

Palladium-Catalyzed Arylation of C(sp²)-H Bonds with 2-(1-Methylhydrazinyl)pyridine as the Bidentate Directing Group

Jian Wei, Xiaoru Shao, Hua Zhao, Hongjian Yang, Shuxian Qiu, and Hongbin Zhai*

Cite This: *ACS Omega* 2021, 6, 25151–25161

Read Online

ACCESS |



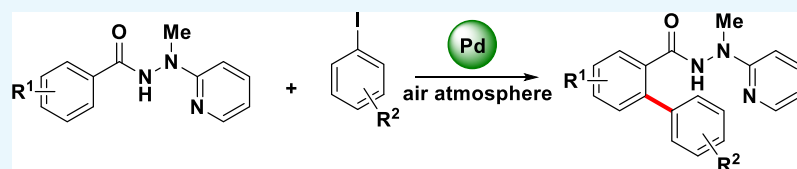
Metrics & More



Article Recommendations



Supporting Information



● 27 examples

● broad substrate scope

● operationally simple

● reductively removable directing group

ABSTRACT: Palladium-catalyzed C(sp²)-H arylation of *ortho* C-H bonds involving 2-(1-methylhydrazinyl)pyridine (MHP) as the directing group has been investigated. The reaction proceeds smoothly under an air atmosphere to generate biaryl derivatives in an environmentally friendly manner while tolerating a wide range of functional groups. Notably, the directing group present in the product could be easily removed under mild reductive conditions.

INTRODUCTION

Transition-metal-catalyzed C-H functionalization is a powerful tool for the transformation of an inert C-H bond into a C-C or C-X (X = N, O, S, F, B, etc.) bond that enables efficient construction of structurally diverse natural products and pharmaceutical compounds.^{1,2} In particular, C-H arylation, as an environmentally benign and economically attractive alternative to the traditional cross-coupling reactions that require prefunctionalized substrates and generate stoichiometric metallic wastes, has garnered considerable attention.^{3,4} In 2005, Daugulis and co-workers reported their pathbreaking example of 8-aminoquinoline-directed C(sp²)-H and C(sp³)-H arylation by utilizing Pd(OAc)₂ as the catalyst.⁵ It is believed that the bidentate auxiliaries could bind to a metal center and allow direct insertion of a metal catalyst into a proximal C-H bond, followed by functionalization of the resulting organometallic intermediate. Since then, a variety of dual-chelation-assisted C-H arylations have been developed, in which 8-aminoquinoline, picolinic acid, and other related compounds are the most used directing moieties.⁶

The biaryl unit widely occurs in natural products, pharmaceuticals, agrochemicals, and conjugated materials,⁷ as exemplified by hippadine,⁸ azilsartan,⁹ and bifenthrin¹⁰ (Figure 1). By exploiting C(sp²)-H arylation, the biaryl motifs have been constructed successfully with the aid of metal catalysts such as palladium,^{5,11} ruthenium,¹² cobalt,¹³ nickel,¹⁴ copper,¹⁵ and iron¹⁶ complexes (Scheme 1).

Recently, we have developed a novel removable bidentate directing group, 2-(1-methylhydrazinyl)pyridine (MHP),¹⁷ which exhibited superior reactivity in the functionalization of aromatic C(sp²)-H bonds. This directing group can be easily

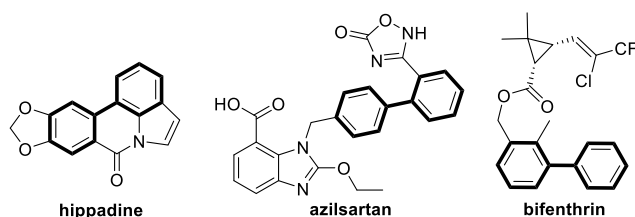


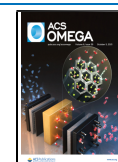
Figure 1. Representative bioactive biaryl derivatives.

synthesized from commercially available materials. So far, we have achieved highly efficient C-H functionalization of benzoyl hydrazides with alkynes,^{17a} CO,^{17b} allenes,^{17c} maleimides,^{17d,e} oxabicyclic alkenes,^{17f} and isocyanides.^{17g} Herein, we report a palladium-catalyzed direct C(sp²)-H arylation with MHP as the directing group. The current synthetic approach to biaryl derivatives features a broad substrate scope, great functional group tolerance, and operational simplicity. In particular, the directing group present in the products could be easily removed under mild reductive conditions in our case.¹⁷

Received: May 11, 2021

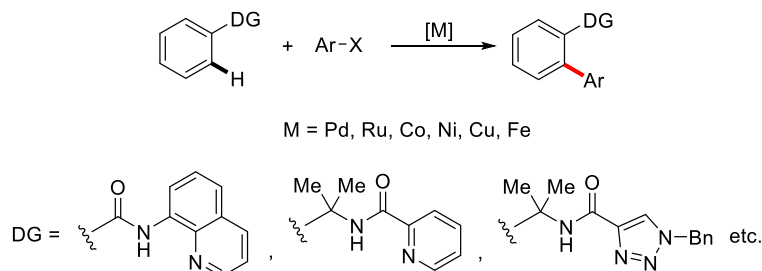
Accepted: September 16, 2021

Published: September 27, 2021

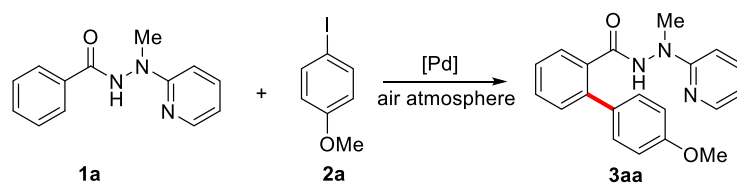


Scheme 1. Transition Metal-Catalyzed C(sp²)-H Arylation

Previous work

a) bidentate-directed C(sp²)-H bond arylation reaction

This work

b) MHP-directed C(sp²)-H bond arylation reaction

RESULTS AND DISCUSSION

N'-Methyl-*N'*-(pyridin-2-yl)benzohydrazide **1a** and 4-iodoanisole **2a** were chosen as the model substrates to survey the optimal reaction parameters. For the baseline experiment, treatment of **1a** with **2a** in the presence of Pd(OAc)₂ (10 mol %), AgOAc (1.5 equiv), and NaOPiv (2.0 equiv) in PhCl at 130 °C for 24 h afforded the desired biphenyl product **3aa** in 10% yield (Table 1, entry 1). Similar results were obtained when AgOAc was replaced with other silver salts such as Ag₂CO₃ and AgNO₃ (entries 2 and 3), suggesting that the type of silver salt has little effect on the reaction. A significant enhancement of the reaction efficiency (46%) was realized when the reaction time was extended to 48 h in the absence of any silver salt additives (entry 4). A brief screening of catalysts revealed that Pd(OAc)₂ is the optimal catalyst (Table S1). The yield of **3aa** was further improved as the catalyst load was increased to 20 mol % (entry 5). Various bases including NaOAc, Na₂CO₃, NaHCO₃, K₂CO₃, and KOAc were evaluated and NaOAc was found to be the best in terms of the reaction outcome (entries 6–10). Comparison of the solvents revealed that *m*-xylene, mesitylene, and dimethylformamide (DMF) were inferior to PhCl (entries 11–13). The yield of **3aa** was increased to 77% when the reaction was conducted at 140 °C, while the yield of **3aa** dropped significantly at 120 °C (entries 14 and 15). The control experiments indicated that the presence of the base was critical to the reaction (entry 16) and that the reaction was completely inhibited in the absence of the Pd(OAc)₂ catalyst (entry 17).

Having established the optimized reaction conditions, we next explored the scope of the hydrazide substrates (Scheme 2). Gratifyingly, benzoyl hydrazides bearing either an electron-donating group (e.g., Me, ^tBu, OMe, SMe, OPh, OCF₃, and Ph) or an electron-withdrawing group (e.g., F, Cl, Br, CF₃, and CO₂Me) at the para-position of the aromatic ring were well accommodated, furnishing the expected products in moderate to high yields (**3aa**–**3ma**), which indicated that the reaction might not be sensitive to the electronic effect. Substrates with a

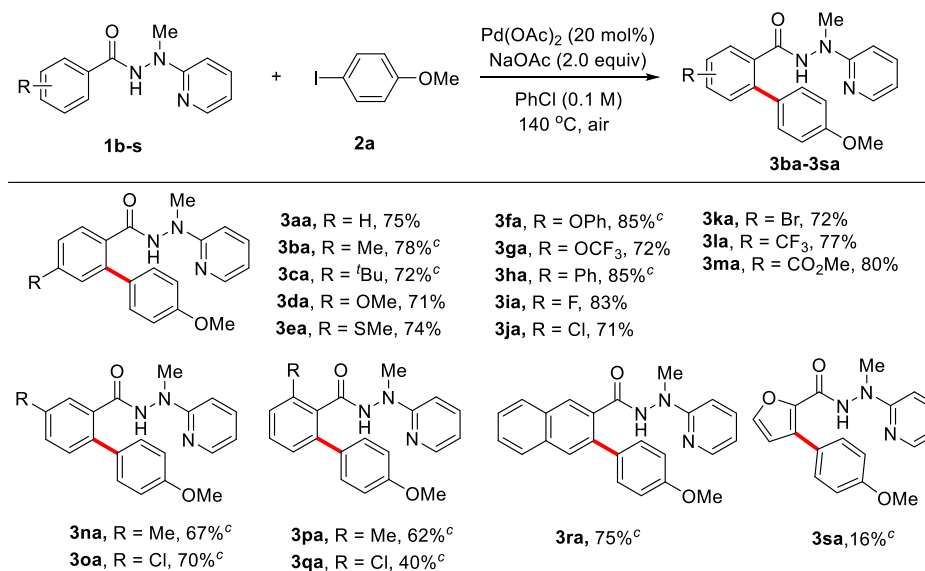
Table 1. Optimization of Reaction conditions^a

| entry | solvent | additive | base | yield(%) ^b |
|-------------------|------------------|---------------------------------|---------------------------------|-----------------------|
| 1 ^{c,d} | PhCl | AgOAc | NaOPiv | 10 |
| 2 ^{c,d} | PhCl | Ag ₂ CO ₃ | NaOPiv | trace |
| 3 ^{c,d} | PhCl | AgNO ₃ | NaOPiv | 8 |
| 4 ^d | PhCl | / | NaOPiv | 46 |
| 5 | PhCl | / | NaOPiv | 51 |
| 6 | PhCl | / | Na ₂ CO ₃ | 57 |
| 7 | PhCl | / | NaOAc | 69 |
| 8 | PhCl | / | NaHCO ₃ | 53 |
| 9 | PhCl | / | K ₂ CO ₃ | ND |
| 10 | PhCl | / | KOAc | 26 |
| 11 | <i>m</i> -xylene | / | NaOAc | 35 |
| 12 | mesitylene | / | NaOAc | 38 |
| 13 | DMF | / | NaOAc | ND |
| 14 ^e | PhCl | / | NaOAc | 54 |
| 15 ^f | PhCl | / | NaOAc | 77 (75 ^h) |
| 16 ^g | PhCl | / | / | 30 |
| 17 ^{f,g} | PhCl | / | NaOAc | ND |

