Data in brief 27 (2019) 104605



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Data Article

Structural data of phenanthrene-9,10dicarbonitriles



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ARTICLE INFO

Article history: Received 28 August 2019 Received in revised form 25 September 2019 Accepted 26 September 2019 Available online 5 October 2019

Keywords: Aromatic stacking interactions Non-covalent interactions Phenanthrene-9,10-dicarbonitriles Single-crystal X-ray diffraction

ABSTRACT

In this data article, we present the single-crystal XRD data of phenanthrene-9,10-dicarbonitriles. Detailed structure analysis and photophysical properties were discussed in our previous study, "Intermolecular interactions-photophysical properties relationships in phenanthrene-9,10-dicarbonitrile assemblies" (Afanasenko et al., 2020). The data include the intra- and intermolecular bond lengths and angles.

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DOI of original article: https://doi.org/10.1016/j.molstruc.2019.07.036.

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https://doi.org/10.1016/j.dib.2019.104605

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Specifications Table

Subject	Chemistry
Specific subject area	Single crystal data of phenanthrene-9,10-dicarbonitriles
Type of data	Table
	Figure
How data were acquired	Single-crystal XRD: The X-ray diffraction data were collected on an Agilent Technologies Excalibur Eos and Supernova Atlas diffractometers. The structures have been solved by the direct methods and refined by means of the SHELXL–97 program incorporated in the OLEX2 program package.
Data format	Raw and Analyzed
Parameters for data collection	Crystals were analysed at a temperature of 100 K. Single crystal X-ray diffraction data was collected with CuK α (3,6-difluorophenanthrene-9,10-dicarbonitrile, 3,6-dimethylphenanthrene-9,10-dicarbonitrile) and MoK α (3,6-dimethoxyphenanthrene-9,10-dicarbonitrile) radiation.
Description of data collection	Crystals of compounds were immersed in cryo-oil, mounted in a nylon loop.
Data source location	Saint Petersburg State University, Saint Petersburg, Russia
Data accessibility	Crystal data have been deposited at the Cambridge Crystallographic Data Centre (CCDC) with deposition numbers CCDC 1821025, CCDC 1821026, and CCDC 1820117 (http://www.ccdc.cam.ac.uk/conts/retrieving.html, e-mail: deposit@ccdc.cam.ac.uk). With the article
Related research article	Anastasiia M. Afanasenko, Alexander S. Novikov, Tatiana G. Chulkova, Yakov M. Grigoriev, Ilya E. Kolesnikov, Stanislav I. Selivanov, Galina L. Starova, Andrey A. Zolotarev, Anatoly N. Vereshchagin, Michail N. Elinson. Intermolecular interactions- photophysical properties relationships in phenanthrene-9,10-dicarbonitrile assemblies. J. Mol. Struct. 1199 (2020) 126789, https://doi.org/10.1016/j.molstruc.2019.07.036.

Value of the Data

• This data would be valuable for other properties studies of phenanthrene-9,10-dicarbonitriles.

• The data in this article will be useful for researchers who study non-covalent interactions.

• This data provide a new strategy to control the association pattern in the crystal state.

1. Data

In this article, the X-ray information for 3,6-difluorophenanthrene-9,10-dicarbonitrile, 3,6-dimethylphenanthrene-9,10-dicarbonitrile, and 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile is represented. The structures of phenanthrene-9,10-dicarbonitriles are shown in Figs. 1–3. Fractional atomic coordinates and equivalent isotropic displacement parameters for 3,6-difluorophenanthrene-9,10-dicarbonitrile, 3,6-dimethylphenanthrene-9,10-dicarbonitrile, and 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile, and 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile are listed in Table 1, Table 2, and Table 3, respectively. The bond lengths, angles, and torsion angles for phenanthrene-9,10-dicarbonitriles are listed in Tables 4–12. The crystal packing of phenanthrene-9,10-dicarbonitriles is shown in Figs. 4, 6 and 8. Information about π -stacking for phenanthrene-9,10-dicarbonitriles is presented in Figs. 5, 7 and 9. The data in Tables 13–15 show the intermolecular distances for phenanthrene-9,10-dicarbonitriles.

2. Experimental design, materials, and methods

Phenanthrene-9,10-dicarbonitriles were obtained by the previously published procedures [1–5]. The crystals of appropriate quality were obtained at room temperature from ethanol solution. The X-ray diffraction data were collected on an Agilent Technologies Excalibur Eos and Supernova Atlas diffractometers. The temperature for all experiments was kept at 100 K. The structures have been solved by the direct methods and refined by means of the SHELXL–97 [6] program incorporated in the OLEX² program package [7]. The carbon-bound H atoms were placed in calculated positions and were included in the refinement in the 'riding' model approximation, with $U_{iso}(H)$ set to $1.5U_{eq}(C)$ and C-H



Fig. 1. Structure of 3,6-difluorophenanthrene-9,10-dicarbonitrile.



