

Article

Photoluminescence of MoS₂ on Plasmonic Gold Nanoparticles Depending on the Aggregate Size

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ABSTRACT: Transition metal dichalcogenides (TMDs) are promising candidates for ultrathin functional semiconductor devices. In particular, incorporating plasmonic nanoparticles into TMD-based devices enhances the light-matter interaction for increased absorption efficiency and enables control of device performance such as electronic, electrical, and optical properties. In this heterohybrid structure, manipulating the number of TMD layers and the aggregate size of plasmonic nanoparticles is a straightforward approach to tailoring device performance. In this study, we use photoluminescence (PL) spectroscopy, which is a commonly employed technique for monitoring device performance, to analyze the changes in electronic and optical properties depending on the number of MoS_2 layers and the size of the gold nanoparticle (AuNP) aggregate under nonresonant and resonant excitation conditions. The PL intensity in monolayer $MoS_2/AuNPs$ increases



as the size of aggregates increases irrespective of the excitation conditions. The strain induced by AuNPs causes a red shift, but as the aggregates grow larger, the effect of p-doping increases and the blue shift becomes prominent. In multilayer $MoS_2/AuNPs$, quenched PL intensity is observed under nonresonant excitation, while enhancement is noted under resonant excitation, which is mainly contributed by p-doping and LSPR, respectively. Remarkably, the alteration in the spectral shape due to resonant excitation is evident solely in small aggregates of AuNPs across all layers.

INTRODUCTION

Two-dimensional (2D) semiconducting materials have unique quantum physical properties, such as large exciton binding energy and a direct electronic band gap, owing to their atomically thin molecular structure. Hence, these materials allow us to design functional quantum electronic and photonic devices with ultrathin thickness. Specifically, 2D transition metal dichalcogenides (TMDs) exhibit extraordinary electronic, optical, and mechanical properties, making them primarily employed in optoelectronic and photonic device applications.^{1–3}

However, 2D TMDs have relatively low absorption efficiency due to their thin thickness, which limits the generation of photoexcited carriers. Since the resulting low photocurrent and photoluminescence (PL) restrict the overall performance of the device, the practical application of TMDs is inefficient. As a promising strategy for reducing absorption loss, combining TMDs with nanophotonic devices, such as meta-⁴⁻⁹ and plasmonic¹⁰⁻¹⁷ structures, has garnered significant attention. The heterohybrid structures have the potential to offer new functionalities and exploit synergistic effects through material interactions, which can highly improve the device performance and control the optical, electronic, and electrical properties.

2D TMDs integrated with plasmonic nanoparticles are one of the most beneficial heterohybrid structures.^{14–17} The plasmonic nanoparticles contribute to the control of device performance by enhancing the absorption efficiency through plasmonic light-matter interactions and providing additional charge channels and local strain. By placing nanoparticles underneath the 2D TMDs, it becomes feasible to induce the strain. The magnitude and direction of the strain applied to them can be regulated by exercising control over the size, shape, and arrangement of the nanoparticles. The induced strain to 2D TMDs leads to alternations in the crystal lattice with modified interatomic interactions, which give rise to the changed electronic and optical properties.¹⁸⁻²¹ Furthermore, plasmonic nanoparticles can also change the electrical properties of adjacent materials at contact interfaces. The band alignment that arises during contact introduces changes in the charge transport behavior and the redistribution of the

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Figure 1. Characterization of the $MoS_2/AuNPs$ hybrid structure. (a) Optical image of $MoS_2/AuNPs$ on the SiO_2/Si substrate. (b) Relative reflectance spectra of MoS_2 and AuNPs. The inset shows the SEM image of AuNPs. (c) Contact potential difference (CPD) image of $MoS_2/AuNPs$. The inset shows the CPD image of single AuNPs. (d) Layer-dependent CPD of MoS_2 and $MoS_2/AuNPs$.

electric charge within the system.^{22–25} In addition, under resonant optical excitation, the interaction between photons and plasmonic nanoparticles gives rise to a localized surface plasmon resonance (LSPR) effect. This plasmonic excitation is accompanied by nonradiative energy transfer processes such as hot energy transfer and plasmon resonant energy transfer, which enhance carrier generation and plasmonic field.^{26–28}

