

Mapping pollution exposure and chemistry during an extreme air quality event (the 2018 Kīlauea eruption) using a low-cost sensor network

Ben Crawford^{a,1,2}, David H. Hagan^{a,3}, llene Grossman^b, Elizabeth Cole^b, Lacey Holland^c, Colette L. Heald^a, and Jesse H. Kroll^{a,2}

^aDepartment of Civil and Environmental Engineering, Massachusetts Institute of Technology, Cambridge, MA 02139; ^bThe Kohala Center, Waimea, Hawai'i Island, HI 96743; and ^cSchool of Ocean and Earth Science and Technology, University of Hawai'i at Mānoa, Honolulu, HI 96822

Edited by Ronald C. Cohen, University of California, Berkeley, CA, and accepted by Editorial Board Member Akkihebbal R. Ravishankara May 4, 2021 (received for review December 11, 2020)

Extreme air quality episodes represent a major threat to human health worldwide but are highly dynamic and exceedingly challenging to monitor. The 2018 Kilauea Lower East Rift Zone eruption (May to August 2018) blanketed much of Hawai'i Island in "vog" (volcanic smog), a mixture of primary volcanic sulfur dioxide (SO₂) gas and secondary particulate matter (PM). This episode was captured by several monitoring platforms, including a low-cost sensor (LCS) network consisting of 30 nodes designed and deployed specifically to monitor PM and SO₂ during the event. Downwind of the eruption, network stations measured peak hourly PM_{2.5} and SO₂ concentrations that exceeded 75 μ g m⁻³ and 1,200 parts per billion (ppb), respectively. The LCS network's high spatial density enabled highly granular estimates of human exposure to both pollutants during the eruption, which was not possible using preexisting air quality measurements. Because of overlaps in population distribution and plume dynamics, a much larger proportion of the island's population was exposed to elevated levels of fine PM than to SO₂. Additionally, the spatially distributed network was able to resolve the volcanic plume's chemical evolution downwind of the eruption. Measurements find a mean SO2 conversion time of ~36 h, demonstrating the ability of distributed LCS networks to observe reaction kinetics and quantify chemical transformations of air pollutants in a real-world setting. This work also highlights the utility of LCS networks for emergency response during extreme episodes to complement existing air quality monitoring approaches.

air quality | low-cost sensors | volcanoes

utdoor air pollution leads to the deaths of millions of Dependence of the single largest environmental risk factor for premature mortality worldwide (1). Air quality (AQ) monitoring is critical to understand and ultimately minimize people's exposure to harmful air pollutants; however, surface-based measurements remain relatively sparse in much of the world (2). Moreover, a substantial (though poorly quantified) fraction of humans' exposure to air pollutants occurs during extreme AQ events in which pollution levels are dramatically elevated relative to mean levels. Examples include the 1948 Donora, Pennsylvania smog event (3), the London Fog of 1952 (4), the Great Smog of Delhi (5), the Beijing "Airpocalypse" in 2013 (6), and recent severe wildfires in North America and Australia (7, 8). During such events, hazardous pollutants such as particulate matter (PM) can be primary (emitted directly) or secondary (formed via atmospheric reactions) (9). The variable confluence of primary emissions, meteorological transport dynamics, and complex secondary chemical processes makes AQ monitoring during extreme episodes exceptionally challenging. There is currently no established approach or strategy to monitor pollutant distribution or human exposure during these episodes.

Here, we track and characterize a recent extreme AQ event, the 2018 lower East Rift Zone (LERZ) eruption of Kīlauea Volcano (Island of Hawai'i, USA, May to August 2018), using a low-cost air quality sensor network. Prior to this event, Kīlauea had been continuously erupting since 1983 (10). As the nearly constant northeasterly trade winds transported the plume downwind around the southern coast of the Island, the primary sulfur dioxide (SO₂) emissions oxidized to form sulfuric acid, leading to elevated levels of fine PM on the island's downwind western side (the Kona coast) (11). The SO₂ and PM (collectively known as "vog," for "volcanic smog") from this effusive eruption had for decades been recognized as a local AQ nuisance and health hazard (12, 13) for the island's ~175,000 residents and even for residents in neighboring islands. Prior to May 2018, both satellite measurements and a regulatory network (five stations operated by Hawai'i Department of Health) showed levels of SO₂ and PM that were substantially elevated above those of background (marine) air (Fig. 1).

