

# Higher-Order Structural Analysis of a Transparent and Flexible High Thermal Conductive Liquid Crystalline Elastomer Sheet and Its Composite

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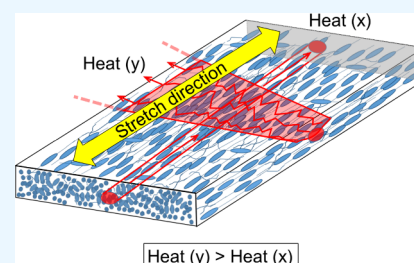
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**ABSTRACT:** Transparency, flexibility, and high thermal conductivity are trade-offs. Specifically, we have investigated a cross-linked acrylic liquid crystal elastomer (LCE) that exhibits both transparency and flexibility while maintaining a high level of thermal conductivity. The transparent monodomain LCE sheet was achieved through a process of stretching an initially opaque polydomain sheet to 80% elongation and subsequently subjecting it to photocuring. The thermal conductivity in the stretching direction ( $x$ ) of the monodomain LCE sheet was found to be 1.8 times higher than that of the prestretched polydomain sheet, consistent with findings from previous studies. However, in the orthogonal direction ( $y$ ) to the stretching ( $x$ ) direction, the thermal conductivity exhibited an even higher value, being 1.7 times greater than in the  $x$ -direction, with a value of 3.0 W/(m·K). This unique observation prompted us to conduct further investigation through higher-order structural analysis of these LCE sheets using 2D wide-angle X-ray scattering (WAXS) analysis. In the transparent sheet, the LCE molecules were aligned in the sheet in the stretching  $x$ -direction (monodomain structure) for the out-of-plane direction. However, in the in-plane  $x$ -direction, the molecular plane spacing exhibited random orientation at a period of 0.45 nm. In contrast, within the  $y$ -direction of the inner layer, the molecular plane spacing exhibited a uniaxial horizontal orientation at the same period length as in the  $x$ -direction. The heat energy entering into the  $y$ -direction once spreads to the  $x$ -direction, but it was considered that the reason for the higher thermal conductivity to the  $y$ -direction would be forming covalent bonds that function as new heat transmission paths, in the direction intersecting to the  $x$ -direction during photocuring. Therefore, we concluded that the synergistic effect of the high level of the ordered inner structure and covalent bonding structure due to cross-linking in the  $y$ -direction contributes to its higher thermal conductivity compared to that in the  $x$ -direction, which exhibits a random in-plane structure. Additionally, we have fabricated an LCE composite sheet filled with 75 vol % of alumina particles using a polydomain-type LCE as the base material. The composite sheet exhibits remarkable thermal conductivity in the thickness direction, measuring at 9.8 W/(m·K), while maintaining a flexibility characterized by an elastic modulus of 70 MPa. This thermal conductivity surpasses that of a nonmesogenic acrylic composite sheet with identical alumina particle filling, which measured at 3.9 W/(m·K), more than twice as much. The presence of the mesogen skeleton has been demonstrated to enhance heat transfer, even within soft composites, by facilitating the formation of an ordered structure.



## 1. INTRODUCTION

As electronic devices become more sophisticated and compact, the heat generation density in circuits is increasing. Consequently, there is an increasing demand for enhanced heat dissipation, especially for electric insulating materials.<sup>1–4</sup> Although polymers are excellent electric insulating materials, they are basically adiabatic and do not conduct heat effectively. This is because, unlike metals, which have free electrons that facilitate heat conduction, in insulating polymers, phonons dominate thermal conduction.<sup>5</sup> Polymers are broadly classified into thermoplastic and thermosetting polymers, and it is known that the thermal conductivity of thermoplastic polymers increases when they are stretched.<sup>6–14</sup> Recently, it has been reported that the thermal conductivity of polyethylene fiber, a thermoplastic polymer, was increased to 104 W/(m·K), which is equivalent to that of metal, by stretching it to the utmost

limit.<sup>15,16</sup> In contrast, thermosetting polymers that form a three-dimensional network structure, such as epoxy resin, which is an essential material for electronic circuits, cannot be stretched or otherwise oriented like thermoplastic polymers, making it difficult to significantly improve their thermal conductivity. Therefore, to reduce phonon scattering and achieve high thermal conductivity, it is effective to form a highly ordered higher-order structure inside the polymer by self-alignment of the mesogen backbone,<sup>4,17–38</sup> and then to

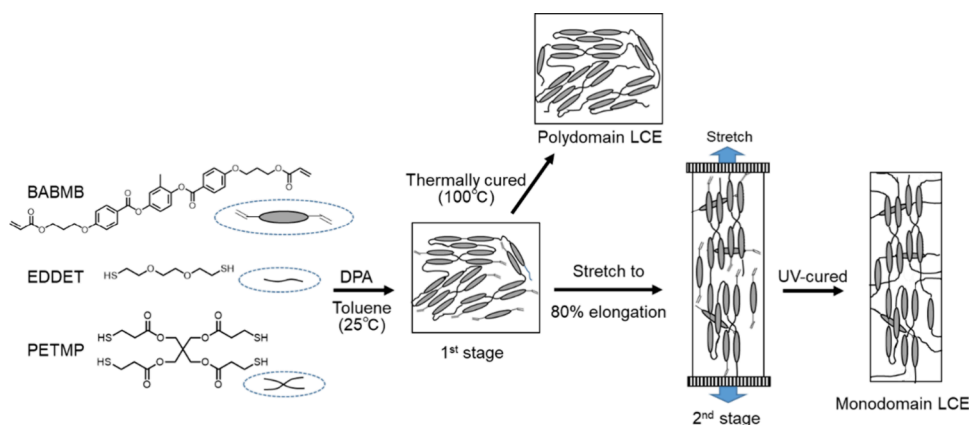
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**Figure 1.** Schematic fabrication process of two types of LCE polydomain and monodomain sheets.