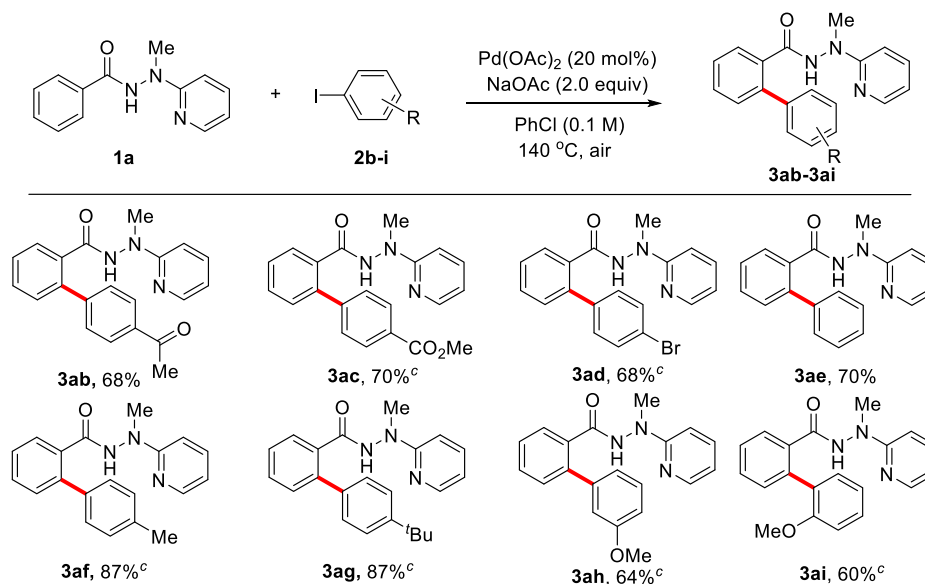
^aReaction conditions: **1a** (0.2 mmol), **2a** (0.6 mmol), Pd(OAc)₂ (0.04 mmol), base (0.4 mmol), solvent (2.0 mL), 48 h, air, 130 °C.

^bDetermined by ¹H NMR using 1,3,5-trimethoxybenzene as the internal standard. ^cThe reaction time was 24 h, under Ar. ^dPd(OAc)₂ (0.02 mmol) was used. ^eAt 120 °C. ^fAt 140 °C. ^gIn the absence of Pd(OAc)₂. ^hIsolated yield. ND = not detectable.

meta-substituent (e.g., Me and Cl) on the benzene ring afforded **3na** and **3oa** in 67 and 70% yields, respectively. ortho-Substituted hydrazides were also compatible with this protocol (**3pa**, **3qa**). In addition, the 2-naphthamide derivative proved to be a suitable substrate, giving the product in 75% yield (**3ra**). However, the hydrazide with a heteroaromatic moiety (such as furan) seemed to be less efficient under the

Scheme 2. Substrate Scope of the Benzoylhydrazone^{a,b}

^aReaction conditions: **1** (0.2 mmol), **2a** (0.6 mmol), Pd(OAc)₂ (20 mol %), NaOAc (2.0 equiv), PhCl (2.0 mL), air, 140 °C, 48 h. ^bIsolated yields. ^cThe reaction time was 72 h.

Scheme 3. Substrate Scope of the Iodobenzenes^{a,b}

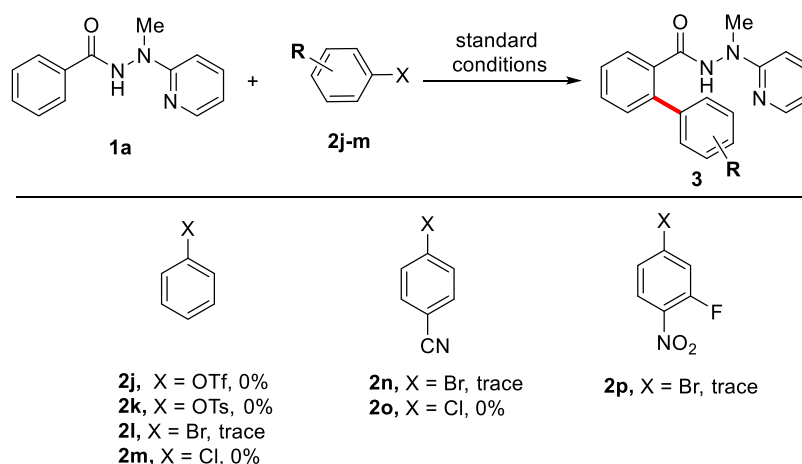
^aReaction conditions: **1a** (0.2 mmol), **2** (0.6 mmol), Pd(OAc)₂ (20 mol %), NaOAc (2.0 equiv), PhCl (2.0 mL), air, 140 °C, 48 h. ^bIsolated yields. ^cThe reaction time was 72 h.

standard conditions, delivering the corresponding product in only 16% yield (**3sa**).

The scope of the aryl iodide substrates was then investigated and the results are summarized in Scheme 3. The experiments showed that aryl iodides with electron-withdrawing substituents at the para-position afforded the corresponding coupling products in 68–70% yields (**3ab–3ad**). Notably, aryl iodides with a Br atom on the phenyl ring (see **3ad**) worked well and showed excellent chemoselectivity in this reaction. In comparison, aryl iodides with an electron-donating substituent at the para-position performed even better (**3af**, **3ag**). Moreover, when the electron-donating group was located at the ortho- or meta-position of aryl iodides, the reaction still

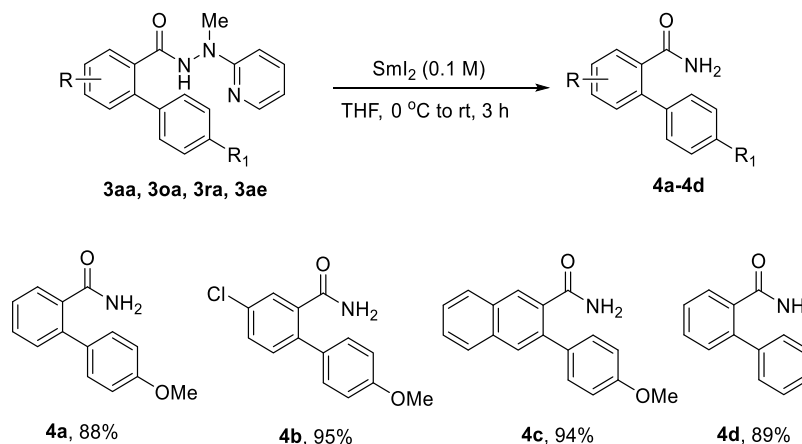
worked but gave the products in slightly lower yields (**3ah**, **3ai**).

Next, replacement of Ar–I with Ar–OTf, Ar–OTs, Ar–Br, and Ar–Cl was evaluated for the reaction (Scheme 4). With Ar–OTf (**2j**), Ar–OTs (**2k**), and Ar–Cl (**2m**, **2o**) as the substrates, no desired products were obtained, while the reaction of aryl bromides **2l**, **2n**, and **2p** did give a trace amount of the products under the standard conditions. This can be accounted for based on the barrier for oxidative addition of aryl halides, which increases in the order of ArI < ArBr < ArCl, consistent with the reactivity order. The speculation is in line with the mechanism proposed in the literature.¹⁸

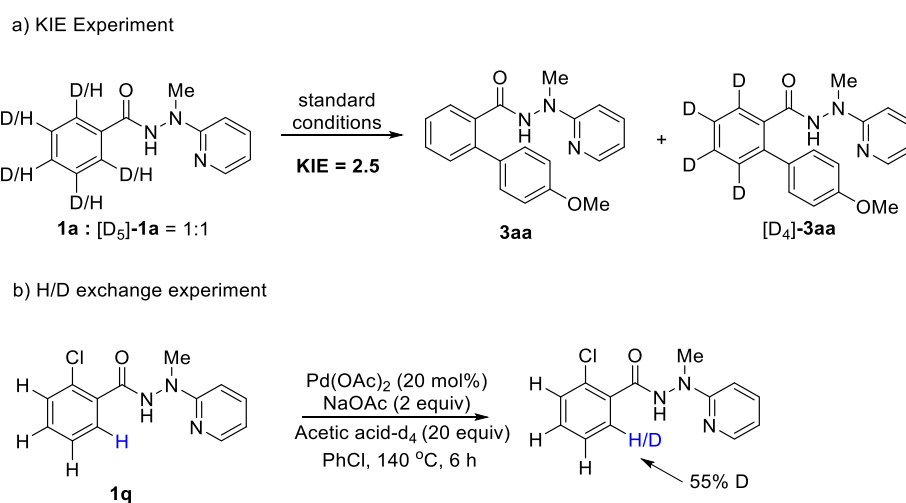
Scheme 4. Substrate Scope of Other Coupling Partners^{a,b}

^aReaction conditions: **1a** (0.2 mmol), **2** (0.6 mmol), Pd(OAc)₂ (20 mol %), NaOAc (2.0 equiv), PhCl (2.0 mL), air, 140 °C, 48 h. ^bDetermined by ¹H NMR using 1,3,5-trimethoxybenzene as the internal standard.

Scheme 5. Reductive Removal of the Directing Group



Scheme 6. Mechanistic Studies

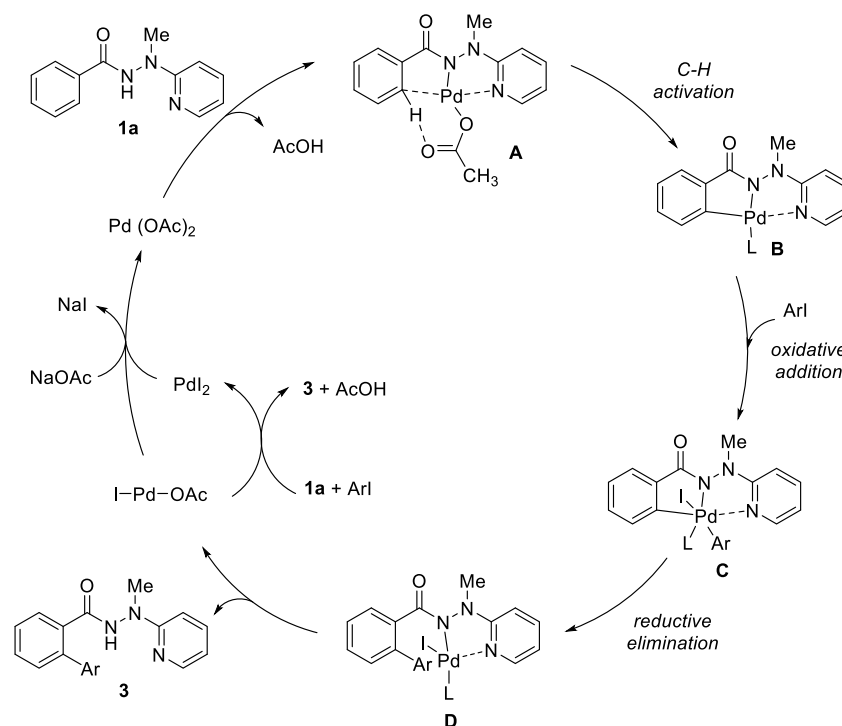


Similar to the transformations developed previously in this laboratory,¹⁷ the N–N bond present in the products was easily cleaved under mild reductive conditions. For example, treatment of 2-aryl benzoyl hydrazines **3aa**, **3oa**, **3ra**, and **3ae** with SmI₂ afforded benzamides **4a–4d** in 88–95% yields

(Scheme 5). Therefore, the current reaction features easy N-deprotection that allows further N-derivatization.

