

Article

Seaweed Essential Oils as a New Source of Bioactive Compounds for Cyanobacteria Growth Control: Innovative Ecological Biocontrol Approach

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Abstract: The application of natural compounds extracted from seaweeds is a promising eco-friendly alternative solution for harmful algae control in aquatic ecosystems. In the present study, the anti-cyanobacterial activity of three Moroccan marine macroalgae essential oils (EOs) was tested and evaluated on unicellular *Microcystis aeruginosa* cyanobacterium. Additionally, the possible anti-cyanobacterial response mechanisms were investigated by analyzing the antioxidant enzyme activities of *M. aeruginosa* cells. The results of EOs GC–MS analyses revealed a complex chemical composition, allowing the identification of 91 constituents. Palmitic acid, palmitoleic acid, and eicosapentaenoic acid were the most predominant compounds in *Cystoseira tamariscifolia*, *Sargassum muticum*, and *Ulva lactuca* EOs, respectively. The highest anti-cyanobacterial activity was recorded for *Cystoseira tamariscifolia* EO (ZI = 46.33 mm, MIC = 7.81 µg mL⁻¹, and MBC = 15.62 µg mL⁻¹). The growth, chlorophyll-*a* and protein content of the tested cyanobacteria were significantly reduced by *C. tamariscifolia* EO at both used concentrations (inhibition rate >67% during the 6 days test period in liquid media). Furthermore, oxidative stress caused by *C. tamariscifolia* EO on cyanobacterium cells showed an increase of the activities of superoxide dismutase (SOD) and catalase (CAT), and malondialdehyde (MDA) concentration was significantly elevated after 2 days of exposure. Overall,



these experimental findings can open a promising new natural pathway based on the use of seaweed essential oils to the fight against potent toxic harmful cyanobacterial blooms (HCBs).

Keywords: anti-cyanobacterial activity; bio-control; seaweed essential oils; Microcystis aeruginosa

Key Contribution: *Microcystis* species are among the most important worldwide freshwater bloom-forming cyanobacteria. Seaweed essential oils were chosen for inhibitory experiment on *Microcystis aeruginosa*. The inhibition rate and oxidative stress caused by *C. tamariscifolia* EO on cyanobacterium cells were quite similar to that obtained by copper sulphate (CuSO₄).

1. Introduction

On the grounds of climate warming and increased nutrient inputs due to anthropogenic activities, harmful cyanobacterial blooms (HCBs) become a severe hazard for freshwater ecosystems [1–4]. Due to the critical economic and public health issues caused by HCBs, extensive research on this topic has been conducted aiming to disclose the detrimental effects of HCBs and mitigation strategies. Recent research has been focused on the strategies applied in HCBs control including chemical, physical, mechanical, and biological methods [5–7]. Cyanobacteria blooms are controlled with ultrasound, artificial mixing, and ultraviolet irradiation in the case of physical and mechanical strategies [8–11]. Chemical products such as photosensitizers (molecules that can be activated by light in order to generate reactive oxygen species which may damage cell structures), metals, and other chemical molecules are among the most commonly used chemical methods [12–14]. The introduction of grazers and competitors of cyanobacteria, such as zooplankton, microorganisms (viruses, pathogenic bacteria, or fungi), and macrophytes have been proposed as the most popular biomanipulation for the bio-control of toxic cyanobacteria [15–19]. However, the application of these strategies is not recommended because of their unforeseen ecological consequences, high costs, energy-intensive, and low efficiency [20]. In order to develop effective anti-HCBs agents that are more eco-friendly to the environment, scientists are looking for natural substances released by other aquatic organisms with activity against the growth of cyanobacteria. The marine environment is an excellent source of natural bioactive compounds with unique structures, different from those found in terrestrial natural compounds [21]. Among marine organisms, seaweeds produce active secondary metabolites with a wide range of biological activities, including antibacterial, antifungal, antioxidant, and anti-cyanobacterial compounds [7].

Owing to its specific geographical position, from the Mediterranean Sea to the North and the Atlantic Ocean to the West, Morocco holds a large bio-ecological diversity of seaweeds. This diversity was documented for instance by Chalabi et al. (2015) [22] who reported a particular richness of 489 species distributed at the Mediterranean Coast (381 species) and the Atlantic Coast (323 species). Several compounds with anti-cyanobacterial properties have been purified from the extracts of some seaweed species, such as palmitelaidic acid and 2,3 dihydroxypropyl ester extracted from the methanol extract of Ulva prolifera [6], and gossonorol and margaric acid purified from Gracilaria lemaneiformis ethanolic extract [23]. Currently and according to our knowledge, unlike the essential oils of plants and plant parts that have been evaluated for their potential anti-cyanobacterial properties, there is no scientific report describing the anti-prokaryotic activity of the essential oils from seaweeds [24–28]. In this respect, the present study aims to uncover for the first time the possible inhibitory effects of essential oils (EOs) extracted from three Moroccan seaweeds that are broadly known by their antimicrobial activities namely Cystoseira tamariscifolia, Sargassum muticum, and Ulva lactuca, on the growth of Microcystis aeruginosa a cyanobacteria species that commonly form HABs in Moroccan freshwaters. In addition, this study also provides first insights regarding the anti-cyanobacterial mechanism of EOs, analyzing the growth inhibition power through the following indicators: measurement of chlorophyll-a, protein contents, and activity of cellular stress response enzymes of the stated strain.

2. Results

2.1. Chemical Composition of Moroccan Seaweed EOs

The EO total content based on the dry weight of the seaweed materials are presented in Table 1. The highest total content was achieved with the EO extracted from the green macroalgae *U. lactuca* (0.187% \pm 0.078%) followed by the brown seaweed *S. muticum* EO (0.106% \pm 0.017%). While, the lowest total content was recorded by the brown seaweed *C. tamariscifolia* EO (0.062% \pm 0.018%).

Table 1. Total content of studied seaweed essential oils.

| Species | EO Total Content (%, v/w) |
|---------------------------|---------------------------|
| Ulva lactuca | 0.19 ± 0.08 |
| Sargassum muticum | 0.11 ± 0.02 |
| Cystoseira tamariscifolia | 0.06 ± 0.02 |

The chemical composition of seaweed EOs was identified qualitatively and quantitatively by GC-MS analysis. The content, expressed in percentage, of the individual components of each seaweed, and retention indices, are summarized in Table 2. Forty constituents were determined in *C. tamariscifolia* EO, corresponding to 59.6% from the total compounds in this species. The major components in this EO were palmitic acid (7.7%) followed by dihydroactinidioide (6.57%), hexahydrofarnesyl acetone (5.1%), heptadecane (4.14%), and phytol (4.1%), while other compounds were present below 4%. In total, 41 compounds (corresponding to 45.7% from the total compounds in this species) were identified in the total *S. muticum* EO composition, the most predominant compounds were found to be palmitoleic acid (7.8%), dihydroactinidioide (6.97%) and benzeneacetaldehyde (4.62%). Whereas, the EO extracted from *U. lactuca* revealed the presence of the highest number of compounds; 45 compounds (corresponding to 55% from the total compounds in this species), with dominance of eicosapentaenoic acid (8%), dihydroactinidioide (7.8%), and β-ionone (7.6%).

Table 2. Chemical composition of essential oils extracted from Moroccan seaweeds (%). Values in bold represent the major compounds present in each sample.

