1,2-Dithiol-3-thione and dithioester analogues: potential radioprotectors

B.A. Teicher¹, J. Stemwedel¹, T.S. Herman^{1,2}, P.K. Ghoshal³ & A. Rosowsky¹

 1 Dana-Farber Cancer Institute, and 2 Joint Center for Radiation Therapy, 44 Binney Street, Boston, MA 02115, USA; and 3Macrochem Corporation, Billerica, MA 01821, USA.

> Summary Several 1,2-dithiol-3-thione and dithioester compounds were assayed for radioprotective capabilities in EMT6 cells in vitro. The 1,2-dithiol-3-thiones were generally more cytotoxic than the dithioesters and in some instances were more cytotoxic toward hypoxic cells than toward normally oxygenated cells. When the drugs were present at a concentration of 500μ M for 1 h prior to and during radiation delivery, the 5-(2-thienyl)-1,2-dithiol-3-thione produced a radiation protection factor (RPF) of 2.7 at ¹ log of cell kill. The 4-methyl analogue of this same compound was, however, much less effective, producing a RPF of only 1.2. The 4-ethoxycarbonyl analogue was moderately active, producing ^a RPF of 1.7. The 4-methyl-5-(2-pyrazinyl)- 1,2-dithiol-3-thione (Oltipraz) was least effective, yielding ^a RPF of only 1.1. Of the dithioesters tested, methyl 3-pyrrolidino-2-phenylpropene dithiocarboxylate produced a RPF of 2.6, methyl 3-piperidino-2 phenylpropenedithiocarboxylate a RPF of 2.7, and the corresponding 3-morpholino and 3-thiomorpholino derivatives RPF values of 2.7 and 2.9, respectively. The iodide salt of 4-ethoxycarbonyl-5-(2-thienyl)-1,2 dithiol-3-thione produced a RPF of 2.6 Methyl 3-cyclohexylamino-2-phenylpropenedithiocarboxylate was equally effective (RPF = 2.6). Finally, methyl 3-morpholino-3-thienyl-2-methylpropenedithiocarboxylate and methyl 3-morpholino-3-(2-pyrazinyl)-2-methylpropenedithiocarboxylate were less effective, producing RPF values of 2.0 and 1.6, respectively. These results demonstrate that several of these compounds are highly effective radioprotectors. In vitro and in vivo studies testing their efficacy are underway.

In the search for methods to improve cancer therapy, one experimental strategy that has been somewhat successful in recent years is the use of normal tissue protectors. In the case of radiation, these protective agents may improve the therapeutic index by reducing toxicity and allowing higher doses of radiation to be delivered to a tumour in the vicinity of sensitive normal tissue. A large body of data exists on the use of thiol compounds as both radiation and chemoprotective drugs. One such agent, WR-2721 (S-2-(3-aminopropylamino)ethylphosphorothioic acid) (Yuhas & Storer, 1969; Yuhas et al., 1977, 1982), is currently undergoing clinical trial (Constine et al., 1986; Glover et al., 1989). The monoethyl ester of glutathione, an uncharged analogue of the most abundant glutathione non-protein sulphydryl compound in cells, also shows potential application as a normal tissue protector (Teicher et al., 1988).

In 1982, Bueding noted that treatment with the antischistosomal agent 4-methyl-5-(2-pyrazinyl)-1 ,2-dithiol-3 thione (Oltipraz) raised the cellular thiol levels in several tissues in mice (Bueding et al., 1982). Oltipraz and other 1,2-dithiol-3-thiones have been shown to be potent inducers of enzymes involved in maintaining levels of reduced glutathione (De Long *et al.*, 1986; Davies *et al.*, 1987; Kensler et al., 1987). Specifically, administration of Oltipraz or related compounds has been shown to produce increases in hepatic glutathione levels and glutathione-S-transferase activities and to provide protection from the hepatotoxicity of tetrachloromethane and acetaminophen (Ansher et al., 1986; Davies et al., 1987). This class of compounds also has potent anti-carcinogenic capacity providing protection against hepatic tumorigenesis from aflatoxin B_1 (Kensler *et* al., 1987).