The reaction kinetics plot revealed that the reaction rate was fast at the start of the reaction and gradually decreased with time (see the Supporting Information). In order to better

Scheme 7. Proposed Mechanism



understand the reaction mechanism, the kinetic isotope effect of the catalytic reaction was evaluated by the reaction of hydrazide **1a** and isotopically labeled counterpart [D₅]-**1a** with **2a** under the standard reaction conditions (Scheme 6). The $k_{\text{H}}/k_{\text{D}}$ value was determined to be 2.5, indicating that *ortho* C(sp²)-H bond cleavage took place in the rate-determining step (see the Supporting Information). Furthermore, by employing acetic acid-*d*₄ as a cosolvent, H/D scrambling was observed at the *ortho*-position of the carbonyl group in **1q**, suggesting the reversibility of the C-H cleavage step under the reaction conditions.

Based on our preliminary mechanistic studies and the relevant literature reports,¹⁹ a plausible reaction mechanism is proposed in Scheme 7. First, **1a** reacts with Pd(OAc)₂ to generate the palladium amidate **A**, facilitating the subsequent C-H insertion to produce the palladium chelate **B**. Oxidative addition of **B** to aryl iodide forms Pd(IV) species **C**, which undergoes reductive elimination to deliver the desired product **3** along with PdIOAc. The latter (i.e., PdIOAc) could initiate a second catalytic cycle, leading to the unreactive species, PdI₂. However, both PdI₂ and PdIOAc could react with NaOAc to regenerate the catalyst, Pd(OAc)₂.

CONCLUSIONS

We have accomplished an efficient palladium-catalyzed C-(sp²)-H arylation involving the easily accessible and highly efficient MHP as the directing group. With this unique directing group developed by our laboratory, arylations of C-H bonds were carried out under an air atmosphere with good regioselectivity, and a wide range of aryl iodides and benzoic hydrazides were successfully applied. In spite of the involvement of the palladium catalyst, our protocol should still be useful in the facile construction of valuable biaryl scaffolds, considering the superb accessibility and removability of the MHP directing group as well as the potential

recyclability of the catalyst in large-scale applications. Studies on the application of MHP as the directing group to other related transformations and to achieve a clearer understanding of the reaction mechanism are ongoing.

EXPERIMENTAL SECTION

General Information. All reactions were carried out under an air atmosphere, unless otherwise noted. All the chemicals were purchased commercially, and used without further purification. Thin-layer chromatography (TLC) was conducted with 0.25 mm Tsingdao silica gel plates (60F-254) and visualized by exposure to UV light (254 nm). Flash column chromatography was performed on the Tsingdao silica gel (200–300 mesh). ¹H NMR spectra were recorded on Bruker spectrometers (at 400 or 500 MHz) and reported relative to deuterated solvent signals or tetramethylsilane internal standard signals. Data for ¹H NMR spectra were reported as follows: chemical shift (δ /ppm), multiplicity (*s* = singlet, *d* = doublet, *t* = triplet, *q* = quartet, *m* = multiplet, and *br* = broad), coupling constant (*J*/Hz), and integration. ¹³C NMR spectra were recorded on Bruker spectrometers (101 or 126 MHz). Data for ¹³C NMR spectra were reported in terms of chemical shift. ¹⁹F NMR spectra were recorded on Bruker spectrometers (376 MHz). High-resolution mass spectrometry (HRMS) was conducted on Bruker Apex IV RTMS.

General Procedure for the Synthesis of Starting Materials (1a–1s). To a stirred solution of MHP (1.0 equiv, 5.0 mmol) and Et₃N (5.0 equiv) in dry dichloromethane (DCM) (0.5 M) was added benzoyl chloride (1.05 equiv) dropwise under an Ar atmosphere at 0 °C. After stirring for 30 min, the resulting mixture was warmed to room temperature and stirred overnight at this temperature. Upon completion of the reaction indicated by TLC, the reaction mixture was washed with H₂O and extracted with DCM (50 mL × 3). The combined organic phases were washed with brine, dried over

with anhydrous Na_2SO_4 , filtered, and concentrated under reduced pressure. The residue was purified by column chromatography (petroleum ether/EtOAc = 3:1 to 1:1) to afford the corresponding product. All spectroscopic data of **1a–1s** are in good agreement with the literature reported data.¹⁷

General Procedure for Palladium-Catalyzed $\text{C}(\text{sp}^2)$ –H Arylation. A mixture of N' -methyl- N' -(pyridin-2-yl)-benzohydrazide (0.2 mmol, 1.0 equiv), aryl iodides (0.6 mmol, 3.0 equiv), $\text{Pd}(\text{OAc})_2$ (9 mg, 0.2 equiv), NaOAc (33 mg, 2.0 equiv), and PhCl (2.0 mL, 0.1 M) was added to a 10 mL reaction tube. It was stirred at 140 °C for 48 h under air conditions. After cooling to room temperature, the reaction mixture was filtered through a plug of Celite, followed by washing with 10 mL of DCM. The combined residue was concentrated under reduced pressure, and then the resulting crude product was purified by column chromatography to provide **3**.

General Procedure for Reductive Removal of the Directing Group. An oven-dried 25 mL two-neck round bottom flask was charged with **3** (0.1 mmol). After purging with Ar three times, 5 mL fresh distilled THF was added, followed by the dropwise addition of SmI_2 (0.1 M in THF, 5.0 equiv) at 0 °C. After 5 min, the mixture was warmed to room temperature and stirred for 3 h. After that, the mixture was quenched with 5 mL saturated aqueous $\text{Na}_2\text{S}_2\text{O}_3$ and extracted with DCM, dried over Na_2SO_4 , filtered, concentrated under reduced pressure, and **4** was obtained via column chromatography.

Characterization Data of Products. 4'-Methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3aa** (50 mg, yield = 75%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.11 (ddd, J = 5.0, 1.8, 0.8 Hz, 1H), 7.71 (dd, J = 7.6, 1.1 Hz, 1H), 7.51 (td, J = 7.5, 1.4 Hz, 1H), 7.46–7.37 (m, 4H), 7.34 (ddd, J = 8.8, 7.2, 1.9 Hz, 1H), 7.23 (s, 1H), 6.98 (d, J = 8.7 Hz, 2H), 6.64 (ddd, J = 7.1, 5.0, 0.8 Hz, 1H), 6.30 (d, J = 8.5 Hz, 1H), 3.85 (s, 3H), 3.13 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.8, 159.6, 159.1, 147.5, 139.5, 137.4, 133.6, 132.2, 130.6, 130.2, 128.9, 127.4, 114.6, 114.3, 107.2, 55.4, 37.5. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{20}\text{N}_3\text{O}_2$, 334.1477; found, 334.1550.

4'-methoxy- N' ,5-dimethyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ba** (54 mg, yield = 78%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (500 MHz, CDCl_3): δ 8.13 (dd, J = 4.9, 1.0 Hz, 1H), 7.66 (d, J = 7.8 Hz, 1H), 7.41 (d, J = 8.7 Hz, 2H), 7.35 (ddd, J = 8.8, 7.3, 1.8 Hz, 1H), 7.25 (d, J = 9.5 Hz, 1H), 7.19 (s, 1H), 7.09 (s, 1H), 6.99 (d, J = 8.7 Hz, 2H), 6.64 (dd, J = 6.7, 5.3 Hz, 1H), 6.33 (d, J = 8.5 Hz, 1H), 3.85 (s, 3H), 3.15 (s, 3H), 2.43 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3): δ 168.7, 159.6, 159.1, 147.4, 140.9, 139.5, 137.4, 132.4, 131.0, 130.6, 130.1, 129.1, 128.1, 114.5, 114.2, 107.2, 55.4, 37.4, 21.4. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{21}\text{H}_{22}\text{N}_3\text{O}_2$, 348.1707; found, 348.1706.

5-(*tert*-butyl)-4'-methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ca** (55 mg, yield = 72%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white solid; mp 81–83 °C. ^1H NMR (400 MHz, CDCl_3): δ 8.11 (ddd, J = 4.9, 1.8, 0.8 Hz, 1H), 7.69 (d, J = 8.1 Hz, 1H), 7.47 (dd, J = 8.2, 2.0 Hz, 1H), 7.42 (d, J = 8.7 Hz, 2H), 7.38–7.31 (m, 2H), 7.15 (s, 1H), 7.00 (d, J = 8.7 Hz, 2H), 6.64 (ddd, J = 7.1, 5.0, 0.7 Hz, 1H), 6.32 (d, J = 8.5 Hz, 1H), 3.86

(s, 3H), 3.14 (s, 3H), 1.36 (s, 9H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.7, 159.6, 159.2, 154.1, 147.5, 139.3, 137.3, 132.9, 130.6, 130.2, 128.9, 127.3, 124.6, 114.5, 114.3, 107.2, 55.4, 37.5, 34.9, 31.2. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{24}\text{H}_{28}\text{N}_3\text{O}_2$, 390.2176; found, 390.2178.

4',5-dimethoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3da** (52 mg, yield = 71%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (500 MHz, CDCl_3): δ 8.12 (dd, J = 4.9, 1.0 Hz, 1H), 7.73 (d, J = 8.6 Hz, 1H), 7.40 (d, J = 8.5 Hz, 2H), 7.35 (ddd, J = 8.8, 7.3, 1.8 Hz, 1H), 7.10 (s, 1H), 6.99 (d, J = 8.7 Hz, 2H), 6.95 (dd, J = 8.6, 2.6 Hz, 1H), 6.85 (d, J = 2.5 Hz, 1H), 6.64 (dd, J = 6.7, 5.3 Hz, 1H), 6.36 (d, J = 8.5 Hz, 1H), 3.87 (s, 3H), 3.85 (s, 3H), 3.14 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3): δ 168.2, 161.2, 159.8, 159.2, 147.5, 141.5, 137.3, 132.4, 131.2, 130.1, 125.9, 115.6, 114.5, 114.3, 112.9, 107.2, 55.5, 55.4, 37.5. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{21}\text{H}_{22}\text{N}_3\text{O}_3$, 364.1656; found, 364.1655.

