



REVIEW ARTICLE

Ferroptosis: An emerging therapeutic opportunity for cancer

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Abstract Ferroptosis, a new form of non-apoptotic, regulated cell death characterized by iron dependency and lipid peroxidation, is involved in many pathological conditions such as neurodegenerative diseases, heart ischemia/reperfusion injury, acute renal failure, and cancer. While metabolic dysfunctions can lead to excessive lipid peroxidation culminating in ferroptotic cell death, glutathione peroxidase 4 (GPX4) resides in the center of a network that functions to prevent lipid hydroperoxides from accumulation, thereby suppressing ferroptosis. Indeed, RSL3 and other small-molecule GPX4 inhibitors can induce ferroptosis in not only cultured cancer cells but also tumor xenografts implanted in mice. Similarly, erastin and other system Xc⁻ inhibitors can deplete intracellular glutathione required for GPX4 function, leading to lipid peroxidation and ferroptosis. As therapy-resistant cancer cells are sensitive to GPX4-targeted therapeutic regimens, the agents capable of inducing ferroptosis hold great promises to improve current cancer therapy. This review will outline the molecular basis of ferroptosis, but focus on the strategies and the agents developed in recent years for therapeutic induction of ferroptosis. The potentials of these ferroptosis-inducing agents, which include system Xc⁻ inhibitors, GPX4 inhibitors, and iron-based nanoparticles, in cancer therapy will be subsequently discussed.

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Introduction

Despite great advances in the molecular understanding of cancer, cancer remains as a leading cause of human death worldwide. A major challenge faced with the current cancer therapy is cancer recurrence owing to re-growth of therapy-resistant malignant cells. While many cancer therapeutics intend to kill cancer cells by inducing apoptotic cell death, resistant or persister cells often find ways to evade apoptosis.¹ It is thus thought that therapeutic agents that can induce cancer cells to undergo non-apoptotic forms of death would hold promises to improve the clinical outcomes of current cancer therapy.² Indeed, recent studies have demonstrated that therapy-resistant cancer cells are more sensitive to agents that can induce ferroptosis, an iron-dependent, non-apoptotic form of cell death.^{3,4}

Ferroptosis has not been characterized until 2012 when erastin - a small molecule thought to selectively inhibit the growth of H-Ras-transformed BJ fibroblasts⁵ - was found to induce cancer cells to undergo a new type of iron-dependent, programmed death characterized by shrinking mitochondria with decreased crista and condensed/ruptured membranes.⁶ Ferroptosis lacks the hallmarks of apoptosis (e.g., chromatin condensation and caspase activation), and is also distinct from necroptosis as knockdown of necroptosis mediators RIPK1 and RIPK3 does not impair ferroptosis.⁶ Rather, ferroptosis is the final fate of cells with excessive lipid peroxidation, and can be regulated by multiple metabolic pathways that intersect to maintain cellular redox hemostasis.⁷⁻¹⁰ Although ferroptosis was first characterized in transformed cells, this non-apoptotic cell death is also involved in other pathological conditions, such as neurodegenerative diseases, heart and hepatic ischemia/reperfusion injury, acute renal failure, and drug-induced hepatotoxicity.⁹ Intriguingly, while ferroptosis induced by loss of the major ferroptosis-regulatory gene *Slc7a11* leads to suppression of growth of pancreatic ductal adenocarcinoma in mice,¹¹ conventional cancer therapeutics, such as cisplatin,¹² irradiation,¹³⁻¹⁵ and checkpoint blockage,^{13,16} can also induce ferroptosis in cancer cells. Moreover, a number of ferroptosis inducers (FINs) have been developed and shown to effectively kill cancer cells in various preclinical animal models.^{10,17} Therefore, ferroptosis-inducing agents have potentials to become a new class of therapeutic agents for treating cancers, particularly advanced, therapy-resistant cancers. This article will outline the strategies for inducing ferroptosis, and then focus on the diverse array of ferroptosis-inducing agents and their potentials in cancer therapy.

The molecular basis of ferroptosis

Ferroptosis is defined as a form of programmed cell death that is dependent on iron and caused by excessive peroxidation of membrane phospholipids.⁶ Lipid peroxidation occurs when free radicals, such as the hydroxyl radical (OH·) produced by H₂O₂-catalyzed oxidation of soluble ferrous ions (Fe²⁺) (i.e., the Fenton reaction), react with polyunsaturated fatty acids (PUFAs) inserted in the membrane phospholipids (L in Fig. 1A), yielding lipid radicals

(L·) (Fig. 1A). Lipid radicals then react with oxygen (O₂) to yield lipid peroxy radicals (LOO·), which in turn react with an adjacent PUFA-phospholipid to produce a lipid hydroperoxide (LOOH) as well as a new lipid radical (Fig. 1A). While the new lipid radical can initiate a new lipid radical reaction, LOOH can also be converted into an alkoxyl radical (LO·) in the presence of Fe²⁺ and subsequently oxidize an adjacent PUFA-phospholipid to yield a lipid radical for another lipid radical reaction (Fig. 1A). This auto-amplifying chain reaction thus causes accumulation of lipid peroxides, leading to loss of membrane fluidity/integrity, damage of cellular proteins, and eventually ferroptotic cell death.¹⁸ As Fe²⁺ can provide free radicals to promote lipid peroxidation, overloading cells with iron by treating cells with ammonium sulfate ((NH₄)₂Fe(SO₄)₂) or iron chloride (FeCl₂) can induce ferroptosis.^{19,20} Likewise, heme and hemoglobin can be metabolized to Fe²⁺ through a reaction catalyzed by heme oxygenase 1 (HMOX1), thereby inducing cell death via ferroptosis.^{19,20} In addition to the Fe²⁺-dependent reactions, lipogenesis, e.g., arachidonate 5-lipoxygenase (ALOX5) can catalyze dioxygenation of PUFAs, yielding lipid hydroperoxides to promote ferroptosis²¹ (Fig. 1B).

While intracellular lipophilic anti-oxidants (e.g., vitamin E) can neutralize lipid radicals by donating electrons and break the lipid radical chain reactions, a major mechanism utilized by cells to prevent lipid peroxidation and subsequent ferroptosis is mediated by glutathione peroxidase 4 (GPX4).²² As a selenocysteine-containing proteins, GPX4 is the only enzyme that can reduce membrane lipid hydroperoxides (LOOH) into non-toxic lipid alcohols (LOH) at the expense of reduced glutathione (GSH)²² (Fig. 1B). It is thus not surprising that genetic depletion of GPX4 or inhibition of GPX4 activity by small molecules (e.g., (1S,3R)-RSL3, or RSL3 in short) results in excessive lipid peroxidation and subsequent ferroptotic cell death.²² RSL3 and other GPX4-binding compounds (e.g., ML210) belong to a major class of small molecules that can induce ferroptosis by inactivating the lipid peroxidase^{17,23} (Fig. 1B).