| | | | | | | Relative % c | |
|----|--------------------------------------|----------|------------------|------------------|-------------------|---------------------|---------------------|
| N° | Compound | RT (min) | LRI ^a | LRI ^b | Ct | Sm | Ul |
| 1 | (E)-2-Pentenal | 6.27 | 750 | 744 | - | - | 0.086 ± 0.002 |
| 2 | 4-Methyl-2-pentanol | 6.42 | 754 | 745 | - | - | 0.0063 ± 0.0002 |
| 3 | Toluene | 6.69 | 763 | 756 | - | - | 0.143 ± 0.004 |
| 4 | Hexanal or <i>n</i> -Caproylaldehyde | 7.92 | 800 | 801 | 0.196 ± 0.001 | 0.44 ± 0.02 | 0.23 ± 0.01 |
| 5 | Furfural | 9.24 | 828 | 828 | 0.19 ± 0.01 | 0.069 ± 0.002 | - |
| 6 | 4-Hexen-3-one | 9.41 | 832 | - | - | - | 0.075 ± 0.001 |
| 7 | 3-Hexen-2-one | 9.73 | 839 | 834 * | - | - | 0.046 ± 0.003 |
| 8 | (E)-2-Hexenal | 10.17 | 848 | 846 | 0.085 ± 0.003 | 0.35 ± 0.01 | 0.347 ± 0.003 |
| 9 | 2-Furanmethanol | 10.37 | 852 | 853 * | 0.57 ± 0.01 | - | - |
| 10 | 1-Hexanol | 11.11 | 868 | 863 | - | 0.063 ± 0.005 | 0.023 ± 0.001 |
| 11 | 4-Cyclopentene-1,3-dione | 11.67 | 880 | 880 * | 0.6 ± 0.01 | 0.02 ± 0 | - |
| 12 | 2-Heptanone | 12.05 | 888 | 889 | - | 0.029 ± 0.001 | 0.35 ± 0.01 |
| 13 | cis-4-Heptenal | 12.47 | 897 | 893 | - | 0.115 ± 0.005 | 0.28 ± 0.01 |
| 14 | n-Heptanal | 12.57 | 900 | 901 | 0.29 ± 0.01 | 0.136 ± 0.003 | 0.293 ± 0.002 |
| 15 | Acetylfuran | 12.99 | 908 | 909 | 0.2 ± 0.005 | 0.176 ± 0.002 | - |
| 16 | 2-Cyclohexen-1-one | 13.98 | 927 | 927 * | 0.102 ± 0.002 | - | - |
| 17 | α-Pinene | 14.20 | 931 | 932 | - | 0.0151 ± 0.0001 | - |
| 18 | Cyclohexen-2-one | 14.22 | 931 | - | - | - | 0.07 ± 0.001 |
| 19 | Hept-3-en-2-one | 14.37 | 934 | 927 | - | 0.038 ± 0.004 | 0.052 ± 0.003 |
| 20 | Benzaldehyde | 15.00 | 956 | 952 | 0.31 ± 0.01 | 0.38 ± 0.01 | 0.569 ± 0.004 |
| 21 | 5-Methyl-furfural | 15.71 | 960 | 957 | 0.737 ± 0.001 | 0.988 ± 0.02 | 3.39 ± 0.05 |
| 22 | 3,5,5-Trimethyl-2-hexene | 16.47 | 975 | - | - | - | 0.192 ± 0.004 |
| 23 | 1-Octen-3-ol | 16.63 | 978 | 974 | 0.103 ± 0.004 | 0.2036 ± 0.0001 | - |
| 24 | 2-methyl-3-Octanone | 16.91 | 983 | 985 * | - | 0.318 ± 0.004 | - |
| 25 | 6-Methyl-5-heptene-2-one | 16.99 | 985 | 986 * | - | 0.056 ± 0.004 | - |
| 26 | 3-Methyl-3-cyclohexen-1-one | 17.00 | 985 | - | - | - | 0.69 ± 0.02 |
| 27 | Octanal | 17.81 | 1000 | 998 | 0.206 ± 0.003 | - | - |

Table 2. Cont.

