



Structure Elucidation

How to cite: Angew. Chem. Int. Ed. 2021, 60, 21778-21783 International Edition: doi.org/10.1002/anie.202107032 German Edition: doi.org/10.1002/ange.202107032

Monitoring Dynamics, Structure, and Magnetism of Switchable Metal-Organic Frameworks via ¹H-Detected MAS NMR

Jan Blahut, Arthur L. Lejeune, Sebastian Ehrling, Irena Senkovska, Stefan Kaskel, Florian M. Wisser,* and Guido Pintacuda*

Abstract: We present a toolbox for the rapid characterisation of powdered samples of paramagnetic metal-organic frameworks at natural abundance by ¹H-detected solid-state NMR. Very fast MAS rates at room and cryogenic temperatures and a set of tailored radiofrequency irradiation schemes help overcome the sensitivity and resolution limits often associated with the characterisation of MOF materials. We demonstrate the approach on DUT-8(Ni), a framework containing Ni²⁺ paddle-wheel units which can exist in two markedly different architectures. Resolved ¹H and ¹³C resonances of organic linkers are detected and assigned in few hours with only 1-2 mg of sample at natural isotopic abundance, and used to rapidly extract information on structure and local internal dynamics of the assemblies, as well as to elucidate the metal electronic properties over an extended temperature range. The experiments disclose new possibilities for describing local and global structural changes and correlating them to electronic and magnetic properties of the assemblies.

Crystalline porous coordination polymers or metal-organic frameworks (MOFs) represent a versatile family of materials with efficiently tunable properties by a combination of various metal-based nodes with an inexhaustible plethora of organic ligands. [1-4] The possibility to control the pore size and shape by design, as well as facile material functionalisation make MOFs attractive for adsorption-based applications. Following the seminal report by Kitagawa, [5] a large class of MOFs features highly flexible architectures, capable of undergoing structural changes dependent on the presence of guest molecules. [6] These guest-dependent transitions affect, in turn, the optical and magnetic properties of the materials, opening new perspectives for the use of MOFs in catalysis,^[7] gas separation, [8-10] or sensing. [11-13]

Atomic-level characterisation of MOFs represents the essential link capable of explaining and controlling the macroscopic properties of these efficient materials in terms of their microscopic structures and dynamics.^[14-17] Given the difficulties associated with the growth of single crystals and their manipulation during guest adsorption, diffraction methods are significantly limited in this area. [18] Solid-state nuclear magnetic resonance (NMR) provides a powerful alternative, capable of interrogating the local structures around metals, the configurations of organic ligands, their dynamics and the environments experienced by guest molecules.^[19-28] Despite the tremendous recent technological and methodological advances, however, this technique still falls short in many aspects, as it requires large sample quantities, long experimental measurements and data analyses, and often complex isotopic labelling schemes.

The direct detection of ¹H signals is the most obvious way to circumvent these difficulties by enhancing the sensitivity of the NMR experiments, due to the high gyromagnetic ratio γ of ¹H spins, their natural isotopic abundance, and their ubiquitous presence in MOF organic constituents. In solids, however, the very same properties of ¹H spins form a dense network of strong dipolar couplings, which broadens ¹H resonances and hampers their constructive use. ^[29] The direct detection of ¹H resonances in MOFs has been the object of notable proof-of-concept reports, [30-34] but spectra often lacked the resolution required for the complete

[*] Dr. J. Blahut, A. L. Lejeune, Dr. G. Pintacuda

Centre de Résonance Magnétique Nucléaire à Très Hauts Champs (UMR 5082 CNRS, ENS Lyon, UCBL)

Université de Lyon

69100 Villeurbanne (France)

E-mail: guido.pintacuda@ens-lyon.fr

NMR Laboratory, Faculty of Science, Charles University Hlavova 8, 12842 Prague (Czech Republic)

A. L. Lejeune IFP Energies Nouvelles 69360 Solaize (France)

Dr. S. Ehrling, Dr. I. Senkovska, Prof. Dr. S. Kaskel Chair of Inorganic Chemistry I Technische Universität Dresden

01069 Dresden (Germany)

Dr. F. M. Wisser

IRCELYON (UMR 5256 CNRS, UCBL), Université de Lyon

69100 Villeurbanne (France)

and

Institute of Inorganic Chemistry University of Regensburg

93040 Regensburg (Germany) E-mail: florian.wisser@chemie.uni-regensburg.de

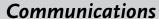
Dr. S. Ehrling

Present address: 3P Instruments GmbH & Co. KG Rudolf-Diesel-Strasse 12, 85235 Odelzhausen (Germany)

Supporting information and the ORCID identification number(s) for the author(s) of this article can be found under:

https://doi.org/10.1002/anie.202107032.