In May 2018, the eruption entered an intense phase with Kīlauea experiencing its largest rift eruption in more than 200 y (14). On May 3, eruptive fissures opened in a residential neighborhood in the LERZ, pumping out lava and emitting substantially elevated amounts of SO_2 (>50,000 tons a day⁻¹) (14) directly into a populated area. During the course of the 3-mo–long eruption, lava covered 35.5 km² of land, more than 700 homes

Significance

Poor air quality is a global public health issue, contributing to millions of premature deaths per year worldwide. Low-cost air quality sensors are a promising tool to improve monitoring capabilities. In this study, we built and deployed a low-cost sensor network for emergency response during an extreme air quality event, the 2018 Kīlauea Lower East Rift Zone eruption. This network was used to estimate fine-scale population exposures to multiple pollutants, to measure the chemical transformation of volcanic emissions, and to provide real-time observations as part of emergency management efforts.

Author contributions: B.C., D.H.H., E.C., C.L.H., and J.H.K. designed research; B.C., D.H.H., I.G., L.H., and J.H.K. performed research; B.C. and C.L.H. analyzed data; and B.C. and J.H.K. wrote the paper with contributions from D.H.H., I.G., E.C., L.H., and C.L.H.

The authors declare no competing interest.

This article is a PNAS Direct Submission. R.C.C. is a guest editor invited by the Editorial Board.

This open access article is distributed under Creative Commons Attribution License 4.0 (CC BY).

¹Present address: Department of Geography and Environmental Sciences, University of Colorado, Denver, CO 80204.

²To whom correspondence may be addressed. Email: benjamin.crawford@ucdenver.edu or jhkroll@mit.edu.

³Present address: QuantAQ, Inc., Somerville, MA 02143.

This article contains supporting information online at https://www.pnas.org/lookup/suppl/ doi:10.1073/pnas.2025540118/-/DCSupplemental.

Published June 21, 2021.



Fig. 1. Satellite- and ground-based monitoring of air quality before, during, and after the LERZ eruption. (*A* and *B*) Satellite observations of column integrated SO₂ and aerosol optical depth (AOD) from May to July, comparing the average of 3 y prior to the eruption (2015 to 2017), the year of the eruption (2018), and the year following the eruption (2019). SO₂ (shown in Dobson Units [DU]) and AOD measurements are taken from the Ozone Mapping and Profiler Suite (OMPS) instrument aboard Suomi National Polar-orbiting Partnership (NPP) (50 km product, Version 2) and the Moderate Resolution Imaging Spectroradiometer (MODIS) instrument aboard the Aqua platform (10 km product, Collection 6.1), respectively. Daily satellite observations are gridded and averaged at $0.5^{\circ} \times 0.5^{\circ}$ horizontal resolution. (*C* and *D*) Concentrations of SO₂ and PM_{2.5} for all of 2018, as measured by the Hawai'i Department of Health ground-level regulatory station at Ocean View.

were destroyed, and thousands of residents were displaced. The elevated SO₂ emissions led to exceedingly poor AQ not only in the immediate vicinity of the eruption but also across the wider region. The order-of-magnitude step change in SO₂ emissions and resulting secondary PM was clearly visible from space (Fig. 1 *A* and *B*) and was also measured by the ground-based regulatory network (Fig. 1 *C* and *D*).

The measurements in Fig. 1 established the LERZ eruption as an extreme AQ event, and recent studies have used satellite and in situ measurements (15-17) to explore air quality implications and plume dynamics during the eruption. For example, analysis of regulatory network data found that 24-h average PM_{2.5} concentrations exceeded US Environmental Protection Agency (EPA) AQ thresholds eight times during the eruption in certain locations, compared to zero times during the previous 8 y (17). However, such measurements are typically designed to monitor regional-scale AQ and therefore provide limited details about the fine-scale spatiotemporal distribution of air pollutants. Satellite measurements are limited both spatially and temporally (because of overpass intervals of 1 to 3 d, pixel sizes of ~tens of km, cloud cover, and limited vertical resolution). Ground-based regulatory measurements provide improved temporal resolution and networks are strategically placed to monitor ambient AQ in populated regions but are generally not designed to monitor fine-scale exposure from dynamic plumes during extreme events. On the Island of Hawai'i, the average resident lived ~17 km from the nearest regulatory AQ station (closer than the United States average of 22 km), and while this network provides continuous, high-quality measurements at key locations, this is too sparse for high-resolution estimates of residents' pollutant exposure given the high temporal and spatial variability of the volcanic plume.

