



Valepotriates From the Roots and Rhizomes of *Valeriana jatamansi*Jones as Novel N-Type Calcium Channel Antagonists

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The roots and rhizomes of Valeriana jatamansi have long been used as folk medicine in Asia and usually named as "Zhizhuxiang" in Chinese for the treatment of abdominal distention and pain. However, its active ingredients and molecular targets for treatment of abdominal pain remain unrevealed. Inhibitors of Ca_v2.2 N-type voltage-gated calcium channels (VGCCs) are actively sought after for their potential in treating pain, especially chronic pain. As far as we know, the method used for seeking analgesic active ingredient from plant material has rarely been reported. The analgesic potentials of the EtOH extract (0.01 mg/ml) of the roots and rhizomes of V. jatamansi and its EtOAc, n-BuOH and H₂O soluble parts (0.01 mg/ml, respectively) were tested herein on Ca_v2.2, using whole-oocyte recordings in vitro by tow-electrode voltage clamp. The results indicated that the EtOAc-soluble part exhibited the most potent inhibition of Ca_v2.2 peak current (20 mv). The EtOAc-soluble part was then subjected to silica gel column chromatography (CC) and giving 9 fractions. Phytochemical studies were carried out by repeated CC and extensive spectroscopic analyses after the fraction (0.01 mg/ml) was identified to be active and got seventeen compounds (1-17). All isolates were then sent for further bioactive verification (1 and 3 at concentration of 10 μ M, others at 30 μ M). In addition, the selectivity of the active compounds **1** and **3** were tested on various ion channels including Ca_v1.2, Ca_v2.1 and Ca_v3.1 VGCCs and Kv1.2, Kv2.1, Kv3.1 and BK potassium channels. The results indicated that compound 1 and 3 (an abundant compound) inhibited $Ca_v2.2$ with an EC_{50} of 3.3 and 4.8 μ M, respectively, and had weaker or no effect on Ca_v1.2, Ca_v2.1 and Ca_v3.1 VGCCs and Kv1.2, Kv2.1, Kv3.1 and BK potassium channels. Compounds 1 and 3 appear to act as allosteric modulators rather than pore blockers of Ca_V2.2, which may play crucial role in attenuating nociception. The results of present research indicated that the

1

ethnopharmacological utilization of V. jatamansi for relieving the abdominal distention and pain may mediate through $Ca_V2.2$ channel. Our work is the first demonstration of inhibition of $Ca_V2.2$ by iridoids, which may provide a fresh source for finding new analgesics.

Keywords: *Valeriana jatamansi*, iridoid, abdominal distention and pain, N-type voltage-gated calcium channels, Ca_v2.2

INTRODUCTION

N-type calcium channels (Ca_v2.2) are highly distributed at nerve terminals and on cell bodies of dorsal root ganglia (DRG) neurons, where they regulate the release of pain-related neuropeptides. Pharmacology and genetic studies showed that Ca_v2.2 inhibiting activity was an effective way to manage or alleviate symptoms of inflammatory, chronic and neuropathic pain (Kerr et al., 1988; Gohil et al., 1994; Saegusa et al., 2001, 2002; McGivern and McDonough, 2004; McGivern, 2006; Altier et al., 2007; Trang et al., 2015). Ca_v2.2 antagonists have therefore long been considered as potential analgesics (Hazelhoff et al., 1982; Mathela et al., 2005; La et al., 2008; Yamamoto and Takahara, 2009; Pajouhesh et al., 2010; Schmidtkco et al., 2010; Swensen et al., 2014). Indeed, a synthetic form of ω-conotoxin MVIIA, a peptide toxin isolated from the marine cone snail, genus Conus, that potently inhibits Ca_v2.2, is in clinical use (marked as ziconotide) to treat severe and chronic pain (Schmidtkco et al., 2010). However, ziconotide can only be delivered intrathecally and even so has numerous serious adverse effects. Therefore, it is clearly desirable for development of new Ca_v2.2 antagonists.

Valeriana jatamansi Jones distributed widely in China and some other Asian countries (Mathela et al., 2005). The roots and rhizomes of this plant is a well-known herbal medicine "Zhizhuxiang" in the Chinese Pharmacopoeia for treating abdominal pain (Editorial Board of Chinese Pharmacopoeia, 2015). Several clinically used medicines for treating abdominal pain in the Chinese market, such as "Xiangguo Jianxiao Pian," are prepared by Zhizhuxiang as a main component. However, the chemical basis and molecular targets of V. jatamansi Jones for the treatment of abdominal pain remain unrevealed.

