

Physico-chemical and microbiological characterization of spontaneous fermentation of Cellina di Nardò and Leccino table olives

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Gianluca Bleve, Unità Operativa di Lecce, Consiglio Nazionale delle Ricerche, Istituto di Scienze delle Produzioni Alimentari, Via Provinciale Lecce-Monteroni, 73100 Lecce, Italy e-mail: gianluca.bleve@ispa.cnr.it Table olives are one of the most important traditional fermented vegetables in Europe and their world consumption is constantly increasing. In the Greek style, table olives are obtained by spontaneous fermentations, without any chemical debittering treatment. Evolution of sugars, organic acids, alcohols, mono, and polyphenol compounds and volatile compounds associated with the fermentative metabolism of yeasts and bacteria throughout the natural fermentation process of the two Italian olive cultivars Cellina di Nardò and Leccino were determined. A protocol was developed and applied aimed at the technological characterization of lactic acid bacteria (LAB) and yeast strains as possible candidate autochthonous starters for table olive fermentation from Cellina di Nardò and Leccino cultivars. The study of the main physic-chemical parameters and volatile compounds during fermentation helped to determine chemical descriptors that may be suitable for monitoring olive fermentation. In both the analyzed table olive cultivars, aldehydes proved to be closely related to the first stage of fermentation (30 days), while higher alcohols (2-methyl-1-propanol; 3-methyl-1-butanol), styrene, and o-cymene were associated with the middle stage of fermentation (90 days) and acetate esters with the final step of olive fermentation (180 days).

Keywords: table olives, yeast, lactic acid bacteria, volatile compounds, fermented food

INTRODUCTION

Olive production is considered one of the major agronomic practices of Mediterranean countries such as Italy, Greece and Spain that together supply almost 30% of world olive annual production (IOC, 2014). Table olives are one of the most important traditional fermented vegetables in Southern European countries and their consumption is constantly increasing. Italy produces about 76,000 tons/year of table olives, which is 3.1% of world and 10.3% of European Union production, respectively. Leccino, a cultivar originary from Tuscany and Umbria regions (Central Italy), is the most widespread dual-purpose olive variety in the world, due to its exceptional adaptability to different growing conditions (Vossen, 2007). Cellina di Nardò is an olive cultivar autochthonous of Salento (Apulia, Southern Italy). The production of this traditional cultivar is about 180,000 tons per year. It is mostly used to produce olive oil and in part it is used for the production of table olives that are being highly appreciated and requested outside the Salento area. Natural black olives in brine (Greek-style), Spanish-style green olives and black oxidized olives (also known as Californian-style) are the three most important commercial preparations of table olives on the international market (Garrido-Fernández, 1997). In the

Greek-style production system, the fruits are placed directly into brine with a salt concentration of about 6-10% (w/v), thus allowing spontaneous fermentation to take place (Balatsouras, 1990). Spontaneous fermentation lasts 8-12 months and is driven by mixed populations of microorganisms, mainly the epiphytic microbial populations of yeasts and lactic acid bacteria (LAB) (Garrido-Fernández, 1997; Brenes et al., 2004; Romero et al., 2004; Hurtado et al., 2008). Lactic acid fermentation by LAB is considered the key step in spontaneous fermentation processes particularly in those related to directly brined olives (green or black), since it promotes: (i) debittering of the olives through oleuropein hydrolysis, (ii) lowering of brine pH, which prevents any growth of spoilage and pathogenic microorganisms (Spyropoulou et al., 2001; Caggia et al., 2004; Cawthorne et al., 2005), (iii) the enhancement of a correct flavor and texture profile in the final product (Ciafardini et al., 1994; Garrido-Fernández, 1997; Sanchez et al., 2000). However, the important role of yeasts in the table olive production process has recently been considered (Bevilacqua et al., 2013; Heperkan, 2013). The beneficial roles of yeasts consist of: (i) the production of volatile compounds and metabolites able to improve the organoleptic characteristics of the final product (Garrido Fernández et al.,

1995), (ii) the release of nutritive compounds that enhance LAB growth (Arroyo-López et al., 2008; Nisiotou et al., 2010; Bautista-Gallego et al., 2011), (iii) activity against undesired microorganisms (Psani and Kotzekidou, 2006), and bioreduction of phenolic compounds (Ettayebi et al., 2003). On the other hand, yeasts can cause several problems during table olive production, such as the formation of gas pockets, the softening of olive tissue, package bulging, clouding of brines, and production of off-flavors (Turantas et al., 1999). At present, industrial production of black olives and several cultivars of green olives in local industries is carried out by spontaneous fermentation processes which are not predictable and are strongly influenced by the autochthonous microbiota, the physico-chemical conditions, the availability of fermentable substrates and salt content (De Castro et al., 2002; Tassou et al., 2002; Alvarez et al., 2003; Abriouel et al., 2011). Among the possible available technological approaches for controlling fermentation process, the use of Lactobacillus plantarum and Lactobacillus pentosus as starter cultures has been proposed (De Castro et al., 2002; Leal-Sánchez et al., 2003; Panagou et al., 2003, 2008; Marsilio et al., 2005; Servili et al., 2006; Sabatini and Marsilio, 2008), in order to avoid the unpredictability of spontaneous fermentations and to improve the production process. However, in recent years, the importance and the potential applications of yeasts as starters for table olive processing has been also recognized Arroyo-López et al., 2008, 2012b; Bevilacqua et al., 2013; Bonatsou et al., 2015.

The main aims of the present study were: (i) to characterize microbial population associated to Cellina di Nardò and Leccino table olive fermentations by technological and molecular approaches; (ii) to investigate the evolution of the main physico-chemical parameters and volatiles during fermentations for identifying specific molecules to be used as tools for fermentation monitoring; (iii) to correlate the outcome of physic-chemical analyses with the microbial growth.

MATERIALS AND METHODS

OLIVE SAMPLES AND FERMENTATION PROCEDURES

The lab-scale fermentations were performed in triplicate on olive samples of Cellina di Nardò and Leccino cultivars at an industrial plant belonging to Agricola Nuova Generazione (Martano, Lecce, Italy). Healthy black olives (90 kg) were collected at the black stage of ripening and washed with tap water to eliminate plant materials (residues of leaves, branches) and superficial contaminants. The olives were then selected (caliber above 10–12 mm), washed and placed in plastic vessels of 30 kg capacity filled with 20 L of 13% NaCl (wt/vol). The olives were allowed to ferment at ambient temperature, adopting, when required, correction of salinity by addition of salt.

ISOLATION OF MICROBIAL POPULATION

To isolate epiphytic yeasts, *Enterobacteriaceae* and LAB, a sample consisting of 15 olives for Leccino (32,41 g) and 20 olives of Cellina di Nardò (32,62 g) and 50 ml of brine was stirred on a rotary shaker at 200 rpm for 30 min. The sediment was recovered after centrifugation at 5000x g for 10 min at room temperature, and suspended in 0.5 ml of 0.1% (wt/vol) peptone water.

Salinity, pH, temperature and the formation of mold layers on top of the brine were evaluated during fermentation, at the following time points: 0, 5, 10, 15, 30, 45, 60, 75, 90, 105, 120, 135, 150, 165, 180 days. At each different fermentation time, olives and brines were collected, incubated together for 30 min in a rotary shaker (200 rpm) and then aliquots of brines were collected, diluted with one volume of sterile 100% glycerol and stored at -80°C for further analysis. In order to quantify LAB, Enterobacteriaceae and yeasts present in olives and brines, samples were serially diluted with 0.1% (wt/vol) peptone water and applied to agar plates containing the following media: Man, Rogosa and Sharpe Agar (MRS), added with 0.05 g/L of nystatin (incubation at 30°C under anaerobic conditions for 48–72 h); Violet Red Bile Glucose Agar (VRBGA; incubation at 37°C for 18-24 h); Sabouraud Dextrose Agar, added with 0.1 g/of ampicillin and 0.05 g/L of kanamycin (incubation at 25°C for 2–4 days). The population on each agar plate was then subjected to microbial count in order to quantify the LAB, Enterobacteriaceae and yeast concentrations in each sample. Thirty-five colonies were randomly selected from the agar plates (specific for LAB and yeasts) at the sampling times indicated above.

YEASTS AND LAB TECHNOLOGICAL CHARACTERIZATION

The set up of a protocol for the technological characterization of autochthonous starters for table olive production preliminarily required the formulation of a synthetic "model brine," in order to use it as reliable substrate for the in vitro technological assays. For this purpose, two different synthetic brines were formulated, for yeast and LAB selection, respectively. For yeast selection, brine samples deriving from each olive cultivar were collected at days 15, 30, and 60 of fermentation and analyzed, being this period representative of selective conditions introduced by olives and brine after the yeasts started to increase their number. As described by Bleve et al. (2015), the following phenolic compounds were detected and quantified in the following ranges: tyrosol (11.39–256.18 mg/L), pyrocatechol (14.69–18.14 mg/L), caffeic acid (7.54-62.11 mg/L), oleuropein (26.30-900.04 mg/L), verbascoside (7.95-422.21 mg/L). The results helped to formulate the composition of the model brine as follows: 100 mg/L tyrosol, 30 mg/L caffeic acid, 500 mg/L oleuropein, 200 mg/L verbascoside, 10% NaCl, 3 g/L glucose, 0.5 g/L yeast extract, 20 g/L agar, pH 4-4.5. Microbial isolates were grown at 15°C. For LAB selection, brine samples were collected every 2 weeks starting from day 120 until day 180. Samples were analyzed to determine the qualitative and quantitative profile of phenolic compounds by reversed-phase HPLC-DAD, as described above. The main phenolic compounds present in several brines of table olive fermentations were identified and quantified at the specific time point when the LAB appeared: tyrosol (13.68-200.42 mg/L), caffeic acid (0-32.77 mg/L), oleuropein (51.58-667.33 mg/L), verbascoside (32.82–190.47 mg/L). The concentration of each phenolic compound detected was then considered to formulate model brines: 100 mg/L tyrosol, 15 mg/L caffeic acid, 300 mg/L oleuropein, 100 mg/L verbascoside, 8% NaCl, 3 g/L glucose, 0.5 g/L yeast extract, 20 g/L agar, pH 4.2. Bacterial isolates for each of the two olive cultivars were applied on MB and incubated for 15 days at 15°C. Beta-glucosidase activity was determined by replica

plating the yeast onto selective media, for yeast [SC = 0.67%yeast nitrogen base (YNB), 0.5% arbutin, 2% agar, pH 5.0] and for bacteria (MRS medium, 0.5% arbutin, 2% agar, pH 5.0). Two milliliters of a filter-sterilized 1% ammonium ferric citrate solution were added to 100 ml medium before pouring the plates. The plates were incubated for 5 days at 30°C and colonies showing beta-glucosidase activity were identified by a brown surrounding halo. The absence of extracellular protease production was determined by replica plating yeast or bacterial colonies onto YPD plates containing 2% casein or onto MRS plates containing 2% casein, respectively. The plates were incubated for 5 days d at 25°C. A clear zone around the colony identified protease activity. Lipase activity was evaluated by replica plating yeast colonies onto agar plates containing 5% peptone, 0.5% glucose, 0.1% NaNO₂, 0.1% KH2PO4, 0.1% MgSO4, 2% Tween 80, and 0.01% rhodamine B. Lypolitic activity was determined by the presence of a darker halo surrounding the colony which was detectable after 10 days at 25°C. Biogenic amine formation was determined using a modified method of Nikolaou et al. (2006). Yeast and bacteria strains were inoculated on YPD or MRS agar plates, respectively, supplemented with bromocresol purple 0.006 and 1% (w/v) of the amino acids histidine, tyrosine, phenylalanine, tryptophan, lysine, leucine, and arginine. The plates were incubated at 25°C for 3-4 days and growth was examined daily. At the beginning, a yellow halo was observed around the colonies because of glucose fermentation, followed by pH reduction, while in the case of amino acid decarboxylation a purple halo appeared.