Fig. 2. Structure of 3,6-dimethylphenanthrene-9,10-dicarbonitrile.

0.96 Å for CH₃ groups, $U_{iso}(H)$ set to $1.2U_{eq}(C)$ and C–H 0.93 Å for the CH groups, and $U_{iso}(H)$ set to $1.2U_{eq}(N)$ and N–H 0.86 Å for the NH groups. Empirical absorption correction was applied in CrysA-lisPro program complex [8] using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.



Fig. 3. Structure of 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile.

Fractional atomic coordinates (×10⁴) and equivalent isotropic displacement parameters (Å² × 10³) for 3,6-difluorophenanthrene-9,10-dicarbonitrile. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	х	У	Z	U (eq)
F1	5032 (4)	6348.1 (6)	8033.0 (9)	24.5 (3)
F2	8212 (4)	3296.2 (6)	6013.3 (9)	29.0 (3)
C2	6390 (6)	6326.4 (11)	7253.2 (13)	18.7 (5)
C14	10137 (6)	5050.9 (11)	4824.8 (14)	17.0 (5)
C8	8202 (6)	4447 (1)	6053.4 (14)	18.1 (5)
C11	8250 (5)	5687.3 (10)	6074.6 (13)	15.6 (5)
N2	13427 (6)	5638.6 (9)	2923.6 (13)	25.2 (4)
C4	8453 (6)	6900.1 (10)	6064.8 (14)	19.2 (5)
C9	10378 (5)	6260.3 (11)	4819.5 (13)	17.0 (5)
C6	10079 (6)	3856.3 (10)	4805.0 (14)	22.0 (5)
C13	8857 (6)	5060.3 (11)	5656.8 (13)	16.9 (5)
N1	12149 (6)	7336.7 (9)	4065.6 (14)	31.2 (5)
C16	12274 (6)	5658.6 (10)	3583.9 (13)	18.8 (5)
C3	7151 (6)	6923.7 (11)	6862.3 (14)	21.1 (5)
C15	11317 (6)	6864.7 (11)	4397.5 (14)	21.3 (5)
C1	6897 (6)	5723.0 (11)	6893.2 (14)	17.8 (5)
C12	9010 (5)	6288.3 (11)	5658.6 (13)	17.4 (5)
C7	8849 (6)	3878.4 (11)	5621.8 (14)	22.0 (5)
C5	10711 (6)	4444 (1)	4409.5 (14)	20.1 (5)
C10	10884 (6)	5669.2 (10)	4421.3 (13)	17.6 (5)

Fractional atomic coordinates (×10⁴) and equivalent isotropic displacement parameters (Å² × 10³) for 3,6-dimethylphenanthrene-9,10-dicarbonitrile. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	x	у	Ζ	U (eq)
N1	-82.7 (11)	4030.6 (6)	3353.3 (9)	22.7 (3)
N2	3096.7 (11)	5002.8 (7)	4957 (1)	25.8 (3)
N1A	4515.2 (12)	5029.2 (7)	8469.4 (11)	27.2 (3)
N2A	6832.0 (12)	3359.2 (7)	9233.7 (10)	25.4 (3)
C14	4028.7 (11)	3116.8 (7)	5481.9 (9)	13.5 (3)
C13	3954.8 (11)	2321.8 (7)	5456.0 (9)	13.3 (3)
C14A	4025.3 (11)	2335.2 (7)	7989.6 (9)	13.9 (3)
C10	2926.2 (11)	3548.2 (7)	4936.7 (9)	13.7 (3)
C12A	2366.9 (12)	3591.6 (7)	7243.0 (9)	14.7 (3)
C13A	2743.0 (12)	2209.4 (7)	7371.7 (9)	14.0 (3)
C11	2756.8 (11)	1951.2 (7)	4904.0 (9)	13.3 (3)
C11A	1910.4 (12)	2851.1 (7)	6978.5 (9)	13.9 (3)
C9A	3658.6 (12)	3694.6 (7)	7885.3 (9)	14.9 (3)
C12	1685.3 (11)	2394.2 (7)	4381.8 (9)	13.6 (3)
C4A	1561.1 (12)	4214.9 (7)	6859.8 (10)	17.4 (3)
C2	1485.8 (12)	812.0 (7)	4344.1 (9)	16.5 (3)
C3A	342.4 (13)	4106.0 (7)	6219 (1)	18.7 (3)
C10A	4463.0 (11)	3092.9 (7)	8222.5 (9)	14.9 (3)
C1	2623.3 (12)	1161.5 (7)	4865.2 (9)	15.1 (3)
C6A	4397.2 (12)	992.0 (7)	8146.7 (10)	17.5 (3)
C7A	3131.3 (12)	855.0 (7)	7526.7 (10)	17.0 (3)
COAA	4128.1 (12)	4439.5 (7)	8195.5 (10)	18.6 (3)
C15	748.9 (12)	3658.0 (7)	3822.9 (10)	16.6 (3)
C1A	645.1 (12)	2763.4 (7)	6325.9 (9)	15.3 (3)
C5A	4832.2 (12)	1711.3 (7)	8369.6 (10)	16.7 (3)
C2A	-133.3 (12)	3374.6 (7)	5937 (1)	16.8 (3)
C9	1803.5 (11)	3199.0 (7)	4397.2 (9)	14.0 (3)
C8A	2337.8 (12)	1460.1 (7)	7154.6 (9)	15.9 (3)
C16	3009.0 (11)	4356.8 (8)	4942.2 (10)	17.6 (3)
C7	6192.6 (12)	2264.9 (7)	6493.5 (9)	17.1 (3)
C5	5183.5 (12)	3477.2 (7)	6021.4 (10)	16.5 (3)
C16A	5777.6 (12)	3233.9 (7)	8800.7 (10)	17.6 (3)
C8	5063.2 (12)	1911.6 (7)	5967.0 (9)	15.2 (3)
C4	519.7 (12)	2034.8 (7)	3859.7 (10)	17.2 (3)
C17	1375.5 (13)	-33.5 (7)	4287.7 (11)	21.8 (3)
C6	6239.4 (12)	3058.7 (8)	6520.8 (9)	17.4 (3)
C3	421.5 (12)	1268.5 (8)	3844.7 (10)	18.8 (3)
C15A	2653.5 (14)	62.2 (7)	7283.1 (11)	23.2 (3)
C18A	-1455.5 (12)	3268.0 (8)	5208.6 (11)	22.0 (3)
C18	7359.6 (13)	1813.5 (8)	7031.6 (11)	25.2 (3)

