sup> and pseudoanomeric stabilization of the allyl hydroxy group<sup>67</sup> influence the equilibrium between **III** and **III'**. Based upon the *A*-values of the individual substituents and the pseudoanomeric stabilization afforded **III**, we can rationalize the modest preference for that pathway. Nevertheless, rearrangement via the energetically less favorable intermediate **III'** would also be feasible. Upon undergoing the Cope rearrangement, two of the stereogenic centers found in the cyclopentane (**10**), C(3) and C(4), have been installed. As can be seen from comparison of **9'** and *ent*-**9'**, the competing chair-like transition states give rise to enantiomeric intermediates. Thus, enantiomers of the cyclopentane, **10** and *ent*-**10**, arise from the competition between **III** and **III'**. The 82% ee observed for the cyclopentane **10** synthesis is due to approximately a ten-to-one preference for **III** over **III'**.

### Rationale for 2<sup>nd</sup> Generation Synthesis

Having analyzed the reaction responsible for partial racemization in the cascade sequence, we honed our efforts toward a general solution to the problematic step. On the basis of the hypothesis that competing chair-like transition states during the oxy-Cope rearrangement were the source of poor chirality transfer, we designed new systems that would suppress this equilibration and strongly favor one chair-like transition state over the other.

In a recent article, we described the tandem ylide formation/[2,3]-sigmatropic rearrangement of readily available chiral allyl alcohols with donor/acceptor substituted rhodium carbenes.<sup>75</sup> For alcohols with variable substitution at C(3) (Figure 4a, **11**, R<sup>1</sup> R<sup>2</sup>), rearrangement products bearing vicinal stereocenters (**12**) were generated with very high enantiocontrol (>99% ee) and moderately high diastereocontrol (12 – >20 : 1). Specifically, the rhodium catalyst dictated the configuration of the hydroxyester stereocenter (green sphere) while the allyl alcohol controlled the configuration of the second chiral center (blue sphere). Thus, we demonstrated that the proper combination of each enantiomer of Rh<sub>2</sub>(DOSP)<sub>4</sub> and allyl alcohol enabled the synthesis of all four diastereomers of **12**. We envisioned that the [2,3]-sigmatropic rearrangement to generate a product with adjacent stereocenters could provide the necessary solution to chirality transfer during the problematic oxy-Cope rearrangement, as shown in Scheme 4b. Specifically, the proper configuration of a 3-hydroxy-1,5-hexadiene

(**12**) would enable a single chair-like intermediate, wherein both chiral centers would orient cooperatively (**12'**), with the substituents having larger *A*-values positioned equatorially ( $R^1 > R^2$ ). Herein we report the synthesis-based exploration of that hypothesis and the development of a highly stereoselective cyclopentane synthesis.

## DISCUSSION

Our investigations into an improved cyclopentane synthesis began with a series of control reactions. First, the diastereo- and enantiomerically pure 3-hydroxy-1,5-hexadiene **13**<sup>75</sup> was heated in a sealed tube for 12 h. Direct purification by flash chromatography afforded a small amount of the ketoester **14** as a single stereoisomer (Figure 5a). The modest isolated yield of **14** is because the crude reaction mixture contains a mixture of starting material (**13**), ketoester (**14**), and cyclopentane (**15**). Prolonged heating of **13** furnished the desired cyclopentane **15** in excellent yield and with no loss in the level of enantioselectivity, as shown in Figure 5b. The product **15** was, however, formed as an 88 : 12 mixture of diastereomers. Having established that the keto ester **14** is formed as a single diastereomer, it appears that the carbonyl ene reaction to form **15** is no longer highly diastereoselective with these substrates. Comparison with literature precedent<sup>22</sup> and our own analyses of the nuclear Overhauser effects (see Supplementary Information/Supplementary Figures 1 and 2) in **15** are indicative of inversion in the configuration at the C(1)- and C(2)-stereocenters in the minor product.

In order for the Cope rearrangement to proceed with high enantiocontrol, the two chiral centers of the hexadiene must *cooperatively* reinforce a single chair-like transition state, as illustrated in Schemes 4a and b. In the negative control reaction with the epimeric material *epi*-**13**, the two chiral centers are orientated *competitively*, such that only the carbomethoxy *or* the methyl group can occupy an equatorial position in the two available chair-like forms (Figure 5c). The reaction with *epi*-**13** demonstrated the detrimental effect on enantiocontrol, as the cyclopentane **15** was obtained with considerable racemization (39% ee) due to competing chair-like transition states. The identical diastereomeric ratio of product cyclopentanes observed for the matched and mismatched reactions (Figure 5b and c) was again consistent with a problematic carbonyl ene reaction. These studies revealed that a second stereocenter could be manipulated for enantiocontrol in the oxy-Cope rearrangement, but generated new problems with regard to the diastereoselectivity of the ene reaction.

The control studies revealed that the rearrangement of the hexadienol **13** to cyclopentanol **15** could be achieved under thermal conditions without the use of scandium triflate as a Lewis acid catalyst. Therefore, modified conditions were developed for the one-pot process.  $\text{Rh}_2(\text{R-DOSP})_4$ -catalyzed reaction of vinyl diazoacetate **6** with the allyl alcohol **16** followed by heating of the crude mixture for 24 h afforded the desired product in a gratifying 71% yield (Figure 5d). While the diastereomeric ratio for the one-pot process was identical to that for the control reaction shown in Figure 5b, an unanticipated decrease in the level of enantioselectivity was observed in the matched reaction to form **15**. The level of enantioselectivity was 91% ee, which is an improvement over our previous studies,<sup>52</sup> but still not ideal. The one-pot mismatched reaction (Figure 5e), conducted by implementing the opposite enantiomer of catalyst  $[\text{Rh}_2(\text{S-DOSP})_4]$ , resulted in an overall less efficacious

synthesis of **15**, as expected. The enantioselectivity of the transformation was consistent with that observed during the tandem oxy-Cope/ene study (Figure 5c).