Inspired by the analgesic effect of V. jatamansi Jones, bioassay-guided isolation and characterization of the active constituents of the plant that target Cav2.2 were carried out herein. Initially, the ethanol extract and the EtOAc, n-BuOH and H2O fractions of the plant were tested for their inhibition of Ca_v2.2. Subsequent function-guided phytochemical studies on the active EtOAc part resulted in the isolation and identification of thirteen iridoids (1-13) and four sesquiterpenes (14-17) (Figure 1), including two new valepotriates, namely jatamanvaltrate T and U (1-2). The inhibitory activities on Ca_v2.2 of all the compounds were tested. Among them, the new compound 1 and the main constituent 3 (8.45 g out of 32.5 kg of plant material) exhibited significant inhibitory effects on Ca_v2.2 with EC₅₀ values of 3.3 and 4.8 µM, respectively. In addition, compound 1 and 3 showed noticeable selectivity over Ca_v1.2, Ca_v2.1 and

 $\text{Ca}_{\text{v}}3.1\ \text{VGCCs}$ and Kv1.2, Kv2.1, Kv3.1 and BK potassium channels.

MATERIALS AND METHODS

General Experimental Procedures

Optical rotations were measured with a JASCO model 1020 polarimeter (Horiba, Tokyo, Japan). UV spectra were recorded on a Shimadzu UV-2401PC spectrophotometer (Shimadzu, Kyoto, Japan) and IR (KBr) spectra were obtained using a Bruker Tenor 27 FT-IR spectrophotometer (Bruker Optics GmbH, Ettlingen, Germany). ESIMS and HRESIMS were performed on an Agilent G6230 TOF MS (Agilent Technologies, Palo Alto, CA, United States) or an API QSTAR time-of-flight spectrometer (AB-MDS Sciex, Concord, ON, Canada) or a VG Autospec-3000 spectrometer (VG, Manchester, United Kingdom). 1D and 2D NMR spectra were recorded on Bruker AM-400 and DRX-500 spectrometer (Bruker, Bremerhaven, Germany) with TMS as the internal standard. Chemical shifts (δ) are expressed in ppm with reference to the solvent signals. MPLC was run on a Dr-Flash-S MPLC system (Lisui, Suzhou, China). Silica gel (200-300 mesh) for column chromatography (CC) and TLC was obtained from Qingdao Marine Chemical Factory, Qingdao, China. Sephadex LH-20 was purchased from Amersham Biosciences, Sweden. RP-C18 gel (40-63 µm, Merck, Darmstadt, Germany) and MCI gel (75-150 µm, Mitsubishi Chemical Corporation, Japan). Fractions were monitored by TLC (GF254, Qingdao Haiyang, Qingdao, China), and spots were visualized by sprayed with 5% sulfuric acid in EtOH, followed by heating.

Plant Material

The roots and rhizomes of *V. jatamansi* were collected at Kunming, Yunnan Province, People's Republic of China, in July 2012, and identified by Prof. En-De Liu, and a voucher specimen (KUN NO. 0864803), has been deposited in State Key Laboratory of Phytochemistry and Plant Resource in West China, Kunming Institute of Botany, The Chinese Academy of Sciences, Kunming, China.

The Preparation of Crude Extract and Fractions

The air-dried roots (450 g) of V. jatamansi were powdered and extracted with 95% ethanol at room temperature (3L \times four times, each time 24 h) with occasional shaking. The extracted was filtered and concentrated with a rotary evaporator at 45–50°C, giving 75.5 g residues (The yield is approximate 16.8%). The

ethanol extract (43.62 g) was suspended in distilled (600 ml) water and partitioned successively with EtOAc and n-BuOH (five times, each time 0.5 h), then evaporated on rotary evaporator to afford EtOAc-soluble part (12.73 g) and n-BuOH layer (5.38 g). The H₂O-soluble part was passed through a macroporous absorbent resin (D-101) with H₂O and ethanol, then concentrated under reduced pressure to obtain H₂O layer (0.51 g). The EtOAc part was subjected to silica gel column chromatography eluted with a gradient of petroleum ether/acetone (1:0-0:1, v/v) to afford 9 fractions (Fr.1–Fr.9).

Extraction and Isolation

Based on the observed effect with the EtOAc portion on $\text{Ca}_{v}2.2$, further extraction and isolation of the EtOAc portion were carried out herein.

The air-dried, powdered roots of V. jatamansi (32.50 kg) were extracted three times with 95% EtOH (3 \times 60 L, each 28 h) at room temperature. The combined EtOH extracts were concentrated in vacuum to afford a crude residue (3.65 kg), which was then suspended in water (4.0 L) and

