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## Data Article

## Data on binding of L-tryptophan and bovine serum albumin by novel gold nanoparticles capped with amphiphilic sulfonatomethylated calixresorcinarenes

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## ABSTRACT

The data provided in this paper are associated with the data in the «Binding of L-tryptophan and bovine serum albumin by novel gold nanoparticles capped with amphiphilic sulfonatomethylated calixresorcinarenes» paper (Shalaeva et al., 2019). Here, we represent i) the characterization data of calixresorcinarenes capped gold nanoparticles obtained by TEM and Vis- and IR spectroscopy; ii) the data giving the information about the interaction of modified AuNPs with L-tryptophan and bovine serum albumin by dynamic light scattering, spectrophotometry and fluorescence spectroscopy.

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E-mail address: [shalaeva@iopc.ru](mailto:shalaeva@iopc.ru) (Y.V. Shalaeva).<https://doi.org/10.1016/j.dib.2019.104241>2352-3409/© 2019 The Author(s). Published by Elsevier Inc. This is an open access article under the CC BY license (<http://creativecommons.org/licenses/by/4.0/>).

## Specifications table

Subject area	Chemistry
More specific subject area	Nanoparticles, calixresorcinarenes, protein
Type of data	Tables, microscopy images, graphs
How data was acquired	Spectrophotometry: Perkin Elmer Instruments TEM: Libra 120 IR: Bruker Vector 22 Fluorimetry: Varian Cary Eclipse DLS: Zetasizer Nano instrument
Data format	Raw, analyzed, processed
Experimental factors	The nanoparticles were synthesized using calixresorcinarenes both as reducing and stabilizing agents in an aqueous solution at 25 °C in the air atmosphere with stirring during 2 hours.
Experimental features	Optical properties and sizes of nanoparticles, IR data, emission spectra of L-tryptophan and bovine serum albumin
Data source location	Arbuzov Institute of Organic and Physical Chemistry, FRC Kazan Scientific Center of RAS, Kazan, Russian Federation Kazan Federal University, Kazan, Russian Federation
Data accessibility	Data are available within this article
<b>Related research article</b>	Ya.V. Shalaeva, Ju. E. Morozova, A. M. Shumatbaeva, I. R. Nizameev, M. K. Kadirov, I.S. Antipin, A. I. Konovalov, Binding of L-tryptophan and bovine serum albumin by novel gold nanoparticles capped with amphiphilic sulfonatomethylated calixresorcinarenes, <i>J. Mol. Liq.</i> , 286 (2019) 110879. <a href="https://doi.org/10.1016/j.molliq.2019.110879">https://doi.org/10.1016/j.molliq.2019.110879</a>

**Value of the data**

- The full study of multicomponent nanosized systems on the base of gold nanoparticles stabilized by supramolecular macrocycles, such as, sulfonatomethylated calixresorcinarenes, capable of multiple intermolecular interactions with a bovine serum albumin with the formation of «macrocycle-protein» complexes and the creation of cooperative supra-molecular assemblies with a protein in an aqueous solution is demonstrated.
- Data obtained could help scientists who investigate the interactions of proteins with synthetic compounds to find out binding and recognition possibilities, any structural and functional changes in protein during the binding in particular under various external conditions and stability of proteins.
- The TEM images, DLS, spectrophotometry and fluorimetry data presented here confirmed the formation of hybrid systems on the base of gold nanoparticles and amphiphilic sulfonatomethylated calixresorcinarenes with controlled size, aggregation, electro-optical and binding properties and could be useful for the understanding of the interaction mechanisms between protein and nanoparticles, give information about structural and functional changes of BSA and the nanoparticles.
- This investigation has a value due to the promising possibility to use such systems for protein transport and visualization in biological media.