The discrepancy in enantioselectivity for the cyclopentane formation in the Cope/ene sequence (Figure 5b) versus the complete one-pot sequence (Figure 5d) is attributed to a slight diastereomeric “leakage” occurring during the [2,3]-sigmatropic rearrangement (Figure 5f). In our previous study, we rationalized the driving force for diastereoselection in the ylide formation/[2,3]-rearrangement to be minimization of  $A^{1,3}$ -strain in the ylide intermediate. While the strain-minimized transition state **IV** is preferred, for the pentenol **16**, the epimeric product (*epi-13*) is also formed (**13** : *epi-13* ratio, ~11 : 1), because the strain formed between a *syn*pentane methyl and hydrogen group in transition state **V** is not severe. According to the control study described in Figure 5c, the minor [2,3]-product *epi-13* will contribute to formation of the enantiomer of **15**.

Another challenge with the second-generation substrates was the formation of diastereomers during the ene reactions. The modest level of diastereoselectivity in this step can be rationalized by a transition state analysis. Placing the ene and enophile components of **14** in a *syn* orientation to allow for requisite orbital overlap renders two feasible transition states (Figure 6a, **VI** and **VII**). As with the [2,3]-sigmatropic rearrangement, minimization of  $A^{1,3}$ -strain is a governing factor in the carbonyl ene reaction. Since the ene consists of a *trans*-1,2-disubstituted alkene, the origins of any  $A^{1,3}$ -strain are from interaction between the vinylic proton and the pseudoequatorial allylic methyl group. Interestingly, a similar strain interaction occurring during the [2,3]-rearrangement ene reactions produces similar levels of diastereoselectivity. Thus, while diastereomer **15**, arising from cyclization via **VI** is preferred, some of the product *epi-15* originating from the slightly more strained transition state **VII** is also formed.

These studies verified both the feasibility of generating cyclopentanes in high levels of enantioselectivity via a one-pot process and the necessity of matching catalyst and alcohol chirality. We decided to explore whether the trend we observed with the allyl alcohol **16** would be extended to other secondary allyl alcohols. The diazoacetate **6** was chosen as the standard carbene precursor as it has been studied extensively in tandem ylide formation/[2,3]-sigmatropic rearrangement chemistry.<sup>52,55,68,69</sup> We found that incorporating superstoichiometric quantities of calcium chloride enabled a ten-fold reduction in the loading of dirhodium tetracarboxylate catalyst without any loss in asymmetric induction, and so, it was incorporated as part of the standard reaction conditions. We believe  $\text{CaCl}_2$  helps minimize competitive coordination of the alcohol to the vacant axial site of the dirhodium tetracarboxylate catalyst; a process which impedes denitrogenative decomposition of the diazoacetate (see Supplementary Information/Supplementary Table 1).

The first series of alcohols (**17a–d**) examined, like those in our previous cyclopentane synthesis,<sup>52</sup> varied in carbinol substitution (Table 1). We envisioned that an increase in steric bulk at the carbinol (compared with **16**) could enhance the diastereoselectivity of the [2,3]-sigmatropic rearrangement and, therefore, the overall enantioselectivity of the cyclopentane synthesis. Increasing the length of the alkyl chain (entries 1–2, **17a–b**) had no effect on the stereocontrol of the reaction. Introduction of more encumbered secondary

carbon or benzylether substituents (entries 3–4, **17c–d**, respectively) adjacent to the carbinol provided comparable yields and levels of enantioselectivity with a slight improvement in the levels of diastereoselectivity.

We then explored the effect of increasing the bulk of the alcohol C(3) substituent, which participates in  $A^{1,3}$ -interactions during the [2,3]-sigmatropic rearrangement. A modest improvement in enantioselectivity was observed when a C(3)-Me substituent (Scheme 5a, **16**) was replaced by an *isopropyl* moiety (Table 2, entry 1, **19a**,  $R^1 = R^2 = \text{Me}$ ). The diastereoselectivity, however, was similar to that observed for the previous class of allyl alcohols. For alcohols where  $R^1$  and  $R^2$  were not equivalent, an added element of complexity in the form of olefin geometry was introduced (entries 2–4, **19b–d**). The yields and levels of stereoselectivity were fairly consistent throughout the group; however, (*E*)-selective alkene formation was correlated with the size of  $R^1$ . Specifically, linear alkyl chains (**19b–c**) provided comparable ratios of *E*- and *Z*-diastereomers (entries 2–3, *ca.* 3 : 1, *E* : *Z*), but a benzyl substituent afforded cyclopentane (*E*)-**20d** in greater than 7 : 1 *E/Z* mixture.

The preference for the formation of the *E*-isomers of **20** is consistent with the transition state model presented for the ene reaction as illustrated in Scheme 8. Minimization of  $A^{1,3}$ -interactions dictates the major reaction pathway. By rotating the methylene carbon of the ene fragment (Figure 6b, **VIII** vs. **IX**), the allylic R-group would be positioned either *trans* or *cis* to the vinylic proton. The former, less strained intermediate would give rise to the (*E*)-cyclopentane **20**, and the latter to the less preferred (*Z*)-**20**.