4'-methoxy- N' -methyl-5-(methylthio)- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ea** (56 mg, yield = 74%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.12 (dd, J = 5.0, 1.1 Hz, 1H), 7.67 (d, J = 8.2 Hz, 1H), 7.43–7.31 (m, 3H), 7.27 (dd, J = 8.1, 2.0 Hz, 1H), 7.18 (d, J = 1.9 Hz, 1H), 7.17 (s, 1H), 6.99 (d, J = 8.7 Hz, 2H), 6.64 (dd, J = 6.8, 5.2 Hz, 1H), 6.32 (d, J = 8.4 Hz, 1H), 3.85 (s, 3H), 3.13 (s, 3H), 2.52 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.2, 159.8, 159.1, 147.5, 142.4, 140.1, 137.4, 131.9, 130.1, 129.7, 129.6, 127.1, 124.4, 114.6, 114.3, 107.1, 55.4, 37.5, 15.1. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{21}\text{H}_{22}\text{N}_3\text{O}_2\text{S}$, 380.1427; found, 380.1430.

4'-methoxy- N' -methyl-5-phenoxy- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3fa** (72 mg, yield = 85%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white solid; mp 149–151 °C. ^1H NMR (400 MHz, CDCl_3): δ 8.12 (dd, J = 5.0, 1.1 Hz, 1H), 7.73 (d, J = 8.5 Hz, 1H), 7.43–7.35 (m, 5H), 7.29 (s, 1H), 7.17 (t, J = 7.4 Hz, 1H), 7.11–7.06 (m, 2H), 7.02 (dd, J = 8.5, 2.5 Hz, 1H), 6.93–6.99 (m, 3H), 6.67 (dd, J = 6.6, 5.2 Hz, 1H), 6.39 (d, J = 8.6 Hz, 1H), 3.84 (s, 3H), 3.16 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.2, 159.8, 159.3, 159.1, 155.9, 147.5, 141.7, 137.4, 131.8, 131.1, 130.0, 130.0, 127.9, 124.2, 119.7, 119.5, 116.8, 114.6, 114.3, 107.1, 55.4, 37.5. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{26}\text{H}_{24}\text{N}_3\text{O}_3$, 426.1812; found, 426.1813.

4'-methoxy- N' -methyl- N' -(pyridin-2-yl)-5-(trifluoromethoxy)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ga** (60 mg, yield = 72%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a light yellow foam. ^1H NMR (400 MHz, CDCl_3): δ 8.12 (dd, J = 5.0, 1.0 Hz, 1H), 7.76 (d, J = 8.5 Hz, 1H), 7.41 (d, J = 8.7 Hz, 2H), 7.39–7.33 (m, 1H), 7.30–7.26 (m, 2H), 7.24 (s, 1H), 7.00 (d, J = 8.7 Hz, 2H), 6.67 (ddd, J = 7.1, 5.0, 0.4 Hz, 1H), 6.33 (d, J = 8.5 Hz, 1H), 3.86 (s, 3H), 3.14 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 167.6, 160.1, 158.9, 150.4, 147.6, 141.9, 137.5, 132.0, 130.9, 130.8, 130.1, 122.2, 120.4 (q, J = 256.8 Hz), 119.4, 114.8, 114.5, 107.1, 55.4, 37.7. ^{19}F NMR (376 MHz, CDCl_3): δ –57.60. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{21}\text{H}_{19}\text{F}_3\text{N}_3\text{O}_3$, 418.1373; found, 418.1374.

4''-methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1':3',1''-terphenyl]-4'-carbohydrazide. Compound **3ha** (65 mg, yield = 80%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white solid; mp 80–82 °C. ^1H NMR (400 MHz, CDCl_3): δ 8.13 (dd, J = 5.0, 1.0 Hz, 1H), 7.81 (d, J = 8.0 Hz, 1H), 7.69–

7.58 (m, 4H), 7.51–7.43 (m, 4H), 7.43–7.36 (m, 2H), 7.35 (s, 1H), 7.01 (d, $J = 8.7$ Hz, 2H), 6.66 (dd, $J = 6.5, 5.1$ Hz, 1H), 6.35 (d, $J = 8.5$ Hz, 1H), 3.86 (s, 3H), 3.16 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.5, 159.8, 159.1, 147.5, 143.5, 140.1, 139.9, 137.4, 132.3, 132.2, 130.2, 129.7, 129.0, 128.9, 128.0, 127.2, 126.0, 114.6, 114.3, 107.2, 55.4, 37.6. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{26}\text{H}_{24}\text{N}_3\text{O}_2$, 410.1863; found, 410.1862.

5-fluoro-4'-methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ia** (51 mg, yield = 73%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.10 (ddd, $J = 4.8, 1.6, 0.8$ Hz, 1H), 7.70 (dd, $J = 8.5, 5.8$ Hz, 1H), 7.42–7.29 (m, 4H), 7.14–7.03 (m, 2H), 6.98 (d, $J = 8.7$ Hz, 2H), 6.65 (ddd, $J = 7.1, 5.0, 0.6$ Hz, 1H), 6.31 (d, $J = 8.5$ Hz, 1H), 3.84 (s, 3H), 3.12 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 167.9, 163.5 (d, $J = 251.6$ Hz), 160.0, 159.0, 147.5, 142.2 (d, $J = 8.4$ Hz), 137.4, 131.3 (d, $J = 9.1$ Hz), 131.1 (d, $J = 1.6$ Hz), 130.0, 129.7 (d, $J = 3.1$ Hz), 117.0 (d, $J = 21.9$ Hz), 114.7, 114.4 (d, $J = 3.1$ Hz), 114.3, 107.1, 55.4, 37.6. ^{19}F NMR (376 MHz, CDCl_3): δ -109.23. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{19}\text{FN}_3\text{O}_2$, 352.1456; found, 352.1458.

5-chloro-4'-methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ja** (50 mg, yield = 68%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.09 (dd, $J = 5.0, 1.0$ Hz, 1H), 7.62 (dd, $J = 7.6, 1.0$ Hz, 1H), 7.48 (s, 1H), 7.40–7.31 (m, 5H), 6.96 (d, $J = 8.7$ Hz, 2H), 6.65 (dd, $J = 7.2, 4.8$ Hz, 1H), 6.29 (d, $J = 8.5$ Hz, 1H), 3.84 (s, 3H), 3.10 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 167.8, 160.0, 158.9, 147.5, 141.3, 137.4, 136.5, 131.9, 130.9, 130.4, 130.2, 130.1, 127.5, 114.8, 114.4, 107.1, 55.4, 37.6. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{19}\text{ClN}_3\text{O}_2$, 368.1160; found, 368.1159.

5-bromo-4'-methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ka** (59 mg, yield = 72%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white solid; mp 86–88 °C. ^1H NMR (500 MHz, CDCl_3): δ 8.10 (d, $J = 3.9$ Hz, 1H), 7.58–7.52 (m, 3H), 7.39–7.32 (m, 4H), 6.97 (d, $J = 8.6$ Hz, 2H), 6.65 (dd, $J = 6.8, 5.2$ Hz, 1H), 6.28 (d, $J = 8.5$ Hz, 1H), 3.84 (s, 3H), 3.11 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 167.9, 159.9, 158.9, 147.4, 141.4, 137.4, 133.0, 132.3, 130.7, 130.4, 130.3, 130.1, 124.8, 114.7, 114.3, 107.1, 55.4, 37.6. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{19}\text{BrN}_3\text{O}_2$, 412.0655; found, 412.0658.

4'-methoxy- N' -methyl- N' -(pyridin-2-yl)-5-(trifluoromethyl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3la** (62 mg, yield = 77%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.10 (dd, $J = 5.0, 1.0$ Hz, 1H), 7.80 (d, $J = 7.9$ Hz, 1H), 7.66 (d, $J = 9.7$ Hz, 1H), 7.65 (s, 1H), 7.48 (s, 1H), 7.41 (d, $J = 8.7$ Hz, 2H), 7.37 (ddd, $J = 8.8, 7.2, 1.9$ Hz, 1H), 7.00 (d, $J = 8.7$ Hz, 2H), 6.67 (ddd, $J = 7.1, 5.0, 0.7$ Hz, 1H), 6.30 (d, $J = 8.5$ Hz, 1H), 3.86 (s, 3H), 3.13 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 167.6, 160.1, 158.8, 147.5, 140.4, 137.5, 136.8, 132.5 (q, $J = 32.7$ Hz), 130.7, 130.2, 129.5, 127.1 (d, $J = 3.8$ Hz), 124.0 (d, $J = 3.7$ Hz), 123.6 (q, $J = 267.9$ Hz), 114.9, 114.4, 107.1, 55.4, 37.8. ^{19}F NMR (376 MHz, CDCl_3): δ -62.87. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{21}\text{H}_{19}\text{F}_3\text{N}_3\text{O}_2$, 402.1424; found, 402.1423.

Methyl 4'-methoxy-6-(2-methyl-2-(pyridin-2-yl)hydrazine-1-carbonyl)-[1,1'-biphenyl]-3-carboxylate. Compound **3ma** (63 mg, yield = 80%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (400 MHz, CDCl_3):

δ 8.13 (ddd, $J = 5.0, 1.8, 0.8$ Hz, 1H), 8.09–8.05 (m, 2H), 7.78 (d, $J = 8.4$ Hz, 1H), 7.44 (d, $J = 8.8$ Hz, 2H), 7.40–7.33 (m, 1H), 7.28 (s, 1H), 7.00 (d, $J = 8.7$ Hz, 2H), 6.67 (ddd, $J = 7.1, 5.0, 0.8$ Hz, 1H), 6.31 (d, $J = 8.5$ Hz, 1H), 3.95 (s, 3H), 3.86 (s, 3H), 3.16 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.0, 166.2, 160.0, 158.9, 147.6, 139.8, 137.5, 137.5, 132.0, 131.4, 131.2, 130.2, 129.1, 128.2, 114.9, 114.4, 107.2, 55.4, 52.5, 37.8. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{22}\text{H}_{22}\text{N}_3\text{O}_4$, 392.1605; found, 392.1603.

4'-methoxy- N' ,4-dimethyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3na** (47 mg, yield = 67%) was isolated (petroleum ether/EtOAc = 10:1 to 4:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.12 (ddd, $J = 5.0, 1.8, 0.8$ Hz, 1H), 7.52 (d, $J = 0.5$ Hz, 1H), 7.38 (d, $J = 8.7$ Hz, 2H), 7.37–7.30 (m, 2H), 7.27 (d, $J = 7.1$ Hz, 1H), 7.17 (s, 1H), 6.98 (d, $J = 8.7$ Hz, 2H), 6.64 (ddd, $J = 7.1, 5.0, 0.8$ Hz, 1H), 6.31 (d, $J = 8.5$ Hz, 1H), 3.84 (s, 3H), 3.13 (s, 3H), 2.43 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.9, 159.5, 159.1, 147.4, 137.4, 137.3, 136.6, 133.3, 132.2, 131.4, 130.2, 130.1, 129.4, 114.6, 114.2, 107.2, 55.4, 37.5, 20.9. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{21}\text{H}_{22}\text{N}_3\text{O}_2$, 348.1707; found, 348.1708.

