SCIENTIFIC REPORTS

Received: 05 May 2015 Accepted: 13 August 2015 Published: 10 September 2015

OPEN From the Au nano-clusters to the nanoparticles on 4H-SiC (0001)

Ming-Yu Li¹, Quanzhen Zhang¹, Puran Pandey¹, Mao Sui¹, Eun-Soo Kim¹ & Jihoon Lee^{1,2}

The control over the configuration, size, and density of Au nanoparticles (NPs) has offered a promising route to control the spatial confinement of electrons and photons, as a result, Au NPs with a various configuration, size and density are witnessed in numerous applications. In this work, we investigate the evolution of self-assembled Au nanostructures on 4H-SiC (0001) by the systematic variation of annealing temperature (AT) with several deposition amount (DA). With the relatively high DAs (8 and 15 nm), depending on the AT variation, the surface morphology drastically evolve in two distinctive phases, i.e. (I) irregular nano-mounds and (II) hexagonal nano-crystals. The thermal energy activates adatoms to aggregate resulting in the formation of self-assembled irregular Au nano-mounds based on diffusion limited agglomeration at comparatively low annealing temperature, which is also accompanied with the formations of hillocks and granules due to the dewetting of Au films and surface reordering. At high temperature, hexagonal Au nano-crystals form with facets along {111} and {100} likely due to anisotropic distribution of surface energy induced by the increased volume of NPs. With the small DA (3 nm), only dome shaped Au NPs are fabricated along with the variation of AT from low to elevated temperature.

Due to its wide band-gap, high current tolerance and high electron mobility, SiC is widely used in high power devices¹⁻⁵. α -SiC (3C-SiC) show zinc blende structure while β -SiC (4H- and 6H-SiC) consists of wurtzite. Among the poly-types, the 4H-SiC possesses the highest band-gap of 3.26 eV and the 6H shows 3.02 eV while 3C shows much lower value at 2.39 eV⁶. Recently, SiC are extensively applied for the fabrication of the high quality epitaxial graphene layers due to their thermal decomposition with a preferential sublimation of Si⁷, and the C-terminated surface generally requires lower temperature than Si-terminated surface to grow a graphene film with an identical thickness as a result of a more rapid sublimation⁸. On the other hand, owing to the localized surface plasmon resonance and large surface to volume ratio, Au NPs has received extensive research attentions for the optical⁹⁻¹¹, electric¹² and biological¹³ applications. The variation of shape, size and density of Au NPs can provide an proficient way to optimize the performance of the corresponding devices such as enhanced light absorption in the solar cells¹⁴, the performance of the localized surface plasmon resonance transducers by determining the surface plasmon decay and refractive index sensitivity¹⁵, controlling the memory window of FETs¹². Also, the Au NPs with a remarkable catalytic capacity¹⁶ can serve as nucleation sites for NWs by absorbing the vaporized target materials based on the vapor-liquid-solid growth mechanism¹⁷ and the diameter and length¹⁸, density^{19,20}, direction²¹, and shape²² of the NWs can be inherently determined by that of the Au NPs. Recently, Au nanoparticles (NPs) have been applied to control the Schottky barrier height with a variation of its size on 4H-SiC²³⁻²⁵. Au NPs have a potential of being applied in the various applications, however, the research on 4H-SiC is still relatively deficient and therefore, in this work we systematically investigate the controlled evolution of the various self-assembled Au nanostructures on 4H-SiC (0001) by the variation of annealing temperature (AT) with various deposition amounts (DAs). As shown in Fig. 1, depending on the DA, various nanostructures are fabricated, and evolve along with the increased AT. For example, with the 15 nm DA, Au nanostructures undergo drastic evolution in configurations with two distinctive phases: (I) irregular Au nano-mounds and (II) hexagonal Au nano-crystals. Phase

¹College of Electronics and Information, Kwangwoon University, Nowon-gu Seoul 139-701, South Korea. ²Institute of Nanoscale Science and Engineering, University of Arkansas, Fayetteville AR 72701, USA. Correspondence and requests for materials should be addressed to J.L. (email: jihoonleenano@gmail.com)



Figure 1. Illustration of the fabrication of self-assembled Au nanostructures on 4H-SiC (0001) by the control of annealing temperature (AT) with various Au deposition amounts (DAs). (a) Scanning electron microscopy (SEM) image $(5 \times 5 \mu m^2)$ of a sample annealed at 600 °C with the 15 nm DA. (a-1)–(a-2) SEM images of a hillocks $(1.6 \times 1.6 \mu m^2)$ and pinholes $(1.3 \times 1.3 \mu m^2)$. (b) Atomic force microscope (AFM) sideview of hexagonal Au nano-crystals at 850 °C ($5 \times 5 \mu m^2$). (b-1) AFM top-view ($1 \times 1 \mu m^2$) of a hexagonal nano-crystal. (c,d) Dome-shaped Au nano-particles fabricated with a DA of 3 nm at 800 °C and 900 °C. (c,d) AFM top-views of $500 \times 500 \text{ nm}^2$.