partitioned with EtOAc (4.0 L × 5). Then got the EtOAc part (1.72 kg) and chromatographed on a silica gel column chromatography (CC, 200-300 mesh), eluting with a gradient of petroleum ether-acetone (1:0-0:1) to give nine fractions (Fr.1-Fr.9). Fr.1 (192.4 g) was resubmitted to silica gel CC (petroleum ether-acetone gradient, 1:0-10:1) and then purified by Sephadex LH-20 column (MeOH) to yield compound 5 (94.0 mg) and 17 (17 mg). Similarly, 8 (53 mg) and 6 (135 mg) were got from Fr.2 (41.10 g). Fr.3 (55.50 g) was decolorized on a MCI gel (MeOH-H₂O gradient, 60:40-95:5), then subjected to MPLC (RP-18 gel, MeOH-H₂O gradient, 40:60-90:10) and separated by Sephadex LH-20 column (MeOH) to give 7 (246 mg) and 10 (34 mg). The bioactive Fr.4 (100.30 g) was chromatographed on silica gel CC (200-300 mesh) (petroleum ether-acetone, 50:1-1:1) to yield 4 fractions (Fr.4.1-Fr.4.4). Fraction 4.2 (28.3 g) was applied to a MCI gel (MeOH-H2O gradient, 60:40-85:15), then by repeated silica gel CC (petroleum ether-acetone, 20:1-5:1) to obtain 14 (70 mg) and 15 (1.65 g), then further by RP-C18 (MeOH-H₂O gradient, 50:50-85:15) and followed by Sephadex LH-20 (MeOH) to give 1 (148 mg) and 9 (95 mg). Fr.4.3

(22.017 g) was successively separated by MPLC (MeOH-H₂O gradient, 30:60-50:50) and repeated silica gel CC (petroleum ether-EtOAc, 10:1-5:1), purified by Sephadex LH-20 (MeOH) to obtain **2** (81 mg) and **11** (5 mg). The bioactive Fr.6 (193.70 g) was passed through a silica gel CC and eluted with a gradient of petroleum ether-acetone (20:1-0:1) to yield 11 fractions (Fr.6.1–Fr.6.11). Fr.6.8 (70.0 g) was subjected to silica gel CC (petroleum ether-acetone gradient, 10:1-0:1) to afford 8 sub-fractions (Fr.6.8.1–Fr.6.8.8). Fr.6.8.5 (33.64 g) was repeatedly subjected to silica gel CC (petroleum ether-acetone gradient, 10:1-0:1), then purified by Sephadex LH-20 column (MeOH) to give **3** (8.45 g) and **4** (50 mg) and **16** (60 mg). Fr.6.9 (73.5 g) was treated as Fr.6.8.5 and get compounds **12** (20 mg) and **13** (51 mg).

Physical and Spectroscopic Data of New Compounds

Jatamanvaltrate T (1)

Colorless oil, $[a]_{23}^{23} = +152.60$ (c 0.26, MeOH, **Supplementary Figure S12**). UV (MeOH) λ_{max} (log ϵ_{max}): 255 (5.22) nm (**Supplementary Figure S11**); IR (KBr) ν_{max} (cm⁻¹): 3441, 2963, 2875, 1742, 1640, 1467, 1371, 1294, 1250, 1187, 1166, 1101, 1027, 963 cm⁻¹; ¹H NMR and ¹³C NMR data, see **Table 1**; positive ESIMS m/z 647 [M + Na]⁺; HRESIMS m/z 647.3040 [M + Na]⁺ (calculated for $C_{32}H_{48}O_{12}Na$, 647.3043).

Jatamanvaltrate U (2)

White amorphous powder, [a] $_{\rm D}^{23}=+131.61$ (*c* 0.45, MeOH) (**Supplementary Figure S22**). UV (MeOH) $\lambda_{\rm max}$ (log $\epsilon_{\rm max}$):: 255 (4.59) nm (**Supplementary Figure S21**); IR (KBr) $\nu_{\rm max}$ (cm $^{-1}$): 3444, 2923, 1733, 1642, 1467, 1381, 1296, 1260, 1183, 1154, 1101, 1082, 1016, 987 cm $^{-1}$; ¹H NMR and ¹³C NMR data, see **Table 1**; positive ESIMS m/z 701 [M + Na] $^+$; HRESIMS m/z 701.4234 [M + Na] $^+$ (calculated for $C_{38}H_{62}O_{10}Na$, 701.4241).

TABLE 1 | $^{1}\mathrm{H}$ NMR (400 MHz) and $^{13}\mathrm{C}$ NMR (100 MHz) data of compounds 1–2 in CD₃OD.

	1		2		
Position	δн	δc	δ_{H}	δc	
1	6.11, d (10.0)	94.0, CH	6.12, d (10.1)	94.0, CH	
3	6.78, s	149.5, CH	6.77, s	149.4, CH	
4		110.4, C		110.4, C	
5		140.6, C		140.7, C	
6	5.76, dd (2.6, 2.5)	118.6, CH	5.74, dd (2.7, 2.6)	118.4, CH	
7	5.48, d (2.7)	84.5, CH	5.49, d (2.7)	84.5, CH	
8		80.6, C		80.6, C	

The ^{1}H NMR data of the substituents at C-10 of **1** (α -(isovaleroxy)isovaleroxy) group): 2.27 (2H, m, H-7"), 2.11 (1H, m, H-8"), and 0.98 (6H, d, J = 6.4 Hz, H-9"), 10"); and of **2** (palmitoy)]: 1.28 (20H, m, H-6" \sim 15") and 0.88 (3H, t, J = 5.6 Hz, H-16"). The 13 C NMR data of the substituents at C-10 of **1** (α -(isovaleroxy)isovaleroxy) group): $\delta_{\rm C}$ 174.0 (C, C-6"), 43.8, (CH₂, C-7"), 26.9, (CH, C-8") and 22.7 (CH₃, C-9", 10"); and of **2** (palmitoy)]: $\delta_{\rm C}$ 30.2-30.7 (CH₂, C-6"-13"), 33.0 (CH₂, C-14"), 23.5 (CH₂, C-15") and 14.1 (CH₃, C-16").