4-chloro-4'-methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3oa** (51 mg, yield = 70%) was isolated (petroleum ether/EtOAc = 10:1 to 4:1) as a white solid; mp 96–98 °C. ^1H NMR (400 MHz, CDCl_3): δ 8.13 (dd, $J = 4.9, 1.0$ Hz, 1H), 7.71 (d, $J = 2.2$ Hz, 1H), 7.48 (dd, $J = 8.3, 2.2$ Hz, 1H), 7.43–7.35 (m, 3H), 7.33 (d, $J = 8.3$ Hz, 1H), 7.19 (s, 1H), 6.99 (d, $J = 8.7$ Hz, 2H), 6.68 (dd, $J = 6.9, 5.2$ Hz, 1H), 6.32 (d, $J = 8.5$ Hz, 1H), 3.85 (s, 3H), 3.14 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 167.3, 159.9, 158.9, 147.6, 138.0, 137.5, 134.9, 133.6, 131.7, 131.0, 130.7, 130.1, 129.0, 114.9, 114.4, 107.2, 55.4, 37.7. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{19}\text{ClN}_3\text{O}_2$, 368.1160; found, 368.1160.

4'-methoxy- N' ,3-dimethyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3pa** (43 mg, yield = 62%) was isolated (petroleum ether/EtOAc = 10:1 to 4:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.09 (ddd, $J = 5.0, 1.8, 0.8$ Hz, 1H), 7.41 (d, $J = 8.8$ Hz, 2H), 7.36 (t, $J = 7.6$ Hz, 1H), 7.25–7.16 (m, 4H), 6.97 (d, $J = 8.8$ Hz, 2H), 6.62 (ddd, $J = 7.1, 5.0, 0.8$ Hz, 1H), 5.93 (d, $J = 8.5$ Hz, 1H), 3.85 (s, 3H), 3.10 (s, 3H), 2.49 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.9, 159.4, 159.0, 147.4, 139.5, 137.3, 136.2, 133.9, 132.7, 130.3, 129.5, 129.0, 127.3, 114.6, 114.0, 107.1, 55.4, 37.5, 19.4. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{21}\text{H}_{22}\text{N}_3\text{O}_2$, 348.1707; found, 348.1707.

3-chloro-4'-methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3qa** (43 mg, yield = 40%) was isolated (petroleum ether/EtOAc = 10:1 to 4:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.10 (ddd, $J = 5.0, 1.8, 0.8$ Hz, 1H), 7.47–7.37 (m, 4H), 7.33–7.27 (m, 2H), 7.21 (ddd, $J = 8.8, 7.2, 1.9$ Hz, 1H), 6.97 (d, $J = 8.8$ Hz, 2H), 6.62 (ddd, $J = 7.1, 5.0, 0.8$ Hz, 1H), 5.99 (d, $J = 8.5$ Hz, 1H), 3.85 (s, 3H), 3.16 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 166.0, 159.8, 159.0, 147.3, 141.8, 137.3, 133.5, 132.1, 131.1, 130.6, 130.4, 128.4, 128.2, 114.7, 114.1, 107.3, 55.4, 37.5. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{19}\text{ClN}_3\text{O}_2$, 368.1160; found, 368.1161.

3-(4-methoxyphenyl)- N' -methyl- N' -(pyridin-2-yl)-2-naphthohydrazide. Compound **3ra** (57 mg, yield = 75%) was isolated (petroleum ether/EtOAc = 10:1 to 4:1) as a white solid; mp 188–190 °C. ^1H NMR (400 MHz, CDCl_3): δ 8.34 (s, 1H), 8.08 (d, $J = 4.5$ Hz, 1H), 7.98 (br, 1H), 7.93 (d, $J = 7.9$ Hz, 1H), 7.85 (d, $J = 7.9$ Hz, 1H), 7.80 (s, 1H), 7.61–7.47

(m, 4H), 7.39 (t, $J = 7.2$ Hz, 1H), 6.99 (d, $J = 8.5$ Hz, 2H), 6.62 (t, $J = 6.0$ Hz, 1H), 6.38 (d, $J = 8.6$ Hz, 1H), 3.85 (s, 3H), 3.20 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.6, 159.6, 159.1, 147.5, 137.4, 136.4, 134.0, 132.4, 131.9, 131.7, 130.3, 129.5, 129.3, 128.3, 127.9, 127.8, 126.8, 114.6, 114.3, 107.2, 55.4, 37.6. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{24}\text{H}_{22}\text{N}_3\text{O}_2$, 384.1707; found, 384.1706.

3-(4-methoxyphenyl)- N' -methyl- N' -(pyridin-2-yl)furan-2-carbohydrazide. Compound **3sa** (11 mg, yield = 16%) was isolated (petroleum ether/EtOAc = 10:1 to 4:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.35 (s, 1H), 8.20 (dd, $J = 5.0, 1.0$ Hz, 1H), 7.72 (d, $J = 8.9$ Hz, 2H), 7.53–7.45 (m, 2H), 6.92 (d, $J = 8.9$ Hz, 2H), 6.79 (d, $J = 8.5$ Hz, 1H), 6.73–6.66 (m, 2H), 3.81 (s, 3H), 3.43 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 159.8, 159.3, 158.1, 147.6, 143.4, 139.7, 137.6, 132.7, 130.7, 123.4, 114.6, 114.4, 113.7, 107.1, 55.3, 38.8. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{18}\text{H}_{18}\text{N}_3\text{O}_3$, 324.1343; found, 324.1342.

4'-acetyl- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide compound **3ab** (47 mg, yield = 68%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.12 (dd, $J = 5.0, 1.0$ Hz, 1H), 8.03 (d, $J = 8.4$ Hz, 2H), 7.74 (dd, $J = 7.6, 1.3$ Hz, 1H), 7.61 (d, $J = 8.4$ Hz, 2H), 7.60–7.54 (m, 1H), 7.51 (td, $J = 7.5, 1.3$ Hz, 1H), 7.44 (dd, $J = 7.6, 1.0$ Hz, 1H), 7.43 (s, 1H), 7.36 (ddd, $J = 8.7, 7.2, 1.9$ Hz, 1H), 6.67 (dd, $J = 6.5, 5.0$ Hz, 1H), 6.36 (d, $J = 8.5$ Hz, 1H), 3.16 (s, 3H), 2.64 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 197.6, 168.3, 158.9, 147.6, 144.7, 139.0, 137.5, 136.4, 133.8, 130.8, 130.2, 129.3, 128.8, 128.7, 128.4, 114.9, 107.1, 38.0, 26.7. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{21}\text{H}_{20}\text{N}_3\text{O}_2$, 346.1550; found, 346.1546.

Methyl 2'-(2-methyl-2-(pyridin-2-yl)hydrazine-1-carbonyl)-[1,1'-biphenyl]-4-carboxylate. Compound **3ac** (51 mg, yield = 70%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.12 (d, $J = 7.9$ Hz, 3H), 7.75 (d, $J = 7.4$ Hz, 1H), 7.61–7.54 (m, 3H), 7.51 (t, $J = 7.3$ Hz, 1H), 7.43 (d, $J = 7.5$ Hz, 1H), 7.40–7.29 (m, 2H), 6.67 (dd, $J = 6.8, 5.2$ Hz, 1H), 6.35 (d, $J = 8.6$ Hz, 1H), 3.95 (s, 3H), 3.13 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.3, 166.7, 158.9, 147.5, 144.5, 139.0, 137.5, 133.7, 130.7, 130.2, 130.0, 129.6, 129.0, 128.8, 128.3, 114.8, 107.1, 52.3, 37.8. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{21}\text{H}_{20}\text{N}_3\text{O}_3$, 362.1499; found, 362.1500.

4'-bromo- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ad** (52 mg, yield = 68%) was isolated (petroleum ether/EtOAc = 10:1 to 4:1) as a yellow oil. ^1H NMR (400 MHz, CDCl_3): δ 8.13 (ddd, $J = 5.0, 1.8, 0.8$ Hz, 1H), 7.70 (dd, $J = 7.6, 1.2$ Hz, 1H), 7.58 (d, $J = 8.5$ Hz, 2H), 7.57–7.52 (m, 1H), 7.48 (td, $J = 7.5, 1.4$ Hz, 1H), 7.44–7.34 (m, 5H), 6.69 (ddd, $J = 7.2, 5.0, 0.8$ Hz, 1H), 6.26 (d, $J = 8.5$ Hz, 1H), 3.18 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.4, 158.9, 147.5, 138.8, 138.8, 137.6, 133.7, 131.9, 130.7, 130.6, 130.1, 128.8, 128.0, 122.4, 114.9, 107.1, 37.9. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{19}\text{H}_{17}\text{BrN}_3\text{O}$, 382.0550; found, 382.0549.

N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ae** (42 mg, yield = 70%) was isolated (petroleum ether/EtOAc = 10:1 to 5:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.12 (ddd, $J = 4.9, 1.8, 0.8$ Hz, 1H), 7.74 (dd, $J = 7.6, 1.2$ Hz, 1H), 7.55 (td, $J = 7.5, 1.5$ Hz, 1H), 7.51–7.40 (m, 7H), 7.35 (ddd, $J = 8.9, 7.2, 1.9$ Hz, 1H), 7.12 (s, 1H), 6.64 (ddd, $J = 7.1, 5.0, 0.7$ Hz, 1H), 6.29 (d, $J = 8.5$ Hz, 1H), 3.08 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3):

δ 168.5, 159.1, 147.5, 140.0, 140.0, 137.4, 133.6, 130.6, 130.2, 128.97, 128.95, 128.9, 128.1, 127.8, 114.6, 107.1, 37.4. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{19}\text{H}_{18}\text{N}_3\text{O}$, 304.1444; found, 304.1445.

$N',4'$ -dimethyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3af** (55 mg, yield = 87%) was isolated (petroleum ether/EtOAc = 10:1 to 4:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.12 (dd, $J = 5.0, 1.0$ Hz, 1H), 7.74 (dd, $J = 7.6, 1.2$ Hz, 1H), 7.53 (td, $J = 7.5, 1.4$ Hz, 1H), 7.45 (td, $J = 7.5, 1.3$ Hz, 1H), 7.41–7.32 (m, 4H), 7.29 (s, 1H), 7.28 (d, $J = 7.9$ Hz, 1H), 7.17 (s, 1H), 6.65 (ddd, $J = 7.1, 5.0, 0.7$ Hz, 1H), 6.31 (d, $J = 8.5$ Hz, 1H), 3.12 (s, 3H), 2.43 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.6, 159.1, 147.4, 139.8, 137.9, 137.3, 137.0, 133.5, 130.6, 130.2, 129.5, 128.9, 128.8, 127.5, 114.5, 107.1, 37.3, 21.2. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{20}\text{N}_3\text{O}$, 318.1601; found, 318.1595.

