

Article

# Fluorescence Spectroscopic Investigation of Competitive Interactions between Quercetin and Aflatoxin B<sub>1</sub> for Binding to Human Serum Albumin

Hongxia Tan <sup>1</sup>, Lu Chen <sup>1</sup>, Liang Ma <sup>1,2,\*</sup>, Shuang Liu <sup>1</sup>, Hongyuan Zhou <sup>1</sup> , Yuhao Zhang <sup>1,2</sup>, Ting Guo <sup>1</sup>, Wei Liu <sup>1</sup>, Hongjie Dai <sup>1</sup> and Yong Yu <sup>1</sup>

<sup>1</sup> College of Food Science, Southwest University, Chongqing 400715, China; tanhongxia@mail.swu.edu.cn (H.T.); chenluch@email.swu.edu.cn (L.C.); liushuang9995@stu.ouc.edu.cn (S.L.); zhouhy@swu.edu.cn (H.Z.); zhy1203@swu.edu.cn (Y.Z.); guoting06@swu.edu.cn (T.G.); lwissue@email.swu.cn (W.L.); daihongjie@swu.edu.cn (H.D.); yuyong@swu.edu.cn (Y.Y.)

<sup>2</sup> Biological Science Research Center, Southwest University, Chongqing 400715, China

\* Correspondence: zhyhml@swu.edu.cn; Tel.: +86-131-0128-2977

Received: 28 February 2019; Accepted: 3 April 2019; Published: 9 April 2019



**Abstract:** Aflatoxin B<sub>1</sub> (AFB<sub>1</sub>) is a highly toxic mycotoxin found worldwide in cereals, food, and animal feeds. AFB<sub>1</sub> binds to human serum albumin (HSA) with high affinity. In previous experiments, it has been revealed that reducing the binding rate of AFB<sub>1</sub> with HSA could speed up the elimination rate of AFB<sub>1</sub>. Therefore, we examined the ability of quercetin to compete with AFB<sub>1</sub> for binding HSA by fluorescence spectroscopy, synchronous spectroscopy, ultrafiltration studies, etc. It was shown that AFB<sub>1</sub> and quercetin bind to HSA in the same Sudlow site I (subdomain IIA), and the binding constant (K<sub>a</sub>) of the quercetin-HSA complex is significantly stronger than the complex of AFB<sub>1</sub>-HSA. Our data in this experiment showed that quercetin is able to remove the AFB<sub>1</sub> from HSA and reduce its bound fraction. This exploratory work may be of significance for studies in the future regarding decreasing its bound fraction and then increasing its elimination rate for detoxification. This exploratory study may initiate future epidemiological research designs to obtain further in vivo evidence of the long-term (potential protective) effects of competing substances on human patients.

**Keywords:** aflatoxin B<sub>1</sub>; quercetin; human serum albumin; competitive interaction; fluorescence spectroscopy

**Key Contribution:** This is the first time evidence that the addition of quercetin can remove AFB<sub>1</sub> from HSA and reduce the bound fraction of AFB<sub>1</sub> has been illustrated.

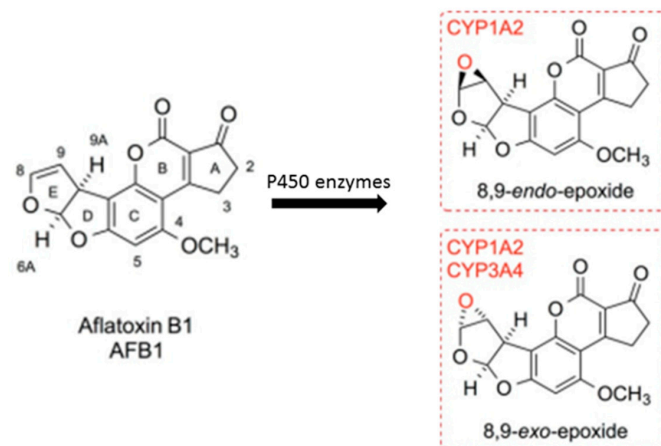
## 1. Introduction

Aflatoxin B<sub>1</sub> (AFB<sub>1</sub>), a kind of carcinogenic toxin, which had been classified as a group I carcinogen [1] by the International Agency for Research on Cancer, is often found in agricultural products, feed, and food such as grain and oil through various channels. According to the statistics, an estimated five billion people worldwide are exposed to high levels AFB<sub>1</sub>, and human health is in serious dangers [2–5]. How to reduce or degrade the toxicity of AFB<sub>1</sub> in agricultural production is one of the key research directions in the food safety field. Many techniques for reducing toxicity or degrading the structure of AFB<sub>1</sub> have been researched and reported. For instance, the toxicity of AFB<sub>1</sub> can be reduced by vitamin E through lowering the activities of plasma aspartate aminotransferase, alanine aminotransferase, and alkyne phosphatase [6]. A macromolecular complex was formed by β-D-glucan with AFB<sub>1</sub> in vivo to attenuate its toxicity [7]. The toxicity of AFB<sub>1</sub> could be decreased by

papaya extract through decreasing the oxidative stress reaction [8]. Most detoxification in the reports was taken as the effects after the toxin reacted with the target organ [9–12]. However, there is still a very high risk of damaging the target organ by AFB<sub>1</sub>. Therefore, it is of great significance to study the detoxification to AFB<sub>1</sub> during the transport process of AFB<sub>1</sub> in vivo.