Results and Discussion

In order to complement existing measurements and provide improved estimates of the pollutants' spatial variability, human exposures, and rate of interconversion, we built and deployed a network of low-cost sensor (LCS) nodes to measure SO2 and PM throughout the region. The relative affordability and small size of LCS enables many network nodes to be deployed within a small area, thereby providing distributed measurements with a much higher spatial resolution than is possible with traditional AQ networks (18-20). LCS networks have recently been deployed in different locations worldwide (20-23); while LCSs are generally less accurate and precise than regulatory AQ instruments, recent work has demonstrated that LCS calibration via colocation with regulatory-grade monitors can enable robust measurements (24-26). Our deployment during the eruption occurred in two phases. First, an initial small-scale deployment in the LERZ, enabling civil authorities and local residents to monitor SO₂ levels and make emergency management decisions, was carried out beginning May 14. These sensors had already been built and used in Hawai'i (24) and were among the first SO_2 measurements in the area. Next, a total of 30 new sensor units were built (measuring SO₂ and PM), calibrated using the regulatory stations' measurements, and deployed throughout the Island beginning May 23. Details of the sensor unit design and calibration are provided in the Materials and Methods and SI Appendix. After this deployment, the network had one node per ~5,800 people and the average resident lived 4.6 km from an AQ measurement (SI Appendix). By comparison, the US- and globally averaged distances to a regulatory-grade AQ measurement are 22 km and 220 km, respectively (2).

Average SO₂ and PM levels measured by a subset of the network during July 15 to August 1 are shown in Fig. 2. This subset (20 PM_{2.5} and 17 SO₂ sensors) includes all sensors that operated nearly continuously throughout this measurement period (the full timeseries is shown in the *SI Appendix*). Both the regulatory and LCS networks measured similar overall trends in volcanic pollution downwind of the volcano; the highest SO₂ concentrations were observed just downwind of the eruption, whereas the highest PM_{2.5} levels were further downwind, along the western (Kona) coast. LCS network stations measured peak hourly SO₂ concentrations >1,200 parts per billion (ppb) and peak hourly PM_{2.5} concentrations >75 μ g · m⁻³ (*SI Appendix*). The LCS network also revealed pollution gradients in populated areas that were imperceptible to the regulatory network, with substantial variability on finer (5 to 20 km) scales. For example, along the Kona coast, average PM_{2.5} loadings at nearby sites varied by more than a factor of two (from 9 to 25 μ g \cdot m⁻³) during the measurement period, likely because of differences in elevation, topography, and local winds. Sensors were also placed in areas that previously had been unaffected by "vog" and were without long-term monitoring stations, revealing elevated levels of pollutants there as well. For instance, in areas of the northern Kona coast, average PM_{2.5} ranged from 5 to 18 μ g \cdot m⁻³ with several short-lived episodes (<8 h) in which PM_{2.5} exceeded $30 \ \mu g \cdot m^{-3}$.

The high spatial resolution of the LCS network also enables fine-grained estimates of residents' exposure to pollutants in terms of both average exposures (Fig. 3 A and B) and hourly exposure distributions (Fig. 3 C and D). Fine-grained exposure estimates have long been viewed as a primary advantage of distributed sensor networks (e.g., ref. 18), but there exist few, if any, examples of such networks used to quantitatively estimate population-wide exposure distributions. With this sensor-based analysis (covering the >70,000 people within 5 km of a sensor node), we are able to resolve the fine structure of pollutant exposure during this extreme AQ event, analogous to previous global population-scaled annual exposure estimates based on satellite data and models (27). The choice of a 5-km buffer is somewhat arbitrary but is intended as a compromise between population coverage (SI Appendix) and spatial variability of point measurements as the plume chemically transforms and dynamically adjusts to underlying surface conditions. During the eruption, the highest SO_2 levels were experienced primarily by those just downwind of the vents (Fig. 3A), with 5.3% of the sampled population exposed to elevated (>35 ppb) average levels of SO₂. The LCS SO₂ hourly exposure distribution (Fig. 3C) shows an extended high-concentration tail, with 2% of all hourly exposures exceeding 75 ppb (the US EPA 1-h standard, though Hawai'i uses the pre-2010 threshold of 140 ppb). In comparison, cumulative hourly SO₂ exposures from the regulatory network (composed of five stations covering nearly 30,000 people) are 25% of the LCS network's cumulative total (SI Appendix, Fig. S9). However, the population-weighted average SO_2 concentrations measured by the two networks are not significantly different (9.8 ppb and 9.2 ppb) because of highest SO₂ concentrations occurring primarily in sparsely populated areas (SI Appendix, Fig. S8).

Exposure to secondary $PM_{2.5}$ was more widespread; elevated average levels (>10 µg · m⁻³, World Health Organization annual standard) were experienced by nearly half (46.7%) of the sampled population (Fig. 3*B*), with 6.7% of hourly exposures exceeding 35 µg · m⁻³ (the US EPA 24 h standard). The LCS PM_{2.5} population-weighted distribution (Fig. 3*D*) also displays a clear bimodal structure, representing the overlap of spatial population and plume dynamics; a large fraction of the sampled population (mostly in the Hilo area) lives upwind of the volcano and therefore mostly experienced background levels of PM_{2.5}, whereas population centers downwind of the vents experienced uniformly elevated PM levels over the entire course of the eruption. Because of the greater number of LCS stations in densely populated



Fig. 2. Average concentrations of SO₂ (*A*) and PM_{2.5} (*B*), as measured by the LCS network (colored circles) and the regulatory network stations (gray circles). Data are from a 15-d period from July 15 to August 1, 2018; only the LCS nodes that were in near-continuous operation during this time are shown. For SO₂, 17 sensors are shown, accounting for 70,414 people within 5 km. For PM_{2.5}, 20 are shown, accounting for 86,856 people within 5 km. In total, there are 16 stations with both SO₂ and PM_{2.5} measurements, accounting for 73,013 people within 5 km (Fig. 3). The full time series for all sensors are shown *SI Appendix*, Figs. S3 and S5.