I: With the thermal energy supplied, Au adatoms can gradually diffuse and aggregated at the pinholes perforated by the voids to form the irregular Au nano-mounds along with the formation of the hillocks and granules at relatively low annealing temperature which can be described with the diffusion limited agglomeration model²⁶, as shown in Fig. 1(a),(a-1) and (a-2). Phase II: With the AT increase, all the Au structures gradually develop into the hexagonal nano-crystals owing to the enhanced surface diffusion, and the truncated facets formed to minimize the anisotropic surface energy, as shown with Fig. 1(b),(b-1). On the other hand, at relatively low DA (3 nm), the agglomeration process immediately proceed to the formation of dome-shaped Au NPs based on the Volmer-Weber growth model^{27,28}, and the Au NPs evolve with the increased size at the expense of the small Au NPs as a function of the AT.

Results and Discussion

Figure 2 shows the fabrication of self-assembled Au nano-mounds on 4H-SiC (0001) by the variation of the annealing temperature (AT) between 500 and 700 °C. Corresponding SEM images are provided in Fig. S4. In general, with the AT variation, the heterogeneous dewetting of Au film gradually occurred as a function of surface energy, resulting in a drastic surface morphology evolution from a continuous Au thin film to isolated irregular Au nano-mounds, which can be described in conjunction with a diffusion limited agglomeration (DLA) model²⁶. Initially, since the deposited Au thin film can generally possess a high vacancy concentration^{26,29}, being providing with the thermal energy, Au adatoms can spontaneously respond to diffuse, which can cause the nucleation of the vacancies, and in turn to



Figure 2. Evolution of the self-assembled Au nano-mounds on 4H-SiC (0001) at various annealing temperatures (AT) between 500 and 700 °C with 8 nm of Au deposition. (a,d) AFM side-views of 1×1 and $3 \times 3 \mu m^2$. (a-1)–(d-1) Cross sectional line-profiles. (a-2)–(d-2) Two dimensional (2-D) Fourier filter transform (FFT) power spectra.

form voids at random nucleation sites including highly strained sites and grain boundaries induced by the thermal expansion coefficient mismatch between Au and the SiC. Subsequently, the voids were kept forming with increased vacancy nucleation, and can perforated the Au film to form pinholes. Meanwhile, the agglomeration of Au can be initiated in the pinholes with a radius (R_p) bigger than the critical size (R_c) , which can be expressed as $R_C = \frac{t_{Au}}{\sin \Theta}^{30}$, where t_{Au} is the thickness of Au layer. Also, the equilibrium contact angle of the Au nano-particle (Θ) can be described with the interfacial energy densities of $\gamma_{SiC/Vac}$ (between SiC and vacuum), $\gamma_{Au/SiC}$ (between Au and SiC), and $\gamma_{Au/Vac}$ (between Au and vacuum) as $\Theta = \cos^{-1} \left(\frac{\gamma_{SiC/Vac} - \gamma_{Au/SiC}}{\gamma_{au/SiC}} \right)$. Namely, the pinholes with an R_p bigger than $\gamma_{Au/Vac}$ the R_C can possess a stronger driving force for the agglomeration (dewetting), which can eventually result in the formation of the irregular nano-mounds. Meanwhile, based on the thermodynamic diffusion theory, the l_D can be expressed as $l_D = \sqrt{D_S t}$, where t is residence time of Au adatoms. And the D_s (diffusion coefficient) can be given by $D_s = D_0 e^{\frac{-E_A}{kT} 31}$, where k is Boltzmann constant, D_0 and E_A (the diffusion barrier) are with certain values under an identical growth condition, therefore, the l_D can be determined by the variation of the T (surface temperature). Consequently, an enhanced diffusion length can be expected with more thermal energy supplied, and as a result, when the AT was increased, more pinholes can be formed with a proper size $(R_p > R_c)$, which can correspondingly enhance the agglomeration initiation. On the other hand, with the increased l_D , the connected irregular nano-mounds can have a tendency to expand and separate into isolated ones due to the Rayleigh instability^{32,33} as shown with AFM side- and top-views in Fig. 2, S3. In specific, the surface morphology with the 8 nm-thick Au deposition appeared quite smooth with only a few of nanometers of surface modulation, as shown in Fig. 2(a),(a-1). After being annealed at 500 °C, partial formation of the connected nano-mounds occurred in the Au film due to the limited Au adatom diffusion as shown



Figure 3. Formation of hillocks, voids, granules and Au nano-mounds with 15 nm Au deposition by annealing between 500–750 °C on 4H-SiC (0001). SEM images are of 100 (x) \times 76.67 (y) μ m².