| N° 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 | CompoundIPyrrole-2-carboxaldehyde(E,E)-2,4-Heptadienal4-Oxohex-2-enal2,2,6-Trimethyl-CyclohexanoneBenzeneacetaldehydey-Hexalactone2,4,4-Trimethyl-2-cyclohexen-1-ol(E)-2-Octenal(R)-3,5,5-Trimethylcyclohex-3-en-1-ol3-Methyl-benzaldehyde1-Octanol3,5-Octadien-2-onePhenylethyl AlcoholIsophorone4-OxoisophoroneIsomenthone2,6-Nonadienal, (E,Z)(E)-2-Nonenal1-Phenyl-1-propanone2,2,6-Trimethyl-1,4-cyclohexanedione2,4-Dimethyl-benzaldehyde1-(4-Methylphenyl)-ethanonep-Methylacetophenone | XT (min) 18.12 18.16 19.49 19.56 19.93 20.39 20.42 20.66 21.09 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 26.35 | LRI ^a 1006 1007 1033 1034 1041 1050 1051 1055 1063 1067 1069 1093 1110 1120 1141 1152 1153 1157 1165 | LRI b 1008 * 1005 - 1036 * 1036 1049 - 1064 1063 1093 1115 * 1118 1142 * 1162 * 1157 | Ct 0.29 ± 0.01 1.43 ± 0.03 - 1.9 ± 0.03 - 0.65 ± 0.01 0.104 ± 0.001 - 0.23 ± 0.004 0.29 ± 0.01 | $\begin{array}{c} \mathbf{Sm} \\ 0.17 \pm 0.01 \\ 0.09 \pm 0.01 \\ - \\ - \\ \mathbf{4.62 \pm 0.04} \\ 0.42 \pm 0.01 \\ - \\ 0.149 \pm 0.01 \\ - \\ - \\ - \\ 1.14 \pm 0.02 \\ - \\ 0.813 \pm 0.005 \\ 0.485 \pm 0.003 \end{array}$ | U1 0.466 ± 0.003 0.83 ± 0.04 0.822 ± 0.003 1.14 ± 0.01 - 0.187 ± 0.003 0.26 ± 0.01 0.63 ± 0.02 0.51 ± 0.01 0.341 ± 0.01 |
|--|---|--|---|---|--|---|---|
| 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 | Pyrrole-2-carboxaldehyde (E,E)-2,4-Heptadienal 4-Oxohex-2-enal 2,2,6-Trimethyl-Cyclohexanone Benzeneacetaldehyde γ -Hexalactone 2,4,4-Trimethyl-2-cyclohexen-1-ol (E)-2-Octenal (R)-3,5,5-Trimethyl-2clohex-3-en-1-ol 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | $18.12 \\18.16 \\19.49 \\19.56 \\19.93 \\20.39 \\20.42 \\20.66 \\21.09 \\21.27 \\21.39 \\22.60 \\23.49 \\23.95 \\25.01 \\25.56 \\25.61 \\25.61 \\25.81 \\26.19 \\26.33 \\26.57 \\25.57 \\26.57 \\$ | $\begin{array}{c} 1006 \\ 1007 \\ 1033 \\ 1034 \\ 1041 \\ 1050 \\ 1051 \\ 1055 \\ 1063 \\ 1067 \\ 1069 \\ 1093 \\ 1110 \\ 1120 \\ 1141 \\ 1152 \\ 1153 \\ 1157 \\ 1165 \end{array}$ | 1008 * 1005 - 1036 * 1036 * 1047 - 1049 - 1064 1063 1093 1115 * 1118 1142 * 1162 * 1157 | $\begin{array}{c} & & - \\ 0.29 \pm 0.01 \\ 1.43 \pm 0.03 \\ & - \\ 1.9 \pm 0.03 \\ & - \\ 0.65 \pm 0.01 \\ 0.104 \pm 0.001 \\ & - \\ 0.23 \pm 0.004 \\ & - \\ & - \\ 0.29 \pm 0.01 \\ & - \\ \end{array}$ | $\begin{array}{c} 0.17 \pm 0.01 \\ 0.09 \pm 0.01 \\ \hline \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ -$ | $\begin{array}{c} & -\\ 0.466 \pm 0.003 \\ 0.83 \pm 0.04 \\ 0.822 \pm 0.003 \\ 1.14 \pm 0.01 \\ -\\ & -\\ 0.187 \pm 0.003 \\ 0.26 \pm 0.001 \\ 0.63 \pm 0.02 \\ -\\ & -\\ 0.51 \pm 0.01 \\ 0.341 \pm 0.01 \\ -\\ \end{array}$ |
| 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 | (E,E)-2,4-Heptadienal 4-Oxohex-2-enal 2,2,6-Trimethyl-Cyclohexanone Benzeneacetaldehyde γ -Hexalactone 2,4,4-Trimethyl-2-cyclohexen-1-ol (E)-2-Octenal (R)-3,5,5-Trimethylcyclohex-3-en-1-ol 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 18.16 19.49 19.56 19.93 20.39 20.42 20.66 21.09 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 | $\begin{array}{c} 1007 \\ 1033 \\ 1034 \\ 1041 \\ 1050 \\ 1051 \\ 1055 \\ 1063 \\ 1067 \\ 1069 \\ 1093 \\ 1110 \\ 1120 \\ 1141 \\ 1152 \\ 1153 \\ 1157 \\ 1165 \end{array}$ | 1005 - 1036* 1046 1047 - 1049 - 1064 1063 1093 1115* 1118 1142* 1162* 1157 | $\begin{array}{c} 0.29 \pm 0.01 \\ 1.43 \pm 0.03 \\ - \\ 1.9 \pm 0.03 \\ - \\ 0.65 \pm 0.01 \\ 0.104 \pm 0.001 \\ - \\ - \\ 0.23 \pm 0.004 \\ - \\ - \\ 0.29 \pm 0.01 \\ - \\ - \\ 0.29 \pm 0.01 \\ - \\ - \\ \end{array}$ | $\begin{array}{c} 0.09 \pm 0.01 \\ \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - $ | $\begin{array}{c} 0.466 \pm 0.003 \\ 0.83 \pm 0.04 \\ 0.822 \pm 0.003 \\ 1.14 \pm 0.01 \\ - \\ 0.187 \pm 0.003 \\ 0.26 \pm 0.01 \\ - \\ 0.63 \pm 0.02 \\ - \\ 0.51 \pm 0.01 \\ 0.341 \pm 0.01 \\ - \\ 0.341 \pm 0.01 \\ - \\ \end{array}$ |
| 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 | 4-Oxohex-2-enal 2,2,6-Trimethyl-Cyclohexanone Benzeneacetaldehyde y-Hexalactone 2,4,4-Trimethyl-2-cyclohexen-1-ol (E)-2-Octenal (R)-3,5,5-Trimethylcyclohex-3-en-1-ol 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone p-Methylacetophenone | 19.49 19.56 19.93 20.39 20.42 20.66 21.09 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.61 25.81 26.19 26.33 | $\begin{array}{c} 1033 \\ 1034 \\ 1041 \\ 1050 \\ 1051 \\ 1055 \\ 1063 \\ 1067 \\ 1069 \\ 1093 \\ 1110 \\ 1120 \\ 1141 \\ 1152 \\ 1153 \\ 1157 \\ 1165 \end{array}$ | - 1036 * 1046 1047 - 1049 - 1064 1063 1093 1115 * 1118 1142 * 1162 * 1157 | $ \begin{array}{c} 1.43 \pm 0.03 \\ 1.9 \pm 0.03 \\ 0.65 \pm 0.01 \\ 0.104 \pm 0.001 \\ 0.23 \pm 0.004 \\ - \\ 0.29 \pm 0.01 \\ - \\ 0.29 \pm 0.01 \\ - \\ - \\ 0.29 \pm 0.01 \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ -$ | $\begin{array}{c} - \\ \textbf{4.62 \pm 0.04} \\ 0.42 \pm 0.01 \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ $ | $\begin{array}{c} 0.83 \pm 0.04 \\ 0.822 \pm 0.003 \\ 1.14 \pm 0.01 \\ - \\ 0.187 \pm 0.003 \\ 0.26 \pm 0.01 \\ - \\ 0.63 \pm 0.02 \\ - \\ 0.51 \pm 0.01 \\ 0.341 \pm 0.01 \\ - \\ \end{array}$ |
| 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 | 2,2,6-Trimethyl-Cyclohexanone Benzeneacetaldehyde γ -Hexalactone 2,4,4-Trimethyl-2-cyclohexen-1-ol (E)-2-Octenal (R)-3,5,5-Trimethylcyclohex-3-en-1-ol 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 19.56 19.93 20.39 20.42 20.66 21.09 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 | 1034 1041 1050 1051 1055 1063 1067 1069 1093 1110 1120 1141 1152 1153 1157 1165 | 1036 * 1036 1047 - 1049 - 1064 1063 1093 1115 * 1118 1142 * 1162 * 1150 | 1.9 ± 0.03 0.65 ± 0.01 0.104 ± 0.001 $-$ 0.23 ± 0.004 $-$ $-$ 0.29 ± 0.01 | 4.62 ± 0.04 0.42 ± 0.01 - - - - - - - - | $\begin{array}{c} 0.83 \pm 0.04 \\ 0.822 \pm 0.003 \\ 1.14 \pm 0.01 \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$ |
| 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 | Benzeneacetaldehyde γ-Hexalactone 2,4,4-Trimethyl-2-cyclohexen-1-ol (E)-2-Octenal (R)-3,5,5-Trimethylcyclohex-3-en-1-ol 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone 4-Oxoisophorone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone p-Methylacetophenone | 19.93 20.39 20.42 20.66 21.09 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 | $\begin{array}{c} 1041\\ 1050\\ 1051\\ 1055\\ 1063\\ 1067\\ 1069\\ 1093\\ 1110\\ 1120\\ 1141\\ 1152\\ 1153\\ 1157\\ 1165\end{array}$ | 1036 1047 - 1064 1063 1093 1115 * 1118 1142 * 1162 * 1150 1157 | 1.9 ± 0.03 0.65 ± 0.01 0.104 ± 0.001 $-$ 0.23 ± 0.004 $-$ 0.29 ± 0.01 | 4.62 ± 0.04 0.42 ± 0.01 - 0.149 ± 0.01 - - 1.14 ± 0.02 - 0.813 ± 0.005 0.485 ± 0.003 | $\begin{array}{c} 0.822 \pm 0.003 \\ 1.14 \pm 0.01 \\ - \\ - \\ 0.187 \pm 0.003 \\ 0.26 \pm 0.01 \\ - \\ 0.63 \pm 0.02 \\ - \\ 0.51 \pm 0.01 \\ 0.341 \pm 0.01 \\ - \\ - \end{array}$ |
| 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 | y-Hexalactone 2,4,4-Trimethyl-2-cyclohexen-1-ol (E)-2-Octenal (R)-3,5,5-Trimethylcyclohex-3-en-1-ol 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i>-Methylacetophenone | 20.39 20.42 20.66 21.09 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.56 25.61 25.81 26.19 26.33 26.67 | $\begin{array}{c} 1050\\ 1051\\ 1055\\ 1063\\ 1067\\ 1069\\ 1093\\ 1110\\ 1120\\ 1141\\ 1152\\ 1153\\ 1157\\ 1165\end{array}$ | 1047 - 1064 1063 1093 1115 * 1118 1142 * 1162 * 1150 | $\begin{array}{c} - & 0.01 \\ 0.65 \pm 0.01 \\ - & - \\ 0.23 \pm 0.004 \\ - & - \\ 0.29 \pm 0.01 \\ - & - \\ 0.29 \pm 0.01 \end{array}$ | 0.42 ± 0.01 | $\begin{array}{c} 1.14 \pm 0.01 \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$ |
| 34 35 36 37 38 39 40 41 42 43 44 45 46 47 40 | 2,4,4-Irimethyl-2-cyclohexen-1-ol (E)-2-Octenal (R)-3,5,5-Trimethylcyclohex-3-en-1-ol 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,4-Dimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 20.42 20.66 21.09 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.51 25.81 26.19 26.33 | 1051 1055 1063 1067 1069 1093 1110 1120 1141 1152 1153 1157 1165 | 1049 - 1064 1063 1093 1115 * 1118 1142 * 1162 * 1150 1157 | $\begin{array}{c} 0.65 \pm 0.01 \\ 0.104 \pm 0.001 \\ \hline \\ 0.23 \pm 0.004 \\ \hline \\ 0.29 \pm 0.01 \\ \hline \\ 0.29 \pm 0.01 \\ \hline \end{array}$ | $\begin{array}{c} - \\ 0.149 \pm 0.01 \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ $ | $0.187 \pm 0.003 \\ 0.26 \pm 0.01 \\ 0.63 \pm 0.02 \\ 0.51 \pm 0.01 \\ 0.341 \pm 0.01 \\ 0.341 \pm 0.01 \\ 0.0$ |