4'-(*tert*-butyl)- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ag** (58 mg, yield = 81%) was isolated (petroleum ether/EtOAc = 10:1 to 4:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.11 (dd, $J = 4.8, 2.0$ Hz, 1H), 7.74 (dd, $J = 7.5, 0.9$ Hz, 1H), 7.53 (td, $J = 7.5, 1.4$ Hz, 1H), 7.51–7.40 (m, 6H), 7.37 (ddd, $J = 8.7, 7.2, 1.9$ Hz, 1H), 7.07 (s, 1H), 6.64 (ddd, $J = 7.2, 5.2, 0.4$ Hz, 1H), 6.41 (d, $J = 8.5$ Hz, 1H), 3.03 (s, 3H), 1.36 (s, 9H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.7, 159.1, 151.3, 147.4, 139.8, 137.4, 136.9, 133.6, 130.6, 130.2, 129.0, 128.6, 127.6, 125.8, 114.6, 107.2, 37.1, 34.6, 31.3. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{23}\text{H}_{26}\text{N}_3\text{O}$, 360.2070; found, 360.2071.

3'-methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ah** (43 mg, yield = 64%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a white foam. ^1H NMR (400 MHz, CDCl_3): δ 8.11 (dd, $J = 4.9, 1.1$ Hz, 1H), 7.74 (dd, $J = 7.6, 1.3$ Hz, 1H), 7.53 (td, $J = 7.5, 1.4$ Hz, 1H), 7.47 (td, $J = 7.5, 1.3$ Hz, 1H), 7.43–7.33 (m, 3H), 7.19 (s, 1H), 7.06 (d, $J = 7.6$ Hz, 1H), 7.02–6.95 (m, 2H), 6.64 (dd, $J = 6.8, 5.3$ Hz, 1H), 6.31 (d, $J = 8.5$ Hz, 1H), 3.81 (s, 3H), 3.10 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.5, 159.8, 159.1, 147.5, 141.3, 139.7, 137.4, 133.6, 130.6, 130.0, 130.0, 129.0, 127.8, 121.2, 114.6, 114.4, 113.8, 107.1, 55.3, 37.3. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{20}\text{N}_3\text{O}_2$, 334.1550; found, 334.1550.

2'-methoxy- N' -methyl- N' -(pyridin-2-yl)-[1,1'-biphenyl]-2-carbohydrazide. Compound **3ai** (40 mg, yield = 60%) was isolated (petroleum ether/EtOAc = 10:1 to 3:1) as a light yellow foam. ^1H NMR (400 MHz, CDCl_3): δ 8.12 (dd, $J = 4.9, 1.1$ Hz, 1H), 7.81 (dd, $J = 7.6, 1.3$ Hz, 1H), 7.57–7.39 (m, 4H), 7.36–7.27 (m, 3H), 7.08 (t, $J = 7.4$ Hz, 1H), 7.01 (d, $J = 8.3$ Hz, 1H), 6.63 (dd, $J = 6.9, 5.2$ Hz, 1H), 6.25 (d, $J = 8.5$ Hz, 1H), 3.78 (s, 3H), 3.06 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 168.0, 159.2, 156.2, 147.4, 137.4, 136.1, 134.3, 131.0, 131.0, 130.5, 129.8, 129.2, 128.7, 127.8, 121.2, 114.4, 111.1, 107.1, 55.5, 37.3. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{20}\text{N}_3\text{O}_2$, 334.1550; found, 334.1546.

4'-methoxy-[1,1'-biphenyl]-2-carboxamide. Compound **4a** (20 mg, yield = 88%) was isolated (petroleum ether/EtOAc = 5:1 to 3:1) as a white solid; mp 108–110 °C. ^1H NMR (400 MHz, CDCl_3): δ 7.76 (dd, $J = 7.7, 1.2$ Hz, 1H), 7.48 (td, $J = 7.5, 1.5$ Hz, 1H), 7.42–7.32 (m, 4H), 6.96 (d, $J = 8.8$ Hz, 2H), 5.70 (br, 1H), 5.30 (br, 1H), 3.84 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 171.5, 159.4, 139.5, 134.2, 132.4, 130.5, 130.4, 129.9, 129.1, 127.2, 114.1, 55.3. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{14}\text{H}_{14}\text{NO}_2$, 228.1019; found, 228.1018.

4-chloro-4'-methoxy-[1,1'-biphenyl]-2-carboxamide. Compound **4b** (25 mg, yield = 95%) was isolated (petroleum ether/EtOAc = 5:1 to 3:1) as a white solid; mp 154–156 °C. ^1H NMR (400 MHz, CDCl_3): δ 7.74 (d, J = 2.3 Hz, 1H), 7.42 (dd, J = 8.3, 2.3 Hz, 1H), 7.31 (d, J = 8.8 Hz, 2H), 7.24 (s, 1H), 6.94 (d, J = 8.8 Hz, 2H), 5.63 (br, 1H), 5.28 (br, 1H), 3.83 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 169.8, 159.7, 137.9, 135.5, 133.4, 131.8, 131.2, 130.5, 129.9, 129.2, 114.3, 55.3. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{14}\text{H}_{13}\text{ClNO}_2$, 262.0629; found, 262.0629.

3-(4-methoxyphenyl)-2-naphthamide. Compound **4c** (26 mg, yield = 94%) was isolated (petroleum ether/EtOAc = 5:1 to 3:1) as a white solid; mp 224–226 °C. ^1H NMR (400 MHz, CDCl_3): δ 8.33 (s, 1H), 7.92 (d, J = 7.9 Hz, 1H), 7.85 (d, J = 8.0 Hz, 1H), 7.79 (s, 1H), 7.60–7.50 (m, 2H), 7.46 (d, J = 8.7 Hz, 2H), 6.99 (d, J = 8.7 Hz, 2H), 5.77 (br, 1H), 5.43 (br, 1H), 3.86 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 171.2, 159.5, 136.5, 134.0, 132.5, 132.4, 131.7, 130.2, 129.8, 129.4, 128.5, 127.8, 127.6, 126.6, 114.2, 55.3. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{18}\text{H}_{16}\text{NO}_2$, 278.1176; found, 278.1174.

[1,1'-biphenyl]-2-carboxamide. Compound **4d** (18 mg, yield = 89%) was isolated (petroleum ether/EtOAc = 5:1 to 3:1) as a white solid. ^1H NMR (400 MHz, CDCl_3): δ 7.77 (dd, J = 7.6, 1.2 Hz, 1H), 7.50 (td, J = 7.5, 1.5 Hz, 1H), 7.46–7.38 (m, 6H), 7.36 (dd, J = 7.6, 1.1 Hz, 1H), 5.79 (s, 1H), 5.29 (s, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3): δ 171.4, 140.2, 139.8, 134.3, 130.5, 130.4, 129.0, 128.7, 128.6, 127.9, 127.6. HRMS m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{13}\text{H}_{11}\text{NO}$, 198.0913; found, 198.0914.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.1c02481>.

Copies of ^1H , ^{13}C , and ^{19}F NMR spectra for all new compounds (PDF)

■ AUTHOR INFORMATION

Corresponding Author

Hongbin Zhai – The State Key Laboratory of Chemical Oncogenomics, Guangdong Provincial Key Laboratory of Nano-Micro Materials Research, School of Chemical Biology and Biotechnology, Shenzhen Graduate School of Peking University, Shenzhen 518055, China; Institute of Marine Biomedicine, Shenzhen Polytechnic, Shenzhen 518055, China; orcid.org/0000-0003-2198-1357; Email: zhaih@pku.edu.cn

Authors

Jian Wei – The State Key Laboratory of Chemical Oncogenomics, Guangdong Provincial Key Laboratory of Nano-Micro Materials Research, School of Chemical Biology and Biotechnology, Shenzhen Graduate School of Peking University, Shenzhen 518055, China

Xiaoru Shao – The State Key Laboratory of Chemical Oncogenomics, Guangdong Provincial Key Laboratory of Nano-Micro Materials Research, School of Chemical Biology and Biotechnology, Shenzhen Graduate School of Peking University, Shenzhen 518055, China

Hua Zhao – Institute of Drug Discovery Technology, QianXuesen Collaborative Research Center of Astrochemistry

and Space Life Sciences, Ningbo University, Ningbo 315211 Zhejiang, China; orcid.org/0000-0003-1169-6149

Hongjian Yang – The State Key Laboratory of Chemical Oncogenomics, Guangdong Provincial Key Laboratory of Nano-Micro Materials Research, School of Chemical Biology and Biotechnology, Shenzhen Graduate School of Peking University, Shenzhen 518055, China

Shuxian Qiu – The State Key Laboratory of Chemical Oncogenomics, Guangdong Provincial Key Laboratory of Nano-Micro Materials Research, School of Chemical Biology and Biotechnology, Shenzhen Graduate School of Peking University, Shenzhen 518055, China

Complete contact information is available at:

<https://pubs.acs.org/10.1021/acsomega.1c02481>

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We thank the State Key Basic Research Program of the PRC (2018YFC0310900), National Natural Science Foundation of China (21871018, 21732001, and 21672017), Guangdong Science and Technology Program (2017B030314002), Shenzhen Science and Technology Program (grant no. KQTD20190929174023858), Shenzhen Science and Technology Innovation Committee (GXWD202021231165807007-20200812100115001 and JCYJ20180504165454447), Industry and Information Technology Bureau of Shenzhen Municipality (201806151622209330), the third foster plan in 2019 “Molecular Imaging Probe Preparation and Characterization of Key Technologies and Equipment” for the development of key technologies and equipment in major science and technology infrastructure in Shenzhen, and the National Ten Thousand Talent Program (the Leading Talent Tier) for the financial support.

■ REFERENCES

- (1) For selected reviews on transition-metal-catalyzed C–H activations, see: (a) Chen, X.; Engle, K. M.; Wang, D.-H.; Yu, J.-Q. Palladium(II)-Catalyzed C–H Activation/C–C Cross-coupling Reactions: Versatility and Practicality. *Angew. Chem., Int. Ed.* **2009**, *48*, 5094–5115. (b) Daugulis, O.; Do, H.-Q.; Shabashov, D. Palladium- and Copper-Catalyzed Arylation of Carbon-Hydrogen Bonds. *Acc. Chem. Res.* **2009**, *42*, 1074–1086. (c) Colby, D. A.; Bergman, R. G.; Ellman, J. A. Rhodium-Catalyzed C–C Bond Formation via Heteroatom-Directed C–H Bond Activation. *Chem. Rev.* **2010**, *110*, 624–655. (d) Mkhaliid, I. A. I.; Barnard, J. H.; Marder, T. B.; Murphy, J. M.; Hartwig, J. F. C–H Activation for the Construction of C–B Bonds. *Chem. Rev.* **2010**, *110*, 890–931. (e) Cho, S. H.; Kim, J. Y.; Kwak, J.; Chang, S. Recent Advances in the Transition Metal-Catalyzed Twofold Oxidative C–H Bond Activation Strategy for C–C and C–N Bond Formation. *Chem. Soc. Rev.* **2011**, *40*, 5068–5083. (f) Neufeldt, S. R.; Sanford, M. S. Controlling Site Selectivity in Palladium-Catalyzed C–H Bond Functionalization. *Acc. Chem. Res.* **2012**, *45*, 936–946. (g) Rouquet, G.; Chatani, N. Catalytic Functionalization of $\text{C}(\text{sp}^2)\text{-H}$ and $\text{C}(\text{sp}^3)\text{-H}$ Bonds by Using Bidentate Directing Groups. *Angew. Chem., Int. Ed.* **2013**, *52*, 11726–11743. (h) Zhang, M.; Zhang, Y.; Jie, X.; Zhao, H.; Li, G.; Su, W. Recent Advances in Directed C–H Functionalizations Using Monodentate Nitrogen-Based Directing Groups. *Org. Chem. Front.* **2014**, *1*, 843–895. (i) Chen, Z.; Wang, B.; Zhang, J.; Yu, W.; Liu, Z.; Zhang, Y. Transition Metal-Catalyzed C–H Bond Functionalizations by the Use of Diverse Directing Groups. *Org. Chem. Front.* **2015**, *2*, 1107–1295. (j) Gandeepan, P.; Cheng, C.-H. Transition-Metal-Catalyzed π -Bond-Assisted C–H Bond Functionalization: An Emerg-

