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Research Article

Identification of anti-adipogenic withanolides from the roots of Indian ginseng (*Withania somnifera*)Seoung Rak Lee ^{a, b}, Bum Soo Lee ^a, Jae Sik Yu ^a, Heesun Kang ^a, Min Jeong Yoo ^a, Sang Ah Yi ^a, Jeung-Whan Han ^a, Sil Kim ^a, Jung Kyu Kim ^c, Jin-Chul Kim ^d, Ki Hyun Kim ^{a, *}^a School of Pharmacy, Sungkyunkwan University, Suwon, Republic of Korea^b Department of Chemistry, Princeton University, NJ, United States^c School of Chemical Engineering, Sungkyunkwan University, Suwon, Republic of Korea^d KIST Gangneung Institute of Natural Products, Natural Product Informatics Research Center, Gangneung, Republic of Korea

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ABSTRACT

Background: *Withania somnifera* (Solanaceae), generally known as Indian ginseng, is a medicinal plant that is used in Ayurvedic practice for promoting health and longevity. This study aims to identify the bioactive metabolites from Indian ginseng and elucidate their structures.

Methods: Withanolides were purified by chromatographic techniques, including HPLC coupled with LC/MS. Chemical structures of isolated withanolides were clarified by analyzing the spectroscopic data from 1D and 2D NMR, and HR-ESIMS experiment. Absolute configurations of the withanolides were established by the application of NMR chemical shifts and ECD calculations. Anti-adipogenic activities of isolates were evaluated using 3T3-L1 preadipocytes with Oil Red O staining and quantitative real-time PCR (qPCR).

Results: Phytochemical examination of the roots of Indian ginseng afforded to the isolation of six withanolides (**1–6**), including three novel withanolides, withasilolides G–I (**1–3**). All the six compounds inhibited adipogenesis and suppressed the enlargement of lipid droplets, compared to those of the control. Additionally, the mRNA expression levels of *Fabp4* and *Adipsin*, the adipocyte markers decreased noticeably following treatment with 25 μM of **1–6**. The active compounds (**1–6**) also promoted lipid metabolism by upregulating the expression of the lipolytic genes *HSL* and *ATGL* and downregulating the expression of the lipogenic gene *SREBP1*.

Conclusion: The results of our experimental studies suggest that the withasilolides identified herein have anti-adipogenic potential and can be considered for the development of therapeutic strategies against adipogenesis in obesity. Our study also provides a mechanistic rationale for using Indian ginseng as a potential therapeutic agent against obesity and related metabolic diseases.

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1. Introduction

Withania somnifera (Solanaceae), widely recognized as Indian ginseng and ashwagandha, is widely distributed in the dry regions of India, Sri Lanka, South Africa, and the Mediterranean [1,2]. Indian ginseng has been used as a functional food for promoting health and longevity via activation of the immune system against diseases, revitalizing the organ systems in weakened conditions, and curbing the process of aging [2,3]. As it is medically safe to use Indian

ginseng, the herb has been used for a long time by people of all ages and sexes, and even during pregnancy, and has not been reported to produce any side effects [3,4]. Indian ginseng is commercially cultivated in the Indian states of Uttar Pradesh, Madhya Pradesh, Punjab, and Gujarat, and India is exporting the roots of this herb to other countries [5]. The herb is commonly known as rasayana in Ayurvedic practice. As the Queen of Indian herbs, the roots of Indian ginseng have been traditionally used in Indian Ayurveda as a tonic for vitality, longevity, and the maintenance and restoration of health [6]. As these properties are quite similar to those of ginseng, *W. somnifera* is also referred to as Indian ginseng [7]. The herb has been widely used for therapeutic purposes owing to its anti-oxidative, analgesic, antiepileptic, antiulcerative, and antibacterial

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properties [8]. Pharmacological and clinical studies have shown that this plant can be used for treating anxiety, cognitive and neurological disorders, inflammatory diseases, hyperlipidemia, and Parkinson's disease [9–11]. Interestingly, Indian ginseng has also been prescribed for alleviating tolerance and chronic dependence on various psychotropic medications [12]. Phytochemical studies on this herb have reported the presence of biologically active chemical constituents, including alkaloids, steroids, saponins and withanolides, having diverse pharmacological properties [13–15]. The main metabolites of Indian ginseng are alkaloids and steroidal lactones, known as withanolides, which are the most important constituents responsible for the medicinal properties of Indian ginseng [16]. It has been reported that the herb also contains other chemical metabolites, including acylsteryl glucosides, starch, reducing sugars, hantreacotane, ducitol, and various amino acids [13–15]. Withaferin A, one of the most representative withanolides found in *W. somnifera* has been reported to exhibit anti-tumor effects including pro-apoptotic and anti-angiogenesis activity [17]. Recently, withaferin A was also found to show anti-adipogenesis effects in 3T3-L1 adipocytes by reducing lipid accumulation and downregulating the expression of key activators of adipogenesis, peroxisome proliferator-activated receptor gamma (PPAR γ) and CCAAT/enhancer binding protein alpha (C/EBP α) as well as adipocyte fatty acid binding protein [17]. Despite the high nutritive and therapeutic values of Indian ginseng, there are few reports on the phytochemical investigation of the bioactive constituents of Indian ginseng. Therefore, we aimed to discover the bioactive metabolites in this promising natural source in this study.

In our ongoing endeavors to identify structurally and/or bioactive metabolites from a variety of natural sources [18–21], we conducted a phytochemical exploration of the MeOH extracts of the Indian ginseng (*W. somnifera*) roots. We have previously identified numerous secondary metabolites, including 13 withanolides, 5 phenolic compounds, and 1 alkaloid by phytochemical examination of the MeOH extracts of Indian ginseng. Of these metabolites, 6 were novel withanolides (withasilolides A–F), and 4 of these withasilolides showed cytotoxicity against several human cancer cells including SK-MEL-2, HCT-15, A549 and SK-OV-3) [22]. We also identified new phenylpropanoid esters, namely withaninsam A and withaninsam B, in addition to phenolic compounds and alkaloids with anti-inflammatory potential [23]. In this study, we further performed the phytochemical analysis of Indian ginseng coupled with the liquid chromatography/mass spectrometry (LC/MS) for identifying the novel bioactive metabolites in Indian ginseng. Our analyses afforded to the isolation and structural elucidation of 3 novel withanolides (1–3) and 3 known compounds (4–6). The chemical structures of three new withanolides were verified by interpretation and analysis of the data obtained by 1D and 2D NMR data, high-resolution mass spectrometry, gauge-including atomic orbital (GIAO) nuclear magnetic resonance (NMR) chemical shift and electronic circular dichroism (ECD) calculations. This study describes the separation/isolation and chemically structural determination of all the isolated withanolides (1–6) along with their effects on *de novo* adipogenesis and lipid metabolism in 3T3-L1 adipocytes.