Most xenobiotics will bind and be transported by serum albumin before they react with the target organ in the blood circulation in vivo [13]. AFB<sub>1</sub> follows a similar behavior as serum albumin in vivo, and many types of DNA damage can be induced by 8,9-epoxy AFB<sub>1</sub>, which is produced in the metabolic process of AFB<sub>1</sub> through the catalytic action of the cytochrome P450 enzyme in the target organ (Figure 1) [14]. Recently, it has been reported that the time of the toxin molecules binding to serum albumin could affect the half-life of the toxin in the organism [15–17]. Koszegi et al. [18] reported that some flavonoids such as quercetin could compete with and displace ochratoxin A (OTA) from human serum albumin (HSA). After the competitive interaction between flavonoids and OTA to bind with HSA, it was found that the bound fraction of OTA was decreased, and then, its half-life was decreased [18]. Hence, it is speculated that the toxicity of AFB<sub>1</sub> to target organs will be reduced and its elimination rate will be accelerated during the bio-transportation process due to the broken interaction, or its displacement, between AFB<sub>1</sub> and HSA. It has been reported that quercetin had high affinity to HSA in the pH 7.4 condition in vitro, and the primary binding site on albumin was the same as that of the AFB<sub>1</sub>-HSA complex (subdomain IIA) [19–23]. It has been reported that quercetin is a highly-active flavonoid that regulates the uptake of toxins to reduce body damage [24]. Therefore, quercetin was thought to be a very effective competitor in the binding of AFB<sub>1</sub> to HSA.

In this proposal, the effect of quercetin on the AFB<sub>1</sub>-HSA complex was studied by fluorescence spectroscopy, synchronous spectroscopy, ultrafiltration studies, etc. The aim is to explore the ability of quercetin to bind competitively to HSA with AFB<sub>1</sub>. Our results showed that quercetin is able to remove the AFB<sub>1</sub> from HSA, and this is the first time this molecular displacement has been described as a new type of interaction of quercetin affecting the biological behavior of AFB<sub>1</sub>.



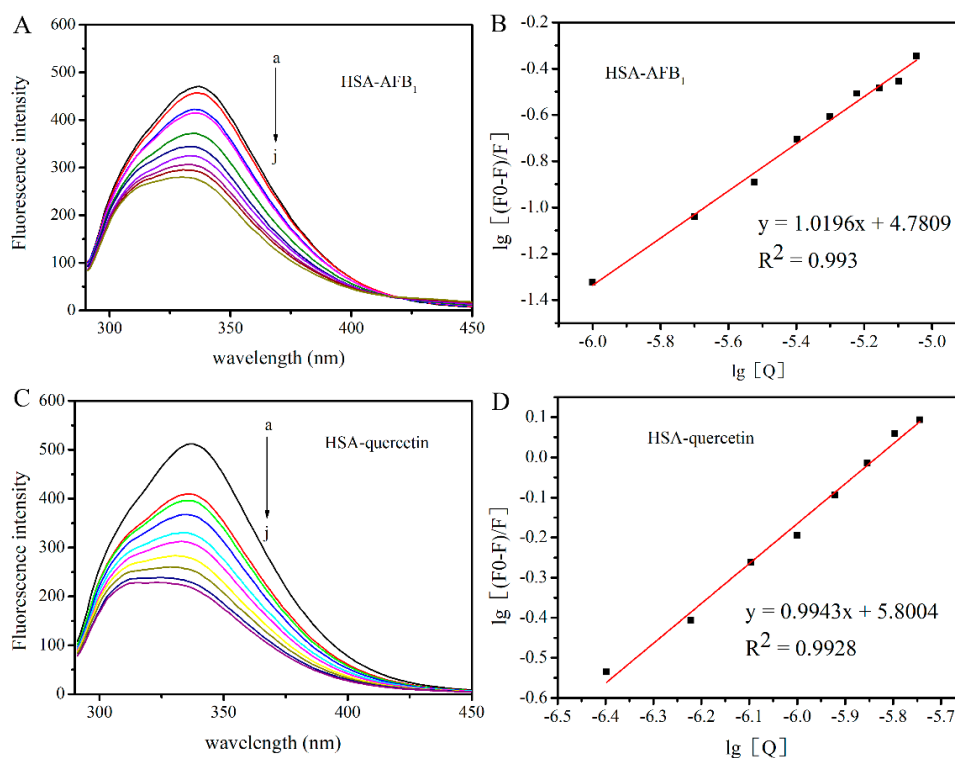
**Figure 1.** Structure of AFB<sub>1</sub> and its metabolites.

## 2. Results

### 2.1. Investigation of the Interaction between HSA with AFB<sub>1</sub> and Quercetin

To analyze the ability of quercetin to compete with AFB<sub>1</sub> for binding to HSA, we first studied the interaction mechanism of the AFB<sub>1</sub>-HSA system and the quercetin-HSA system, respectively. In this study, fluorescence emission spectra of albumin were recorded in the presence of increasing AFB<sub>1</sub> and quercetin concentration (Figure 2). In order to exclude the inner-filter effect, emission intensities were corrected by Equation (1). In a concentration-dependent fashion, AFB<sub>1</sub> and quercetin induced the decrease of fluorescence at 340 nm (emission maximum of HSA), resulting from the quenching effects of AFB<sub>1</sub> and quercetin. Based on the fluorescence quenching (Figure 2), the  $K_{sv}$ ,  $K_q$ , and  $K_a$  of the

AFB<sub>1</sub>-HSA complex and quercetin-HSA complex were calculated (Table 1) by Equations (2) and (4). In practice,  $k_q$  cannot be larger than  $2 \times 10^{10} \text{ L mol}^{-1} \text{ S}^{-1}$  for a collisional quenching process [25,26]. We found that the static quenching mechanism occurred for the AFB<sub>1</sub>-HSA system ( $K_q = 1.623 \times 10^{12}$ ) and the quercetin-HSA system ( $K_q = 1.83 \times 10^{13}$ ), which corresponded to the results of previous studies such as [27,28] and [29]. Fluorescence quenching studies denoted the results of a static quenching mechanism for the AFB<sub>1</sub>-HSA and quercetin-HSA system. The  $K_a$  of the quercetin-HSA complex was one order of magnitude larger than the AFB<sub>1</sub>-HSA complex, indicating that the fluorescence quenching ability of quercetin to HSA is significantly stronger than AFB<sub>1</sub> in the same condition.