© 2021 The Authors. Angewandte Chemie International Edition published by Wiley-VCH GmbH. This is an open access article under the terms of the Creative Commons Attribution License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited.







identification of individual ¹H sites. The problem is alleviated by fast magic-angle spinning (MAS) at high magnetic fields, which weakens the ¹H dipolar coupling networks and sharpens their NMR lines.[35,36] However, MAS rates achievable with commercially available NMR probes today are not sufficient to yield fully resolved ¹H spectra of MOFs, and the only site-resolved ¹H-detected NMR structural studies relied on ¹H spin dilution involving (complex and costly) partial replacement of ¹Hs with ²Hs, which significantly limits the advantages of the approach.[37]

MAS NMR of MOFs containing open-shell metal ions carries an additional layer of experimental difficulties. [38-42] In these samples, the hyperfine interactions between NMRactive nuclei and the unpaired electrons of the paramagnetic metals introduce important problems in the acquisition and interpretation of the spectra.^[43] At the same time, however, these interactions encode important information of the geometry and the electronic structure of the metal environments. Also in this case, fast MAS is a crucial factor for the detection of nuclei in close proximity to a metal centre. [44,45] However, as some paramagnetic effects are larger on high-γ ¹H nuclei, MAS NMR of paramagnetic MOFs has up to now focused preferentially on low-γ nuclei (¹³C, ¹⁵N, ¹⁷O, ²H, etc.) which suffer from low natural abundance with only limited applicability of isotope labelling.

Here, we show that with the help of very fast (60 kHz) MAS rates and a set of tailored radiofrequency (RF) irradiation schemes, [46-49] paramagnetic effects can become an asset for the rapid characterisation of powdered samples of open-shell MOFs at natural abundance by ¹H-detected solidstate NMR. This includes sensitive detection and assignment of resolved ¹H and ¹³C resonances of organic linkers via 2D ¹H-¹H and ¹H-¹³C correlations, analysis of structure and local internal dynamics, as well as elucidation of the metal electronic properties over an extended temperature range.

We showcase this methodology on a prototypical switchable pillared layer MOF offering a variable metal substitution chemistry, namely DUT-8(M) (DUT = Dresden University of Technology), [50] composed of (M²⁺)₂ paddle-wheel (PW) units linked through 2,6-naphthalenedicarboxylate (NDC) into 2D layers, which are in turn interconnected by 1,4-diazabicyclo-[2.2.2]octane (DABCO) pillars (Scheme 1). We notably focus on DUT-8(Ni) containing (Ni²⁺)₂ PWs, which exhibits strong hysteresis during guest adsorption/desorption between an open-pore (op) including a guest molecule (DMF in this

Scheme 1. DUT-8(M) with labelling used in the text.

study) and a closed-pore (cp) structure. This reversible, guestdependent structural change is associated with a significant reorientation of the DABCO pillars with respect to the PW plane (the Ni···Ni-N angle changes from approx. 178° to 154°), which maintains the Ni-Ni distance almost unaltered but strongly deforms the square-pyramidal coordination of the nickel and produces an unpreceded decrease (by 60%) of the unit-cell volume.[50,51]

Figure 1 A,B shows 1D ¹H spin-echo spectra of **cp** and **op** DUT-8(Ni) powders, respectively, acquired on a 700 MHz spectrometer at 60 kHz MAS. These spectra illustrate the challenge associated with ¹H detection and paramagnetic NMR in these paddle-wheel-based frameworks. Due to the combined effect of ¹H-¹H homonuclear interactions and hyperfine coupling with the unpaired electrons of the paramagnetic metal ions, the spectra contain highly overlapped centre-band lines (inset for Figure 1B) and a pronounced pattern of rotational sidebands. The richness of paramagnetic NMR effects nonetheless provides a way to monitor the

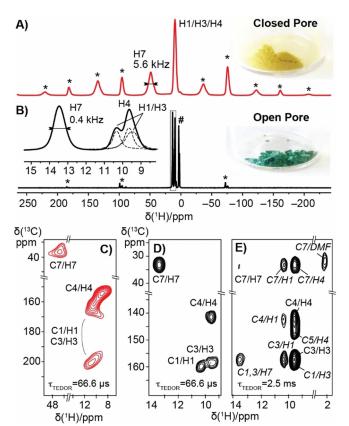


Figure 1. Solid-state MAS ¹H NMR spectra of A) cp and B) op DUT-8(Ni) with expansion including deconvolution of the NDC spectral region (dashed line). Spectra acquired with rotor-synchronised spinecho detection at 60 kHz MAS (700 MHz, 325 K of sample temperature). Rotational sidebands and residual solvent signals are indicated by an asterisk (*) and a hash mark (#), respectively. The displayed line-width of the H7 resonance corresponds to the full-width at half maximum. For the labelling of resonances, see Scheme 1. C-E) ¹H-¹³C HSQC-TEDOR spectra of **cp** (C, red contours) and **op** DUT-8(Ni) (D,E, black contours), acquired with short (66.6 µs, C and D) and long (2.5 ms, E) τ_{TEDOR} recoupling. In (E) long-range correlations are labelled in italics.