2. Materials and methods

2.1. General experimental procedure and plant material

The related information was included in Supplementary materials.

2.2. Extraction and separation/isolation

The dried and crushed roots (1.1 kg) of *W. somnifera* were extracted using 80% MeOH (3.0 L \times 3 days) under reflux and filtered. Resultant filtrate was evaporated *in vacuo* to generate 180 g of crude MeOH extract, which was directly dissolved in 700 mL distilled H₂O and then solvent-partitioned with hexane, dichloromethane, ethyl acetate, and butanol, yielding 3.1, 4.2, 1.9, and 17.8 g of residue, respectively. The LC/MS analysis was performed using an in-house-built UV spectra library, and the results revealed that the withanolides were mainly present in the dichloromethane-soluble fraction. The isolation procedure to obtain the compounds 1–6 from dichloromethane fraction was described in Supplementary materials.

2.2.1. Withasilolide G (1)

White powder; $[\alpha]_D^{25}$ -27.1 (c 0.11, methanol); UV (methanol) λ_{\max} (log ϵ) 220 (3.4) nm; IR (KBr) ν_{\max} 3716, 2977, 2841, 2502, 1718, 1524, and 1032 cm⁻¹; ECD (methanol) λ_{\max} ($\Delta\epsilon$) 216 (-1.1), 260 (+2.1), and 304 (-4.6) nm; ¹H and ¹³C NMR (800 MHz and 200 MHz, respectively), refer Table 1; positive HR-ESIMS m/z 455.2785 [M + H]⁺ (calculated for C₂₈H₃₉O₅, 455.2797).

2.2.2. Withasilolide H (2)

White powder; $[\alpha]_D^{25}$ +39.3 (c 0.15, methanol); UV (methanol) λ_{\max} (log ϵ) 230 (2.9) nm; IR (KBr) ν_{\max} 3713, 3330, 2957, 1760, 1459, 1133, and 1038 cm⁻¹; ECD (methanol) λ_{\max} ($\Delta\epsilon$) 223 (-0.6), 255 (+0.3), and 337 (-4.1) nm; ¹H and ¹³C NMR (800 MHz and 200 MHz, respectively), refer Table 1; positive HR-ESIMS m/z 489.2852 [M + H]⁺ (calculated for C₂₈H₄₁O₇, 489.2852).

2.2.3. Withasilolide I (3)

White powder; $[\alpha]_D^{25}$ +14.2 (c 0.10, methanol); UV (methanol) λ_{\max} (log ϵ) 225 (3.1) nm; IR (KBr) ν_{\max} 3713, 2972, 2838, 2497, 1713, 1520, and 1029 cm⁻¹; ECD (methanol) λ_{\max} ($\Delta\epsilon$) 256 (+2.4) and 336 (-3.1) nm; ¹H and ¹³C NMR (800 MHz and 200 MHz, respectively), refer Table 1; positive HR-ESIMS m/z 487.2689 [M + H]⁺ (calculated for C₂₈H₃₉O₇, 487.2696).

2.3. Computational analyses

All the conformers reported in this study were generated using the MacroModel module (version 2019–3, Schrödinger LLC) [24]. The detailed process for geometry optimization was included in Supplementary materials.

The ECD calculations for conformers 1a/2a/3a and 1b/2b/3b were conducted at identical theoretical levels and basis sets, the detailed process of which was described in Supplementary materials. The ECD was visualized using SigmaPlot, version 14.0.

The calculations for the optimized conformers of 17R-2/17R-3 and 17S-2/17S-3 were performed with GIAO magnetic shielding constants at the B3LYP/6-31 + G(d) level of theory [25]. The detailed process for the NMR chemical shifts calculation was inserted in Supplementary materials.

2.4. Cell culture and cell differentiation

The related information was included in Supplementary materials.

Table 1
¹H (800 MHz) and ¹³C NMR (200 MHz) data of compounds **1–3** in CDCl₃.^{a,b}

Position	1		2		3	
	δ_C	δ_H (J in Hz)	δ_C	δ_H (J in Hz)	δ_C	δ_H (J in Hz)
1	209.6 s		202.9 s		203.0 s	
2 α	39.5 t	2.77 dd (20.0, 4.5)	129.0 d	5.82 dd (10.0, 2.5)	128.9 d	5.85 dd (10.0, 2.5)
2 β		3.29 dd (20.0, 2.5)				
3	125.6 d	5.79 m	139.7 d	6.57 ddd (10.0, 5.0, 2.0)	139.5 d	6.58 ddd (10.0, 5.0, 2.0)
4 α	129.1 d	6.07 d (10.5)	36.6 t	2.51 dd (19.0, 5.0)	36.6 t	2.53 dd (19.0, 5.0)
4 β				2.65 m		2.68 m
5	144.7 s		73.1 s		73.2 s	
6	127.1 d	5.80 d (4.5)	56.1 d	3.01 d (3.5)	55.9 d	3.05 d (4.0)
7	64.5 d	3.94 m	57.0 d	3.29 dd (3.5, 2.0)	56.8 d	3.35 m
8	36.2 d	1.50 m	35.3 d	1.75 m	36.4 d	1.92 m
9	33.6 d	2.01 m	34.9 d	1.56 m	35.1 d	1.54 m
10	52.8 s		50.9 s		50.9 s	
11 α	21.8 t	1.97 m	21.3 t	2.82 m	21.7 t	2.78 m
11 β		1.54 m		1.28 m		1.32 m
12 α	39.7 t	1.35 m	32.5 t	1.42 m	33.2 t	1.74 m
12 β		2.00 m		1.77 m		2.16 m
13	43.3 s		48.1 s		50.5 s	
14	50.1 d	1.51 m	46.0 d	2.24 m	46.6 d	2.00 m
15 α	21.6 t	1.97 m	22.3 t	1.90 m	21.5 t	1.56 m
15 β		1.54 m		1.29 m		1.40 m
16 α	23.7 t	1.23 m	32.2 t	1.64 m	33.4 t	1.74 m
16 β		1.79 m		2.32 m		2.35 m
17	54.4 d	1.48 m	89.4 s		87.0 s	
18	13.4 q	0.90 s	16.2 q	0.89 s	17.2 q	1.02 s
19	18.4 q	1.32 s	14.7 q	1.15 s	14.6 q	1.19 s
20	75.5 s		76.0 s		78.1 s	
21	20.7 q	1.27 s	21.9 q	1.24 s	19.9 q	1.37 s
22	80.7 d	4.19 dd (13.5, 3.5)	80.6 d	4.45 dd (11.0, 3.5)	82.3 d	4.63 dd (11.0, 3.0)
23 α	31.6 t	2.11 m	30.4 t	1.56 m	31.5 t	2.28 m
23 β		2.37 m		2.27 m		2.79 m
24	149.1 s		30.8 d	1.74 m	150.7 s	
25	122.1 s		40.3 d	2.12 dq (9.5, 6.5)	120.9 s	
26	166.3 s		174.4 s		164.8 s	
27	12.3 q	1.86 s	14.0 q	1.21 d (6.5)	12.1 q	1.87 s
28	20.4 q	1.93 s	20.9 q	1.11 d (7.0)	20.5 q	1.97 s

^a The coupling constants (in parentheses) are in Hz.^b The ¹³C NMR data were assigned on the basis of the data obtained in the HSQC and HMBC experiments.