Crawford et al.

Mapping pollution exposure and chemistry during an extreme air quality event (the 2018 Kīlauea eruption) using a low-cost sensor network



Fig. 3. Population exposure to volcanic pollutants, measured by the LCS network over the 15-d study period. (*A* and *B*) Mean pollutant distribution as a function of cumulative near-node population (residents living within 5 km of each node: 73,013 total). Bar width is proportional to nearby population, and bar height is the average pollutant concentration measured by each node. Sensor nodes are differentiated by color, as shown on the inset map. Stations are arranged from lowest to highest average concentration. (*C* and *D*) Population distribution as a function of hourly exposure frequency to SO₂ and PM_{2.5}. Here, the distribution of hourly concentrations experienced by each sensor node is weighted by population within 5 km of the node and arranged by average concentration. Estimation of near-node population is given in the *SI Appendix*. Hourly population-weighted time series data to create (*C* and *D*) is shown in *SI Appendix*, Fig. S9. An equivalent figure using regulatory network data are shown for comparison in *SI Appendix*, Fig. S10.

regions exposed to high levels of $PM_{2.5}$, cumulative hourly $PM_{2.5}$ exposures from the regulatory network during the study period are 28% of the LCS network (*SI Appendix*, Fig. S9). Additionally, the population-weighted average $PM_{2.5}$ concentration from the LCS network is substantially higher than that of the regulatory network (12.9 µg · m⁻³ to 8.2 µg · m⁻³; *SI Appendix*, Fig. S8), demonstrating the importance of dense monitoring networks to complement existing measurements for accurate population-wide exposure distributions.

The highly variable concentrations (Fig. 2) and exposures (Fig. 3) throughout the island arise not only from differences in location relative to the fissures and the plume but also from the dynamic chemical evolution of the volcanic pollutants (the oxidation of SO_2 to form PM). Such secondary transformations of

pollutants represent major challenges in AQ monitoring, which spatially dense LCS networks are well-suited to address. The high spatial resolution of our LCS network, with stations placed at different distances (i.e., plume transport times) downwind of the fissure, enables this chemical transformation to be mapped. Fig. 4 shows the evolution of the plume chemistry (described by the measured SO₂/PM_{2.5} ratio) as a function of plume age, which is calculated using an atmospheric dispersion model (*SI Appendix*). The ratio is highest at stations closest to the eruption, with a clear decay as the plume ages.

The rate of change in plume composition (Fig. 4) implies a mean SO₂ loss rate (k_{SO2}) of $7.6 \times 10^{-6} \cdot s^{-1} (\pm 4.8 \times 10^{-7} \cdot s^{-1})$ for an atmospheric SO₂ oxidation lifetime (τ_{ox}) of 36 h. While there are several uncertainties and assumptions in this calculation



Fig. 4. Chemical evolution of the vog plume (SO₂/PM_{2.5} ratio), as measured by LCS nodes (circles) and regulatory stations (triangles) downwind of the volcano during the July 15 to August 1 study period. This quantifies the rate of the chemical transformation of SO₂ (gas) to sulfuric acid (PM), yielding an estimated timescale of $7.6 \times 10^{-6} \cdot s^{-1}$ (lifetime of 36 h). Measurement uncertainties (vertical error bars) were determined separately for SO₂ and PM_{2.5} sensors during instrument calibration against reference instruments (*SI Appendix*). Plume age uncertainties (horizontal error bars) are the interquartile range of model parcel travel times between the LERZ and measurement location during each hour of the study period.