**1. Data**

This article contains the data about the synthesis of gold nanoparticles using sulfonatomethylated calixresorcinarenes with methyl (**C1S**) and pentyl (**C5S**) substituents on the lower rim both as reducing and stabilizing agents in an aqueous solution at different component ratio and structural characteristic (Table 2, IR data) of **C1S** and **C5S** macrocycles and obtained Au@**C1S** and Au@**C5S** nanoparticles. The values of averaged hydrodynamic diameters ( $d$ , nm) of Au@**C1S** and Au@**C5S** nanoparticles and their associates with **Trp** and BSA molecules are presented in Table 3, Table 4, Fig. 4 and Fig. 9. Fig. 3 illustrates the TEM images of Au@**C1S** and Au@**C5S** nanoparticles. There is also information about the spectral characteristic of nanoparticles and their changes during the interactions with **Trp** and BSA molecules (Table 1, Fig. 1 and Figs. 5–8). Also, the data presented in this paper illustrate the stability of Au@**C1S** and Au@**C5S** nanoparticles during the time (Fig. 2). Fig. 10 demonstrates the stoichiometry of binding of BSA with Au@**C5S** (Job's plot).

**Table 1**

The values of the wavelength corresponding to the SPR maximum absorption intensity of gold nanoparticles ( $\lambda_{\max}$ , nm) modified by **C1S** and **C5S** macrocycles in an aqueous solution.

System	$C_{\text{calix}}/C_{\text{HAuCl}_4}$ , mM	$\lambda_{\max}$ , nm
		fresh
<b>C1S</b> + HAuCl <sub>4</sub>	1.5/0.5	529
	1/0.5	529
	0.5/0.5	529
<b>C5S</b> + HAuCl <sub>4</sub>	0.1/0.5	552
	2/0.5	535
	1/0.5	533
	1.5/0.5	531
	0.5/0.5	532
	0.1/0.5	545

**Table 2**

IR data of **C1S**, Au@**C1S**, **C5S** and Au@**C5S** nanoparticles (in KBr pallets, at macrocycle/HAuCl<sub>4</sub> ratio of 0.5/0.5, mM).

System	$\nu_{\text{OH}}$ , cm <sup>-1</sup>	$\nu_{\text{C=O}}$ , cm <sup>-1</sup>	$\nu_{\text{C=C}}$ , cm <sup>-1</sup>	$\nu_{\text{C-O}}$ , cm <sup>-1</sup>	$\nu_{\text{as S=O}}$ , cm <sup>-1</sup>	$\nu_{\text{s S=O}}$ , cm <sup>-1</sup>
<b>C1S</b>	3438	–	1609	1476	1193	1044
Au@ <b>C1S</b>	3212	1718	1618	1476	1189	1042
<b>C5S</b>	3431	–	1610	1472	1193	1047
Au@ <b>C5S</b>	3423	1714	1627	1468	1209	1043

**Table 3**

The values of averaged hydrodynamic diameters of Au@**C1S**, «Au@**C1S** + **Trp**», Au@**C5S**, «Au@**C5S** + **Trp**» nanoparticles ( $d$ , nm) in an aqueous solutions, their intensities of scattering ( $I$ , %), polydispersity index ( $PDI$ ) obtained by dynamic light scattering method.

System	$C_{\text{calix}}/C_{\text{HAuCl}_4}$ , mM	$d$ , nm ( $I$ , %)	$PDI$
Au@ <b>C1S</b>	0.1/0.5	91 (22)	0.169
Au@ <b>C1S</b>	0.5/0.5	7 (4), 68 (15)	0.431
Au@ <b>C1S</b> + <b>Trp</b>	0.5/0.5	105 (13)	0.479
Au@ <b>C1S</b> + <b>Trp</b> (pH 9.4)	0.5/0.5	78 (15)	0.415
Au@ <b>C5S</b>	0.1/0.5	79 (15)	0.253
Au@ <b>C5S</b>	0.5/0.5	9 (2), 59 (18)	0.350
Au@ <b>C5S</b> + <b>Trp</b>	0.5/0.5	79 (14)	0.292
Au@ <b>C5S</b> + <b>Trp</b> (pH 9.5)	0.5/0.5	79 (13)	0.415

The error of the hydrodynamic particle size determination was <2%.