Recently, several reports on heterohybrid-structured devices have attracted attention, in which TMDs are integrated with plasmonic nanoparticles. In these studies, the design and characterization processes play pivotal roles in optimizing the device and improving our understanding of materials by controlling and evaluating their performance. In the design step, modifying the material or shape of device components is a widely used approach to fabricate devices with the desired performance by leveraging alterations in electrical and optical responses resulting from changes in their physical properties.^{29–33} However, while adjusting the layer number of TMDs and the aggregate size of nanoparticles is another straightforward approach to control their properties, there is still a lack of studies due to the intricate nature associated with interactions between layers or particles. In the characterization step, PL spectroscopy is useful for analyzing changes in electronic structures caused by local strain and electrical doping and is, therefore, one of the most adopted analysis methods.¹ Especially in plasmonic heterohybrid structures, device performance varies depending on the interaction between the plasmonic nanoparticle and light, so the energy of the excitation light is a crucial controlling factor. Therefore, comparing the PL response under nonresonant and resonant

conditions offers additional analytical insights for a better understanding and utilization of the impact of LSPR on device performance.

In this work, we investigated the optical properties of monolayer (1L), bilayer (2L), and trilayer (3L) MoS_2 on both small (<5 particles) and large (>40 particles) aggregates of plasmonic gold nanoparticles (AuNPs); see the Supporting Information, Figure S1. To characterize the optical response of this heterohybrid structure, we measured PL spectra in the energy range of 1.57-1.93 eV using excitation lasers with energies of 1.96 eV (resonance) and 2.33 eV (nonresonance). The PL intensity in 1L MoS₂/AuNPs increases with a higher concentration of AuNPs, regardless of the excitation energy. This trend holds valid for 2L and 3L MoS₂/AuNPs under resonant excitation conditions. Conversely, the PL spectra in 2L and 3L MoS₂/AuNPs exhibit more prominent quenching at larger aggregates of AuNPs under nonresonant excitation conditions. Differences in the spectral shape due to excitation conditions are observed only in MoS₂ on small aggregates of AuNPs. Under resonant excitation conditions, it becomes broader due to the excitonic emission of higher order charge complexes such as trions and biexcitons with lower energies.

RESULTS AND DISCUSSION

Optical and Electrical Characterization of MoS_2/AuNP Heterohybrid Structures. Figure 1a displays the color optical image of the $MoS_2/AuNP$ heterohybrid structures on the SiO₂/Si substrate. As the number of MoS_2 layers increases, the reflection of red light decreases, allowing for the distinction between 1L, 2L, and 3L MoS_2 through changes in the color.



Figure 2. Photoluminescence (PL) of the $MoS_2/AuNPs$ hybrid structure. (a) Integrated PL intensity image under nonresonant excitation at 2.33 eV. Nonresonance PL spectra of (b) 1L, (c) 2L, and (d) 3L MoS_2 depending on the size of AuNP aggregates. (e) Integrated PL intensity image under resonant excitation at 1.96 eV. Resonance PL spectra of (f) 1L, (g) 2L, and (h) 3L MoS_2 depending on the size of AuNP aggregates.



Figure 3. Electronic band structure of (a) 1L $MoS_2/AuNPs$ and (b) 3L $MoS_2/AuNPs$. Because the A exciton emission occurs at the K-point, a pronounced doping effect is observed in 1L MoS_2 , where the conduction band minimum is located at the K-point, but not in 3L MoS_2 .

Figure 1b shows the optical reflectance spectra observed in MoS_2 and AuNPs. The SiO_2/Si substrate was used as a reference surface. In MoS_2 , the reduction in reflection is attributed to absorption occurring at 1.90 and 2.04 eV, corresponding to the A and B excitonic emission energies, respectively. The spectrum observed in the ring-shaped AuNPs with a diameter of 100 nm exhibits a spectral dip spanning from 1.60 to 2.10 eV, with the center located at 1.85 eV. This spectral dip arises from the absorption caused by LSPR.

