

Heavy Enzymes and the Rational Redesign of Protein Catalysts

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An unsolved mystery in biology concerns the link between enzyme catalysis and protein motions. Comparison between isotopically labelled “heavy” dihydrofolate reductases and their natural-abundance counterparts has suggested that the coupling of protein motions to the chemistry of the catalysed reaction is minimised in the case of hydride transfer. In alcohol dehydrogenases, unnatural, bulky substrates that induce additional electrostatic rearrangements of the active site enhance

coupled motions. This finding could provide a new route to engineering enzymes with altered substrate specificity, because amino acid residues responsible for dynamic coupling with a given substrate present as hotspots for mutagenesis. Detailed understanding of the biophysics of enzyme catalysis based on insights gained from analysis of “heavy” enzymes might eventually allow routine engineering of enzymes to catalyse reactions of choice.

Introduction

Enzymes are naturally occurring protein catalysts that control the chemistry of life.^[1] Their catalytic power is exploited by industry for, inter alia, waste management and the production of food, pharmaceuticals, textiles and fine chemicals.^[2] Because they are able to catalyse difficult synthetic reactions without the need for extreme temperature, high pressure or toxic chemicals and require simplified downstream processing (due to fewer unwanted side products), enzymes are increasingly becoming viewed as sustainable alternatives to synthetic catalysts, offering reductions in energy costs and in environmental damage.^[2,3]

However, the selectivity of enzymes is often for substrates that are not optimal for the needs of industry, a major limitation that holds back their widespread use.^[4] This problem can be addressed by the redesign of natural enzymes, either through rational design or through directed evolution. Though some key achievements have been seen, redesigning enzymes is not a simple task.^[4c,5] A recent review of enzyme engineering showed that only $\approx 5\%$ of modified enzymes in a literature sample (60 enzymes produced by directed evolution and 15

computationally designed/redesigned enzymes) gave more than 10^4 -fold increases in catalytic efficiency ($k_{\text{cat}}/K_{\text{M}}$) over the native proteins.^[4c] To understand the cause of this failure and to become better engineers of enzymes, it is essential further to improve our understanding of the principles by which natural enzymes operate.

In a conventional understanding of enzymology, the catalytic power of enzymes comes from their ability to stabilise transition states through binding interactions, thus lowering reaction activation energies.^[6] However, antibodies that possess binding sites that are complementary to transition states either fail to catalyse reactions or display significantly lower efficiency than the natural enzymes.^[7] There are many possible explanations for this failure, including poor design of the hapten.^[7] It has also been noted that antibodies lack the residues required in order to participate in catalysis: for acid–base proton shuffling, for example.^[7b] Also, unlike most enzymes, catalytic antibodies lack the ability to mediate conformational changes for the binding and release of substrates and products.^[7] To explain these observations, controversial enzymology models have been developed. Notably, it has been suggested that the catalytic power of enzymes is mediated through dynamic motions in the protein.^[8] This has been a topic of intense debate, and some researchers have disputed the need to invoke such “promoting motions” and have proposed that transition state theory alone can explain the catalytic power of enzymes.^[6b,e,9]

Heavy Enzymes

The observation that isotopically labelled enzymes sometimes show reduced rates of catalysis was first made in 1969.^[10] In 2011, Schramm and co-workers pioneered the utilisation of such isotopically labelled enzymes to probe the possible roles of protein motions in catalysis.^[11] In this method, all non-

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exchangeable carbon, nitrogen and hydrogen atoms in an enzyme are replaced with their heavy counterparts— ^{15}N , ^{13}C and ^2H —to generate a modification with an increased mass and slower motions. The increased atomic mass alters the vibrational frequencies but according to the Born–Oppenheimer approximation leaves the potential energy surface (PES) unaltered. The rate of the chemical step is measured, and the ratio of the rate constants for the light enzyme to those for its heavy counterpart gives an enzyme kinetic isotopic effect (KIE). If a significant fraction of protein atoms has been isotopically substituted, an enzyme KIE of unity implies no significant coupling of dynamics to the chemical step, whereas a KIE above or below unity is taken to imply significant coupling of protein motions to catalysis.