On the basis of the experimental observations and the transition state analysis, it became apparent that a highly stereoselective entry to the cyclopentanes would require effective control of chair transition states in the oxy-Cope rearrangement and maximization of the  $A^{1,3}$ -strain control in the [2,3]-sigmatropic rearrangement and the ene reaction. Therefore, we examined a series of alcohols **23** bearing geminal disubstitution at the C(3) position (Table 3). One of the substituents was methyl to maximize chair selectivity in the oxy-Cope rearrangement and to minimize the formation of isomers in the ene reaction. The desired chiral alcohols (**23**) were readily prepared from commercially available enals or by means of carbometallation<sup>68,70</sup> of the requisite alkynes and trapping with acetaldehyde. The results of the cyclopentane formation from **23** are described in Table 3. In each case the reaction proceeds in very high yield (85–89%) and with excellent levels of stereoselectivity. The cyclopentanes **24** were formed in >30 : 1 dr and in 99% ee. The reaction is likely to be applicable to a range of vinyl diazoacetates as the alkyl-substituted vinyl diazoacetate (**22**,  $R = \text{Et}$ ), was similarly effective in the net transformation. The absolute configuration of cyclopentane **24a** was verified by X-ray crystallographic analysis and this assignment was applied to the other products by analogy. Moreover, the absolute configuration was consistent with the assignments in our earlier studies.<sup>58</sup>

Although the cyclopentane bearing a quaternary hydroxycarbonyl stereocenter is a common structural motif among numerous polyterpenoid natural products (e.g. jatrophane diterpenoids), we sought to demonstrate the ease with which the products could be converted to the corresponding cyclopentanones. To this end, reduction of the cyclopentane



carboxylate to the 1,2-diol was affected by lithium borohydride. Quenching with pH 7.0 buffer followed by direct treatment of the crude reaction mixture with excess sodium periodate furnished the cyclopentanone **25** in good yield with no observable epimerization of the  $\alpha$ -stereocenter (Table 3).

A detailed summary and analysis of the processes involved for stereoselective synthesis of **(R,R,R,S)-28** from alcohols such as **23** is presented in Figure 7. The four discrete steps involved are the following: oxonium ylide formation (**Step 1**), [2,3]-sigmatropic rearrangement (**Step 2**), oxy-Cope rearrangement (**Step 3**), and carbonyl ene reaction (**Step 4**).

The reaction sequence commences with nucleophilic addition of **23** to the rhodium-bound carbene intermediate derived from diazoacetate **6**. A high degree of enantiocontrol is exerted by the rhodium catalyst, dictating *si* approach to the metallocarbene.<sup>61,75,76</sup> Therefore, the analysis shown in Figure 7 is limited to compounds generated from *si* face attack. Due to the severe  $A^{1,3}$ -strain which develops between the allylic methyl substituents in **TS<sub>2,3</sub>1**, the reaction proceeds selectively through **TS<sub>2,3</sub>2**, affording the hexadiene diastereomer **(S,R)-26**. Rendering the reaction intermediate into a chair-like transition state produces **TS<sub>3,3</sub>3**, where the bulkiest substituents (CO<sub>2</sub>Me and R) are occupying equatorial positions. Proceeding through an oxy-Cope rearrangement with tandem enol-keto tautomerization provides the  $\alpha$ -ketoester **(S,S,E)-27**. As with the [2,3]-sigmatropic rearrangement, the driving force for diastereoselectivity in the carbonyl ene reaction is minimization of  $A^{1,3}$ -interactions. Thus, the transition state where the allylic methyl groups are oriented on opposite faces of the ensuing cyclopentane (**TS<sub>e</sub>-5**) is the dominant pathway. The result is the formation of the observed stereoisomer of the cyclopentane **(R,R,R,S)-28**.

In our previous study where the [2,3]-sigmatropic rearrangement products lacked a second stereocenter (**26**, R = Me), the 1,5-hexadiene is forged with excellent enantiocontrol. Nonetheless, during the oxy-Cope rearrangement **TS<sub>3,3</sub>4** becomes a viable, albeit minor, reaction pathway to give rise to **(R,R)-27** (R = Me). Since the allylic (ene) portion of the molecule is trisubstituted, it is still able to exploit a high degree of  $A^{1,3}$ -strain, resulting in dramatic preference for carbonyl ene reaction through **TS<sub>e</sub>8** rather than **TS<sub>e</sub>7**. As a result, the enantiomeric cyclopentane **(S,S,S,R)-28** is formed competitively, resulting in moderate enantiomeric excess for the overall synthesis. When the allylic alcohol was not 3,3-disubstituted (e.g. **16**, **17**, and **19**), **TS<sub>2,3</sub>1** becomes viable in the 2,3-sigmatropic rearrangement leading to the formation of some of the enantiomeric product **(S,S,S,R)-28**. Also, **TS<sub>e</sub>2** becomes viable in the ene reaction leading to the formation of some of the diastereomeric product **(S,S,R,S)-28**.

Thus, a domino reaction of rhodium vinylcarbenes and enantiopure allyl alcohols provides a convergent strategy for the synthesis of cyclopentane carboxylates. Employing just 0.1 mole % of chiral catalyst, three new bonds are formed, while installing four contiguous stereocenters in high yield with excellent diastereo- and enantioselectivity. A range of functional groups can be introduced at three of the stereogenic centers by variations in the allyl alcohol and vinyl diazoacetate. The fourth hydroxyester stereocenter is readily manipulated to generate complex cyclopentanones via a one-pot reduction/oxidation

procedure. Each discrete step in the reaction sequence involves distinct chirality transfer events, which can be manipulated by appropriate substitution of the alcohol and ensuing intermediates. A detailed understanding of the stereochemical processes involved in the cascade has been developed, allowing for exceptional control over each stereochemical transfer. The low catalyst loadings, readily available starting materials and in-depth understanding of the mechanistic details make this an attractive method for accessing substituted cyclopentanes.