## 2. Experimental design, materials and methods

### 2.1. Materials

To synthesize the Au@**C1S** and Au@**C5S** nanoparticles we used the tetramethylsulfonated calixresorcinarenes with methyl (**C1S**) and pentyl (**C5S**) substitutes on the lower rim, which were obtained according to the previously reported procedure [2]. L-Tryptophan (**Trp**) and BSA from Sigma-Aldrich (Moscow, Russia) without any additional purification was obtained. HAuCl<sub>4</sub>·4H<sub>2</sub>O was kindly provided by Prof. E. Kh. Kazakova. All experiments were done in deionized water (3.5 μOm/cm).

### 2.2. Common synthetic procedure of Au@**C1S** and Au@**C5S** nanoparticles

To the aliquot of **CS** aqueous solution ( $C_{\text{CS}}$ , mM;  $V_{\text{CS}}$ , ml), deionized water ( $V$ , ml) and aliquot of HAuCl<sub>4</sub> aqueous solution ( $C_{\text{Au}}$ , mM;  $V_{\text{Au}}$ , ml) were vigorously stirred at 25 °C. For 2 hours the reaction was complete finished that were confirmed by spectrophotometry method (Table 5). 2 ml of solution with final concentration of the components  $C_{\text{CS}}/C_{\text{HAuCl}_4}$ , mM was obtained.

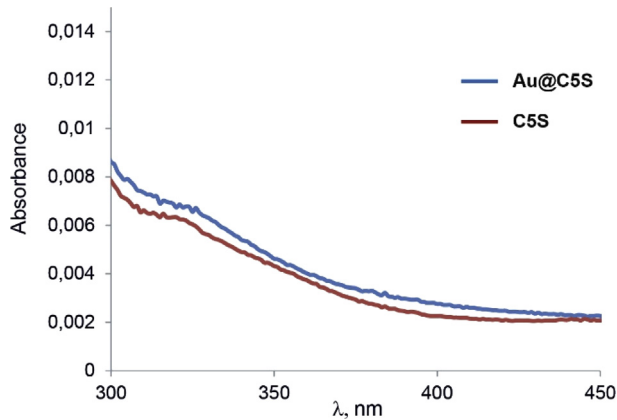
**Table 4**

The values of averaged hydrodynamic diameters of Au@C5S+BSA nanoparticles ( $d$ , nm) in an aqueous and PBS buffered solutions (pH = 7.4), their intensities of scattering ( $I$ , %), polydispersity index (PDI) obtained by dynamic light scattering method (fresh, after 2 and 7 days) and pH values of solutions.

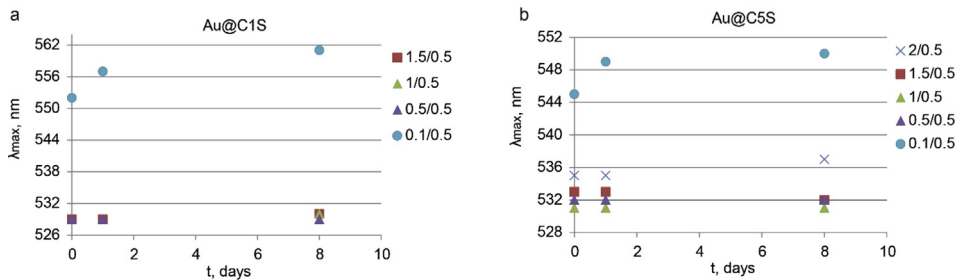
System	pH	$C_{NPS}$ , mM	$C_{BSA}$ , $\mu$ M	fresh		2 days		7 days	
				$d$ , nm ( $I$ , %)	PDI	$d$ , nm ( $I$ , %)	PDI	$d$ , nm ( $I$ , %)	PDI
Au@C5S + BSA		0.5/0.5	0	9 (2), 59 (18)	0.226	9 (2), 59 (18)		9 (2), 59 (18)	
	5.59	0.5/0.5	2.5	68 (22)	0.270	91 (21)	0.191	91 (20)	0.208
	5.65	0.5/0.5	5	79 (20)	0.183	91 (25)	0.177	91 (19)	0.231
	6.26	0.5/0.5	10	91 (18)	0.255	91 (21)	0.182	91 (19)	0.182
	6.54	0.5/0.5	50	91 (20)	0.164	91 (21)	0.174	91 (21)	0.181
	6.76	0.5/0.5	100	91 (18)	0.166	91 (18)	0.181	91 (21)	0.179
Au@C5S + BSA	7.4	0.25/0.25	0	16 (1), 79 (19)	0.228	16 (1), 79 (19)		16 (1), 79 (19)	
	7.4	0.25/0.25	2.5	21 (2), 91 (15)	0.182	14 (2), 91 (20)	0.206	91 (22)	0.193
	7.4	0.25/0.25	10	91 (22)	0.243	91 (21)	0.184	91 (18)	0.198
	7.4	0.25/0.25	50	106 (20)	0.197	106 (19)	0.190	106 (19)	0.208
	7.4	0.25/0.25	100	106 (15)	0.228	106 (21)	0.198	106 (22)	0.243