2.5. Oil Red O staining

The staining of lipid droplets in the differentiated adipocytes was conducted using Oil Red O working solution for visualization, the procedure of which was described in Supplementary materials in detail.

2.6. Cell viability

The 3T3-L1 preadipocytes were seeded in 6-well plates and treated with compounds **1–6** at concentrations of 12.5 and 25 μ M for 24 h. The cells were detached with trypsin/EDTA solution (TE) and diluted with PBS. Counting of the number of cells was conducted using an equipment of LUNA-II™ Automated Cell Counter (Logos Biosystems).

2.7. Reverse transcription and quantitative real-time PCR (RT-qPCR)

In order to extract the total RNA from the mature adipocytes, we used the Easy-Blue reagent (Intron Biotechnology, Seongnam, Korea) based on the manufacturer's instructions. The related process was included in Supplementary materials in detail.

2.8. Statistical analyses

All the results and data presented as the standard error of the mean (SEM) for $n = 3$. Statistical significance was analyzed using

Student's *t*-test (two-tailed) in Excel and was assessed on the basis of the *P*-value (**P* < 0.05, ***P* < 0.01, ****P* < 0.001 vs. the control group).

3. Results and discussion

3.1. Compound isolation

LC/MS-based examination of the solvent-partitioned fractions from the Indian ginseng extracts combined with our in-house UV library revealed that the CH₂Cl₂-soluble fraction was rich in withanolides. This led to an intensive phytochemical exploration of the fraction using successive column chromatographic purification and preparative and semi-preparative HPLC, which afford to the isolation of 6 withanolides (**1–6**), of which 3 were identified as novel withanolides (**1–3**) (Fig. 1A).

3.2. Structural identification of the compounds

Compound **1** was separated as a white powder. The molecular formula of **1** was deduced to be C₂₈H₃₈O₅, based on the data obtained from NMR (Table 1) and positive-ion mode HR-ESIMS, which displayed one protonated ion peak at *m/z* 455.2785 [M + H]⁺ (calculated for C₂₈H₃₉O₅, 455.2797). The UV data of **1** displayed a distinctive feature at λ_{\max} 220 nm, indicating the existence of an α,β -unsaturated ketone group. The ¹H NMR data (Table 1) of **1** combined with the data from HSQC showed the existence of proton

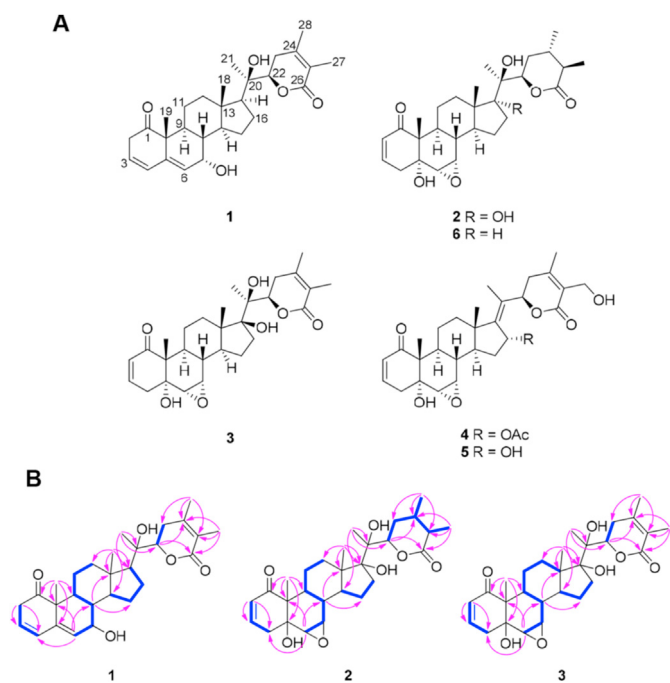


Fig. 1. (A) Chemical structures of withanolides (**1–6**). (B) Important HMBC (→) and COSY (→) correlations for compounds **1–3**.

signals for five methyls [δ_{H} 0.90 (3H, s), 1.27 (3H, s), 1.32 (3H, s), 1.86 (3H, s), and 1.93 (3H, s)], six methylenes [δ_{H} 1.23/1.79 (each 1H, m), 1.35/2.00 (each 1H, m), 1.54/1.97 (each 1H, m, overlap), 1.54/1.97 (each 1H, m, overlap), 2.11/2.37 (each 1H, m), 2.77 (1H, dd, $J = 20.0$, 4.5 Hz)/3.29 (1H, dd, $J = 20.0$, 2.5 Hz)], and nine methines [δ_{H} 1.48 (1H, m), 1.50 (1H, m), 1.51 (1H, m), 2.01 (1H, m), 3.94 (1H, m), 4.19 (1H, dd, $J = 13.5$, 3.5 Hz), 5.79 (1H, m), 5.80 (1H, d, $J = 4.5$ Hz), and 6.07 (1H, d, $J = 10.5$ Hz)]. The ^{13}C NMR data, combined with the results of HSQC/HMBC data analyses, revealed 28 carbon resonance signals that were classified into five methyl groups (δ_{C} 12.3, 13.4, 18.4, 20.4, and 20.7), six methylenes (δ_{C} 21.6, 21.8, 23.7, 31.6, 39.5, and 39.7), nine methines (δ_{C} 33.6, 36.2, 50.1, 54.4, 64.5, 80.7, 125.6, 127.1, and 129.1), and eight non-protonated carbons (δ_{C} 43.3, 52.8, 75.5, 122.1, 144.7, 149.1, 166.3, and 209.6). Comprehensive scrutiny of the NMR spectral data suggested that the chemical structure of **1** was very similar to that of withasilolide A, previously identified from Indian ginseng, but the apparent difference in the planar structures of **1** and withasilolide A was in the position of the olefinic group in ring A [22]. From the COSY correlations from H₂-2 to H-4 and the key HMBC correlations of H₂-2/C-3, H₂-2/C-4, and H-6/C-4, we deduced that the double bond of C-2/C-3 in withasilolide A was shifted to between C-3 and C-4 in **1**, leading to the formation of the conjugated 3,5-diene system in rings A and B (Fig. 1B). The complete gross structure of **1** was further verified by the detailed inspection of the COSY and HMBC data (Fig. 1B).