(see Materials and Methods), this value is consistent with known atmospheric SO₂ oxidation kinetics (28) and suggests some role of not only gas phase oxidation but also conversion within cloud droplets or particles as well. Moreover, these reaction kinetics are comparable to those of other volcanoes worldwide (29), within bounds of previous measurements at Kilauea (11, 17, 30-32), and in broad agreement with a recent estimate from the same eruption (17). Previous studies have used a range of measurement approaches-single-point, ground-based SO₂ photometer observations (31), time-dependent downwind SO₂ and sulfate measurements (30), satellite SO_2 measurements (32), and SO_2 and PM measurements from AQ stations (17)-which varied substantially in temporal and spatial coverage. The range of lifetimes of SO₂ from Kīlauea reported in the literature, $\tau_{ox} = \sim 9 h (31)$ to ~22 d (30), likely results from complex plume SO_2 dynamics, the measurement approach used, and differences in atmospheric conditions (e.g., solar radiation, temperature, and moisture) and plume composition (e.g., aerosols, water, and oxidants). An advantage of using LCS measurements in this case is that in situ measurements over multiple days from a relatively large number of spatially distributed stations are used to aggregate highly variable plume meteo-chemical conditions. This study validates the use of distributed LCS networks for monitoring not only air pollutant concentrations but also plume chemical evolution.

In all, the LERZ eruption lasted for ~13 wk before ending abruptly in early August 2018. After the sudden cessation of volcanic activity, AQ conditions improved appreciably across the island as SO₂ levels fell immediately and PM_{2.5} returned to background levels after 7 to 10 d (Fig. 1 *C* and *D*). While unexpected, this extreme AQ event provided the opportunity to rapidly implement an LCS network, demonstrating the strengths of LCS for aiding in emergency response, measuring populations' exposure to pollution, and characterizing regional atmospheric chemistry. Key features of this network were the individual nodes' low cost, small physical footprint, stand-alone power, and real-time communications, all of which allowed for rapid, flexible deployment. Two nodes were even lost in the lava flow, highlighting the resilience of networks composed of multiple low-cost nodes.

This environment-with its small number of pollutants, point sources emitting into a clean environment, and relatively simple chemistry and meteorology-is in many ways an ideal scenario for monitoring AQ using LCS. Nonetheless, this general approach can be extended to other, more complex environments as well. The present work highlights the need for multipollutant LCS nodes that are already calibrated and readily deployable; the characterization of other extreme AQ events, such as wildfires and urban smog, requires that these nodes measure a number of additional pollutants (not just PM and SO₂ but also O₃, NO₂, and CO) as well as species that can provide insight into pollutant sources and secondary chemical processes (such as NO, VOCs, and CO₂). Furthermore, knowledge of background pollutant levels and atmospheric dynamics is necessary to isolate contributions from the event and constrain reaction kinetics. These are included in the present analysis and will be even more important in regions with higher baseline concentrations or more complex meteorology. This underscores the need for improved characterization of regional background pollutant levels (from prior regulatory, satellite, or LCS measurements) and local meteorological conditions (from wind measurements and dispersion models) in such cases. Because of the chemical complexity of wildfire and urban smog pollution, LCS measurements will also benefit from future technological improvements, such as in-line dryers to obviate the need for uncertain relative humidity corrections, and low-cost techniques for measuring PM and VOC composition to provide insight into pollutant sources, chemistry, and impacts. LCS networks thus offer the potential for characterizing pollutant exposure and chemistry under a wide range of conditions and represent an important high-resolution component of multiplatform systems to monitor and characterize extreme AO events.

Materials and Methods

LCSs and Network Design. Custom multipollutant AQ sensor (MPAQS) sensor nodes were built to measure sulfur dioxide gas (Alphasense SO₂-B4 electrochemical sensor) and PM (Alphasense OPC-N2) concentrations, as well as auxiliary measurements of air temperature (T_A) and relative humidity (*RH*). Additionally, six SO₂-only nodes (24) and five PM-only nodes (Plantower PMS5003 nephelometer) were used to supplement the MPAQS network. Communications and data transmission are via 3G cellular microcontroller (Particle Electron). Sensor units are powered by rechargeable batteries and solar panels (Voltaic Systems). All sensors sample at 1 Hz and data are recorded to local storage, with 1-min averages transmitted to a custom cloud database. Total materials cost for each MPAQS node is approximately 1,000 US dollars.

The SO₂ and PM sensors were calibrated via colocation outdoors with regulatory-grade monitors run by the Hawai'i Department of Health (33). SO₂ calibration algorithms are based on sensor-specific nonlinear relations to air temperature and linear sensitivity to ambient SO₂ concentrations (*SI Appendix*). PM mass measurements are statistically corrected for ambient *RH* due to hygroscopic uptake by sulfate aerosols (*SI Appendix*). This type of field calibration approach is advantageous because instruments are exposed to dynamic and realistic ambient environmental conditions, which can be challenging to achieve under controlled laboratory settings (e.g., ref. 24).