Because GPX4 mainly utilizes anti-oxidant GSH to reduce lipid hydroperoxides, the GPX4 activity is largely dependent on the intracellular GSH level. GSH is synthesized in the cytosol through two-step reactions that require cysteine. While cysteine can be obtained through reduction from cystine in the cytosol, uptake of cystine from the extracellular environment through system Xc⁻ serves as the rate-limiting step in providing cysteine for GSH synthesis in many cancer cells.²⁴ As a cystine/glutamate antiporter, system Xc⁻ is composed of a light-chain subunit (xCT) encoded by *SLC7A11* and a heavy-chain subunit (4F2hc) encoded by *SLC3A2*, and mediates the exchange of extracellular cystine and intracellular glutamate across the plasma membrane^{24,25} (Fig. 1B). In support of the important role of system Xc⁻-mediated cystine uptake in the prevention of ferroptosis, cystine deprivation,²⁶ or treatments with erastin, an inhibitor of system Xc⁻,²⁷ induces accumulation of lipid peroxides and subsequent ferroptosis. Therefore, by regulating cysteine availability and GSH synthesis, system Xc⁻ acts together with GPX4 to drive a pathway that serves to reduce excessive lipid peroxidation and suppress ferroptosis (Fig. 1B). The agents that can inhibit system Xc⁻ or limit cystine availability (e.g., erastin) are thus classified as

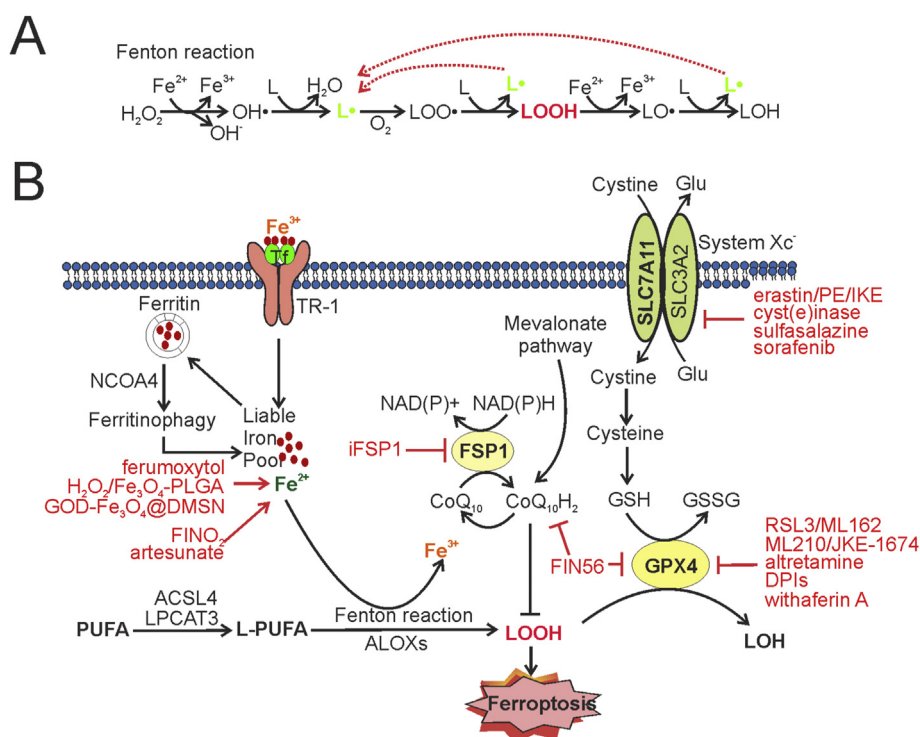


Figure 1 The molecular basis of ferroptosis and the strategies for therapeutic induction of ferroptosis. (A) The lipid radical chain reactions triggered by ferrous irons (Fe^{2+}) yield and amplify lipid radicals ($\text{L}\cdot$) for the production of lipid hydroperoxides (LOOH). Free radicals produced by Fenton reactions oxidize phospholipids to yield $\text{L}\cdot$, which in turn reacts with O_2 to generate lipid peroxy radicals ($\text{LOO}\cdot$) for subsequent production of LOOH. These reactions can re-generate lipid radicals leading to excessive lipid peroxidation if remaining unblocked. (B) While the increase of the liable iron pool caused by ferric import through transferrin (Tf) and its receptor (TR-1), or ferritin degradation via ferritinophagy, can promote lipid peroxidation and subsequent ferroptosis, two major mechanisms respectively mediated by GPX4 and FSP1 serve to prevent lipid hydroperoxides from accumulation thereby suppressing ferroptosis. GPX4 is the only peroxidase that can reduce lipid peroxides, and its activity is regulated by the intracellular GSH level. System Xc^- imports cystine for the production of cysteine and GSH, and thus plays a crucial role in the regulation of GPX4 activity. FSP1, on the other hand, may reduce CoQ_{10} for the suppression of lipid peroxidation. Multiple metabolic pathways, such as the mevalonate pathway and those for lipid synthesis or iron hemostasis, intersect to regulate ferroptosis. Accordingly, system Xc^- inhibitors (e.g., erastin), GPX4 inhibitors (e.g., RSL3), and iron-based nanoparticles (e.g., ferumoxytol) represent the 3 major classes of ferroptosis-inducing agents that hold great promises for cancer therapy.

a class of ferroptosis inducers that can indirectly inactivate GPX4 by depleting intracellular GSH^{10,17} (Fig. 1B).

Coenzyme Q_{10} (CoQ_{10}), also known as ubiquinone, is a potent intracellular lipophilic anti-oxidant that can be produced from isopentenyl-5-pyrophosphate, the end product of the mevalonate metabolic pathway. Ubiquinol ($\text{CoQ}_{10}\text{H}_2$), the reduced form of CoQ_{10} , can trap lipid peroxy radicals capable of catalyzing lipid peroxidation, thereby suppressing ferroptosis by preventing accumulation of lipid peroxides.²⁸ Ferroptosis suppressor protein 1 (FSP1), formerly known as apoptosis-inducing factor mitochondria-associated 2 (AIFM2), can reduce CoQ_{10} and catalyze regeneration of $\text{CoQ}_{10}\text{H}_2$ using NAD(P)H , thereby promoting $\text{CoQ}_{10}\text{H}_2$ -mediated suppression of lipid peroxidation.^{29,30} Indeed, genetic screens identified FSP1 as one of the top hits that can prevent ferroptosis induced by GPX4 inactivation.^{29,30} Although a recent study argues against the role of CoQ_{10} in FSP1-mediated ferroptosis suppression,³¹ FSP1 likely mediates an alternative pathway that suppresses ferroptosis independent of GPX4 (Fig. 1B).

