# AIBr<sub>3</sub>-Promoted stereoselective *anti*-hydroarylation of the acetylene bond in 3-arylpropynenitriles by electron-rich arenes: synthesis of 3,3-diarylpropenenitriles

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#### Full Research Paper

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#### **Abstract**

Reactions of 3-arylpropynenitriles (ArC=CCN) with electron-rich arenes (Ar'H, benzene and its polymethylated derivatives) under the action of aluminum bromide (AlBr<sub>3</sub>, 6 equiv) at room temperature for 0.5–2 h result in the stereoselective formation of 3,3-diarylpropenenitriles (Ar(Ar')C=CHCN) in yields of 20–64%, as products of mainly *anti*-hydroarylation of the acetylene bond. The obtained 3,3-diarylpropenenitriles in triflic acid  $CF_3SO_3H$  (TfOH) at room temperature for 1 h are cyclized into 3-arylindenones in yields of 55–70%.

#### Introduction

Conjugated acetylene nitriles (propynenitriles, R-C=C-C=N) are versatile building blocks in organic synthesis for the preparation of a plethora of functionalized compounds and heterocycles. The presence of conjugated acetylene and nitrile bonds in these compounds leads to an enhancement of reactivity of both functional groups. Thus, propynenitriles take part in electrophilic [1,2] and nucleophilic [3-6] addition reactions onto the

acetylene bond leading to various substituted nitriles. Reactions onto both acetylene and nitrile groups are widely used for the construction of various heterocyclic systems [7-15].

Recently, we have shown that reactions of 3-arylpropenenitriles (cinnamonitriles, ArCH=CHCN) with arenes (Ar'H) under the superelectrophilic activation by the Brønsted super-

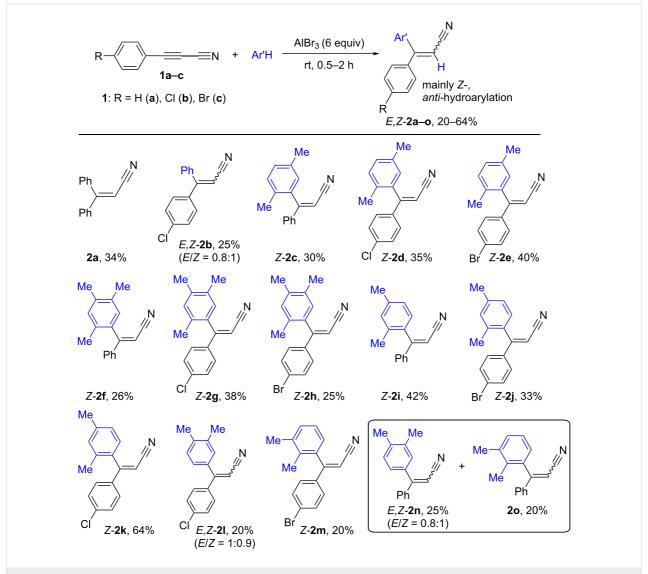
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acid  $CF_3SO_3H$  (TfOH, triflic acid) or the strong Lewis acid  $AlBr_3$  result in the formation of 3,3-diarylpropanenitriles ( $Ar(Ar')CHCH_2CN$ ) through the regioselective hydroarylation of the carbon–carbon double bond. In TfOH, the reactions proceed further to 3-arylindanones, as products of the intramolecular aromatic acylation of the 3,3-diarylpropanenitriles by the electrophilically activated nitrile group [16,17]. Based on this study and our work on electrophilic transformations of alkynes [18-20], we investigated reactions of 3-arylpropynenitriles under electrophilic activation conditions (see [21] for the chemistry of superelectrophilic species). The goal of this work was to study the reactions of 3-arylpropynenitriles with arenes under the action of the strong Lewis acids, aluminum halogenides  $AlX_3$  (X = Cl, Br) and the Brønsted superacid TfOH ( $CF_3SO_3H$ ).

#### Results and Discussion

It was found that acetylene nitriles  $1\mathbf{a}$ — $\mathbf{c}$  reacted with arenes in the presence of excess AlBr<sub>3</sub> (6 equiv) at room temperature for 0.5–2 h to afford E,Z-3,3-diarylpropenenitriles  $2\mathbf{a}$ — $\mathbf{o}$  as products of the regioselective hydroarylation of the acetylene bond (Scheme 1). These reactions proceeded only with an excess of AlBr<sub>3</sub> (6 equiv). Thus, running the reaction of nitrile  $1\mathbf{a}$  with benzene under the action of less amount AlBr<sub>3</sub> (1–4 equiv) resulted in an incomplete conversion of the starting compound  $1\mathbf{a}$  and a low yield of the target product  $2\mathbf{a}$  (<10%).

The yields of the compounds **2** were moderate (20–64%) that may be caused by possible transformations of the nitrile group in the electrophilic medium leading to oligomeric material [22]. The reaction proceeded rather regionselectively giving mainly



Scheme 1: AlBr<sub>3</sub>-promoted hydroarylation of the acetylene bond of 3-arylpropynenitriles 1a-c by arenes with the formation of 3,3-diarylpropenenitriles E-,Z-2a-o.