| 35 36 37 38 39 40 41 42 43 44 45 46 47 40 | (E)-2-Octenal (R)-3,5,5-Trimethylcyclohex-3-en-1-ol 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone p-Methylacetophenone | 20.66 21.09 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 | 1055 1063 1067 1069 1093 1110 1120 1141 1152 1153 1157 1165 | 1049 - 1064 1063 1093 1115 * 1118 1142 * 1162 * 1150 1157 | 0.104 ± 0.001 0.23 ± 0.004 0.23 ± 0.004 0.29 ± 0.01 | 0.149 ± 0.01 - - 1.14 ± 0.02 - 0.813 ± 0.005 0.485 ± 0.003 | $0.187 \pm 0.003 \\ 0.26 \pm 0.01 \\ 0.63 \pm 0.02 \\ 0.51 \pm 0.01 \\ 0.341 \pm 0.01 \\ -$ |
| 36 37 38 39 40 41 42 43 44 45 46 47 | (R)-3,5,5-Irimethylcyclohex-3-en-1-ol 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone p-Methylacetophenone | 21.09 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 | 1063 1067 1069 1093 1110 1120 1141 1152 1153 1157 1165 | 1064 1063 1093 1115 * 1118 1142 * 1162 * 1150 | 0.23 ± 0.004 - - - - - - - - - - - - - - - - - - - | - - - - - - - - - - - - - - - - - - - | $\begin{array}{c} 0.187 \pm 0.003 \\ 0.26 \pm 0.01 \\ \hline \\ 0.63 \pm 0.02 \\ \hline \\ 0.51 \pm 0.01 \\ 0.341 \pm 0.01 \\ \hline \\ \end{array}$ |
| 37 38 39 40 41 42 43 44 45 46 47 48 | 3-Methyl-benzaldehyde 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone p-Methylacetophenone | 21.27 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 26.67 | 1067 1069 1093 1110 1120 1141 1152 1153 1157 1165 | 1064 1063 1093 1115 * 1118 1142 * 1162 * 1150 | 0.23 ± 0.004 - 0.29 ± 0.01 | $\begin{array}{c} - \\ - \\ - \\ - \\ 0.813 \pm 0.005 \\ 0.485 \pm 0.003 \end{array}$ | 0.26 ± 0.01 |
| 38 39 40 41 42 43 44 45 46 47 48 | 1-Octanol 3,5-Octadien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone p-Methylacetophenone | 21.39 22.60 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 26.67 | 1069 1093 1110 1120 1141 1152 1153 1157 1165 | 1063 1093 1115 * 1118 1142 * 1162 * 1150 | 0.23 ± 0.004 - - 0.29 ± 0.01 - | - 1.14 ± 0.02 - 0.813 ± 0.005 0.485 ± 0.003 | 0.63 ± 0.02 0.51 ± 0.01 0.341 ± 0.01 |
| 40 41 42 43 44 45 46 47 48 | 3,5-Octatien-2-one Phenylethyl Alcohol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone p-Methylacetophenone | 22.60 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 26.67 | 1093 1110 1120 1141 1152 1153 1157 1165 | 1093 1115 * 1118 1142 * 1162 * 1150 | - - 0.29 ± 0.01 - | 1.14 ± 0.02 0.813 ± 0.005 0.485 ± 0.003 | 0.63 ± 0.02 - 0.51 ± 0.01 0.341 ± 0.01 |
| 40 41 42 43 44 45 46 47 48 | Interlytertyl Alconol Isophorone 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 23.49 23.95 25.01 25.56 25.61 25.81 26.19 26.33 | 1110 1120 1141 1152 1153 1157 1165 | 1115 4 1118 1142 * 1162 * 1150 | 0.29 ± 0.01 | 1.14 ± 0.02 - 0.813 ± 0.005 0.485 ± 0.003 | 0.51 ± 0.01 0.341 ± 0.01 |
| 41 42 43 44 45 46 47 48 | 4-Oxoisophorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 25.95 25.01 25.56 25.61 25.81 26.19 26.33 | 1120 1141 1152 1153 1157 1165 | 1118 1142 * 1162 * 1150 | 0.29 ± 0.01 | 0.813 ± 0.005 0.485 ± 0.003 | 0.31 ± 0.01 0.341 ± 0.01 |
| 42 43 44 45 46 47 48 | I-Oxologhiorone Isomenthone 2,6-Nonadienal, (E,Z) (E)-2-Nonenal I-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 25.01 25.56 25.61 25.81 26.19 26.33 | 1141 1152 1153 1157 1165 | 1142 * 1162 * 1150 | 0.29 ± 0.01 | 0.813 ± 0.003 0.485 ± 0.003 | 0.341 ± 0.01 |
| 45 44 45 46 47 | 2,6-Nonadienal, (E,Z) (E)-2-Noneal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 25.36 25.61 25.81 26.19 26.33 | 1152 1153 1157 1165 | 1150 | - | 0.465 ± 0.005 | - |
| 44 45 46 47 | 2,6-Rohadehal, (2,2) (E)-2-Nonenal 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 25.81 26.19 26.33 | 1155 1157 1165 | 1150 | | | 0.25 ± 0.005 |
| 45 46 47 | 1-Phenyl-1-propanone 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 25.81 26.19 26.33 | 1157 | / | - | - | 0.35 ± 0.005 |
| 40 47 | 2,2,6-Trimethyl-1,4-cyclohexanedione 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 26.19 | 1100 | 1157 | 0.279 ± 0.005 | - | - |
| 47 | 2,2,5,0 minetry1-1,4-Cyclonexanethole 2,4-Dimethyl-benzaldehyde 1-(4-Methylphenyl)-ethanone <i>p</i> -Methylacetophenone | 26.33 | 1167 | - | - | - | 0.49 ± 0.01 |
| | 1-(4-Methylphenyl)-ethanone p-Methylacetophenone | · · · · · · · · · · · · · · · · · · · | 110/ | - | - | - | 0.274 ± 0.003 |
| 40 | <i>p</i> -Methylacetophenone | 20.07 | 11/5 | 11/5 | - 0.42 + 0.01 | - | 0.505 ± 0.01 |
| 49 50 | <i>p</i> -memylacetophenone | 27.02 | 1101 | 1102 | 0.45 ± 0.01 | - | 0 54 + 0.02 |
| 50 E1 | Cafuanal | 27.15 | 1184 | 11/9 | - | - | 0.54 ± 0.03 |
| 51 | Salfallal | 27.02 | 1190 | 1210 | 0.97 ± 0.01 | - | 1.6 ± 0.1 |
| 52 | p-Cyclocitrai Ethydraethydraelainyida | 28.84 | 1219 | 1219 | -1 = 62 + 0.02 | 0.479 ± 0.001 | 0.614 ± 0.002 |
| 53 E4 | Ethylmethylmaleimide | 29.67 | 1237 | 1234 | 1.362 ± 0.03 | 3.69 ± 0.03 | 0.59 ± 0.01 |
| 54 | r ulegone | 29.75 | 1236 | 1255 | - | 0.375 ± 0.005 | - |
| 55 | 2,6,6-Irimethyl-1-Cyclonexene- 1-acetaldehyde | 30.70 | 1259 | 1253 * | - | - | 0.53 ± 0.01 |
| 56 | 2.3.6-Trimethyl-7-octen-3-ol | 31.53 | 1277 | - | - | 1.846 ± 0.01 | - |
| 57 | Indole | 32 47 | 1297 | 1290 | - | - | 1.026 ± 0.003 |
| 58 | Carvacrol | 32.69 | 1302 | 1298 | 0.98 ± 0.02 | - | - |
| 59 | v-Amylbutyrolactone | 35.29 | 1361 | 1362 * | - | 1.03 ± 0.02 | - |
| 60 | Capric acid | 35.97 | 1376 | - | 1.15 ± 0.02 | - | - |
| 61 | Fumaric acid, ethyl 2-methylallyl | 26.27 | 1292 | | | 1.45 ± 0.02 | |
| 61 | ester | 30.27 | 1565 | - | - | 1.45 ± 0.05 | - |
| 62 | β-Caryophyllene | 37.91 | 1421 | 1417 | - | 0.02 ± 0.004 | - |
| 63 | α-Ionone | 38.22 | 1428 | 1428 | 3.23 ± 0.02 | 3.07 ± 0.03 | 1.1 ± 0.01 |
| 64 | Nerylacetone | 39.27 | 1454 | 1434 | - | - | 0.26 ± 0.01 |
| 65 | β-Ionone | 40.88 | 1492 | 1488 | 1.3 ± 0.03 | - | 7.6 ± 0.2 |
| 66 | Dihydroactinidiolide | 42.49 | 1536 | 1538 * | 6.577 ± 0.004 | 6.971 ± 0.003 | 7.8 ± 0.2 |
| 67 | Lauric acid | 43.66 | 1562 | 1565 | 2.9 ± 0.1 | 0.49 ± 0.01 | 1.2 ± 0.1 |
| 68 | Fumaric acid, ethyl 2-Methylallyl ester | 44.46 | 1583 | - | - | - | 3.0 ± 0.1 |
| 69 | Tridecanoic acid | 47.61 | 1666 | 1662 | - | - | 0.1979 ± 0.000 |
| 70 | 3-Keto-β-ionone | 47.77 | 1670 | 1661 * | - | 1.29 ± 0.02 | - |
| | 4-(4-hydroxy-2 2 6-trimethyl-7-oxabicyclo | 1 | 1070 | 1001 | | 112/ 2 0102 | |
| 71 | [4.1.0]hept-1-v])-3-Buten-2-one | 48.40 | 1687 | 1690 | - | 1.2 ± 0.1 | - |
| 72 | Heptadecane | 48.77 | 1697 | 1700 | 4.14 ± 0.04 | - | - |
| 73 | Pentadecanal | 49.37 | 1711 | 1713 | | - | 0.27 ± 0.01 |
| 74 | Myristic acid | 50.61 | 1769 | 1765 * | 22 ± 01 | 216 ± 0.01 | 1.855 ± 0.001 |
| 75 | Pentadecanoic acid | 51.81 | 1820 | 1869 | - | 2.10 ± 0.01 | 0.121 ± 0.004 |
| 76 | Hexabydrofarnesyl acetone | 52.07 | 1847 | - | 51 ± 01 | _ | - |
| 77 | 2-Pentadecanone, 6,10,14-trimethyl | 52.11 | 1843 | 1847 | - | - | 0.23 ± 0.01 |
| 70 | Methyl | ED (9 | 1005 | | | | 0.15 + 0.01 |
| 10 | 4,7,10,13-hexadecatetraenoate | 52.00 | 1003 | - | - | - | 0.13 ± 0.01 |
| 79 | Eicosane | 52.80 | 1895 | - | 0.22 ± 0.02 | - | - |
| 80 | Palmitoleic acid | 53.41 | 1948 | 1953* | - | 7.8 ± 0.1 | - |
| 81 | Eicosapentaenoic acid | 53.57 | 1962 | - | - | - | 8.0 ± 0.2 |
| 82 | Palmitic acid | 53.64 | 1968 | 1959 | 7.7 ± 0.1 | 0.73 ± 0.01 | 2.887 ± 0.02 |
| 83 | Phytol | 55.00 | 2113 | 2111 * | 4.1 ± 0.1 | 0.38 ± 0.03 | 0.23 ± 0.01 |
| 84 | Linolenic acid | 55.36 | 2159 | 2134 * | - | - | 1.2 ± 0.1 |
| 85 | Eicosanal | 55.83 | 2223 | 2224 | 0.7 ± 0.1 | - | - |
| 86 | 1-Hexacosanol | 56.25 | 2283 | 2906 | 1.39 ± 0.05 | - | - |
| 87 | Henicosanal | 56.52 | 2325 | 2329 | 0.89 ± 0.04 | - | - |
| 88 | Docosanal | 57.15 | 2427 | 2434 | 1.38 ± 0.04 | - | - |
| 89 | 1-Docosanol | 57.51 | 2488 | 2470 | 2.423 ± 0.004 | - | - |
| 90 | Tricosanal | 57.75 | 2529 | 2534 | 2.8 ± 0.1 | - | - |
| 91 | Bis (2-ethylhexyl) phthalate | 57.95 | 2562 | 2550 * | _ | - | 0.5 ± 0.1 |
| | Total identified (%) | | | | 59.6 ± 0.1 | 457 ± 01 | 55 + 1 |
| | Not identified (%) | | | | 40.4 ± 0.1 | 54.3 ± 0.1 | 45 ± 1 |