LAB AND YEAST MOLECULAR IDENTIFICATION

Total genomic DNA from the yeast strains was prepared according to the method used by Querol et al. (1992) and diluted to 50 ng/µl. ITS1-5,8S-ITS2 region was amplified with primers ITS1 (5'-TCC GTA GGT GAA CCT GCG G-3') and ITS4 (5'-TCC TCC GCT TAT TGA TAT GC-3') (White et al., 1990). The polymerase chain reaction (PCR) conditions were as described by Esteve-Zarzoso et al. (1999) with the following modifications: initial denaturation at 94°C for 5 min, followed by 40 cycles consisting of 30s at 94°C, 30s at 52°C and 1 min at 72°C, followed by a final extension at 72°C for 10 min and subsequently cooled to 8°C. The amplified DNA products were visualized by agarose gel electrophoresis. DNA from pelleted bacterial cells, grown for 3-5 days at 25°C under vacuum in liquid culture media, was extracted as described by Wilson (2001). The nearly full-length 16S rRNA gene was amplified for all isolates by using the Universal 16S forward primer (5'-GGAGAGTTAGATCTTGGCTCAG-3'), and Universal 16S reverse primer (5'-AGAAAG GAGGTGATCCAGCC-3'). A reaction mixture (50 μ l) containing 1 μ l (50 ng/ μ l) genomic DNA, 10× PCR buffer (Euroclone), 2 mM MgCl₂, 200 µM each dATP, dTTP, dCTP and dGTP, primers Universal 16S forward and reverse, 0.5 µM each, and 1 U DNA polymerase (Euroclone, Italy) was prepared. Genomic DNA was amplified with a 2-min denaturation step at 94°C, followed by 35 cycles of 30 s denaturation at 94°C, 30 s primer annealing at 55°C and 1 min DNA chain extension at 72°C. The PCR was completed by 5 min DNA chain extension at 72°C. DNA sequencing was performed as previously described (Bleve et al., 2011). The obtained sequences, corresponding to a total of 194 yeasts and to 140 bacteria isolates were all identified by a database similarity search in the GENBANK Collection using the BLAST software (http://www. ncbi.nlm.nih.gov/BLAST/).

HPLC ANALYSES

HPLC analyses of phenolic compounds in the brine were achieved by direct injection of filtered (through a 0.45 µm filter) brine into the chromatographic system. The HPLC apparatus consisted of an Agilent 1100 equipped with a photodiode array detector. The wavelengths used for quantification of phenol compounds were 280, 295, and 320 nm. Separation was achieved according to Li et al. (2008) with some modifications using a Phenomenex Luna 5μ C18 (2) 100 Å column (250 × 4.6 mm), with the temperature of the column set to 30°C. A gradient elution program was utilized with a mobile phase consisting of acetonitrile (solution A) and 1% (v/v) phosphoric acid in water (solution B) as follows: isocratic elution, 5% B, 0-30 min; linear gradient from 5 B to 15% B, 30-50 min; linear gradient from 5 B to 50% B, 50-55 min; linear gradient from 50 B to 100% B, 55-65 min; post time, 10 min before the next injection. The flow rate during the mobile phase was 1.0 ml/min, and the injection volume was 20 µl. All phenol compounds were quantified using calibration curves of authentic phenolic standards. Quantification of sugars, organic acids and alcohols in the brine was achieved by directly injecting the brine (filtered as stated above) into the chromatographic system (Agilent 1100) equipped with a RID-10A refractive index detector (for sugar and alcohol analysis) and with a photodiode array detector set to 210 nm (for the analysis of organic acids). Sugars, organic acids, and alcohols were simultaneously analyzed, according to De Benedictis et al. (2011), on an Aminex HPX-87H column (300×7.8 mm) (Bio-Rad) and kept at 55°C. The analytical conditions used were as follows: flow 0.3 ml/min, eluent 0.045 N H₂SO₄ with 6% acetonitrile (v/v).

VOLATILE COMPOUND EXTRACTION FROM OLIVE FRUIT AND BRINES

Volatile compounds of fermentations were identified by Solid-Phase Micro-Extraction technique in Head Space followed by Gas Chromatography/Mass Spectrometry (HS-SPMEGC/MS) (Pawliszyn, 1997). Analysis of samples was carried out by homogenizing 5 g of stoned drupes and transferring into 20-ml vials covered with a polytetrafluoroethylene (PTFE)/silicone rubber septum containing a micro stirring bar. 8 µL of 4-methyl-2pentanol in methanol (final concentration 50 µg/ml) were used as an internal standard. The 65 µm DVB/CAR/PDMS coated fiber (Supelco, Spain) was used according to Malheiro et al. (2011). The vials were heated to a controlled temperature $(40 \pm 0.5^{\circ}C)$ in order to reach equilibrium. On the basis of preliminary tests, 30 min exposure time proved suitable for fiber saturation and for reproducibility of the extraction procedure. Before the first daily analysis, the fiber was conditioned in the injector for 10 min at 250°C to remove any volatile contaminants. Fibers were desorbed in a split/splitless injector at 250°C for 5 min. All samples were analyzed in triplicate. For the analysis of volatile compounds in olive brines, the fiber used was a CAR/PDMS-75 µm (needle length 1 cm, needle size 24 ga) (Supelco, Spain). The headspace SPME sampling conditions used were as follows: 10 ml of brine and 0.5 g NaCl were transferred into a 20 ml glass vial and spiked with 20 μ L of a 4-methyl-2-pentanol (final concentration 50 μ g/ml). The vial was tightly capped with a PTFE-faced silicone septum. The sample was equilibrated for 15 min at 35°C, and then the fiber was exposed to the headspace for 30 min under the same conditions. Desorption of volatiles took place in the injector of the GC/MS for 5 min. Before the first daily analysis, the fiber was conditioned in the injector for 10 min at 250°C to remove any volatile contaminants. All samples were analyzed in triplicate.

GC/MS ANALYSIS

HS-SPME analyses were performed using an AGILENT 6890N gas chromatograph coupled to an AGILENT 5973 mass spectrometer (Agilent, USA). Helium was used as a carrier gas at a constant flow rate of 1 ml/min. The injection port was equipped with an SPME liner $(0.75 \times 6.35 \times 78.5 \text{ mm})$. Separation of compounds was performed on a DB-WAX column (60 m, 0.25 mm i.d., 0.25 mm film thickness, Agilent). The injections were performed in splitless mode. Oven temperature was maintained at 40°C for 5 min, programmed at 3°C/min to 150°C for 20 min. The mass spectrometer was operated in electron impact mode with the electron energy set to 70 eV and a scan range of 30-350 m/z. The temperature of MS source and quadrupole were set to 230 and 150°C. Analyses were performed in full-scan mode. Compounds were identified by comparing the retention times of the chromatographic peaks with those of authentic standards analyzed under the same conditions and by comparison of the retention indices (as Kovats indices) with literature data. MS fragmentation patterns were compared with those of pure compounds, and a mass spectrum database search was performed using the National Institute of Standards and Technology (NIST) MS 98 spectral database. Semi-quantitative determination was carried out by the internal standard method (IS, 4-methyl-2-pentanol). The volatile compounds were quantified by comparison of peak areas to those of internal standards.

STATISTICAL ANALYSIS

Chemical data were subjected to One-Way factor analysis of variance (ANOVA). Significant differences were separated using the Duncan test. The level of significance was set at P < 0.05. The comparison of volatile classes of compounds during fermentation was achieved by principal component analysis (PCA). All statistical analyses were carried out using the STATISTICA 7.0 software (StatSoft software package, Tulsa, OK, USA).

RESULTS

POPULATION DYNAMICS OF THE MICROBIOTA ASSOCIATED TO OLIVES FERMENTATIONS

The microbial count values obtained by analyzing the yeast associated with the fresh fruit surface of the two cultivars (Cellina di Nardò and Leccino) was similar, i.e., 6.4×10^4 CFU/g. At the same sampling time, LAB and *Enterobacteriaceae* in Cellina di Nardò were 5.4×10^2 CFU/g and 6.4×10^3 CFU/g (data not shown), respectively. In Leccino table olives, LAB were present at a concentration of 3.3×10^2 CFU/g, while *Enterobacteriaceae* were not detectable. Olives were then subjected to fermentation and brine samples were collected at the specific time intervals. *Enterobacteriaceae* were detected only in olives and brines derived from the Cellina di Nardò cultivar until the 5th day of fermentation and then they resulted undetectable (data not shown). Yeast population showed a decrease during the first 15 days of fermentation and then increased from 10^3 CFU/ml to 10^4-10^5 CFU/ml in both Cellina di Nardò and Leccino. However, yeasts in Cellina di Nardò brines showed a continuous increase in total count until the 90th day of fermentation and then a slow decrease until the 150th day and a new limited increase at the 180th day. By contrast, in Leccino brines, yeast populations showed a rapid increase at the 150th day (6.7×10^4 CFU/ml). The number of viable cells dropped rapidly at the 180th day (9.3×10^3 CFU/ml) of fermentation (**Figure 1A**).

The dominant species during fermentation time were *Debaryomyces hansenii*, *Wickerhamomyces anomalus* and *Pichia membranifaciens* for Cellina di Nardò, whereas *Saccharomyces cerevisiae* and *P. membranifaciens* for Leccino olives (**Table 1**). In particular, the following yeast species were identified in Cellina di Nardò olives: *D. hansenii* (18%), *W. anomalus* (40%), *P. membranifaciens* (28%), *Debaryomyces carsonii* (5%), and *Candida tartarivorans* (4%). From Leccino fermentations, yeast isolates identified belonged to *S. cerevisiae* (51%), *P. membranifaciens* (20%), *Debaryomyces etchellsii* (3%), *Candida boidinii* (3%), and





Table 1 | Yeast isolates identified during spontaneous fermentation of Cellina di Nardò and Leccino olives.