The absolute stereochemistry of **1** was determined by examination of the spectrum obtained by ROESY, vicinal proton coupling constants, and data obtained from ECD. The negative Cotton effect at around 304 nm in the ECD data of **1** led to the confirmation of the *trans*-linkage in the A/B ring [26]. ROESY correlations of H₃-19/H-8, H-7/H-8, and H-8/H₃-18 revealed their β -orientation in the partial structure of **1**. Meanwhile, the α -orientations at H-9, H-14, H-17, H₃-21, and H-22 were also confirmed by the ROESY correlations, as depicted in Fig. 2. In particular, the ROESY correlations of H₃-21/H-17 and H₃-21/H-23 α determined the *R*-form of the hydroxylated quaternary carbon at C-20. According to the spectroscopic values

reported in the previous study, the ^1H NMR spectrum of H-22 α possesses a doublet of doublets with two different coupling constants ($J = 9.0$ – 13.8 and 0.5 – 4.0 Hz) [22,26], while the ^1H NMR spectrum of H-22 β shows two similar coupling constants ($J = 2.5$ – 7.0 , 2.0 – 5.0 Hz) [27]. Based on the two different coupling constants observed as $J = 13.5$ and 3.5 Hz, the orientation of H-22 was determined to be α -form, which was also verified by the ROESY correlations of H-16 α /H-22 and H-17/H-22. In fact, the absolute configuration of C-22 was unambiguously elucidated to be *R*-form by the positive Cotton effect at 250–260 nm derived from the $n \rightarrow \pi^*$ transition of the α,β -unsaturated δ -lactone [22,27]. Finally, for the absolute configuration determination of **1**, the ECD data of two possible isomers, **1a** (7*S*,20*R*,22*R*) and **1b** (7*R*,20*S*,22*S*), were subjected to quantum mechanics calculations. The results of calculation of the ECD data revealed that the ECD curve of **1a** (blue line) was consistent with the experimentally determined ECD spectrum of **1** (Fig. 3A). Therefore, the chemical structure of **1** was successfully determined as illustrated in Fig. 1 and its trivial name was withasilolide G.

Compound **2**, isolated as a white powder, had the molecular formula of C₂₈H₄₀O₇, based on the data obtained by positive-ion mode HR-ESIMS, exhibiting a protonated ion peak at m/z 489.2852 [M + H]⁺ (calculated for C₂₈H₄₁O₇, 489.2852). The ^1H and ^{13}C NMR spectral data (Table 1) of **2**, combined with the evidence obtained by HSQC, revealed the presence of a distinctive A/B ring pattern observed in 6 α ,7 α -epoxy-5 α -hydroxy-1-oxowitha-2-enolide [21], as determined from the NMR signals at C-1 (δ_{C} 202.9), C-2 [δ_{H} 5.82 (1H, dd, $J = 10.0$ and 2.5 Hz) and δ_{C} 129.0], and C-3 [δ_{H} 6.57 (1H, ddd, $J = 10.0$, 5.0 and 2.0 Hz) and δ_{C} 139.7] for an α,β -unsaturated ketone; C-5 (δ_{C} 73.1) for a hydroxylated quaternary carbon; and C-6 [δ_{H} 3.01 (1H, d, $J = 4.0$ Hz) and δ_{C} 56.1] and C-7 [δ_{H} 3.29 (1H, dd, $J = 3.5$ and 2.0 Hz) and δ_{C} 57.0] responsible for an epoxy functional group. The partial structure of **2** was also supported by interpretation of the key HMBC correlations from H-2/C-3, H-3/C-1, H-3/C-5, H₃-19/C-5, H₃-19/C-1 and H-6/C-10, as well as the key COSY correlations from H-2 to H-4 and H-6 to H-9 (Fig. 1B). The existence of two angular methyl units at C-18 [δ_{H} 0.89 (3H, s) and δ_{C} 16.2] and C-19 [δ_{H} 1.15 (3H, s) and δ_{C} 14.7] and three methyls at C-21 [δ_{H} 1.24 (3H, s) and δ_{C} 21.9], C-27 [δ_{H} 1.21 (3H, d, $J = 6.5$ Hz) and δ_{C} 14.0], and C-28 [δ_{H} 1.11 (3H, d, $J = 7.0$ Hz) and δ_{C} 20.9] were also clearly detected in the NMR data of **2**. Moreover, the presence of a 3,4-dimethyltetrahydropyran-2-one moiety was elucidated by the key COSY correlations from H-22 to H-25, and the HMBC correlations for H-22/C-26, H₃-27/C-26, H₃-27/C-24, H₃-28/C-23 and H₃-28/C-25 (Fig. 1B). A thorough analysis of the 2D NMR data of **2** indicated that the NMR spectroscopic values of **2** were quite similar to those of withasilolide C identified from the same natural source in a previous study [22], with the difference of a hydroxyl group at C-20 (δ_{C} 76.0) in **2**. This inference was verified by important HMBC correlations from H₃-21, H-22, H₂-23 and H₂-16 to C-20 (Fig. 1B), which led to the final structure of **2**.

By comparing the spectroscopic values and analyzing the ROESY experiment and ECD data, the absolute configuration of **2** was elucidated to be identical to that of withasilolide C. The α -position of the epoxy group between C-6 and C-7 was assigned by the ROESY correlations of H-7/H-8, H-6/H-7 and H₃-19/H-8, and the coupling constant of 3.5 Hz between H-6 and H-7 (Fig. 2), which also indicated the *trans*-linkage in the conformation of A/B ring in **2**, because a smaller coupling constant as 0–2 Hz for H-6 is typical in *cis*-linkages [22,26,27]. The *trans*-linkage for the A/B ring was further verified by the key negative Cotton effect around 340 nm in the ECD experiment of **2** [22], and the configurations at C-5, C-6, and C-7 were assigned accordingly, as depicted in Fig. 2. Additionally, the weak positive Cotton effect in the region of 250–260 nm provided critical evidence for confirming the 22*R* configuration [22,27]. The