dynamic structural transformation between the two states associated with distinct spectral responses. Both spectra are characterised by an unresolved or partially resolved set of lines (approx. 9.5 ppm) from the NDC ligands (H1/H3/H4) and shifted downfield a baseline-resolved signal from the ¹Hs in the DABCO pillars (H7). The latter, however, resonates at 13.5 ppm in the **op** form and at 48.8 ppm in the **cp** structure, which immediately reveals significantly different paramagnetic contributions, and thus different electronic structures and spin-density distributions, in the two samples. Moreover, ¹H longitudinal and transverse relaxation rates are similarly enhanced in the two spectra, but linewidths feature a ten-fold difference in the two samples (e.g. 440 and 5600 Hz for the DABCO ¹Hs (H7) in the **op** and **cp** sample, respectively, see Table S1). This effect is a clear signature of the presence of a large inhomogeneous broadening in the latter spectrum, compatible with a more diverse structure with frequent irregularities for the **cp** sample indicated by the powder X-ray diffraction analysis.^[51] Finally, the very different chemical shift anisotropies (about twice as high for the cp structure) indicate largely different magnetic susceptibilities in the two samples, in line with an antiferromagnetic coupling between neighboring Ni²⁺ ions in the **op** structure (see below).

An unambiguous identification of all individual isotropic ¹H resonances can be achieved by extending the ¹H spin-echo spectra into 2D ¹H-¹³C correlations. While in diamagnetic samples the transfer of polarisation from protons to nearby hetero-nuclei is performed routinely by cross-polarisation, pulsed techniques such as the Transferred Echo DOuble Resonance recoupling (TEDOR)^[52] allow a more efficient magnetisation transfer for nuclei exposed to strong paramagnetic shift and fast paramagnetic-induced relaxation. [45] TEDOR uses rotor-synchronised π pulses to reintroduce, during an interval τ_{TEDOR} , the $^{1}H-^{13}C$ dipolar couplings otherwise averaged out by MAS, and is usually acquired as a ¹³C-detected experiment at intermediate MAS rates in the 30 kHz range. Thanks to the improved resolution and the shorter recoupling periods possible at 60 kHz MAS, we adopted here a ¹H-detected variant (Heteronuclear Single-Quantum Correlation through TEDOR or HSQC-TEDOR, Figure S1). This scheme was recently proposed for the characterisation of the coordination sphere of a paramagnetic metalloenzyme, [49] and provides comparable sensitivity to the ¹³C-detected spectrum acquired at slower MAS, with a fivefold reduction in the sample volume. The uniformly short T_1 relaxation times, together with the absence of the requirement for high-power RF irradiations during the experiment, [44] allow here the use of very short recycle delays between acquisitions (5 ms). Combined with the high sensitivity of ¹H detection, this produces a sensitivity boost, and a 2D correlation can be acquired within 30 minutes on 1–2 mg of sample at natural $^{13}\mathrm{C}$ abundance. Depending on the τ_{TEDOR} recoupling period employed, ¹H-¹³C correlations can be observed either over short distances corresponding to a single bond (Figure 1 C,D), or over longer distances, including correlations between ¹Hs and quaternary ¹³Cs (Figure 1E). These TEDOR spectra resolve and assign the individual frequencies of all the ¹H and ¹³C nuclei in the sample.

As illustrated in Figure 2, the sensitive detection of resolved ¹H resonances represents a powerful handle for the characterisation of the framework architecture and its local dynamics at atomic resolution. Figure 2 A,B shows an example of 2D ¹H-¹H correlations obtained with Radio-Frequency-Driven Recoupling (RFDR), [54] as well as the buildup curve of the cross-signals between NDC and DABCO ligands. This experiment plays a similar role as the ¹H-¹H NOESY in solution, where the dependence of the volume of ¹H-¹H cross-signals with the mixing time as a function of ¹H-¹H distances forms the basis of NMR structure determination. The faster initial slope in case of **cp** geometry encodes shorter distances between ¹H nuclei of NDC and DABCO (2.1 Å vs. 2.6 Å for **cp** and **op** forms, respectively), and thus reflects a more compact assembly architecture with respect to the op sample.