## METHODS

### General Considerations

All reactions were conducted in oven-dried glassware under an inert atmosphere of dry argon. All chemicals were purchased from either Sigma-Aldrich, TCI America, Acros or Alfa-Aesar, and were used as received. Pentane, hexanes, tetrahydrofuran and diethyl ether were obtained from a Grubbs-type solvent purification system. Heptanes were purchased from Macron Fine Chemicals, and were used as received. Calcium chloride was dried at 200 °C under vacuum for 12 h and stored in a desiccator. <sup>1</sup>H NMR spectra were recorded at either 400 MHz on an INOVA-400 spectrometer or at 600 MHz on an INOVA-600 spectrometer. <sup>13</sup>C NMR spectra were recorded at 100 MHz on an INOVA-400 spectrometer. NMR spectra were recorded in deuterated chloroform (CDCl<sub>3</sub>) solutions, with residual chloroform ( $\delta$  7.27 ppm for <sup>1</sup>H NMR and  $\delta$  77.23 ppm for <sup>13</sup>C NMR) or tetramethylsilane ( $\delta$  0.00 ppm for <sup>1</sup>H NMR) taken as the internal standard, and were reported in parts per million (ppm). Abbreviations for signal coupling are as follows: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet. Coupling constants were taken from the spectra directly and are uncorrected. IR spectra were collected on a Nicolet iS10 FT-IR spectrometer as neat films. Mass spectra determinations were carried out on a Thermo Finnigan LTQ-FTMS spectrometer with electrospray (ESI) or atmospheric pressure chemical (APCI) ionization. Optical rotations were measured on JASCO P-2000 polarimeter. Gas chromatography (GC) analysis was performed on an Agilent 7890A; column conditions: 30 °C for 1 min, then increasing to 180 °C at a rate of 5 °C/min, then 180 °C for 5 min. Analytical thin layer chromatography (TLC) was performed on silica gel plates using UV light or stained with 10% vanillin/1% sulfuric acid/ethanol solution. Flash column chromatography was performed with silica gel 60 A (230-400 mesh) according to the literature procedure.<sup>71</sup> Substrates **2**,<sup>72</sup> Rh<sub>2</sub>(SDOSP)<sub>4</sub> and Rh<sub>2</sub>(R-DOSP)<sub>4</sub>,<sup>73</sup> **16**, **17b-d**, **19a** and **c**, **23b**,<sup>68</sup> **17a**,<sup>74</sup> **19b**,<sup>75</sup> **19d**,<sup>76</sup> and **23a**<sup>77</sup> were all synthesized according to published procedures. For <sup>1</sup>H and <sup>13</sup>C NMR data see Supplementary Figures 3–34. For HPLC traces see Supplementary Figures 35–47. For an ORTEP diagram of compound **24a** see Supplementary Figure 48.

### General Procedure

A 35 mL pressure tube fitted with a rubber septum was evacuated and backfilled with dry Ar (3×). The reaction vessel was charged with Rh<sub>2</sub>(R-DOSP)<sub>4</sub> (0.9 mg, 0.005 mmol, 0.1 mol %) and anhydrous CaCl<sub>2</sub> (111 mg, 1.0 mmol, 2.0 equiv) before adding a heptanes solution (1.0 mL) of **16** (43 mg, 0.50 mmol, 1.0 equiv). The suspension was cooled to 0 °C in an ice-bath. A heptanes solution (5 mL) of **6** (121 mg, 0.60 mmol, 1.2 equiv) was added dropwise over 0.5 h accompanied by vigorous stirring. The reaction was further stirred at 0 °C for 2 h,