Bioassay

Oocyte Preparation and Expression

Oocytes were obtained from adult *Xenopus* by digesting its ovarian lobes with collagenase A (sigma) for 2–3 h under 180 rpm shaking in OR2 [MgCl₂ (1 mM), NaCl (82.4 mM), KCl (2.5 mM), HEPES (5 mM) and NaOH (pH 7.6)]. The best Stages V–VI oocytes were selected, injected with cRNA (50–100 ng), and then incubated at 18°C for 3–6 days depending on cRNA expression in ND96 [CaCl₂ (1.8 mM), MgCl₂ (1 mM), NaCl (96 mM), KCl (2.5 mM), HEPES (5 mM), streptomycin (100 μ g/mL), penicillin (100 units/mL) and NaOH (pH 7.6)].

HEK 293T Cell Culture and Expression

Human embryonic kidney (HEK) 293T cells were grown in DMEM (HyClone) plus 10% newborn calf serum (Gibco) and streptomycin (0.1 mg/ml)/penicillin (100 U/ml) (Biological Industries). The cells were transiently transfected with pCDNA3.1- $\alpha_2\delta$, pCDNA3.1- β_3 , pCDNA3.1-EGFP and pCDNA3.1-N type plasmids together using LipoD293TM (SignaGen Laboratories) and recorded in 48 h.

Electrophysiology

All experiments were performed at 20-22°C. Whole-oocyte recordings by performed with two-electrode voltage clamp (TEVC). Electrodes were filled with KCl (3 mM) and had resistances of 0.3-1 M Ω . The bath solution used to record calcium channel currents contained KCl (2 mM), BaCl₂ (1.8 mM), NaOH (50 mM), Ba(OH)₂ (40 mM) and HEPES (5 mM). pH 7.4 was adjusted with methanesulfonic acid and the solution was filtered. L-type (Ca_v1.2), N-type (Ca_v2.2), and P/Q-type (Ca_v2.1) calcium channel currents were evoked from a holding potential of -80 mV by 50-ms depolarizations ranging from -30 to 70 mV in 10-mV increments at 3-s intervals. Currents through T-type (Ca_v3.1) calcium channels were elicited by voltage pulses (50 ms) from -50 to 60 mV with a holding potential of -80 mV in 10-mV increments at 3-s intervals. Kv1.1 currents were recorded from a holding potential of -80 mV by 200-ms depolarizations ranging from -60 to 60 mV in 10-mV increments at 15-s intervals. Kv2.1 and Kv3.1 currents were obtained from a holding potential of -80 mV by 200-ms depolarizations ranging from −40 to 60 mV in 10-mV increments at 15-s intervals. BK channel currents were obtained from a holding potential 0f -50 mV by 60-ms depolarization ranging from 0 to 100 mV in 10-mV increments at 3-s intervals. All the currents were sampled and filtered, respectively.

In whole-cell recordings of HEK 293T cells, pipettes were fabricated from borosilicate glass (World Precision Instruments) using a micropipette puller (P-1000, Sutter Instrument), and were fire-polished to resistances of $\sim\!3$ MW. Whole-cell currents were elicited by 20-ms voltage steps from -60 to 80 mV with 10-mV increments, with a holding potential of -80 mV. Currents were amplified by Axopatch 200B and digitized by Digidata 1440A (Molecular Devices). Currents were low-pass filtered at 2 kHz and sampled at 10 kHz. The extracellular solutions contained (in mM) 5 CsCl, 10 BaCl₂, 140 TEA.Cl, 10 Glucose and 10 HEPES. pH was adjusted to 7.4 with CsOH. The intracellular solutions contained

TABLE 2 | Effect of extracts and compounds 1-9 on Ca_v2.2 N-type VGCC.