The change in the doping level of MoS_2 due to contact with AuNPs was investigated using Kelvin probe force microscopy (KPFM) as shown in Figure 1c. The contact potential difference (CPD) decreases as the number of layers increases in MoS₂ on the SiO₂/Si substrate. MoS₂/AuNPs exhibit a lower CPD compared to MoS₂, which indicates that the contact with AuNPs induces p-doping.^{16,34-36} In particular, pdoping is observed in regions where strain is applied, leading to a weak doping effect in single AuNPs with a simple structure but a strong doping effect in large aggregates of AuNPs with a complex structure. This doping effect of MoS₂/AuNPs is most prominent in 1L rather than in 2L and 3L as presented in Figure 1d. This is because the potential difference between 1L MoS₂ and Au is larger than that between 2L or 3L MoS₂ and Au, and electrons can transfer more easily for lower layer numbers of MoS₂.³⁷

AuNP Aggregate Size-Dependent Photoluminescence of MoS₂. Figure 2a illustrates the total PL intensity image of the MoS₂/AuNPs heterohybrid structures obtained by integrating over the spectral range from 1.57 to 1.93 eV under nonresonant excitation at 2.33 eV. It exhibits a decrease in PL intensity as the number of MoS₂ layers increases, enabling a clear distinction between MoS₂ layer numbers, while revealing almost no change in PL intensity associated with AuNPs except for 1L MoS₂. Enhanced PL is only prominent in 1L MoS₂/AuNPs (see Figure 2b) and the 3L MoS₂/Au-masked structure. Conversely, 2L and 3L MoS₂/ AuNPs exhibit either the same or quenched PL, depending on the size of AuNP aggregates as shown in Figure 2c,d. The same process was performed under resonance excitation conditions. Figure 2e displays an image representing the integrated PL intensity under excitation at 1.96 eV. Because the energy of the exciting laser is close to the energy of an allowed electronic absorption, the resonance Raman spectrum is measured in the energy range of 1.87 eV and more, along with the PL spectrum; see the Supporting Information, Figure S2. In contrast to the case of nonresonant excitation, the 1.96 eV laser excites the LSPR, which enhances the PL intensity of MoS₂/ AuNPs across all layers. Even in multilayers, higher enhancements, rather than quenching, are observed at large aggregates of AuNPs as shown in Figure 2f-2h. In addition to alterations in the PL intensity, changes in the peak position and spectral width are observed. In 1L MoS₂/AuNPs, a broader spectrum is observed at lower energy when the aggregates of AuNPs are small, while a sharper spectrum is seen at higher energy when the aggregates of AuNPs are large as presented in Figure 2b,f. However, even with large aggregates of AuNPs, the PL spectra of multilayer MoS₂/AuNPs exhibit emissions at lower energy as shown in Figure 2c,d,2g,2h.

The changes in the PL spectra of MoS₂/AuNPs are primarily influenced by strain, doping, and LSPR. Figure 3a depicts the schematic electronic band structure of 1L MoS₂/AuNPs. For smaller aggregates of AuNPs, it is difficult to expect spectral changes due to the low degree of p-doping of MoS₂. However, because the induced biaxial strain deforms the conduction band minimum at the K-point, a funnel effect occurs.^{19,20} In particular, under LSPR excitation conditions, additional electrons and excitons are generated in MoS₂ by plasmoninduced charge transfer and field enhancement.²⁶⁻²⁸ Because the funnel effect, extra charge, and excitons promote the formation of higher order charge complexes, higher PL emission at lower energies can be observed.^{38,39} With large aggregates of AuNPs, MoS₂ experiences a substantial p-doping effect (see Figure 1d), resulting in a decrease in electron density and a reduction in band gap renormalization.^{40,41} Moreover, the decrease in the electron density of MoS₂ reduces the electrical conductivity and associated reflectivity, consequently enhancing the PL intensity. The resultant PL exhibits no discernible blue shift when compared to the absence of AuNPs, attributed to a reduction in band gap energy caused by strain. Despite the increase in excitonic emission, there was no corresponding increase in the emission of higher order charge complexes by the funnel effect and LSPR.

Multilayer MoS_2 has an indirect band gap because its conduction band minimum is at the Λ (also denoted as Q or Σ) point, and its valence band maximum is at the Γ point as illustrated in Figure 3b. Because the electron density mainly decreases at the Λ point due to p-doping, there is no significant change in the electron density at the K-point, making it difficult to expect a noticeable variation in A exciton emission.⁴² When electrons are excited by light, they preferentially occupy the Λ state, where a radiative indirect transition can occur, but a non-negligible portion transitions to the K state, leading to a radiative direct transition.⁴³ Under nonresonant excitation conditions, AuNPs induce p-doping in MoS₂, leading to quenching in PL emission that scales