The error of the hydrodynamic particle size determination was <2%.

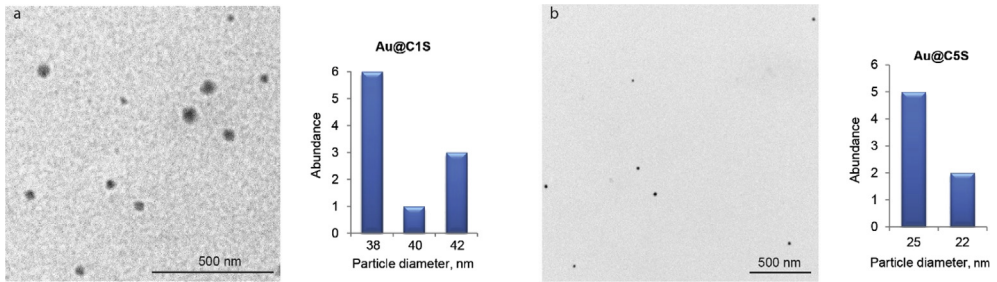
<sup>a</sup> – precipitation.



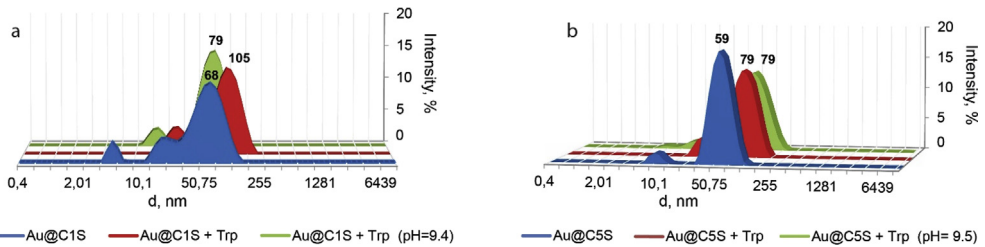
**Fig. 1.** UV–Vis-spectra of aqueous solutions, containing C5S and Au@C5S, respectively ( $C = 10 \mu$ M).



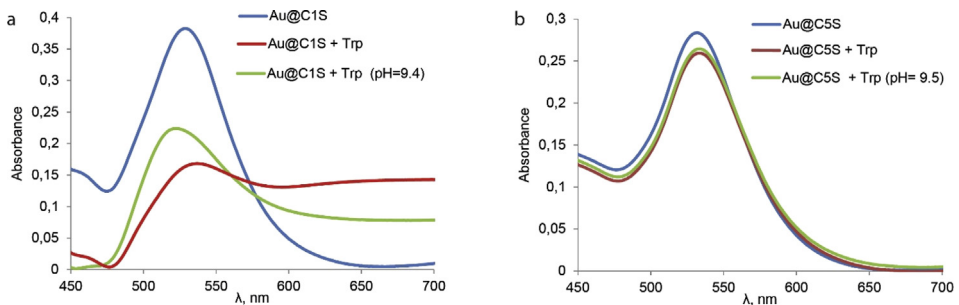
**Fig. 2.** Time dependence of SPR maximum adsorption intensity of gold ( $\lambda_{max}$ , nm) for Au@C1S (a) and Au@C5S (b).