	Time (day)	Species	Number of isolates	Percentage (%)	Sum of isolates		Time (day)	Species	Number of isolates	Percentage (%)	Sum of isolates
Cellina di Nardò	0	Debaryomyces carsonii	2	10	20	Leccino	0	Candida boidinii	4	20	20
		Debaryomyces hansenii	14	70				Candida sp.	4	20	
		Debaryomyces sp.	2	10				Saccharomyces cerevisiae	12	60	
		Wickerhamomyces anomalus	2	10			15	, Candida boidinii	5	13	40
	15	Debaryomyces carsonii	4	10	40			Saccharomyces cerevisiae	26	65	
		Debaryomyces hansenii	27	68				Zygosaccharomyces mrakii	6	15	
		Debaryomyces sp.	9	22				Zygosaccharomyces sp.	3	7	
	30	Debarvomvces hansenii	21	53	40		30	Candida sp.	2	5	40
		Debarvomvces sp.	3	7				Saccharomvces cerevisiae	29	73	
		Pichia sp.	1	2				Zvgosaccharomyces mrakii	7	17	
		Wickerhamomyces anomalus	15	- 38				Zvgosaccharomyces sp	2	5	
	45	Debarvomvces carsonii	3	7	40		45	Candida boidinii	2	5	40
	40	Debaryomyces bansenii	19	48	40		-10	Saccharomyces cerevisiae	28	70	40
		Pichia sn	2	5				Zvgosaccharomyces mrakii	6	15	
		Wickerhamomyces anomalus	16	40				Zygosaccharomyces miaki	4	10	
	60	Dobaryomycos bansonii	10	28	40		60	Pichia mombranifacions	4	7	40
	00	Debai yonnyces nansenn Dobaryomyces sp	1	10	40		00	Saccharomycos corovisiao	26	65	40
		Debai yonnyces sp.	5	10				Zvananohromvana mrakii	11	20	
			20	12			75	Dishia mambranifasiana	2	20	40
	75	Debaryamyana paraanii	20	20	40		75		2	5	40
	/5	Debaryomyces carsonii	8	20	40			Saccharomyces cerevisiae	2/	10	
			4	10				Zygosaccharomyces mrakii	5	13	
	00	Vvickernamomyces anomaius	28	70	10		00	Zygosaccharomyces sp.	0	15	40
	90	Debaryomyces carsonii	4	10	40		90		3	7	40
			11	27				Saccharomyces cerevisiae	28	70	
	105	Wickerhamomyces anomalus	25	63				Zygosaccharomyces mrakii	8	20	
	105	Pichia mempranitaciens	10	25	40			Zygosaccharomyces sp.	-	3	
		Pichia sp.	3	8			105	Pichia membranitaciens	5	13	40
		Wickerhamomyces anomalus	27	67				Saccharomyces cerevisiae	24	60	
	120	Pichia membranifaciens	11	27	40			Zygosaccharomyces mrakıı	11	27	
		Wickerhamomyces anomalus	29	/3			120	Pichia membranifaciens	25	63	40
	135	Pichia membranifaciens	16	40	40			Saccharomyces cerevisiae	15	37	
		Wickerhamomyces anomalus	24	60			135	Pichia membranifaciens	18	45	40
	150	Candida tartarivorans	8	20	40			Pichia sp.	6	15	
		Debaryomyces carsonii	4	10				Saccharomyces cerevisiae	16	40	
		Pichia membranifaciens	22	55			150	<i>Candida</i> sp.	7	17,5	40
		Wickerhamomyces anomalus	6	15				Debaryomyces ethcellsii	2	5	
	175	Candida tartarivorans	6	15	40			Debaryomyces sp.	5	12,5	
		Pichia membranifaciens	31	77				Pichia membranifaciens	16	40	
		Wickerhamomyces anomalus	3	8				Saccharomyces cerevisiae	10	25	
	180	Candida tartarivorans	6	15	40		175	Candida boidinii	3	8	40
		Pichia membranifaciens	32	80				Candida sp.	7	17,5	
		Wickerhamomyces anomalus	2	5				Debaryomyces ethcellsii	7	17,5	
								Debaryomyces sp.	8	20	
								Pichia membranifaciens	9	22	
								Saccharomyces cerevisiae	6	15	
							180	Candida boidinii	1	2,5	40
								Debaryomyces ethcellsii	5	12,5	
								Debaryomyces sp.	8	20	
								Pichia membranifaciens	18	45	
								Saccharomyces cerevisiae	8	20	

Zygosaccharomyces mrakii (11%). When the identity percentage of the obtained sequences resulted <99%, the isolates were identified only at genus level (*Debaryomyces* sp. and *Pichia* sp. obtained from Cellina di Nardò and of *Candida* sp., *Debaryomyces* sp., *Pichia* sp. and *Zygosaccharomyces* sp. from Leccino). Additional molecular tests (i.e., by RFLP, 5.8 ITS or sequencing D1–D2 domains 26S) will be further performed for their unequivocal identification at species level.

Concerning LAB detection, the results obtained are reported in **Figure 1B**. In Cellina di Nardò after the starting time, LAB were no more detectable until the 120th day of fermentation $(3.2 \times 10^4 \text{ CFU/ml})$ and then they increased $(2 \times 10^6 \text{ CFU/ml})$ until the 180th day of fermentation. In Leccino cultivar, LAB were detectable $(3.9 \times 10^3 \text{ CFU/ml})$ at the 135th day of fermentation and then they increased up to $1.4 \times 10^5 \text{ CFU/ml}$ at the 180th day of fermentation. Bacterial isolates obtained from Cellina di Nardò fermentation belonged to species *L. plantarum* (68%), *Kocuria* spp. (5%), and *Swaminathania salitolerans* (13%). The bacterial isolates identified at species level (sequence identity percentage > 98%) from Leccino fermentation belonged to the *L. plantarum* (89%). Identification at species level of *Lactobacillus* sp. isolates obtained from Cellina di Nardò and Leccino, is now forthcoming by using multiplex PCR of recA gene (Torriani et al., 2001). Moreover, although LAB were not detectable until the 120th and 135th day of fermentation for Cellina di Nardò and Leccino, respectively, the dominant species was *L. plantarum* for both the table olive cultivars (**Table 2**).

YEAST STRAINS TECHNOLOGICAL CHARACTERIZATION

A dedicated procedure was assessed for the isolation and technological characterization of yeasts and bacteria from fermented table olives. As a first step we set up the formulation of two media, for the evaluation of yeasts and LAB technological properties. Both media reproduced the severe physico-chemical constraints usually found into the fermenting brines. Isolated microorganisms were assayed for their ability to grow in a synthetic model brine (MB), that varied in their composition throughout the fermentations, to be used as a reliable and reproducible substrate for the in vitro technological assays. To achieve the formulation of the model brines, brine samples deriving from several fermentations of different black olive cultivars were analyzed by reversed-phase HPLC-DAD to determine their chemical profile, in particular, that corresponding to the phenolic compounds (Bleve et al., 2015). Five hundred yeast isolates for each of the two olive cultivars were applied on MB and incubated for 15 days at 15°C. This temperature represented the average value observed during late autumn and winter period when Leccino and Cellina di Nardò olives were generally fermented. Out of 500 initial yeast isolates, 278 isolates from Cellina di Nardò and 101 isolates from Leccino fermentations were able to survive the above constraints. In the second step of the proposed procedure, the microbial strains were tested by plate assays in order to evaluate (i) the presence of betaglucosidase activity, required to degrade oleuropein, and (ii) the inability to produce biogenic amines. Afterward, 77 yeast isolates from Cellina di Nardò and 53 from Leccino demonstrated that they were able to satisfy the above parameters. These isolates were further evaluated for the presence of protease and lipase activities (not shown). For beta-glucosidase activity it was assigned a score 3 (intense brown), 2 (light brown), 1 (yellow-milky), 0 (white), whereas, for amino acids decarboxylation activities, the value 3 was considered for isolates surrounded by an intense purple halo, 2 for isolates which produced intense blue halo, 1 for isolates producing slight blue halo and the value 0 was assigned for isolates that remained white. The yeast population characterized by intense beta-glucosidase activity (score 3) and absence of amino acids decarboxylation activities (score 0) consisted of 77 Cellina di Nardò isolates and 53 Leccino isolates. These isolates were then evaluated for the presence of protease and lipase activities: 17 isolates from Cellina di Nardò and 11 isolates from Leccino showed protease activity on plate against casein, whereas none of the isolates deriving from the second selection step demonstrated lipase activity. The third step in the proposed protocol consisted of the identification at species level of the yeast isolates that did not showed protease activity by PCR analysis of their rRNA genes.

At the end of this procedure, from Cellina di Nardò one yeast isolate was selected for the species *P. anomala*, identified by the Accession number LK392318, one isolate of *D. hansenii* (Accession number LK322319), one for the species *P. membranifaciens* (Accession number LK322320), whereas, it was selected from Leccino one isolate of *S. cerevisiae* (Accession

	Time	Species	Number of isolates	Percentage (%)	Sum of isolates
Cellina di Nardò	0	Kocuria	12	100	12
	120	Swaminathania salitolerans	26	100	26
	135	Lactobacillus plantarum	35	77	52
		Lactobacillus sp.	12	13	
		Swaminathania salitolerans	5	10	
	150	Lactobacillus plantarum	40	88	50
		Lactobacillus sp.	10	12	
	175	Lactobacillus plantarum	50	100	50
	180	Lactobacillus plantarum	38	94	50
		Lactobacillus sp.	12	6	
Leccino	0	Lactobacillus plantarum	12	100	12
	135	Lactobacillus plantarum	40	77	52
		Lactobacillus sp.	12	23	
	150	Lactobacillus plantarum	39	75	52
		Lactobacillus sp.	13	25	
	175	Lactobacillus plantarum	52	100	52
	180	Lactobacillus plantarum	52	100	52
	150 175 180	Lactobacilius sp. Lactobacillus plantarum Lactobacillus sp. Lactobacillus plantarum Lactobacillus plantarum	12 39 13 52 52	23 75 25 100 100	5 5 5

Table 2 | Bacteria isolates identified during spontaneous fermentation of Cellina di Nardò and Leccino olives.

number LK392321) and one of *P. membranifaciens* (Accession number LK392322).

LAB TECHNOLOGICAL CHARACTERIZATION

As decribed by Bleve et al. (2015) the selection of bacterial isolates was carried out on the MB *ad hoc* formulated. After this step, the bacterial population resistant to the above constraint was isolated, consisting of 110 (out of 240) isolates from Cellina di Nardò and 90 (out of 220) isolates from Leccino. The presence of beta-glucosidase and amino acids decarboxylation activities were tested as described above. After this second selection step, the LAB population which satisfied the above parameters (intense beta-glucosidase activity, score 3, and absence of amino acids decarboxylation activities, score 0) consisted of 18 Cellina di Nardò isolates and 14 Leccino isolates. These isolates were then evaluated for the presence of protease and lipase activities. None of the isolates from Cellina di Nardò isolates and Leccino showed protease and lipase activities.

At the end of this procedure, two LAB isolates both belonging the species *L. plantarum* were selected from Cellina di Nardò and Leccino and identified by the Accession number LK392323 and LK392324, respectively.

CHEMICAL DYNAMICS OF BRINES

Brine temperatures increased from 8 to 17°C during the first 3 months of fermentation and then it reached 30°C at the 180th day (Figure 2A). The pH values quickly decreased (4.4-4.7) within the first 5-10 days of fermentation and, after the 120th, they declined at its minimum values (Figure 2B). Salinity values were checked throughout the fermentation to maintain it almost stable about a value of 10% (w/v) (data not shown). Glucose concentrations increased until the 30th and the 60th of fermentation respectively, in Cellina di Nardò and Leccino samples and then they decreased until the end of the process (Figure 3A). The fructose content should apparently be a constant increase during the fermentation of Leccino table olives, whereas in Cellina di Nardò brines, after a substantial increase during the first 90 days, fructose levels decreased to a final concentration of about 0.7 g/L (Figures 3A,C). The level of total organic acids in Leccino brines was found to be higher than in Cellina di Nardò and these compounds increased during fermentation (Figures 3B,D). Ethanol concentration increased gradually with time and reached a final concentration of about 0.5 g/L (Figures 3A,C). Figure 4 shows the concentration dynamics of the following simple phenolic compounds in the brine: hydroxytyrosol, oleuropein, and tyrosol. In general, Leccino brine samples showed a greater amount of phenolic compounds than Cellina di Nardò brines. In all samples, the main phenolic compound detected during fermentation was hydroxytyrosol, it being about 200 and 800 µg/ml, respectively in Cellina di Nardò and Leccino brines. Oleuropein was undetectable in Cellina di Nardò brines soon after the 30th day of fermentation (Figure 4A) whereas, in Leccino brines, it was observed until 90 days of fermentation (Figure 4B). Tyrosol concentration respectively, increased in the first 60 and 90 days of fermentation in Cellina di Nardò and Leccino brines and then it remained constant until the end of process (Figures 4A,B).