- ing Trend in Organic Synthesis. *Chem.—Asian J.* **2015**, *10*, 824–838.
- (k) Su, B.; Cao, Z.-C.; Shi, Z.-J. Exploration of Earth-Abundant Transition Metals (Fe, Co, and Ni) as Catalysts in Unreactive Chemical Bond Activations. *Acc. Chem. Res.* **2015**, *48*, 886–896.
- (l) Gensch, T.; Hopkinson, M. N.; Glorius, F.; Wencel-Delord, J. Mild Metal-Catalyzed C-H Activation: Examples and Concepts. *Chem. Soc. Rev.* **2016**, *45*, 2900–2936.
- (m) Wang, F.; Yu, S.; Li, X. Transition Metal-Catalyzed Couplings Between Arenes and Strained or Reactive Rings: Combination of C-H Activation and Ring Scission. *Chem. Soc. Rev.* **2016**, *45*, 6462–6477.
- (n) Dong, Z.; Ren, Z.; Thompson, S. J.; Xu, Y.; Dong, G. Transition-Metal-Catalyzed C-H Alkylation Using Alkenes. *Chem. Rev.* **2017**, *117*, 9333–9403.
- (o) Ma, W.; Gandeepan, P.; Li, J.; Ackermann, L. Recent Advances in Positional-Selective Alkenylations: Removable Guidance for Twofold C-H Activation. *Org. Chem. Front.* **2017**, *4*, 1435–1467.
- (p) Gandeepan, P.; Müller, T.; Zell, D.; Cera, G.; Warratz, S.; Ackermann, L. 3d Transition Metals for C-H Activation. *Chem. Rev.* **2019**, *119*, 2192–2452.
- (2) (a) Gutekunst, W. R.; Baran, P. S. C-H Functionalization Logic in Total Synthesis. *Chem. Soc. Rev.* **2011**, *40*, 1976–1991.
- (b) McMurray, L.; O'Hara, F.; Gaunt, M. J. Recent Developments in Natural Product Synthesis Using Metal-Catalyzed C-H Bond Functionalisation. *Chem. Soc. Rev.* **2011**, *40*, 1885–1898.
- (c) Yamaguchi, J.; Yamaguchi, A. D.; Itami, K. C-H Bond Functionalization: Emerging Synthetic Tools for Natural Products and Pharmaceuticals. *Angew. Chem., Int. Ed.* **2012**, *51*, 8960–9009.
- (3) (a) Alberico, D.; Scott, M. E.; Lautens, M. Aryl-Aryl Bond Formation by Transition-Metal-Catalyzed Direct Arylation. *Chem. Rev.* **2007**, *107*, 174–238.
- (b) Seregin, I. V.; Gevorgyan, V. Direct Transition Metal-Catalyzed Functionalization of Heteroaromatic Compounds. *Chem. Soc. Rev.* **2007**, *36*, 1173–1193.
- (c) Shi, Z.-J.; Li, B.-J.; Yang, S.-D. Recent Advances in Direct Arylation via Palladium-Catalyzed Aromatic C-H Activation. *Synlett* **2008**, *2008*, 949–957.
- (d) Ackermann, L.; Vicente, R. n.; Kapdi, A. R. Transition-Metal-Catalyzed Direct Arylation of (Hetero)Arenes by C-H Bond Cleavage. *Angew. Chem., Int. Ed.* **2009**, *48*, 9792–9826.
- (e) Bellina, F.; Rossi, R. Transition Metal-Catalyzed Direct Arylation of Substrates with Activated sp³-Hybridized C-H Bonds and Some of Their Synthetic Equivalents with Aryl Halides and Pseudohalides. *Chem. Rev.* **2010**, *110*, 1082–1146.
- (f) Sun, C.-L.; Li, B.-J.; Shi, Z.-J. Pd-Catalyzed Oxidative Coupling with Organometallic Reagents via C-H Activation. *Chem. Commun.* **2010**, *46*, 677–685.
- (4) (a) Suzuki, A. Carbon-Carbon Bonding Made Easy. *Chem. Commun.* **2005**, 4759–4763.
- (b) Corbet, J.-P.; Mignani, G. Selected Patented Cross-Coupling Reaction Technologies. *Chem. Rev.* **2006**, *106*, 2651–2710.
- (c) Biffis, A.; Centomo, P.; Del Zotto, A.; Zecca, M. Pd Metal Catalysts for Cross-Couplings and Related Reactions in the 21st Century: A Critical Review. *Chem. Rev.* **2018**, *118*, 2249–2295.
- (5) Zaitsev, V. G.; Shabashov, D.; Daugulis, O. Highly Regioselective Arylation of sp³ C-H Bonds Catalyzed by Palladium Acetate. *J. Am. Chem. Soc.* **2005**, *127*, 13154–13155.
- (6) (a) Daugulis, O.; Roane, J.; Tran, L. D. Bidentate, Monoanionic Auxiliary-Directed Functionalization of Carbon-Hydrogen Bonds. *Acc. Chem. Res.* **2015**, *48*, 1053–1064.
- (b) Rej, S.; Ano, Y.; Chatani, N. Bidentate Directing Groups: An Efficient Tool in C-H Bond Functionalization Chemistry for the Expedient Construction of C-C Bonds. *Chem. Rev.* **2020**, *120*, 1788–1887.
- (7) Hassan, J.; Sévignon, M.; Gozzi, C.; Schulz, E.; Lemaire, M. Aryl-Aryl Bond Formation One Century after the Discovery of the Ullmann Reaction. *Chem. Rev.* **2002**, *102*, 1359–1470.
- (b) Bringmann, G.; Price Mortimer, A. J.; Keller, P. A.; Gresser, M. J.; Garner, J.; Breuning, M. Atroposelective Synthesis of Axially Chiral Biaryl Compounds. *Angew. Chem., Int. Ed.* **2005**, *44*, 5384–5427.
- (c) Martin, R.; Buchwald, S. L. Palladium-Catalyzed Suzuki-Miyaura Cross-Coupling Reactions Employing Dialkylbiaryl Phosphine Ligands. *Acc. Chem. Res.* **2008**, *41*, 1461–1473.
- (8) Chattopadhyay, S.; Chattopadhyay, U.; Mathur, P.; Saini, K.; Ghosal, S. Effects of Hippadine, an Amaryllidaceae Alkaloid, on Testicular Function in Rats. *Planta Med.* **1983**, *49*, 252–254.
- (9) (a) Dargad, R. R.; Parekh, J. D.; Dargad, R. R.; Kukrety, S. Azilsartan: Novel Angiotensin Receptor Blocker. *J. Assoc. Physicians India* **2016**, *64*, 96–98.
- (b) Kurtz, T. W.; Kajiya, T. Differential Pharmacology and Benefit/Risk of Azilsartan Compared to Other Sartans. *Vasc. Health Risk Manage.* **2012**, *8*, 133–143.
- (c) Rakugi, H.; Enya, K. Azilsartan: a new angiotensin receptor blocker. *Nihon Rinsho* **2012**, *70*, 1615–1620.
- (10) Ham, J.; You, S.; Lim, W.; Song, G. Bifenthrin Impairs the Functions of Leydig and Sertoli Cells in Mice via Mitochondrion-Endoplasmic Reticulum Dysregulation. *Environ. Pollut.* **2020**, *266*, 115174–115185.
- (11) (a) Huang, L.; Li, Q.; Wang, C.; Qi, C. Palladium(II)-Catalyzed Regioselective Arylation of Naphthylamides with Aryl Iodides Utilizing a Quinolinamide Bidentate System. *J. Org. Chem.* **2013**, *78*, 3030–3038.
- (b) Nadres, E. T.; Santos, G. I. F.; Shabashov, D.; Daugulis, O. Scope and Limitations of Auxiliary-Assisted, Palladium-Catalyzed Arylation and Alkylation of sp² and sp³ C-H Bonds. *J. Org. Chem.* **2013**, *78*, 9689–9714.
- (c) Pearson, R.; Zhang, S.; He, G.; Edwards, N.; Chen, G. Synthesis of Phenanthridines via Palladium-Catalyzed Picolinamide-Directed Sequential C-H Functionalization. *Beilstein J. Org. Chem.* **2013**, *9*, 891–899.
- (d) Han, J.; Liu, P.; Wang, C.; Wang, Q.; Zhang, J.; Zhao, Y.; Shi, D.; Huang, Z.; Zhao, Y. Oxalyl Amide Assisted Palladium-Catalyzed Arylation of C(sp²)-H Bond at the δ Position. *Org. Lett.* **2014**, *16*, 5682–5685.
- (e) Misal Castro, L. C.; Chatani, N. Palladium(II)-Catalyzed *ortho*-C-H Arylation/Alkylation of N-Benzoyl α -Amino Ester Derivatives. *Chem.—Eur. J.* **2014**, *20*, 4548–4553.
- (f) Liu, Y.; Zhang, Y.; Huang, M.; Wan, J.-P. Step Economical Synthesis of *o*-aryl Benzamides via C-H Activation Relayed by the *in situ* Installation of Directing Group: a Multicomponent Method. *RSC Adv.* **2015**, *5*, 46192–46196.
- (g) Zhou, X.; Wang, Q.; Zhao, W.; Xu, S.; Zhang, W.; Chen, J. Palladium-Catalyzed *ortho*-Arylation of Benzoic Acid Derivatives via C-H bond Activation Using an Aminoacetic Acid Bidentate Directing Group. *Tetrahedron Lett.* **2015**, *56*, 851–855.
- (h) Han, J.; Zhang, L.; Zhu, Y.; Zheng, Y.; Chen, X.; Huang, Z.-B.; Shi, D.-Q.; Zhao, Y. Highly Regioselective *meta* Arylation of Oxalyl Amide-Protected β -Arylethylamine via the Catellani Reaction. *Chem. Commun.* **2016**, *52*, 6903–6906.
- (i) Li, Q.; Knight, B. J.; Ferreira, E. M. Palladium(II)-Catalyzed *ortho*-Arylation of Aromatic Alcohols with a Readily Attachable and Cleavable Molecular Scaffold. *Chem.—Eur. J.* **2016**, *22*, 13054–13058.
- (j) Liu, W.; Wang, D.; Zhao, Y.; Yi, F.; Chen, J. Palladium-Catalyzed Mono-Selective *ortho* C-H Arylation of Aryl Sulfonamides in Water: A Concise Access to Biaryl Sulfoamide Derivatives. *Adv. Synth. Catal.* **2016**, *358*, 1968–1974.
- (k) Santrač, D.; Cella, S.; Wang, W.; Ackermann, L. Palladium-Catalyzed C-H Arylation of Amides by Triazole Assistance. *Eur. J. Org. Chem.* **2016**, 5429–5436.
- (l) Li, Q.; Ferreira, E. M. Meta-Selective C-H Arylation of Aromatic Alcohols with a Readily Attachable and Cleavable Molecular Scaffold. *Chem.—Eur. J.* **2017**, *23*, 11519–11523.
- (m) Reddy, M. D.; Blanton, A. N.; Watkins, E. B. Palladium-Catalyzed, *N*-(2-Aminophenyl)acetamide-Assisted *ortho*-Arylation of Substituted Benzamides: Application to the Synthesis of Urolithins B, M6, and M7. *J. Org. Chem.* **2017**, *82*, 5080–5095.
- (n) Shen, Y.; Cindy Lee, W.-C.; Gutierrez, D. A.; Li, J. J. Palladium-Catalyzed Direct C(sp²)-H *ortho*-Arylation of Anilides Using 2-Aminophenylpyrazole as the Directing Group. *J. Org. Chem.* **2017**, *82*, 11620–11625.
- (o) Ji, Y.-F.; Hu, Y.-H.; Xu, Z.; Shao, L.-Y. Palladium-Catalyzed Arylation of Aromatic Amides Directed by a [4-Chloro-2-(1H-pyrazol-1-yl)-phenyl]amine Auxiliary. *Synlett* **2018**, *29*, 1875–1880.
- (p) Zeng, W.; Nukeyeva, M.; Wang, Q.; Jiang, C. Synthesis of Unnatural α -Amino Acid Derivatives via Selective *o*-C-H Functionalization. *Org. Biomol. Chem.* **2018**, *16*, 598–608.
- (12) (a) Aihara, Y.; Chatani, N. Ruthenium-Catalyzed Direct Arylation of C-H Bonds in Aromatic Amides Containing a Bidentate Directing Group: Significant Electronic Effects on Arylation. *Chem. Sci.* **2013**, *4*, 664–670.
- (b) Al Mamari, H. H.; Diers, E.; Ackermann, L. Triazole-Assisted Ruthenium-Catalyzed C-H Arylation of Aromatic Amides. *Chem.—Eur. J.* **2014**, *20*, 9739–9743.