Strategies previously employed in the literature by various groups include isotopically labelling the entire enzyme^[11b,12] or labelling either of single amino acid residues^[13] or of particular segments, such as mobile loops (Figure 1).^[14] The effects of protein isotope labelling on transition states have been characterised in different enzymes including purine nucleotide phosphorylase (PNP),^[11a,13,15] HIV protease (HIV-1 PR),^[11b] alanine racemase,^[12c] dihydrofolate reductase (DHFR),^[12b,h,j,14a,16] pentaerythritol tetranitrate reductase (PETNR),^[12d,e] formate dehydrogenase (FDH),^[12f] lactate dehydrogenase (LDH)^[12g] and alcohol dehydrogenase.^[12k] In most of these cases, isotope labelling reduced the rate of the chemical step.^[11,12c–e,g–j,13a,14a,16a] These observations are often interpreted to demonstrate that protein motions couple to active-site chemistry by engaging a probabilistic search for conformations that promote the crossing of the energy barrier.^[11,12c–e,13] Nevertheless, for most of these enzymes, mechanistic studies in greater depth are needed, because computational analysis, mutagenesis studies and comparisons between homologues are often lacking.

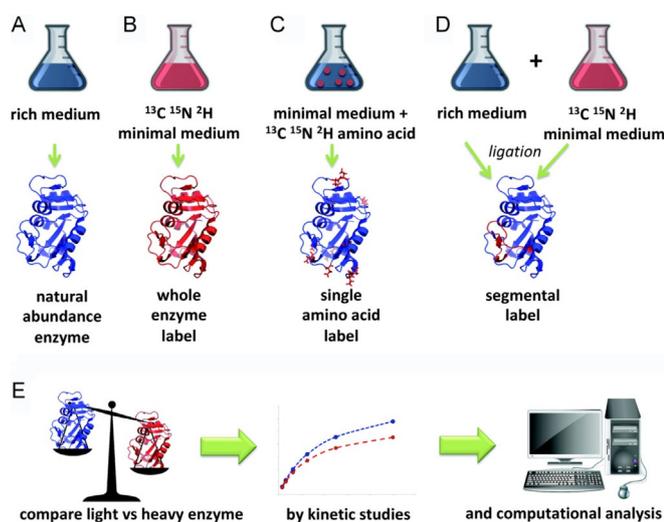
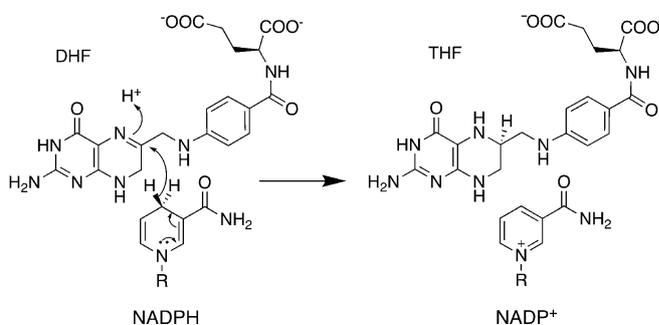


Figure 1. Strategies for the production and analysis of heavy enzymes. A) to D) show A) the production of natural-abundance enzyme, B) whole-enzyme isotope labelling, C) single-residue isotope labelling, and D) segmental isotope labelling by production of two peptides, only one of which is labelled with heavy isotopes, that are ligated together and refolded. E) How heavy enzymes are analysed by kinetics and computational analysis.

Dihydrofolate Reductase

Intensive studies on enzyme KIEs have been performed on dihydrofolate reductase (DHFR), including combined experimental and computational approaches to investigate different homologues and variants.^[12h,j,14a,16] DHFR catalyses hydride transfer from NADPH to tetrahydrofolate (THF, Scheme 1).^[17]



Scheme 1. Reaction catalysed by dihydrofolate reductase.