Strategies for therapeutic induction of ferroptosis

Given that it is an iron-dependent process associated with excessive lipid peroxidation, ferroptosis can be induced by targeting the major mediators of the ferroptosis-suppressing pathways (e.g., GPX4 and system Xc^-), or increasing the liable iron pool (Fig. 1B). Indeed, a number of ferroptosis inducers (FINs) have been identified or developed that can either directly inhibit the GPX4 activity by binding to (e.g., RSL3, ML210) or depleting (e.g., FIN56) GPX4, or indirectly do so by inhibiting system Xc^- (e.g., erastin, sulfasalazine, and sorafenib) or depleting extracellular cystine/cysteine (i.e., cyst(e)inase)^{10,17} (Fig. 1B). In addition, agents that can cause iron overload (e.g., FeCl_2 and heme) or regulate iron oxidation (FINO₂) can also induce ferroptosis in cancer cells. As iron plays an important role in the regulation of redox hemostasis in normal cells, systemic administration of the agents regulatory for iron metabolism likely causes severe side effects. Although a number of

iron-based nanoparticles/nanomaterials have been developed that can deliver and release irons selectively at tumor sites by virtue of their dimensions and the defective tumor neovascularity – an effect often referred to as enhanced permeability and retention (EPR) effect - in animal models (Fig. 1B), the EPR effect is often compromised in human cancers, particularly in advanced cancers, due to their high heterogeneity and/or complexity.³² Therefore, the GPX4-mediated pathway remains as the ideal target for therapeutic induction of ferroptosis in human cancers. In support of this notion, overexpression of *SLC7A11*, the gene encoding the unique subunit (xCT) of system Xc⁻,^{24,25} often occurs in human cancers and is associated with reduced patient survival.¹¹

As genetic knockout of *Slc7a11* in mice does not result in lethality or obvious defects,^{33,34} system Xc⁻ inhibitors are expected to be well tolerable. Intriguingly, although T cell proliferation in culture is dependent on *SLC7A11* expression, system Xc⁻ deficiency does not appear to abolish anti-cancer immunity *in vivo*.³⁵ However, while the main function of system Xc⁻ is to import cystine and supply cysteine, cysteine can not only be synthesized through the transsulfuration pathway,³⁶ but also be imported through system ASC transporters such as ASCT1 and ASCT2.³⁷ Moreover, GPX4 can use protein thiols as donors of electrons to reduce lipid hydroperoxides in the absence of GSH.³⁸ Therefore, only cancer cells whose growth is largely dependent on cystine uptake, such as pancreatic cancer cell lines (MIA PaCa-2, PANC-1 and BxPC-3) and a subset of triple-negative breast cancer lines, are sensitive to system Xc⁻ inhibition. On the other hand, *Gpx 4* knockout mice are embryonic lethal,³⁹ suggesting the importance of defining therapeutic windows for safely administering GPX4 inhibitors in patients. Of note, although ferroptosis can be suppressed by FSP1, depletion or inhibition of FSP1 alone is not sufficient to induce ferroptosis, but rather sensitize cancer cells to undergo ferroptosis induced by GPX4 inactivation.^{29,30} In the following sections, we will summarize the pharmacological properties and the therapeutic potentials of major classes of ferroptosis-inducing agents (Figs. 1B and 2).

Agents depleting GSH (system xc⁻ inhibitors and cyst(e)inase)

Erastin and its derivatives

Erastin (2-[1-[4-[2-(4-Chlorophenoxy)acetyl]-1-piperazinyl]ethyl]-3-(2-ethoxyphenyl)-4(3H)-Quinazolinone) (Fig. 2) was first identified as a small molecule that selectively causes death of transformed cells expressing oncogenic HRAS, KRAS, or BRAF in a high-throughput drug screening.⁵ While subsequent studies failed to confirm such selectivity, erastin was shown to bind the outer mitochondrial membrane proteins - voltage-dependent anion channel 2 and 3 (VDAC2/3) - to induce oxidative stress-dependent, non-apoptotic cell death,⁴⁰ and was later identified as the first FIN that can inhibit system Xc⁻ and decrease the intracellular GSH level at low micromolar concentrations.⁶ Cancer cell lines originated from diffuse large B cell lymphoma and renal cell carcinoma cells are often sensitive to erastin-

induced ferroptosis.²² Erastin can also induce ferroptosis in head & neck, gastric, melanoma, lung cancer cells,^{41–44} and muscle-derived embryonal and alveolar rhabdomyosarcoma cells.⁴⁵

As the prototype of FINs, erastin has also been tested for its anti-cancer effects in combination with common therapeutic agents. For example, it was shown that erastin synergizes with cisplatin to induce death of HCT116 and A549 cells.⁴⁶ Likewise, low-cytotoxic doses of erastin significantly enhance the activities of cytarabine and doxorubicin, two first-line anticancer drugs for acute myeloid leukemia, in HL60 cells.⁴⁷ Erastin also enhances the radiosensitivity of non-small cell lung cancer (NSCLC) cells (e.g., A549-R and H460-R), partly because irradiation can induce *SLC7A11* expression and increase the system Xc⁻ activity.⁴⁸

However, erastin has intermediate water solubility (0.086 mM) and is metabolically liable, precluding its *in vivo* applications for cancer therapy. Piperazine erastin (PE) is an analog of erastin, which adds a piperazine moiety on the meta position of the aniline ring to improve water solubility (1.4 mM) while substituting the ethoxyl moiety with isopropoxy to improve metabolic stability (Fig. 2). Initial subcutaneous injection of 40 mg/ml of PE followed by intravenous PE injection (30 mg/mL) starting one day after implantation of HT1080 cells into athymic nude mice significantly inhibits the xenograft growth without overt toxicity in this prevention model.²² However, PE has a limited effect on the growth of established tumors, perhaps due to its intermediate potency.⁴⁹ Imidazole ketone erastin (IKE) is another erastin analog that replaces the piperazine moiety of PE with a carbonyl (Fig. 2), which can form a reversible covalent interaction with a lysine side chain in the binding site of a protein, resulting in more than 100-fold increase in potency.⁴⁹ IKE also adds an imidazole moiety to increase water solubility (1 mM at pH 3; 0.25 mM at pH 7) and the stability of the ketone (Fig. 2). Intraperitoneal injections of IKE (23 and 40 mg/kg) can significantly slow the tumor growth in a xenograft mouse model of diffuse large B cell lymphoma (DLBCL).⁵⁰ Tumor samples from IKE-treated mice express higher levels of ferroptosis markers (e.g., COX-2, MDA), demonstrating that IKE induces ferroptosis *in vivo*.⁵⁰ As IKE is more soluble under acidic conditions than neutral conditions, nanoparticles (NP) based on biocompatible and biodegradable PEG-PLGA di-block copolymers have also been tested for their utility as an IKE carrier system.⁵⁰ The PEG block was used to prevent clearance and prolong circulation lifetime while the PLGA block was used to provide a hydrophobic core to incorporate and sustainably release IKE. Although it does not improve the anti-tumor efficacy, the IKE NP formulation appears to exert less toxicity.⁵⁰ These results have demonstrated the potentials of erastin analogs in the treatment of cancer.