before removing the ice-bath and allowing to warm to ambient temperature. The rubber septum was removed and the vessel was sealed with a Teflon screwcap. The sealed tube was immersed in an oil bath preheated to 125 °C. After 48 h, the reaction was cooled to ambient temperature and the suspension was filtered through a fritted glass funnel, washing the filter cake with pentane (3 x 10 mL), and the filtrate was concentrated *in vacuo*. Column chromatography (SiO<sub>2</sub>, pentane/diethyl ether, 10 : 1) afforded a single diastereomer of the pure product **15** (92 mg, 71% yield). For additional procedures see Supplementary Methods.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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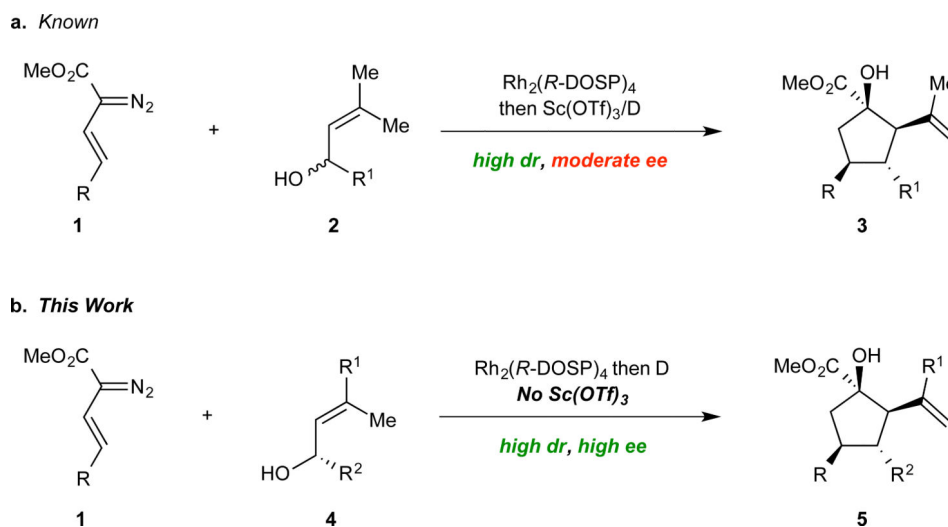
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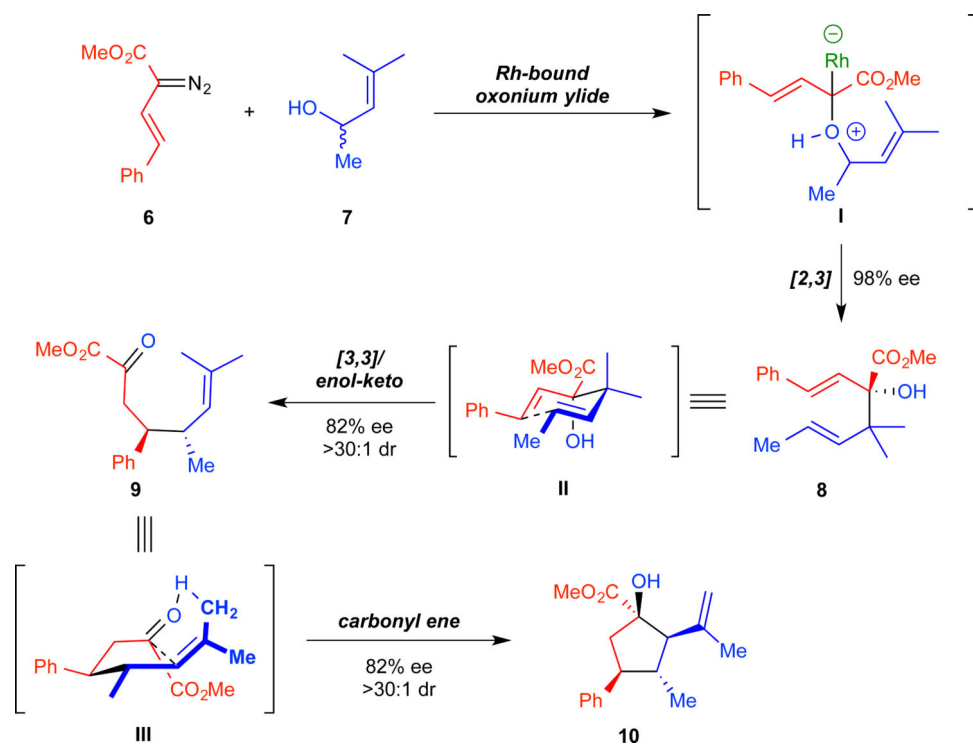
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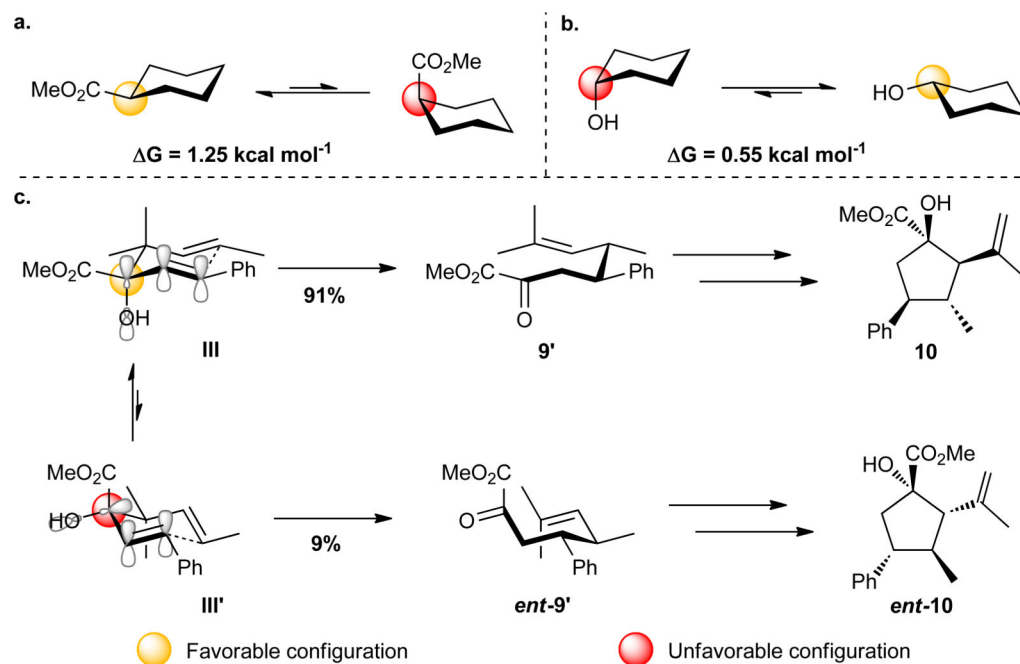


**Figure 1.** Evolution of a Convergent Cyclopentane Synthesis. (a) One-pot synthesis of a cyclopentane (**3**) from a vinyl diazoacetate (**1**) and racemic allyl alcohol (**2**) in modest enantioselectivity via a rhodium-initiated cascade. (b) A second generation cyclopentane (**5**) synthesis from analogous starting materials in high stereoselectivity.