Because of the small mass of hydrogen, this reaction has a significant tunnelling component and is therefore particularly sensitive to changes in protein dynamics. The temperature dependence of the heavy enzyme KIE on hydride transfer has been investigated, by using pre-steady state kinetics, for DHFRs from organisms that have adapted to live at different temperatures.^[12h–j,16b] At physiological pH, DHFR from *Escherichia coli* (EcDHFR) shows an enzyme KIE on the hydride transfer rate of 0.93; it rises to 1.18 as the temperature is increased from 10 to 40 °C.^[12j] The DHFR from the psychrophilic *Moritella profunda* (MpDHFR) shows an enzyme KIE that rises from 1.07 at 5 °C to 1.45 at 30 °C.^[12j] The DHFR from the thermophile *Geobacillus* (formerly *Bacillus*) *stearothermophilus* shows the reverse trend, with the KIE falling from 1.65 at 5 °C to 1.09 at 45 °C.^[12h] In each case, the KIE approaches unity close to the physiological temperature of the host organism; this thus strongly suggests that dynamic coupling might be at a minimum around physiological temperature (Figure 2).^[12h–j,16b] Computational studies illustrate that the observed KIEs in DHFR are not due to hindered “promoting motions” but rather to increased re-

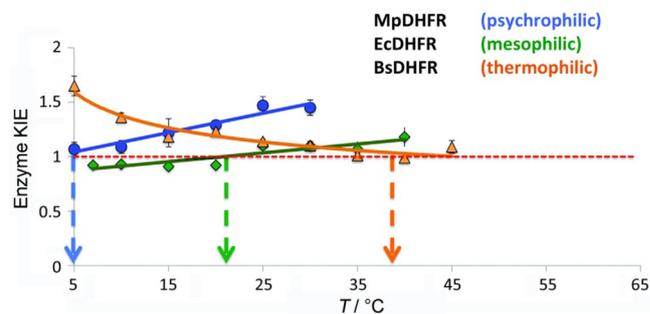


Figure 2. Temperature dependent pre-steady-state KIEs for different DHFRs at pH 7.0. Data show MpDHFR (●),^[12j] EcDHFR (◆)^[12j] and BsDHFR (▲).^[12h] The arrows indicate the temperature at which the KIE tends to unity for psychrophilic MpDHFR, mesophilic EcDHFR and thermophilic BsDHFR.

small nonconjugated molecules with values of k_{cat} ranging from 2 to 8 s^{-1} (Figure 3A). “Bad” substrates were bulky and highly conjugated, with k_{cat} values below 2 s^{-1} . Because hydride transfer is partially rate-limiting, the heavy enzyme KIE

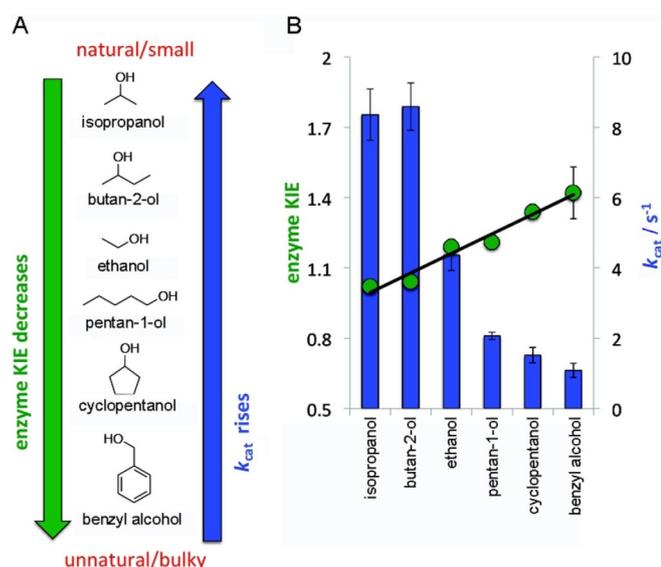


Figure 3. BsADH heavy enzyme KIE for a range of substrates correlated with k_{cat} at 20°C . A) Structures of substrates tested. B) Enzyme KIEs are shown in circles; k_{cat} values are shown as bars. Data from ref. [12k].