ANALYSIS OF VOLATILE COMPOUNDS IN OLIVES AND BRINES

The analytical SPME/GC-MS method used in this work allowed the correct identification and quantification of 54 and 44 volatile compounds in Cellina di Nardò and Leccino olives and brines, respectively. All the volatile compounds identified were grouped into aldehydes/ketones, esters, alcohols, hydrocarbons, volatile phenols, fatty acids, monoterpenes, and lactones classes. Aldehydes and ketones content was significantly different in the two olive cultivars, ranging from $88.61 \pm 7.25 \,\mu$ g/kg in Cellina di Nardò olives and 104.25 \pm 6.01 µg/kg in Leccino olives at 30 days. After 180 days of fermentation they resulted 30.26 \pm 4.21 and 61.93 \pm 4.09 μ g/kg for Cellina di Nardò and Leccino at, respectively (Table 3). With the exception of benzaldehyde, all identified aldehydes showed a decrease in concentration during fermentation. Alcohol and ester contents increased during fermentation, with higher concentrations in Cellina di Nardò than in Leccino (Table 3). In fact, the levels of ethyl acetate, isoamyl acetate, ethyl hexanoate (green apple aroma), and ethyl octanoate (sweet aroma) increased during fermentation in both table olive cultivars (Table 3). Short chain fatty acids (acetic, propanoic, 2-methylpropanoic) were detected in both fermentations, with acetic acid being the major representative of this group. Acetic acid was present throughout the fermentation period and in both cultivars and is representative of yeast and



LAB metabolism following oxidation of ethanol (Table 3). Levels of acetic acid increased progressively from 7.73 \pm 0.04 to 59.32 \pm 1.30 μ g/kg in Cellina di Nardò olives, and from 5.22 \pm 0.30 to $25.06 \pm 0.93 \,\mu$ g/kg in Leccino olives. In both fermentations, total amounts of hydrocarbons (octane, toluene, styrene, trimethyl benzene, minor volatile compounds) increased from the initial to the middle stages of fermentation and then a slight decrease during the final period (Table 3). Terpene concentrations also increased during fermentation in both table olive cultivars. There were five different terpenes, of which 3 were monoterpenes (linalol, linalolox, 3,7 dimethyl- 2,6 octadienal) and two sesquiterpenes (cymene and α -pinene). The highest amounts of all these terpenes were detected in Cellina di Nardò olives (Table 3). The principal free volatile classes showed different distributions in olives from brines (Table 4). In fact, in Cellina di Nardò brines collected after 90 and 180 days of fermentation, a greater concentration of esters, alcohols and short chain fatty acids was observed than in the corresponding drupes. The most representative ester species was ethyl acetate, followed by methyl acetate. Propionic esters were detected in the last fermentation stage only in Cellina di Nardò brine. Alcohols detected in higher concentrations in Cellina di Nardò brines were 2+3-methyl-1-butanol, followed by

1-propanol and cis 3 hexen-ol (Z), whereas in Leccino brines there was a higher concentration of hexan-1-ol, followed by 2+3methyl-1-butanol and heptanol. Analogously to what observed in drupes, in Cellina di Nardò brines there was a higher amount of fatty acids than in Leccino brines.

PRINCIPAL COMPONENT ANALYSIS

In order to correlate chemical data with microbial population dynamics and to identify particular compounds to be suitable for monitoring olive fermentation, a PCA analysis was performed on the complete SPME/GC-MS data matrix of each olive sample. The analysis was carried out by plotting the mean values of each volatile compound (variables) at 3 different fermentation times (30–90–180 days of brining). Two bi-plots displaying PC1 vs. PC2 are illustrated in **Figures 5**, **6**, which show the projection of the variables on the plane defined by the first and second principal components. In Cellina di Nardò and Leccino, the first principal component (PC1) explained 81.12 and 75.52%, respectively, of the total variability, between volatile compounds produced during fermentation and the second principal component (PC2) accounted for an additional 18.88% (Cellina di Nardò) and 24.48% (Leccino). The two planes made using the



first two PCs (**Figures 5, 6**) indicated that the molecules were divided into three groups. In both table olive cultivars, one group, consisting of aldehydes, resulted closely related to the first stage of fermentation (T30); the second one consisted of higher alcohols (2-methyl-1-propanol; 3-methyl-1-butanol), styrene, and o-cymene associated with the middle stage of fermentation (T90) and the third one contained acetate esters linked to the final step in olive fermentation (T180).

DISCUSSION

Olive fermentation is a complex process, where the enzymes from olives interact with the metabolic activities of microorganisms through various biochemical pathways (McFeeters, 2004). The volatile compounds are produced in the brine by fermentation processes operated by microorganisms and in the fruit matrix by the action of either endogenous and microbial enzymes (produced by yeasts, lactic acid bacteria, etc.) (Sabatini et al., 2009). Volatile compounds in table olives belong to very heterogeneous group including higher alcohols, aldehydes, ketones, esters, fatty acids, monoterpenes, and hydrocarbons (Malheiro et al., 2011). However, limited data are present in literature on the compounds responsible for the organoleptic properties of table olives.

In the present paper, we have produced the molecular and technological characterization of LAB and yeast populations in Cellina di Nardò and Leccino spontaneous fermentations, as a first step for the production of autochthonous fermentation starters. Moreover, the evolution of the main physic- chemical parameters and volatile compounds during fermentations were studied in order to determine chemical descriptors, correlated with microbiological activities, that may be suitable for monitoring the outcome of the fermentation.

The microbiota of processed olives or brines include members of *Enterobacteriaceae*, *Clostridium*, *Pseudomonas*, *Staphylococcus*,

LAB, yeasts, and occasionally molds (Heperkan, 2013). In agreement with data previously reported by Alves et al. (2012), our results indicated that, during the fermentations of both Cellina di Nardò and Leccino olives, some Enterobacteriaceae were initially present but after few days they resulted undetectable and were not detected at the end of the process. No Clostridium and Pseudomonas were revealed, probably because they were unable to survive till the end of the process due to the low pH values (<4.3, the maximum pH value established by the IOOC for preserving olives physicochemical characteristics (Montano et al., 2010). By contrast, yeasts were detectable throughout the fermentation process and finally, in the last month, the hitherto undetectable LAB appeared and carried out lactic fermentation. According to Arroyo-López et al. (2012a) and Bevilacqua et al. (2012), yeasts play a substantial role in the production of both green-treated olives and black naturally-fermented olives. Also, during Leccino and Cellina di Nardò olive fermentations, they could exert a fundamental role. Yeasts increased their concentration throughout both fermentation processes from 3.0 log CFU/ml (at the beginning) to 4.0-5.0 log CFU/ml (at the end). Since several studies have demonstrated that the action of indigenous yeasts during olive production process is able to enhance the quality of the final product (Aponte et al., 2012; Bevilacqua et al., 2012, 2013), a pre-selection and technological characterization of yeast populations associated with Cellina di Nardò and Leccino olives was carried out. Yeasts and LAB associated to Cellina di Nardò and Leccino fermented table olives were isolated and characterized. In a first step, isolated microorganisms were assayed in a "model brine," formulated ad hoc for yeasts to replicate the industrial conditions of brine fermentation. To this purpose, starting from previous indications by Servili et al. (2006), an improved synthetic mixture of glucose, sodium chloride, and the mono-cyclic and poly-cyclic aromatic acids was used to grow yeast and LAB in low pH and temperature conditions. Then, the microbial isolates able to grow in model brine, were assayed for their the capacity to secrete beta-glucosidase (thus being able to rapidly eliminate oleuropein) and the inability to produce biogenic amines. Indeed beta-glucosidase contribute to olives debittering by degrading polyphenols such as oleuropein, and it was considered an interesting positive technological trait (Bevilacqua et al., 2012, 2013). Isolates that showed the above described abilities were evaluated also for the presence of lipase activity, that could improve the aromatic profile of fermented olives by increasing their free fatty acid content (Rodríguez-Gómez et al., 2012), and the absence of proteolytic activity, which could have a negative impact on olive quality because it is related to olive softening (Arroyo-López et al., 2012a). The third step in the proposed procedure consisted of the identification of the selected strains by sequencing a significant portion of their rRNA genes. In agreement with previous studies in Cellina di Nardò fermentations we identified D. hansenii (Nisiotou et al., 2010), W. anomalus (Nisiotou et al., 2010), P. membranifaciens (Coton et al., 2006; Bautista-Gallego et al., 2011) D. carsonii and C. tartarivorans, whereas in Leccino fermentations were detected S. cerevisiae, P. membranifaciens, D. etchellsii, C. boidinii and Z. mrakii (Arroyo López et al., 2006; Nisiotou et al., 2010).

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Cellina di Nardò table olives