**Fig. 3.** TEM images and histograms of size distribution of Au@C1S (a) and Au@C5S (b) nanoparticles (scale: 500 nm).



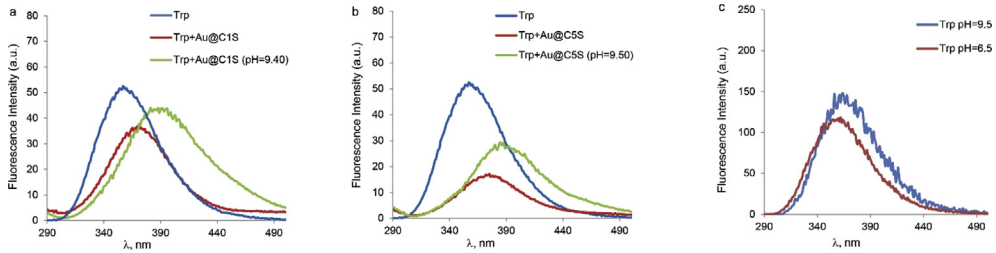
**Fig. 4.** The intensity-averaged size distribution for aqueous solutions, containing (a) Au@C1S, Au@C1S + Trp (pH = 2.41) and Au@C1S + Trp (pH = 9.4) and (b) Au@C5S, Au@C5S + Trp (pH = 2.53) and Au@C5S + Trp (pH = 9.5) at 25 °C.



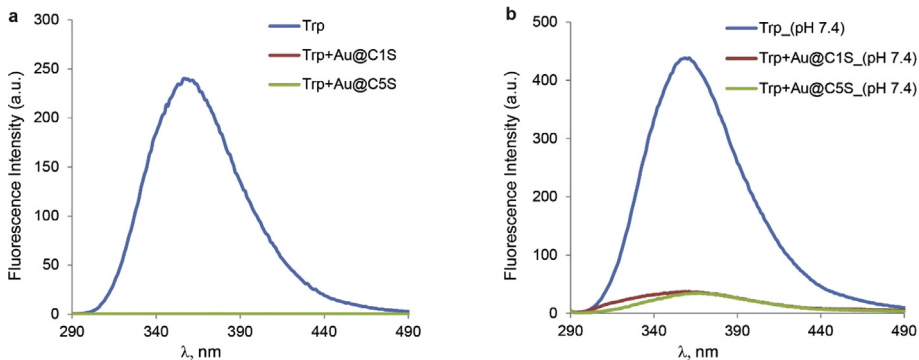
**Fig. 5.** Vis-spectra of aqueous solutions, containing (a) Au@C1S, Au@C1S + Trp (pH = 2.41) and Au@C1S + Trp (pH = 9.4) and (b) Au@C5S, Au@C5S + Trp (pH = 2.53) and Au@C5S + Trp (pH = 9.5) at 25 °C (0.2 cm quartz path length cuvettes).