The present studies were undertaken to examine the cytotoxicity and potential of a series of 1,2-dithiol-3-thiones and related dithioesters to act as radioprotective agents in normally oxygenated and hypoxic EMT6 mouse mammary tumour cells in vitro. Some of these compounds proved highly effective protectors of normally oxygenated cells whereas others were fair protectors of normally oxygenated cells, but paradoxically also radiosensitised hypoxic cells.

Correspondence: B.A. Teicher. Received 17 October 1989; and in revised form 9 January 1990.

Materials and methods

Drugs

The 1,2-dithiol-3-thiones and dithioesters shown in Table ^I were synthesised by the general method reported earlier (Foye et al., 1987). Details of the chemistry of these compounds will be published later.

Cell line

The EMT6 mammary tumour cell line has been widely used for the study of antineoplastic agents (Rockwell $&$ Kallman, 1973; Rockwell et al., 1972; Teicher et al., 1981, 1985). The experiments described here were performed using asynchronous EMT6 cells in monolayers in exponential growth in Waymouth's medium supplemented with antibiotics (Grand Island Biological Co., Grand Island, NY, USA) and 15% newborn calf serum (Hyclone Laboratories, Logan, UT, USA). This cell line has a plating efficiency of 65-80% and a doubling time of $16-19$ h in vitro (Rockwell et al., 1972).

Drug cytotoxicity

EMT6 cells were placed into T-25 flasks (Falcon Plastics). Each flask contained approximately 1×10^6 cells at the time of treatment. The drugs, in a small volume of sterile 0.9% phosphate-buffered saline, were added to the cells in complete medium in concentrations ranging from 5 to 500 μ M.

Production of hypoxia

To produce hypoxia, flasks containing cells in complete medium plus serum were fitted with rubber sleeve serum stoppers and exposed to ^a continuously flowing 95% nitrogen/5% $CO₂$ humidified atmosphere for 4 h at 37°C. Parallel flasks were maintained in 95% air/5% CO₂. At this time, the drug (0.10 ml) or vehicle $(0.9\% \text{ PBS}, 0.10 \text{ ml})$ was added to the flasks by injection through the rubber stopper without disturbing the hypoxia (Teicher et al., 1981; Teicher & Holden, 1987).

Radiation treatment

For studies involving radiation, the flasks were placed in an insulated chamber filled with 95% nitrogen/5% CO₂ (hypoxic) or 95% air/5% $CO₂$ (normally oxygenated) atmo-

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^aCells in exponential growth were exposed to the drugs for 1 h and survival was measured by colony formation. IC₅₀ is the drug concentration which reduced the survival of the cells by 50% ; "Protection factor is defined as the ratio of the X-ray dose with protector to the X-ray dose without protector at the level of ^I log of cell killing.

spheres. The cells were irradiated with a $137Cs$ radiation unit (Gammacell 40, Atomic Energy of Canada Ltd) at a dose rate of 1.05 Gy min-'. Drug-treated cells were irradiated so that the irradiation was completed ¹ h after addition of the drug. X-ray doses of 2.5, 5.0, ¹⁰ and ¹⁵ Gy were used. A drug concentration of 500 μ M was used so that there would be only limited toxicity from the 1,2-dithiol-3-thiones and dithioesters.

Cell viability measurements

Cell viability was measured by the ability of single cells to form colonies in vitro, as described previously (Teicher et al., 1981, 1985). Following treatment, suspensions of known cell numbers were plated in plastic Petri dishes and allowed to grow in a 37°C incubator under standard culture conditions for 8-10 days. After this time, macroscopic colonies were stained with crystal violet in methanol containing 3.7% formaldehyde and were counted manually. Each experiment was repeated three to five times, and each data point per experiment represents the results of three different dilutions of cells plated in triplicate.