Sulfasalazine

Sulfasalazine (SAS, 2-Hydroxy-5-((4-((2-pyridinylamino)sulfonyl)phenyl)azo)benzoic acid) (Fig. 2) is a FDA-approved drug commonly used for the treatment of rheumatoid arthritis and inflammatory bowel disease.⁵¹

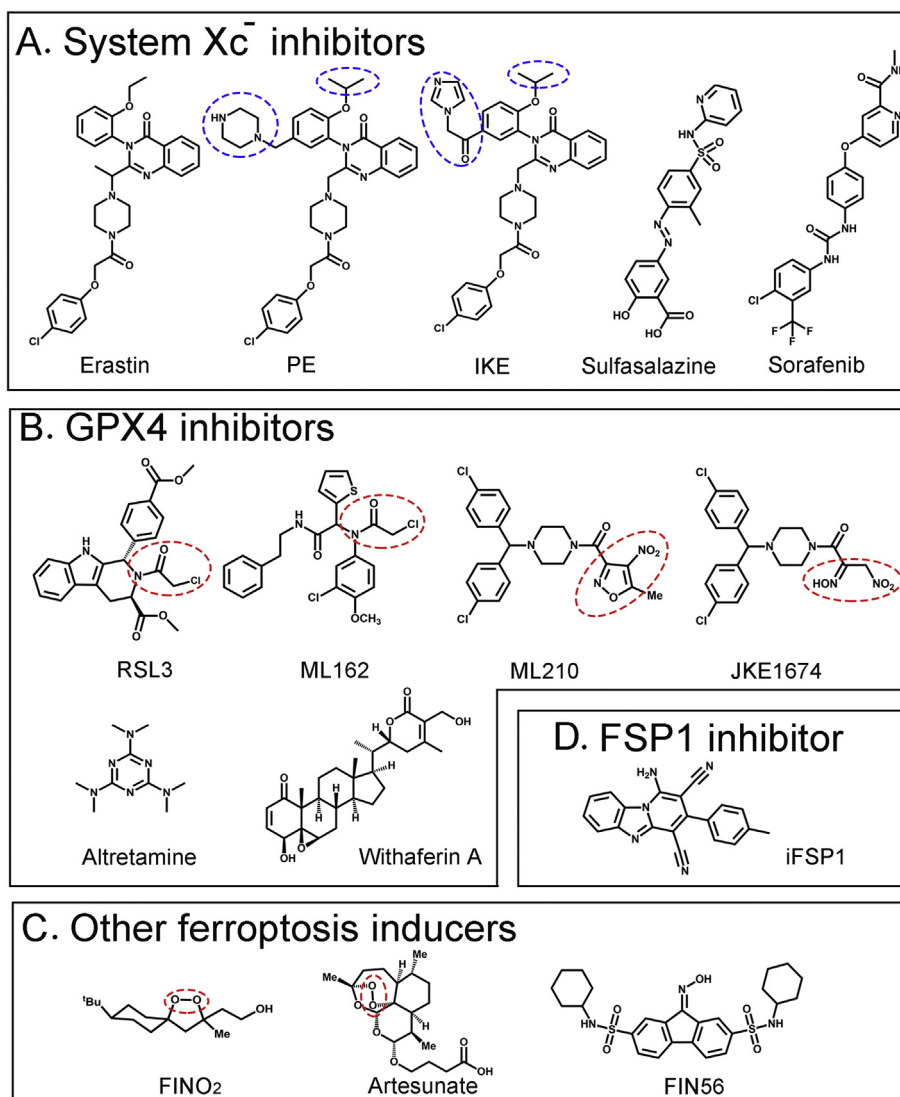


Figure 2 Chemical structures of representative ferroptosis-inducing small molecules. (A) System Xc⁻ inhibitors. PE, piperazine erastin; IKE, Imidazole ketone erastin. The groups distinct from erastin are indicated by blue dashed circles (B) GPX4 inhibitors. The moieties indicated by red dashed circles are proposed warheads, i.e., chloroacetamide for RSL3 and ML162, and nitroketoxime for ML210 and JKE1674 (C) Other ferroptosis inducers. The endoperoxide moiety is indicated by red circles (D) FSP1 inhibitor.

Taken by mouth, SAS is metabolized by intestinal bacteria into two active metabolites, sulfapyridine and 5-aminosalicylic acid. In a 2001 study, SAS, but not its colonic metabolites, was shown to inhibit cystine uptake at 0.3 mM, and consequently identified as a potent system Xc⁻ inhibitor that can markedly inhibit the growth of lymphoma without major toxicity to the hosts when administered intraperitoneally at a dose of 200 or 250 mg/kg in a rat xenograft model.⁵² SAS was later shown to induce ferroptosis, but this activity is much lower than that of erastin.^{6,27} SAS-induced cell death can be reversed by β-mercaptoethanol, a compound which can reduce extracellular cystine to cysteine and thus bypass the function of system Xc⁻,²⁷ confirming that SAS acts as a system Xc⁻ inhibitor to induce ferroptosis. SAS has been shown to inhibit the growth of several cancer cells, including prostate, small cell lung, liver, head & neck, and

triple-negative breast cancer cells.^{41,53–56} Similar to erastin, SAS can sensitize cancer cells to first-line therapies. For example, as human glioblastomas (GBM) express *SLC7A11*, intraperitoneal injection of SAS (30–40 mg/kg) synergizes with gamma knife radiosurgery to prolong survival of nude rats carrying GBM xenografts.⁵⁷ Likewise, while *SLC7A11* expresses in one-third of triple negative breast tumors, SAS increases the efficacy of carboplatin against these tumors.⁵⁶ Moreover, SAS can potentiate the cytotoxicity of cisplatin to colorectal, and head & neck cancer.^{41,58} However, clinical trials that test SAS efficacy in treatments of gliomas and lymphomas were unsuccessful,⁵⁹ likely due to its low potency and metabolic instability. SAS is rapidly metabolized through cleavage of the diazo bond. Analogs with an alkyne group replacing the diazo group were synthesized, but their potencies in inhibiting system Xc⁻ need improvement.⁶⁰

Sorafenib

Sorafenib (4-[4-[[[4-chloro-3-(trifluoromethyl)phenyl]amino]carbonyl]amino]phenoxy]-N-methyl-2-pyridinecarboxamide, BAY43-9006) (Fig. 2) is a broad-spectrum protein kinase inhibitor that can inhibit kinases including VEGFR, PDGFR, and Raf kinases, and is approved by FDA for the treatment of advanced renal cell carcinoma, advanced hepatocellular carcinoma, and thyroid carcinoma. While it remains uncertain whether the anti-cancer activity of sorafenib is due to its inhibition of kinases,⁶¹ sorafenib was found to induce a form of caspase-independent, but reactive oxygen species-dependent death in cancer cells.^{62,63} It was soon found that sorafenib is a potent system Xc^- inhibitor, and it acts alone to induce ferroptosis in a range of cancer cell lines, including lung, kidney, pancreatic, colon, and liver cancer cells.^{27,64,65} Unlike erastin, however, sorafenib causes non-ferroptotic death at higher concentrations and thus triggers ferroptosis only within a narrow concentration window. While it is the only known protein kinase inhibitor that can induce ferroptosis, sorafenib may inhibit an unknown kinase whose activity is necessary for system Xc^- activity, or bind to a non-kinase target regulatory for system Xc^- activity.²⁷