design molecules that increase the cross-linking density by selecting a curing agent.<sup>2</sup> Recently, a research study utilized a mesogen epoxy resin film alone, without the addition of a thermally conductive ceramic filler, and achieved a thermal conductivity of 10 W/(m·K).<sup>39</sup> Although thermoplastic polymers exhibit a significant decrease in the thermal conductivity above  $T_g$ , thermosetting polymers experience a minor decrease in thermal conductivity. This is attributed to the preserved cross-linked orientation order in thermosetting polymers even when above  $T_g$ .<sup>1,2</sup>

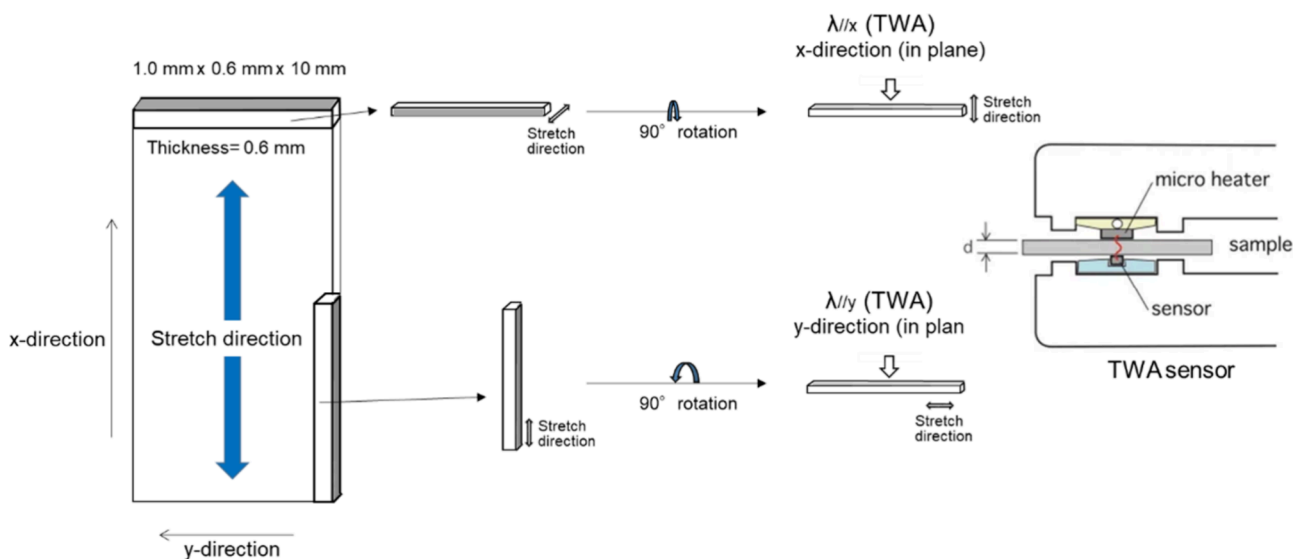
However, transparency, flexibility, and high thermal conductivity are in a trade-off relationship within polymers. Thus far, a polymer that successfully embodies all three characteristics simultaneously has not yet been developed. Several papers had reported transparent polymer films with high thermal conductivity, but not enough transparency properties.<sup>40,41</sup> The introduction of a mesogen skeleton is effective in increasing thermal conductivity, but its high orientation and birefringence anisotropy result in a rigid and opaque structure.<sup>2,18,35,36,39</sup> Recently, a new liquid crystalline acrylic elastomer (LCE) sheet with transparency, flexibility, and impact strength resistance has been published in papers.<sup>42–44</sup> In this paper, monodomain and polydomain liquid crystalline elastomers (LCEs), as well as LCE composites, are fabricated, and their thermal conductivities are measured. The study aims to investigate the relationship between higher-order structural analysis and thermal conductivity by utilizing XRD. The fabrication process of the monodomain LCE involves two stages. First, the LCE polydomain is achieved by employing Michael addition of polyether thiols and multifunctional thiols to an acrylic monomer containing mesogens (1st stage). Subsequently, the material undergoes stretching and photo curing to transition into the monodomain state (2nd stage). Additionally, a flexible composite sheet of polydomain LCE with a high alumina filler content was fabricated without the stretching process. Subsequently, its thermal conductivity was evaluated.

## 2. EXPERIMENTAL SECTION

**2.1. Preparation of LCE Polydomain, Monodomain, and Composite Sheets.** LCE polydomain and monodomain sheets were fabricated in accordance with methods previously published,<sup>42,43</sup> and the schematic fabrication process is shown in Figure 1. For the LCE synthesis described here, 1,4-bis[4-(3-acryloyloxypropoxy)benzoyloxy]-2-methylbenzene (BABMB: M.W. 588 g/mol) with 99% purity was purchased