Sensor nodes were placed primarily at schools, public health clinics, and community centers in residential areas, and each individual setting and placement was unique according to specific site characteristics and constraints (*SI Appendix*). Nodes were placed 3 to 15 m above ground level on rooftops (on tripods, electrical masts, etc.) or in open areas on facility grounds (on masts, tree trunks, and utility poles). Although sensors were deployed to more than 30 locations around the Island (*SI Appendix*), several nodes experienced intermittent communications interruptions, power issues, or sensor component failures (in particular, several SO₂ electrochemical

sensors failed, likely because of high ambient humidity). In total, there were 16 locations during the 2-wk study period with sufficient data quality and continuity for both SO₂ and PM_{2.5} (Figs. 2 and 3; 16 locations with PM_{2.5} and SO₂, 1 location with SO₂ only, and 4 locations with PM_{2.5} only).

Population Exposure Data and Analysis. Spatial population data are 2015 Census Designated Place (CDP) level populations from the Hawai'i State Data Center. The CDP polygons are rasterized assuming uniform density within each CDP (*SI Appendix*). The data are based on night-time residential population, and there are likely uncertainties based on nonresident populations (e.g., tourists and temporary workers), diurnal commuter patterns, and spatial settlement patterns (e.g., variations in housing density and informal encampments).

In Fig. 3, the 5-km buffer is intended as a compromise between spatial variability and representativeness of point measurements and population coverage. To avoid population double counting when stations are within 5 km, one half of the total number of residents living in the overlapping buffer zone is assigned to each station.

There are two methods used to calculate the 5-km population-weighted exposures shown in Fig. 3. First, total population exposure $\langle x_i \rangle$ for each node $\langle i \rangle$ (Fig. 3 *A* and *B*) is calculated by multiplying the number of residents within 5 km of the sensor node (p_i) by the mean observed concentrations $([c_i])$ of SO₂ (person ppb) and PM_{2.5} (person µg m⁻³): $x_i = p_i[c_i]$. Second, time-integrated exposure distributions (Fig. 3 *C* and *D* and *SI Appendix*, Fig. S9) for each node $\langle x_{t,i} \rangle$ are calculated as the summed product of population and smoothed (three-bin moving window) binned frequency counts (j) of hourly SO₂ (person ppb hours) and PM_{2.5} concentrations (person µg · m⁻³ hours):

$$x_{t,i} = \sum_{j=1}^{n} p_i[c_j].$$

SO₂ to PM Conversion Rate. The mean SO₂ reaction rate is calculated based on observed SO₂ and PM_{2.5} at 13 stations downwind of the eruption. The SO₂ and PM_{2.5} concentrations are converted to mass concentrations of sulfur (S in units of $\mu g \cdot m^{-3}$) and the fraction of total gas phase S (f_s) is

$$f_{\rm S} = S_{\rm SO2} / (S_{\rm SO2} + S_{\rm PM2.5}).$$
 [1]

This procedure assumes all remaining PM_{2.5} is secondary sulfuric acid aerosols, originating from volcanic SO₂ emissions. (The mass concentration of liquid water had been subtracted already as part of the PM RH-correction; *SI Appendix.*) This approach neglects background levels of nonvolcanic aerosols (~5 μ g · m⁻³ based on the network average PM_{2.5} measured from September 1 to 30, 2018 after the eruption had ceased). This fractional conversion is advantageous because reaction rates calculated using SO₂ measurements alone can be overestimated due to nonoxidative SO₂ losses from deposition or dilution (30).

An exponential decay function (λ) is then fit to the downwind measurement points:

$$f_{\rm S} = f_{\rm S0} {\rm e}^{-k_{\rm SO2}t}$$
, [2]

where t is the mean plume travel time between the LERZ and measurement locations from July 15 to August 1, 2018, calculated using a particle dispersion meteorological model (*SI Appendix*). Here, k_{SO2} represents the first-order

- S. S. Lim *et al.*, A comparative risk assessment of burden of disease and injury attributable to 67 risk factors and risk factor clusters in 21 regions, 1990-2010: A systematic analysis for the global burden of disease study 2010. *Lancet* 380, 2224–2260 (2012).
- 2. R. V. Martin *et al.*, No one knows which city has the highest concentration of fine particulate matter. *Atmos. Environ. X* **3**, 100040 (2019).
- B. Roueche, The fog. The New Yorker (1950). https://www.newyorker.com/magazine/ 1950/09/30/the-fog. Accessed 1 June 2021.
- M. L. Bell, D. L. Davis, Reassessment of the lethal London fog of 1952: Novel indicators of acute and chronic consequences of acute exposure to air pollution. *Environ. Health Perspect.* **109** (suppl. 3), 389–394 (2001).
- 5. S. Singh, *The Great Smog of India* (Penguin Random House India Private Limited, 2018).
- J. M. Ferreri *et al.*, The January 2013 Beijing "Airpocalypse" and its acute effects on emergency and outpatient visits at a Beijing hospital. *Air Qual. Atmos. Health* 11, 301–309 (2018).
- P. Yu, R. Xu, M. J. Abramson, S. Li, Y. Guo, Bushfires in Australia: A serious health emergency under climate change. *Lancet Planet. Health* 4, e7–e8 (2020).
- C. G. Jones et al.; The CARES Surveillance Group, Out-of-hospital cardiac arrests and wildfire-related particulate matter during 2015-2017 California wildfires. J. Am. Heart Assoc. 9, e014125 (2020).