Compounds

Image: SD (a) (b) (b) (c) (c) (c) (c) (c) (c) (c) (c) (c) (c		30 days	90 days	180 days	30 days	90 days	180 days	
MCHTORS No No </th <th></th> <th>mean ± SD μg/kg*</th> <th>mean ± SD µg/kg*</th> <th></th>		mean ± SD μg/kg*	mean ± SD µg/kg*					
Terriny functional 1516 ± 4.06 ND ND Field ± 100 ND ND Condectarmed Terriny functional 179 ± 2.00 370 ± 1.00 773 ± 1.01 236 ± 4.01 170 ± 2.02 170 ± 2.02 170 ± 2.02 170 ± 2.02 170 ± 0.02	ALDEHYDES-KETONS							
Terrehydnenel 179 ± 10 ± 10 ± 377 ± 10 ± 10 ± 10 ± 10 ± 10 ± 10 ± 10 ±	2-methylpropanal	$13.16 a \pm 0.90$	ΟN	QN	$16.54 b \pm 0.80$	QN	ŊŊ	cooked-caramel
Smethylamati 130 ± 0.55 85.5 ± 1.02 715 ± 1.05 715 ± 1.05 715 ± 1.05 710 710 700<	2-methylbutanal	$17.19 c \pm 0.20$	$9.37 b \pm 0.60$	7.12 a ± 0.14	$23.66 d \pm 0.91$	$14.13 b \pm 0.23$	$13.70 b \pm 0.22$	malty
Hondi 0519-2/20 570±140 ND ND ND ND ND ND ND ND ND Hondi 14.00-12.20 57.0±4.07 ND 0.000±100 150.0±100 ND	3-methylbutanal	$14.80 b \pm 0.55$	$8.53 a \pm 1.02$	7.51 a ± 0.75	$31.50 c \pm 2.20$	$20.41 \text{ b} \pm 2.11$	$17.65 b \pm 0.45$	malty
3 differ 3 differ 1 diff 1 diff <th1 diff<="" th=""> <th1 diff<="" th=""> <th1 diff<="" td=""><td>Hexanal</td><td>$10.51 b \pm 2.02$</td><td>5.70a ± 1.40</td><td>QN</td><td>ND</td><td>ND</td><td>ND</td><td>green-sweet-green apple</td></th1></th1></th1>	Hexanal	$10.51 b \pm 2.02$	5.70a ± 1.40	QN	ND	ND	ND	green-sweet-green apple
Heptend 150 b ± 005 315 a ± 021 ND 440 b ± 025 ND	2,6-dimethyl-4-heptanon	14.40 b ± 2.20	$7.33 \text{ a} \pm 0.77$	QN	$10.30ab \pm 0.05$	12.51b ± 1.23	8.05a ± 1.80	oily-fatty-woody
Nomed 11801±120 16321±08 91±312 664±05 977±122 0.23±0.20 Bithy-servpungent treat-deniny (bit management) NO 3(3±4.0) </td <td>Heptanal</td> <td>$4.50 b \pm 0.05$</td> <td>$3.15 a \pm 0.21$</td> <td>QN</td> <td>$4.49b \pm 0.20$</td> <td>ND</td> <td>ND</td> <td>tallowy-pungent</td>	Heptanal	$4.50 b \pm 0.05$	$3.15 a \pm 0.21$	QN	$4.49b \pm 0.20$	ND	ND	tallowy-pungent
Bornaldehyde 2.00 ab 10.03 2.00 b 10.04 3.35 c + 0.02 1.82 a + 0.04 3.35 c + 0.03 and trans 2.00 mol 8.00 ± 0.30 1.72 ± 0.11 8.00 ± 0.30 ND 8.56 ± 1.23 pinylikyvatt trans 2.00 mol 8.30 ± 1.25 5.31 ± 5.55 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.31 ± 5.50 5.52 ± 5.51 5.51 ± 5.50 5.52 ± 5.51 5.51 ± 5.50 5.51 ± 5.50 5.52 ± 5.51 5.51 ± 5.50 5.51 ± 5.50 5.51 ± 5.50 5.51 ± 5.50 5.51 ± 5.50 5.51 ± 5.50 </td <td>Nonanal</td> <td>11.90 b ± 1.30</td> <td>$16.92 \text{ b} \pm 0.85$</td> <td>$9.1 a \pm 3.12$</td> <td>$6.64 a \pm 0.85$</td> <td>9.77 a ± 1.22</td> <td>$10.43 a \pm 0.20$</td> <td>fatty-waxy-pungent</td>	Nonanal	11.90 b ± 1.30	$16.92 \text{ b} \pm 0.85$	$9.1 a \pm 3.12$	$6.64 a \pm 0.85$	9.77 a ± 1.22	$10.43 a \pm 0.20$	fatty-waxy-pungent
Instruction ND 312a±01 800±080 ND 855±082 painty-faity-faity Instancencis 8801±725 53.87±525 32.6±121 104.25±601 864±239 1033±409 Instancencis 8801±726 53.87±525 32.6±1421 104.25±601 864±239 1133±409 Instancencis 9.00±052 1950±1022 10220d±480 ND ND 1412b±152 Ethereid-sweet Ethyl acetate ND ND ND ND ND ND ND Stortyl acetate ND ND ND ND ND ND ND ND Stortyl acetate ND ND <td>Benzaldehyde</td> <td>$2.20 \text{ ab} \pm 0.03$</td> <td>$2.90 \text{ bc} \pm 0.40$</td> <td>$3.43 c \pm 0.20$</td> <td>$2.32 \text{ ab} \pm 0.20$</td> <td>1.82 a ± 0.04</td> <td>$3.35 c \pm 0.60$</td> <td>almond</td>	Benzaldehyde	$2.20 \text{ ab} \pm 0.03$	$2.90 \text{ bc} \pm 0.40$	$3.43 c \pm 0.20$	$2.32 \text{ ab} \pm 0.20$	1.82 a ± 0.04	$3.35 c \pm 0.60$	almond
Instant 861±325 537±525 5.02±4.21 (4.05±6.01 56.64±2.08 61.93±4.06 Heat No 1 1 1 1 1 1 1 Heat No 1 3.03±4.05 1 5.03±4.480 NO 1 1.12.1±1.5 Ether Hyl active 1 <td>trans 2-decenal</td> <td>QN</td> <td>ND</td> <td>$3.12a \pm 0.1$</td> <td>$8.80b \pm 0.80$</td> <td>ND</td> <td>$8.55b \pm 0.82$</td> <td>painty-fishy-fatty</td>	trans 2-decenal	QN	ND	$3.12a \pm 0.1$	$8.80b \pm 0.80$	ND	$8.55b \pm 0.82$	painty-fishy-fatty
Herest No No <th< td=""><td>Total amounts</td><td>88.61 ± 7.25</td><td>53.87 ± 5.25</td><td>30.26 ± 4.21</td><td>104.25 ± 6.01</td><td>58.64 ± 2.83</td><td>61.93 ± 4.09</td><td></td></th<>	Total amounts	88.61 ± 7.25	53.87 ± 5.25	30.26 ± 4.21	104.25 ± 6.01	58.64 ± 2.83	61.93 ± 4.09	
Methyl accate ND 4,56 ± 0.61 15,30 ± 0.62 15,30 ± 1.62 ND N1 11,12 ± 1.12 Etheralisment Ethyl accate 9,00 ± 0.02 31,90 ± 1.062 13,90 ± 1.02 13,90 ± 2.01 554 ± 2.00 100 ND Ethyl accate ND ND 771 ± 1.02 13,90 ± 2.05 100 ND Terthyl accate ND ND 771 ± 1.02 13,90 ± 0.50 100 ND Terthyl accate ND ND 195 ± 0.43 ND ND ND ND propolyl accate ND ND 1156 ± 0.90 350 ± 0.02 500 ± 0.50 storky-soncet Ethyl propanoate ND ND <td>Esters</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>	Esters							
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rpropyl acetate ND ND 195 ± 0.43 ND ND ND soburyl acetate ND ND ND ND ND ND soburyl acetate ND ND ND 1280 ± 0.63 ND ND soburyl acetate ND ND 510 ± 0.50 sweetfurity Propyl porpanoate ND ND ND ND ND ND Fthyl 2-methyl butanoate ND ND ND ND ND ND ND Sthyl 3-methyl butanoate ND ND 1156 ± 0.90 3.80 ± 0.04 3.90 ± 0.32 5.52 ± 0.24 fruity Sthyl 3-methyl butanoate ND ND 1156 ± 0.40 3.80 ± 0.04 5.60 ± 0.51 fruity Sthyl 3-methyl butanoate ND ND As2 ± 0.24 fruity fruity Sthyl 3-methyl butanoate ND ND As2 ± 0.24 fruity fruity Sthyl 3-methyl butanoate ND ND As2 ± 0.24 fruity fruity	2 methyl ethyl propanoate	ND	DN	$7.71 \text{ a} \pm 0.50$	ND	ΠN	ND	
Isoburyl acetate ND ND ND ND ND ND Furthyl butanoate ND	n-propyl acetate	ND	DN	19.5 a ± 0.43	ND	ND	ND	
Ethyl butanoateNDNDNDS01 b± 0.50sweetfuilyFropyl propanoateNDNDNDNDNDNDNDNDNDFropyl propanoateNDNDNDNDNDNDNDNDNDNDEthyl 2-methyl butanoateNDNDND1150 ± 0.00 3.80 ± 0.03 3.00 ± 0.32 5.22 ± 0.24 fruilyEthyl 2-methyl butanoateNDND 1150 ± 0.00 3.80 ± 0.03 3.00 ± 0.32 5.50 ± 0.50 fruilySomyl aestiteNDND 9.00 ± 0.43 ND 2.36 ± 0.11 16.72 ± 5.24 fruilySamthyl 1 butanol propanoateNDND 3.90 ± 0.50 3.80 ± 0.50 3.90 ± 0.50 sweetfullySamthyl 1 butanol propanoateNDND 2.36 ± 0.11 16.72 ± 5.24 fruilySamthyl 1 butanol propanoateNDND 2.36 ± 0.04 3.50 ± 0.50 weetfullySamthyl 1 butanol propanoateNDND 2.36 ± 0.11 16.72 ± 5.24 fruilySamthyl 1 butanol propanoateNDND 2.36 ± 0.13 NDNDSamthyl 1 butanol propanoateNDND 2.36 ± 0.23 NDNDSamthyl 1 butanol propanoateNDND 1.86 ± 0.24 5.00 ± 0.50 weetfullylingSamthyl 1 butanol propanoateNDND 1.86 ± 0.24 3.70 ± 0.50 NDNDSamthyl 1 butanol propanoateNDND 1.88 ± 0.24 7.70 ± 0.24 8.76 ± 0.50	Isobutyl acetate	ND	QN	12.80 a ± 0.80	ND	ΩN	ND	
Propy propandate ND ND ND ND ND ND ND Ethyl 2-methyl butanoate ND ND 1150 \pm 0.90 380 \pm 0.04 390 \pm 0.32 5.22 \pm 0.24 fuity Ethyl 3-methyl butanoate ND ND 1150 \pm 0.90 380 \pm 0.04 5.00 \pm 0.31 fuity Ethyl 3-methyl butanoate ND 940 \pm 0.91 54.20 \pm1.220 ND 5.26 \pm0.24 fuity Isoamyl acetate ND 900 c \pm 0.43 ND 2.36 a \pm0.11 16.72 c \pm 5.32 fuity Sommyl acetate ND ND 2.36 a \pm 0.70 8.00 c \pm 0.43 ND ND 2-methyl 3-butenoic acid ethyl ND ND 2.36 a \pm 0.10 1.52 c \pm 5.32 s.20 \pm 0.50 sweetfuily 2-methyl 3-butenoic acid ethyl ND ND ND ND ND ND 2-methyl 3-butenoic acid ethyl ND 1.01 2.36 a \pm 0.10 1.52 c \pm 5.05 sweetfuily-fuily Ethyl 3-butenoic acid ethyl ND ND ND ND ND ND <td>Ethyl butanoate</td> <td>ND</td> <td>DN</td> <td>$5.12 a \pm 0.83$</td> <td>ND</td> <td>ΠN</td> <td>$5.01 b \pm 0.50$</td> <td>sweet-fruity</td>	Ethyl butanoate	ND	DN	$5.12 a \pm 0.83$	ND	ΠN	$5.01 b \pm 0.50$	sweet-fruity