Table 3

Fractional atomic coordinates $(\times 10^4)$ and equivalent isotropic displacement parameters $(Å^2 \times 10^3)$ for 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	x	у	Z	U (eq)
C1	9340 (2)	7965.2 (7)	3471 (2)	12.7 (4)
C2	8802 (2)	8553.7 (7)	3703 (2)	12.5 (4)
C3	7248 (2)	8670.0 (7)	4359 (2)	13.7 (4)
C4	6265 (2)	8196.0 (8)	4780 (2)	13.7 (4)
C5	8313 (2)	5746.8 (7)	3835 (2)	15.2 (4)
C6	9789 (2)	5618.4 (7)	3145 (2)	14.8 (4)
C7	10816 (2)	6098.3 (7)	2686 (2)	13.6 (4)
C8	10357 (2)	6697.4 (7)	2934 (2)	12.8 (3)
C9	5764 (2)	7083.5 (7)	5013 (2)	12.6 (3)
C10	6262 (2)	6491.2 (7)	4808 (2)	13.1 (4)
C11	8336 (2)	7468.7 (7)	3890 (2)	11.2 (3)
C12	6783 (2)	7585.2 (7)	4561 (2)	12.0 (3)

(continued on next page)

Table 3 (c	continued)
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Atom	x	ν	7	U (ea)
		5	~	0 (04)
C13	8852 (2)	6838.9 (7)	3638 (2)	12.0 (4)
C14	7810 (2)	6353.4 (7)	4096 (2)	12.7 (4)
C15	4209 (2)	7206.7 (7)	5739 (2)	13.9 (4)
C16	5235 (2)	6005.2 (7)	5362 (2)	14.0 (4)
C17	11302 (2)	8964.3 (8)	2703 (2)	17.3 (4)
C18	12774 (2)	5410.8 (8)	1616 (2)	19.5 (4)
N1	2976.3 (19)	7306.4 (7)	6322 (2)	19.8 (3)
N2	4438.9 (19)	5619.1 (7)	5838 (2)	19.1 (3)
01	9690.6 (15)	9060.9 (5)	3343.7 (15)	16.0 (3)
02	12295.8 (15)	6021.9 (5)	1975.2 (15)	16.1 (3)

Bond lengths	for 3,6-difluoro	phenanthrene-9,1	0-dicarbonitrile.

Atom	Atom	Length/Å
F1	C2	1.354 (2)
F2	C7	1.357 (3)
C2	C3	1.393 (3)
C2	C1	1.364 (3)
C14	C13	1.417 (3)
C14	C5	1.413 (3)
C14	C10	1.438 (3)
C8	C13	1.417 (3)
C8	C7	1.365 (3)
C11	C13	1.453 (3)
C11	C1	1.411 (3)
C11	C12	1.418 (3)
N2	C16	1.147 (3)
C4	C3	1.371 (3)
C4	C12	1.414 (3)
C9	C15	1.446 (3)
C9	C12	1.442 (3)
C9	C10	1.368 (3)
C6	C7	1.388 (3)
C6	C5	1.369 (3)
N1	C15	1.141 (3)
C16	C10	1.442 (3)

Table 5Bond angles for 3,6-difluorophenanthrene-9,10-dicarbonitrile.