from Synthron Chemicals (Germany). Pentaerythritol tetrakis-(3-mercaptopropionate) (PETMP: M.W. 488 g/mol), 2,2-(ethylenedioxy)diethanethiol (EDDET: M.W. 182 g/mol), dipropylamine (DPA: M.W. 101 g/mol), and 2-hydroxy-4'-(2-hydroxyethoxy)-2-methylpropiophenone (HHMP: M.W. 224 g/mol) were purchased from Sigma-Aldrich. The LCE material is made using a chemical composition (BABMB/EDDET/PETMP = 7/6/1 (mol ratio)), with the addition of toluene to facilitate the Michael addition reaction. Once the solution returned to room temperature (25 °C), 1 mol % DPA was added for the first-stage polydomain LCE polymerization as depicted in Figure 1. The polydomain LCE sheets were made by coating the solution on a PET film with a PTFE spacer under vacuum, subsequently postcured at 100 °C. For the monodomain LCE sheets, 1 mol % HHMP photoinitiator was added to first-stage material; the mixture was coated on a PET film with a PTFE spacer. After vacuuming to remove the solvent at room temperature, the precured LCE sheet was peeled from the PET film. Next, mechanical stretching was applied to the LCE sheet until it reached 80% elongation, as depicted in Figure 1 (2nd stage). In this study, the stretching condition was standardized at 80% elongation to ensure stability and prevent breakage, resulting in final thicknesses of 300 and 600  $\mu\text{m}$ . After stretching, the width of the LCE sheet is reduced to approximately 70%, while the thickness is reduced to around 80%. Next, the LCE sheet is exposed to a 365 nm UV light source at an intensity of 10 mW/cm<sup>2</sup> for 10 min by holding a UV lamp approximately 150 mm from the sheet. The resulting cured sheet for evaluation measures 300 and 600  $\mu\text{m}$  in thickness.

The LCE alumina composite sheet was fabricated following the procedure outlined in Figure S1. The LCE alumina varnish is made using the above LCE chemical composition (BABMB/EDDET/PETMP = 7/6/1 (mol ratio)), with the inclusion of toluene to aid the Michael addition reaction. The mixture of alumina powder (Advanced Alumina AA series; AA18 (mean diameter: 20  $\mu\text{m}$ ): AA3 (3.5  $\mu\text{m}$ ): AA04 (0.47  $\mu\text{m}$ ) = 66:24:10 (weight ratio)) purchased from Sumitomo Chemical was added. Since the alumina particle size has a distribution and the bulk density varies, the amount of alumina prepared for compositing is controlled by the weight ratio. Given the density of alumina and LCE are 3.98 and 1.25 g/cm<sup>3</sup>, respectively, the weight ratio of LCE:alumina = 9.5:90.5 is equivalent to a volume ratio of LCE:alumina = 25:75. The LCE and alumina mixture varnish is mixed by a planetary centrifugal mixer (3 cycles for 5 min at 2000 rpm) at room



**Figure 2.** Processing method of LCE sheets used for measuring thermal diffusivities of each in-plane direction by the TWA method.

temperature, and toluene is added in minimal amounts necessary to facilitate the mixing of the alumina powder. Finally, 1 mol % DPA relative to the total chemicals amount (excluding alumina) was added, and the mixture was mixed by a planetary centrifugal mixer (1 cycle for 5 min at 2000 rpm). This compound varnish is injected into a mold using a PTFE spacer and release PET film as shown in Figure S1; subsequently, the compound is vacuum degassed at room temperature for 30 min and vacuum pressed for 30 min at 100 °C under 2 MPa to cure.

As a reference base resin, nonmesogenic multifunctional acrylic elastomer was used. For the reference base resin synthesis described here, *n*-butyl acrylate (BA: M.W. 128 g/mol) was purchased from Mitsubishi Chemical Co., trimethylpropane triacrylate (TMPA: M.W. 296 g/mol) was purchased from Daicel-Allnex Ltd., 2-hydroxyethyl acrylate (HEA: M.W. 116 g/mol) was purchased from Nippon Shokubai Co. Ltd., and *t*-butyl peroxy-2-ethylhexanoate (PBO: M.W. 216 g/mol) as a polymerization initiator was purchased from the NOF corporation. Using the above chemical composition (BA/TMPA/HEA = 12/3/1 (mol ratio)), and incorporating the same alumina powder mixture as the LCE composite, a solution is prepared. Finally, 0.3 mol % PBO relative to the total chemical amount is added and thoroughly mixed.

## 2.2. Thermal Conductivity Measurement Method.

The thermal conductivity ( $\lambda$ ) of the LCE films was calculated as

$$\lambda = \alpha \times \rho \times C_p \quad (1)$$

where  $\alpha$ ,  $\rho$ , and  $C_p$  express the thermal diffusivity, density, and specific heat capacity of the LCE films, respectively. The thermal diffusivities of the LCE films were measured by temperature wave analysis (TWA)<sup>45–47</sup> using an apparatus (ai-Phase Mobile M3type1) based on ISO 22007. The configuration of the TWA apparatus and its features and principles are described in Figure S2. The  $\rho$  and  $C_p$  values were measured by the electronic hydrometer MDS-3000 (Alfa Mirage Co., Ltd.) and DSC 250 (TA Instrument), respectively.

The thermal diffusivity  $\alpha$  of the LCE sheet was measured by cutting small pieces of the sheet against the thickness and in-

plane direction as illustrated in Figure 2. The LCE monodomain sheets were measured for two directions: the stretch direction ( $x$ ) and the orthogonal direction ( $y$ ).

**2.3. Higher-Order Structure Analysis Method.** The higher-order structure of the obtained sheets was measured by transmission wide-angle X-ray scattering (WAXS) using the Rigaku SmartLab X-ray diffractometer and by transmission 2D-WAXS using the HyPix-3000 detector, which can measure two-dimensional images. The radiation source was CuK $\alpha$  radiation ( $\lambda = 0.15418$  nm). It was used at 40 kV and 30 mA output. The diameter of the collimator was 0.1 mm. The camera length was 65 mm, and integrated measuring time was 120 min each. Using the Rigaku 2D Data Processing software "2DP", the  $2\theta$  diagram was described from the 2D data corrected for the scattering by the air atmosphere.

Polarized interference images of LCE polydomain and monodomain sheets were also qualitatively observed under crossed Nicols spectroscopy using a Nikon ECLIPSE LV100POL polarized optical microscope (POM).