Calculated value

The radioprotection factor (RPF) was calculated as the ratio of the dose of radiation to kill ¹ log of normally oxygenated cells with the drug present to dose of radiation without drug to kill ¹ log of normally oxygenated cells corrected for the cytotoxicity of each drug.

Results

In general, ¹ h exposure to the substituted 1,2-dithiol-3 thiones and dithioesters had a relatively low level of cytotoxicity toward both normally oxygenated and hypoxic EMT6 cells as measured by the IC_{50} (Table I). The 4-methyl-5-(2thienyl) analogue (compound 2) was slightly more cytotoxic towards hypoxic EMT6 cells. The 2-ethoxycarbonyl analogue of compound 2 (compound 3) and Oltipraz (compound 4) were 16-fold and 5-fold more cytotoxic toward hypoxic than normally oxygenated EMT6 cells. In contrast, of the dithioesters, only the 3-morpholino-3-(2-thienyl) analogue (compound 11) was significantly more cytotoxic toward hypoxic EMT6 cells than toward normally oxygenated EMT6 cells.

Radiation survival curves for normally oxygenated and hypoxic EMT6 cells in vitro in the presence and absence of 1,2-dithiol-3-thiones are shown in Figure 1. The concentration of each compound used for these radiation studies was 500μ M and the agents were present for 1 h prior to and during radiation delivery. The curves are corrected for the small amount of drug killing. The simple 5-(2-thienyl)-1,2 dithiol-3-thione (compound 1) was an excellent radiation protector of normally oxygenated EMT6 cells, affording ^a protection factor of 2.7 at ¹ log of cell killing. The 4-methyl derivative of compound *I* (compound 2) was a much less effective radiation protector, with a protection factor of only 1.2; however, compound 2 increased the cytotoxicity of the radiation treatments in hypoxic EMT6 cells. It appeared that the main effect of compound 2 on the radiation response of hypoxic EMT6 cells was to eliminate the shoulder of the radiation survival curve. The 4-ethoxycarbonyl derivative

Figure 1 Radiation survival of EMT6 mouse mammary tumour cells in the presence of 1,2-dithiol-3-thiones (compounds $1-4$ from Table 1). No drug, normally oxygenated cells $(①)$; no drug, hypoxic cells $(①)$; 500 μ M drug, normally oxygenated cells $(②)$; ⁵⁰⁰ gM drug, hypoxic cells (0). The data are presented as means ± SEM (bars) for three independent experiments.

(compound 3) produced a good level of radiation protection in normally oxygenated EMT6 cells, with ^a protective factor of 1.7. Compound 3, like compound 1, had no effect on the radiation survival of hypoxic EMT6 cells. The 4-methyl-5-(2 pyrazinyl)-1,2-dithiol-3-thione (compound 4; Oltipraz), like the other 4-methyl derivative, compound 2, increased the killing of hypoxic EMT6 cells by radiation. The main effect of compound 4 in hypoxic EMT6 cells also appeared to be elimination of the shoulder of the radiation survival curve, although there was a small increase in the slope of the hypoxic radiation survival curve as well.

Radiation survival curves of normally oxygenated and hypoxic EMT6 cells in the presence and absence of dithioesters are shown in Figure 2. Methyl 3-(1-pyrrolidino)- 2-phenylpropenedithiocarboxylate (compound δ) was an excellent radioprotector of normally oxygenated EMT6 cells, with a protection factor of 2.6, but had no effect on the radiation survival of hypoxic EMT6 cells. Methyl 3-(1 piperidino)-2-phenylpropenedithiocarboxylate (compound 7) was also an excellent radiation protector of normally oxygenated EMT6 cells, with ^a radiation protection factor of 2.7, but again there was no significant effect on the radiation

Figure 2 Radiation survival of EMT6 mouse mammary tumour cells in the presence of dithioesters (compounds $6-9$ from Table 1). No drug, normally oxygenated cells (\bullet) ; no drug, hypoxic cells (O) ; 500 μ M drug, normally oxygenated cells (\bullet) ; 500 μ M drug, hypoxic cells (\square). The data are presented as means \pm SEM (bars) for three independent experiments.

survival of hypoxic EMT6 cells. The 3-morpholino derivative (compound δ) similarly produced excellent radiation protection in normally oxygenated EMT6 cells, with ^a protection factor of 2.6. Unlike compounds 6 and 1 , compound 8 increased the cytotoxicity of radiation towards hypoxic EMT6 cells to ^a small extent. The 3-thiomorpholino derivative (compound 9) was the most effective radioprotector of normally oxygenated EMT6 cells tested, with ^a protection factor of 2.9, and compound 9 did not alter the radiation response of hypoxic EMT6 cells.

