CYSTEAMINE IMPROVES THE BACTERICIDAL EFFICACY OF INTRA-CANAL MEDICAMENTS AGAINST ENTEROCOCCUS FAECALIS

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Abstract

Aim. The aim of this study was to compare the antimicrobial efficacy of cysteamine, calcium hydroxide [Ca(OH)2], triple antibiotic paste (TAP), chlorhexidine (CHX) and their combinations against Enterococcus faecalis (E. Faecalis).

Methods. The E. Faecalis eradication capacity of cysteamine, Calcium hydroxide (Ca[OH]2), TAP, CHX, and their combinations was tested on E. Faecalis by Kirby Brauer disc diffusion method.

Results. Cysteamine in combination with TAP was able to completely eradicate E. Faecalis within 24 hours. Ca(OH)2 was unable to show its effect on E. Faecalis in the given time.

Conclusion. Cysteamine increased the E. Faecalis eradicating capacity of TAP and also showed positive results when used in combination with Ca(OH)2, which if used alone was unable to show any action in 24 hours.

Keywords: calcium hydroxide, chlorhexidine, cysteamine, Enterococcus Faecalis, triple antibiotic paste

Introduction

The invasion of the root canal system by bacteria leads to the onset and persistence of periapical infection. Thus, the objective of root canal treatment is to completely eliminate the bacteria from the root canals and its extensions. Although chemo-mechanical preparation effectively reduces the microbial load of the root canal, it fails to eliminate the bacteria from accessory canals, isthmi, lateral canals and apical deltas [1]. Hence, it is recommended to utilize intra-canal medicaments between the appointments to reduce bacterial counts [2].

One of the commonly isolated bacteria from persistent root canal infections is Enterococcus Faecalis (*E. Faecalis*). Studies have shown that *E. Faecalis* can penetrate deep into the dentinal tubules, hence escaping the effects of chemo-mechanical preparation during the root canal. *E. Faecalis* enters a non-cultivable state during root canal treatment and survives the chemo-mechanical procedures that are proven to be bactericidal. After return to a favorable environment, *E. Faecalis* returns into a cultureable state [3,4].

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Calcium hydroxide $(Ca[OH]_2)$ is the most widely used intra-canal medicament and shows its bactericidal activity due to high alkaline pH. However, studies have proven that its antibacterial efficacy has been reduced in the presence of dentin [5,6].

Chlorhexidine (CHX) is a broad-spectrum antibiotic and shows the property of substantivity in which it shows its antibacterial activity for a long duration when it adheres to anionic substrates [7]. Because of such properties, it is commonly used during endodontic treatments. A combination of CHX with Ca(OH)₂ has been suggested previously, which has shown better results than CHX used alone [8].

Triple antibiotic paste (TAP), a mixture of metronidazole, ciprofloxacin and minocycline has been used as an intra-canal medicament to disinfect the root canal system for revascularization procedures. TAP is formed by mixing equal parts of all three antibiotics in powder form with saline or sterile water or glycerine as vehicle. This paste is sealed in the canal for 3-weeks [9].

Cysteamine is derived from cysteine and is the simplest aminothiol. The pKa value of cysteamine is 9.42. It deprotonates in alkaline environment and forms thiolate ions which are responsible for breaking the disulfide bond of bacterial proteins due to which proteins are denatured

and bacteria lose their structural integrity. One possible future implication of cysteamine in endodontics is due to its mucolytic property, making it highly effective against different types of bacteria present in biofilms in the root canal. Breaking of mucopolysaccharides by cysteamine will disrupt the structural integrity of biofilm rendering its inhabitants susceptible to various intracanal medicaments. Guo et al. recently used cysteamine and cysteamine with $Ca(OH)_2$ in a study and found it to be highly effective against *E. Faecalis* [10]. Hence, in this study cysteamine and its combinations with various other intra-canal medicaments were used against *E. Faecalis*.