Z-isomers of nitriles 2, as products of an anti-addition of hydrogen and the aryl group to the carbon-carbon triple bond. Only in three cases, mixtures of E,Z-isomers (2b,l,n) in a ratio of ≈1:1 were obtained. The E,Z-configuration of compounds 2a-o was determined by NOESY correlations between the vinyl proton and the aromatic protons or methyl groups in neighboring aryl substituents (see Supporting Information File 1). However, the configuration of nitrile 20 was unclear. Benzene, o-, m-, p-xylenes, and 1,2,4-trimethylbenzene (mesitylene) were included in the hydroarylation of nitriles 1. Reactions of nitriles **1a-c** with o-xylene led to the formation of regioisomers derived from the electrophilic substitution at different positions of this arene. Thus, nitrile 1a gave two types of regioisomers 2n and 20. After reactions of nitriles 1b,c with o-xylene, compounds 2l and 2m correspondingly were obtained; other regioisomers were not isolated in amounts that were high enough for their identification.

This transformation was also tested with another strong Lewis acid, aluminum chloride (AlCl<sub>3</sub>), for the reaction of nitrile **1a** with benzene. However, in this case, mainly oligomeric material was obtained and the yield of the target reaction product **2a** was lower (20%) than in the same reaction with AlBr<sub>3</sub>. This result revealed the less effectiveness of AlCl<sub>3</sub> for the hydroarylation of nitriles **1**.

One may propose the following reaction mechanism (Scheme 2). The coordination of AlBr<sub>3</sub> to both the nitrile and acetylene bonds of the starting compound 1 furnishes the highly electrophilic species A bearing a positive charge on the acetylenic carbon atom C3. The subsequent reaction of species A with the arene molecule via electrophilic aromatic substitution results in the formation of species B. The most probably, this stage proceeds stereoselectively due to spatial factors, with the incoming arene Ar'H attacking the acetylene bond in an *anti*-position to the bulky AlBr<sub>3</sub>, that determines the final predominant formation of the mainly *anti*-hydroarylation products of the starting nitriles 1. At the last step of the reaction, a proton substitutes AlBr<sub>3</sub>, and final hydrolysis of the reaction mixture gives rise to nitriles 2.

It should be noted that this AlBr<sub>3</sub>-promoted hydroarylation of acetylene nitriles 1 (Scheme 1) is a novel transition-metal (Pd,

Pt, Rh, etc.)-free stereoselective way for the synthesis of compounds **2**. These compounds can be alternatively obtained by a Pd-catalyzed Heck reaction of 3-arylpropenenitriles with iodoarenes [23] or by a Cu-catalyzed hydroarylation of 3-arylpropynenitriles with arylboronic acid [24,25]. There is one example of use of dicyanoacetylene in a similar AlCl<sub>3</sub>-catalyzed hydroarylation of an acetylene bond in the synthesis of cyclophanes [26].

Additionally, the cyclization of two selected nitriles **2c**,g into 3-arylindanones **3a**,b correspondingly was carried out in triflic acid TfOH (CF<sub>3</sub>SO<sub>3</sub>H) at room temperature for 1 h (Scheme 3). The intramolecular aromatic substitution by the electrophilically activated nitrile group took place in the more electron-rich methylated aryl ring. A similar cyclization of 3,3-diarylpropanenitriles into 3-arylindanones in TfOH was described by us previously [17]. It should be specially emphasized that the synthesis of indenones is an important goal in organic chemistry since this structural motif is associated with interesting chemical and biological properties [27-32].

We also conducted reactions of nitriles **1a-c** with arenes in TfOH at room temperature, which, however, led to complex mixtures of reaction products. Only in the cases of the reaction of nitriles **1a,b** with benzene in TfOH, the hydrophenylation

$$Ar \xrightarrow{\frac{3}{2}} \frac{1}{=} N \xrightarrow{2AlBr_3} Ar \xrightarrow{AlBr_3} Ar \xrightarrow{Ar'} Ar' \xrightarrow{$$

Scheme 2: Plausible mechanism for reaction of acetylene nitriles 1 with arenes leading to nitriles 2.

products, i.e., nitriles **2a**,**b**, were obtained (Scheme 4; compare with synthesis of **2a**,**b** in Scheme 1).

In general, the comparison of reactions of 3-arylpropenenitriles (cinnamonitriles; ArCH=CHCN) (see our work [17]) and 3-arylpropynenitriles 1 (this work) under electrophilic activation reveals unambiguously that the electrophilic intermediates generated from acetylene nitriles 1 possess higher reactivity. Thus, the hydroarylation of 3-arylpropynenitriles 1 promoted by AlBr<sub>3</sub> proceeds at room temperature (Scheme 1), whereas the same reaction of 3-arylpropenenitriles needs elevated temperature up to 80 °C [17]. Reactions of acetylene nitriles 1 with arenes in TfOH have complex character, contrary to 3-arylpropenenitriles, which react smoothly with arenes in TfOH at room temperature [17].

#### Conclusion

We have developed a novel transition-metal (Pd, Pt, Rh, etc.)-free procedure for the regio- and stereoselective hydroarylation of the carbon-carbon triple bond in 3-arylpropynenitriles by arenes under electrophilic activation by aluminum bromide AlBr<sub>3</sub>. The obtained 3,3-diarylpropenenitriles were cyclized into 3-arylindenones in triflic acid CF<sub>3</sub>SO<sub>3</sub>H (TfOH).

## Supporting Information

#### Supporting Information File 1

Experimental procedures, compound characterization, and  $^1\mathrm{H}$  and  $^{13}\mathrm{C}$  NMR spectra of compounds.

[https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-17-180-S1.pdf]

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