### 2.3. Methods

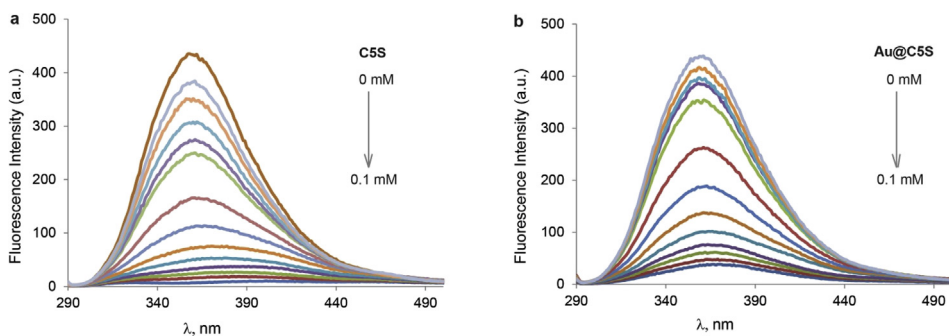
**Uv–vis spectra** were recorded on Lambda 35 UV–vis spectrometer (PerkinElmer Instruments, Shelton, CT, USA) in 0.2 cm quartz path length cuvettes with optical background correction. **Transmission electron microscopy (TEM)** images were obtained with Libra 120 (Carl Zeiss). The images were acquired at an accelerating voltage of 100 kV. Samples were dispersed on 300 mesh copper grids with continuous carbon-formvar support films. **IR-spectra** were recorded with Bruker Vector 22 FTIR Spectrometer (Bruker, Karlsruhe, Germany) with the wavelength range of 4000–400  $\text{cm}^{-1}$ . **pH** of aqueous solutions was measured at 25 °C with Thermo pH-meter (Thermo Electron, USA). **Dynamic light scattering (DLS)** measurements were carried out on Zetasizer Nano instrument (Malvern Instruments, USA) with 10 mW 633 nm He–Ne laser light source and the light scattering angle of 173°.



**Fig. 6.** Fluorescence emission spectra of (a) **Trp** (0.1 mM), **Trp** in the presence of **C1S@Au** (0.5/0.5, mM) at spontaneous pH and at pH = 9.40; (b) **Trp** (0.1 mM), **Trp** in the presence of **C5S@Au** (0.5/0.5, mM) at spontaneous pH and at pH = 9.50; (c) **Trp** (0.1 mM) at spontaneous pH and at pH = 9.50; Ex. and Em. slits for **Trp** and **Trp** in the presence of nanoparticles have different values for clarity,  $V = 600$  V.

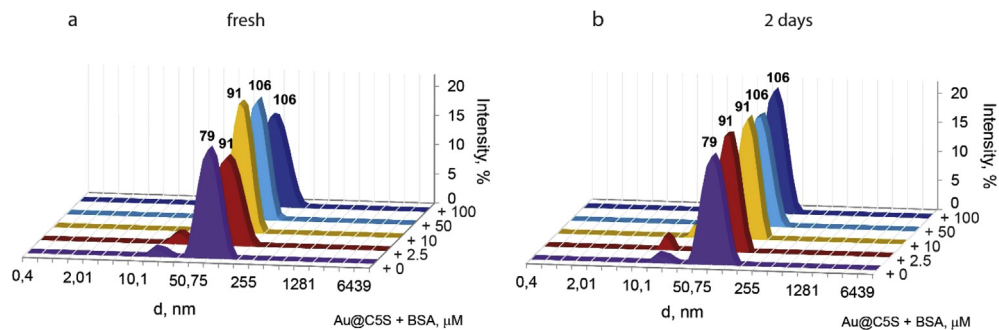


**Fig. 7.** Fluorescence emission spectra of (a) **Trp** (0.1 mM), **Trp** in the presence of **Au@C1S** (0.5/0.5, mM) and **Au@C5S** (0.5/0.5, mM) at spontaneous pH (Ex. and Em. slits are 5 and 2.5 nm, respectively,  $V = 600$  V); (b) **Trp** (0.01 mM), **Trp** in the presence of **Au@C1S** (0.1/0.1, mM) and **Au@C5S** (0.1/0.1, mM) in phosphate buffer, pH 7.4 (Ex. and Em. slits are 10 and 5 nm, respectively,  $V = 600$  V).