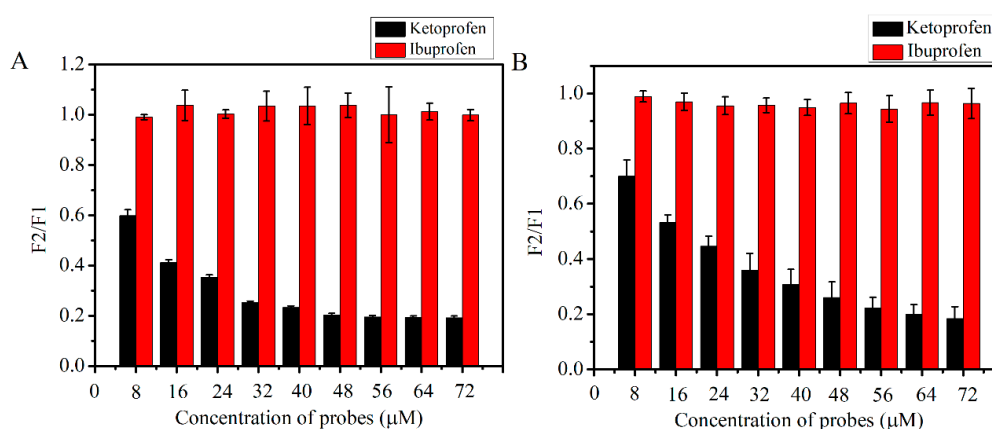
**Figure 2.** (A) Fluorescence emission spectra of HSA (5 μM) in the presence of increasing AFB<sub>1</sub> concentration. The concentration of AFB<sub>1</sub> was 0, 1, 2, 3, 4, 5, 6, 7, 8, 9 μM from a–j. (inset: the Stern–Volmer plot of the AFB<sub>1</sub>-HSA system). Conditions: T = 25 °C, pH = 7.4,  $\lambda_{ex}$  = 280 nm. (B) Plots of  $\log[(F_0 - F)/F]$  versus  $\log[Q]$  for the interaction of HSA and AFB<sub>1</sub>. (C) Fluorescence emission spectra of HSA (5 μM) in the presence of increasing quercetin concentration. The concentration of quercetin was 0.4, 0.8, 1.2, 1.6, 2.0, 2.4, 2.8, 3.2, 3.6, 4.0 μM from a–j. (inset: the Stern–Volmer plot of the quercetin-HSA system). Conditions: T = 25 °C, pH = 7.4,  $\lambda_{ex}$  = 280 nm. (D) Plots of  $\log[(F_0 - F)/F]$  versus  $\log[Q]$  for the interaction of HSA and quercetin.

**Table 1.** The values of  $K_{sv}$ ,  $K_q$ ,  $K_a$ , and the possible number of binding sites (n) for various systems are estimated.

System	$K_{sv} (\text{Lmol}^{-1})$	$K_q (\text{Lmol}^{-1}\text{S}^{-1})$	$K_a (\text{M}^{-1})$	n	$R^2$
AFB <sub>1</sub> -HSA	$1.62 \times 10^4$	$1.62 \times 10^{12}$	$6.02 \times 10^4$	1	0.993
Quercetin-HSA	$1.83 \times 10^5$	$1.83 \times 10^{13}$	$6.31 \times 10^5$	1	0.992

To determine the binding site, site marker competitive experiments were carried out. Typically, ketoprofen and ibuprofen, as the most commonly-applied site markers of Sudlow's site I (located on subdomain IIA) and Sudlow's site II (located on subdomain IIIA) respectively, were applied in this experiment. Complex formation of the ligand (AFB<sub>1</sub> or quercetin) with HSA resulted in the increased fluorescence signal of the ligand, and there was no fluorescence spectral interference of site markers

(ketoprofen and ibuprofen as competitive probes) at the maximum emission wavelength of the ligand (AFB<sub>1</sub> or quercetin) [25,30]. Therefore, displacement of the ligand from HSA by site markers could lead to the decreased fluorescence signal at 440 nm (AFB<sub>1</sub>) or 535 nm (quercetin) [25,30]. As we can see from Figure 3, In the AFB<sub>1</sub>-HSA and quercetin-HSA system, ibuprofen, a specific ligand for Sudlow's site II, had little effect on the fluorescence intensity of HSA, indicating that ibuprofen could not substitute the AFB<sub>1</sub> and quercetin. With the addition of ketoprofen, as a specific ligand for Sudlow's site I, the fluorescence intensity of HSA in both complexes was greatly reduced, indicating that ketoprofen replaced AFB<sub>1</sub> and that quercetin bound to HSA. Based on the above studies, we can conclude that AFB<sub>1</sub> and quercetin bind to HSA in Sudlow's site I (subdomain IIA), which is in good agreement with the modeling studies previously reported [25,26].