Atom	Atom	Atom	Angle/°
F1	C2	СЗ	117.92 (18)
F1	C2	C1	118.19 (19)
C1	C2	C3	123.9 (2)
C13	C14	C10	118.7 (2)
C5	C14	C13	120.3 (2)
C5	C14	C10	121.0 (2)
C7	C8	C13	118.74 (19)
C1	C11	C13	122.00 (18)
C1	C11	C12	117.92 (18)
C12	C11	C13	120.08 (18)
C3	C4	C12	120.78 (19)
C12	C9	C15	119.7 (2)
C10	C9	C15	119.1 (2)
C10	C9	C12	121.2 (2)
C5	C6	C7	117.8 (2)
C14	C13	C11	119.8 (2)

Atom	Atom	Atom	Angle/*
C8	C13	C14	118.0 (2)
C8	C13	C11	122.13 (18)
N2	C16	C10	178.5 (2)
C4	C3	C2	117.74 (19)
N1	C15	C9	178.2 (2)
C2	C1	C11	119.30 (19)
C11	C12	C9	118.6 (2)
C4	C12	C11	120.36 (18)
C4	C12	C9	121.0 (2)
F2	C7	C8	117.84 (18)
F2	C7	C6	117.8 (2)
C8	C7	C6	124.3 (2)
C6	C5	C14	120.8 (2)
C14	C10	C16	118.6 (2)
C9	C10	C14	121.6 (2)
C9	C10	C16	119.7 (2)

Table 5 (c	ontinued)
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Table 6Torsion angles for 3,6-difluorophenanthrene-9,10-dicarbonitrile.

А	В	С	D	Angle/
F1	C2	C3	C4	-178.63 (18)
F1	C2	C1	C11	178.91 (17)
C13	C14	C5	C6	-0.7 (3)
C13	C14	C10	C9	-0.2 (3)
C13	C14	C10	C16	178.45 (17)
C13	C8	C7	F2	179.88 (18)
C13	C8	C7	C6	-0.8 (3)
C13	C11	C1	C2	179.90 (18)
C13	C11	C12	C4	-179.49 (18)
C13	C11	C12	C9	0.1 (3)
C3	C2	C1	C11	-0.4 (3)
C3	C4	C12	C11	-0.4 (3)
C3	C4	C12	C9	179.99 (18)
C15	C9	C12	C11	-177.39 (18)
C15	C9	C12	C4	2.2 (3)
C15	C9	C10	C14	177.43 (18)
C15	C9	C10	C16	-1.2 (3)
C1	C2	C3	C4	0.7 (3)
C1	C11	C13	C14	178.46 (18)
C1	C11	C13	C8	-1.4 (3)
C1	C11	C12	C4	0.7 (3)
C1	C11	C12	C9	-179.70 (17)
C12	C11	C13	C14	-1.3 (3)
C12	C11	C13	C8	178.78 (17)
C12	C11	C1	C2	-0.3 (3)
C12	C4	C3	C2	-0.3 (3)
C12	C9	C10	C14	-1.1 (3)
C12	C9	C10	C16	-179.69 (18)
C7	C8	C13	C14	0.4 (3)
C7	C8	C13	C11	-179.76 (18)
C7	C6	C5	C14	0.3 (3)
C5	C14	C13	C8	0.4 (3)
C5	C14	C13	C11	-179.5 (2)
C5	C14	C10	C9	-179.30 (19)
C5	C14	C10	C16	-0.7 (3)
C5	C6	C7	F2	179.81 (19)

(continued on next page)

Table 6	(continued)
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-				
А	В	С	D	Angle/
C5	C6	C7	C8	0.5 (3)
C10	C14	C13	C8	-178.74 (19)
C10	C14	C13	C11	1.4 (3)
C10	C14	C5	C6	178.40 (19)
C10	C9	C12	C11	1.1 (3)
C10	C9	C12	C4	-179.31 (19)

Table 7Bond lengths for 3,6-dimethylphenanthrene-9,10-dicarbonitrile.