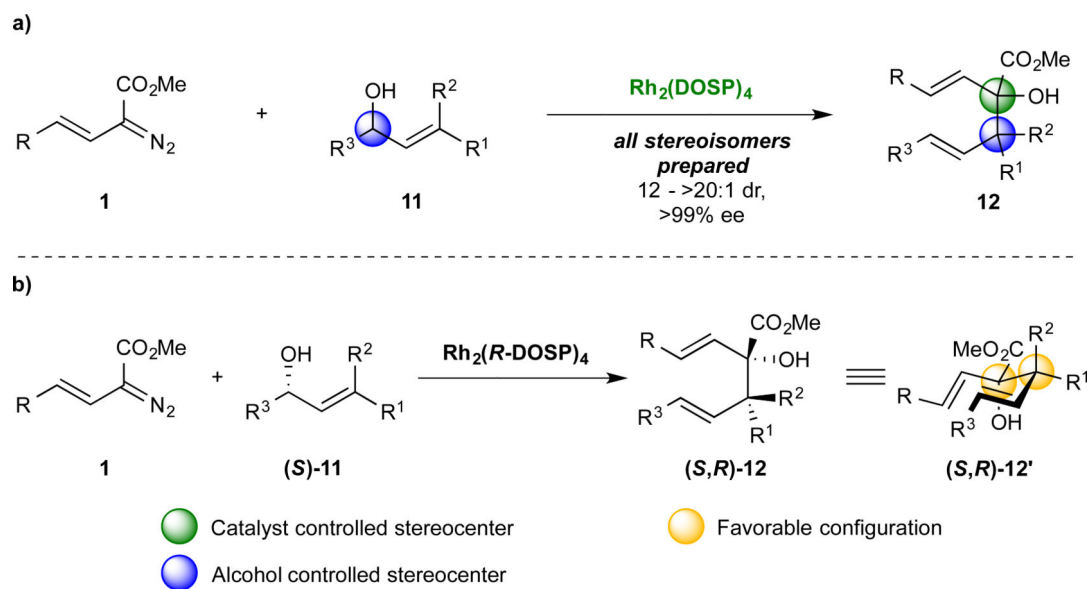


**Figure 2.** Mechanism of cyclopentane synthesis. Mechanism of the one-pot domino sequence for the synthesis of a cyclopentane (**10**) from a diazoacetate (**6**) and a racemic allyl alcohol (**7**).