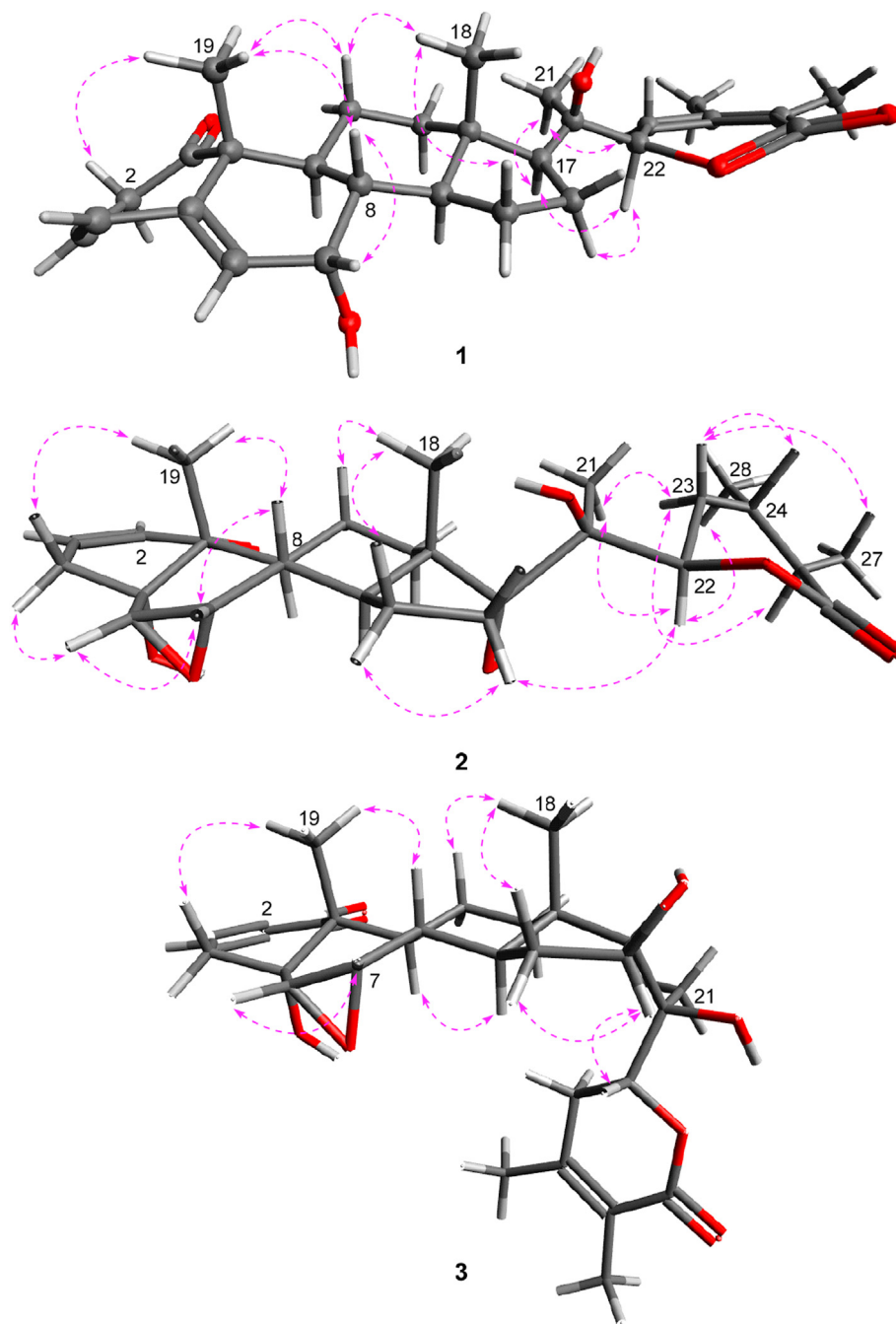


Fig. 2. Important ROESY correlations for compounds 1–3.

stereochemistry of the δ -lactone ring was assigned from the ROESY data analysis (Fig. 2), where the correlations from H-23 β to H-24, H-23 α to H-25, and H-22 to H₃-28 suggested that H-24 and H-25 were in β - and α -orientations, respectively. H-24 and H-25 were confirmed to be in axial orientation, as revealed by the coupling constant as 9.5 Hz [22]. The negative Cotton effect at around 220 nm in the ECD experiment supported the half-chair confirmation of the saturated δ -lactone of **2** [22,27], which indicated a configuration of 22*R*,24*S*,25*R*. Considering the ROESY correlations from H-22 and H-23 α to H₃-21, the chiral center at C-20 was determined to be in *S*-form (Fig. 2). In a previous study, ¹³C NMR γ -gauche effects on the carbon chemical shifts of C-18 and C-12, as well as C-21 were used for distinguishing epimeric withanolides

with OH-17 α and OH-17 β groups [22,27]. The ¹³C NMR data of the epimers of 17-hydroxywithanolides indicated that the alteration in the OH-17 orientation from β to α induced an upfield shift in the C-12 and C-18 chemical shifts, but a downfield shift in the C-21 chemical shift in the same CDCl₃ solvent owing to the γ -gauche effects of the OH-17 group [27]. The differences in the C-12, C-18, and C-21 carbon chemical shifts were determined for identifying the stereochemistry at C-17. The differences between **2** and philadelphicalactone C were $\Delta\delta_{C-12}$ (+0.8 ppm), $\Delta\delta_{C-18}$ (+0.4 ppm), and $\Delta\delta_{C-21}$ (−0.1 ppm), while the differences between **2** and 17-*epi*-philadelphicalactone A were $\Delta\delta_{C-12}$ (−0.9 ppm), $\Delta\delta_{C-18}$ (−0.5 ppm), and $\Delta\delta_{C-21}$ (+2.0 ppm) in the same CDCl₃ solvent. These results provided significant evidence that the absolute stereochemistry of