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in Fig. 2(b),(b-1). When the AT was increased to 600 and 700 °C, Au adatoms aggregated more compactly, resulting in the shape transition from the connected Au nano-mounds to the isolated ones with a drastic vertical size expansion as shown in Fig. 2(c-1)–(d-1). In addition, the morphology evolution can be clearly observed with the sharp increases in both the root-mean-squared roughness (R_{RMS}) and surface area ratio (SAR), suggesting that the significant vertical size increase along with the AT increase, as shown in Fig. S3(e). As summarized in Table SI, the R_{RMS} increased × 47.7 times from ~1 to ~47.7 nm, and the SAR increased from 0.09% to 8.27%, correspondingly. As a result, the drastic changes in the two dimensional (2-D) Fourier filter transform (FFT) power spectra along with the surface morphology evolution also can be similarly witnessed in Fig. 2(a-2)–(d-2). The symmetric bright spot indicating random distributed height drastically shrunk into a small spot caused by the reduction in the height distribution along with the vertical size increase. In brief, during the annealing between 500 and 700 °C the surface morphology underwent a drastic evolution from the flat Au thin film with only few nanometer modulation to the irregular Au nano-mounds with several hundred nanometers in height due to the enhanced surface diffusion as a function of temperature, which also can be equally observed with large-scale scanning electron microscopy (SEM) images in Fig S4.



Figure 4. Initial stages of the fabrication of Au nanostructure annealed at 500 and 600 °C with a DA of 15 nm. (a,b) AFM side-views of $10 \times 10 \mu m^2$. (a-1)–(b-1) Enlarged AFM side-views of $3 \times 3 \mu m^2$. (a-2)–(b-2) Cross sectional line-profiles. (a-3)–(b-3) FFT power spectra. (c,d) SEM images ($18.2 \times 13.9 \mu m^2$) of the samples annealed at 500 °C and 600 °C.

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Similar evolution of the irregular Au nano-mounds can be observed on quartz substrate³⁴, various GaAs^{35,36}, sapphire³⁷ and soft polymeric substrates³⁸.

Figure 3 shows the evolution of the self-assembled irregular Au nano-mounds with 15 nm DA on 4H-SiC (0001) controlled by the variation of the AT between 500 and 750 °C. The detailed surface morphology changes at the initial stage of the Au thin film agglomeration are presented in Fig. 4 and S5. Similar to the 8nm Au deposition, Au adatoms gradually aggregated and developed into the isolated irregular nano-mounds with the incremental variation of AT. Meanwhile, the formation of pinholes and granules was simultaneously witnessed during the evolution of Au nano-mounds, as shown with SEM images in Fig. 3. More specifically, as shown in Fig. 3(a), hillocks were formed with diameters of several micrometers or even larger than 10 micrometers at 500 °C possibly because of the thermal expansion of Au films. The hillocks appeared to the prior process of the pinhole formation at increased thermal energy, and the hillocks would eventually evolve into the pinholes with the formation of nano-mounds subsequently, as mentioned. At 600 °C of annealing, with the increased thermal energy, the number of hillocks and pinholes were further increased, as clearly shown in Fig. 3(b). Also, the granules started forming on the hillocks due to the tendency of the Au film to release the compressive stress, which resulted from the higher thermal coefficient of the Au film than the substrate. The hillock formation was also observed with the Au films on Y₂O₃-doped ZrO₂ (YSZ)^{39,40}. Finally, when the AT was reached 750 °C, all the Au structures agglomerated into the isolated Au nano-mounds with a uniform distribution owing to the enhanced diffusion of Au adatoms as shown in Fig. 3(c). At the stage of the pinholes formation, with the temperature increase, the size of the pinholes noticeably extended because of the more drastic agglomeration, as shown with the AFM side-views, line-profiles and top-views and SEM images in Fig. 4 and S5. Figure 5 shows the elemental analysis of the irregularly connected Au nano-mounds by the energy-dispersive X-ray spectroscopy (EDS). As shown by the combined EDS phase map in Fig. 5(b),



Figure 5. Elemental analysis of self-assembled Au nanostructures on 4H-SiC (0001) by the energy-dispersive X-ray spectroscopy (EDS). (a) SEM image of the sample with the 8 nm DA annealed at 600 °C.
(b) Combined EDS phase map of Au (yellow) and Si (red). (c) Line-profiles of element counts of Si (green) and Au (blue) denoted with the yellow line in the SEM image. (d,e) 3-D top-view phase maps of Si and Au.

the Au (yellow color) and Si phases (red color) were clearly matched the surface morphology shown by the SEM image in Fig 5(a). The Si (green line) can be observed everywhere on the surface, whereas, Au counts (blue line) mainly distributed along the nano-mounds, suggesting the agglomeration occurred with the limited diffusion, as shown by line-profiles indicating by the yellow line in the enlarged image in Fig. 5(c). Similarly, more noticeable counts of Au existed at the area with nano-mounds, and rest area were full of the Si counts, as shown in Fig. 5(d-e).