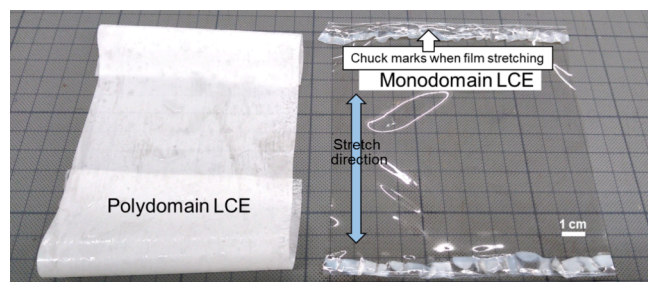
**2.4. Measurement of Elastic Modulus.** The elongation at break and tensile modulus of the cured material were measured at 25 °C using a tensile tester (Autograph EZ-TEST EZ-S manufactured by Shimadzu Corporation). In the measurements, cured materials of dimensions 0.5 mm (thickness)  $\times$  5 mm (width)  $\times$  30 mm (length) were used. These samples were measured under the conditions of a 20 mm distance between chucks and a tensile speed of 5 mm/min. The elastic modulus is determined by the following equation based on the values of strain specified at two points.

$$E = (\sigma_2 - \sigma_1) / (\varepsilon_2 - \varepsilon_1) \quad (2)$$

where  $E$  expresses the elastic modulus (MPa),  $\sigma_1$  expresses the stress (MPa) at strain  $\varepsilon_1 = 0.0005$  (0.05%), and  $\sigma_2$  expresses the stress (MPa) at strain  $\varepsilon_2 = 0.0025$  (0.25%).

## 3. RESULTS AND DISCUSSIONS

**3.1. LCE Polydomain and Monodomain Sheets.** Figure 3 displays the appearance of the fabricated polydomain and monodomain LCE sheets, each with a thickness of 300  $\mu$ m. The monodomain sheet exhibits transparency in the visible region. The transmittance spectrum of each LCE sheet was



**Figure 3.** Appearance of the opaque polydomain and transparent monodomain LCE sheets (thickness = 300  $\mu\text{m}$ ).

measured using a Hitachi UV–vis–NIR Spectrophotometer (U-4000), as depicted in Figure S2.

First, the planar orientation of the LCE polydomain and monodomain sheet birefringence was qualitatively confirmed by rotating the stage under a crossed Nicols using a polarizing microscope (POM). The results are shown in Figure 4. For polydomain sheets, the brightness of the interference image did not change when the stage was rotated, whereas for monodomain sheets, the bright and dark fields repeated every  $45^\circ$  when the stage was rotated, indicating that the sheets were uniaxially oriented in the stretching direction within the plane.

Measurements of the in-plane thermal conductivity of LCE sheets were performed on 600  $\mu\text{m}$  thick sheets. This choice was made due to the sensor size of the TWA apparatus, which is  $0.25 \times 0.5$  mm (as illustrated in Figure S1). Table 1 shows the measurement results of thermal conductivities of each LCE sheets. The orientation of the thermal conductivity measurements is denoted by  $\parallel$  for in-plane and  $\perp$  for the out-of-plane (thickness) direction. The thermal conductivity of 0.35 W/(m·K) in the out-of-plane (thickness) direction for the polydomain LCE sheet is approximately three times higher than that of the monodomain LCE or reference acrylic elastomer sheet. It is hypothesized that the increase in the thermal conductivity of the polydomain LCE is attributed to the formation of an ordered structure facilitated by the presence of mesogens. In contrast, for the in-plane direction, both sheets exhibited thermal conductivities more than three times higher than those observed in the thickness direction. The reason for this phenomenon is assumed that the in-plane orientation would be formed by the effect of the hydrophobic

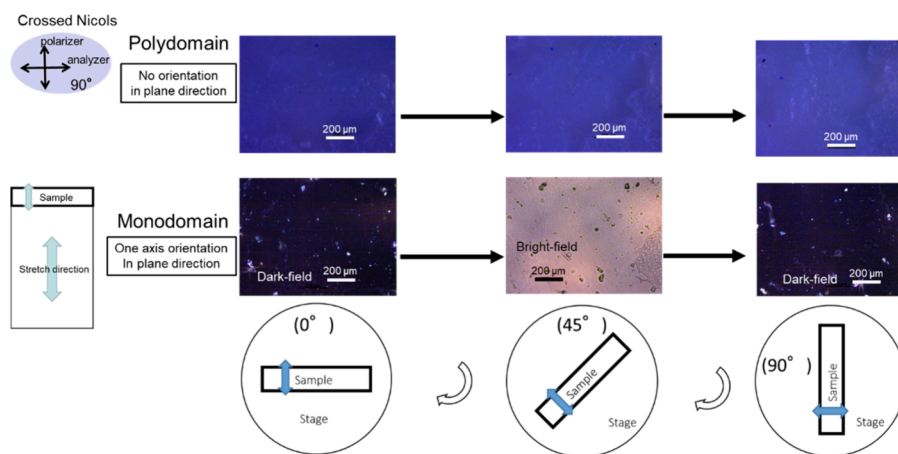
surface of the PET film and the surface tension on the surface in contact with air. This is the typical characteristic of thermal conductivity anisotropy of polymer films; the longer molecular chains lie down randomly in the in-plane direction, as a result, facilitating heat transfer along the direction of the molecular chains rather than between molecules.<sup>4,14,38,39,48–50</sup>

In particular, the in-plane directional thermal conductivities of LCEs were over 1 W/(m·K), surpassing the in-plane thermal conductivity of polyimide (0.7–2.1 W/(m·K)).<sup>4,38,50</sup> The in-plane thermal conductivity of the monodomain was 1.8 W/(m·K) in the  $x$ -direction and 3.0 W/(m·K) in the  $y$ -direction, demonstrating very high values not seen in transparent and flexible polymers. In general, the thermal conductivity of polymer films increases in the direction of stretching and decreases tremendously in the thickness direction, as has been reported for PE and acrylic resins in many cases.<sup>7–17</sup> For this LCE, the thermal conductivity increased in the stretch direction as above and was even 1.7 times higher in the in-plane  $y$ -direction, which is orthogonal to the stretch direction. This unique phenomenon prompted the study of higher-order structural analysis for these LCE sheets by using X-ray diffraction (XRD) analysis.