Cyst(e)inase

System Xc^- inhibitors induce ferroptosis by blocking cystine uptake and consequently limiting the supply of cysteine required for GSH synthesis. While it has been demonstrated that culturing cancer cells in cystine-free medium sufficiently induces ferroptosis,²⁶ an alternative strategy for decreasing the intracellular cysteine level in cancer cells is pharmacological depletion of extracellular cystine and cysteine by cyst(e)inase, an engineered cystathionine- γ -lyase (CGL) conjugated to methoxy PEG succinimidyl carboxyl ester.⁶⁶ The wildtype CGL can degrade both cysteine and cystine, but the kinetic rates are too slow to be relevant for clinical applications. The engineered enzyme has two substitutions (E59T and E339V) that largely increases the binding affinity to cysteine and cystine.⁶⁶ While the half-life of cyst(e)inase can reach to 25 h in mice, a single dose (8 mg/kg, i.p.) of cyst(e)inase results in a marked reduction of the blood cysteine/cystine level with no apparent toxicity in cynomolgus monkeys, indicating its potential application in cancer therapy.⁶⁶ Indeed, while cyst(e)inase potently decreases the viability of mouse (HMVP2) and human (PC3 and DU145) prostate and breast (MDA-MB-361) cancer cells, treating mice harboring xenografts from these cells with cyst(e)inase (50 and 100 mg/kg) every 4 days for 4 weeks dramatically inhibits the growth of tumors with no overt adverse effects.⁶⁶ Cyst(e)inase also prolongs the survival of genetically-engineered mice mimicking human chronic lymphocytic leukemia.⁶⁶ Based on the observation that the growth of pancreatic ductal adenocarcinoma (PDAC) is critically dependent on system Xc^- to import cystine and suppress ferroptosis, a recent study treated mice harboring established PDACs with cyst(e)inase, and showed that this treatment induces PDACs to undergo ferroptosis, resulting in tumor regression or stabilization.¹¹ Of note, this study also reveals that PDAC cells can use cysteine to synthesize coenzyme A and its derivative CoQ₁₀, two metabolites which

can suppress ferroptosis in PDAC cells.¹¹ Thus, system Xc^- inhibitors may also induce ferroptosis through a mechanism independent of GSH depletion.

Agents inhibiting GPX4 activity (GPX4 inhibitors)

RSL3 and ML162

RSL3 ((1S,3R)-Methyl 2-(2-chloroacetyl)-2,3,4,9-tetrahydro-1-[4-(methoxycarbonyl)phenyl]-1H-pyrido [3,4-b]indole-3-carboxylate) (Fig. 2) was first identified in a high-throughput screening as a compound that can selectively induce non-apoptotic, iron-dependent, oxidative cell death in transformed cells harboring activated H-Ras.⁶⁷ However, unlike erastin, RSL3-induced cell death is independent of VDAC2/VDAC3 and system Xc^- . Affinity purification experiments identifies GPX4 as a direct target of RSL3.²² RSL3 via its electrophilic chloroacetamide moiety covalently binds to the selenocysteine residue of GPX4, and thus directly inhibits the catalytic activity of the peroxidase in an irreversible manner.²¹ A number of cancer cell lines in the NCI60 panel (e.g., HOP-92, H522, CAKI-1, DU145 and PC3) are sensitive to RSL3, and nanomolar concentrations of RSL3 are often sufficient to decrease cell viability.⁶⁷ RSL3 fails to induce lipid peroxidation and ferroptosis in GPX4-knockout cells, indicating that RSL3-induced ferroptosis is due to its inhibition of GPX4.²² Interestingly, recent studies show that drug-tolerant persistent cells and therapy-resistant cancer cells with mesenchymal characteristics are sensitive to RSL3 and other GPX4 inhibitors.^{3,4} As drug-tolerant persister cells represent surviving cells from which fully drug-resistant cells may eventually emerge to cause tumor relapse, these results support the potentials of these GPX4 inhibitors in the treatment of recurrent cancer. Moreover, the cells from clear-cell carcinomas (CCCs), which are highly aggressive malignancies commonly originating in the kidney and ovary, exhibit substantially higher sensitivity to GPX4 inhibition than normal cells, indicating the presence of a therapeutic window for inducing ferroptosis to treat CCC.⁶⁸ Although it has been shown that intratumoral injection of RSL3 (100 mg/kg) inhibits the growth of xenografted BJeLR cells in mice,²² the poor solubility and unfavorable pharmacokinetic property of RSL3 preclude its systematic use *in vivo*.

ML162 (DPI7, α -[(2-chloroacetyl) (3-chloro-4-methoxyphenyl)amino]-N-(2-phenylethyl)-2-thiopheneacetamide) (Fig. 2) is another chloroacetamide-containing GPX4 inhibitor identified in a high-throughput screening for selective H-Ras-targeting compounds.⁶⁹ Co-crystallization experiments demonstrate that ML162 binds the selenocysteine site of GPX4 via its reactive α -chloroacetamide group.²³ ML162 exhibits a similar ferroptosis-inducing activity as RSL3.

ML210 and JKE-1674

The GPX4 inhibitors RSL3 and ML162 bind GPX4 via a chloroacetamide moiety. In addition to poor pharmacokinetic property, a major limitation of this group of GPX inhibitors

lies in the fact that the highly reactive chloroacetamide moiety mediates promiscuous binding to cellular proteins, resulting in substantial off-target effects.²³ Indeed, ferrostatin-1, a lipophilic radical-trapping antioxidant which specifically rescues cellular effects caused by the loss of GPX4 activity, only partly rescues cell death caused by RSL3 or ML162.²³ ML210 ([4-[bis(4-chlorophenyl)methyl]-1-piperazinyl](5-methyl-4-nitro-3-isoxazolyl)-methanone) (Fig. 2), identified along with ML162, represents another class of GPX4 inhibitors that do not harbor a chloroacetamide group. ML210 exhibits a pattern of cell killing across 821 cancer cell lines that is similar to that of RSL3 and ML162. Although ML210 via a nitroisoxazole group can bind the selenocysteine residue of GPX4 in cells, it does not bind GPX4 *in vitro*, suggesting that ML210 is rather a prodrug. Indeed, cellular activation of ML210 generates the α -nitroketoxime JKE-1674, which retains the same GPX4-binding and—inhibitory activity.²³ While ML210 and JKE-1674 are metabolically stable and bind a significantly-reduced number of cellular proteins, cell death caused by these GPX4 inhibitors can be fully rescued by ferrostatin-1, indicating their high selectivity toward GPX4 inhibition. Importantly, the α -nitroketoxime group of JKE-1674 significantly improves the water solubility. As a result, JKE-1674 can be detected in the serum of mice when orally administrated in a PEG400/ethanol solution.²³ Therefore, JKE-1674 and its optimized derivatives have great potentials in cancer therapy.

Other GPX4 inhibitors

Altretamine (2,4,6-Tris (dimethylamino)-1,3,5-triazine) (Fig. 2) is an oral drug approved by FDA for palliative treatment of patients with persistent or recurrent ovarian cancer. Altretamine is structurally similar to the alkylating agent triethylenemelamine. While the precise mechanism underlying its anti-cancer effects remains unknown, N-demethylation of altretamine may produce reactive intermediates that covalently bind to DNA, causing DNA damage. In a network perturbation analysis, altretamine and SAS were identified as an anti-cancer compound pair that shares a same mechanism of action.⁷⁰ However, unlike SAS, altretamine (0.5 mM) does not decrease the intracellular GSH level, but rather inhibits the GPX4 activity, resulting in lipid peroxidation.⁷⁰ It remains elusive whether altretamine can induce ferroptosis *in vivo* and to what extent the ferroptosis-inducing activity contributes to the anti-cancer effects of altretamine in patients.