measurements were based on k_{cat} .^[24] No dynamic coupling was observed for any of the substrates at physiological temperature (40°C), at which enzyme KIEs were around unity. However, at lower temperature (20°C), the enzyme KIEs rose with inverse correlation to k_{cat} (Figure 3B). The absence of dynamic coupling at physiological temperature is consistent with earlier work on DHFR and shows that dynamics do not contribute to the reaction under physiological conditions.^[12h-j,16,19] The KIEs observed at lower temperature correlated with k_{cat} and confirmed the hypothesis that unnatural, bulky substrates require assistance from protein dynamics to produce a larger reorganisation of the active site.^[12k] As a result, slightly greater protein friction is generated on the chemical system along the evolution of the reaction coordinate. This means more protein movements (femtosecond mass-dependent protein motions) that can be coupled to the crossing of the transition state dividing surface, an event that occurs on a timescale of the same order of magnitude as the protein motions (femtoseconds).

Future Outlook for Protein Engineering

Our growing understanding of protein dynamics could help to engineer better enzymes in the future. Theoreticians have proposed that introducing a new “promoting motion” would improve the activity of aromatic amine dehydrogenase but this has never been tested experimentally.^[25] Similarly, work on PNP has shown that the efficiency of barrier crossing in a heavy enzyme can be modified by mutations that enhance promot-

ing vibrations.^[15] Experimentally, this resulted in the inversion of the enzyme KIE from a normal KIE of 1.31 to an inverse KIE of 0.75.^[15] However, the mutant enzyme was less catalytically efficient than the wild type. An alternative or complementary approach for engineering dynamics emerges from the study of DHFR and BsADH.^[12h-k,16,19] In these enzymes there are no “promoting motions” but rather protein dynamics are involved in active site reorganisation under non-physiological conditions or when poorly tolerated substrates are used. Identification and mutation of residues responsible for dynamic effects in BsADH could therefore provide a route towards the rational re-engineering of this enzyme for unnatural substrates. It is thus now necessary to locate the region of BsADH and the particular amino acid residues responsible for dynamic effects in order to test the hypothesis that these residues are hotspots for re-engineering of the enzyme’s substrate profile. A number of experimental and computational techniques for identifying residues responsible for dynamic effects exist; they include labelling of a single amino acid (a technique previously applied to PNP)^[13] or the production of hybrid isotopomers in which one particular loop or domain of a protein is isotopically labelled.^[14] Such hybrids can be constructed by using a variety of techniques including chemical ligation, in which a peptide containing a C-terminal thioester can be ligated to a peptide containing a free N-terminal cysteine residue. This technique has been applied to EcDHFR.^[14] Other approaches involve the use of peptide ligases^[26] or protein *trans* splicing with split inteins.^[27] The best approach for a particular protein has to be experimentally determined and often requires time-consuming optimisation of the ligation and subsequent refolding of the ligated chain. For the technique to become practical as a routine method for enzyme engineering, further advances need to be made in protein ligation and refolding technologies to enable easy construction of hybrid isotopomers. An alternative, complementary approach is to use computer simulations to predict the residues responsible for friction along the reaction coordinate.^[18]

It may be questioned whether the gains in catalytic efficiency from such engineering will be large, given the small enzyme KIEs observed. Although the enzyme KIEs appear small, there is a significant kinetic difference between “good” and “bad” substrates. Hence, mutation of residues that hinder the progression of the reaction can translate into a measurable increase in catalytic turnover. Nevertheless, because only a small number of enzymes have to date been studied by the heavy enzyme methodology, it is currently unclear which enzymes obey the rule of minimised dynamic coupling under physiological conditions. Hence investigations of protein dynamics over a broad range of enzyme families with different chemistries and cofactors are essential.