Atom	Length/Å
C15	1.1529 (18)
C16	1.151 (2)
COAA	1.1496 (19)
C16A	1.1516 (19)
C13	1.4138 (17)
C10	1.4358 (16)
C5	1.4125 (17)
C11	1.4603 (16)
C8	1.4089 (16)
C13A	1.4233 (17)
C10A	1.4317 (17)
C5A	1.4154 (17)
C9	1.3776 (17)
C16	1.4387 (18)
C11A	1.4147 (17)
C9A	1.4360 (17)
C4A	1.4148 (17)
C11A	1.4579 (16)
C8A	1.4059 (17)
C12	1.4156 (16)
C1	1.4093 (17)
C1A	1.4142 (17)
C10A	1.3741 (17)
COAA	1.4366 (17)
C9	1.4344 (17)
C4	1.4150 (17)
C3A	1.3732 (19)
C1	1.3836 (18)
C17	1.5063 (16)
C3	1.4163 (18)
C2A	1.4095 (18)
C16A	1,4426 (17)
C7A	1.4129 (18)
C5A	1.3654 (18)
C8A	1.3799 (17)
C15A	1 5041 (17)
(9	1 4393 (17)
C2A	1 3842 (17)
C18A	1.5034 (17)
C8	1.3818 (17)
C6	1 4104 (18)
C18	1 5018 (17)
6	1 3723 (18)
C3	1 3647 (10)
	Atom C15 C16 C0AA C16A C13 C10 C5 C11 C8 C13A C10A C5A C9 C16 C11A C9A C4A C11A C10A C0A C11A C14 C10A C12 C1 C1A C10A C0AA C12 C1 C1A C10A C0AA C9 C4 C3A C1 C17 C3 C2A C16A C7A C5A C8A C15A C9 C2A C18A C6 C18 C6 C18 <tr td=""></tr>

Table 8Bond angles for 3,6-dimethylphenanthrene-9,10-dicarbonitrile.

Atom	Atom	Atom	Angle/*
C13	C14	C10	119.12 (11)
C5	C14	C13	120.07 (11)
C5	C14	C10	120.79 (11)
C14	C13	C11	119.92 (11)
C8	C13	C14	117.99 (11)
C8	C13	C11	122.08 (11)
C13A	C14A	C10A	118.98 (11)
C5A	C14A	C13A	119.46 (11)
C5A	C14A	C10A	121.55 (11)
C14	C10	C16	119.19 (11)
(9	C10	C14	121.00 (12)
(9	C10	C16	119.80 (11)
CIIA	CI2A	C9A	118.85 (11)
CHA	CI2A	C4A	119.93 (11)
C4A	CI2A	CIA	121.20 (11)
CI4A CRA	CI3A		117.02 (11)
CSA	CI3A C12A	CI4A C11A	117.83 (11)
C8A	CI3A C11		122.39 (11)
C12	C11	C13	119.45 (11)
CI	C11	C13	122.55 (11)
C12A		C12 C12A	110.22 (11)
C12A		CI3A C13A	117.01 (11)
CIA CIA		C12A	117.31 (11)
C124		CISA	122.22 (11)
C10A	C9A	C12A	121 45 (11)
C10A	C9A	COAA	118 72 (11)
C11	C12	C9	119.72 (11)
C4	C12	C11	119 37 (12)
C4	C12	(9	121 39 (11)
C3A	C4A	C12A	120.36 (12)
C1	C2	C17	121.30 (12)
C1	C2	C3	118.42 (12)
C3	C2	C17	120.26 (12)
C4A	C3A	C2A	120.87 (12)
C14A	C10A	C16A	119.85 (11)
C9A	C10A	C14A	121.18 (11)
C9A	C10A	C16A	118.95 (11)
C2	C1	C11	122.22 (12)
C5A	C6A	C7A	120.60 (12)
C6A	C7A	C15A	120.51 (12)
C8A	C7A	C6A	118.94 (12)
C8A	C7A	C15A	120.54 (12)
N1A	COAA	C9A	178.19 (15)
N1	C15	C9	178.78 (15)
C2A	C1A	C11A	122.04 (12)
C6A	C5A	C14A	120.83 (12)
C3A	C2A	C18A	120.06 (12)
C1A	C2A	C3A	118.87 (12)
C1A	C2A	C18A	121.06 (12)
C10	C9	C12	121.24 (11)
C10	C9	C15	118.61 (11)
C12	C9	C15	120.14 (11)
	CSA	CI3A	122.34 (12)
NZ	C16	C10	178.82 (14)
			119.16(11)
	C7		120.74 (12)
			120.11(11)
			120.27 (12)
NZA	CIDA	CIUA	177.01 (15)

(continued on next page)

Table 8	(continued)
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Atom	Atom	Atom	Angle/°
C7	C8	C13	121.87 (12)
C3	C4	C12	120.84 (12)
C5	C6	C7	120.63 (11)
C4	C3	C2	120.90 (11)

Table 9

Torsion angles for 3,6-dimethylphenanthrene-9,10-dicarbonitrile.