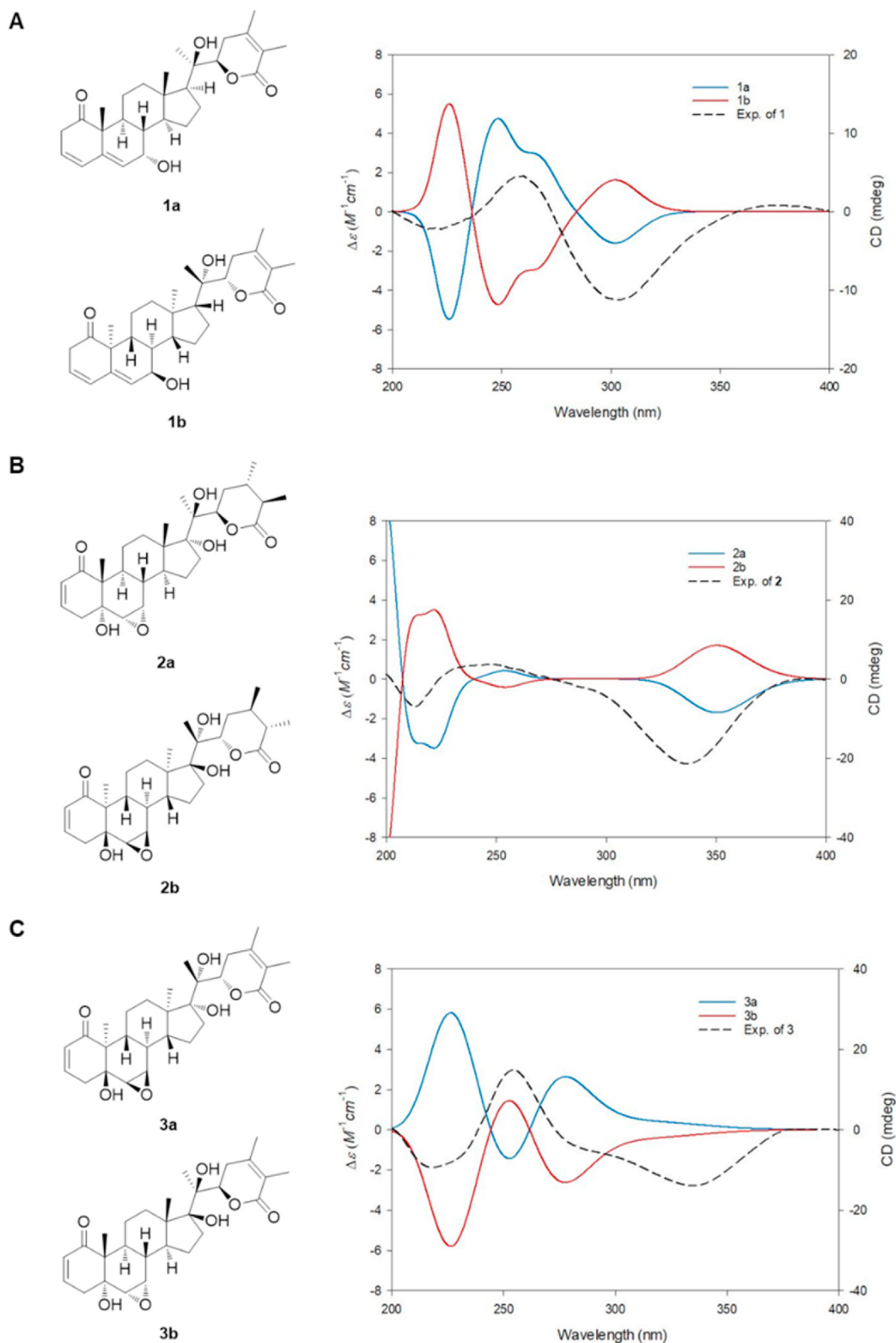


Fig. 3. Experimental and computed ECD data of compound (A) **1**, (B) **2** and (C) **3**.

C-17 was the *R*-form [27]. Additionally, the proton chemical shifts of H₂-12, H₃-18, and H-21 of **2** were similar to those of the 17 α -hydroxywithanolide, philadelphicalactone C [27]. To prove this assignment, the NMR chemical shifts of 17*R*-**2** and 17*S*-**2** as the two

possible diastereomers were calculated, and the calculated NMR chemical shifts were empirically scaled for removing system errors, which is a common practice in such calculations [28–30]. The calculated NMR chemical shifts of 17*R*-**2**/17*S*-**2** and the

experimental values obtained for **2** were compared. The differences ($\Delta\delta$) were resolved based on the formula $\Delta\delta = \delta_{\text{calcd}} - \delta_{\text{exptl}}$ and are accessible in Table S1 (Supplementary data). The correlation coefficient (R^2) obtained by linear regression analysis, largest absolute deviation (LAD), and the mean absolute deviation (MAD) for 17R-**2** were 0.9970 (Fig. 4A), 3.96, and 0.99 (Fig. 4B), respectively, while the R^2 , LAD, and MAD values for 17S-**2** were 0.9965, 5.73, and 1.18, respectively, which indicated that the structure of 17R-**2** was highly accurate and consistent with the 17R assignment determined by NMR analysis. For its absolute stereochemistry, an experiment for ECD calculations were achieved by comparing the experimental ECD data of **2** with the computed ECD data of **2a** (17R,20S,22R,24S,25R) and **2b** (17S,20R,22S,24R,25S) as the two possible enantiomers (Fig. 3B). The experimental data of **2** showed good correlation with the computed curve of **2a**. Accordingly, the chemical structure of **2** was determined to be (17R,20S,22R,24S,25R)-6 α ,7 α -epoxy-5 α ,17,20-trihydroxy-1-oxowitha-2-enolide and the trivial name was withasilolide H.

Compound **3**, obtained as a white powder showed the HR-ESIMS data displaying a protonated ion peak at m/z 487.2689 [M + H]⁺ (calculated for C₂₈H₃₉O₇, 487.2696), which suggested the molecular formula of C₂₈H₃₈O₇. The NMR spectral data (Table 1) of **3** contained signals similar to those of **2**, with the exception of the signals corresponding to the double bond of C-24 (δ_C 150.7)/C-25 (δ_C 120.9) in **3**. The partial structure of the δ -lactone ring was verified by the key HMBC correlations of H-22/C-24, H₃-27/C-24, H₃-27/C-26, H₃-27/C-25, H₃-28/C-24, H₃-28/C-23 and H₃-28/C-25 (Fig. 1B), which verified the complete planar structure of **3** as depicted in Fig. 1B.

The stereochemistry of **3** was also established using the ROESY data, proton coupling constants, ECD calculations, and NMR chemical shifts calculations. Overall, the stereochemistry of **3** was identical to that of **2** because most of the data, including the ROESY and ECD data, and the characteristic J values were identical to those of compound **2**. However, there were noticeable differences in the signals corresponding to C-17 (δ_C 87.0 for **3**; δ_C 89.4 for **2**), which indicated that the chirality of C-17 differed between compounds **2** and **3**. For the stereochemistry at C-17 in **3**, the NMR spectroscopic values of the related withanolides were compared. As aforementioned, the key chemical shifts in the ¹H and ¹³C NMR spectra for C-12 and C-18 were relatively deshielded, but those of C-21 were shifted upfield by the presence of the 17 β -hydroxy group, compared to those of the 17 α -hydroxy group [27]. This difference was distinct between compounds **2** and **3** as compound **2** is a 17 α -hydroxywithanolide. The differences in the ¹³C NMR chemical shifts of **2** and **3** ($\Delta = \delta_2 - \delta_3$) were -0.7, -1.0, and +2.0 ppm for C-12, C-18, and C-21, respectively, which confirmed the absolute configuration of C-17 in **3** to be S-form. To elucidate the absolute configuration of C-17, NMR chemical shifts of 17R-**3** and 17S-**3** as the two possible diastereomers were calculated, and the NMR chemical shifts of 17R-**3** and 17S-**3** that had been calculated computationally were compared with the corresponding experimentally determined values for **3**. The differences ($\Delta\delta$) were calculated (Table S2, Supplementary data), and the results showed that the values of R^2 , LAD, and MAD were 0.9983 (Fig. 4C), 6.35 and 1.08 (Fig. 4D), respectively, for 17R-**3**, and 0.9986, 4.14, and 0.95, respectively, for 17S-**3**. The results indicated that the 17S-**3** form was more reasonable, as it agreed with the 17S assignment determined by NMR analysis. Finally, the experimental ECD curve of **3** agreed with the computed ECD spectrum for **3b** (17S,20S,22R) rather than with that of **3a** (17R,20R,22S) (Fig. 3C). Consequently, the chemical structure of **3** was finally elucidated as shown in Fig. 1 and the trivial name was withasilolide I.