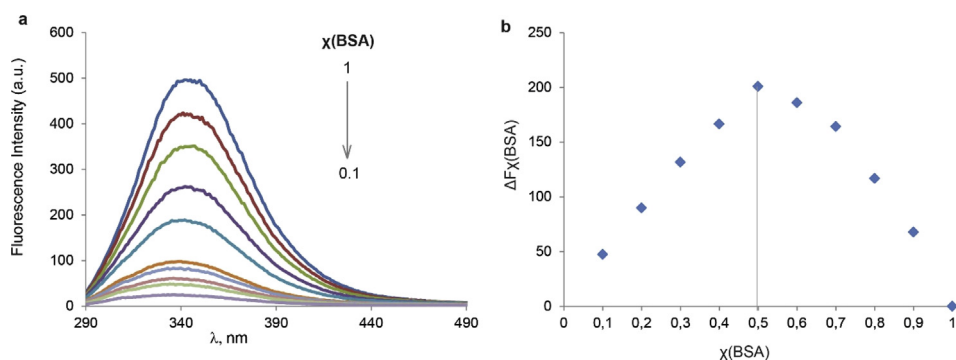


**Fig. 8.** Fluorescence emission spectra of **Trp** (10  $\mu$ M) in the absence and in the presence of **C5S** (from 0.001 to 0.1 mM) (a) and **Au@C5S** (from 0.001 to 0.1 mM) (b) (pH 7.4).

**Emission spectra** of L-tryptophan (**Trp**) and L-tryptophan-residues of BSA molecule were recorded on Varian Cary Eclipse spectrofluorometer (Agilent Technologies company production, USA) with the excited wave length at 279 nm using 1 cm path length quartz cuvettes at 25 °C.



**Fig. 9.** The intensity-averaged size distribution for aqueous solutions, containing Au@C5S (0.25/0.25, mM) in the presence of different amount of BSA (from 0 to 100  $\mu\text{M}$ ) at 25  $^{\circ}\text{C}$ : fresh (a) and in 2 days (b) (pH 7.4). For unbuffered solutions see [1].



**Fig. 10.** Fluorescence emission spectra for the Job's plot of BSA - Au@C5S systems (a) and Job's plot for the determination of stoichiometry of binding of BSA with Au@C5S (b), total concentration is 0.01 mM (pH 7.4).

**Table 5**

Amounts of components during the synthesis of Au@C1S and Au@C5S nanoparticles.

Macrocycle	$C_{CS}$ , mM	$V_{CS}$ , ml	$V$ , ml	$C_{Au}$ , mM	$V_{Au}$ , ml	$C_{CS}/C_{HAuCl_4}$ , mM
<b>C1S</b>	4	0.75	1.123	7.87	0.127	1.5/0.5
	4	0.5	1.373	7.87	0.127	1/0.5
	4	0.25	1.623	7.87	0.127	0.5/0.5
	4	0.05	1.823	7.87	0.127	0.1/0.5
<b>C5S</b>	4	1	0.873	7.87	0.127	2/0.5
	4	0.75	1.123	7.87	0.127	1/0.5
	4	0.5	1.373	7.87	0.127	1.5/0.5
	4	0.25	1.623	7.87	0.127	0.5/0.5
	4	0.05	1.823	7.87	0.127	0.1/0.5

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sulfonatomethylated calixresorcinarenes and gold nanoparticles were carried out by Ya.V. Shalaeva, Ju. E. Morozova and I.S. Antipin). The authors gratefully acknowledge the Assigned Spectral-Analytical Center of FRC Kazan Scientific Center of RAS).

### **Conflict of interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

### **Appendix A. Supplementary data**

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.dib.2019.104241>.

### **References**

- [1] Ya.V. Shalaeva, Ju. E. Morozova, A.M. Shumatbaeva, I.R. Nizameev, M.K. Kadirov, I.S. Antipin, A.I. Kononov, Binding of L-tryptophan and bovine serum albumin by novel gold nanoparticles capped with amphiphilic sulfonatomethylated calixresorcinarenes, *J. Mol. Liq.* 286 (2019), 110879, <https://doi.org/10.1016/j.molliq.2019.110879>.
- [2] E.Kh Kazakova, N.A. Makarova, A.Yu Ziganshina, L.A. Muslinkina, A.A. Muslinkin, W.D. Habicher, Novel water-soluble tetrasulfonatomethyl calixresorcinarenes, *Tetrahedron Lett.* 41 (2000) 10111–10115. [https://doi.org/10.1016/S0040-4039\(00\)01798-6](https://doi.org/10.1016/S0040-4039(00)01798-6).