^a LRI, linear retention index determined on a DB-5 MS fused silica column relative to a series of *n*-alkanes (C8–C40). ^b Linear retention index reported in literature (Adams, 2017). ^c Relative % is given as mean \pm SD, *n* = 3. * NIST Standard Reference Database 69: NIST Chemistry WebBook. Ct. *Cystoseira tamariscifolia;* Sm. *Sargassum muticum;* UI. *Ulva lactuca*.

2.2. Screening of Anti-Cyanobacterial Activity

The potential anti-cyanobacterial properties of seaweed EOs was evaluated qualitatively using the disk diffusion methods. After 1 week of incubation, the inhibition zones were measured and the results are presented in Table 3 and Figure 1. The results show that the tested green macroalgae U. lactuca did not reveal any inhibitory activity against M. aeruginosa, while both brown algae showed algicidal activity against the tested cyanobacteria. The most relevant activity was observed with C. tamariscifolia EO with zones of inhibition greater than 46 mm. Notably, the growth inhibition of C. tamariscifolia EO was approximately similar to the positive control, copper sulphate (CuSO₄), that presented a growth inhibition diameter of 45.3 mm. Furthermore, S. muticum EO showed moderate activity (zone of inhibition was 32 mm). The activity of seaweed EOs against *M. aeruginosa* was determined quantitatively by means of broth microdilution technique. Calculation of minimal inhibitory concentrations (MIC) and minimal bactericidal concentrations (MBC) was performed after 1 week of incubation and the results are summarized in Table 4. CuSO₄ as positive control displayed a high potency (MIC = MBC = $3.12 \ \mu g \ mL^{-1}$). The greatest effectiveness was achieved with the EO extracted from *C. tamariscifolia*, with MIC being equal to 7.81 μ g mL⁻¹ and the MBC equal to 15.62 µg mL⁻¹. Whereas, S. muticum EO showed a moderate potency with MIC and MBC values of 62.5 and 125 μ g mL⁻¹, respectively.

Table 3. Inhibition-zone diameters, minimal inhibitory concentrations (MIC) and minimal bactericidal concentrations (MBC) of Moroccan seaweed essential oils (EOs).

| Treatments | Inhibition Zone (mm) | MIC ($\mu g m L^{-1}$) | MBC ($\mu g \ mL^{-1}$) |
|-------------------|----------------------|--------------------------|---------------------------|
| C. tamariscifolia | 46.3 ± 0.6 *** | 7.81 | 15.62 |
| S. muticum | 32.3 ± 0.6 *** | 62.5 | 125 |
| U. lactuca | n.a | n.a | n.a |
| CuSO ₄ | 45.3 ± 0.6 *** | 3.12 | 3.12 |
| DMSO | n.a | n.a | n.a |

Each value representing mean \pm SD of six replicates, *** p < 0.001 indicates significant differences compared with DMSO, n.a not active.



Figure 1. Anti-cyanobacterial activity of the active tested EOs against *M. aeruginosa* on solid media. (A) *C. tamariscifolia;* (B) *Sargassum muticum*.

| | Inhibition Rate (%) | | | | | | | |
|-------------------|---------------------|--------------------|------------------|----------------------|----------------------|----------------------|--|--|
| Treatments | Time (Days) | | | | | | | |
| | 1 | 2 | 3 | 4 | 5 | 6 | | |
| MIC | 68.0 ± 0.4 *** | 87.6 ± 0.4 *** | 90.2 ± 0.5 *** | 95.4 ± 0.1 *** | 96.16 ± 0.08 *** | 97.85 ± 0.05 *** | | |
| MBC | 74 ± 1 *** | 89.9 ± 0.2 *** | 94.4 ± 0.3 *** | 97.8 ± 0.1 *** | 98.81 ± 0.07 *** | 99.24 ± 0.07 *** | | |
| CuSO ₄ | 71 ± 2 *** | 88.9 ± 0.2 *** | 94.12 ± 0.07 *** | $97.54 \pm 0.05 ***$ | 98.5 ± 0.02 *** | 98.87 ± 0.01 *** | | |
| DMSO | -0.3 ± 0.5 | -0.08 ± 1.67 | -0.15 ± 0.39 | -1.0 ± 0.6 | -1 ± 2 | -1.1 ± 0.9 | | |

Table 4. Inhibitory rate of Cystoseira tamariscifolia EO on Microcystis aeruginosa.

MIC: minimum inhibitory concentration of *C. tamariscifolia* EO (7.81 µg mL⁻¹) and MBC: minimum bactericidal concentration of *C. tamariscifolia* EO (15.62 µg mL⁻¹). Each value representing mean \pm SD of three replicates, *** *p* < 0.001 indicate significant differences compared with DMSO.

2.3. Physiological Effects of C. tamariscifolia EO on M. aeruginosa

2.3.1. Inhibitory and Growth Rates of C. tamariscifolia EO on Tested Cyanobacteria

The physiological effects caused by the action of the essential oil on the cyanobacteria cells were accessed only for the seaweed EO that showed the highest activity on the qualitative assay (Table 3 and Figure 1). The algicidal effects of C. tamariscifolia EO at the MIC and MBC concentrations (7.81 and 15.68 μ g mL⁻¹, respectively) against *M. aeruginosa* are shown as the inhibition and growth rates in Table 4 and Figure 2. The results indicate that C. tamariscifolia EO had a significant inhibitory effect on the tested cyanobacteria compared with the negative control (DMSO), which did not show any inhibitory effect. On the first day of the experience, C. tamariscifolia EO showed strong growth inhibition for both tested concentrations in a concentration-dependent way. The IR was $67.95\% \pm 0.38\%$ and 73.93% \pm 0.98% at the MIC (7.81 µg mL⁻¹) and MBC (15.68 µg mL⁻¹), respectively. Thereafter, the inhibition rates increased and set at more than 85% along the experiment, for both used concentrations. The maximum IR was recorded at the last day of treatment with IR of $99.24\% \pm 0.07\%$ at the MBC concentration. The generation time of M. aeruginosa was 1.23/day with the growth rates of 0.75/day under standard culturing conditions and negative control treatment. Under treatment with CuSO₄, the growth rate and the generation time of the tested cyanobacteria decreased significantly with μ value of -0.18/day and generation time of -3.77/day. The C. tamariscifolia EO revealed a strong effect on the growth of *M. aeruginosa* at both tested concentration (the μ value was less than -0.08/day).



Figure 2. Effect of MIC and MBC, *C. tamariscifolia* EO on the growth rate of *M. aeruginosa*. MIC: minimum inhibitory concentration and MBC: minimum bactericidal concentration. Each value representing mean \pm SD of three replicates. *** p < 0.001 indicate significant differences compared with the untreated culture.

2.3.2. Morphological Changes of M. aeruginosa Cells

The visual observation of the tested unicellular *M. aeruginosa* cultures under *C. tamariscifolia* EO at both used concentrations showed that after 3 days of exposure, a blue color appeared in the treated groups and became colorless after the fifth day of treatment. While, microscope observation at a magnification of ×40 showed that the unicellular cyanobacteria strain becomes colonial on the second day of treatment under stress with *C. tamariscifolia* EO. Moreover, similar morphological changes were also observed under treatment with the positive control (CuSO₄) (Figure 3).



Figure 3. Micrographs of *M. aeruginosa* (Gr. \times 40) (**A**) At the first day of treatment; (**B**) untreated culture at the end of treatment; (**C**) culture under treatment with *C. tamariscifolia* EO at second day of the experience; (**D**) culture under treatment with CuSO₄ at second day of the experience.

2.3.3. Effects of C. tamariscifolia EO on M. aeruginosa Chlorophyll-a and Protein Contents

To explore whether chlorophyll-*a* synthesis of the treated cells was inhibited by *C. tamariscifolia* EO, the chlorophyll-*a* content of *M. aeruginosa* cells after exposure to *C. tamariscifolia* EO are shown in Figure 4. At the first day of experience, the chlorophyll-*a* content was the same at all treatments (including control). With exposure time, the chlorophyll-*a* content increased significantly for the untreated culture and the negative control. While, *M. aeruginosa* chlorophyll-*a* content was significantly decreased by both used concentrations of *C. tamariscifolia* EO. Compared to CuSO₄ used as positive control, *C. tamariscifolia* EO recorded a similar effect on *M. aeruginosa* chlorophyll-*a* content during the 6 days of experiment.



Figure 4. Effect of *C. tamariscifolia* EO on *M. aeruginosa* chlorophyll-*a* concentration. Results are presented as mean \pm SD of three independent assays (*** indicates *p* < 0.001 relative to the untreated culture by ANOVA).

To investigate the effects of *C. tamariscifolia* EO on *M. aeruginosa* cells, the protein content of *M. aeruginosa* cells was determined after exposure to *C. tamariscifolia* EO, and the results are shown in Figure 5. Similar to chlorophyll-*a*, for the untreated culture and the negative control, the protein content increased significantly with exposure time. On day 2, the differences in protein content for the MIC and MBC concentrations of *C. tamariscifolia* EO were significantly different, compared with the negative control (untreated culture and DMSO). Similar results were recorded on days 4 and 6, *C. tamariscifolia* EO (at both used concentrations) and CuSO₄ (positive control) recorded a significant reducing effect on the protein content of *M. aeruginosa* cells. Furthermore, the protein contents were highly coherent with the cell density and the chlorophyll-*a* results.