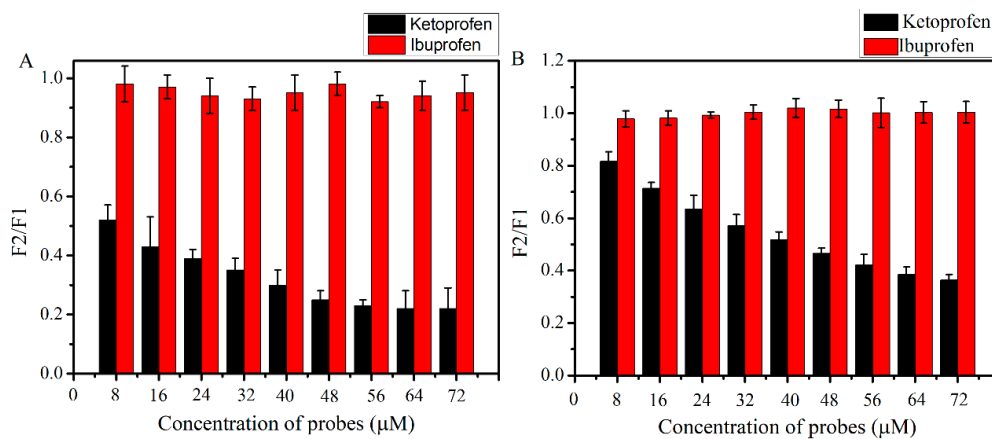
**Figure 3.** (A) Effects of probes on the fluorescence intensity of AFB<sub>1</sub> in the system of AFB<sub>1</sub>-HSA. ( $\lambda_{ex} = 365$  nm,  $\lambda_{em} = 440$  nm). (B) Effects of probes on the fluorescence intensity of quercetin in the system of quercetin-HSA. ( $\lambda_{ex} = 455$  nm,  $\lambda_{em} = 535$  nm). F1: Fluorescence intensity of AFB<sub>1</sub> or quercetin in the absence of probes. F2: Fluorescence intensity of AFB<sub>1</sub> or quercetin in the presence of probes.

## 2.2. Investigation of the Competitive Interaction between Quercetin and AFB<sub>1</sub> for HSA

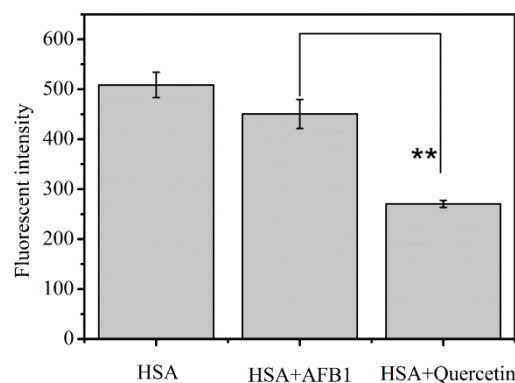
In order to explore the competitive interaction between quercetin and AFB<sub>1</sub> to bind with HSA, site marker experiments were carried out with fluorescence probes (ketoprofen and ibuprofen). Ibuprofen had no effect on the fluorescence spectra of AFB<sub>1</sub> (red columns in Figure 4A) or quercetin (red columns in Figure 4B) in the AFB<sub>1</sub>-HSA-quercetin system, while the fluorescence signals of AFB<sub>1</sub> (black columns in Figure 4A) or quercetin (black columns in Figure 4B) were decreased by the ketoprofen probe. These results indicated that AFB<sub>1</sub> and quercetin could bind to HSA in the same Sudlow's site I.

From the above experiments, both AFB<sub>1</sub> and quercetin could specifically bind to a known site on HSA in Sudlow's site I, and the  $K_a$  of the quercetin-HSA complex was much greater than that of the AFB<sub>1</sub>-HSA complex; thus, we hypothesized that the affinity of quercetin with HSA was much greater than AFB<sub>1</sub> with HSA.

For the sake of verification of this hypothesis, competitive interaction studies between quercetin and AFB<sub>1</sub> for HSA were carried out. As shown in Figure 5, there was a significant difference between the system (HSA + AFB<sub>1</sub>) and the system (HSA + quercetin) in the fluorescence intensity of HSA. It was confirmed that the fluorescence quenching ability of quercetin to HSA was significantly stronger than AFB<sub>1</sub>, and quercetin had stronger binding ability to bind with HSA compared to AFB<sub>1</sub>. Consequently, we further speculated that quercetin had the potential to compete for AFB<sub>1</sub> from HSA, thereby hindering the binding of AFB<sub>1</sub> to HSA.



**Figure 4.** (A) Effects of the probes on the fluorescence intensity of AFB<sub>1</sub> in the system of AFB<sub>1</sub>-HSA-quercetin ( $\lambda_{ex} = 365$  nm,  $\lambda_{em} = 440$  nm). (B) Effects of probes on the fluorescence intensity of quercetin in the system of AFB<sub>1</sub>-HSA-quercetin ( $\lambda_{ex} = 455$  nm,  $\lambda_{em} = 535$  nm). F1: Fluorescence intensity of AFB<sub>1</sub> or quercetin in the absence of probes. F2: Fluorescence intensity of AFB<sub>1</sub> or quercetin in the presence of probes.