Fraction	Ca _v 2.2 ^a (N-type)	Compound	Ca _v 2.2 ^b (N-type)
EtOH extract	36.3 ± 10.2%	1	43.8 ± 1.7%
EtOAc-part	$48.6 \pm 7.7\%$	2	$3.2 \pm 2.1\%$
n-BuOH-part	_	3	$38.9 \pm 6.7\%$
H ₂ O-part	$1.1 \pm 1.7\%$	4	$4.1 \pm 2.4\%$
Fr.1	$1.3 \pm 6.3\%$	5	$9.0 \pm 16.2\%$
Fr.2	$56.5 \pm 4.8\%$	6	$8.5 \pm 7.0\%$
Fr.3	$10.5 \pm 6.2\%$	7	$0.1 \pm 1.1\%$
Fr.4	$68.6 \pm 3.1\%$	8	$3.4 \pm 1.9\%$
Fr.5	$55.2 \pm 3.5\%$	9	$5.4 \pm 1.5\%$
Fr.6	$60.1 \pm 4.0\%$		
Fr.7	$34.1 \pm 6.3\%$		
Fr.8	$28.4 \pm 5.2\%$		
Fr.9	$14.1 \pm 2.0\%$		

^aPeak current inhibition ratio of extracts at 0.01 mg/ml; ^bPeak current inhibition ratio of compounds at 10 μ M (1 and 3) or 30 μ M (other compounds); Data represent the mean \pm SEM of 3 or 4 cells.

(in mM) 4 MgCl $_2$, 140 CsCl, 10 EGTA and 10 HEPES (pH 7.4 adjusted with CsOH).

The effect on $\text{Ca}_{\text{v}}2.2$ of the ethanol extract (0.01 mg/ml) and the EtOAc, n-BuOH and H_2O soluble parts (0.01 mg/ml) of the EtOH extract was tested by TEVC of *Xenopus* oocytes (**Supplementary Figure S23**). The EtOAc-soluble part (**Supplementary Figure S24**), which exhibited a stronger inhibitory activity, was subjected to silica gel column chromatography to produce 9 fractions. All of the fractions (0.01 mg/ml), together with the isolates (1–17) from the fractions, were tested for their inhibitory effects on $\text{Ca}_{\text{v}}2.2$, with a concentration of 10 μ M (1 and 3) or 30 μ M (2, and 4–17). Compounds 1 and 3 were further tested for their activities on $\text{Ca}_{\text{v}}1.2$, $\text{Ca}_{\text{v}}2.1$ and $\text{Ca}_{\text{v}}3.1$ VGCCs (10 μ M) and on Kv1.2, Kv2.1, Kv3.1 and BK potassium channels (30 μ M) (**Supplementary Figure S25**).

In a positive control experiment, the effect on $Ca_v2.2$ of 100 μ M CdCl₂ and 0.2–1.0 μ M ω -conotoxin MVIIA was tested by TEVC in *Xenopus* oocytes and by whole-cell recording in HEK 293T cells.

Data Analysis and Statistics

Data acquisition and analysis of the whole-oocyte recordings were carried out by using pClamp 10 (Molecular Devices Corporation, San Jose, CA, United States). Data fitting and statistical analyses were performed by PRISM 5.0 (GraphPad Software Inc., San Diego, CA, United States). EC₅₀ values and Hill slopes were determined by fitting the data points to a sigmoidal dose-response equation ($Y = \text{Min} + (\text{Max-Min})/(1+10^{\circ}((\text{LogEC50-}X)^*n)))$, where Y is % Inhibition, X is the concentration of the compounds, Min is minimal inhibition, Max is maximum inhibition, and n is the Hill coefficient. All data were presented as mean \pm SEM, and statistical analysis was performed using Student's t-test. P-values of < 0.05 were considered as significant, and levels of significance were marked by asterisks (**P < 0.01, *P < 0.05).

RESULTS AND DISCUSSION

Structural Elucidation of Isolated Compounds

Jatamanvaltrate T (1) was obtained as colorless oil with the molecular formula of C₃₂H₄₈O₁₂ assigned by positive HRESIMS (Supplementary Figure S1) at m/z $647.3040 [M + Na]^+$ (calcd for C₃₂H₄₈O₁₂Na, 647.3043), indicating nine degrees of unsaturation. The IR absorption (Supplementary Figure S10) bands at 3441, 1742, and 1640 cm⁻¹ revealed the presence of hydroxy, ester carbonyl, and double bond groups, respectively. The ¹H NMR and ¹³C-NMR spectroscopic data (**Table 1** and Supplementary Figures S2-S4) displayed signals for two trisubstituted olefinic bonds [δ_H 6.78 (s, H-3); δ_C 149.5 (d, C-3) and 110.4 (s, C-4); δ_H 5.76 (dd, J = 2.6, 2.5 Hz, H-6); δ_C 118.6 (d, C-6) and 140.6 (s, C-5)], a hemiketal methine [δ_H 6.11 (d, J = 10.0 Hz), H-1); δ_C 94.0 (d, C-1)], and two oxymethylenes $[\delta_{\rm H} \ 4.39, \ 4.35 \ ({\rm each} \ {\rm d}, \ J=11.3 \ {\rm Hz}, \ {\rm H}\text{-}10), \ 4.77 \ ({\rm d}, \ J=11.4 \ {\rm Hz}, \ {\rm H}$ H-11a) and 4.65 (d, J = 12.3 Hz, H-11b); δ_C 67.5 (t, C-10) and 62.0 (t, C-11)]. These data, together with the resonances of five ester carbonyls at δ_C 171.0, 172.1, 172.6, 173.8, and 174.0, led to the assumption that compound 1 is a valtrate hydrin-type iridoid (Demirezer et al., 1999; Tang et al., 2002; Wang et al., 2008, 2014; Lin et al. 2010b, 2013). It could be seen that the