Summary

Heavy-isotope labelling of proteins combined with detailed computational work is a useful tool for studying the contribution of protein motions to the catalytic step. Studies on DHFR variants have shown that dynamic effects are only significant

under nonphysiological conditions that require reorganisation of the active site.^[12h–j,16,19] This leads to the hypothesis that dynamic coupling should also be increased when unnatural substrates are used because the active site architecture is not optimised for the corresponding chemical transformations.^[12k] By this argument, dynamic coupling indicates the extent to which an active site is suited to a particular substrate. This was confirmed through heavy enzyme studies on an alcohol dehydrogenase from *G. stearothermophilus*. Amino acid residues responsible for dynamic effects might be useful as targets for mutagenesis to create an active site optimally suited for a designed substrate. Unlike directed evolution, which requires a high-throughput screen to assay a large number of variants, such a rational approach based on insight into dynamic effects only requires a small number of mutants to be analysed.^[28]

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Conflict of Interest

The authors declare no conflict of interest.

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- T. Bugg, in *Introduction to Enzyme and Coenzyme Chemistry: Second Edition*, Blackwell Publishing, Ltd., Oxford, UK, **2004**, p. 8.
- R. Singh, M. Kumar, A. Mittal, P. K. Mehta, *3 Biotech* **2016**, *6*, 174–189.
- A. Tarczykowska, B. Kochański, K. Kaluźny, W. Zukow, *J. Educ. Health Sport* **2017**, *7*(9), 217–223.
- a) R. Zhang, Y. Xu, R. Xiao, *Biotechnol. Adv.* **2015**, *33*, 1671–1684; b) M. T. Reetz, *Chem. Rec.* **2016**, *16*, 2449–2459; c) M. Goldsmith, D. S. Tawfik, *Curr. Opin. Struct. Biol.* **2017**, *47*, 140–150.
- a) X. Huang, M. Garcia-Borràs, K. Miao, S. B. J. Kan, A. Zutshi, K. N. Houk, F. H. Arnold, *ACS Cent. Sci.* **2019**, *5*, 270–276; b) F. H. Arnold, *Angew. Chem. Int. Ed.* **2018**, *57*, 4143–4148; *Angew. Chem.* **2018**, *130*, 4212–4218; c) H. Renata, Z. J. Wang, F. H. Arnold, *Angew. Chem. Int. Ed.* **2015**, *54*, 3351–3367; *Angew. Chem.* **2015**, *127*, 3408–3426.
- a) R. Wolfenden, *Annu. Rev. Biophys. Bioeng.* **1976**, *5*, 271–306; b) A. J. Adamczyk, J. Cao, S. C. L. Kamerlin, A. Warshel, *Proc. Natl. Acad. Sci. USA* **2011**, *108*, 14115–14120; c) M. Garcia-Viloca, J. Gao, M. Karplus, D. G. Truhlar, *Science* **2004**, *303*, 186–195; d) J. Gao, S. Ma, D. T. Major, K. Nam, J. Pu, D. G. Truhlar, *Chem. Rev.* **2006**, *106*, 3188–3209; e) M. Roca, M. Oliva, R. Castillo, V. Moliner, I. Tuñón, *Chem. Eur. J.* **2010**, *16*, 11399–11411; f) L. C. Pauling, *Chem. Eng. News* **1946**, *24*, 1375–1377.