Figure 6 shows the transition phase between the Au nano-mounds and Au nano-crystals at a higher AT range between 750 and 950 °C with a DA of 15 nm on 4H-SiC (0001). The corresponding AFM top-views and SEM images are shown in Fig. 7, S6 and S7 respectively. Generally, Au nanostructure fabrication was quite sensitive to the AT, namely, above a certain AT, the hexagonal Au nano-crystals can be successfully synthesized. The equilibrium shape of NPs can be decided by the Wulff construction resulted from the orientation dependence of surface energy^{41,42}, and thus, with each fixed growth condition the shape of NPs tend to minimize the surface energy within a certain volume. The face-centered-cubic (fcc) materials, such as Au, have a tendency to truncate facets to reduce the surface energy and consequently, with the accumulation of Au adatoms, the anisotropy gradually appears more obviously. Thus, as soon as reaching the critical volume, the facet truncation can happen along each {111} and {100}, finally resulting in the formation of the hexagonal nano-crystals⁴¹. In more detail, at 750 °C, the agglomeration (Au nano-mounds separation) was still dominant due to the insufficient diffusion, which can be evidenced with some elongated nano-mounds (phase I), as shown in Figs 6(a),(a-1) and 7. Being providing with a sufficient thermal energy at 850 °C, the nano-mounds gradually separated into Au nano-crystals with



Figure 6. Self-assembled Au nano-crystals fabricated at various ATs between 750 and 950 °C with a DA of 15 nm. (a-c) AFM side-views of $10 \times 10 \,\mu$ m². (a-1)–(c-1) $3 \times 3 \,\mu$ m². (a-2)–(c-2) Cross sectional line-profiles. (a-3)–(b-3) FFT power spectra.

a noticeable size increase, as shown in Fig. 6(b),(b-1). As mentioned, owing to the anisotropic surface energy, the Au nano-crystals were turned into the hexagonal shape as shown in Fig. 6(a-1) (phase II). When the AT reached 950 °C, the hexagonal nano-crystals can still be fabricated with a slight decrease in both the size and the density, as shown in Fig. 6(c),(c-1). The size reduction at a higher surface temperature associated with the density decrease is against the general trend of surface diffusion, and this can be possibly due to the enhanced evaporation of nanoscale Au as a function of AT, which can similarly witnessed on Si⁴³, MgO, SrTiO₃, and Al₂O₃⁴⁴. To specify the size and density evolution, the average height (AH), lateral diameter (LD) and average density (AD) are summarized in Fig. 7(a,b) and Table SII, which can be divided into two phases: phase I between 750 and 850 °C and phase II above 850 °C.





The size and density evolution can also be clearly witnessed with SEM images in a larger scale, as shown in Fig. 7(c-e). Owing to the further enhanced diffusion energy, the AH kept increasing $\times 1.48$ times and as a result of the compact aggregation of Au adatoms, the LD decreased by 17%, whereas, the AD increased $\times 1.39$ times in phase I. The AD increase with an increase temperature from 750 to 850 °C can be due to the phase transition from the Au mounds to the hexagonal NPs. In phase II at 950 °C, the AD further decreased 47.4% due to the enhanced surface diffusion as expected, clearly shown in Fig. 7(b,e). However, both the AH and LD decreased 2.82% and 3.03% respectively at 950 °C likely due to the nanoscale Au evaporation as discussed. Accordingly, the R_{RMS} and SAR initially increased with the growth of nano-crystals in vertical size, and went down due to the decreased size and density of the nano-crystals, as shown in Fig. S6 and Table SII. Similarly, the bright spot in the 2-D FFT power spectra radically shrunk into smaller ones with a hexagonal pattern, suggesting the formation of the Au nanocrystals in Fig. 6(a-3)-(c-3). The self-assembled nano-crystal formation can be also witnessed with the elementary analysis shown in Fig. 8. As clearly shown by phase map in Fig. 8(b), the Au (yellow) distributes among the NPs, which means the shape transition happened with the formation of the Au NPs. As a result, the Au can be only detected in the area with NPs, whereas, the Si evenly exists in the areas with/without NPs, as shown with line-profiles in Fig. 8(c). Similarly, the Si and C K α peaks can be equally witnessed in the both locations, whereas, the M α 1 peak at 2.123 KeV only occurred in the area with NPs, as shown in Fig. 8(d-e). The Au NPs are equally evidenced with the pillars indicating higher concentration in 3-D side-views of the Au map, which matched the holes in the Si map, as shown in Fig. 8(f,g).