Ethyl JutanoateNDND1150 ± 0.90 3.80 ± 0.04 3.90 ± 0.32 5.22 ± 0.24 futivEthyl JutanoateNDND $10.96 \varepsilon \pm 0.93$ 4.20 ± 4.04 4.52 ± 0.04 5.02 ± 6.031 futivStamtyl JutanoateND $9.40 b \pm 0.91$ $54.20 d \pm 2.20$ ND 2.36 ± 4.011 $16.72 \varepsilon \pm 5.32$ futivSamtyl Jutanol propanoateND $9.00 \varepsilon \pm 0.43$ ND 4.84 ± 0.24 $5.20 b \pm 0.50$ NDSmethyl Jutanol propanoateND $9.00 \varepsilon \pm 0.43$ ND ND 2.36 ± 4.012 $5.20 b \pm 0.50$ Smethyl Jutanol propanoateND ND 2.80 ± 0.71 $16.72 \varepsilon \pm 5.32$ ND StattNDND 2.80 ± 4.031 NDND ND StattND 1.86 ± 4.032 8.76 ± 4.042 $5.20 b \pm 0.50$ ND StattNDND ND ND ND ND StattND 1.520 ± 0.43 ND ND ND StattND ND ND ND ND StattND ND ND ND ND Statt $1.55 \pm 4.0.4$ $1.55 \pm 4.0.4$ $1.55 \pm 4.0.2$ $4.11b \pm 0.43$ Hox/A canoateND $1.35 \pm 4.0.1$ $1.55 \pm 4.0.4$ ND ND StattND ND ND ND ND StattND ND ND ND ND StattND ND ND ND ND Statt $1.35 \pm 4.0.4$ $3.72 \pm 4.0.$	Propyl propanoate	ND	DN	$11.88 a \pm 0.61$	ND	ND	ND	
Ethyl JanacteNDND10.96 c ± 0.93 4.20 a ± 0.40 5.60 b ± 0.31 furityIsoamyl acetateND9.40 b ± 0.91 54.20 d ± 2.20 ND2.36 a ± 0.11 16.72 c ± 5.32 3 methyl I butanol propanateNDND9.00 c ± 0.43 ND4.84 a ± 0.24 5.20 b ± 0.50 3 methyl I butanol propanateNDND9.00 c ± 0.43 ND4.84 a ± 0.24 5.20 b ± 0.50 3 methyl I butanol propanateNDND2.80 a ± 0.31 NDNDND2.methyl 3 butanoic acid ethylND4.86 a ± 0.73 8.76 a ± 0.70 NDNDEthyl hexanateND1.58 a ± 0.04 7.70 c ± 0.20 NDNDNDEthyl hexanateNDND0.74 a ± 0.05 NDNDND3 hexanateNDNDNDNDNDNDEthyl actateNDNDNDNDNDAthorabacateND1.75 a ± 0.14 1.55 b ± 0.14 NDAthyl actanoateND1.75 a ± 0.14 1.55 b ± 0.14 NDAthyl actanoateNDNDNDNDAthyl actanoateNDNDNDNDAthyl actanoateNDNDNDNDAthyl actanoateNDNDNDNDAthyl actanoateNDNDNDNDAthyl actanoateNDNDNDNDAthyl actanoateNDNDNDNDAthyl actanoate <td< td=""><td>Ethyl 2-methyl butanoate</td><td>ND</td><td>QN</td><td>11.50 ± 0.90</td><td>3.80 ± 0.04</td><td>3.90 ± 0.32</td><td>5.22 ± 0.24</td><td>fruity</td></td<>	Ethyl 2-methyl butanoate	ND	QN	11.50 ± 0.90	3.80 ± 0.04	3.90 ± 0.32	5.22 ± 0.24	fruity
	Ethyl 3-methyl butanoate	ND	ND	$10.96 c \pm 0.93$	$4.20 a \pm 0.40$	$4.52 a \pm 0.04$	$5.60 b \pm 0.31$	fruity
3 methyl 1 butanol propanoate ND ND ND 4.84 a \pm 0.24 5.20 b \pm 0.50 2-methyl 3-butenoic acid ethyl ND ND ND ND ND ND 2-methyl 3-butenoic acid ethyl ND ND 2.80 a \pm 0.31 ND ND ND ND ester ND 4.86 a \pm 0.73 8.76 a b \pm 0.70 ND 6.10 b \pm 1.11 12.35 c \pm 1.09 Ethyl hexanoate ND 1.58 a \pm 0.04 7.70 c \pm 0.20 ND ND ND Hexyl acetate ND ND ND ND ND ND Shexenol acetate ND 1.55 a \pm 0.14 1.57 a \pm 0.43 ND ND Shexenol acetate ND 1.35 a \pm 0.1 1.55 b \pm 0.14 ND ND Ethyl heptanoate ND 1.11 b \pm 0.43 4.60b \pm 0.51 ND Shexenol acetate ND 1.15 b \pm 0.14 1.55 b \pm 0.50 sweet/furity-floral Ethyl heptanoate ND 1.55 b \pm 0.50 ND ND ND Ethyl scicylate </td <td>Isoamyl acetate</td> <td>ND</td> <td>9.40 b ± 0.91</td> <td>$54.20 d \pm 2.20$</td> <td>ND</td> <td>$2.36 a \pm 0.11$</td> <td>$16.72 c \pm 5.32$</td> <td></td>	Isoamyl acetate	ND	9.40 b ± 0.91	$54.20 d \pm 2.20$	ND	$2.36 a \pm 0.11$	$16.72 c \pm 5.32$	
2-methyl 3-butenoic acid ethyl ND ND ND ND ester ND ND ND	3 methyl 1 butanol propanoate	ND	DN	$9.00 c \pm 0.43$	ND	$4.84 a \pm 0.24$	$5.20 b \pm 0.50$	
ester ester Ethyl hexanoate ND $4.86 a \pm 0.73$ $8.76 a b \pm 0.70$ ND $6.10 b \pm 1.11$ $12.35 c \pm 1.09$ Hexyl acetate ND $1.58 a \pm 0.04$ $7.70 c \pm 0.20$ ND $0.73 c \pm 0.50$ sweet-fruity-floral Hexyl acetate ND ND $0.74 a \pm 0.05$ ND $0.74 a \pm 0.65$ ND $0.74 a \pm 0.61$ $3.55 b \pm 0.50$ sweet-fruity-floral Thyl acetate ND ND $0.74 a \pm 0.05$ ND $0.71 a \pm 0.43$ $4.60b \pm 0.51$ 3-hexenol acetate ND ND $1.55 a \pm 0.14$ $1.55 b \pm 0.14$ ND ND Ethyl heptanoate ND $1.75 a \pm 0.24$ $3.72 b \pm 0.50$ ND ND ND Ethyl octanoate ND ND ND ND ND ND Benzyl acetate ND ND ND ND ND ND Benzyl acetate ND ND ND ND ND ND Benzyl	2-methyl 3-butenoic acid ethyl	ND	DN	$2.80 a \pm 0.31$	ND	ΟN	ND	
Ethyl hexanoateND $4.86 a \pm 0.73$ $8.76 ab \pm 0.70$ ND $6.10 b \pm 1.11$ $12.35 c \pm 1.09$ Hexyl acetateND $1.58 a \pm 0.04$ $7.70 c \pm 0.20$ NDND $3.55 b \pm 0.50$ sweet-fruity-floralHexyl acetateNDND $0.74a \pm 0.05$ ND $0.74a \pm 0.05$ ND $0.74a \pm 0.63$ NDShexenol acetateNDND $0.74a \pm 0.05$ ND $0.74a \pm 0.05$ ND ND $3.55 b \pm 0.50$ Shexenol acetateNDND $1.52 a \pm 0.13$ $1.52 a \pm 0.43$ NDND ND Ethyl heptanoateND $1.35 a \pm 0.1$ $1.55 b \pm 0.14$ NDNDNDEthyl octanoateND $1.75 a \pm 0.24$ $3.72 b \pm 0.50$ NDNDNDBenzyl acetateNDND $1.75 a \pm 0.24$ $3.72 b \pm 0.50$ NDNDNDMethyl salicylateNDND $1.75 a \pm 0.24$ $3.72 b \pm 0.50$ NDND $10.14 c \pm 1.22$ $12.30 d \pm 0.02$ Methyl salicylateNDND $1.75 a \pm 0.30$ NDND $0.14 c \pm 1.22$ $12.30 d \pm 0.16$ Methyl salicylateNDND $2.82 a \pm 0.30$ NDND $0.14 c \pm 1.22$ $12.30 d \pm 0.16$ Methyl salicylateNDNDND $0.14 c \pm 1.22$ $12.30 d \pm 0.16$ $10.14 c \pm 1.22$ $12.30 d \pm 0.16$ Methyl salicylateNDNDND $0.14 c \pm 1.22$ $12.30 d \pm 0.16$ $10.14 c \pm 1.22$ $12.30 d \pm 0.16$ Methyl salicylateNDND2.65 a \pm	ester							
Hexyl acetate ND 1.58 a ± 0.04 7.70 c ± 0.20 ND ND 3.55 b ± 0.50 sweetfruity-floral Ethyl 3-hexenoate ND ND 0.74a ± 0.05 ND 4.11b ± 0.43 4.60b ± 0.51 3-hexenol acetate ND ND 0.74a ± 0.05 ND 4.11b ± 0.43 4.60b ± 0.51 3-hexenol acetate ND 1.520 a \pm 0.14 ND ND ND Ethyl heptanoate ND 1.35 a \pm 0.1 1.55 b \pm 0.14 ND ND ND Ethyl octanoate ND 1.75 a \pm 0.24 3.72 b \pm 0.50 ND ND ND Benzyl acetate ND ND 10.14 c \pm 1.22 12.30 d \pm 0.02 Sto a \pm 0.16 Methyl salicylate ND ND ND ND 2.90 a \pm 0.16 ND Methyl salicylate ND ND ND ND ND ND Sto a \pm 0.16 Methyl salicylate ND ND ND ND ND ND Sto a \pm 0.16 Ethyl chroxybenzoate ND <td>Ethyl hexanoate</td> <td>ND</td> <td>$4.86 a \pm 0.73$</td> <td>$8.76 \text{ ab} \pm 0.70$</td> <td>ND</td> <td>$6.10 b \pm 1.11$</td> <td>12.35 c ± 1.09</td> <td></td>	Ethyl hexanoate	ND	$4.86 a \pm 0.73$	$8.76 \text{ ab} \pm 0.70$	ND	$6.10 b \pm 1.11$	12.35 c ± 1.09	
Ethyl 3-hexenoate ND ND 0.74a ± 0.05 ND 4.11b ± 0.43 4.60b ± 0.51 3-hexenol acetate ND ND ND ND ND ND 3-hexenol acetate ND ND 15.20 a ± 0.43 ND ND ND 5thyl heptanoate ND 1.35 a ± 0.1 1.55 b ± 0.14 ND ND ND Ethyl octanoate ND 1.35 a ± 0.24 3.72 b ± 0.50 ND ND ND Benzyl acetate ND ND 10.14 c ± 1.22 12.30 d ± 0.02 Benzyl acetate 10.14 c ± 1.22 12.30 d ± 0.02 Methyl salicylate ND ND 2.82 a ± 0.30 ND ND 2.90 a ± 0.16 Ethyl 2-hydroxy-benzoate ND ND 2.45 a \pm 0.04 ND ND 2.90 a \pm 0.16	Hexyl acetate	ND	1.58 a ± 0.04	$7.70 c \pm 0.20$	ND	ΠN	$3.55 b \pm 0.50$	sweet-fruity-floral
3-hexenol acetate ND ND ND ND ND ND Ethyl heptanoate ND 1.35 a ± 0.1 1.52 0 a ± 0.43 ND ND ND Ethyl heptanoate ND 1.35 a ± 0.1 1.55 b ± 0.14 ND ND ND Ethyl octanoate ND 1.75 a ± 0.24 3.72 b ± 0.50 ND 10.14 c ± 1.22 12.30 d ± 0.02 Benzyl acetate ND ND 2.82 a ± 0.30 ND ND 2.90 a ± 0.16 Methyl salicylate ND ND 4.12 a ± 0.06 ND ND ND Ethyl 2-hydroxy-benzoate ND ND 2.45 a ± 0.04 ND 2.50 a ± 0.17	Ethyl 3-hexenoate	ND	DN	0.74a ± 0.05	ND	4.11b 土 0.43	$4.60b \pm 0.51$	
Ethyl heptanoate ND 1.35 a ± 0.1 1.55 b ± 0.14 ND ND ND Ethyl octanoate ND 1.75 a ± 0.24 3.72 b ± 0.50 ND 10.14 c ± 1.22 12.30 d ± 0.02 Benzyl acetate ND ND 2.82 a ± 0.30 ND ND 2.90 a ± 0.16 Methyl salicylate ND ND 4.12 a ± 0.06 ND ND ND Ethyl 2-hydroxy-benzoate ND ND 2.45 a ± 0.04 ND ND 2.50 a ± 0.16	3-hexenol acetate	ND	ND	15.20 a ± 0.43	ND	ND	ND	
Ethyl octanoate ND 1.75 a ± 0.24 3.72 b ± 0.50 ND 10.14 c ± 1.22 12.30 d ± 0.02 Benzyl acetate ND ND 2.82 a ± 0.30 ND ND 290 a ± 0.16 Methyl salicylate ND ND 4.12 a ± 0.06 ND ND 2.90 a ± 0.16 Ethyl salicylate ND ND 2.45 a ± 0.06 ND ND 2.50 a ± 0.17	Ethyl heptanoate	ND	1.35 a ± 0.1	$1.55 b \pm 0.14$	ND	ΔN	ND	
Benzyl acetate ND ND 2.82 a ± 0.30 ND ND 2.90 a ± 0.16 Methyl salicylate ND ND 4.12 a ± 0.06 ND ND ND Ethyl 2-hydroxy-benzoate ND ND 2.45 a ± 0.04 ND ND 2.50 a ± 0.17	Ethyl octanoate	ND	1.75 a ± 0.24	$3.72 b \pm 0.50$	ND	$10.14 c \pm 1.22$	$12.30 d \pm 0.02$	
Methyl salicylate ND ND 4.12 a ± 0.06 ND ND ND Ethyl 2-hydroxy-benzoate ND ND 2.45 a ± 0.04 ND 2.50 a ± 0.17	Benzyl acetate	ND	DN	$2.82 a \pm 0.30$	ND	ND	$2.90 a \pm 0.16$	
Ethyl 2-hydroxy-benzoate ND ND 2.45 a \pm 0.04 ND 2.45 a \pm 0.04 ND 2.50 a \pm 0.17	Methyl salicylate	ND	DN	$4.12 a \pm 0.06$	ND	ND	ND	
	Ethyl 2-hydroxy-benzoate	ND	DN	2.45 a ± 0.04	ND	ND	$2.50 a \pm 0.17$	