Figure 5. Effect of *C. tamariscifolia* EO on *M. aeruginosa* protein content. Results are presented as mean \pm SD of three independent assays (*** indicates *p* < 0.001 relative to the untreated culture by ANOVA).

2.3.4. Effects of *C. tamariscifolia* EO Superoxide Dismutase (SOD) and Catalase (CAT) Activities and Malondialdehyde (MDA) Concentration in *M. aeruginosa* Cells

In order to determine whether the cellular oxidative defense system was activated, the superoxide dismutase (SOD) activity, as the first defense against Reactive Oxygen Species (ROS) among antioxidant systems, which catalyzes the superoxide anion into H_2O_2 and O_2 , was investigated (Figure 6A). The results demonstrate that the differences in the SOD activities in cyanobacterial cells between control and treatment groups were significant and visible from the second day of treatment. From the second day of experiment, the SOD activity under the MBC concentration (15.62 µg mL⁻¹) of *C. tamariscifolia* EO treatment became higher than the positive control (CuSO₄) and reached the peak of 150.25 U/mg protein. Thereafter, the SOD activity began to decrease gradually in the following time in the control and treatment groups.



Figure 6. Superoxide dismutase (SOD) (**A**) and catalase (CAT) (**B**) activities in *M. aeruginosa* cells after treatment with *C. tamariscifolia* EO. Results are presented as mean \pm SD of three independent assays (*** indicates *p* < 0.001 relative to the untreated culture by ANOVA).

The second defense against reactive oxygen species (ROS) is catalase (CAT), which can convert H_2O_2 into H_2O and directly eliminates H_2O_2 in the peroxisome. As shown in Figure 6B the CAT activity in *M. aeruginosa* cells exposed to *C. tamariscifolia* EO exhibited a significant increase, while the CAT activity in the untreated culture and the negative control remained unchanged over time. The differences between treated and control groups were apparent from the second day of treatment. The CAT activity at 15.62 µg mL⁻¹ of *C. tamariscifolia* EO (MBC concentration) was higher than that at 7.81 µg mL⁻¹ of *C. tamariscifolia* EO (MIC concentration); however, the differences between them were not significant. After 4 days of exposure, the activity of CAT greatly increased and reached a maximum value with the exposure to 15.62 µg mL⁻¹ *C. tamariscifolia* EO. At the sixth day of treatment, the CAT activity with *C. tamariscifolia* EO decreased but was still significantly different compared with the untreated culture.

Malondialdehyde (MDA, the final product of lipid peroxidation) was used as an indicator of lipid peroxidation, as indicated in Figure 7, the MDA content increased significantly at both tested concentrations of *C. tamariscifolia* EO from the second day of treatment. While, the MDA level in the negative control groups remained unchanged over time. After 48 h of exposure, 15.62 and 7.81 μ g mL⁻¹ *C. tamariscifolia* EO induced an increase in MDA compared with the control. The MDA concentrations increased with the increased concentration of *C. tamariscifolia* EO. The maximal MDA value was

94 μ mol/L at day 4 of treatment with 15.62 μ g mL⁻¹ *C. tamariscifolia* EO, which was 4.95 times higher than that in the negative control groups.



Figure 7. Malondialdehyde (MDA) concentration in *M. aeruginosa* cells after treatment with *C. tamariscifolia* EO. Results are presented as mean \pm SD of three independent assays (*** indicates p < 0.001 relative to the untreated culture by ANOVA).

3. Discussion

Seaweeds are one of the most primitive and dominant organisms in aquatic ecosystems. They could provide an eco-friendly approach for HCBs control due to their ability to produce a large range of bioactive compounds [7,29,30]. Therefore, we proceeded to assess the anti-cyanobacterial activity of three Moroccan seaweed EOs. To the best of our knowledge, the present study constitutes the first attempt to extract and characterize Moroccan seaweed EOs. The total content percentage of Moroccan seaweed EOs was lower compared to that reported previously by Patra et al. (2017a, 2015) and Patra and Baek (2016a, 2016b) [31–34] who found that the total content of seaweeds collected from the Korean coast was usually higher than 0.26%. This difference in seaweed total content percentage could be due to the geographical locations, the species used, harvesting time, and the used extraction method. An important richness and variability of compounds was observed after seaweed EOs chemical analysis. Among the total of 91 compounds identified in the three selected algal EOs, 14 compounds showed to belong to the terpenes group. In general, terpenoids are compounds that have been associated with several bioactive properties, including antimicrobial activity [35].

The most abundant constituents in the two brown seaweeds *C. tamariscifolia* and *S. muticum* EOs were fatty acids, namely palmitic acid and palmitoleic acid, respectively. Previous studies on EOs from other species of seaweeds have shown the presence of hexadecenoic acid (palmitoleic acid). In particular, Patra et al. (2017a) [36] reported the presence of this unsaturated fatty acid as one of the main compounds (22.39%) of the brown edible seaweed *Undaria pinnatifida* EO collected from the Korean coast. A high content of palmitic acid (9.2% and 16.57%) was also found by Patra et al. (2017b, 2015a) [31,37] on *Porphyra tenera* and *Laminaria japonica* EOs, respectively. Previously, the EO composition of seaweed species of *Cystoseira* genus, other than *C. tamariscifolia*, has been described. Ozdemir et al. (2006) [38] reported that the brown seaweed *C. barbata* EO (*Cystoseira* genus) consists of several compounds different from those we have determined in *C. tamariscifolia* EO, such as docosane (7.61%), tetratriacontane (7.47%), eicosane (5.05%), tricosane (4.43%), hexadecane (4.16%), and heptadecane (1.35%) as major compounds. Additionally, 1-chloro-2,2-diethoxyethane (2%) were identified in the volatile compounds

composition of *C. crinite* (*Cystoseira* genus) collected from the eastern Mediterranean [39]. Nevertheless, it should be noticed that research on the chemical composition of seaweed EOs is still very scarce. Regarding the chemical composition of the green macroalgae *U. lactuca* EO, the polyunsaturated fatty acid eicosapentaenoic acid (8%) was the most dominant constituent. The presence of other polyunsaturated fatty acids, such as cis- and trans-5,8,11,14-eicosatetraenoic acids were also detected in the *Dictyopteris polypodioides* EO collected from the Algerian coast [40].

Among the identified compounds, some terpenes such as β -caryophyllene (0.02%) and α -pinene (0.015%) were only found in minor amounts, while others such as dihydroactinidiolide (6.6–7.8%), β -Ionone (not detected to 7.6%), and phytol (0.23–4.1%) were present in higher amounts. Terpenes such as safranal and others have been recently reported in the essential oil extracted from the brown algae *D. polypodioides* [40]. Similarly, the presence of the terpenic compounds dihydroactinidiolide, β -Ionone and phytol has been previously described in marine algae [41]. Besides terpenes, several compounds belonging to different groups such as alcohols, aldehydes, and ketones were present in minor amounts. The presence of such compounds is in good agreement with Gressler et al. (2009) [41], who mentioned the ability of marine algae to produce a wide range of metabolites including hydrocarbons, terpenes, fatty acids, esters, alcohols, aldehydes, and ketones.

The qualitative screening using the paper disk diffusion method in solid medium demonstrated that the green macroalgae *U. lactuca* did not show any inhibitory activity against the tested Gram-negative bacteria *M. aeruginosa*. These results are in good agreement with those found by Zerrifi et al. (2019) [30] who investigated the anti-cyanobacterial activity of *U. lactuca* methanolic extract collected from the Moroccan coast. Their results showed that the *U. lactuca* extract also did not have an effect against M. aeruginosa. Similar results were obtained by Salvador et al. (2007) [42] who reported that the seaweeds of the genus *Ulva* did not show antibacterial activity against any assayed Gram-negative bacteria. On the contrary, Mishra (2018) [43] observed that methanol, butanol, and ethyl acetate extract of *U. lactuca* display moderate activity against *Pseudomonas aeruginosa*. Additionally, Begum et al. (2018) [44] reported that the methanolic extract of *U. reticulate* (*Ulva* genus) revealed the maximum inhibitory activity against Gram-negative bacteria P. aeruginosa. Furthermore, we found that S. muticum EO showed a strong activity against *M. aeruginosa* (32.33 mm). Our results are in agreement with those found by Kumaresan et al. (2018) [45] who investigated the antimicrobial activity of S. wightii (Sargassum genus) aqueous extract and showed that this extract had an important antibacterial activity against Gram-negative bacteria Escherichia coli with inhibition zone of 13 mm. Sujatha et al. (2019) [46] observed that S. swartzii ethanolic extract exhibited a high antibacterial activity against Gram-negative bacteria. S. muticum extract was active against Enterobacter aerogenus, Proteus vulgaris, Salmonella typhymurium, and Salmonella paratyphi [47]. In what concerns the activity against M. aeruginosa, Zerrifi et al. (2019) [30] evaluated the methanolic extract of *S. muticum*, which did not reveal any inhibiting capacity, most probably due to the different chemical composition between the methanolic extract and that obtained by hydrodistillation. As mentioned, in the present study the highest activity was recorded for C. tamariscifolia EO (46.33 mm). However, this finding is in disagreement with those found by Farid et al. (2009) [48]. These authors investigated the antibacterial activities of C. tamariscifolia collected from Morocco, and their results suggest that C. tamariscifolia dichloromethane/methanol extract did not show antimicrobial activity against the Gram-negative bacteria assayed (E. coli). Salvador et al. (2007) [42] observed that C. tamariscifolia extract did not inhibit the growth of any tested Gram-negative bacteria. On the other hand, Ozdemir et al. (2006) [38] reported that the volatile oil of the genus Cystoseira recorded a moderate effect on the tested Gram-negative bacteria (7 mm against E. coli and S. typhimurium). Contrarily, Zerrifi et al. (2019) [30] observed that C. tamariscifolia extract conferred an important activity against Gram-negative bacteria *M. aeruginosa* with inhibition zone equal to 13.33 mm. In another report, Chiheb et al. (2009) [49] reported that different Cystoseira species show a potent antibacterial activity against tested Gram-negative bacteria Salmonella typhi, E. coli, P. aeruginosa, and Klebsiella sp. Ainane et al. (2014) [50] found that C. tamariscifolia extract produce interesting zones of inhibition against both tested Gram-negative bacteria, Enterobacter cloacae and Klebsiella' pneumoniae