- a) P. Wentworth, K. D. Janda, *Cell Biochem. Biophys.* **2001**, *35*, 63–87; b) D. Hilvert, *Annu. Rev. Biochem.* **2000**, *69*, 751–793.
- a) S. D. Schwartz, V. L. Schramm, *Nat. Chem. Biol.* **2009**, *5*, 551–558; b) D. Antoniou, M. R. Abolfath, S. D. Schwartz, *J. Chem. Phys.* **2004**, *121*, 6442–6447; c) E. Z. Eisenmesser, O. Millet, W. Labeikovsky, D. M. Korzhnev, M. Wolf-Watz, D. A. Bosco, J. J. Skalicky, L. E. Kay, D. Kern, *Nature* **2005**, *438*, 117–121; d) P. K. Agarwal, S. R. Billeter, P. T. R. Rajagopalan, S. J. Benkovic, S. Hammes-Schiffer, *Proc. Natl. Acad. Sci. USA* **2002**, *99*, 2794–2799; e) S. Caratzoulas, J. S. Mincer, S. D. Schwartz, *J. Am. Chem. Soc.* **2002**, *124*, 3270–3276.
- a) K. Świderek, J. Javier Ruiz-Pernía, V. Moliner, I. Tuñón, *Curr. Opin. Chem. Biol.* **2014**, *21*, 11–18; b) E. J. Loveridge, E. M. Behiry, J. Guo, R. K. Allemann, *Nat. Chem.* **2012**, *4*, 292–297; c) A. V. Pislakov, J. Cao, S. C. L. Kamerlin, A. Warshel, *Proc. Natl. Acad. Sci. USA* **2009**, *106*, 17359–17364; d) M. H. M. Olsson, W. W. Parson, A. Warshel, *Chem. Rev.* **2006**, *106*, 1737–1756; e) M. Roca, V. Moliner, I. Tuñón, J. T. Hynes, *J. Am. Chem. Soc.* **2006**, *128*, 6186–6193; f) J. J. Ruiz-Pernía, I. Tuñón, V. Moliner, J. T. Hynes, M. Roca, *J. Am. Chem. Soc.* **2008**, *130*, 7477–7488.
- S. Rokop, L. Gajda, S. Parmeter, H. L. Crespi, J. J. Katz, *Biochim. Biophys. Acta Enzymol.* **1969**, *191*, 707–715.
- a) R. G. Silva, A. S. Murkin, V. L. Schramm, *Proc. Natl. Acad. Sci. USA* **2011**, *108*, 18661–18665; b) D. R. Kipp, R. G. Silva, V. L. Schramm, *J. Am. Chem. Soc.* **2011**, *133*, 19358–19361.
- a) D. Antoniou, X. Ge, V. L. Schramm, S. D. Schwartz, *J. Phys. Chem. Lett.* **2012**, *3*, 3538–3544; b) Z. Wang, P. Singh, C. M. Czekster, A. Kohen, V. L. Schramm, *J. Am. Chem. Soc.* **2014**, *136*, 8333–8341; c) M. D. Toney, J. N. Castro, T. A. Addington, *J. Am. Chem. Soc.* **2013**, *135*, 2509–2511; d) J. E. Longbotham, S. J. O. Hardman, S. Görlich, N. S. Scrutton, S. Hay, *J. Am. Chem. Soc.* **2016**, *138*, 13693–13699; e) C. R. Pudney, A. Guerriero, N. J. Baxter, L. O. Johannissen, J. P. Waltho, S. Hay, N. S. Scrutton, *J. Am. Chem. Soc.* **2013**, *135*, 2512–2517; f) C. Ranasinghe, Q. Guo, P. J. Sapienza, A. L. Lee, D. M. Quinn, C. M. Cheatum, A. Kohen, *J. Am. Chem. Soc.* **2017**, *139*, 17405–17413; g) Z. Wang, E. P. Chang, V. L. Schramm, *J. Am. Chem. Soc.* **2016**, *138*, 15004–15010; h) L. Y. P. Luk, J. J. Ruiz-Pernía, W. M. Dawson, E. J. Loveridge, I. Tuñón, V. Moliner, R. K. Allemann, *J. Am. Chem. Soc.* **2014**, *136*, 17317–17323; i) J. J. Ruiz-Pernía, E. Behiry, L. Y. P. Luk, E. J. Loveridge, I. Tuñón, V. Moliner, R. K. Allemann, *Chem. Sci.* **2016**, *7*, 3248–3255; j) L. Y. P. Luk, J. Javier Ruiz-Pernía, W. M. Dawson, M. Roca, E. J. Loveridge, D. R. Glowacki, J. N. Harvey, A. J. Mulholland, I. Tuñón, V. Moliner, R. K. Allemann, *Proc. Natl. Acad. Sci. USA* **2013**, *110*, 16344–16349; k) E. Behiry, J. J. Ruiz-Pernía, L. Luk, I. Tuñón, V. Moliner, R. K. Allemann, *Angew. Chem. Int. Ed.* **2018**, *57*, 3128–3131; *Angew. Chem.* **2018**, *130*, 3182–3185.