Figure 4. Nonresonance and resonance PL spectra of (a) 1L MOS_2 , (b) 1L MOS_2 on small aggregates of AuNPs, (c) 1L MOS_2 on large aggregates of AuNPs, (d) 2L MOS_2 , (e) 2L MOS_2 on small aggregates of AuNPs, (f) 2L MOS_2 , (e) 2L MOS_2 , (h) 3L MOS_2 on small aggregates of AuNPs, (g) 3L MOS_2 , (h) 3L MOS_2 on small aggregates of AuNPs, (a) 3L MOS_2 on large aggregates of AuNPs. Significant changes in the spectral shape are observed only in small aggregates of AuNPs.

proportionally with the size of AuNP aggregates. However, in the multilayer MoS_2/Au -masked structure, enhancement is observed rather than quenching. This is because of the PL of the Au sidewall at 1.95 eV; see the Supporting Information, Figure S3. Under resonant excitation conditions, despite the induced p-doping, the PL emission shows proportionality to the size of AuNP aggregates due to efficient light absorption and plasmon-induced charge transfer by AuNPs. Contrary to the blue shift of PL observed in 1L MoS₂/AuNPs resulting from substantial p-doping, multilayer MoS₂/AuNPs exhibit a red shift primarily attributed to strain because the alteration in charge density at the K-point due to doping is small.

Resonant and Nonresonant Excitation Effects in the Photoluminescence Spectral Shape. To analyze the PL spectral shape under different excitation conditions, normalized PL spectra for nonresonant and resonant excitations are plotted based on the number of MoS_2 layers and the size of AuNP aggregates, as shown in Figure 4. Interestingly, in the absence (see Figure 4a,4d,4g) or large aggregates (see Figure 4c,4f,4i) of AuNPs, little change in the spectral shape was observed, whereas in the small aggregates of AuNPs (see Figure 4b,e,4h), an increase in relative PL emission at lower energy was observed. These differences may occur due to local strain. The strain in MoS_2 , induced by small aggregates of AuNPs, primarily originates from the height difference between the AuNPs and the substrate. Because of the small amount of AuNPs, the strain area is narrow, concentrating the funnel effect. In MoS_2 with large aggregates of AuNPs, additional strain can be induced by the height disparity among the AuNPs. In this scenario, while stronger strain may be induced, it results in a larger strain area and a dispersed funnel effect. The concentration of the funnel effect provides an opportunity for the extra charge or exciton to form more trions or biexcitons.

CONCLUSIONS

In conclusion, we analyzed the PL spectra of $MoS_2/AuNPs$, considering the layer number of MoS_2 and size of AuNP aggregates under both nonresonant and resonant excitation conditions. In 1L $MoS_2/AuNPs$, the PL intensity proportional to the size of AuNP aggregates was observed regardless of the excitation conditions. However, depending on the size of the aggregates, the peak shift showed an opposite tendency. In the case of small aggregates of AuNPs, a spectral red shift was observed, primarily resulting from induced strain, while in large aggregates, a blue shift was observed mainly due to p-doping. The changes in PL intensity of 2L and 3L $MoS_2/AuNPs$ were different depending on the excitation conditions. As the size of AuNP aggregates increases, the PL intensity was quenched mainly because of the p-doping, until a direct band gap is formed by strain in MoS2 under nonresonant excitation conditions. On the other hand, under resonant excitation conditions, the PL intensity was enhanced by LSPR. Under resonant excitation conditions, the PL intensity experienced enhancement attributed to the heightened LSPR effect, directly correlated with the illuminated quantity of AuNPs. In all layers, distinct changes in the spectral shape based on excitation conditions were prominent solely in small aggregates of AuNPs. This is likely because localization by AuNPs becomes more pronounced in scenarios with fewer particles. Our results indicate the potential to manipulate the electrical and optical performance of the device by controlling the number of TMD layers and aggregate size of plasmonic nanoparticles. We believe that controlling and manipulating photoresponses through the TMDs and plasmonic nanoparticles, combined with a comprehensive understanding of these processes, will be instrumental in advancing the development of optoelectric and photonic devices.