By comparing the NMR spectroscopic data obtained herein with those previously reported in literatures [22,23] along with LC/MS

analysis, we identified 3 known withanolides in this study, namely, withasilolide F (**4**), withasilolide E (**5**), and withasomniferol B (**6**).

3.3. Evaluation of the anti-adipogenic activity of the isolated compounds

Obesity is a major health problem that is caused by the summation of multiple factors, including genetic, dietary, lifestyle-related, and environmental factors, which lead to the excessive accumulation of body fat in the adipose tissues [31]. The growth of adipose tissues occurs with the differentiation of preadipocytes in the adipose tissues into adipocytes, and the generation and accumulation of lipid droplets in the adipocytes [32,33]. Therefore, the discovery of active compounds that prevent adipogenesis and lipogenesis has been considered to be a potential therapeutic strategy for the prevention of obesity and metabolic diseases, which has attracted attention to the discovery of natural compounds with anti-adipogenic potential.

Before assessing the effect of compounds **1–6** on adipogenesis, the 3T3-L1 preadipocytes were treated with all the compounds at different concentrations of 0, 12.5, and 25 μ M for 24 h, for evaluating their cytotoxicity in the 3T3-L1 preadipocytes. None of the compounds caused cytotoxic effects at concentrations of up to 25 μ M (Figure S22). Therefore, the 3T3-L1 cells were treated with the compounds **1–6** during adipogenesis at a concentration of 25 μ M, for evaluating their anti-adipogenic activities in the subsequent experiments. After 10 days of differentiation, the lipid droplets within the mature adipocytes were stained using Oil Red O working solution [34]. Microscopic examination of the stained adipocytes revealed that compounds **1–6** significantly inhibited adipogenesis and suppressed the enlargement of lipid droplets compared to those of the control setup (Fig. 5B). Furthermore, the mRNA expression levels of *Fabp4* and *Adipsin*, the adipocyte marker genes were markedly reduced following treatment with 25 μ M of compounds **1–6** (Fig. 5C). To assess dose-dependent response of one of the effective compounds, we evaluated the protein level of A-FABP, an adipogenic marker, according to the treatment with diverse concentrations of compound **2**, which was a sufficient amount of the new compound for further evaluation. The immunoblot data showed that treatment with compound **2** reduced the protein level of A-FABP in dose-dependent manner (Fig. 5D). These results indicated that the compounds **1–6** alleviated adipogenesis in the 3T3-L1 preadipocytes.

As compounds **1–6** were able to suppress adipogenesis, we investigated whether they were capable of controlling lipid metabolism (Fig. 5E–G). The effects of compounds **1–6** on lipolysis and lipogenesis were measured by determining the difference in the ratio of expression of the lipolytic genes (*HSL* and *ATGL*) to the lipogenic gene (*SREBP1*) and comparing with that of the control. The mRNA expression of *HSL* and *ATGL*, the lipolytic genes in the 3T3-L1 cells was upregulated by treatment with compounds **1–5**; however, the mRNA expression of *ATGL* was not upregulated by compound **6** (Fig. 5E and F). On the other hand, the mRNA expression of *SREBP1*, the lipogenic gene was downregulated following treatment with 25 μ M of compounds **1–6** during adipogenesis (Fig. 5G). These data demonstrated that compounds **1–6** can enhance lipid metabolism by promoting lipolysis and inhibiting lipogenesis.

4. Conclusions

In conclusion, we identified 6 withanolides, including 3 novel withanolides, withasilolides G–I (**1–3**), from a MeOH extract of Indian ginseng roots by LC/MS. The effects of compounds **1–6** on lipid metabolism and adipogenesis were evaluated during