In parallel, the cross-signal intensity in the TEDOR correlations is exquisitely sensitive to local dynamics, providing a picture which is independent of static distortions and disorder, hardly accessible with diffraction data. When the duration of the recoupling interval au_{TEDOR} is incremented, the intensity of each correlation signal experiences a modulation whose frequency is proportional to the effective ¹H₋¹³C dipolar coupling constant. Such a constant can be extracted by directly fitting the oscillatory build-up in the time domain, and order parameters S_{ord} can be calculated as a ratio between

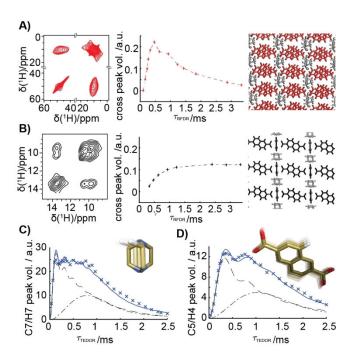


Figure 2. A,B) $^1H-^1H$ RFDR correlation spectra for cp (red) and op (black) DUT-8(Ni) acquired with τ_{RFDR} = 0.3 ms, and $H_{DABCO} \rightarrow H_{NDC}$ build-up curves. Intensities were scaled to a maximum of diagonal H_{DABCO} signal equals one. Note that the accumulation of pulse imperfections (more pronounced for the broader spectrum of the cp sample) is responsible for the decrease observed for the build-up curve of **cp** in $\tau_{RFDR} > 0.5$ ms. C,D) $^{1}H^{-13}C$ TEDOR build-up curves (x) for heteronuclear H7-C7 (C) and H4-C5 (D) correlations in the op DUT-8(Ni). Data fit (solid line) was obtained using an analytical model consisting of two relaxation-truncated Bessel functions (black lines; [53] see SI for further details).



Communications



measured effective coupling values and those calculated from H and C distances in the X-ray structure with DFT-reoptimised proton positions. Figure 2 C,D shows the experimental build-up curves in the **op** form for a ${}^{1}H-{}^{13}C$ correlation within the NDC ligand (H4-C5) and one in the DABCO moiety (H7-C7; note that given the symmetry of the two organic groups, each correlation encodes a proximal and a distal ¹H-¹³C dipolar coupling, see SI). By fitting these profiles, order parameters S_{ord} of 0.7 and 0.25, respectively, are obtained for the two spin pairs. The former corresponds to a highly restricted motion of the rigid NDC ligand, [55,56] while the latter reports on very fast internal dynamics associated with the DABCO rotation around its N-N axis, in agreement with previous IR results,[51] as well as with ²H MAS NMR and ¹H T₁ relaxation studies on analogous diamagnetic materials.[57,58] This description of ligand dynamics is essential for understanding the tunability of guest-adsorption properties.[14,15,56,59] The different mobility of the NDC and DABCO ligands seems to be independent of the nature of the metal centre, as demonstrated by equivalent measurements on the analogue DUT-8(Cu) op (see Section 9 of the

The NDC and DABCO molecules are the key building blocks connecting the global structural features of the periodic framework during the pore-closing process to a local deformation of the metal coordination geometry, and thus ultimately to the electronic fine structure of the PW node. In the following, we shed light on this connection, which has remained elusive despite tremendous efforts through the standard characterisation methods. The possibility to detect resolved paramagnetically shifted ¹H signals indeed allows to measure their temperature dependence, which in turn is a direct reporter of the fine details of the electronic structure of the metal ions. The very recent development of MAS NMR probes capable of fast spinning at cryogenic temperatures^[60–62] allows to extend the sensitivity benefits in a broad temperature range between 100 K and 305 K. In this way, MAS NMR can be used in a practical way to monitor the temperature dependence of resolved ¹H shifts thus providing a microscopic insight into the magnetic properties of the system.

Magnetometric measurements and computational models revealed an antiferromagnetic coupling between the Ni^{II} centres in the cp form, resulting in a diamagnetic ground state $(S = |S_1 - S_2| = 0)$ and in a thermal admixture of lowlying paramagnetic excited states (S = 1, 2).^[51,63] As mentioned above, this configuration is mirrored in the smaller shift anisotropy of the ¹H spectrum and the modest paramagnetic shift of the DABCO ¹Hs. In line with this, we observed fundamental differences in the temperature behaviour of DABCO ¹H NMR shifts in the two samples (Figure 3). While in the cp form the paramagnetic shift decreases with rising temperature in agreement with Curie law, in the op sample the shift has the opposite ("anti-Curie") temperature behaviour. The wide temperature range allows a quantitative analysis of the data with a simple Heisenberg model (see SI), which allows to extract both the Heisenberg coupling constant (J) within a Ni···Ni ion pair and the hyperfine coupling constant between Ni and DABCO ¹Hs (A_M). ^[64,65]

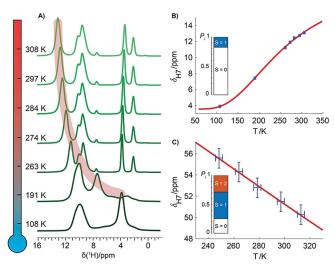


Figure 3. Temperature dependence H7 NMR shift of the A,B) op and C) cp DUT-8(Ni) sample. In (A) ¹H spectra (40 kHz MAS) of op DUT-8(Ni) are reported, while in (B),(C) peak positions (blue + with error bars) are fitted by a Heisenberg model (red line). The bar-plots indicate the Boltzmann population of electronic spin levels at 300 K for $J_{\text{NiNi}} = -266$ and -52 cm⁻¹ obtained from the fit in (B) and (C), respectively.