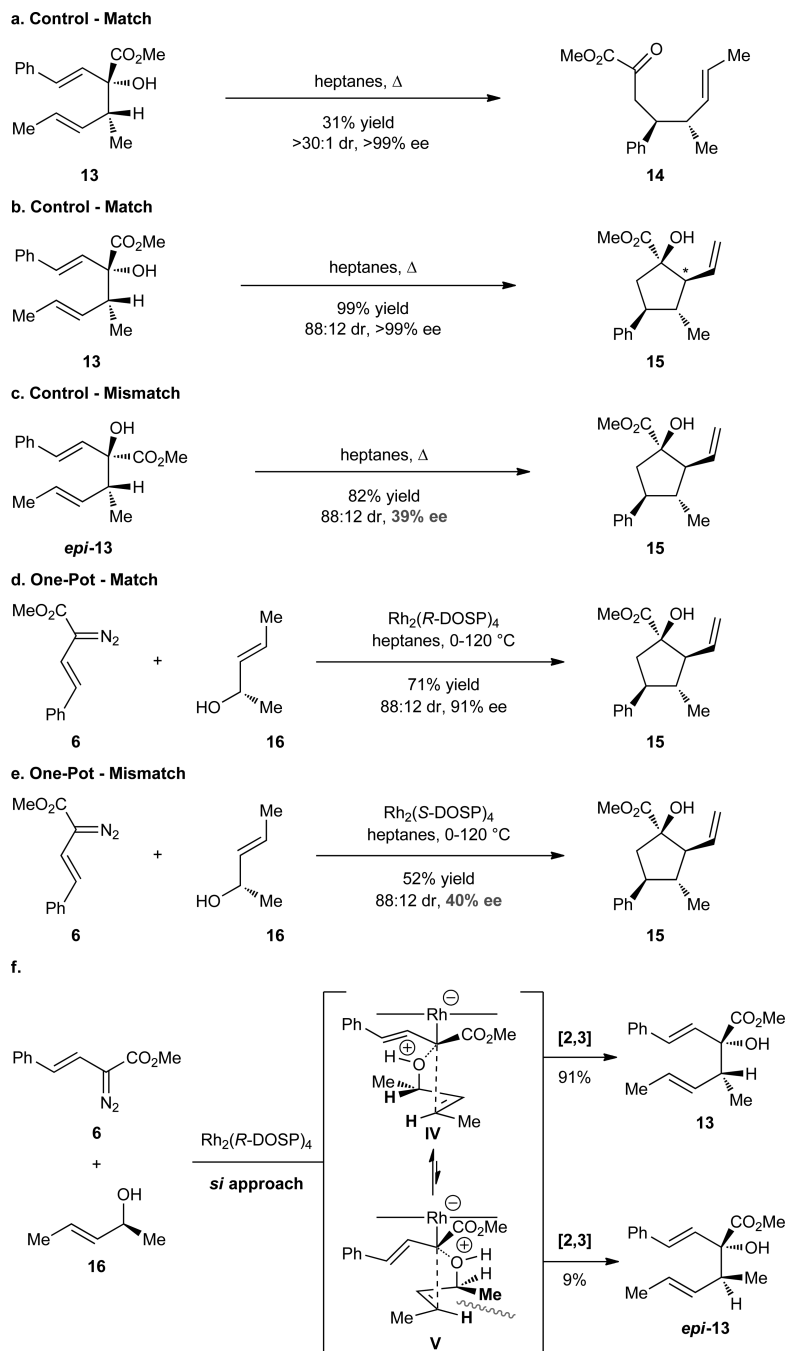


**Figure 3.**

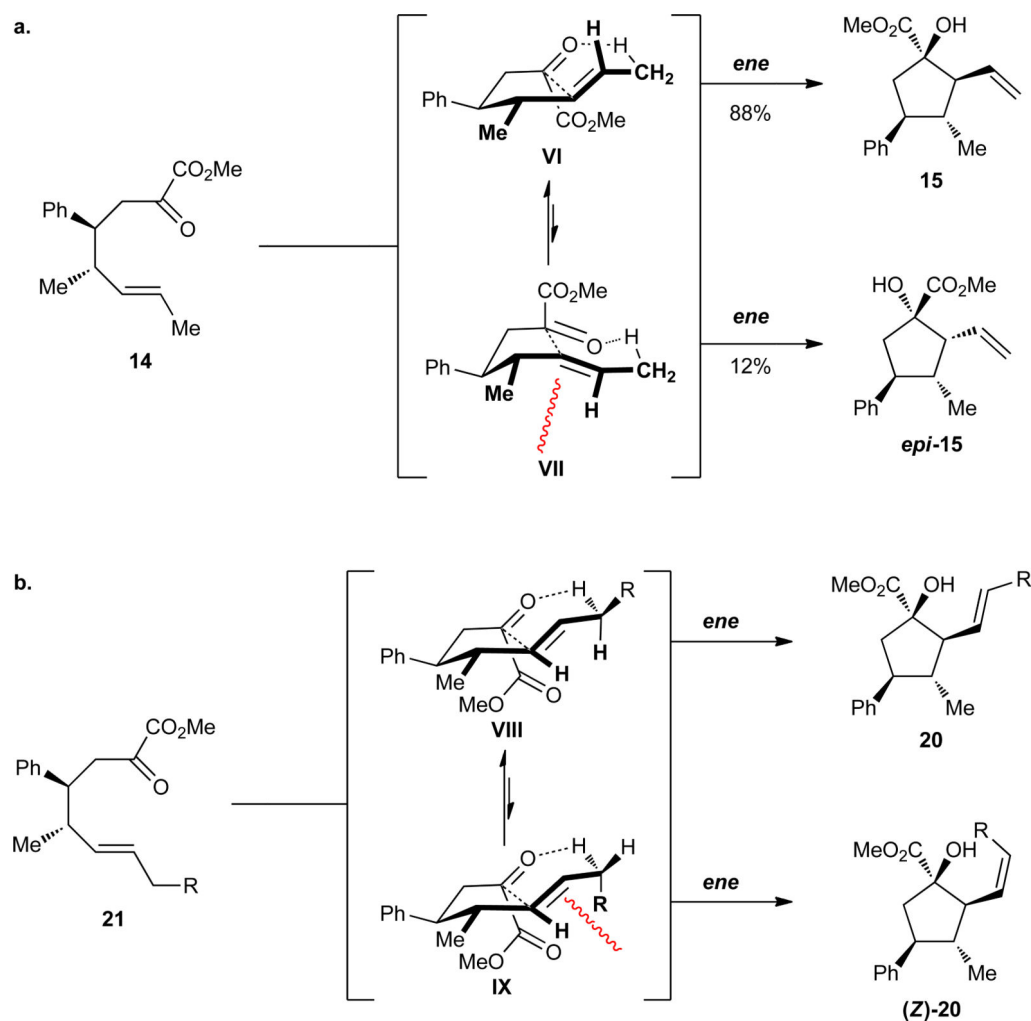
Stereoelectronic considerations for an oxy-Cope rearrangement. Cyclohexane *A*-values for carbomethoxy (a) and hydroxy (b) substituted cyclohexanes. Manifestation of stereoelectronic effects in the competition between two chair-like transition states (III and III').

**Figure 4.**

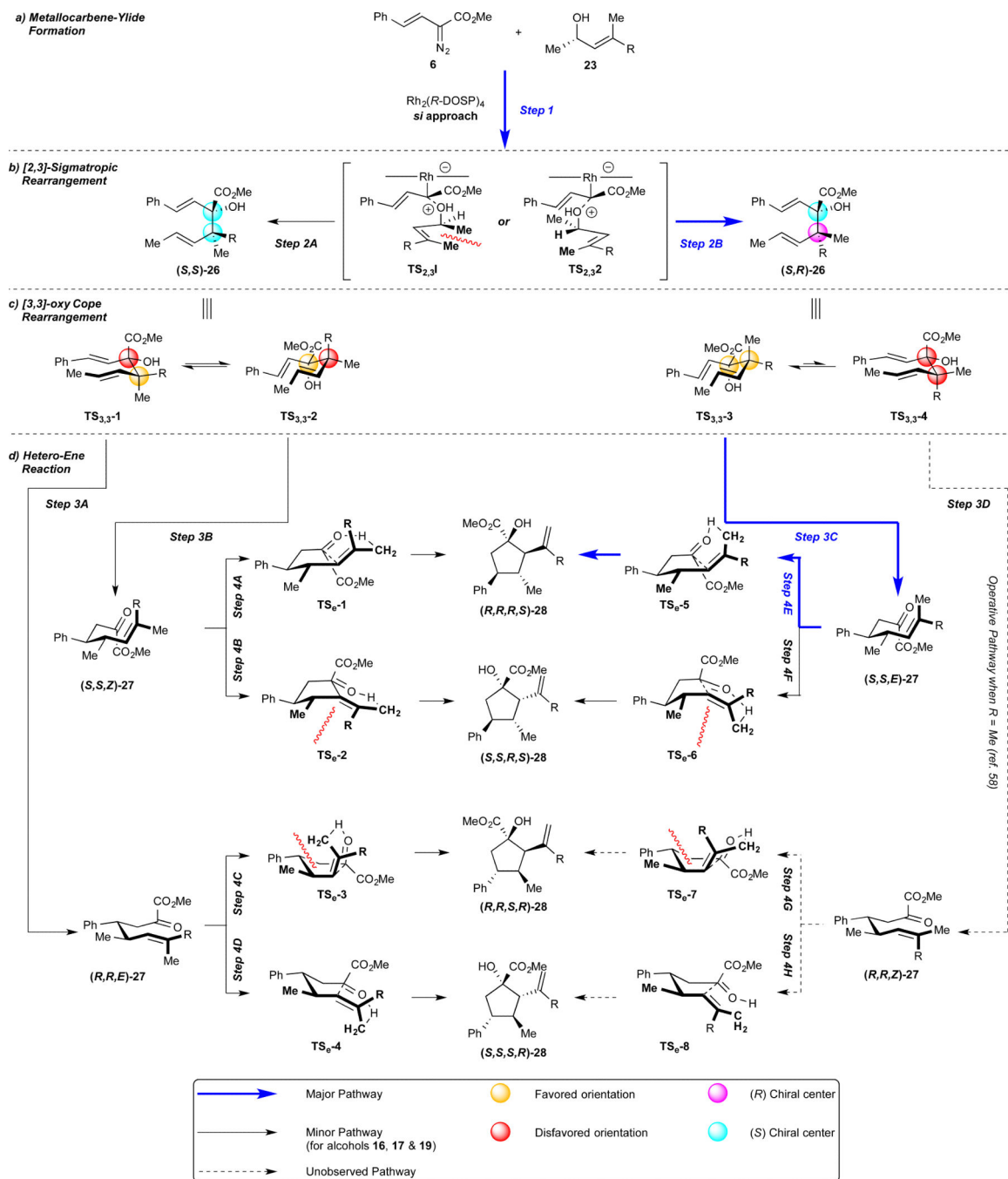
Rationale for improved chirality transfer during an oxy-Cope rearrangement. (a) The tandem-ylide formation/[2,3]-sigmatropic rearrangement between a diazoacetate (**1**) and allyl alcohol (**2**) to generate a 3-hydroxy-1,5-hexadiene (**12**) bearing vicinal stereocenters.