High throughput screening also identifies several other GPX4 inhibitors with diverse structures.²² Among them, DPI3, DPI4, DPI6, DPI8, DPI9, DPI12, DPI13, DPI15, and DPI19 contain a chloroacetamide group like RSL3, while DPI10 is highly similar to ML162 and contains a nitroisoxazole warhead. DPI17 and DPI18 have a triazine ring, which is similar to altretamine. DPI12, DPI13, and DPI19 has been confirmed to inhibit GPX4 activity. These compounds are not well characterized so far. In addition, a natural product withaferin A (WA) was also shown to inhibit GPX4 activity and induce ferroptosis in neuroblastoma IMR-32 and SK-N-SH cells.²⁰ WA can bind GPX4 at a high concentration (10 μ M), causing a decrease in the GPX4 protein level. WA at a low concentration (1 μ M), however, binds

Keap 1 and induces ferroptosis through targeting the Keap1-Nrf2-HMOX1 pathway to increase the liable iron pool.²⁰ WA can also induce apoptosis in cancer cells,⁷¹ which may confound the interpretation of its anti-cancer activity *in vivo*.

Iron-based nanomaterials

Nanoparticles encapsulating iron oxide

Nanomaterials/nanoparticles are designed to specifically accumulate at tumor sites due to the EPR effect, and thus are ideal carriers of iron ions whose release in acidic lysosomes accelerates Fenton reactions, leading to ferroptotic death of cancer cells (Fig. 1B). Although ferumoxytol injection (Feraheme), the iron oxide nanoparticles approved by FDA for the treatment of anemia in patients of chronic kidney disease, can promote leukemia cells to produce hydroxides/hydroperoxides and induce them to die,⁷² this agent rather indirectly kill solid tumor cells by inducing macrophage polarization in the tumor microenvironment.⁷³ Indeed, the intracellular H₂O₂ level in cancer cells is often too low for iron oxide nanoparticles to catalyze Fenton reactions to produce sufficient amounts of hydroperoxides for the induction of ferroptosis.⁷⁴ A solution to this problem is to treat cancer cells with iron-based nanocatalysts encapsulated with H₂O₂. For example, a H₂O₂/Fe₃O₄-PLGA polymersome encapsulates H₂O₂ in the hydrophilic core of a Fe₃O₄-poly (lactic-co-glucolic acid) (PLGA) nanocarrier. Disruption of this polymersome by ultrasound can readily release both H₂O₂ and Fe₃O₄ in cancer cells, thereby triggering Fenton reactions to produce excessive reactive oxygen species to kill cancer cells.⁷⁵ However, premature leakage of the strong oxidant from the carriers may cause oxidative damage to normal tissues. An alternative strategy is to use a sequential catalytic nanosystem to produce H₂O₂ specifically within cancer cells.⁷⁴ The nanocatalyst, referred to as GOD-Fe₃O₄@DMSNs, integrates glucose oxidase (GOD) and ultrasmall Fe₃O₄ nanoparticles into large pore-sized and biodegradable dendritic silica nanoparticles (DMSNs). GOD generates H₂O₂ by oxidizing glucose in cancer cells, thereby providing H₂O₂ for production of hydroperoxides in presence of iron. Intravenous injection of 10 mg/kg of GOD-Fe₃O₄@DMSNs inhibits the growth of 4T1 and U87 xenografts by 65% without obvious toxicities.⁷⁴ Single-atom nanocatalysts have improved catalytic activities, and are often nontoxic. Recently, PEGylated single-atom iron-containing nanocatalysts (PSAF NCs) were developed that isolate the Fe atom in nitrogen-doped carbon nanomaterials.⁷⁶ Although the addition of H₂O₂ is required for PSAF NCs to catalyze the production of hydroperoxides and induce ferroptosis *in vitro*, a single dose (20 mg/kg) of intravenous injection of the single-atom nanocatalysts to mice suppresses the growth of 4T1 xenografts by 63.5%.⁷⁶ A mild photothermal process, *i.e.*, exposing the xenografts to an 808 nm laser for 5 min, can accelerate the Fenton reactions catalyzed by PSAF NCs, leading to complete tumor regression.⁷⁶ However, it is worth noting that the iron-based nanomaterials often induce both apoptosis and ferroptosis in cancer cells.

Complex nanoparticles containing other therapeutics

Iron-based metal–organic networks (MON) can also induce ferroptosis in cancer cells. One such agent, MON-p53, carrying not only ferric irons and tannic acid (TA) but also a plasmid expressing p53, is designed to maximally induce lipid peroxidation and ferroptosis by virtue of MON-induced Fenton reactions and p53-mediated downregulation of *SLC7A11*.⁷⁷ MON-p53 was shown to induce ferroptosis of HT1080, 4T1 and SCC-7 cancer cells, and suppress the growth of HT1080 xenografts with negligible systemic toxicity.⁷⁷ A similar agent SRF@Fe^{III}TA (SFT) consists of a network-like corona self-deposited by Fe³⁺ and TA onto sorafenib nanocores.⁷⁸ The entrapped sorafenib can inhibit system Xc⁻ when released along with the corona dissociation in lysosomes of cancer cells, and thus may cooperate with iron-catalyzed Fenton reactions to induce ferroptosis.⁷⁸ These studies also support the feasibility of a strategy for the development of complex nanomaterials that not only induce ferroptosis, but also encapsulate other therapeutic agents for best therapeutic benefits. For example, the magnetic nanoparticle FeGd-HN@Pt@LF/RGD2 loads Fe₃O₄/Gd₂O₃ hybrid nanoparticles with the common anti-cancer drug cisplatin (Pt), lactoferrin (LF) for ready transportation of the nanoparticles across the blood–brain barrier, and a RGD2 peptide whose binding to integrin $\alpha_v\beta_3$ on the surfaces of cancer cells can facilitate endocytosis of the nanoparticles by cancer cells.⁷⁹ This iron-based agent indeed exhibits a profound anti-cancer activity that not only kills U-87MG cancer cells *in vitro*, but also significantly suppresses the growth of orthotopically-implanted glioblastoma cells in animals.⁷⁹

Other ferroptosis-inducing agents

FIN56 (N²,N⁷-dicyclohexyl-9-(hydroxyimino)-9H-fluorene-2,7-disulfonamide) (Fig. 2) is an analog of a small-molecule ferroptosis-inducer CIL56 identified through a screening for compounds inducing caspase-independent death of HT1080 cells. FIN56 induces ferroptosis likely through depleting GPX4 while binding and inhibiting squalene synthase (SQS) required for CoQ₁₀ synthesis.²⁸ Unlike FIN56, FINO2 ((5 α ,8 α)-8-(1,1-dimethylethyl)-3-methyl-1,2-dioxaspiro [4,5]decane-3-ethanol) (Fig. 2), an analog of platinumic acids, rather indirectly inhibits the GPX4 activity and oxidize ferrous irons via its endoperoxide moiety, leading to widespread lipid peroxidation and ferroptosis.⁸⁰ Artesunate (ART, 1-[(3R,5aS,6R,8aS,9R,10S,12R,12aR)-decahydro-3,6,9-trimethyl-3,12-epoxy-12H-pyrano [4,3-j]-1,2-benzodioxepin-10-yl] ester) (Fig. 2), the first-line medication used to treat malaria, also contains an endoperoxide moiety. While it has long been known that ART can induce cell death in a manner dependent on ferrous irons and ROS generation, this well-tolerate, FDA-approved drug was recently shown to induce ferroptosis in pancreatic, head & neck cancer cells, and Burkitt's lymphoma.^{81–84} However, the *in vivo* ferroptosis-inducing and anti-cancer activities of these small molecules remain unknown.