The fitted temperature dependence curves show an excellent agreement with the experimental data (RMSD = 0.034 and 0.036 ppm for **cp** and **op**, respectively), despite some simplifications in the model, which notably neglects the contributions of pseudo-contact shifts as well as hyperfine effects from distant metals.

The value of the Ni···Ni Heisenberg coupling obtained for the **op** structure $(J_{op} = -266(13) \text{ cm}^{-1} \text{ is significantly higher})$ than thermal energy over the temperature range explored experimentally) explains the almost purely diamagnetic NMR shift (3.85 ppm) at cryogenic temperatures. The fitted value of $J_{\rm on}$ is in good agreement with the value determined through magnetometric measurements (-242 cm^{-1}) , [51] and it matches the value calculated for a periodic crystal model by Seifert and co-workers (-277 cm^{-1}) as well. [63] This reinforces the evidence of a high crystallinity for the op form. In contrast to magnetometric measurements, [51] the temperature dependence of the ¹H NMR shifts provides access to the Heisenberg coupling constant also for the cp form. Here, the experimentally fitted $(J_{cp} = -52(34) \text{ cm}^{-1})$ value reveals a weaker Heisenberg coupling, in line with the longer Ni···Ni distance (2.653(5) Å and 2.735(9) Å for the **op** and **cp** structure, respectively),^[51] and with a more localised unpaired-electron density $(A_{\rm M} = 1.4(2))$ and $(A_{\rm M} = 1.4(2))$ for **op** and **cp**, respectively). Interestingly, the experimental $|J_{cp}|$ value is significantly lower than that calculated for a crystalline system (-108 cm⁻¹), but fits very well with the one calculated for a truncated molecular Ni···Ni dimer model (-59 cm⁻¹).^[63] This result suggests the presence of defects (i.e. the absence of DABCO units) in the cp form. These results demonstrate that ¹H-detected NMR provides the simultaneous elucidation of the bonding character and of the magnetic (super)exchange at the metal node, two essential elements for







understanding phase transitions in switchable MOFs, both at a microscopic and macroscopic level.

In summary, we demonstrated a step forward in the analysis of paramagnetic MOFs, enabled by high MAS rates and carefully tuned pulse sequences. This allows to assign all proton and carbon resonances in paramagnetic DUT-8(Ni), to determine local dynamics of the organic moieties next to the paramagnetic centre, and to disclose electronic properties of the nickel ions. For the first time the Heisenberg coupling of the closed-pore form became experimentally accessible, providing unprecedented insight into switching behaviour and node deformation. The experiments described above constitute an important addition to the conventional repertoire of solid-state NMR, notably allowing a reduction in sample quantities (often a bottleneck in the characterisation of new materials) and experiment times for recording NMR spectra and increasing the amount of information associated with them. Advantageously, with paramagnetic solid-state NMR structural and electronic information can be obtained simultaneously under the same experimental conditions and on the same sample, thus avoiding any impact of different sample preparation on the measured properties and minimising the risk of sample degradation during multiple preparation steps. Due to the high gain in sensitivity, these methodologies will pave the way for elucidating the molecular structure of guest molecules, their interaction with the node and defects as well as resulting changes in the MOF properties in the future, also under in situ conditions, that is, during guest molecule adsorption.^[27] The simultaneous accessibility of this information is of utmost importance for understanding and tuning the materials response during catalytic or adsorptiondriven processes.

Finally, the methodologies are not limited to MOFs, but can also be implemented in the characterisation of other advanced materials containing paramagnetic centres, including silica materials or porous organic polymers. [66-68] They also disclose a wealth of new possibilities for the atomic-level description of materials properties, and we anticipate that paramagnetic solid-state NMR will cover in this context a bridging and highly complementary role between lowresolution techniques reporting on bulk properties (such as magnetometry, gas sorption, and calorimetry) and those reporting on the periodic arrangement of the molecules (PXRD, electron and neutron diffraction).

Acknowledgements

The work was supported by the European Research Council (ERC-2015-CoG GA 648974), as well as by the CNRS (IR-RMN FR3050). F.M.W. gratefully acknowledges financial support from the Deutsche Forschungsgemeinschaft (DFG, grant number WI 4721/1-1 & WI 4721/3-1) and from CNRS through Momentum 2018 excellence grant. S.E., I.S., and S.K. thank the DFG (FOR 2433 MOF Switches, Project No. 279409724) for financial support. The authors thank Dr. Ladislav Benda for fruitful discussions regarding TEDOR build-up. Open access funding enabled and organized by Projekt DEAL.

Conflict of Interest

The authors declare no conflict of interest.