MATERIALS AND METHODS

Synthesis of Au Nanorings. Au nanoprisms were employed as the initial materials for synthesizing gold nanorings. First, the vertices of Au nanoprisms were truncated to form thin gold nanodisks, with a height of approximately 10 \pm 1 nm, using Au³⁺ ions as the etchant. Afterward, during the thickening step, the thin Au nanodisks evolved into Au nanohexagons by depositing Au. Subsequently, the Au nanohexagons were further etched into Au thick nanodisks with a height of approximately 25 ± 2 nm. In the selective edge deposition of Pt, Pt atoms were placed selectively at the periphery of thick Au nanodisks. This decoration is facilitated by the presence of high-index crystalline facets, which lower the activation energy barrier for Pt nucleation. In the next step, the core Au domains were etched by adding Au³⁺ ions, leading to the formation of Au@Pt split frames. Eventually, in the Au regrowth step, Au atoms were homogeneously reduced across the entire surface of split Pt nanorings, resulting in the formation of Au nanorings.⁴⁴

 MoS_2 Growth and Transfer Process. The synthesis of single-layer MoS_2 was achieved through a two-zone furnace chemical vapor deposition (CVD) process. The Mo precursor was prepared by mixing 0.008 M ammonium heptamolybdate (AHM, Sigma-Aldrich, 431346), 0.125 M NaOH, iodixanol (Sigma-Aldrich, Opti Prep, D1556), and deionized (DI) water in 0.3:2:0.2:1 volumetric ratio. The solution was then spread onto a SiO₂/Si substrate, which had been pretreated with oxygen plasma (30 W for 1 min), using a spin coater at 4000 rpm for 40 s.

The coated substrate, cut into a 1×1 cm² dimension, was placed in the downstream zone of a 2 in. tube furnace. Also, 200 mg of S (Sigma-Aldrich, no. 212392) was positioned in the upstream zone. For growth, the temperature of the downstream zone was modified to 190 °C, while the upstream zone was brought up to 780 °C. During this ramping period, which lasted for 12 min, a N₂ gas flow was maintained at 300 standard cubic centimeters per minute (sccm). This flow rate was then increased to 1000 sccm and maintained for an additional 7 min to facilitate the growth of the MoS₂. The entire process occurred at ambient pressure, and a purging step at the onset was determined to be unnecessary. After the growth process, the MoS_2 layer was transferred by using the PMMA method. The MoS_2 was first coated with PMMA, spun at 4000 rpm for 40 s. It was then soaked in a 0.125 M NaOH solution to separate the MoS_2 layer from its growth substrate. Following this, it was rinsed with DI water to eliminate any NaOH residue. Finally, the PMMA/MoS_2 layer was transferred onto a SiO_2/Si substrate, which had been predeposited with gold nanoparticles followed by acetone and isopropyl alcohol (IPA) rinsing to remove the PMMA binder.

Sample Characterization. In optical characterization, the illumination source was focused on the sample via a $50 \times$ objective lens (Nikon, TU Plan Epi ELWD, NA 0.60). The reflected optical signal and photoluminescence (PL) were recorded and analyzed by a spectrometer (Princeton Instruments, SpectraPro 2300i, 150 lpmm) equipped with a CCD (Princeton Instruments, ProEM: 1600²). For reflectance spectroscopy, a white LED was used as the light source. The exposure time of each spectrum was set to 100 ms. For PL spectroscopy, a 532 nm DPSS laser (Optoelectronicstech, MGLIII-532) and a 632.8 nm He-Ne laser (Thorlabs, HNL210LB) are utilized as nonresonant and resonant excitation sources, respectively. The excitation power was set to 200 μ W after the objective lens, and the optical path was controlled by a 2D galvo mirror system (GVSM002, Thorlabs, Inc.) for 2D PL mapping.

In electrical characterization, Kelvin probe force microscopy was employed to measure the surface potential of $MOS_2/AuNPs$ heterohybrid structures. The topographic and surface potential images were recorded simultaneously via commercial scanning probe microscopy (Parksystems, XE-NSOM) with an Au-coated cantilever (MikroMasch, NSC-14–Cr-Au). EFM (EXT) mode was used for the measurement, and the external lock-in amplifier (Stanford Research Systems, SR830) was set to a time constant of 300 μ s, sensitivity of 5 mV, and AC voltage of 0.45 V. The scan was performed at a speed of 0.01 line/s over 60 μ m × 60 μ m (512 pxl × 512 pxl).

ASSOCIATED CONTENT

G Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsomega.4c02442.

Optical images of spectral test points, resonance Raman spectra of MoS_2 , photoluminescence spectra of the gold-masked structures, and photoluminescence of $MoS_2/$ AuNPs at different positions (PDF)

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Notes

The authors declare no competing financial interest.

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