Several other nanoparticles that can induce ferroptosis in cancer cells have also been developed. For example,

low-density lipoprotein docosahexaenoic acid nanoparticles (LDL-DHA) reconstituted with the natural ω -3 fatty acid, docosahexaenoic acid were shown to deplete GSH and inactivate GPX4, leading to accumulation of lipid peroxidation and subsequent ferroptosis in hepatocellular carcinoma cells.⁸⁵ Intratumoral injection of LDL-DHA can also severely suppress the tumor growth in a cohort of mice bearing human HepG2 xenografts.⁸⁵ α MSH-PEG-C' dots, ultrasmall near-infrared fluorescent silica nanoparticles coated by PEG with diameters down to 10 nm and surface-functionalized with alpha-melanocyte stimulating hormone (α MSH), the melanoma-targeting peptide, can also induce ferroptosis in multiple cancer cells in nutrient-deprived conditions while inhibiting the growth of HT1080 and 786-O xenografts when intravenously injected into mice.⁸⁶ It was proposed that the α MSH peptide mediates the internalization and cancer-cell targeting while the silica particles can absorb and incorporate iron from the extracellular environment to promote Fenton reactions.⁸⁶

Regulators of cancer cell sensitivity to ferroptosis

As ferroptosis-inducing agents are often effective in only a subset of cancer cells, great efforts have been put forth to understand the molecular basis of cell sensitivity to ferroptosis. As a result, a large number of genes have been identified that can promote or suppress ferroptosis by regulating the major ferroptosis mediators (*i.e.*, system Xc⁻ and GPX4), or the metabolic pathways crucial for triggering ferroptosis (Fig. 3A). Many of these genes are potential drug targets for sensitizing cancer cells to ferroptosis-based therapies.

SLC7A11 regulators

A major group of such regulators are transcription factors (TFs) that can regulate transcription of *SLC7A11* – the gene encoding the light chain unique to system Xc⁻ (Fig. 3B). Downregulation of *SLC7A11* expression would impair cystine uptake by system Xc⁻, thereby depleting GSH leading to an increase in cancer cell sensitivity to ferroptosis-inducing agents. The tumor suppressor p53 was the first-identified TF that promotes erastin- or oxidative stress-induced ferroptosis by repressing *SLC7A11* transcription.⁸⁷ Although p53 was shown to directly bind to a p53-consensus site (p53-BS) in the mouse *SLC7A11* promoter,⁸⁷ deletion of the corresponding fragment in the human *SLC7A11* promoter (Fig. 3B) does not prevent p53 from repressing the human gene,⁸⁸ suggesting that the binding of p53 to this putative site is unlikely required for p53-mediated *SLC7A11* repression.⁸⁹ P53 can also promote ferroptosis by inducing expression of glutaminase 2 (GLS2) and spermidine/spermine N1-acetyltransferase 1 (SAT1).^{90,91} Paradoxically, p53 was also shown to decrease the sensitivity of cancer cells to ferroptosis by inducing p21 expression to promote GSH synthesis,⁹² or blocking the activity of DPP4 in a transcription-independent manner,⁹³ in other cell contexts. Activating transcription factor 3 (ATF3) is another *SLC7A11* transcriptional repressor that can suppress system Xc⁻ and predispose cells to a state sensitive to ferroptosis induced

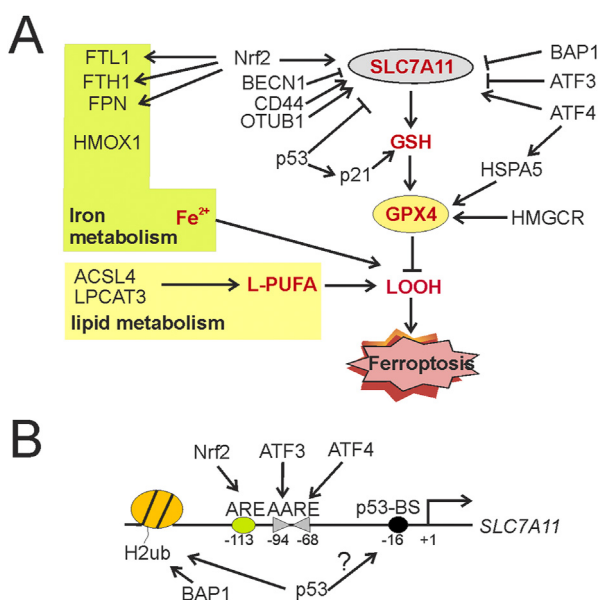


Figure 3 Genetic factors regulate cancer cell sensitivity to ferroptosis-inducing agents. **(A)** Proteins that can regulate GSH synthesis, iron hemostasis, or lipid metabolism alter the sensitivity of cancer cells to ferroptosis induction. These proteins include transcription factors (e.g., p53, ATF3, ATF4, and Nrf 2) and epigenetic factors (e.g., BAP1 and OTUB1) that regulate GPX4 translation/stability or expression of *SLC7A11* and iron-regulatory genes (e.g., *FPN*), and enzymes (e.g., ACSL4) that regulate the synthesis of phospholipids **(B)** Multiple transcription factors and epigenetic factors bind to the human *SLC7A11* promoter and regulate *SLC7A11* transcription. Note that the p53-binding element (p53-BS) appears to be dispensable for the regulation of the activity of the human *SLC7A11* promoter. ARE, antioxidant response element; AARE, amino acid response element; H2ub, ubiquitinated histone H2.

by various FINs.⁸⁸ ATF3 binds the *SLC7A11* promoter at two adjacent sites that are in opposite orientation but coincide with the amino acid response elements (AAREs) (Fig. 3B) presumably bound by transcription activators C/EBP or other ATF/CREB members, thereby likely repressing *SLC7A11* expression by competing with these transcription activators for binding to the promoter.⁸⁸ Indeed, ATF4, a family member of ATF3, was shown to bind these AAREs (Fig. 3B), but the binding rather leads to *SLC7A11* transactivation and suppression of ferroptosis in human glioma cells.⁹⁴ ATF4 can also transactivate *HSPA5*, which encodes a heat shock protein capable of blocking the degradation of GPX4⁹⁵(Fig. 3A). Accordingly, knockdown of ATF4 was shown to indirectly regulate the GPX4 protein level and promote human pancreatic ductal adenocarcinoma (PDAC) cells to undergo ferroptosis.⁹⁵ However, as *Atf4*-knockout mice are embryonic lethal, drugging ATF4 could result in severe side effects. In addition to TFs, *SLC7A11* transcription is also repressed by BRCA1-associated protein 1 (BAP1) – a major component of a deubiquitinase (DUB) complex which catalyzes deubiquitination of histone 2 A (H2A) associated with the *SLC7A11* promoter.⁹⁶ Not surprisingly, BAP1 promotes ferroptosis in a DUB-dependent manner.⁹⁶ Other *SLC7A11*/system Xc⁻ regulators include CD44 and