Figure 9 shows the evolution of the tiny dome-shaped Au NPs with a relatively low DA of 3 nm on 4H-SiC (0001) by the variation of AT between 300 and 900 °C. Figure S8 presents the corresponding samples with AFM top-views. Generally, with the increased AT, the tiny dome-shaped Au NPs were fabricated instantly after the annealing, and developed in size and density without the shape transition, which can be described with the Volmer-Weber growth model^{26,27}. As mentioned, Au adatoms can only freely aggregated with a limited l_D in the pinholes perforated by the voids. Provided that the Au thin film with the DA of 3 nm is much thinner than 8 and 15 nm, the perforation can immediately happen at the initial stage with much less thermal energy. In addition, given that the bonding energy between Au atoms (E_{Au}) was stronger than the bonding energy between Au adatoms can spontaneously nucleate to form the three-dimensional (3-D) islands (NPs). On the other hand, as discussed, the equilibrium shape can be decided by the surface energy for each orientation^{41,42}, and with a small volume, the surface energy can still sustain isotropic, which can result in the smooth Au thin film as soon as treated with



Figure 8. EDS phase maps and spectra of the Au nano-crystals fabricated at 900 °C with the 15 nm DA on 4H-SiC (0001). (a) SEM image of $40(x) \times 30(y) \mu m^2$. (b) Combined phase map of Au (yellow) and Si (red). (c) Line-profiles of element counts of Si (green) and Au (blue). (d,e) Corresponding EDS spectra from the red box (d) and blue (e). (f,g) 3-D top-view phase maps of Si and Au.

sufficient thermal energy from 300 to 500 °C, as shown in Fig. 9(a,b). Accordingly, the surface modulation increased from ~1 nm to ~5 nm, as shown with line-profiles in Fig. 9(a-1),(b-1). Meanwhile, being providing with $E_{Au} > E_{l}$, at elevated temperatures, the Au islands with larger boundaries tend to absorb more surrounding adatoms and merge smaller ones to form bigger in order to minimize the surface energy. Therefore, between 500 and 900 °C, with the enhanced l_D , the dome-shaped Au NPs gradually increased in size at a expense of the Au NPs density, as shown in Fig. 9(c,d), which resulted in the slight increase of the surface modulation indicated by the line-profiles in Fig. 9(c-1),(d-1). As shown in Fig. S8(b)–S8(d), throughout the whole evolution, the tiny dome-shaped NPs were fabricated uniformly with a packed density at each AT, which results the symmetric bright spots instead of the irregular pattern in the 2-D FFT power spectra, as shown in Fig. 9(b-2)–(d-2). The evolved surface morphology can also be witnessed with the increase of the R_{RMS}: Initially, the R_{RMS} drastically increased from 0.7 to 2 nm owing to the Au NPs formation, and subsequently, the R_{RMS} gradually increased as a function of the AT, as summarized in Fig. S8(e) and Table SIII. In short, the dome-shaped can be fabricated above 500 °C, and with the increase of Au NPs increased with decreased density, which can be a conventional



Figure 9. Evolution of the self-assembled Au nano-particles with a DA of 3 nm fabricated between 300 and 900 °C on 4H-SiC (0001). (a–d) AFM side-views $1 \times 1 \mu m^2$. (a-1)–(d-1) Cross-sectional line-profiles acquired from the white lines in the corresponding AFM side-views. (a-2)–(d-2) 2-D FFT power spectra.

behavior for metallic NPs on various substrate, such as Ag NPs on the GaN and sapphire^{45,46}, Au and Ga NPs on the GaAs^{47,48}, and Au NPs on the Si⁴⁹. Figure S9 shows the EDS spectra of the sample with 3 and 15 nm DAs annealed at 800 °C. As shown in Fig. S9(a) and S9(b), the increased DA can be evidenced with nearly 5 times higher counts in the M α 1 (2.123 keV) peaks of Au than that of 3 nm DA, as similarly witnessed with the L α 1 (9.711 keV) peaks, shown in Fig. S9(a-2) and S9(b-2).

Conclusions

In summary, the systematical investigation on the evolution of the self-assembled Au nanostructures on N-type 4H-SiC (0001) controlled by varying the annealing temperature (AT) between 300 and 950 °C was successfully demonstrate with various deposition amounts (DAs): 3, 8, 15 nm. At higher DAs (8 and 15 nm), with the increased AT, the drastic morphology evolution of Au nanostructures was observed into two phases: (I) Au nano-mounds, and (II) hexagonal Au nano-crystals. Below 700 °C, the Au nano-mounds gradually formed and developed in the pinholes perforated by the nucleated voids as a function of the AT, which was discussed as a dewetting process in conjunction with the DLA model. Meanwhile, the hillocks were fabricated at the lower AT range between 500 and 600 °C, caused by the thermal expansion of Au film. Above 750 °C, with a sufficient thermal energy, the hexagonal nano-crystals can be successfully fabricated on 4H-SiC (0001) as a result of anisotropic distribution of surface energy caused by the increased volume. Finally, the size of Au NPs started to decrease above 850 °C likely due to the nanoscale dependent evaporation of Au nanocrystals. For the samples with 3 nm DA, the tiny dome-shaped Au NPs were fabricated based on the Volmer-Weber growth model without the formation of the irregular nano-mounds. With a small volume, the distribution of surface energy of the Au NPs was still isotropic, which eventually resulted in the dome shape rather than the polyhedral shape.