Materials and methods

The present study was carried out following the methods described by Kirby, Bauer, Sherris and Turck [11]. In the Kirby Bauer method, the agar plate is seeded with the bacteria. The antibiotic discs are then placed on the bacteria seeded agar plate. When discs containing known concentration of antimicrobials are placed on the surface of inoculated plate, antimicrobial agents begin to diffuse and thus establish a concentration gradient around the disc. The highest concentration is closest to the disc. After incubation at 37°C for 24 hours, growth of microorganisms are seen on the plate except where the concentration of disc is sufficiently high enough to inhibit the growth. Following incubation, the diameter of the zone of inhibition around each disc is measured in millimeters with the help of an inhibition zone measuring scale. An inhibition zone measuring scale is used to measure the diameter of inhibition zone in millimeters around the disc.

Preparation of sample solutions

Solution of 200 mg/mL cysteamine (LobaChemie, Mumbai, India) was prepared in distilled water (Amanta Healthcare, Gujarat, India). Saturated $Ca(OH)_2$ (Prime Dental, Thane, India) solution was prepared fresh by dissolving $Ca(OH)_2$ powder in distilled water. Combination of cysteamine and $Ca(OH)_2$ was prepared by dissolving cysteamine in $Ca(OH)_2$ solution at a concentration of 10 mg/mL. TAP was formed by combining 200 mg ciprofloxacin, metronidazole 500 mg and minocycline 100 mg using propylene glycol as vehicle. Combination of cysteamine and TAP was done by dissolving cysteamine in TAP at a concentration of 10 mg/mL. Combination of CHX with Ca(OH)_2 was prepared by dissolving Ca(OH)_2 at a concentration of 10 mg/ml in CHX. All the solutions were filter sterilized by using 0.2 mm syringe filter.

E. Faecalis culture

E. Faecalis was coated on Brain heart infusion agar by means of a sterile swab. A sterile forceps were used to dip the blank sensitivity disc in the prepared solutions followed by placing the coated sensitivity disc in the agar plates having *E. Faecalis*. The samples were then incubated for 24-hours at 37 °C. Diameters of zone of inhibition were then measured after 24-hours (Figure 1).



Figure 1. Photograph showing representative samples of all the groups using disc diffusion method.

Statistical analysis

All experiments were performed 3 independent times and each time in triplicate. Results were presented as the mean \pm standard deviation. Statistical significance was determined by One-way analysis of variance (ANOVA) with the Tukey post-hoc test. Differences were considered significant at (p<0.001).

Results

Diameters of zone of inhibition were measured for all the groups in millimeters (mm) (Table I). Maximum zone of inhibition was seen with the combination of TAP and cysteamine while $Ca(OH)_2$ completely failed to prevent the growth of *E. Faecalis* (Figure 2). Differences between all the groups were significant except between Ca(OH)₂ compared to control group, and cysteamine with Ca(OH)₂ compared to CHX with Ca(OH)₂.

 Table I. Antibacterial efficacy of different intracanal medicaments (Values in mm).

	SAMPLE 1(mm)	SAMPLE 2 (mm)	SAMPLE 3(mm)
Cysteamine	8	11	9
Calcium hydroxide	0	0	0
TAP	26	28	25
Cysteamine + Ca(OH) ²	16	21	15
Cysteamine + TAP	29	33	30
$CHX + Ca(OH)^2$	14	16	15
Control	0	0	0

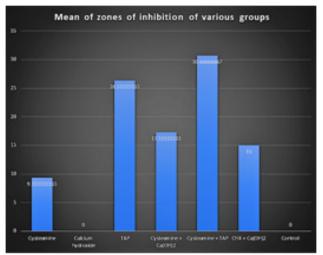


Figure 2. Graph representing comparative results (mean values) of all the groups in terms of antimicrobial efficacy.

Discussion

E. Faecalis is a predominant microorganism found in the cases where the endodontic treatment has failed [12]. Enterococci are well adapted for survival and persistence in a variety of adverse environments. This explains why it survives in root canal infections where nutrients are diminished. E. Faecalis colonized dentinal walls in case of application of endodontic medicaments. Aggregation substance, surface carbohydrates or fibronectin binding moieties favor adherence of organism to host collagen type I and extracellular matrix proteins present in the dentin. In invitro studies, *E. Faecalis* has been shown to invade dentinal tubules. It can colonize root canal and survive without the support of other bacteria. It is resistant to the antimicrobial effects of Ca(OH)₂ [3,12].