Figure 5 shows the results of transmission 2D-WAXS measurements of LCE polydomain and monodomain sheets to the out-of-plane (thickness) direction. In the polydomain sheet (a), the intermolecular periodic structure (about 0.45 nm) around  $2\theta = 20^\circ$  is isotropic and circularly scattered; although in the monodomain sheet (b), the intermolecular periodic structure around  $2\theta = 20^\circ$  is strongly scattered in parallel to the LCE sheet stretching direction.

Focusing on the small angle region, a  $2\theta$  peak was observed at  $2.7^\circ$  ( $d = 3.25$  nm) of polydomain sheet, but no peak existed for the monodomain sheet. Here, the molecular length of BABMB calculated from the linear distance between atoms in the MMFF94s force field using the structure optimization program Conflex is shown in Figure 6. The  $2\theta = 2.7^\circ$  ( $d = 3.25$  nm) value in Figure 5 is close to the molecular length of the BABMB.

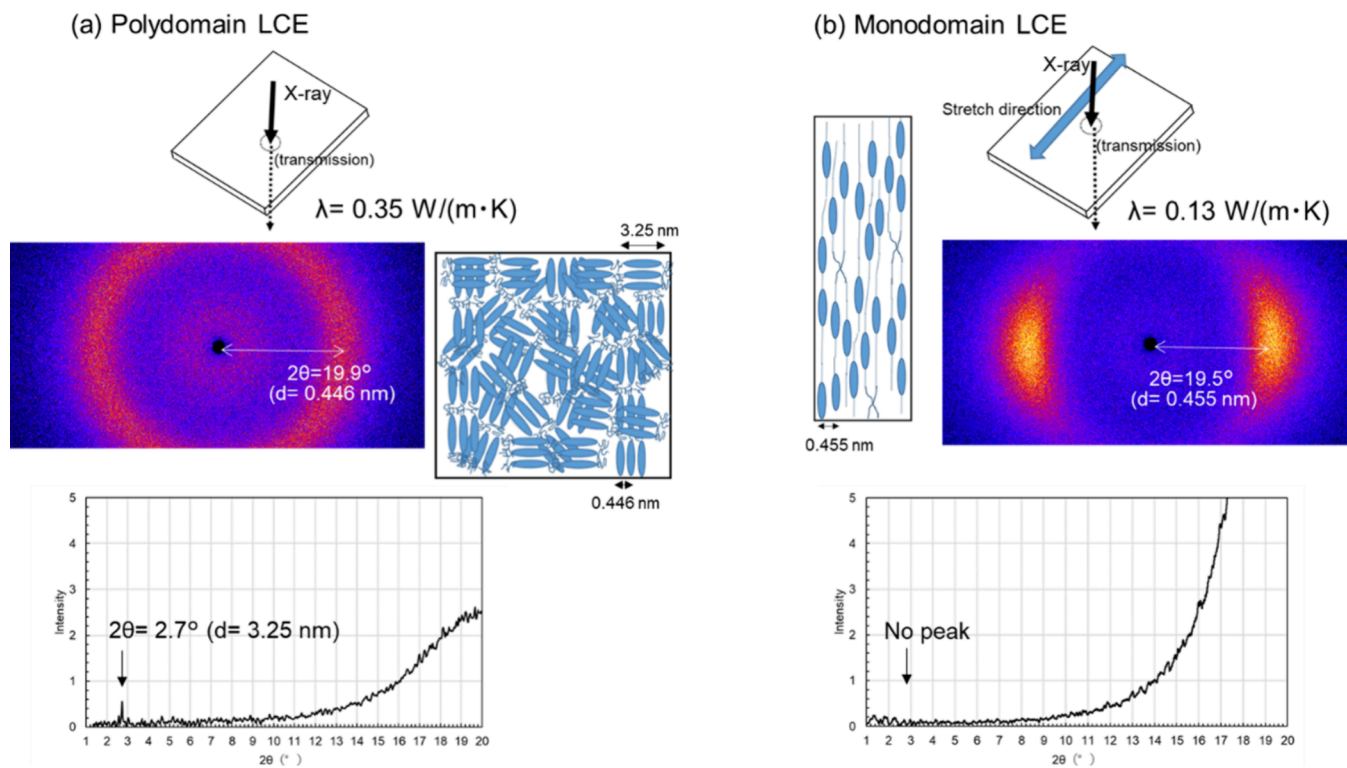
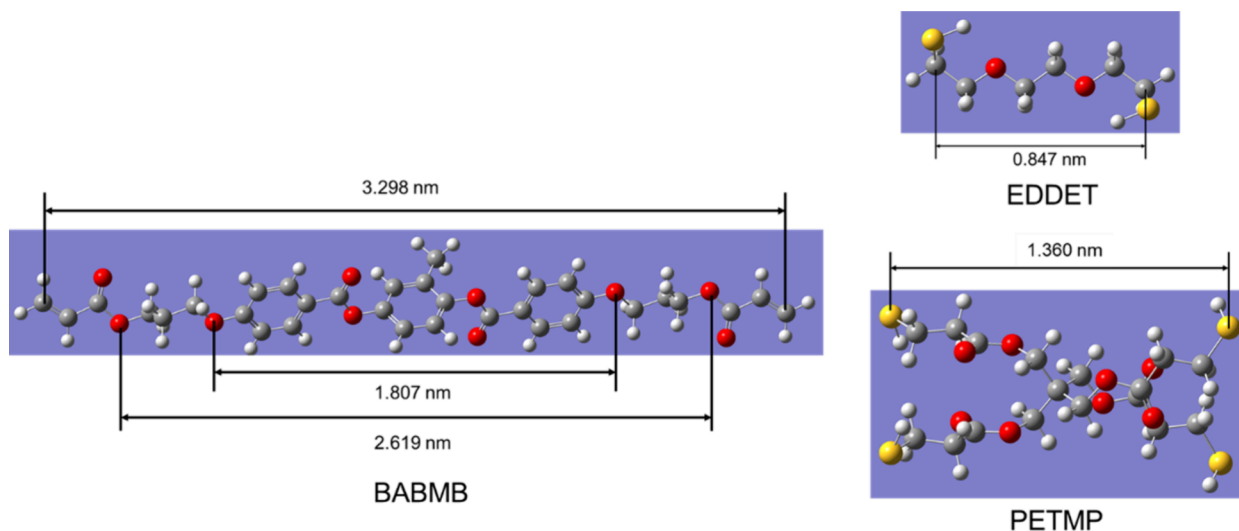
Although the intensity of this peak is weak, it could be attributed to the periodic structure of the BABMB molecule. As shown in the schematic diagram in Figure 5, it is hypothesized that several mesogens would be periodically aligned in all directions, and the alkyl chains (EDDET, PETMP) are connected to BABMB in a random structure.