Methods

In this experiment, the annealing temperature (AT) effect was investigated with 3, 8 and 15 nm deposition amounts (DAs) by the variation of annealing temperature in the pulsed laser deposition (PLD) system. Epi-ready N-type 4H-SiC (0001) substrate was ~250 μ m thick with an off-axis of $\pm 0.1^{\circ}$ from the Technology and Devices International (TDI, USA). Prior to the growth, samples were treated with a chemical cleaning in the hydrofluoric acid (49.0-51.0%) solution for 10 minutes and subsequently flushed with the deionization (DI) water for three times. For each growth, samples were mounted on an Inconel holder with indium solder for a good thermal conduction of samples and degassed at 700 °C for 30 min under a chamber vacuum below 1×10^{-4} Torr. After degassing, the surface was quite flat without any contaminants as confirmed by the morphological and optical characterizations in Fig. S1 and S2. Subsequently, 3, 8 and 15 nm-thick Au thin films were deposited on the sample respectively in a plasma lion-coater at a growth rate of 0.05 nm/s with the ionization current of 3 mA below the vacuum of 1×10^{-1} Torr. To systematically investigate the AT effect, with the fixed DA and annealing duration, samples were systematically annealed at various ATs of 300, 500, 600, 700, 750, 800, 900, 950 °C with a ramping rate of 2 °C/s by a halogen lamp. After reaching each target substrate temperature, the samples were dwelled there for 450s to ensure the uniformity of the Au nanostructures, and all the annealing process was strictly controlled with a computer-operated recipes in the PLD system under 1×10^{-4} Torr. To minimize the effect of the Ostwald ripening⁵⁰, the temperature was immediately quenched down to the ambient temperature after each growth. For the morphological characterization, an atomic force microscope (AFM) was utilized for smaller area scanning with the non-contact (tapping) mode. AFM tips (NSC16/AIBS, μ masch) employed were with a radius of less than 10 nm curvature, made of Si etching. The tips were $17-21\mu m$ long with the spring constant of $\sim 42 \text{ N/m}$ and resonant frequency of ~330kHz. The cantilevers of the tips were back-side coated with ~30nm Al to enhance the laser reflection. The same batch of tips were hired in order to diminish the tip effect for the consistency of the analysis and characterization. A scanning electron microcopy (SEM) was employed for larger area scanning. The acquired original data was processed and analyzed with the XEI software (Park Systems) to generate the AFM images, cross-sectional line-profiles, Fourier filter transform (FFT) power spectra, root-mean-squared (RMS) roughness, size and density plots of the Au NPs. The FFT power spectrum was obtained by converting the height distribution from the spatial domain to the frequency domain by the Fourier filter transform. Additionally, the energy-dispersive X-ray spectroscopy (EDS) with the spectral mode (Thermo Fisher Noran System 7) was employed for the elemental analysis and phase mappings under vacuum. For the optical characterization, the Raman spectrum was excited by a CW diode-pumped solid-state (DPPS) laser of a wavelength of 532 ± 1 nm with an output power of 120 mWand was received with a TE cooled CCD detector.