(inhibition diameter between 10 and 15 mm). Likewise, a recent study showed the high antibacterial activity against Gram-negative bacteria of another species of the *Cystoseira* genus (*C. mediterranea*) [51].

The quantitative screening using the broth microdilution technique confirmed the results of the qualitative assay since C. tamariscifolia EO achieved the greatest effectiveness against M. aeruginosa $(MIC = 7.81 \ \mu g \ mL^{-1} \ and \ MBC = 15.62 \ \mu g \ mL^{-1})$. These results are in accordance with those reported earlier by Wang et al. (2015, 2014) and Zerrifi et al. (2020) [27,28,52] who found that other EOs also showed a high activity against the toxic cyanobacteria *M. aeruginosa*. Our results in the liquid medium revealed that C. tamariscifolia EO recorded a significant anti-cyanobacterial activity against the toxic cyanobacteria M. aeruginosa with a percentage inhibition of more than 67%. Several studies were conducted to investigate growth inhibition by EOs extracted from many aquatic and terrestrial plants and solvent extract of seaweeds on M. aeruginosa. Wang et al. (2014) [52] reviewed the anti-cyanobacterial activity of two emergent plant EOs (Typha latifolia and Arundo donax) on M. aeruginosa. The authors reported inhibition rates of more than 40% at 50.0 mg L^{-1} of both tested EOs. The Rosmarinus officinalis EO recorded significant growth inhibition against *M. aeruginosa* [25]. Moreover, Wang et al. (2015) [27] showed that the growth of *M. aeruginosa* was strongly inhibited by *Vallisneria spinulosa* EO at 50.0 mg L^{-1} with an inhibition rate equal to 41.7%. Xian et al. (2006) [53] found that Ceratophyllum demersum EO composed of fatty compounds, terpenoids, and phenolic compounds, recorded a high inhibitory activity on M. aeruginosa growth. Furthermore, Zerrifi et al. (2019) [30] tested the effect of C. tamariscifolia methanolic extract on the growth of *M. aeruginosa*. Their results show that the reached inhibition rates were higher than 49% after the first day of treatment at 0.6 mg L⁻¹. The morphological changes observed in M. aeruginosa culture in this study are quite similar to those observed by Harada et al. (2009), Huang et al. (2002), and Zerrifi et al. (2020) [28,54,55]. The chlorophyll-a and protein content that reflect *M. aeruginosa* growth, was decreased after *C. tamariscifolia* EO treatment. This decrease could be related to malfunctions of normal physiological metabolism in cyanobacterial cells (e.g., disruption of Photosystem I and destruction of Photosystem II) [56,57]. The mentioned changes can be related to the chemical composition of *C. tamariscifolia* EO, namely to the presence, although in low amounts, of several terpenoids, including oxygenated compounds such as alcohols and aldehydes. Besides, the possibility of synergisms, between the more abundant terpenoids, such as dihydroactinidiolide (6.57%) and hexahydrofarnesyl acetone (5.1%) with compounds present in minor amounts, should also be considered. Finally, all the three studied seaweeds presented a very complex composition, with a large abundance of compounds, several of which were not possible to be identified by the used technique (GC-MS). Some of those unidentified compounds can possibly explain the different activity observed for the three samples.

The activation of SOD and CAT activities were responsible for the protection of *M. aeruginosa* cells against oxidative exposure (eliminate ROS or reduce damaging effects). Our finding of SOD and CAT activities in *M. aeruginosa* were also observed in response to glyphosate treatment [58]. Meng et al. (2015) [59] observed a significant increase in the SOD activity of *M. aeruginosa* exposed to different concentrations of *Ailanthus altissima* extract. Similar results are also found from other treatments, such as rice straw aqueous extract [60], heptanoic acid and benzoic acid [61], fenoxaprop-*p*-ethyl [62], 17b-estradiol [63], and juglone (5-hydroxy-1,4-naphthoquinone) [64] on *M. aeruginosa*. The last product of lipid peroxidation is MDA, which is an indicator of oxidative stress [65]. The increase of MDA concentration is in agreement with Zhang et al. (2017) [66] who detected a significant increase in MDA levels of *M. aeruginosa* exposed to 5 and 10 mg L⁻¹ of glufosinate, comparing to control, 0.5, and 1 mg L⁻¹, indicates the occurrence of damage to the lipid membranes. The treatment of *M. aeruginosa* cells with pyrogallol (polyphenol) caused lipid peroxidation and altered MDA levels [67]. Contrariwise, Xie et al. (2019) [68] found that the MDA concentrations on *M. aeruginosa* cells showed no apparent change under napropamide and acetochlor treatments.

4. Conclusions

After screening of EOs extracted from Moroccan seaweeds for their anti-cyanobacterial activity, our results revealed that marine macroalgae EOs are potential producers of anti-cyanobacteria compounds. Consequently, they should be subject to a comprehensive study as natural sources of bioactive substances. Accordingly, to better understand the potential effects and the mechanisms of action of the studied EOs on *M. aeruginosa*, the search of the active EOs major components effects on *M. aeruginosa* will be the next step of our research. Moreover, further research will need to be conducted using other seaweeds and/or phytoplankton species in macrocosms and natural field conditions, studying the toxicity, nature, and stability of the compounds and their potentially synergistic interactions in the aquatic ecosystem.

5. Materials and Methods

5.1. Seaweed Material Sampling and Extraction of Essential Oils (EOs)

Three seaweed species were selected for the study: *C. tamariscifolia* (Phaephyceae, Sargassaceae), *S. muticum* (Phaephyceae, Sargassaceae), and *U. lactuca* (Ulvophyceae, Ulvaceae). These macro-algae were harvested from two Moroccan coastal regions (Table 5).

| Species | Species Code | Harvesting Place | Date of Harvesting | Latitude/Longitude |
|-------------------|--------------|------------------|-----------------------|--------------------------------|
| C. tamariscifolia | Ct | Souiria Laqdima | February 2019 | N 32°03′04.6″/ W 9°20′30.2″ |
| S. muticum | Sm | El jadida | April 2019 | N 3°15′45.9″/ W 8°30′03.4″ |
| U. lactuca | Ul | El jadida | March 2019 | N 3°15′45.9″/ W 8°30′03.4″ |

Table 5. Date of harvesting and location of the Moroccan seaweeds studied.

The samples were rinsed with seawater and distilled water to remove debris. After identification of each species according to their morphological and histological features [69], seaweed materials were dried in the shade at room temperature (\approx 25 °C) and subjected to hydro-distillation, using a Clevenger-type apparatus for 3 h until total recovery of oil. The EOs obtained were dried over anhydrous sodium sulfate and stored at 4 °C in the dark.

5.2. Gas Chromatography/Mass Spectrometry (GC/MS) Analyses

The seaweed essential oils were analyzed by GC-MS following a protocol previously described by Falcão et al. (2018) [70]. Analyses were performed in a GC-2010 Plus (Shimadzu, Kioto, Japan) gas chromatography system equipped with a AOC-20iPlus (Shimadzu, Kioto, Japan) automatic injector, a SH-RXi-5ms column (30 m × 0.25 mm × 0.25 μ m; Shimadzu, Kioto, Japan), and a mass spectrometry detector, operated using an injector temperature of 260 °C and the following oven temperature profiles: an isothermal hold at 40 °C for 4 min, an increase of 3 °C/min to 175 °C, followed by an increase of 15 °C/min to 300 °C and an isothermal hold for 10 min. The transfer line temperature was set at 280 °C and the ion source at 220 °C; the carrier gas, helium, was adjusted to a linear velocity of 30 cm/s; the ionization energy was 70 eV, the scan range was set at 35–500 u, with a scan time of 0.3 s. A quantity of 1 μ L of each sample diluted in n-hexanewas injected using the split injection mode at 1:10. The identification of the essential oil components was carried out by comparison of the obtained spectra with those from the NIST17 mass spectral library and by determining the linear retention index (LRI) based on the retention times of an n-alkanes mixture (C8–C40, Supelco, Darmstadt, Germany).

published data. Compounds were quantified as relative percentage of total volatiles using relative peak area values obtained from total ion current (TIC).