Odor descriptors

Leccino table olives

Compounds	Cel	llina di Nardò table oli	ves		Leccino table olives		Odor descriptors
	30 days	90 days	180 days	30 days	90 days	180 days	
	mean ± SD μg/kg∗	mean ± SD μg/kg*	mean ± SD μg/kg*	mean ± SD μg/kg*	mean ± SD μg/kg*	mean ± SD μg/kg*	
Phenyl ethyl acetate	QN	ŊŊ	4.56 b ± 0.12	DN	ND	QN	
Total amounts	9.90 ± 0.52	55.87 ± 3.25	352.97 ± 16.62	7.98 ± 0.44	49.81 ± 6.58	147.61 ± 13.44	
ALCOHOLS							
1-propanol	ND	ND	$19.95 b \pm 0.84$	ND	ND	$5.27 a \pm 0.60$	
2 methyl-1-propanol	$2.11 a \pm 0.80$	$13.96 c \pm 1.10$	$9.43 b \pm 0.80$	ND	7.74 b ± 0.54	$8.01 b \pm 0.33$	Penetrating, solvent
2+3-methyl-1-butanol	$10.7 a \pm 0.20$	$86.80 c \pm 6.02$	$60.21 \text{ b} \pm 2.70$	$10.30 a \pm 0.41$	$49.95 b \pm 2.11$	$46.20 \text{ b} \pm 1.24$	woody-whiskey-sweet
3 methyl-3-buten-ol	ND	ND	1.60 a ± 0.07	NDa	ND	ND	
Hexanol	$13.40 \text{ bc} \pm 1.10$	$17.50 \text{ de} \pm 1.32$	$20.20 e \pm 1.21$	8.76 a ± 0.72	$11.26 \text{ ab} \pm 1.03$	$15.00 \text{ cd} \pm 1.80$	fruit-banana
trans 3-hexen-ol (E)	ND	ND	1.21 a ± 0.24	ND	ND	QN	green
cis 3-hexen-ol (Z)	$15.43 b \pm 0.55$	$25.26 c \pm 0.55$	$50.95 d \pm 0.92$	1.74 a ± 0.04	$1.81 a \pm 0.23$	$2.44 \text{ a} \pm 0.04$	green
Heptanol	ND	$5.10 a \pm 0.50$	$6.40 b \pm 0.25$	ND	8.81 cd ± 0.90	$12.20 d \pm 0.25$	earthy-sweefy
Nonanol	ND	ND	$6.85 a \pm 0.10$	ND	ND	ΟN	fatty
Benzylalcohol	ND	$7.80 b \pm 0.80$	ND	ND	$2.31 a \pm 0.13$	$8.30 b \pm 0.70$	
Total amounts	41.64 ± 2.65	156.35 ± 10.29	179.43 土 7.13	20.80 ± 1.17	81.92 ± 4.94	97.42 ± 4.96	
HYDROCARBONS							
Octane	ND	ND	ND	$68.50 a \pm 0.93$	78.90a ± 3.42	$71.30 a \pm 4.10$	
Toluene	6.81 a ± 1.65	ND	ND	ND	ND	ND	
Styrene	ND	43.30 b \pm 5.16	$31.90 b \pm 1.80$	$10.70 a \pm 1.02$	89.80 c ± 3.25	$77.21 \text{ c} \pm 2.10$	
Total amounts	6.81 ± 1.65	43.27 ± 5.16	31.90 ± 1.80	79.17 ± 1.95	168.70 ± 6.67	148.51 ± 6.20	
FATTY ACIDS							
Acetic acid	$7.73 a \pm 0.04$	$13.25 a \pm 0.22$	$59.32 c \pm 1.30$	$5.22 \text{ a} \pm 0.30$	8.7 a ± 1.1	$25.06 \text{ b} \pm 0.93$	sour-vinegary
Propanoic acid	ND	ND	$10.22 a \pm 0.32$	ND	ND	QN	pungent-sour
2-methyl propanoic acid	ND	ND	$1.2 a \pm 0.05$	ND	ND	ND	
Total amounts	7.73 ± 0.04	13.25 ± 0.22	70.72 ± 1.67	5.22 ± 0.30	8.67 ± 1.1	25.06 ± 0.93	
VOLATILES PHENOLS							
Guaiacol	ND	ND	ND	ND	35.55 a ± 3.42	182.20 b ± 1.80	
TERPENES							
3,7 dimethyl 1,3,7 octatriene	$8.01c \pm 0.53$	$10.94 \mathrm{e} \pm 0.90$	$9.50 d \pm 0.23$	2.95 a ± 0.91	$5.65 b \pm 0.22$	$4.70 \text{ b} \pm 0.12$	
(cimene)							
3,7 dimethyl 2,6 octadienal	ND	ND	ND	ND	ND	$0.80 \ a \pm 0.20$	
α pinene	ND	ND	ND	ND	$2.10 a \pm 0.15$	$3.01 b \pm 0.40$	
Linalolox	ND	ND	$3.63 a \pm 0.06$	ND	ND	ND	
Linalol	ND	ND	1.40 a ± 0.05	ND	ND	ND	
Total amounts	8.01 ± 0.53	10.94 ± 0.90	14.50 ± 0.34	2.95 ± 0.91	7.75 ± 0.37	8.51 ± 0.72	
*Results expressed as mean \pm SD For each compound, according to t	i; ND, not detected. he result of the One-Way.	ANOVA test, concentrati	on values that do not sh	are a common superscri	ot are significantly differe	ent (P < 0.05).	

Table 3 | Continued

Table 4 SPME/GC-MS quantita	tive data, including concentratio	n (µg/L) with standard	deviation (SD) of all t	he volatile compounds ide:	entified in Cellina di Nardò	and Leccino brines.
Compounds	Ö	ellina di Nardò brines			Leccino brines	
	30 days	90 days	180 days	30 days	90 days	180 days
	mean ± SD µg/L*	mean ± SD µg/L*	mean ± SD µg/L*	mean ± SD μg/L	mean ± SD μg/L*	mean ± SD μg/L*
ALDEHYDES-KETONS						
2-methyl propanal	$2.50 a \pm 0.23$	ND	QN	QN	QN	QN
2-methyl butanal	$7.93 c \pm 0.90$	ND	DN	$4.02 \text{ ab} \pm 0.32$	$5.70 \text{ bc} \pm 0.20$	$3.40 a \pm 0.20$

	mean ± SD μg/L*	mean ± SD	mean ± SD	mean ± SD μg/L	mean ± SD μg/L*	mean ± SD μg/L*
		μg/L*	μg/L*			
ALDEHYDES-KETONS						
2-methyl propanal	2.50 a ± 0.23	QN	QN	QN	ND	QN
2-methyl butanal	$7.93 c \pm 0.90$	ND	DN	$4.02 \text{ ab} \pm 0.32$	$5.70 \text{ bc} \pm 0.20$	$3.40 a \pm 0.20$
3-methyl butanal	$18.60 c \pm 0.43$	ND	ND	$5.64 \text{ b} \pm 0.51$	$2.10 a \pm 0.9$	$2.91 \text{ ab} \pm 0.05$
Hexanal	$1.70 b \pm 0.10$	DN	DN	$0.60 a \pm 0.20$	ND	ND
Acetoino	ND	DN	$13.02 \text{ b} \pm 0.44$	QN	ND	QN
Nonanal	ND	$1.02 a \pm 0.02$	QN	1.53 a ± 0.34	$7.70 b \pm 0.71$	QN
Decanal	ND	ND	QN	ND	$1.4 b \pm 0.10$	QN
Benzaldehyde	$5.75 b \pm 1.05$	$3.45 b \pm 0.10$	QN	$3.50 b \pm 0.25$	$3.30 a \pm 0.40$	$8.76 c \pm 0.84$
Total amounts	36.48 ± 2.71	4.47 ± 0.12	13.02 ± 0.44	15.29 ± 1.62	20.20 ± 1.50	15.07 ± 1.09
ESTERS						
Methyl acetate	3.76 a ± 0.76	7.82 a ± 1.90	$32.71 c \pm 1.13$	$3.50 a \pm 0.11$	3.24a ± 0.05	16.80 b ± 0.96
Ethyl acetate	33.71 a ± 2.40	174.11 c ± 4.93	226.97 d ± 5.80	21.26 a ± 1.60	$25.63 a \pm 1.54$	114.21 b ± 9.43
Ethyl propanoate	ND	ND	96.97 a ± 2.05	ND	ND	DN
n-propyl acetate	ND	ND	$91.70 a \pm 5.52$	ND	ND	ND
Acetic acid 2 methyl propyl ester	ND	$2.78 a \pm 0.20$	$13.45 \text{ b} \pm 0.51$	ND	ND	1.64 a ± 0.07
Ethyl butanoate	ND	DN	$2.20 a \pm 0.10$	ND	ND	DN
2 methyl ethyl butanoate	ND	ND	6.96 a ± 0.74	ND	ND	QN
3 methyl ethyl butanoate	ND	ND	1.14 a ± 0.10	ND	ND	DN
Isoamyl acetate	0.92 a ± 0.41	$5.04 b \pm 0.66$	$27.22 c \pm 1.10$	ND	ND	$4.55 b \pm 1.03$
Ethyl hexanoate	ND	$1.00 a \pm 0.05$	0.94 a ± 0.10	ND	ND	ND
Ethyl octanoate	ND	ND	QN	ND	ND	$0.90 a \pm 0.20$
Methyl salycilate	ND	ND	DN	ND	ND	10.92a ± 0.90
3 hexen-1-ol acetate	ND	ND	$2.06 b \pm 0.95$	ND	ND	QN
Ethyl lactate	ND	ND	$1.64 c \pm 0.72$	ND	ND	$0.80 b \pm 0.10$
Total amounts	38.40 ± 3.57	190.75 ± 7.54	503.96 ± 18.82	24.76 土 1.71	28.87 ± 1.59	149.82 ± 12.60
ALCOHOLS						
2 butanol	ND	ND	$3.34 a \pm 0.10$	ND	ND	ND
1 propanol	1.60 a ± 0.11	$4.05 a \pm 0.21$	$36.50 b \pm 2.40$	ND	1.10 a ± 0.10	3.45 a ± 1.01
2 methyl-1-propanol	$6.26 a \pm 0.70$	$15.61 b \pm 0.71$	$15.82 b \pm 1.65$	$0.89 a \pm 0.12$	$1.63 a \pm 0.10$	3.50 a ± 0.44
1 butanol	1.42 ± 0.06	1.32 ± 0.10	1.02 ± 0.03	1.05 ± 0.03	1.27 ± 0.12	4.12 ± 0.60
2+3 methyl-1-butanol	$33.60 b \pm 0.81$	204.33 d ± 3.50	$109.66 c \pm 5.32$	ND	18.25 ab ± 3.20	$15.17 \text{ ab} \pm 1.51$
Pentan-1-ol	ND	DN	7.07 a ± 1.60	ND	ND	QN
2 penten-ol	ND	$0.34a \pm 0.04$	$3.70 b \pm 0.60$	DN	ND	ND
3 methyl-2-buten-ol	ND	$1.06 a \pm 0.30$	QN	QN	ND	ND
						(Continued)