Keywords: DUT-8(Ni) · fast magic-angle spinning · metalorganic framework · paramagnetic NMR · proton detection

- [1] S. Kitagawa, R. Kitaura, S. I. Noro, Angew. Chem. Int. Ed. 2004, 43, 2334-2375; Angew. Chem. 2004, 116, 2388-2430.
- A. Bavykina, N. Kolobov, I. S. Khan, J. A. Bau, A. Ramirez, J. Gascon, Chem. Rev. 2020, 120, 8468-8535.
- [3] Ü. Kökçam-Demir, A. Goldman, L. Esrafili, M. Gharib, A. Morsali, O. Weingart, C. Janiak, Chem. Soc. Rev. 2020, 49, 2751 -
- [4] S. L. Griffin, N. R. Champness, Coord. Chem. Rev. 2020, 414, 213295
- [5] S. Kitagawa, M. Kondo, Bull. Chem. Soc. Jpn. 1998, 71, 1739-1753.
- [6] G. Férey, C. Serre, Chem. Soc. Rev. 2009, 38, 1380.
- [7] S. Yuan, L. Zou, H. Li, Y.-P. Chen, J. Qin, Q. Zhang, W. Lu, M. B. Hall, H.-C. Zhou, Angew. Chem. Int. Ed. 2016, 55, 10776-10780; Angew. Chem. 2016, 128, 10934-10938.
- [8] E. J. Carrington, C. A. McAnally, A. J. Fletcher, S. P. Thompson, M. Warren, L. Brammer, Nat. Chem. 2017, 9, 882-889.
- [9] M. K. Taylor, T. Runčevski, J. Oktawiec, J. E. Bachman, R. L. Siegelman, H. Jiang, J. A. Mason, J. D. Tarver, J. R. Long, J. Am. Chem. Soc. 2018, 140, 10324-10331.
- [10] J. Y. Kim, J. Park, J. Ha, M. Jung, D. Wallacher, A. Franz, R. Balderas-Xicohténcatl, M. Hirscher, S. G. Kang, J. T. Park, et al., J. Am. Chem. Soc. 2020, 142, 13278-13282.
- [11] N. Yanai, K. Kitayama, Y. Hijikata, H. Sato, R. Matsuda, Y. Kubota, M. Takata, M. Mizuno, T. Uemura, S. Kitagawa, Nat. Mater. 2011, 10, 787-793.
- [12] P. Freund, I. Senkovska, S. Kaskel, ACS Appl. Mater. Interfaces **2017**, 9, 43782 – 43789.
- [13] X.-Y. Dong, H.-L. Huang, J.-Y. Wang, H.-Y. Li, S.-Q. Zang, Chem. Mater. 2018, 30, 2160-2167.
- [14] S. Horike, R. Matsuda, D. Tanaka, S. Matsubara, M. Mizuno, K. Endo, S. Kitagawa, Angew. Chem. Int. Ed. 2006, 45, 7226-7230; Angew. Chem. 2006, 118, 7384-7388.
- [15] M. Inukai, M. Tamura, S. Horike, M. Higuchi, S. Kitagawa, K. Nakamura, Angew. Chem. Int. Ed. 2018, 57, 8687 - 8690; Angew. Chem. 2018, 130, 8823-8826.
- [16] V. Bon, E. Brunner, A. Pöppl, S. Kaskel, Adv. Funct. Mater. 2020, 30, 1907847.
- [17] E. Ploetz, H. Engelke, U. Lächelt, S. Wuttke, Adv. Funct. Mater. **2020**, *30*, 1909062
- [18] R. I. Walton, F. Millange in The Chemistry of Metal-Organic Frameworks: Synthesis Characterization, and Applications (Ed.: S. Kaskel), Wiley-VCH, Weinheim, 2016, pp. 729-764.
- [19] H. Hoffmann, M. Debowski, P. Müller, S. Paasch, I. Senkovska, S. Kaskel, E. Brunner, *Materials* **2012**, *5*, 2537 – 2572.
- [20] X. Kong, H. Deng, F. Yan, J. Kim, J. A. Swisher, B. Smit, O. M. Yaghi, J. A. Reimer, Science 2013, 341, 882-885.
- [21] A. M. Fracaroli, H. Furukawa, M. Suzuki, M. Dodd, S. Okajima, F. Gándara, J. A. Reimer, O. M. Yaghi, J. Am. Chem. Soc. 2014, 136, 8863 - 8866.
- [22] T. M. McDonald, J. A. Mason, X. Kong, E. D. Bloch, D. Gygi, A. Dani, V. Crocellà, F. Giordanino, S. O. Odoh, W. S. Drisdell, et al., Nature 2015, 519, 303-308.
- [23] V. J. Witherspoon, J. Xu, J. A. Reimer, Chem. Rev. 2018, 118, 10033 - 10048.
- [24] B. E. G. Lucier, S. Chen, Y. Huang, Acc. Chem. Res. 2018, 51,