Α	В	С	D	Angle/
C14	C13	C11	C12	1.27 (17)
C14	C13	C11	C1	-179.03 (11)
C14	C13	C8	C7	0.90 (18)
C14	C10	C9	C12	1.66 (19)
C14	C10	C9	C15	-176.83 (11)
C14	C5	C6	C7	0.75 (19)
C13	C14	C10	C9	0.34 (18)
C13	C14	C10	C16	-178.50 (11)
C13	C14	C5	C6	0.16 (18)
C13	C11	C12	C9	0.68 (17)
C13	C11	C12	C4	-179.05 (11)
C13	C11	C1	C2	179.80 (11)
C14A	C13A	C11A	C12A	-2.31 (18)
C14A	C13A	C11A	C1A	177.20 (11)
C14A	C13A	C8A	C7A	0.71 (19)
C10	C14	C13	C11	-1.79 (17)
C10	C14	C13	C8	177.72 (11)
C10	C14	C5	C6	-178.51 (12)
C12A	C11A	C1A	C2A	0.96 (19)
C12A	C9A	C10A	C14A	-3.33 (19)
C12A	C9A	C10A	C16A	175.19 (11)
C12A	C4A	C3A	C2A	0.5 (2)
C13A	C14A	C10A	C9A	1.94 (18)
C13A	C14A	C10A	C16A	-176.56 (11)
C13A	C14A	C5A	C6A	0.31 (19)
C13A	CIIA	CIA	C2A	-178.57 (12)
C11	C13	C8	C7	-179.60 (11)
C11	C12	<u>C9</u>	C10	-2.16 (18)
	C12	C9	C15	1/6.31 (11)
	C12	C4	3	-0.75 (19)
CIIA	CI2A	CGA	CIUA	1.81 (19)
CTTA C11A	CI2A CI2A	CGA	CUAA	-1/7.59(11)
CTTA C11A	CI2A CI2A	C4A	CJA	-1.02 (19)
CTTA C11A	CI3A CIA	CBA	C7A	1/9.82 (12)
CTTA C11A		C2A C2A	CIRA	-1.46(19)
COA	CI2A	C2A C11A		1/7.27 (12)
COA	CI2A	CIIA CIIA		179 52 (11)
COA	CI2A	CIA		-178.32(11)
C9A C12	CI2A C11	C4A C1	CA	0.50 (12)
C12 C12	C1		C2	-0.50 (18)
C12	C12A	C11A	C12A	-0.3(2)
C4A	C12A	C11A		0.20 (18)
C4A	C12A	COA		176.00 (12)
C4A	C12A	COA	COAA	-170.33(12) 3.61(19)
C4A	C34	C3A	C1A	0.7(13)
C4A	C3A	C2A	C18A	-178.04(12)
C10A	C14A	C13A	C11A	-170.04(12) 0.87(17)
C10A	C14A	C13A	C8A	180.00 (11)
C10A	C14A	C5A	C6A	179 49 (12)
C1	C11	(12	(9	_179 (12) _179 (12)
C1	C11	C12	C4	1 24 (17)

Α	В	С	D	Angle/
C1	C2	СЗ	C4	1.3 (2)
C6A	C7A	C8A	C13A	-0.11 (19)
C7A	C6A	C5A	C14A	0.3 (2)
COAA	C9A	C10A	C14A	176.07 (11)
COAA	C9A	C10A	C16A	-5.41 (18)
C5A	C14A	C13A	C11A	-179.93 (11)
C5A	C14A	C13A	C8A	-0.80 (18)
C5A	C14A	C10A	C9A	-177.24 (12)
C5A	C14A	C10A	C16A	4.26 (18)
C5A	C6A	C7A	C8A	-0.4(2)
C5A	C6A	C7A	C15A	-179.92 (13)
C9	C12	C4	C3	179.53 (12)
C8A	C13A	C11A	C12A	178.60 (11)
C8A	C13A	C11A	C1A	-1.88 (19)
C16	C10	C9	C12	-179.51 (11)
C16	C10	C9	C15	2.00 (18)
C5	C14	C13	C11	179.53 (11)
C5	C14	C13	C8	-0.96 (17)
C5	C14	C10	C9	179.02 (11)
C5	C14	C10	C16	0.18 (18)
C8	C13	C11	C12	-178.22 (11)
C8	C13	C11	C1	1.48 (18)
C8	C7	C6	C5	-0.82 (19)
C4	C12	C9	C10	177.57 (12)
C4	C12	C9	C15	-3.97 (19)
C17	C2	C1	C11	178.18 (12)
C17	C2	C3	C4	-177.67 (12)
C6	C7	C8	C13	-0.03 (19)
C3	C2	C1	C11	-0.73 (19)
C15A	C7A	C8A	C13A	179.40 (12)
C18	C7	C8	C13	179.98 (12)
C18	C7	C6	C5	179.17 (12)