Compounds	C	llina di Nardò brines			Leccino brines	
	30 days	90 days	180 days	30 days	90 days	180 days
	mean ± SD μg/L∗	mean ± SD µg/L∗	mean ± SD µg/L*	mean ± SD μg/L	mean ± SD μg/L∗	mean ± SD μg/L*
Hexan-1-ol	14.80 ab	31.60 c ± 2.11	$19.62 \text{ ac} \pm 2.85$	5.80 a ± 0.70	10.14 ab ± 0.10	21.98 ac ± 0.90
trans 3 hexen-ol (E)	DN	$0.93 \ a \pm 0.50$	$1.50 b \pm 0.50$	ND	ND	ND
cis 3 hexen-ol (Z)	$18.66 \text{ ab} \pm 2.96$	$39.46 \text{ bc} \pm 2.60$	65.80 c ± 2.32	$11.60 \text{ ab} \pm 0.04$	1.48 a ± 0.42	5.42 a ± 1.05
Heptanol	ND	QN	ND	$2.80 a \pm 0.20$	$2.41 a \pm 0.20$	$9.53 b \pm 0.60$
Benzylalcohol	ND	1.75 a ± 0.90	$5.65 b \pm 0.63$	$0.68 a \pm 0.14$	$0.5 a \pm 0.04$	$1.06 a \pm 0.25$
Total amounts	76.34 ± 6.18	300.40 ± 10.96	428.42 ± 18.00	163.03 ± 1.23	145.71 土 4.38	241.94 ± 6.36
HYDROCARBONS						
Trimethyl benzen	1.30 ab \pm 0.20	0.64 a ± 0.13	$1.20 \text{ ab} \pm 0.32$	$14.98 c \pm 0.22$	$2.10 \text{ ab} \pm 0.34$	$2.35 \text{ ab} \pm 0.60$
Styrene	ND	1.60 a ± 0.30	$0.50 a \pm 0.03$	$2.20 \text{ a} \pm 1.01$	$24.53 b \pm 3.80$	ND
Total amounts	1.30 ± 0.20	2.22 ± 0.33	1.70 ± 0.35	17.18 ± 1.23	26.63 ± 4.14	2.35 ± 0.60
FATTY ACIDS						
Acetic acid	$2.72 a \pm 0.10$	11.50b ± 2.10	164.30d ± 8.64	1.30a ± 0.24	ND	$36.31c \pm 3.60$
Propanoic acid	ND	DN	$35.01 b \pm 2.12$	ND	ND	ND
2 methyl propanoic acid	ND	0.44a ± 0.10	$1.66 b \pm 0.05$	ND	ND	ND
2 methyl butanoic acid	ND	DN	$2.80 c \pm 0.13$	ND	ND	$1.30 b \pm 0.16$
Total amounts	2.72 ± 0.10	11.93 ± 2.20	203.74 ± 10.94	1.30 ± 0.24		36.31 ± 3.76
VOLATILES PHENOLS						
Guaiacol	2.60 a ± 0.41	0.93a ± 0.23	$22.91 b \pm 0.53$	ND	ND	81.51 c ± 3.22
Terpenes				ND	ND	
Terpineol	ND	QN	ND	ND	ND	1.10 b ± 0.14
Lactones						
Butyrolacton	ND	$0.92 \text{ c} \pm 0.15$	$0.55 b \pm 0.14$	ND	ND	ND
*Results expressed as means \pm SD; ND, not For each compound, according to the result o	detected. f the One-Way ANOVA test, c	oncentration values that c	to not share a common su	verscript are significantly diff	erent (P < 0.05).	

Table 4 | Continued



Lactic acid bacteria are the most important group of bacteria in olives. They are able to convert fermentable sugars to lactic acid and other organic acids depending on their metabolic pathways. In fermented olives, homo fermentative LAB such as Lactobacillus, Streptococcus, and Pediococcus and hetero fermentative LAB such as Leuconostoc and some members of Lactobacillus (Abriouel et al., 2012; Randazzo et al., 2012) can be found. In particular, Lactobacillus spp. play a major role, while Leuconostoc and Pediococcus are involved to a lesser degree (Abriouel et al., 2011; Corsetti et al., 2012). In accordance with the above previous studies, bacterial isolates obtained from both Leccino and Cellina di Nardò fermentation belonged to L. plantarum species with a very limited number of Kocuria and S. salitolerans found in the second one. However, because the demonstrated beneficial effects for human health deriving by the application of microbial starters in table olive processing (Silva et al., 2011; Argyri et al., 2013; Bautista-Gallego et al., 2013; Blana et al., 2014), the probiotic activity of the characterized yeasts and LAB will be further investigated. Moreover,



another trait that will be examined is the ability of the identified isolates to produce bacteriocins, since this feature can increase the competitiveness of the producer strain in food and contribute to prevent food spoilage obtaining longer shelf life and safety of the products reducing the use of chemical preservatives (Holzapfel, 2002; Leroy and DeVuyst, 2004; Arroyo-López et al., 2005).

The differences observed in microbial population in the two fermentations could be related to the different physico-chemical composition of the two olive cultivars. Preliminary data obtained in our laboratory indicated that Cellina di Nardò and Leccino olives differed in their chemical profile because Leccino is richer of organic acids (citric, tartaric, malic, succinic, acetic acid) and sugars (glucose, fructose, and sucrose) than Cellina di Nardò. Moreover, significant differences were detected also in volatile profiles of the two olive varieties: Leccino showed a 1.5-fold higher content of aldehydes than Cellina di Nardò and the presence of a high level of hydrocarbons, whereas Cellina di Nardò olives revealed a 2-fold and 3-fold higher contents of alcohols and terpens than Leccino, respectively (data not shown). The results of this study showed that yeasts are responsible of the first part of fermentation (at least 90 days) and then LAB, together with yeasts, complete the process for the period ranging from 90th day to 180th day. Although different yeasts and bacteria species are responsible of the process, these data confirmed the microbial evolution described in spontaneous fermentation of Conservolea and Kalamàta olives (Bleve et al., 2015).

In order (i) to identify specific molecules suitable to be proposed as tools for process monitoring among the chemical compounds produced during fermentation and (ii) to correlate the results of physico-chemical analyses with the microbial metabolism, we optimized a novel procedure for the analysis of table olive and brine chemical composition. In particular, we used the solid-phase micro extraction (SPME), a technique widely used for the analysis of olive oil (Kanavouras et al., 2005; Temime et al., 2006; Baccouri et al., 2008) and table olive aroma compounds (Kalua et al., 2007; Aponte et al., 2010; Malheiro et al., 2011). The statistical analysis (PCA) of the outcome of the chemical assays allowed us to establish a correlation among the profiles of volatile molecules and microbial growth also giving indications of the molecules that can be considered helpful descriptors of each fermentation stage. Soluble sugars, such as glucose and fructose, are substrates for microbial fermentation giving primary and secondary metabolites responsible for the good organoleptic characteristics and the distinctive flavor of the final product. The prevalence of residual glucose over fructose could be related, besides varietal differences, to the fructophilic character of some non-Saccharomyces yeasts commonly found in brine. Independently of the total number of LAB, the great variability of lactic acid content could be related to many factors, such as the content of fermentable sugars, as well as the balance among species/strains present and their acidifying activity.

Analogously to the data observed in Conservolea and Kalamàta table olives, the first stage (30 days) of fermentation is mainly characterized by a high concentration of aldehydes (2+3-methylbutanal, 2-methylpropanal), that are very important compounds in fruits and vegetables and can contribute to characteristic herbaceous flavors. Italian cultivars presented lower concentrations of C2-, C4-, and C6-aldehydes and ketons than the two Greek varieties (Bleve et al., 2015).

The second fermentation period (90 days) was mainly characterized by the presence of higher alcohols, styrene, and terpenes. According to data reported in Tables 3, 4, olives and brines, during the fermentation process, showed changes in their sensory profile, with a decrease in the number of aldehydes and the appearance of new compounds. Indeed, aldehydes undergo a series of enzymatic transformations mediated by microbial isomerases and alcohol dehydrogenases, leading to C6 alcohols (Cavalli et al., 2004). The increase in isoamyl alcohols during the first fermentation step indicates the role of yeasts in driving this part of the process and suggests that these molecules are important markers of yeast metabolism. In particular, among alcohols, 2+3 methyl-1-butanol (isoamyl alcohol, fruity-winey notes), hexanol (fruity-green notes) and cis 3-hexen-1-ol (green notes) prevailed both in olives and brines. Higher alcohols (1-propanol and 2-methyl-1 propanol) derive from the reduction process of aldehydes, but can also be linked to the microbial deamination

process of amino acids (McFeeters, 2004). In the two Italian table olive cultivars Cellina di Nardò and Leccino there was the absence of sesquiterpenes (farnesene and cubebene) and of monoterpene (cimene) found in the two Greek varieties Conservolea and Kalamàta. Moreover, Greek cultivars were richer of alcohols than Italian ones (Bleve et al., 2015).

The third fermentation stage (180 days) was characterized by a significant increase in alcohols/esters/fatty acids/hydrocarbons, probably due to the different pathway undertaken by enzymes produced by LAB, yeasts and/or other microorganisms. There was an increase in acetic acid, which is indicative of a heterolactic fermentation process to the microbial action, mainly operated by LAB. A considerable presence of acetate esters, acetic acid, propanoic, and 2-methylpropanoic acid negatively correlated to the PC1 semi-axis in Cellina di Nardò (80.44% variance) and Leccino (77.20% variance) olive maps, characterizing the sixth month of the process (T180) linked to the appearance of bacteria. Acetic acid is produced by yeast activity, but its concentration increased due to the activity of bacteria in the last stage of fermentation. Although the presence of short chain fatty acids is usually related to the appearance of negative odors, they can be very important for aromatic equilibrium because they are opposed to the hydrolysis of the corresponding esters, as already reported in wines (Bertrand, 1981; Edwards et al., 1990). Moreover, Greek cultivars were richer of esters than Italian ones, and the comparison of chemical profiles revealed the absence of ethyl lactate and the presence of 3-hexenol-acetate in Italian table olives (Bleve et al., 2015). Compounds identified as chemical descriptors are common not only for the two fermentations of Cellina di Nardò ad Leccino, but also for the two Greek cultivars of table olives (Bleve et al., 2015) and are strongly related to the activity and evolution of microorganisms (yeasts and LAB) during the process. Several authors reported the accumulation of hydroxytyrosol in the brine, as the main simple phenolic compound in olives at the end of fermentation (Romero et al., 2004; Ben Othman et al., 2008; Pistarino et al., 2013). Accumulation of hydroxytyrosol in the brine can be attributed to the ability of microorganisms to hydrolyze complex compounds such as oleuropein, that decreased during the process, by means of beta-glucosidase (Servili et al., 2006). Indeed in our samples, oleuropein was completely absent after the 30th day in Cellina di Nardò and after 90 days of fermentation in Leccino brines. According to the here reported results, yeasts play an important role in the olive debittering process. The different physico-chemical characteristics of each olive variety can be responsible of the differences observed at the end of fermentation in Italian and Greek cultivars. In fact, the Italian olives showed a richer profile of phenolic compounds than Greek olives (Bleve et al., 2015).

In conclusion, the chemical descriptors identified in this investigation represent the first evidence that specific molecules can be suggested as predictors for monitoring the fermentation process of black table olives produced by Greek method. For the first time, it has been described the technological and molecular characterization of yeast and LAB isolates associated to Cellina di Nardò and Leccino and it represents the first stage of a selection procedure for the production of mixed autochthonous fermentation starters. After assessment of the enzymatic traits and their ability to grow in model brines, two starter formulations, composed of *P. anomala/L. plantarum* (Cellina di Nardò) and *S. cerevisiae/L. plantarum* (Leccino) strains are going to be validated through olive industrial-scale fermentation.

Further work is now forthcoming in order to identify organoleptic descriptors and to set up a panel test for the evaluation of the Cellina di Nardò and Leccino olives.

AUTHOR CONTRIBUTIONS

Substantial contributions to the conception and design of the work, acquisition, analysis, and interpretation of data (Gianluca Bleve, Francesco Grieco, Giovanni Mita, Maria Tasioula-Margari, Antonio F. Logrieco); acquisition, analysis and interpretation of data (Maria Tufariello, Miriana Durante, Ezio Perbellini, Francesca A. Ramires, Maria S. Cappello, Stefania De Domenico); drafting the work and revising it critically for intellectual content (Gianluca Bleve, Francesco Grieco, Giovanni Mita, Maria Tasioula-Margari). All authors approved the final version of the manuscript to be submitted for publication and agreed to be accountable for all aspects of the work in ensuring that questions related to the accuracy and integrity of any part of the work are appropriately investigated and resolved.

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