Communications



- [25] J. T. Damron, J. Ma, R. Kurz, K. Saalwächter, A. J. Matzger, A. Ramamoorthy, Angew. Chem. Int. Ed. 2018, 57, 8678-8681; Angew. Chem. 2018, 130, 8814-8817.
- [26] C. Martineau-Corcos, Curr. Opin. Colloid Interface Sci. 2018, 33, 35 - 43.
- [27] C. Paula, D. Wisser, M. Rangus, W. Schwieger, M. Hovestadt, M. Kriesten, K. Vanatalu, A. Oss, M.-L. Org, A. Samoson, et al., J. Phys. Chem. C 2020, 124, 19136-19145.
- [28] E. Brunner, M. Rauche, Chem. Sci. 2020, 11, 4297 4304.
- [29] M. M. Maricq, J. S. Waugh, J. Chem. Phys. 1979, 70, 3300 3316.
- [30] A. Krajnc, T. Kos, N. Zabukovec Logar, G. Mali, Angew. Chem. Int. Ed. 2015, 54, 10535 – 10538; Angew. Chem. 2015, 127, 10681 –
- [31] J. Tang, S. Li, Y. Chu, Y. Xiao, J. Xu, F. Deng, J. Phys. Chem. C **2019**, 123, 24062 - 24070.
- [32] J. Tang, S. Li, Y. Su, Y. Chu, J. Xu, F. Deng, J. Phys. Chem. C **2020**, 124, 17640 – 17647.
- [33] A. C. Forse, P. J. Milner, J.-H. Lee, H. N. Redfearn, J. Oktawiec, R. L. Siegelman, J. D. Martell, B. Dinakar, L. B. Porter-Zasada, M. I. Gonzalez, et al., J. Am. Chem. Soc. 2018, 140, 18016-
- [34] V. Y. Mao, P. J. Milner, J. Lee, A. C. Forse, E. J. Kim, R. L. Siegelman, C. M. McGuirk, L. B. Porter-Zasada, J. B. Neaton, J. A. Reimer, et al., Angew. Chem. Int. Ed. 2020, 59, 19468-19477; Angew. Chem. 2020, 132, 19636-19645.
- [35] A. Böckmann, M. Ernst, B. H. Meier, J. Magn. Reson. 2015, 253, 71 - 79.
- [36] L. B. Andreas, T. Le Marchand, K. Jaudzems, G. Pintacuda, J. Magn. Reson. 2015, 253, 36-49.
- [37] J. Xu, V. V. Terskikh, Y. Chu, A. Zheng, Y. Huang, Chem. Mater. **2015**, 27, 3306 – 3316.
- [38] D. M. Dawson, L. E. Jamieson, M. I. H. Mohideen, A. C. McKinlay, I. A. Smellie, R. Cadou, N. S. Keddie, R. E. Morris, S. E. Ashbrook, Phys. Chem. Chem. Phys. 2013, 15, 919-929.
- [39] T. Wittmann, A. Mondal, C. B. L. Tschense, J. J. Wittmann, O. Klimm, R. Siegel, B. Corzilius, B. Weber, M. Kaupp, J. Senker, J. Am. Chem. Soc. 2018, 140, 2135-2144.
- [40] L. N. McHugh, M. J. McPherson, L. J. McCormick, S. A. Morris, P. S. Wheatley, S. J. Teat, D. McKay, D. M. Dawson, C. E. F. Sansome, S. E. Ashbrook, et al., Nat. Chem. 2018, 10, 1096-
- [41] D. M. Dawson, C. E. F. Sansome, L. N. McHugh, M. J. McPherson, L. J. McCormick McPherson, R. E. Morris, S. E. Ashbrook, Solid State Nucl. Magn. Reson. 2019, 101, 44-50.
- [42] M. Rauche, S. Ehrling, S. Krause, I. Senkovska, S. Kaskel, E. Brunner, Chem. Commun. 2019, 55, 9140-9143.
- [43] A. J. Pell, G. Pintacuda, C. P. Grey, Prog. Nucl. Magn. Reson. Spectrosc. **2019**, 111, 1–271.
- [44] Y. Ishii, N. P. Wickramasinghe, S. Chimon, J. Am. Chem. Soc. **2003**, 125, 3438 – 3439.
- [45] G. Kervern, G. Pintacuda, Y. Zhang, E. Oldfield, C. Roukoss, E. Kuntz, E. Herdtweck, J.-M. Basset, S. Cadars, A. Lesage, et al., J. Am. Chem. Soc. 2006, 128, 13545 – 13552.
- [46] G. Kervern, G. Pintacuda, L. Emsley, Chem. Phys. Lett. 2007, 435, 157 - 162.
- [47] R. J. Clément, A. J. Pell, D. S. Middlemiss, F. C. Strobridge, J. K. Miller, M. S. Whittingham, L. Emsley, C. P. Grey, G. Pintacuda, J. Am. Chem. Soc. 2012, 134, 17178-17185.
- [48] A. Bertarello, T. Schubeis, C. Fuccio, E. Ravera, M. Fragai, G. Parigi, L. Emsley, G. Pintacuda, C. Luchinat, *Inorg. Chem.* 2017, 56, 6624-6629.