Radiation survival curves for normally oxygenated and hypoxic EMT6 cells in the presence and absence of the remaining compounds are shown in Figure 3. Compound 5, the salt formed by the reaction of methyl iodide with the corresponding 1,2-dithiol-3-thione, was a highly effective radioprotective agent of normally oxygenated EMT6 cells, with a protection factor of 2.6 Compound 5 had no effect on the radiation survival of hypoxic EMT6 cells. Methyl 3 cyclohexylamino-2-phenylpropenedithiocarboxylate (compound 10) also had a highly effective radiation protection factor of

Figure 3 Radiation survival of EMT6 mouse mammary tumour cells in the presence of a 1,2-dithiolium salt (compound 5 from Table 1) and of substituted dithioesters (compounds $10-12$ from Table 1). No drug, normally oxygenated cells (\bullet); no drug, hypoxic cells (O); 500 μ M drug, normally oxygenated cells (\blacksquare); 500 μ M drug, hypoxic cells (\square). The data are presented as means ± SEM (bars) for three independent experiments.

2.6 while producing no change in the radiosensitivity of hypoxic EMT6 cells. Methyl 3-morpholino-3-(2-thienyl)-2 methylpropenedithiocarboxylate (compound 11) was an effective radiation protective agent, with a radioprotective factor of 2.0, but again did not alter the radiation sensitivity of hypoxic EMT6 cells. Methyl 3-morpholino-3-(2-pyrazinyl)- 2-methylpropenedithiocarboxylate (compound 12) was a moderately effective radioprotective agent, with a resulting radiation protection factor of 1.6 However, this compound also increased the radiosensitivity of hypoxic EMT6 cells, although to a lesser extent.

Discussion

1,2-Dithiol-3-thiones have demonstrated chemoprotective actions against a wide variety of structurally diverse carcinogens (Wattenberg & Bueding, 1986; Kensler et al., 1987; Helmes et al., 1989). These compounds have been used medicinally as antischistosomal agents, choleretics, and to stimulate salivary secretion (Lozac'h & Stavaux, 1980; Bueding et al., 1982; Hausler, 1979; Archer, 1985).

We have examined the radioprotective potential of four 1,2-dithiol-3-thiones and seven dithioesters which are chemically derived from dithiolthiones. Two of the 1,2-dithiol-3 thiones were effective radioprotectors of normally oxygenated EMT6 cells and two showed very little radioprotective

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activity. Both ineffective compounds had 4-methyl substituents and both of these agents increased the radiation sensitivity of hypoxic EMT6 cells quite markedly. It may be that having a hydrogen in the 4-position allows ring opening of the 1,2-dithiol-ring to occur providing a substantial increase in non-protein sulphydryl within cells and thus radioprotection. A methyl group in the 4-position, on the other hand, may block the ring opening, leading to an inactive molecule. The mechanism for the hypoxic cell selectivity of compounds 3 and 4 remains to be elucidated. The dithioesters were, in general, less cytotoxic than the 1,2-dithiol-3-thiones but also highly effective radiation protective agents. It may be envisioned that oxidative metabolism of the dithioesters could lead to the release of a thiomethylating species, a potent radical scavenger.

Since most normal tissues are well oxygenated, it is the normally oxygenated EMT6 cells which are the most appropriate model for a radioprotector in a cell culture system. Our work with these agents is continuing and we are examining the efficacy of selected examples as normal tissue radiation protectors in vivo.

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