5.3. Screening for Anti-Cyanobacterial Activity

5.3.1. Cyanobacteria Strain

In this study, the cyanobacteria strain *M. aeruginosa* was sampled from the eutrophic reservoir Lalla Takerkoust (31°21′36″ N; 8°7′48″ W), Morocco, in bloom period (October 2017) and then the strain was isolated, separated into single cells, and maintained in culture in BG11 medium under a controlled culture chamber endowed with the following conditions: temperature of 26 ± 2 °C, light intensity of 63 µmol m⁻² s⁻¹, and a light/dark cycle of 15 h/9 h [30].

5.3.2. Disc Diffusion Method

In vitro anti-cyanobacterial activity of seaweed EOs of each of the three algae was evaluated using the agar diffusion method [71]. The suspension of tested *M. aeruginosa*, containing about 10^8 cells/mL using a Malassez counting cell, was spread on BG11 medium with 4% of agarose. Subsequently, 10 µL of each EO and CuSO₄, prepared at a concentration of 50 µg mL⁻¹ in ultrapure water, as positive control was dropped on sterile filter paper discs, 6 mm in diameter (Whatman no. 1, Little Chalfont, UK) and placed on the agar surface. Before incubation in the culture chamber under the described condition, all treated plates were stored in a refrigerator at 4 °C for more than 4 h to prevent the cyanobacteria growth and allow the diffusion of the bioactive substances contained in the EOs into the medium. Each experiment was repeated six times to statistically confirm the results.

5.3.3. Determination of the Minimum Inhibitory Concentration (MIC) and Minimum Bactericidal Concentration (MBC)

The determination of the MIC values of the EOs that showed activity in the disc diffusion assay, was carried out in a 96-well microplate using the microdilution assay according to the NCCLS guidelines M7-A4 [72]. The MIC values represent the lowest EO concentration that prevents the cyanobacteria growth. Succinctly, 200 μ L of tested cyanobacteria culture with density of 3 × 10⁶ cells/mL (exponential growth phase) was added to each microplate well. The EOs were dissolved in DMSO (1%) and added to the tested culture to obtain final concentrations from 4000 to 1.953 μ g mL⁻¹. Subsequently, the prepared microplates were incubated for 5 days under the described controlled conditions in the culture chamber. In order to determine the MBC values, which represent the lowest EOs concentration that induces 100% cell death of incubated cyanobacteria, 100 μ L of each wells without visible cyanobacteria growth was spread on BG11 and incubated for 5 days in the culture chamber.

5.4. Determination of Cyanobacteria Growth Rates

The effects of the most bioactive EO on *M. aeruginosa* strain, namely the EO of *C. tamariscifolia*, were accessed by measuring the inhibition and growth rates estimation. The growth test was conducted, in triplicate, under the determined MIC and MBC concentration of the most bioactive EO and CuSO₄ (positive control) in Erlenmeyer flasks (150 mL) containing 9 mL of cyanobacteria inoculum and 71 mL of BG11 medium. The initial density of the tested cyanobacteria culture was adjusted by addition of BG11 medium and counting cells until a value of 2×10^6 cells/mL (the exponential growth phase). DMSO was employed as negative control. Whereas, another untreated cyanobacteria culture was used for the performance of all calculations necessary for the results treatment. The inhibition (IR) and growth rates were estimated by cells counting using a hemocytometer under a microscope every 24 h [73] and calculated using the following Equations (1) and (2), respectively:

where, Nc and Nt represent the cell concentrations (cells/mL) in the control and treatment samples, respectively [74].

$$\mu = (\ln Ne - \ln Nb)/\Delta t \tag{2}$$

In which μ is the average growth rate; Ne and Nb (cells/mL) are the cell densities on the last day and the first day of the experiment, respectively, and Δ t denotes the duration of the experiment.

5.5. Biochemical Parameters in M. aeruginosa

5.5.1. Determination of Chlorophyll-a and Total Protein Contents

Chlorophyll-*a* (Chl-*a*) concentration was measured in triplicate and calculated following the method previously described by Lichtenthaler and Wellburn (1983) [75]. Shortly, 5 mL of the culture sample was centrifuged at 4000× *g* for 15 min to collect algal cells. Cells were then re-suspended with boiling ethanol (95%). The three replicas were incubated at 4 °C for 48 h. Subsequently, another centrifugation for 5 min at 3400× *g* was performed to eliminate the pellet. The supernatant optical density (OD) was read at different wavelengths absorbance (649 and 665 nm). Chlorophyll-*a* concentration was calculated by the following Equation (3).

$$[Chl-a] = 13.95 \times DO665 - 6.88 \times DO649 \tag{3}$$

The enzyme extracts were prepared according to Li et al. (2016) protocol [76]. Briefly, *M. aeruginosa* cells were collected by centrifugation of each culture (5 mL) at 4000× *g* for 25 min. The pellet was re-suspended in 0.1 M phosphate buffer (pH 6.5) containing 1% (*w*/*v*) polyvinylpyrrolidone (PVP). Then the cells were disrupted and homogenized by an ultrasonic cell pulverizer for 5 min in an ice bath. The homogenate was then centrifuged 10,000× *g* at 4 °C for 10 min. The supernatant was used for total protein measurement and antioxidant enzyme activity assays. The total protein content was determined by the application of Bradford (1976) method [77]. Briefly, 100 µL of the enzyme extract was added to 2 mL of Bradford's reagent and incubated at room temperature in obscurity for 20 min. Furthermore, a mixture of the assay buffer (100 µL) and the Bradford's reagent (2 mL) was used as a blank. The absorbance was read at 595 nm and the protein content was calculated from a calibration curve of Bovine Serum Albumin (BSA).

5.5.2. Activity of Antioxidant Response Enzymes, CAT and SOD

The SOD activity was assayed in triplicate according to Beauchamp and Fridovich (1971) method [78]. The reaction mixture contained 0.8 mL PBS solution (50 mM, pH 7.8), 0.3 mL methionine solution (130 mM), 0.3 mL Na₂EDTA solution (100 µM), 0.3 mL riboflavin solution (20 µM), 0.3 mL nitroblue tetrazolium (NBT) solution (750 μ M), and 1 mL enzyme extract for a total volume of 3 mL. As SOD has the ability to inhibit the photochemical reduction of NBT, this assay utilized negative controls (silver paper wrapped around the test tube to mimic fully dark condition without any photochemical reduction of NBT), positive controls (deficiency of SOD activity in light with full photochemical reduction of NBT), and treatment groups (in light with SOD inhibition on photochemical reduction of NBT). The absorbencies of all experimental tubes were measured at 560 nm after a 20 min irradiance of 40–60 mmol photons $m^{-2} s^{-1}$. One unit of SOD activity was defined as the amount of enzyme that inhibited 50% of photochemical reduction of NBT. CAT activity was assayed in triplicate by absorbance decrease being proportional to the breakdown rate of hydrogen peroxide (H_2O_2) at 240 nm according to the method of Rao et al. (1996) [79]. The reaction mixture contained 1 mL H_2O_2 , 1.9 mL H₂O, and 1 mL crude enzyme. Samples were incubated for 2 min at 37 °C and the absorbance of the sample was monitored for 5 min at 240 nm using a Varian Cary® 50 UV-Vis Spectrophotometer (Agilent Technologies, Santa Clara, CA, USA).

5.5.3. Determination of MDA Content

The lipid peroxidation level was reflected by changes of malondialdehyde (MDA) content, which was determined in triplicate, according to Du et al. (2017) [62]. Samples were collected every 2 days and centrifuged at $4000 \times g$ for 20 min. The cell pellets were homogenized with 2 mL of 10% (w/v) trichloroacetic acid (TCA) and centrifuged at $12,000 \times g$ for 10 min at 4 °C. After centrifugation, 2 mL of the supernatant was mixed with 2 mL of 0.6% thiobarbituric acid (in 10% TCA) and heated in boiling water for 15 min. The reaction was stopped by transferring the reaction tubes into an ice bath. Following cooling, the samples were then centrifuged at $12,000 \times g$ for 10 min. The absorbance of the supernatant was measured at 532, 600, and 450 nm, taking a mixture of 2 mL ultrapure water and 2 mL 0.6% TBA as reference. The MDA level (µmol/L) was calculated according to Equation (4):

$$MDA = 6.45 \times OD532 - OD600 - 0.56 \times OD450$$
(4)

5.6. Statistical Analysis

The experiments were done in six replicates in solid medium (n = 6) and three replicates in liquid medium (n = 3) with each independent assay. Statistical analysis between experimental groups and the control were performed by applying a one-way and two-way ANOVA analysis. Post hoc differences between group means was carried out with the Tukey test using Sigma Plot software (sigmaplot 12.5 for windows; Systat Software Inc., San Jose, CA, USA) for Windows. Values of p < 0.001 were considered statistically significant.

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