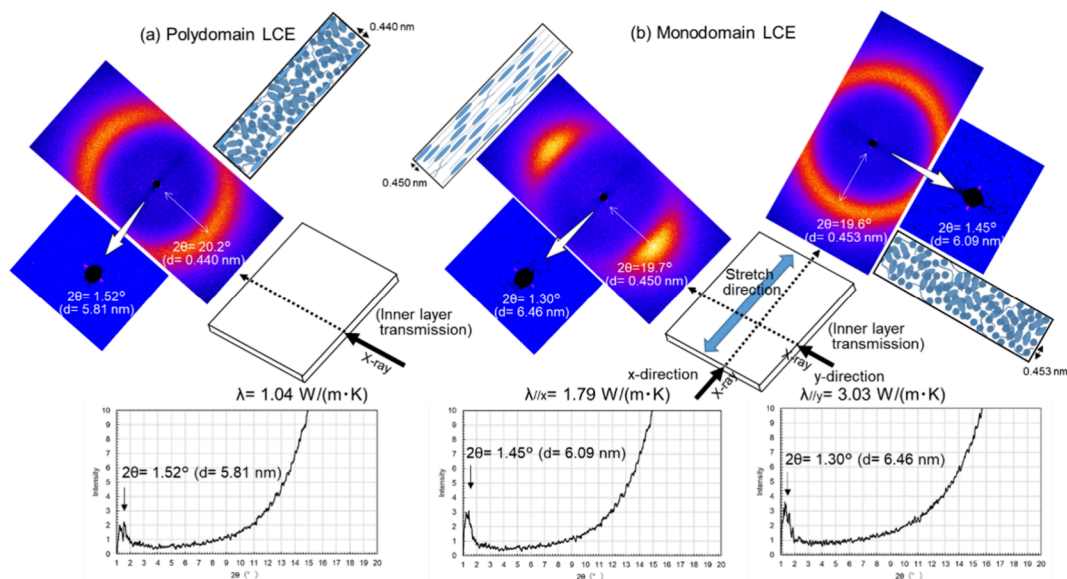


**Figure 4.** Observation of the orientation state of monodomain and polydomain sheets using a polarizing optical microscope (POM).

**Table 1. Thermal Conductivities and Elastic Modulus for LCE Monodomain, Polydomain, and Reference Nonmesogenic Multifunctional Acrylic Elastomer Sheets (Thickness = 600  $\mu\text{m}$ )**

material	monodomain LCE		polydomain LCE		reference	
	$\parallel$ (in plane)	$\perp$ (out of plane)	$\parallel$ (in plane)	$\perp$ (out of plane)	$\parallel$ (in plane)	$\perp$ (out of plane)
thermal conductivity [W/(m·K)]	(1) $x$ -direction: $1.8 \pm 0.15$ (2) $y$ -direction: $3.0 \pm 0.17$	$0.13 \pm 0.02$	$1.0 \pm 0.13$	$0.35 \pm 0.04$	$0.30 \pm 0.03$	$0.10 \pm 0.02$
density [g/cm <sup>3</sup> ]	$1.25 \pm 0.01$		$1.25 \pm 0.01$		$1.06 \pm 0.01$	
specific heat capacity [J/(g·K)]	$1.20 \pm 0.02$		$1.20 \pm 0.02$		$1.93 \pm 0.02$	
elastic modulus [MPa]	$0.15 \pm 0.02$		$0.11 \pm 0.01$		$0.05 \pm 0.02$	

**Figure 5.** Higher-order structural analysis for (a) polydomain LCE and (b) monodomain LCE sheets, using transmission 2D-WAXS (out-of-plane direction) (thickness = 600  $\mu\text{m}$ ).**Figure 6.** Molecular lengths of BABMB, EDDT, and PETMP calculated from linear distances between atoms in the MMFF94s force field using the “Conflex” program.

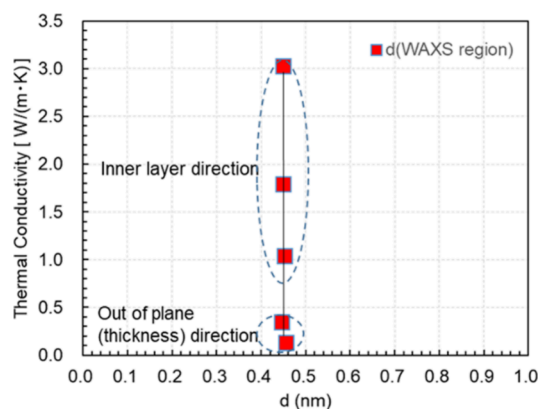


**Figure 7.** Higher-order structural analysis for (a) polydomain LCE and (b) monodomain LCE sheets, using transmission 2D-WAXS (in-plane direction) (thickness = 600  $\mu\text{m}$ ).