References

- 1. Frenzel, H. et al. Recent Progress on ZnO-Based Metal-Semiconductor Field-Effect Transistors and Their Application in Transparent Integrated Circuits. Adv. Mater. 22, 5332–5349 (2010).
- Fiorenza, P. et al. Comparative study of gate oxide in 4H-SiC lateral MOSFETs subjected to post-deposition-annealing in N₂O and POCl₃. Appl. Phys. A 115, 333–339 (2014).
- Okamoto, D. et al. Improved inversion channel mobility in 4H-SiC MOSFETs on Si face utilizing phosphorus-doped gate oxide. IEEE Electron Device Letters 31, 710–712 (2010).
- 4. Fuchs, F. *et al.* Silicon carbide light-emitting diode as a prospective room temperature source for single photons. *Scientific Report* **3**, 1637 (2013).
- 5. Lee, K. K. *et al.* N-channel MOSFETs fabricated on homoepitaxy-grown 3C-SiC films. *IEEE Electron Device Letters* 24, 466–468 (2003).
- 6. Persson, C. & Lindefelt, U. Detailed band structure for 3C-, 2H-, 4H-, 6H-SiC, and Si around the fundamental band gap. *Phys. Rev. B* 54, 10257–10260 (1996).
- 7. Strupinski, W. et al. Graphene epitaxy by chemical vapor deposition on SiC. Nano Lett. 11, 1786-1791 (2011).
- Srivastava, L. N., He, G. & Feenstra, R. M. Comparison of graphene formation on C-face and Si-face SiC (0001) surfaces. Phys. Rev. B 82, 235406 (2010).
- Katayama, I. et al. Ultrafast dynamics of surface-enhanced Raman scattering due to Au nanostructures. Nano Lett. 11, 2648–2654 (2011).
- Chen, S.-C. *et al.* Toward Omnidirectional Light Absorption by Plasmonic Effect for High-Efficiency Flexible Nonvacuum Cu(In, Ga)Se₂ Thin Film Solar Cells. ACS Nano 8, 9341–9348 (2014).
- 11. Lodewijks, K. et al. Boosting the figure-of-merit of LSPR-based refractive index sensing by phase-sensitive measurements. Nano Lett. 12, 1655–1659 (2012).
- 12. Han, S.-T. et al. Layer-by-Layer-Assembled Reduced Graphene Oxide/Gold Nanoparticle Hybrid Double-Floating-Gate Structure for Low-Voltage Flexible Flash Memory. Adv. Mater. 25, 872–877 (2013).
- Zhang, Q. & Wang, H. Facet-Dependent Catalytic Activities of Au Nanoparticles Enclosed by High-Index Facets. ACS Catal. 4, 4027–4033 (2014).
- 14. Kirkeminde, A. *et al.* Surface-passivated plasmonic nano-pyramids for bulk heterojunction solar cell photocurrent enhancement. *Nanoscale* **4**, 4421–4425 (2012).
- 15. Kedem, O., Tesler, A. B., Vaskevich, A. & Rubinstein, I. Sensitivity and optimization of localized surface plasmon resonance transducers. ACS Nano 5, 748–760 (2011).
- 16. Hvolbæk, B. et al. Catalytic activity of Au nanoparticles. Nano Today 2, 14-18 (2007).