Bond lengths for 3	3,6-dimethoxyphenanthrene-9,10-dicarbonitrile.
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Atom	Atom	Length/Å
C1	C2	1.378 (2)
C1	C11	1.408 (2)
C2	C3	1.408 (2)
C2	01	1.3624 (19)
C3	C4	1.361 (2)
C4	C12	1.418 (2)
C5	C6	1.372 (2)
C5	C14	1.411 (2)
C6	C7	1.402 (2)
C7	C8	1.384 (2)
C7	02	1.3643 (19)
C8	C13	1.404 (2)
C9	C10	1.373 (2)
C9	C12	1.433 (2)
C9	C15	1.440 (2)
C10	C14	1.435 (2)
C10	C16	1.439 (2)
C11	C12	1.414 (2)
C11	C13	1.461 (2)
C13	C14	1.418 (2)
C15	N1	1.145 (2)
C16	N2	1.145 (2)
C17	01	1.4361 (19)
C18	02	1.431 (2)

Table 11Bond angles for 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile.

Atom	Atom	Atom	Angle/°
C2	C1	C11	120.24 (15)
C1	C2	C3	120.88 (15)
01	C2	C1	124.35 (15)
01	C2	C3	114.77 (14)
C4	C3	C2	119.73 (15)
C3	C4	C12	120.82 (15)
C6	C5	C14	121.23 (16)
C5	C6	C7	119.45 (15)
C8	C7	C6	120.53 (15)
02	C7	C6	124.23 (15)
02	C7	C8	115.24 (14)
C7	C8	C13	120.95 (15)
C10	C9	C12	121.44 (15)
C10	C9	C15	119.59 (15)
C12	C9	C15	118.95 (14)
C9	C10	C14	120.93 (15)
C9	C10	C16	119.10 (15)
C14	C10	C16	119.95 (15)
C1	C11	C12	118.90 (14)
C1	C11	C13	121.80 (15)
C12	C11	C13	119.31 (14)
C4	C12	C9	121.21 (15)
C11	C12	C4	119.43 (15)
C11	C12	C9	119.36 (14)
C8	C13	C11	121.68 (15)
C8	C13	C14	118.50 (15)
C14	C13	C11	119.82 (15)
C5	C14	C10	121.53 (15)
C5	C14	C13	119.35 (15)
C13	C14	C10	119.12 (15)
N1	C15	C9	179.7 (2)
N2	C16	C10	178.49 (19)
C2	01	C17	116.72 (12)
C7	02	C18	117.18 (13)

 Table 12

 Torsion angles for 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile.

-				
А	В	С	D	Angle/
C1	C2	C3	C4	0.4 (2)
C1	C2	01	C17	-0.9 (2)
C1	C11	C12	C4	0.6 (2)
C1	C11	C12	C9	-179.17 (14)
C1	C11	C13	C8	-1.3 (2)
C1	C11	C13	C14	179.34 (15)
C2	C1	C11	C12	-0.5 (2)
C2	C1	C11	C13	179.13 (15)
C2	C3	C4	C12	-0.2 (2)
C3	C2	01	C17	178.81 (14)
C3	C4	C12	C9	179.49 (15)
C3	C4	C12	C11	-0.3 (2)
C5	C6	C7	C8	0.3 (3)
C5	C6	C7	02	-179.30 (16)
C6	C5	C14	C10	179.62 (16)
C6	C5	C14	C13	-0.2 (3)

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Table	12	(continued)	

A	В	С	D	Angle/
C6	C7	C8	C13	-0.4 (2)
C6	C7	02	C18	2.7 (2)
C7	C8	C13	C11	-179.24 (15)
C7	C8	C13	C14	0.1 (2)
C8	C7	02	C18	-176.96 (14)
C8	C13	C14	C5	0.1 (2)
C8	C13	C14	C10	-179.66 (15)
C9	C10	C14	C5	-178.29 (16)
C9	C10	C14	C13	1.5 (2)
C10	C9	C12	C4	-179.82 (16)
C10	C9	C12	C11	0.0 (2)
C11	C1	C2	C3	0.0 (2)
C11	C1	C2	01	179.62 (15)
C11	C13	C14	C5	179.52 (15)
C11	C13	C14	C10	-0.3 (2)
C12	C9	C10	C14	-1.4(2)
C12	C9	C10	C16	176.95 (15)
C12	C11	C13	C8	178.30 (15)
C12	C11	C13	C14	-1.1 (2)
C13	C11	C12	C4	-178.97 (15)
C13	C11	C12	C9	1.2 (2)
C14	C5	C6	C7	-0.1 (3)
C15	C9	C10	C14	-179.70 (15)
C15	C9	C10	C16	-1.4(2)
C15	C9	C12	C4	-1.5 (2)
C15	C9	C12	C11	178.32 (15)
C16	C10	C14	C5	3.4 (2)
C16	C10	C14	C13	-176.81 (15)
01	C2	C3	C4	-179.32 (14)
02	C7	C8	C13	179.30 (14)



Fig. 4. The crystal packing of 3,6-difluorophenanthrene-9,10-dicarbonitrile.