Next, the results of 2D-WAXS transmission measurements on the inner layer of LCE sheets were conducted and shown in Figure 7 and Figure S4. No directional differences were observed in the polydomain sheets in Figure S4; the circular diffraction scatter image, which exhibits almost isotropic characteristics but with a slight orientation toward the in-plane direction, was observed in the 2D-WAXS. However, significant differences in higher-order structure were observed between the  $x$ - and  $y$ -directions in the monodomain LCE sheets, as depicted in Figure 7b. Specifically, the intermolecular periodic structure (about 0.45 nm) around  $2\theta = 20^\circ$  in the inner layer of the orthogonal  $y$ -direction to the stretching direction was strongly in-plane oriented in parallel to the stretching direction of the sheet. The characteristic four-point pattern to cybotactic nematics with stronger ordering<sup>51–53</sup> was not observed. In contrast, for the  $x$ -direction, a circular diffraction scattering image was observed in the 2D-WAXS image, suggesting that the intermolecular periodic structure is isotropic. No peak was observed at  $2\theta = 2.7^\circ$  ( $d = 3.25$  nm) for all LCE sheets, indicating no periodic structure of the BABMB molecule would exist in the inner layer.

Focusing on the small angle region, a weak but sharp scattering were observed in the vicinity of  $2\theta$  from  $1.3^\circ$  to  $1.5^\circ$  for all the sheets. Although the  $d$ -spacing is around 6 nm, which is nearly equivalent to two molecules of BABMB, the scattering direction is in the thickness direction of the LCE sheet. Additionally, the intensity is very weak compared to that of around  $2\theta = 20^\circ$ ; therefore, it is considered that there is little contribution to the thermal conductivity properties.

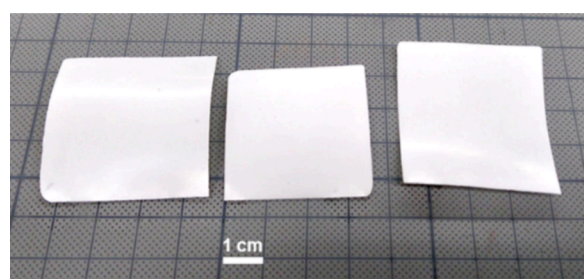
As shown in Figure 8, the slight difference in the periodic structure between the intermolecular planes at the wide-angle region (around  $2\theta = 20^\circ$ ) had little effect on the thermal conductivity. The width and thickness of the LCE sheet after stretching are macroscopically reduced to around 70 and 80%, respectively; however, the  $d$ -spacing is not changed nanoscopically. Therefore, it is considered that the reason for the higher thermal conductivity in the  $y$ -direction is due to the formation of covalent bonds that act as new heat transfer pathways in the direction intersecting with the  $x$ -direction during photo curing.



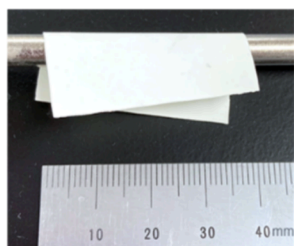
**Figure 8.** Relation between  $d$ -spacing values obtained from XRD analysis and their thermal conductivities ( $\lambda$ ) measured by the TWA method (thickness = 600  $\mu\text{m}$ ).

Consequently, we concluded that the heat energy entering into the  $y$ -direction once spreads to the  $x$ -direction, but the synergistic effect of the high level of ordered inner structure and the covalent bonding structure due to cross-linking in the  $y$ -direction contributes to its higher thermal conductivity compared to that in the  $x$ -direction, which exhibits a random in-plane structure.

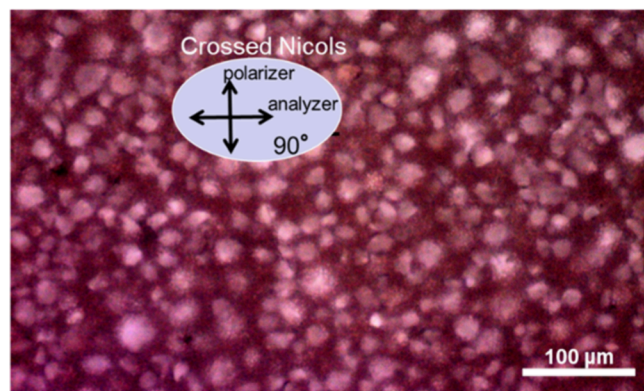
**3.2. LCE Alumina Composite Sheet.** Polydomain LCE, while soft and opaque, possesses high thermal conductivity in the thickness direction, so it might be applied to thermal interface materials (TIM) if it is made into a composite. Therefore, we aimed to fabricate the LCE alumina composite sheet for this purpose. Figure 9 illustrates (a) appearance of the obtained cured LCE alumina composite sheet with an alumina filler content of 75 vol % and a thickness of 300  $\mu\text{m}$ , (b) a photo of the composite sheet wrapped around and adhered to a 6 mm diameter aluminum pipe demonstrating flexibility, and (c) confirmation of alumina filler dispersibility using POM (under crossed Nicols). As depicted in Figure 9b, the cured LCE alumina composite sheets have flexibility even with a 75 vol % alumina filler. The elastic modulus is approximately 70



(a) Appearance of alumina composite sheets



(b) Appearance of wrapped and adhered to a 6 mm-diameter aluminum pipe



(c) POM image of alumina composite sheet under crossed Nicols (thickness= 300 μm)

**Figure 9.** Photographs of each: (a) appearance of the alumina composite sheet using polydomain LCE, (b) appearance of wrapped and adhered to a 6 mm diameter aluminum pipe, and (c) a POM image of the alumina composite sheet under crossed Nicols (thickness= 300 μm).