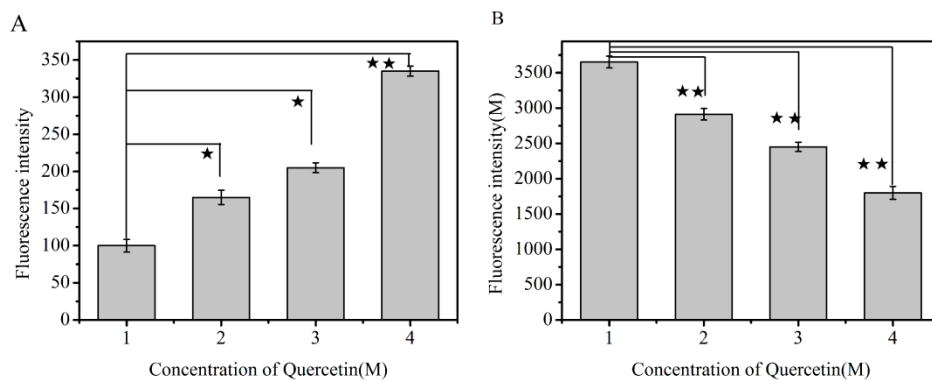


**Figure 5.** Fluorescence intensity of HSA in the absence or the presence of the AFB<sub>1</sub> and quercetin ( $\lambda_{ex} = 280$  nm,  $C_{AFB1} = C_{quercetin} = 4 \times 10^{-6}$  M, \*\*:  $p < 0.01$  between HSA-AFB<sub>1</sub> system and HSA-quercetin system).

### 2.3. Ultrafiltration Studies

Although quercetin and AFB<sub>1</sub> bound to HSA at the same binding site and the binding ability of the quercetin-HSA complex was stronger than the complex of AFB<sub>1</sub>-HSA, it could not directly be proven that AFB<sub>1</sub> could be substituted by quercetin in the presence of AFB<sub>1</sub> and quercetin. Consequently, the investigation of competitive substitution by ultrafiltration studies was carried out, which determined whether quercetin was able to block the binding of AFB<sub>1</sub> and HSA, decreasing the half-life of AFB<sub>1</sub>.

As shown in Figure 6, the addition of quercetin significantly increased the AFB<sub>1</sub> concentration in the free state and decreased the AFB<sub>1</sub> concentration in the bound state, which signified that the addition of quercetin could replace AFB<sub>1</sub> in the AFB<sub>1</sub>-HSA complex, and the AFB<sub>1</sub> was changed from the bound state to the free state.

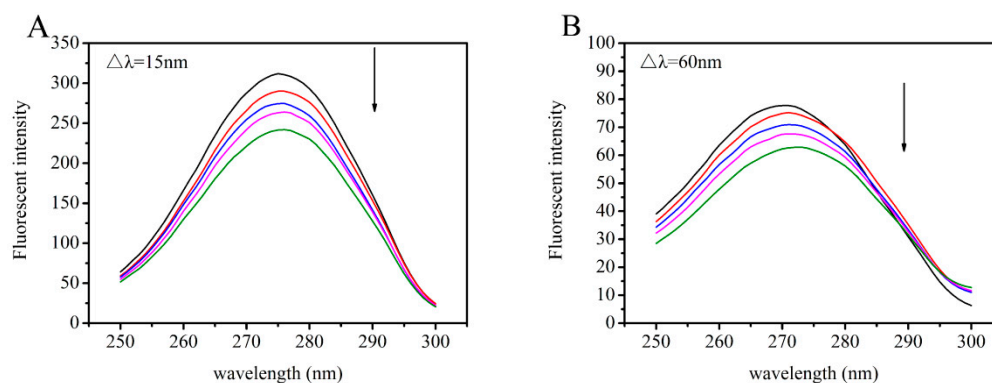


**Figure 6.** Fluorescent intensity of AFB<sub>1</sub> in different systems ( $\lambda_{\text{ex}} = 365 \text{ nm}$ ,  $\lambda_{\text{em}} = 440 \text{ nm}$ , (A) AFB<sub>1</sub> in the free state; (B) AFB<sub>1</sub> in binding state; \*:  $p < 0.05$  compared to another quercetin concentration; \*\*:  $p < 0.01$  compared to another quercetin concentration).

#### 2.4. Synchronous Fluorescence Studies

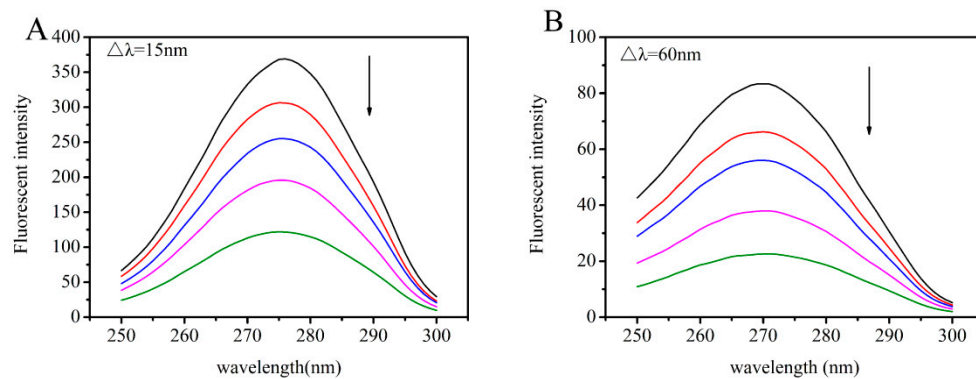
The information about the molecular environment near the chromophore molecules can be obtained by synchronous spectroscopy. The shift at maximum emission, corresponding to the changes of polarity around chromophores, is a useful way to evaluate the microenvironment of amino acid residues.