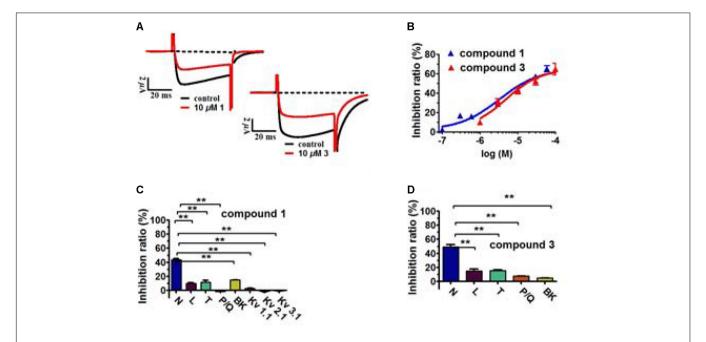


FIGURE 3 | Inhibitory effects of compounds **1** and **3** on $Ca_v2.2$ and their selectivity. **(A)** Representative $Ca_v2.2$ current traces evoked by a 50-ms depolarization to +20 mV at 3-s intervals from a holding potential (HP) of -80 mV in the presence of the indicated compounds. $Ca_v2.2$ was expressed in *Xenopus* oocytes. **(B)** Dose–response relationships of the indicated compounds for $Ca_v2.2$ at a HP of -80 mV. Data points represent mean \pm SEM of three or five measurements. Solid curve represents a fit to a modified Hill equation with an EC_{50} of 3.3 μ M (**1**) and 4.8 μ M (**3**), and a Hill coefficient of 0.82 (**1**) and 0.93 (**3**). **(C,D)** Effects of compounds **1** and **3** on indicated ion channels. The effect on the peak current was determined for each compound at 10 μ M (N-, L-, P/Q- and T-type VGCCs) and 30 μ M (BK, K_v1.1, K_v2.1, and K_v3.1 potassium channels). Data represent mean \pm SEM (n = 3). **P < 0.01 according to the two-tailed Student's t-test.

structure of 1 was similar to those of 10-acetoxy-1-acevaltrate hydrin from the ¹H- and ¹³C-NMR data of 1 (Tang et al., 2002), except for the presence of an isovaleroxy residue [δ_C 172.1 (s, C-1'), 44.0 (t, C-2'), 26.8 (d, C-3'), 22.6 (q, C-4', C-5') and an α -(isovaleroxy)isovaleroxy group [δ_C 171.0 (s, C-1"), 77.8 (d, C-2"), 31.1 (d, C-3"), 19.3 (q, C-4"), 17.5 (q, C-5"), 174.0 (s, C-6"), 43.8, (t, C-7"), 26.9, (d, C-8"), and 22.7 (q, C-9", 10")] in 1. The substituent structure of 1 with two isovaleroxy groups located at C-1 and C-7, respectively, was confirmed by the HMBC spectra (Supplementary Figures S5-S7) as following: the correlations (**Figure 2**) from H-1 (δ_H 6.11, d, J = 10.0 Hz), H-2' (δ_H 2.32, m) and H-3' (δ_H 2.04, m) to C-1' (δ_C 172.1, s); from H-7 (δ_H 5.48, d, J = 2.7 Hz), H-2''' (δ_{H} 2.28, m) and H-3''' (δ_{H} 2.15, m) to C-1''' (δ_C 173.8, s). The connection of the α -(isovaleroxy)isovaleroxy functionality to C-10 was fully determined on the basis of the correlations from H-10 (δ_{H} 4.39, 4.35, each d, J = 11.3 Hz), H-2" $(\delta_{\rm H} 4.78, d, J = 4.4 \text{ Hz})$ and H-3" $(\delta_{\rm H} 2.26, m)$ to C-1" $(\delta_{\rm C} 171.0,$ s), together with connections from H-2" ($\delta_{\rm H}$ 4.78, d, J = 4.4 Hz), H-7" (δ_H 2.27, m) and H-8" (δ_H 2.11, m) to C-6" (δ_C 174.0, s).

The correlations from H-11a ($\delta_{\rm H}$ 4.77, d, J = 11.4 Hz), H-11b ($\delta_{\rm H}$ 4.65, d, J = 12.3 Hz) and H-2"" ($\delta_{\rm H}$ 2.00, s) to C-1"" ($\delta_{\rm C}$ 172.6, s) determined the linkage of the AcO residue to C-11.