- (13) (a) Grigorjeva, L.; Daugulis, O. Cobalt-Promoted Dimerization of Aminoquinoline Benzamides. *Org. Lett.* **2015**, *17*, 1204–1207. (b) Du, C.; Li, P.-X.; Zhu, X.; Suo, J.-F.; Niu, J.-L.; Song, M.-P. Mixed Directing-Group Strategy: Oxidative C-H/C-H Bond Arylation of Unactivated Arenes by Cobalt Catalysis. *Angew. Chem., Int. Ed.* **2016**, *55*, 13571–13575. (c) Hu, L.; Gui, Q.; Chen, X.; Tan, Z.; Zhu, G. Cobalt-Promoted Selective Arylation of Benzamides and Acrylamides with Arylboronic Acids. *Org. Biomol. Chem.* **2016**, *14*, 11070–11075. (d) Lv, N.; Chen, Z.; Liu, Y.; Liu, Z.; Zhang, Y. Cobalt-Catalyzed Aerobic Oxidative C-H/C-H Cross-Coupling of Unactivated Arenes for the Synthesis of Biaryls. *Org. Lett.* **2018**, *20*, 5845–5848. (e) Bu, Q.; Goñka, E.; Kuciński, K.; Ackermann, L. Cobalt-Catalyzed Hiyama-Type C-H Activation with Arylsiloxanes: Versatile Access to Highly *ortho*-Decorated Biaryls. *Chem.—Eur. J.* **2019**, *25*, 2213–2216.
- (14) (a) Yokota, A.; Aihara, Y.; Chatani, N. Nickel(II)-Catalyzed Direct Arylation of C-H Bonds in Aromatic Amides Containing an 8-Aminoquinoline Moiety as a Directing Group. *J. Org. Chem.* **2014**, *79*, 11922–11932. (b) Liu, B.; Zhang, Z.-Z.; Li, X.; Shi, B.-F. Nickel(II)-Catalyzed Direct Arylation of Aryl C-H Bonds with Aryl-Boron Reagents Directed by a Removable Bidentate Auxiliary. *Org. Chem. Front.* **2016**, *3*, 897–900. (c) Zhao, S.; Liu, B.; Zhan, B.-B.; Zhang, W.-D.; Shi, B.-F. Nickel-Catalyzed *ortho*-Arylation of Unactivated (Hetero)aryl C-H Bonds with Arylsilanes Using a Removable Auxiliary. *Org. Lett.* **2016**, *18*, 4586–4589. (d) Honeycutt, A. P.; Hoover, J. M. Nickel-Catalyzed Oxidative Decarboxylative (Hetero)-Arylation of Unactivated C-H Bonds: Ni and Ag Synergy. *ACS Catal.* **2017**, *7*, 4597–4601. (e) Li, P.; Wang, G.-W.; Chen, H.; Wang, L. Nickel-Catalyzed Regioselective Arylation of Aromatic Amides with Aryl Iodides Enabled by an *N,O*-Bidentate Directing Group. *Org. Biomol. Chem.* **2018**, *16*, 8783–8790.
- (15) (a) Shang, M.; Sun, S.-Z.; Dai, H.-X.; Yu, J.-Q. Cu(OAc)₂-Catalyzed Coupling of Aromatic C-H Bonds with Arylboron Reagents. *Org. Lett.* **2014**, *16*, 5666–5669. (b) Gui, Q.; Chen, X.; Hu, L.; Wang, D.; Liu, J.; Tan, Z. Copper-Mediated *ortho*-Arylation of Benzamides with Arylboronic Acid. *Adv. Synth. Catal.* **2016**, *358*, 509–514. (c) Wang, M.; Hu, Y.; Jiang, Z.; Shen, H. C.; Sun, X. Divergent Copper-Mediated Dimerization and Hydroxylation of Benzamides Involving C-H Bond Functionalization. *Org. Biomol. Chem.* **2016**, *14*, 4239–4246. (d) Takamatsu, K.; Hirano, K.; Miura, M. Copper-Mediated Decarboxylative Coupling of Benzamides with *ortho*-Nitrobenzoic Acids by Directed C-H Cleavage. *Angew. Chem., Int. Ed.* **2017**, *56*, 5353–5357.
- (16) (a) Gu, Q.; Al Mamari, H. H.; Graczyk, K.; Diers, E.; Ackermann, L. Iron-Catalyzed C(sp²)-H and C(sp³)-H Arylation by Triazole Assistance. *Angew. Chem., Int. Ed.* **2014**, *53*, 3868–3871. (b) Shang, R.; Ilies, L.; Asako, S.; Nakamura, E. Iron-Catalyzed C(sp²)-H Bond Functionalization with Organoboron Compounds. *J. Am. Chem. Soc.* **2014**, *136*, 14349–14352.
- (17) (a) Zhai, S.; Qiu, S.; Chen, X.; Wu, J.; Zhao, H.; Tao, C.; Li, Y.; Cheng, B.; Wang, H.; Zhai, H. 2-(1-Methylhydrazinyl)pyridine as a Reductively Removable Directing Group in a Cobalt-Catalyzed C(sp²)-H Bond Alkenylation/Annulation Cascade. *Chem. Commun.* **2018**, *54*, 98–101. (b) Qiu, S.; Zhai, S.; Wang, H.; Tao, C.; Zhao, H.; Zhai, H. Efficient Synthesis of Phthalimides *via* Cobalt-Catalyzed C(sp²)-H Carbonylation of Benzoyl Hydrazides with Carbon Monoxide. *Adv. Synth. Catal.* **2018**, *360*, 3271–3276. (c) Zhai, S.; Qiu, S.; Chen, X.; Tao, C.; Li, Y.; Cheng, B.; Wang, H.; Zhai, H. Trifunctionalization of Allenes *via* Cobalt-Catalyzed MHP-Assisted C-H Bond Functionalization and Molecular Oxygen Activation. *ACS Catal.* **2018**, *8*, 6645–6649. (d) Zhao, H.; Shao, X.; Wang, T.; Zhai, S.; Qiu, S.; Tao, C.; Wang, H.; Zhai, H. A 2-(1-Methylhydrazinyl)-pyridine-Directed C-H Functionalization/Spirocyclization Cascade: Facile Access to Spirosuccinimide Derivatives. *Chem. Commun.* **2018**, *54*, 4927–4930. (e) Zhao, H.; Wang, T.; Qing, Z.; Zhai, H. Cobalt-Catalyzed 2-(1-Methylhydrazinyl)pyridine-Assisted Cyclization of Thiophene-2-Carbohydrazides with Maleimides: Efficient Synthesis of Thiophene-Fused Pyridones. *Chem. Commun.* **2020**, *56*, 5524–5527. (f) Qiu, S.; Zhai, S.; Wang, H.; Chen, X.; Zhai, H. One-Pot Synthesis of Benzo[b]fluorenones *via* a Cobalt-Catalyzed MHP-Directed [3+2] Annulation/Ring-Opening/Dehydration Sequence. *Chem. Commun.* **2019**, *55*, 4206–4209. (g) Zhao, H.; Shao, X.; Qing, Z.; Wang, T.; Chen, X.; Yang, H.; Zhai, H. Cobalt-Catalyzed 2-(1-Methylhydrazinyl)pyridine-Assisted Direct C-H/N-H Functionalization of Benzoyl Hydrazide with Isocyanide: Efficient Synthesis of Iminoisoindolinone Derivatives. *Adv. Synth. Catal.* **2019**, *361*, 1678–1682.
- (18) (a) Fitton, P.; Rick, E. A. The addition of aryl halides to tetrakis(triphenylphosphine) palladium(0). *J. Organomet. Chem.* **1971**, *28*, 287–291. (b) Xue, L.; Lin, Z. Theoretical aspects of palladium catalysed carbon-carbon cross-coupling reactions. *Chem. Soc. Rev.* **2010**, *39*, 1692–1705.
- (19) (a) Shabashov, D.; Daugulis, O. Auxiliary-Assisted Palladium-Catalyzed Arylation and Alkylation of sp² and sp³ Carbon-Hydrogen Bonds. *J. Am. Chem. Soc.* **2010**, *132*, 3965–3972. (b) Arroniz, C.; Denis, J. G.; Ironmonger, A.; Rassias, G.; Larrosa, I. An Organic Cation as a silver(I) Analogue for the Arylation of sp² and sp³ C-H Bonds with Iodoarenes. *Chem. Sci.* **2014**, *5*, 3509–3514. (c) Topczewski, J. J.; Sanford, M. S. Carbon-Hydrogen (C-H) Bond Activation at Pd^{IV}: A Frontier in C-H Functionalization Catalysis. *Chem. Sci.* **2015**, *6*, 70–76.