- a) J. Suarez, V. L. Schramm, *Proc. Natl. Acad. Sci. USA* **2015**, *112*, 11247–11251; b) R. K. Harijan, I. Zoi, D. Antoniou, S. D. Schwartz, V. L. Schramm, *Proc. Natl. Acad. Sci. USA* **2018**, *115*, E6209–E6216.
- a) L. Y. P. Luk, J. J. Ruiz-Pernía, A. S. Adesina, E. J. Loveridge, I. Tuñón, V. Moliner, R. K. Allemann, *Angew. Chem. Int. Ed.* **2015**, *54*, 9016–9020; *Angew. Chem.* **2015**, *127*, 9144–9148; b) A. F. Scott, L. Y. P. Luk, R. K. Allemann, *Methods Enzymol.* **2017**, *596*, 23–41.
- R. K. Harijan, I. Zoi, D. Antoniou, S. D. Schwartz, V. L. Schramm, *Proc. Natl. Acad. Sci. USA* **2017**, *114*, 6456–6461.
- a) J. J. Ruiz-Pernía, L. Y. P. Luk, R. García-Meseguer, S. Martí, E. J. Loveridge, I. Tuñón, V. Moliner, R. K. Allemann, *J. Am. Chem. Soc.* **2013**, *135*, 18689–18696; b) L. Y. P. Luk, E. J. Loveridge, R. K. Allemann, *J. Am. Chem. Soc.* **2014**, *136*, 6862–6865.
- S. A. Kholodar, A. K. Ghosh, A. Kohen, *Methods Enzymol.* **2017**, *596*, 43–83.
- a) A. Krzemińska, V. Moliner, K. Świderek, *J. Am. Chem. Soc.* **2016**, *138*, 16283–16298; b) M. Roca, V. Moliner, I. Tuñón, *ACS Catal.* **2018**, *8*, 815–827.
- L. Y. P. Luk, E. J. Loveridge, R. K. Allemann, *Phys. Chem. Chem. Phys.* **2015**, *17*, 30817–30827.
- K. Zinovjev, I. Tuñón, *Proc. Natl. Acad. Sci. USA* **2017**, *114*, 12390–12395.
- Z. Wang, D. Antoniou, S. D. Schwartz, V. L. Schramm, *Biochemistry* **2016**, *55*, 157–166.
- a) H. Radianingtyas, P. C. Wright, *FEMS Microbiol. Rev.* **2003**, *27*, 593–616; b) A. Guagliardi, M. Martino, I. Iaccarino, M. D. Rosa, M. Rossi, S. Bartolucci, *Int. J. Biochem. Cell Biol.* **1996**, *28*, 239–246; c) Z.-X. Liang, T. Lee, K. A. Resing, N. G. Ahn, J. P. Klinman, *Proc. Natl. Acad. Sci. USA* **2004**, *101*, 9556–9561.
- A. Pennacchio, M. Rossi, C. A. Raia, *Appl. Biochem. Biotechnol.* **2013**, *170*, 1482–1490.
- A. Kohen, R. Cannio, S. Bartolucci, J. P. Klinman, *Nature* **1999**, *399*, 496–499.
- I. Zoi, D. Antoniou, S. D. Schwartz, *J. Phys. Chem. B* **2017**, *121*, 7290–7298.
- K. M. Mikula, I. Tascón, J. J. Tommilla, H. Iwaï, *FEBS Lett.* **2017**, *591*, 1285–1294.

[27] a) Y. Minato, T. Ueda, A. Machiyama, I. Shimada, H. Iwai, *J. Biomol. NMR* **2012**, *53*, 191–207; b) A. Ciragan, A. S. Aranko, I. Tascon, H. Iwai, *J. Mol. Biol.* **2016**, *428*, 4573–4588; c) A. E. L. Busche, A. S. Aranko, M. Talebzadeh-Farooji, F. Bernhard, V. Dötsch, H. Iwai, *Angew. Chem. Int. Ed.* **2009**, *48*, 6128–6131; *Angew. Chem.* **2009**, *121*, 6244–6247.

[28] H. Xiao, Z. Bao, H. Zhao, *Ind. Eng. Chem. Res.* **2015**, *54*, 4011–4020.

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