The relative configuration of 1 was further established by ROESY experiment (Figure 2 and Supplementary Figures S8, S9) and comparison the spectroscopic data with those reported valepotriates (Demirezer et al., 1999; Tang et al., 2002; Wang et al., 2008, 2014; Lin et al., 2010b, 2013). Generally, naturally occurring iridoids display α -orientation for H-1 and β -orientation for H-9. In the ROESY spectrum, the correlations of H-9/H-10, OH-8/H-7, not of H-7/H-9 and H-7/H-10 implied that H-9 and H-10 were β -oriented, and H-7 and OH-8 were α -oriented. Comparing the NMR data of 1 with iridoids reported in the literature, the configurations of H-1, H-7, OH-8, H-9, and H-10 were consistent with those valepotriates (Demirezer et al., 1999; Wang et al., 2008, 2014; Lin et al., 2010b, 2013). Thus, 1 was assigned as (1S, 7S, 8R, 9S)-11-acetoxy-10-[a-(isovaleroxy)isovaleroxy]-1, 7-diisovaleroxyvaltrate hydrin, named as jatamanvaltrate T.

TABLE 3 | Effect of 1 and 3 on selected types of VGCCs and potassium channels.

Compound	Ca _v 1.2 ^a (L-type)	Ca _v 2.1 ^a (P/Q-type)	Ca _v 3.1 ^a (T-type)	BK ^b channel	K _v 1.1 ^b	K _v 2.1 ^b	K _v 3.1 ^b
1	10.1 ± 1.2%	$1.3 \pm 0.3\%$	$11.8 \pm 2.9\%$	$15.5 \pm 1.0\%$	$2.8 \pm 1.1\%$	$1.0 \pm 1.1\%$	$0.5 \pm 0.5\%$
3	$13.5 \pm 4.1\%$	$7.8 \pm 1.2\%$	$20.4 \pm 7.4\%$	$6.7\pm7.7\%$	_	_	_

^aPeak current inhibition ratio of compounds at 10 μ M; ^bPeak current inhibition ratio of compounds at 30 μ M; "—" not tested, due to the limited amount of material. Data represent the mean \pm SEM of 3 or 4 cells.

Jatamanvaltrate U (2) was get as an amorphous white powder, possessed the molecular formula of C₃₈H₆₂O₁₀ by positiveion HRESIMS at m/z 701.4234 [M + Na]⁺ (calcd 701.4241) (Supplementary Figure S13). Its IR spectrum (Supplementary Figure S20) revealed the presence of hydroxy, ester carbonyl and double bond groups from absorption bands at 3444, 1733, and 1642 cm⁻¹. Detailed analysis of the ¹H, ¹³C NMR and DEPT data of 2 (Table 1) (Supplementary Figures S14, S15) revealed that the structure of 2 was similar to those of 1 except for the presence of a palmitoyl group [δ_H 2.32 (2H, t, I = 2.4 Hz, H-2''), 1.59 (2H, m, H-3"), 1.28 (each 2H, each m, H-4" \sim 15") and 0.88 (3H, t, J = 5.6 Hz, H-16"); δ_C 175.0 (s, C-1"), 34.5 (t, C-2"), 25.9 (t, C-3"), 30.2-30.7 (t, C-4"-13"), 33.0 (t, C-14"), 23.5 (t, C-15"), and 14.1 (q, C-16")] in 2 rather than an α -(isovaleroxy)isovaleroxy in 1. The HMBC correlations (**Figure 2**) of H-10 (δ_H 4.31, brs), H-2" (δ_H 2.32, t, J = 2.4 Hz) and H-3" (δ_{H} 1.59, m) with C-1" (δ_{C} 175.0, s), of H-11a (δ_H 4.76, d, J = 12.3 Hz), H-11b (δ_H 4.65, d, J = 12.3 Hz), and H-2"" (δ_H 2.00, s) with C-1"" (δ_C 172.6, s) indicated the palmitoyl and acetoxyl residue were located at C-10 and C-11 of 2, respectively. Furthermore, the positions of the other two isovaleroxy groups at C-1 and C-7 were also confirmed based on the HMBC correlations (Supplementary Figures S16– S18).

By comparison of the ROESY correlations (**Supplementary Figure S19**) with that of **1**, the relative configuration of **2** (**Figure 2**) was demonstrated to be identical to that of **1**. All of the key ROESY correlations of H-9/H-10, OH-8/H-7 supporting the structure of **1** were also observed in **2**, suggesting that the H-9 and H-10 were β -oriented and H-7 and OH-8 were α -oriented. Therefore, **2** was established as (1*S*, 7*S*, 8*R*, 9*S*)-11-acetoxy-1,7-di isovaleroxy-10-palmitoyl caltrate hydrin, named as jatamanvaltrate U.