decay constant of SO₂ and so the mean SO₂ lifetime is equal to k_{SO2}^{-1} . To incorporate measurement and plume age model uncertainties into confidence and prediction intervals (95%) (Fig. 4), the decay function is fit to an array of random points uniformly sampled within the uncertainty bounds at each measurement point (n = 10 points at each location; total n = 130). Measurement uncertainties were determined for SO₂ and PM_{2.5} sensors during instrument calibration against reference instruments (SO₂ mean absolute error [MAE] is 7.3 ppb and PM_{2.5} MAE is 4.5 µg · m⁻³) and modeled plume age uncertainties are the interquartile range of calculated parcel travel times between the LERZ and measurement location during each hour of the study period (*SI Appendix*).

The plume SO₂ conversion rate (k_{SO2}) fit to all downwind stations (n = 13) is 7.6 × 10⁻⁶ s⁻¹($\tau_{ox} = 36.3$ h), while the k_{SO2} fit to only the LCS network stations (n = 10) is 7.2 × 10⁻⁶ · s⁻¹ ($\tau_{ox} = 38.3$ h), a statistically insignificant difference demonstrating the ability of the low-cost network to observe reaction kinetics. Additionally, there is not a dramatic difference when k_{SO2} is fitted to measurements that have had approximate background PM_{2.5} levels subtracted ($k_{SO2} = 6.0 \times 10^{-6} \cdot s^{-1}$; $\tau_{ox} = 46.4$ h). However, k_{SO2} fit using plume ages estimated from an observed mean wind speed at a single location and downwind linear distance is $1.9 \times 10^{-5} \cdot s^{-1}$ ($\tau_{ox} = 15$ h), substantially faster than the rate using plume ages calculated from the particle dispersion model. This demonstrates the importance of local meteorological variations, especially in areas of complex terrain, to estimate reaction kinetics.

Data Availability. All study data are included in the article and/or SI Appendix.

ACKNOWLEDGMENTS. This work was supported by Assistance Agreement Grant RD83618301 awarded by the US EPA to the Massachusetts Institute of Technology (MIT). It has not been formally reviewed by the EPA. The views expressed in this document are solely those of the authors and do not necessarily reflect those of the agency. The EPA does not endorse any products or commercial services mentioned in this publication. Additional funding was provided by the MIT Tata Center and MIT's Department of Civil and Environmental Engineering. We acknowledge James Ciszewski and Lisa Wallace at the Hawai'i Department of Health for early logistical support; Tamar Elias, Clive Oppenheimer, John Saffell, Don Thomas, and Steven Businger for support and useful input; Kathy Vandiver (MIT); Liam Kernell (The Kohala Center); and the many people who assisted with this project and provided site access for the sensors: Wendy Baker, Eric Johnson, Steve Hirakami, and Terri York (Hawai'i Academy of Arts and Sciences); Chris King-Gates (Volcano School of Arts and Sciences); Ben Duke and Una Burns (West Hawai'i Explorations Academy); Justin Brown and John Hayes (Kealakehe High); Derek Okawa (Bay Clinic Health Centers); Milton Cortez (Hamakua-Kohala Health Centers); Kelly Lacks (Malamalama Waldorf School); Ted Brattstrom (Ka'u High School); Darlene Javar (Na'alehu Elementary School); Ken Watanabe (Waiakea Elementary School); Joan and Greg Guithues and Rick Ward (Ocean View Community Association); Rachel Silverman (Innovations Public Charter School); Karen Miller and Glenn Gray (Holualoa Elementary); Noreen Kunitomo (Honaunau Elementary); Bill Chen and Nia Lovell (Kahakai Elementary); Kaliko Grace (Kailapa Community Center); Shawn Suzuki (Konawaena High); Wayne Higaki (North Hawaii Community Hospital); Bob Zellner (Laupahoehoe High); Roseanna Konoa and Pancho Aldana (Big Island Noni Processing, Kalapana); David Benitez (US National Park Service); Jonathan and Oliver Grayson; and Christine Phelps (Pohakuloa Training Area).