**Figure 5.** Control studies relevant to the synthesis of cyclopentane **15**. (a) Chirality transfer control for the oxy-Cope rearrangement. Matched (b) and mismatched (c) tandem oxy-Cope rearrangement hetero-ene reactions of stereoisomerically pure 3-hydroxy-1,3-hexadienes (**13** and *epi*-**13**, respectively). Matched (d) and mismatched (e) one-pot cyclopentane (**15**) syntheses. (f) Justification for high, but imperfect, diastereocontrol during the rhodium-catalyzed ylide formation/[2,3]-sigmatropic rearrangement.



**Figure 6.** Rationale for diastereoselectivities in the hetero-ene reaction. (a) Envelope-like transition states **VI** and **VII** for the formation of **15** and *epi*-**15**, respectively. (b) Rotational isomers of an ene transition state **VIII** and **IX** for the formation of **20** and (*Z*)-**20**, respectively.



**Figure 7.**  
Plausible Reaction Pathways in the Cyclopentane Synthesis

Table 1

<sup>a</sup> Variations in Carbinol Substitution.

entry	comp'd	product	yield, % <sup>b</sup>	dr <sup>c</sup>	ee, % <sup>d</sup>
1	a		65	89:11	92
2	b		71 <sup>e</sup>	88:12	92
3	c		70	92:8	92
4	d		55 <sup>e</sup>	93:7	93

<sup>a</sup> Reaction conducted with **17** (0.50 mmol, 1.0 equiv), **6** (0.60 mmol, 1.2 equiv) and Rh<sub>2</sub>(*R*-DOSP)<sub>4</sub> (0.0005 mmol, 0.1 mol %), unless otherwise indicated.

<sup>b</sup> Isolated yield of the major diastereomer.

<sup>c</sup> Determined by <sup>1</sup>H NMR analysis of the crude reaction residue.

<sup>d</sup> Determined by HPLC analysis on a chiral stationary phase.

<sup>e</sup> Yield of the reaction with 0.5 mol % Rh<sub>2</sub>(*R*-DOSP)<sub>4</sub>.



Table 2

<sup>a</sup> Variations in Allylic Substitution.

entry	comp'd	product	yield, % <sup>b</sup>	<i>E:Z</i> <sup>c</sup>	<i>dr</i> <sup>c</sup>	ee, % <sup>d</sup>
1	a		71	–	91:9	95
2	b		65	2.2:1	85:15	96
3	c		67	3.4:1	90:10	94
4	d		80 <sup>e</sup>	7.3:1	90:10	95

<sup>a</sup> Reaction conducted with **19** (0.50 mmol, 1.0 equiv), **6** (0.60 mmol, 1.2 equiv) and Rh<sub>2</sub>(*R*-DOSP)<sub>4</sub> (0.0005 mmol, 0.1 mol %), unless otherwise indicated.

<sup>b</sup> Isolated yield of the major diastereomer.

<sup>c</sup> Determined by <sup>1</sup>H NMR analysis of the crude reaction residue.

<sup>d</sup> Determined by HPLC analysis on a chiral stationary phase.

<sup>e</sup> Yield of the reaction with 0.5 mol % Rh<sub>2</sub>(*R*-DOSP)<sub>4</sub>.

Table 3

<sup>a</sup> Variations in Vinylic Substitution.

entry	comp'd	product	yield, % <sup>b</sup>	dr <sup>c</sup>	ee, % <sup>d</sup>
1	a		87	>97:3	99
2	b		85	>97:3	99
3	c		86	>97:3	99
4	d		89	>97:3	99

<sup>a</sup> Reaction conducted with **23** (0.50 mmol, 1.0 equiv), **6** (0.60 mmol, 1.2 equiv) or **22** (1.0 mmol, 2.0 equiv) and Rh<sub>2</sub>(R-DOSP)<sub>4</sub> (0.0005 mmol, 0.1 mol %), unless otherwise indicated.

<sup>b</sup> Isolated yield of the major diastereomer.

<sup>c</sup> Determined by <sup>1</sup>H NMR analysis of the crude reaction residue.

<sup>d</sup> Determined by HPLC analysis on a chiral stationary phase.