Figures 7–9 present the peaks for the complex when  $\Delta\lambda = 15 \text{ nm}$  and  $\Delta\lambda = 60 \text{ nm}$ . For the AFB<sub>1</sub>-HSA system, the synchronous fluorescence presented a red shift of 1 nm at  $\Delta\lambda = 15 \text{ nm}$  and 2 nm at  $\Delta\lambda = 60 \text{ nm}$ , which denoted that AFB<sub>1</sub> was able to increase polarity around Tyr, as well as around Trp. For the quercetin-HSA system, the synchronous fluorescence showed a small blue shift (0.5 nm) at  $\Delta\lambda = 15 \text{ nm}$ , which signified that hydrophobicity around the Tyr residue increased. However, the small red shift of 0.5 nm in the quercetin-HSA system at  $\Delta\lambda = 60 \text{ nm}$  demonstrated that the hydrophobicity around the Trp residue decreased. For the AFB<sub>1</sub>-quercetin-HSA system, where AFB<sub>1</sub> and quercetin were increasingly added and HSA was at constant concentration, a small red shift of 0.5 nm at  $\Delta\lambda = 60 \text{ nm}$  was observed, but no shift when  $\Delta\lambda = 15 \text{ nm}$ . These results suggested that in ternary system, ligands bind near the Trp residue. It can be inferred from the above experimental phenomena that the presence of quercetin affected the hydrophobicity of the protein microenvironment and changed the interaction of AFB<sub>1</sub> with HSA.

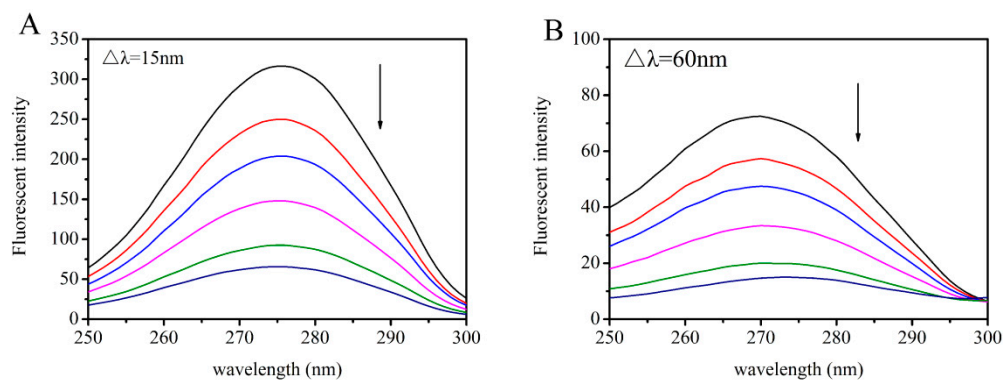


**Figure 7.** (A) Synchronous fluorescence spectra of the AFB<sub>1</sub>-HSA system,  $\Delta\lambda = 15 \text{ nm}$ ; (B) synchronous fluorescence spectra of the AFB<sub>1</sub>-HSA system,  $\Delta\lambda = 60 \text{ nm}$ . The concentration of HSA was  $4 \mu\text{M}$ , and the concentrations of AFB<sub>1</sub> were 0, 2, 4, 8,  $12 \mu\text{M}$  from top to bottom.





**Figure 8.** (A) Synchronous fluorescence spectra of the quercetin-HSA system,  $\Delta\lambda = 15$  nm; (B) synchronous fluorescence spectra of the quercetin-HSA system,  $\Delta\lambda = 60$  nm. The concentration of HSA was  $4 \mu\text{M}$ , and the concentrations of quercetin were 0, 2, 4, 8, 12  $\mu\text{M}$  from top to bottom.



**Figure 9.** (A) Synchronous fluorescence spectra of the competitive interaction between quercetin and  $\text{AFB}_1$  for HSA,  $\Delta\lambda = 15$  nm. (B) Synchronous fluorescence spectra of the competitive interaction between quercetin and  $\text{AFB}_1$  for HSA,  $\Delta\lambda = 60$  nm. The concentration of HSA was  $4 \mu\text{M}$ . The concentrations of quercetin and  $\text{AFB}_1$  were 0, 2, 4, 8, 12  $\mu\text{M}$  from top to bottom.

### 2.5. The Effect of the Ratio of Quercetin to HSA on the Competitive System

For the sake of estimating the effects of different ratios of  $\text{AFB}_1$ : quercetin on the competitive interaction between quercetin and  $\text{AFB}_1$  to bind with HSA, the fluorescence intensity of HSA in different systems was examined (Table 2). Here, and in the tables below, the first column represents different system conditions. The other columns represent the value of the fluorescence intensity of HSA. As shown in the second and third lines in Table 3, the fluorescence intensity of HSA in the system (HSA +  $\text{AFB}_1$  + quercetin) had a significant difference from the  $\text{AFB}_1$ -HSA system ( $p < 0.05$ ), which indicated that the addition of quercetin significantly affected the fluorescence intensity of HSA in the  $\text{AFB}_1$ -HSA system. In the fourth and fifth lines in Table 3, the fluorescence intensity of HSA in the system (HSA + quercetin +  $\text{AFB}_1$ ) had no significant difference from the HSA + quercetin system ( $p < 0.05$ ), which indicated that the addition of  $\text{AFB}_1$  had no influence in the HSA + quercetin system. These results indicated that quercetin is able to remove  $\text{AFB}_1$  from HSA and that  $\text{AFB}_1$  is not able to remove quercetin from HSA. In the system (HSA + ( $\text{AFB}_1$  + quercetin)), the fluorescence intensity had a significant difference compared with the system (HSA +  $\text{AFB}_1$ ), and the fluorescence intensity had no significant difference compared with the system (HSA + quercetin). It was revealed that HSA was prone to binding to quercetin and that quercetin could prevent the binding between  $\text{AFB}_1$  and HSA when the molar concentration ratio of  $\text{AFB}_1$ :quercetin was 1:2.