OTUB1, which suppress ferroptosis by directly binding the *SLC7A11* protein and increasing the stability of the latter.^{97,98} Beclin-1 (BECN1), a key autophagy regulator, on the other hand, inhibits system Xc⁻ by binding *SLC7A11*, and consequently promotes ferroptosis induced by system Xc⁻ inhibitors.⁹⁹

GPX4 regulators

In contrast to *SLC7A11*/system Xc⁻, the GPX4 activity is mainly regulated at the posttranscriptional level. As mentioned earlier, *HSPA5*, also known as GRP78 or BIP, can bind GPX4 and stabilize the peroxidase, and thus negatively regulate ferroptosis induced by erastin.⁹⁵ In line with this activity, knockdown of *HSPA5* or pharmacologic inhibition of *HSPA5* by epigallocatechin gallate (EGCG) promotes GPX4 protein degradation, thereby promoting erastin-induced ferroptosis in pancreatic cancer cells.⁹⁵ GPX4 is a selenoprotein containing the unusual amino acid selenocysteine (Sec) at its active site. The translation/synthesis of GPX4 requires selenocysteine-tRNA^{[Sec]^{Sec}} whose functionality depends on isopentenylation - a biochemical reaction catalyzed by tRNA isopentenyl transferase using isopentenyl pyrophosphate (IPP) as the isopentenyl donor.^{38,100} As IPP is an intermediate of the mevalonate pathway, statins (e.g., fluvastatin, lovastatin, and simvastatin), the inhibitors of the rate-limiting enzyme 3-hydroxy-3-methylglutaryl coenzyme A reductase (HMGCR) in this pathway, can decrease the GPX4 protein level, and sensitize cancer cells to ferroptosis induced by FIN56 or RSL3.^{3,28} Statins can also promote ferroptosis by depleting CoQ₁₀ produced by the mevalonate pathway, or inhibiting tRNA isopentenylation via TRIT1^{3,28,101}.

Nuclear factor erythroid 2-related factor 2 (Nrf 2)

Nrf 2 is a master anti-oxidant TF whose level is mainly regulated by proteasomal degradation mediated by kelch-like ECH-associated protein 1 (Keap 1).¹⁰² Nrf2 along with small Maf proteins bind to antioxidative response elements (ARE) to transactivate a number of genes involved in redox homeostasis, iron metabolism, and lipid metabolism.¹⁰³ *SLC7A11* and *GPX4* are among the genes induced by Nrf2^{103,104} (Fig. 3B). This master antioxidant TF can also transactivate genes encoding both the light chain (FTL1) and heavy chain (FTH1) of ferritin, the major iron storage protein, as well as the only known iron-transporter ferroportin (FPN), and thus suppress ferroptosis by reducing the level of liable iron pool¹⁰³(Fig. 3A). Indeed, Nrf2 overexpression, or inhibition of its negative regulator Keap1, can confer ferroptosis resistance in glioblastoma and head & neck cancer cells, while knockdown of Nrf2 promotes cancer cells to undergo ferroptotic cell death.^{83,105} Accordingly, trigonelline, a coffee alkaloid which can inhibit Nrf2 nuclear translocation and transactivation activity, enhances erastin/sorafenib-induced ferroptosis in hepatocellular carcinoma cells, or artesunate-induced ferroptosis in head & neck cancer cells.^{83,106} These studies thus support a notion that Nrf2-specific inhibitors could be of value in use to promote ferroptosis in cancer cells.

However, it is important to note that trigonelline is not a specific Nrf2 inhibitor.

Ferroptosis suppressor protein 1 (FSP1)

FSP1 mediates a ferroptosis-suppressing pathway that can compensate for GPX4 inactivation in cancer cells (Fig. 1B). Indeed, *FSP1* expression directly correlates with resistance to GPX4 inhibitors (including RSL3, ML162, and ML210) in a panel of 860 cancer cell lines.^{29,30} 3 μ M of iFSP1 (1-amino-3-(4-methylphenyl)-pyrido [1,2-a]benzimidazole-2,4-dicarbonitrile) (Fig. 2), a FSP1 inhibitor identified from a screen of 10,000 drug-like compounds, was shown to sensitize cancer cell lines from different origins to RSL3-induced ferroptosis.²⁹ However, further chemical optimization appears to be necessary before applying this agent for *in vivo* studies.²⁹

Other ferroptosis regulatory proteins

Many other proteins can also regulate the sensitivity to ferroptosis (Fig. 3B). For example, knockdown of acyl-CoA synthase 4 (ACSL4), which can convert long-chain fatty acids into fatty acyl-CoA esters, or lysophosphatidylcholine acyltransferase 3 (LPCAT3), which catalyzes the insertion of acylated fatty acids into phospholipids, prevents ferroptosis induced by erastin or RSL3.^{107–110} Knockdown of cysteinyl-tRNA synthetase (CARS) or several other tRNA synthetases results in upregulation of the transsulfuration pathway and cysteine biosynthesis, thereby suppressing ferroptosis induced by system Xc⁻ inhibitors.¹¹¹ In addition, many proteins regulating iron metabolism, such as iron response element-binding protein 2 (IREB2), heat shock factor binding protein 1 (HSPB1), and CDGSH iron sulfur domain 1 (CISD1) can also regulate cancer cell sensitivity to ferroptosis.^{6,112,113}

Concluding remarks

While therapy-resistant cancer cells are vulnerable to GPX4 inhibition, conventional cancer therapeutics (e.g., cisplatin, radiation, and immunotherapy) can also kill cancer cells by inducing ferroptosis. It has thus become clear that therapeutic induction of ferroptosis holds great promises for improving the outcome of current cancer therapy.¹⁰ Although a number of ferroptosis-inducing small molecules have been developed and demonstrated to induce death of cultured cancer cells, many of them have poor bioavailability, precluding their usages in cancer therapy. While medicinal chemistry optimization can improve pharmacological properties of these small-molecule agents, it is worth noting that the fact that human cancers are more heterogeneous and complex than mouse tumors used in animal models³² may limit the success of ferroptosis-inducing nanoparticles in clinical settings. On the other hand, as only a subset of cancer cells are sensitive to ferroptosis-inducing agents, the success of ferroptosis-based cancer therapies depends on the identification of reliable biomarkers for predicting the responsiveness of a cancer. However, the fact that ferroptosis results from dysregulation of intersected metabolic

pathways⁹ hinders the efforts toward this direction. Indeed, while the intracellular level of NADPH, an essential reductant for lipid radicals, was found to determine cancer cell sensitivity to GPX4 inhibitors,²⁸ it remains unclear how broad this biomarker can be used to predict cancer responsiveness to GPX4 inhibitors. In a similar vein, although FSP1 emerges as a novel ferroptosis suppressor, the FSP1 inhibitor (*i.e.*, iFSP1) appears to only sensitize a subset of cancer cell lines to GPX4 inhibition.²⁹ Notwithstanding these limitations, while we have been making great strides toward a better understanding of ferroptosis and its regulatory network, we have many reasons to be optimistic that therapeutic induction of ferroptosis will eventually become a new therapeutic option for cancer patients. Indeed, the recent identification of transferrin receptor 1 as a ferroptosis marker¹¹⁴ would allow for characterization of ferroptosis *in vivo*, and thus represents a significant step toward the success of ferroptosis-based cancer therapy.

Conflict of interests

The authors declare no conflict of interest.

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