- R. J. Huang et al., High secondary aerosol contribution to particulate pollution during haze events in China. Nature 514, 218–222 (2014).
- T. Elias, A. Jefferson Sutton, "Sulfur dioxide emission rates from Kilauea Volcano, Hawai'i, 2007–2010" (Tech. Rep. 2012-1107, US Geological Survey, 2012).
- A. K. Pattantyus, S. Businger, S. G. Howell, Review of sulfur dioxide to sulfate aerosol chemistry at Kilauea Volcano, Hawai'i. Atmos. Environ. 185, 262–271 (2018).
- 12. S. Businger et al., Observing and forecasting vog dispersion from Kilauea volcano, Hawai'i. Bull. Am. Meteorol. Soc. 96, 1667–1686 (2015).
- B. M. Longo, W. Yang, J. B. Green, F. L. Crosby, V. L. Crosby, Acute health effects associated with exposure to volcanic air pollution (vog) from increased activity at Kilauea Volcano in 2008. J. Toxicol. Environ. Health A 73, 1370–1381 (2010).
- C. A. Neal et al., The 2018 rift eruption and summit collapse of Kilauea Volcano. Science 363, 367–374 (2019).
- J. P. Vernier et al., VolKilau: Volcano rapid response balloon campaign during the 2018 Kilauea eruption. Bull. Am. Meteorol. Soc. 101, E1602–E1618 (2020).
- Y. Tang et al., Air quality impacts of the 2018 Mt. Kilauea volcano eruption in Hawaii: A regional chemical transport model study with satellite-constrained emissions. *Atmos. Environ.* 237, 117648 (2020).
- R. C. Whitty et al., Spatial and temporal variations in SO₂ and PM_{2:5} levels around Kīlauea Volcano, Hawai'i during 2007–2018. Front. Earth Sci. 8, 36 (2020).
- E. G. Snyder et al., The changing paradigm of air pollution monitoring. Environ. Sci. Technol. 47, 11369–11377 (2013).

Crawford et al.

- A. C. Rai et al., End-user perspective of low-cost sensors for outdoor air pollution monitoring. Sci. Total Environ. 607-608, 691–705 (2017).
- A. A. Shusterman et al., The Berkeley atmospheric CO₂ observation network: Initial evaluation. Atmos. Chem. Phys. 16, 13449–13463 (2016).
- R. Subramanian *et al.*, Air quality in Puerto Rico in the aftermath of Hurricane Maria: A case study on the use of lower cost air quality monitors. ACS Earth Space Chem. 2, 1179–1186 (2018).
- F. D. Pope, M. Gatari, D. Ng'ang'a, A. Poynter, R. Blake, Airborne particulate matter monitoring in Kenya using calibrated low-cost sensors. *Atmos. Chem. Phys.* 18, 15403–15418 (2018).
- J. J. Caubel, T. E. Cados, C. V. Preble, T. W. Kirchstetter, A distributed network of 100 black carbon sensors for 100 days of air quality monitoring in west Oakland, California. *Environ. Sci. Technol.* 53, 7564–7573 (2019).
- D. H. Hagan et al., Calibration and assessment of electrochemical air quality sensors by co-location with regulatory-grade instruments. Atmos. Meas. Tech. 11, 315–328 (2018).
- L. R. Crilley et al., Evaluation of a low-cost optical particle counter (Alphasense OPC-N2) for ambient air monitoring. Atmos. Meas. Tech. 11, 709–720 (2018).
- N. Zimmerman et al., A machine learning calibration model using random forests to improve sensor performance for lower-cost air quality monitoring. Atmos. Meas. Tech. 11, 291–313 (2018).

- 27. M. Brauer et al., Ambient air pollution exposure estimation for the global burden of disease 2013. Environ. Sci. Technol. 50, 79–88 (2016).
- R. von Glasow, N. Bobrowski, C. Kern, The effects of volcanic eruptions on atmospheric chemistry. *Chem. Geol.* 263, 131–142 (2009).
- C. Oppenheimer, B. Scaillet, R. S. Martin, Sulfur degassing from volcanoes: Source conditions, surveillance, plume chemistry and earth system impacts. *Rev. Mineral. Geochem.* 73, 363–421 (2011).
- J. H. Kroll et al.; TREX XII; TREX XI, Atmospheric evolution of sulfur emissions from Kīlauea: Real-time measurements of oxidation, dilution, and neutralization within a volcanic plume. Environ. Sci. Technol. 49, 4129–4137 (2015).
- J. N. Porter *et al.*, Sun photometer and lidar measurements of the plume from the Hawaii Kilauea Volcano Pu'u O'o vent: Aerosol flux and SO₂ lifetime. *Geophys. Res. Lett.* 29, 30–31 (2002).
- S. Beirle *et al.*, Estimating the volcanic emission rate and atmospheric lifetime of SO₂ from space: A case study for Kilauea Volcano, Hawaii. *Atmos. Chem. Phys.* 14, 8309–8322 (2014).
- Hawaii Department of Health, Clean Air Branch, Hawaii ambient air quality data. https://health.hawaii.gov/cab/hawaii-ambient-air-quality-data/. Accessed 1 June 2021.