- 17. Dedyulin, S. N. & Goncharova, L. V. Thermodynamic and kinetic control of the lateral Si wire growth. Appl. Phys. Lett. 104, 123103 (2014).
- Dasgupta, N. P. et al. 25th anniversary article: semiconductor nanowires-synthesis, characterization, and applications. Adv. Mater. 26, 2137–2184 (2014).
- 19. Rueda-Fonseca, P. et al. Structure and morphology in diffusion-driven growth of nanowires: the case of ZnTe. Nano Lett. 14, 1877–1883 (2014).
- Hamdou, B. et al. Thermoelectric Characterization of Bismuth Telluride Nanowires, Synthesized Via Catalytic Growth and Post-Annealing. Adv. Mater. 25, 239–244 (2013).
- 21. Schmidt, V., Senz, S. & Go1sele, U. Diameter-dependent growth direction of epitaxial silicon nanowires. *Nano Lett.* **5**, 931–935 (2005).
- 22. Jiang, N. et al. Understanding the True Shape of Au-Catalyzed GaAs Nanowires. Nano Lett. 14, 5865-5872 (2014).
- Gorji, M. S. & Cheong, K. Y. Au nanoparticles embedded at the interface of Al/4H-SiC Schottky contacts for current density enhancement. Appl. Phys. A 118, 315–325 (2015).
- 24. Han, S. Y., Lee, J.-L. & Pearton, S. J. Electrical properties of nanoscale Au contacts on 4H -SiC. J. Vac. Sci. Technol. B 27, 1870-1873 (2009).
- 25. Han, S. Y. & Lee, J.-L. Characteristics of SiC pillar-shaped nanostructure Schottky diode. J. Appl. Phys. 106, 023709 (2009).
- Kwon, J.-Y., Yoon, T.-S., Kim, K.-B. & Min, S.-H. Comparison of the agglomeration behavior of Au and Cu films sputter deposited on silicon dioxide. J. Appl. Phys. 93, 3270–3278 (2003).
- 27. Cheng, Y. & Beresford, R. Epitaxial Silicon Dots Self-Assembled on Aluminum Nitride/Si (111). Nano Lett. 13, 614-617 (2013).
- Sun, Y. et al. Probing Local Strain at MX₂-Metal Boundaries with Surface Plasmon-Enhanced Raman Scattering. Nano Lett. 14, 5329–5334 (2014).
- 29. Lu, W. & Komvopoulos, K. Implanted argon atoms as sensing probes of residual stress in ultrathin films. Appl. Phys. Lett. 76, 3206–3208 (2000).
- 30. Danielson, D. T., Sparacin, D. K., Michel, J. & Kimerling, L. C. Surface-energy-driven dewetting theory of silicon-on-insulator agglomeration. J. Appl. Phys. 100, 083507 (2006).
- 31. Ruffino, F. et al. Self-organization of gold nanoclusters on hexagonal SiC and SiO₂ surfaces. J. Appl. Phys. 101, 064306 (2007).
- 32. Huang, Y.-C. et al. Transformation of polymer nanofibers to nanospheres driven by the Rayleigh instability. ACS Appl. Mater. Interfaces 5, 3134–3142 (2013).
- Mukherjee, R. et al. Stability and dewetting of metal nanoparticle filled thin polymer films: control of instability length scale and dynamics. ACS Nano 4, 3709–3724 (2010).
- Sun, H., Yu, M., Sun, X., Wang, G. & Lian, J. Effective temperature sensing by irreversible morphology evolution of ultrathin gold island films. J. Phys. Chem. C 117, 3366–3373 (2013).
- Mao, S., Li, M.-Y., Kim, E.-S. & Lee, J. Effect of annealing temperature on the fabrication of self-assembled gold droplets on various type-B GaAs surfaces. *Cryst. Eng. Comm.* 16, 4390–4398 (2014).
- 36. Li, M.-Y., Mao, S., Kim, E.-S. & Lee, J. From the nucleation of wiggling Au nanostructures to the dome-shaped Au droplets on GaAs (111) A, (110), (100), and (111) B. Nanoscale Res. Lett. 9, 113 (2014).
- 37. Chen, C.-Y. et al. Fabrication of sphere-like Au nanoparticles on substrate with laser irradiation and their polarized localized surface plasmon behaviors. Opt. Express 17, 14186–14198 (2009).
- 38. Ruffino, F., Torrisi, V., Marletta, G. & Grimaldi, M. G. Growth morphology of nanoscale sputter-deposited Au films on amorphous soft polymeric substrates. *Appl. Phys. A* 103, 939–949 (2011).
- 39. Shaffir, E., Riess, I. & Kaplan, W. D. The mechanism of initial de-wetting and detachment of thin Au films on YSZ. *Acta Mater.* 57, 248–256 (2009).
- 40. Shaffir, E., Kauffmann, Y. & Riess, I. Void formation in gold films on yttrium-doped zirconia in the initial stage of de-wetting. *Acta Mater.* **79**, 59–65 (2014).
- Sau, T. K. & Rogach, A. L. Nonspherical noble metal nanoparticles: colloid-chemical synthesis and morphology control. Adv. Mater. 22, 1781–1804 (2010).
- 42. Vitos, L., Ruban, A.V., Skriver, H. L. & Kolla'r, J. The surface energy of metals. Surface Sci. 411, 186-202 (1998).
- Choi, W. K. et al. A Combined Top-Down and Bottom-Up Approach for Precise Placement of Metal Nanoparticles on Silicon. small 4, 330–333 (2008).
- 44. Meng, G. et al. Pressure-induced evaporation dynamics of gold nanoparticles on oxide substrate. Phys. Rev. E 87, 012405 (2013).
- 45. Jiang, S. *et al.* Resonant absorption and scattering suppression of localized surface plasmons in Ag particles on green LED. *Opt. Express* **21**, 12100–12110 (2013).
- Lazzari, R. & Jupille, J. Growth kinetics and size-dependent wetting of Ag/α-Al₂O₃ (0001) nanoparticles studied via the plasmonic response. *Nanotechnology* 23, 135707 (2012).
- Li, M.-Y., Mao, S., Kim, E.-S. & Lee, J. Nucleation, transition, and maturing of the self-assembled Au droplets on various type-A GaAs substrates. J. Appl. Phys. 116, 084301 (2014).
- Jo, M., Mano, T. & Sakoda, K. Two-step formation of gallium droplets with high controllability of size and density. *Cryst. Growth Des.* 11, 4647–4651 (2011).
- 49. Mao, S., Li, M.-Y., Kim, E.-S. & Lee, J. Annealing temperature effect on self-assembled Au droplets on Si (111). *Nanoscale Res. Lett.* **8**, 525 (2013).
- Wettergren, K., Schweinberger, F. F., Deiana, D., Ridge, C. J., Crampton, A. S., Rötzer, M. D., Hansen, T. W., Zhdanov, V. P., Heiz, U. & Langhammer, C. High sintering resistance of size-selected platinum cluster catalysts by suppressed Ostwald ripening. *Nano Lett.* 14, 5803–5809 (2014).

Acknowledgments

Financial support from the National Research Foundation (NRF) of Korea (no. 2011-0030821 and 2013R1A1A1007118), and in part by the research grant of Kwangwoon University in 2015 is gratefully acknowledged.

Author Contributions

M.L., Q.Z., P.P. and M.S. carried out the experiments. M.L. and J.L. designed the experiments and testing methods. M.L., Q.Z., P.P., M.S., E.K. and J.L. participated in the analysis of data. M.L. and J.L. carried out the writing. All authors helped in drafting and read and approved the final manuscript.

Additional Information

Supplementary information accompanies this paper at http://www.nature.com/srep

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Li, M.-Y. *et al.* From the Au nano-clusters to the nanoparticles on 4H-SiC (0001). *Sci. Rep.* **5**, 13954; doi: 10.1038/srep13954 (2015).

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