**Table 2. Thermal Conductivities and Elastic Modulus for LCE Alumina Composite Sheets and Reference Nonmesogenic Multifunctional Acrylic Elastomer Composite Sheets (Thickness= 300 μm)**

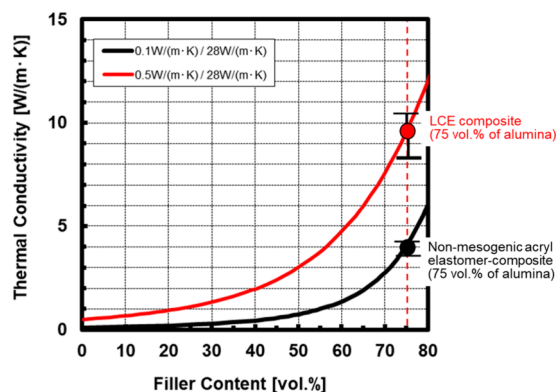
composite material	75 vol % of alumina of LCE		75 vol % of alumina of Reference	
	(in plane)	⊥ (out of plane)	(in plane)	⊥ (out of plane)
thermal conductivity of composite [W/(m·K)]	16.5 ± 2.5	9.8 ± 0.6	8.5 ± 0.5	3.9 ± 0.2
density [g/cm <sup>3</sup> ]	3.24 ± 0.05		3.20 ± 0.05	
specific heat capacity [J/(g·K)]	0.87 ± 0.03		0.88 ± 0.02	
elastic modulus [MPa]	70 ± 1		30 ± 1	

MPa, as shown in Table 2. The POM observation result indicates that different particle sizes of the alumina particles were evenly dispersed throughout the composite LCE material. Achieving good dispersibility is crucial as it greatly influences both thermal conductivity and insulation properties.<sup>2</sup> Table 2 shows the measurement results of thermal conductivities and the elastic modulus, where the orientation of the thermal conductivity measurements is indicated by || for in-plane and ⊥ for out-of-plane (thickness) directions. For comparison, the thermal conductivity of the acrylic elastomer composite without mesogen is also provided in Table 2.

The LCE alumina composite sheet exhibits an average value of 9.8 W/(m·K) in the thickness direction. This value is over twice as high as that of the acrylic elastomer composite sheet without mesogen. Using Bruggeman's equation,<sup>2,54–58</sup> the thermal conductivity of the base resin was estimated to be 0.5 W/(m·K) from Figure 10. The apparent thermal conductivity of the LCE in the composite (0.50 W/(m·K)) being larger than that of the resin-only sheet (0.35 W/(m·K) in Table 1) is assumed to be due to the interfacial interaction and molecular orientation of the LCE molecules at the alumina interface in the composite.

#### 4. CONCLUSIONS

In this paper, higher ordered structures were analyzed for the cross-linking acrylic liquid elastomer (LCE), which exhibits transparency and flexibility while possessing high thermal conductivity in the in-plane direction. This transparent



**Figure 10.** Composite thermal conductivity prediction and validation based on Bruggeman's equation.

monodomain LCE sheet was obtained by photocuring while stretching the polydomain LCE to 80% elongation. The thermal conductivity of the stretching direction ( $x$ ) of this monodomain LCE sheet was 1.8 times higher, measuring 1.79 W/(m·K), compared to that of the prestretched polydomain sheet. Furthermore, regarding the thermal conductivity of the orthogonal direction ( $y$ ) to the  $x$ -direction, which was measured at 3.0 W/(m·K), it was around 1.7 times higher than that of the  $x$ -direction.

From the XRD analysis, for the out-of-plane direction, the molecules in the monodomain LCE sheet were aligned in the

stretching  $x$ -direction. In contrast, for the in-plane direction, the molecular plane spacing was randomly oriented to the  $x$ -direction (i.e., heat transfer direction). However, in contrast, the molecular plane spacing was horizontally uniaxially oriented, exhibiting a highly ordered structure in the  $y$ -direction. Therefore, we conclude that the heat energy entering into the  $y$ -direction once spreads to the  $x$ -direction due to the higher ordered structure; subsequently, the synergistic effect of the covalent bonding structure due to photo-cross-linking in the  $y$ -direction contributes to its higher thermal conductivity compared to the  $x$ -direction.

Next, using the polydomain LCE, composite sheets filled with 75 vol % of alumina particles were fabricated and characterized. The alumina composite sheet exhibited high thermal conductivity along the thickness direction, averaging 9.8 W/(m·K), coupled with flexibility characterized by an elastic modulus of around 70 MPa. The thermal conductivity was over twice the value of 3.9 W/(m·K), which is measured in a nonmesogenic acrylic composite sheet with the same alumina particle filling. The apparent thermal conductivity of the base LCE resin in the composite was estimated to be 0.5 W/(m·K) using Bruggeman's composite's thermal conductivity prediction equation. These flexible composite sheets are expected to be applicable as thermal interface materials (TIM) in semiconductor applications and other related fields.

## ■ ASSOCIATED CONTENT

### SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.3c09550>.

Preparation of LCE composite sheets, thermal diffusivity measurement system (ai-Phase mobile M3 type1), transmittance properties for two types of LCE sheets, higher-order structural analysis for polydomain and monodomain sheets using transmission 2D-WAXS, effect of thickness on XRD 2D-transmit WAXS for inner layer of sheets, and compared with 2D-transmit WAXS image of mesogenic epoxy thin sheet (PDF)

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### Notes

The authors declare no competing financial interest.

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