The known compounds, valtrate hydrin B8 (3) (Demirezer et al., 1999), volvaltrate B (4) (Lin et al., 2010b), deoxido-didrovaltrate (5) (Bounthanh et al., 1983), baldrinal (6) (Denee et al., 1979), homobaldrinal (7) (Denee et al., 1979), desacylbaldrinal (8) (Georg and Michael, 1979), 11-methoxyviburtinal (9) (Chen et al., 2005), jatamanvaltrate R (10) (Dong et al., 2015), jatamanvaltrate S (11) (Dong et al., 2015) rupesin E (12) (Lin et al., 2010a), (1S, 3R, 5S, 7S, 8S, 9S)-3, 8-epoxy-7-hydroxy-1-methoxy-4, 11-dihydronepetane (13) (Thies, 1970), valeriananoid B (14) (Ming et al., 1997), valeriananoid C (15) (Ming et al., 1997), valeriananoid C (15) (Ming et al., 1997), valeriananoid D (16) (Dong et al., 2015), and valeriananoid E (17) (Dong et al., 2015) were identified by analysis of their spectroscopic data and comparison with the literature values. This is the first isolation of iridoids with fatty acid esters (2) from the family of Valerianaceae.

Functional Characterization of Isolated Compounds

The traditional use of the roots and rhizomes of *V. jatamansi* to treat abdominal pain led us to assume that some of their effects may be mediated through inhibition of Ca_v2.2. Therefore, we carried out a systematic functional assay of the various extracts, fractions and isolated compounds obtained from the herbal medicine. Initially, we found that the ethanol extract

and its EtOAc-soluble layer (0.01 mg/ml) showed inhibitory effects of 36 and 49% on Ca_v2.2, respectively, while the n-BuOH and H₂O layers (0.01 mg/ml) exhibited negligible activities (Table 2). Of all the fractions from the ethyl acetate extract, fractions 4 and 6 exhibited strongest activities on $Ca_v 2.2$, with $68.6 \pm 3.1\%$ and $60.1 \pm 4.0\%$ inhibition at 0.01 mg/ml, respectively (Table 2). Further phytochemical isolation from these two fractions led to the discovery of two new valepotriates, jatamanvaltrate T and U (1-2) and fifteen known compounds (3-17). All of the isolates were evaluated for their inhibitory effects on Ca_v2.2. The new compound 1 (jatamanvaltrate T) and the main constituent 3 showed prominent inhibition of Ca_v2.2 (Table 2 and Figure 3). Other compounds produced no (10-17) or much weaker (2, 4-9) effects on Ca_v2.2, even at a concentration of 30 μM (Table 2) (Supplementary Figure S26). Based on the bioassay results and the structure of these compounds, we postulate that the diene-type iridoid is a nucleus for the inhibitory activity on Ca_v2.2.

The inhibition of Ca_v2.2 by compounds 1 and 3 was dose-dependent, with an EC₅₀ of 3.3 μ M (n=5) and 4.8 μ M (n=3), respectively (**Figure 3**). Interestingly, the inhibition by both compounds was incomplete, plateauing at $\sim\!65\%$ at a near saturation concentration of 100 μ M (**Figure 3**). This result suggests that compounds 1 and 3 act allosterically to modulate Ca_v2.2 gating rather than block channel conduction.

We also tested the effect of compounds 1 and 3 on several other types of VGCCs, including L-type ($Ca_v1.2$), P/Q-type ($Ca_v2.1$) and T-type ($Ca_v3.1$), and on several types of potassium channels, including Kv1.1, Kv2.1, Kv3.1 and BK channels (**Table 3** and **Figures 3C,D**). At 10 μ M, 1 and 3 showed substantially weaker inhibitory effects on $Ca_v1.2$, $Ca_v2.1$, and $Ca_v3.1$ (**Table 3** and **Figures 3C,D**). At 30 μ M, compound 1 had negligible effects on Kv1.1, Kv2.1, and Kv3.1 channels (**Table 3** and **Figure 3C**), although both compounds 1 and 3 showed weak inhibition on the BK channel (**Table 3** and **Figure 3D**).

In a positive control experiment, we tested the effect of two known N-type VGCC inhibitors, the divalent cation cadmium (Cd²⁺), which is a broad pore blocker of VGCCs, and the marine cone snail peptide toxin ω -conotoxin MVIIA, which is an N-type VGCC antagonist. As expected, both molecules inhibited Ca_v2.2 currents in *Xenopus* oocytes and HEK 293T cells (**Supplementary Figures S27, S28**).

In summary, our study shows that valepotriates, one of the main constituents of the V. jatamansi, can inhibit $Ca_v2.2$ N-type VGCCs. This activity is consistent with the analgesic effect of V. jatamansi in alleviating abdominal distention and pain. Further bioassay-guided search in V. jatamansi may yield more compounds that target N-type VGCCs.

AUTHOR CONTRIBUTIONS

F-WD and H-HJ are the co-first authors responsible for making experiments, consulting literature, and writing article. LY, YG,

and C-TZ are responsible for editorial assistance with this manuscript. DY, C-JY, and HL are responsible for participating in plasma sample preparation. JZ took part in the topic selection and the experiment advising. JY, YN, and J-MH are corresponding authors of this article will be responsible for conducting the research and all correspondence with the editorial and accept the consultation of the reader.

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SUPPLEMENTARY MATERIAL

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Conflict of Interest Statement: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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