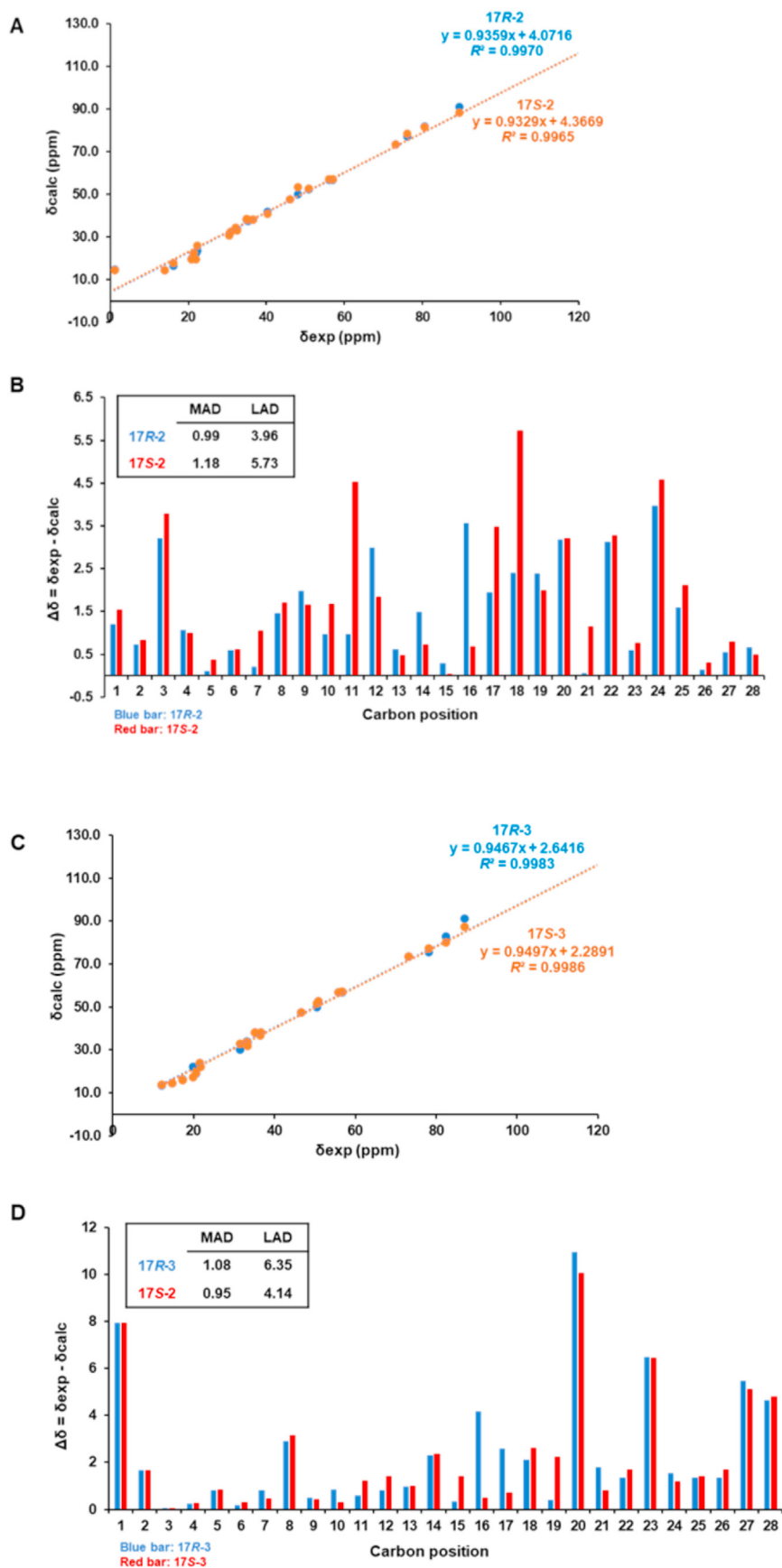


Fig. 4. (A) Regression analysis of the experimental versus computed ^{13}C NMR chemical shifts of 17R-2 and 17S-2. (B) Relative chemical shift errors between the computed and experimental ^{13}C NMR data for 17R-2 and 17S-2. (C) Regression study of the experimental versus computed ^{13}C NMR chemical shifts of 17R-3 and 17S-3. (D) Relative chemical shift errors between the computed and experimental ^{13}C NMR data for 17R-3 and 17S-3.

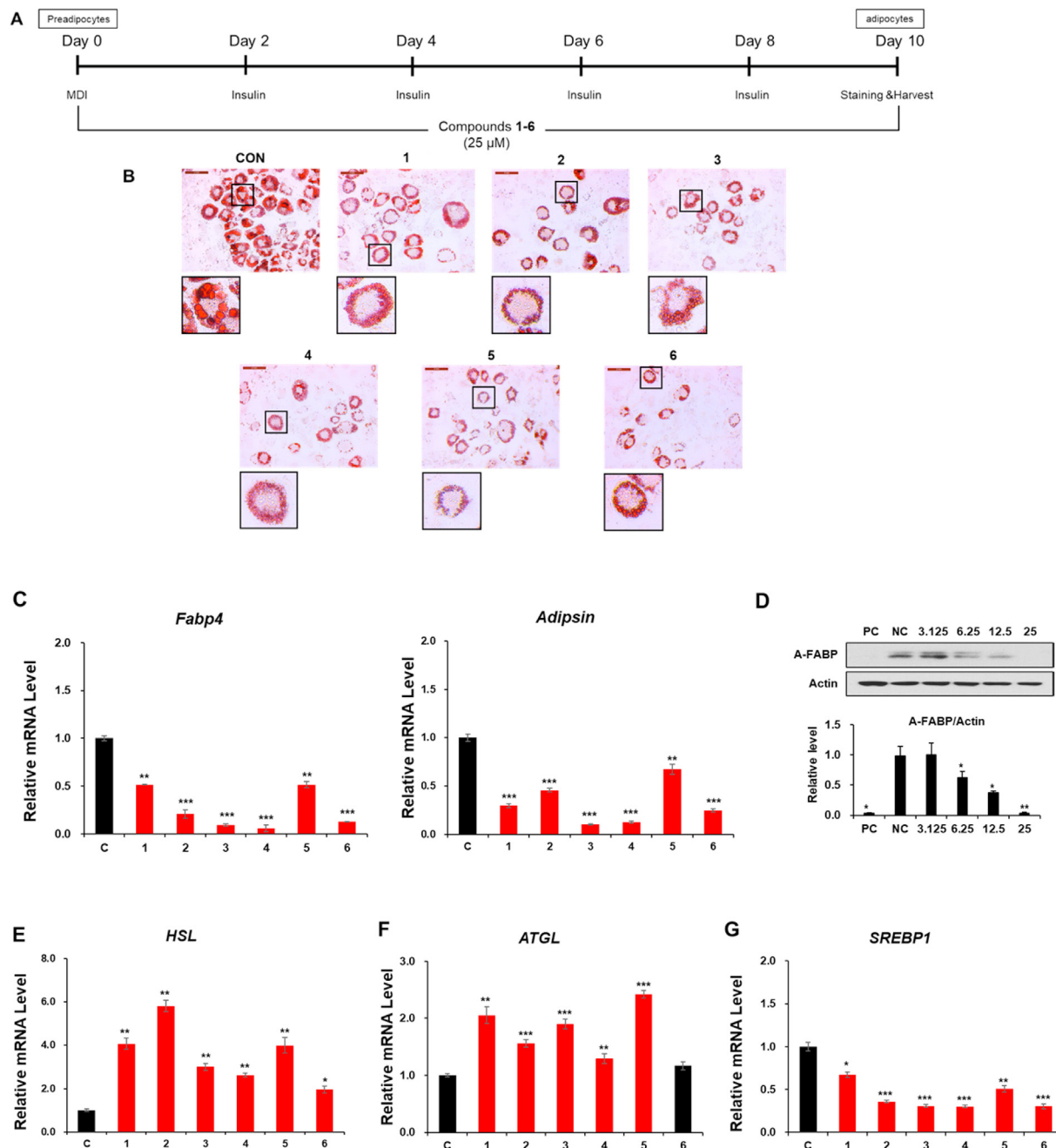


Fig. 5. Inhibitory effects of the compounds 1–6 on adipogenesis. (A) Schematic representation of the differentiation of 3T3-L1 preadipocytes into adipocytes following 10 days of culture. The 3T3-L1 cells were treated with the compounds 1–6 during differentiation (B) Images of adipocytes stained with Oil Red O following incubation with 25 μ M of compounds 1–6 during adipogenesis. (C) The relative mRNA expression of *Fabp4* and *Adipsin* in 3T3-L1 adipocytes incubated with 25 μ M of compounds 1–6 during adipogenesis. The inhibitory effects of compounds 1–6 on lipid metabolism. (D) Immunoblot data of 3T3-L1 adipocytes incubated with 30 μ M of resveratrol as a positive control (PC), DMSO as a negative control (NC) and indicated concentrations of compound 2. (E–G) The relative mRNA expression of *HSL* (E), *ATGL* (F), and *SREBP1* (G) in 3T3-L1 adipocytes incubated with 25 μ M of the compounds 1–6 during adipogenesis. The data are presented as the mean \pm SEM of n = 3 replicates. Statistical significance was compared to control (C, E–G) or negative control (D). *p < 0.05, **p < 0.01, and ***p < 0.001.

adipocyte maturation. Compounds 1–6 effectively inhibited the differentiation of 3T3-L1 preadipocytes to adipocytes by reducing the mRNA expression levels of *Fabp4* and *Adipsin*. We also observed that compounds 1–6 regulated lipid metabolism by upregulating the expression of *ATGL* and *HSL*, the lipolytic genes and down-regulating the expression of *SREBP1*, the lipogenic gene. The results

of our experimental studies demonstrated the anti-adipogenic activity of compounds 1–6, and we propose that these compounds have the potential to prevent adipogenesis in obesity, and can suppress excessive lipid accumulation in obesity-related metabolic disorders.

Declaration of competing interest

The authors declare no conflicts of interest.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.jgr.2021.09.004>.

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