**Table 2.** The fluorescence intensity of HSA in different systems when the molar concentration rate of AFB<sub>1</sub>: quercetin is 1:2 ( $\lambda_{ex}$  = 280 nm,  $\lambda_{em}$  = 345 nm).

Systems	HSA:AFB <sub>1</sub> :Quercetin	
	2:1:2	1:1:2
HSA	494.3 ± 41.6 <sup>a</sup>	498.7 ± 61.8 <sup>a</sup>
HSA + AFB <sub>1</sub>	424.7 ± 58.5 <sup>b</sup>	444.9 ± 19.4 <sup>b</sup>
HSA + AFB <sub>1</sub> + Quercetin	251.4 ± 17.8 <sup>d</sup>	282.7 ± 37.9 <sup>c</sup>
HSA + Quercetin	292.6 ± 31.0 <sup>b, c</sup>	319.8 ± 31.3 <sup>b</sup>
HSA + Quercetin + AFB <sub>1</sub>	254.4 ± 22.9 <sup>c</sup>	265.4 ± 38.5 <sup>b</sup>
HSA + (AFB <sub>1</sub> + Quercetin)	315.6 ± 20.7 <sup>c</sup>	281.8 ± 49.9 <sup>c</sup>

Note: The numbers marked by the same letters (“a”, “b”, “c”, and “d”) represent the existence of statistical significance ( $p < 0.05$ ) in the same column.

As shown in Table 3, the same results were presented in 1:1 system as 1:2 system, indicating that quercetin can displace AFB<sub>1</sub> from HSA and that HSA was more prone to binding to quercetin, which prevented the binding of AFB<sub>1</sub> to HSA.

**Table 3.** The fluorescence intensity of HSA in different systems when the molar concentration rate of AFB<sub>1</sub>: quercetin is 1:1 ( $\lambda_{ex}$  = 280 nm,  $\lambda_{em}$  = 345 nm).

Systems	HSA:AFB <sub>1</sub> :Quercetin	
	2:1:1	1:1:1
HSA	512.8 ± 5.7 <sup>a</sup>	530.9 ± 57.8 <sup>a</sup>
HSA + AFB <sub>1</sub>	433.6 ± 15.2 <sup>b</sup>	497.1 ± 23.9 <sup>b</sup>
HSA + AFB <sub>1</sub> + Quercetin	376.3 ± 14.3 <sup>c</sup>	375.7 ± 64.1 <sup>c</sup>
HSA + Quercetin	374.1 ± 16.7 <sup>b</sup>	407.3 ± 21.6 <sup>b</sup>
HSA + Quercetin + AFB <sub>1</sub>	360.5 ± 27.5 <sup>b</sup>	386.7 ± 64.2 <sup>b</sup>
HSA + (AFB <sub>1</sub> + Quercetin)	398.6 ± 19.9 <sup>c</sup>	414.1 ± 23.4 <sup>c</sup>

Note: The numbers marked by the same letters (“a”, “b”, and “c”) represent the existence of statistical significance ( $p < 0.05$ ) in the same column.

As shown in Table 4, the fluorescence intensity of HSA in the system (HSA + AFB<sub>1</sub> + quercetin) was significantly different from the system (HSA + AFB<sub>1</sub>) ( $p < 0.05$ ). The fluorescence intensity of HSA in the system (HSA + quercetin + AFB<sub>1</sub>) was not significantly different from the system (HSA + quercetin). In the system (HSA + (AFB<sub>1</sub> + quercetin)), the fluorescence intensity had a significant difference compared with the system (HSA + AFB<sub>1</sub>), and the fluorescence intensity had no significant difference compared with the system (HSA + quercetin). These results revealed that quercetin is still able to competitively bind to HSA with AFB<sub>1</sub> when the ratio of AFB<sub>1</sub>:quercetin is 2:1.

**Table 4.** The fluorescence intensity of HSA in different systems when the molar concentration rate of AFB<sub>1</sub>: quercetin is 2:1 ( $\lambda_{ex}$  = 280 nm,  $\lambda_{em}$  = 345 nm).

Systems	HSA:AFB <sub>1</sub> :Quercetin	
	2:2:1	1:2:1
HSA	467.3 ± 40.9 <sup>a</sup>	501.8 ± 31.2 <sup>a</sup>
HSA + AFB <sub>1</sub>	405.7 ± 42.9 <sup>b</sup>	473.8 ± 2.7 <sup>b</sup>
HSA + AFB <sub>1</sub> + Quercetin	315.9 ± 21.6 <sup>c</sup>	37.3 ± 19.4 <sup>c</sup>
HSA + Quercetin	328.0 ± 21.6 <sup>b</sup>	365.5 ± 7.6 <sup>b</sup>
HSA + Quercetin + AFB <sub>1</sub>	362.9 ± 30.4 <sup>b</sup>	72.0 ± 43.4 <sup>b</sup>
HSA + (AFB <sub>1</sub> + Quercetin)	362.9 ± 30.4 <sup>b</sup>	72.0 ± 43.4 <sup>b, c</sup>

Note: The numbers marked by the same letters (“a”, “b”, and “c”) represent the existence of statistical significance ( $p < 0.05$ ) in the same column.