- [49] A. Bertarello, L. Benda, K. J. Sanders, A. J. Pell, M. J. Knight, V. Pelmenschikov, L. Gonnelli, I. C. Felli, M. Kaupp, L. Emsley, et al., J. Am. Chem. Soc. 2020, 142, 16757-16765.
- [50] N. Klein, C. Herzog, M. Sabo, I. Senkovska, J. Getzschmann, S. Paasch, M. R. Lohe, E. Brunner, S. Kaskel, Phys. Chem. Chem. Phys. 2010, 12, 11778-11784.
- [51] S. Ehrling, I. Senkovska, V. Bon, J. D. Evans, P. Petkov, Y. Krupskaya, V. Kataev, T. Wulf, A. Krylov, A. Vtyurin, et al., J. Mater. Chem. A 2019, 7, 21459-21475.
- [52] A. W. Hing, S. Vega, J. Schaefer, J. Magn. Reson. 1992, 96, 205 –
- [53] T. Gullion in Modern Magnetic Resonance (Ed.: G. A. Webb), Springer Netherlands, Dordrecht, 2008, pp. 713-718.
- [54] A. E. Bennett, R. G. Griffin, J. H. Ok, S. Vega, J. Chem. Phys. 1992, 96, 8624-8627.
- [55] P. S. Petkov, V. Bon, C. L. Hobday, A. B. Kuc, P. Melix, S. Kaskel, T. Düren, T. Heine, Phys. Chem. Chem. Phys. 2019, 21, 674-680.
- [56] S. Ehrling, E. Reynolds, V. Bon, I. Senkovska, T. Gorelik, M. Rauche, M. Mendt, M. Weiss, E. Brunner, U. Kaiser, et al., ChemRxiv 2020, https://doi.org/10.26434/chemrxiv.12326165.
- [57] A. Gonzalez-Nelson, F.-X. Coudert, M. van der Veen, Nanomaterials 2019, 9, 330.
- [58] N. C. Burtch, A. Torres-Knoop, G. S. Foo, J. Leisen, C. Sievers, B. Ensing, D. Dubbeldam, K. S. Walton, J. Phys. Chem. Lett. 2015, 6,812-816.
- [59] S. Ehrling, M. Mendt, I. Senkovska, J. D. Evans, V. Bon, P. Petkov, C. Ehrling, F. Walenszus, A. Pöppl, S. Kaskel, Chem. Mater. 2020, 32, 5670-5681.
- [60] S. R. Chaudhari, P. Berruyer, D. Gajan, C. Reiter, F. Engelke, D. L. Silverio, C. Copéret, M. Lelli, A. Lesage, L. Emsley, *Phys.* Chem. Chem. Phys. 2016, 18, 10616-10622.
- [61] K. Jaudzems, A. Bertarello, S. R. Chaudhari, A. Pica, D. Cala-De Paepe, E. Barbet-Massin, A. J. Pell, I. Akopjana, S. Kotelovica, D. Gajan, et al., Angew. Chem. Int. Ed. 2018, 57, 7458-7462; Angew. Chem. 2018, 130, 7580-7584.
- [62] P. Berruyer, S. Björgvinsdóttir, A. Bertarello, G. Stevanato, Y. Rao, G. Karthikeyan, G. Casano, O. Ouari, M. Lelli, C. Reiter, et al., J. Phys. Chem. Lett. 2020, 11, 8386-8391.
- [63] K. Trepte, S. Schwalbe, G. Seifert, Phys. Chem. Chem. Phys. **2015**. 17. 17122 - 17129.
- [64] I. Bertini, C. Luchinat, G. Parigi, E. Ravera, in Solution NMR of Paramagnetic Molecules, Elsevier, Amsterdam, 2017, pp. 347 -
- [65] T. Soda, Y. Kitagawa, T. Onishi, Y. Takano, Y. Shigeta, H. Nagao, Y. Yoshioka, K. Yamaguchi, Chem. Phys. Lett. 2000, 319, 223 – 230.
- [66] H. Phan, T. S. Herng, D. Wang, X. Li, W. Zeng, J. Ding, K. P. Loh, A. T. Shen Wee, J. Wu, Chem 2019, 5, 1223-1234.
- [67] I. B. Moroz, A. Lund, M. Kaushik, L. Severy, D. Gajan, A. Fedorov, A. Lesage, C. Copéret, ACS Catal. 2019, 9, 7476-7485.
- Y. Mohr, M. Alves-Favaro, R. Rajapaksha, G. Hisler, A. Ranscht, P. Samanta, C. Lorentz, M. Duguet, C. Mellot-Draznieks, E. A. Quadrelli, et al., ACS Catal. 2021, 11, 3507-3515.

Manuscript received: May 26, 2021 Revised manuscript received: July 12, 2021 Accepted manuscript online: July 17, 2021 Version of record online: September 2, 2021