The antibacterial properties of $Ca(OH)_2$ are attributed to its high alkalinity and its ability to destroy the cytoplasmic membrane, denature bacterial proteins, and damage bacterial DNA. *E. Faecalis* is resistant to $Ca(OH)_2$ because *E. Faecalis* has a proton pump which can move the protons into the cell and make the cytoplasm acidic, which is necessary to survive in high pH of $Ca(OH)_2$ [13]. These findings contribute to our understanding of the response by *E. Faecalis* to the antimicrobial action of $Ca(OH)_2$.

The antibiotic action of $Ca(OH)_2$ is lowered by the buffering action of dentin [14]. To overcome this, addition of CHX has been introduced [15]. CHX is a cationic bisguanide that destroys the bacterial cytoplasmic membrane, leading to leakage of cellular components. It has been reported that 2% CHX is more efficient than $Ca(OH)_2$ in eliminating *E. Faecalis* biofilm [16]. Thus, CHX appears to be advantageous over $Ca(OH)_2$ to eradicate alkaline-resistant bacteria. CHX causes disturbance in the bacterial membrane after penetrating into the bacteria. Bactericidal activity of CHX and hydroxide ions is by causing disintegration of bacterial membranes. CHX also releases reactive oxygen species which inhibits growth of *E. Faecalis* by destruction of cell wall and plasma membrane. An issue that has been reported is the formation of toxic parachloroaniline when CHX and sodium hypochlorite are used together. On the other hand, studies have reported that the antimicrobial activity of CHX is reduced when it is associated with Ca(OH), [8].

Cysteamine is formed by oxidation of cystamine, which is a radio-protective agent. It can be used intravenously or orally to treat radiation sickness. It is available in bitartarate and hydrochloride forms for the treatment of neuropathic cystinosis in young patients. Cysteamine outperforms DNAse I which is a widely used mucolytic agent in patients with cystic fibrosis. The active thiol group of cysteamine is responsible for breakage of disulfide linkages of bacterial proteins. Cysteamine exerts its biofilm eradication effects through the destruction of disulfide links [17]. In our study cysteamine showed antimicrobial activity against *E. Faecalis* within 24 hours, which Ca(OH), failed to show.

Combination of Ca(OH)₂ and Cysteamine has shown to be bactericidal by acting synergistically. Even in the presence of Ca(OH)2, cysteamine deprotonates and disrupts the bacterial enzymes and membranes [10]. One major implications of this result is that addition of Cysteamine decreases the amount of time required by Ca(OH)₂ to be used as intra-canal medicament. We postulate that at a pH of 9 cysteamine still deprotonates and forms thiolate anions to show a synergistic anti- bacterial effect with Ca(OH)₂.

Although the exact reason for intra canal medicaments enhancing effect of cysteamine is not yet completely understood, it may be attributed to the fact that cysteamine not just only denatures key bacterial enzymes which are important for their metabolism and survival but also causes destruction of disulfide bonds of proteins acting synergistically with the intra-canal medicament [18].

TAP has been shown to be highly effective against *E*. *Faecalis*. Studies [19,20] have shown it to be more effective than the combination of $Ca(OH)_2$ and CHX which was also seen in our study. The efficacy of TAP was enhanced by combining it with cysteamine. This may be attributed to the fact that cysteamine caused the breakdown of bacterial integrity which was further severed by antimicrobial effect of the mixture of other three antibiotics.

Conclusion

The presented study may have a limitation of short duration of growth of *E. Faecalis* as a few weeks old bacterium is found to be more resistant towards medicaments. Nevertheless, cysteamine has shown positive outcomes in in-vitro studies so far; several studies, such as its toxicity to stem cells from apical papillae and using advanced methods for antibacterial analysis, would be required to label cysteamine safe for use as intra-canal medicament for endodontic purposes.

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