Fig. 5. π -Stacking in the crystal structure of 3,6-difluorophenanthrene-9,10-dicarbonitrile.



Fig. 6. The crystal packing of 3,6-dimethylphenanthrene-9,10-dicarbonitrile.



Fig. 7. π -Stacking in the crystal structure of 3,6-dimethylphenanthrene-9,10-dicarbonitrile.



Fig. 8. The crystal packing of 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile.



Fig. 9. π -Stacking in the crystal structure of 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile.

	······	
Atom	Atom	Length/Å
F1	H5#1	2.493 (2)
H1	N2#1	2.577 (2)
H8	N2#1	2.628 (2)
F2	H4#2	2.586 (2)
H6	N1#2	2.658 (2)
H5	F1#3	2.493 (2)
N2	H1#3	2.577 (2)
N2	H8#3	2.628 (2)
N1	H6#4	2.658 (2)
H4	F2#4	2.586 (2)

 Table 13

 Intermolecular distances for 3,6-difluorophenanthrene-9,10-dicarbonitrile.

Intermolecular distances for 3,6-dimethylphenanthrene-9,10-dicarbonitrile.

Atom	Atom	Length/Å
N1	H3A#1	2.728 (2)
N1	H18B#2	2.943 (2)
H8	N1A#3	2.545 (2)
H18B	N1#4	2.943 (2)
N1A	H8#5	2.545 (2)
N2A	H18E#6	2.619 (2)
H18E	N2A#7	2.619 (2)
НЗА	N1#8	2.728 (2)

Table 1	15
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Intermolecular distances for 3,6-dimethoxyphenanthrene-9,10-dicarbonitrile.

Atom	Atom	Length/Å
N1	H1#1	2.637 (2)
N1	H8#1	2.450 (2)
N2	H17b#1	2.591 (2)
H17b	N2#2	2.591 (2)
H1	N1#2	2.637 (2)
H8	N1#2	2.450 (2)

Acknowledgments

This work has been supported by Russian Foundation for Basic Research (project 16-33-60063). We thank the Research Centre for X-ray Diffraction Studies (Saint Petersburg State University) for physicochemical measurements.

Conflict of Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.dib.2019.104605.

References

- A.M. Afanasenko, A.S. Novikov, T.G. Chulkova, Y.M. Grigoriev, I.E. Kolesnikov, S.I. Selivanov, G.L. Starova, A.A. Zolotarev, A.N. Vereshchagin, M.N. Elinson, Intermolecular interactions-photophysical properties relationships in phenanthrene-9,10dicarbonitrile assemblies, J. Mol. Struct. 1199 (2020) 126789, https://doi.org/10.1016/j.molstruc.2019.07.036.
- [2] A. Cook, R. Linstead, Phthalocyanines. Part XI. The preparation of octaphenylporphyrazines from diphenylmaleinitrile, J. Chem. Soc. (1937) 929–933, https://doi.org/10.1039/JR9370000929.
- [3] A.M. Afanasenko, D.V. Boyarskaya, I.A. Boyarskaya, T.G. Chulkova, Y.M. Grigoriev, I.E. Kolesnikov, M.S. Avdontceva, T.L. Panikorovskii, A.I. Panin, A.N. Vereshchagin, M.N. Elinson, Structures and photophysical properties of 3,4-diaryl-1H-pyrrol-2,5-diimines and 2,3-diarylmaleimides, J. Mol. Struct. 1146 (2017) 554–561, https://doi.org/10.1016/j.molstruc.2017.06.048.
- [4] M.N. Elinson, A.S. Dorofeev, S.K. Feducovich, P.A. Belyakov, G.I. Nikishin, Stereoselective electrocatalytic oxidative coupling of phenylacetonitriles: facile and convenient way to trans-α,β-dicyanostilbenes, Eur. J. Org. Chem. (2007) 3023–3027, https://doi.org/10.1002/ejoc.200601108.
- [5] K. Ichimura, S. Watanabe, pH-dependency of photocyclization of diarylfumaronitriles, Bull. Chem. Soc. Jpn. 49 (1976) 2224–2229, https://doi.org/10.1246/bcsj.49.2224.
- [6] G.M. Sheldrick, A short history of SHELX, Acta Crystallogr. A. 64 (2008) 112–122, https://doi.org/10.1107/ S0108767307043930.
- [7] O.V. Dolomanov, LJ. Bourhis, R.J. Gildea, J.A.K. Howard, H. Puschmann, OLEX2: a complete structure solution, refinement and analysis program, J. Appl. Crystallogr. 42 (2009) 339–341, https://doi.org/10.1107/s0021889808042726.
- [8] CrysAlisPro: Agilent, 2014. CrysAlis PRO. Agilent Technologies Ltd; Yarnton; Oxfordshire; England.