### 3. Discussion

In previous experiments, several mycotoxins' ligands (including AFB<sub>1</sub>, OTA, citrinin (CIT), deoxynivalenol (DON), patulin, zearalenone (ZEN), etc. [27,31,32]) formed stable complexes with HSA. HSA forms non-covalent complexes with these ligands, which could significantly affect the distribution and elimination of these ligand molecules [27,31–34]. Especially, OTA binds primarily to albumin with high affinity ( $K_a \sim 10^7$  L/mol) [18], which gives rise to its longer plasma elimination half-life (from a few days to one month) [22,35,36]. Sigrid et al. [36] studied the toxicokinetic profile of OTA (50 ng/g) after the oral or intravenous administration to fish, quail, mouse, rat, and monkey. It was found that the elimination of OTA can be completed through renal and hepatic clearance, both of which can be influenced by the toxin's binding to plasma macromolecules. Irene et al. [37] assessed the toxicokinetic profile of OTA by one human volunteer. This half-life of OTA was approximately eight-times longer than that determined previously in rats, which accounted for the specific OTA binding plasma protein species and respective protein concentrations. Furthermore, Kumagai et al. [16] studied the changes of OTA in both albumin-deficient and normal rats. It was revealed that the concentration of OTA in plasma was reduced to  $<0.5$   $\mu\text{g/mL}$  within 10 min of the injection in albumin-deficient rats, while remaining  $>50$   $\mu\text{g/mL}$  for 90 mins in normal rats [16]. The concentrations of OTA in bile and urine, as well as the excretion rate of OTA in these fluids were 20–70-fold higher in albumin-deficient than in normal rats [16]. In summary, it was demonstrated that the primary effect of albumin binding on OTA is to retard its elimination by restricting the entry of OTA into the hepatic and renal cells [16].

It is well known that natural flavonoids can also bind to HSA at the same binding site as OTA does (site I, subdomain IIA). Miklós Poór et al. [18] studied the competitive interaction between flavonoids and OTA, which found that some flavonoid aglycones are able to remove the toxin from HSA and decrease the bound fraction of OTA. Baudrimonta et al. [17] had found that piroxicam also can bind strongly to plasma proteins and could stop OTA binding with HSA and transporting to target organs. Miklós Poór et al. [24] studied the competitive interactions between OTA and 13 drug molecules for binding to HSA, which demonstrated that some drugs show high competitive capacity. In addition, several extracorporeal dialysis procedures using albumin-containing dialysates have proven to be an effective tool for removing endogenous toxins or overdosed drugs from patients [38–40].

As we all know, AFB<sub>1</sub> can bind to HSA with high affinity ( $K \sim 10^4$  L/mol). The high affinity between AFB<sub>1</sub> and HSA could affect its distribution and elimination [27]. It has not been studied at present whether AFB<sub>1</sub> is removed from HSA and then reduces the bound fraction by competitive interaction. Besides, flavonoids could bind to HSA with high affinity ( $K \sim 10^5$  L/mol) at the same binding site as AFB<sub>1</sub> does (site I, subdomain IIA) [41,42]. Especially, quercetin is one of the most common flavonoids in nature and can bind with human albumin with high affinity [43]. Furthermore, quercetin was the most effective competitor of OTA in the competition experiment between OTA and polyphenols [18]. Therefore, quercetin was adopted to study the competitive interaction between AFB<sub>1</sub> and quercetin when binding with HSA.

However, no data were found for any trial using quercetin in competitive AFB<sub>1</sub>-HSA models. In this experiment, fluorescence quenching studies denoted the results of a static quenching mechanism for the AFB<sub>1</sub>-HSA and quercetin-HSA (Figure 2) complex. The  $K_a$  of quercetin-HSA was one order of magnitude larger than AFB<sub>1</sub>-HSA (Table 1), indicating that the fluorescence quenching ability of quercetin to HSA is significantly stronger than AFB<sub>1</sub> in the same condition. By the competitive probe experiment, we can conclude that Sudlow's site I is a high affinity binding site of the AFB<sub>1</sub>-HSA and quercetin-HSA complex, which is in good agreement with modeling studies previously reported [25,26]. Quercetin and AFB<sub>1</sub> bind to HSA at the same binding site (Figure 3); thus, it is possible that quercetin and AFB<sub>1</sub> could competitively bind with HSA. For the sake of verification of the above hypothesis, research of the competitive interaction between quercetin and AFB<sub>1</sub> to bind with HSA was carried out (Figures 4 and 5). It was revealed that in the competition system, the binding sites of quercetin and AFB<sub>1</sub> on HSA were still on Sudlow's site I. The binding ability of quercetin with HSA was significantly

stronger than that of AFB<sub>1</sub>-HSA, which indicated that quercetin had the ability to replace HSA from AFB<sub>1</sub>. By ultrafiltration studies (Figure 6), it was further confirmed that quercetin was able to remove AFB<sub>1</sub> from HSA and then decreased the bound fraction of AFB<sub>1</sub>.

According to other published papers, it was found that quercetin could affect the AFB<sub>1</sub>-induced negative changes. Choi et al. [44] found that quercetin does not directly protect against AFB<sub>1</sub>-mediated liver damage in vivo, but plays a partial role in promoting antioxidative defense systems and inhibiting lipid peroxidation. Additionally, El-Nekeety et al. [45] found that quercetin has potential antioxidant activity and could regulate the alteration of genes expression induced by AFB<sub>1</sub>. However, in this experiment, based on the competitive interaction between quercetin and AFB<sub>1</sub> binding to HSA, respectively, we can conclude that quercetin is able to remove AFB<sub>1</sub> from HSA and decrease the bound fraction of AFB<sub>1</sub>. The effect of quercetin on AFB<sub>1</sub>-induced negative changes was studied in this experiment from different aspects compared with the above [9–11,44,45] studies.

#### 4. Conclusion

Competitive interaction between quercetin and HSA was investigated by fluorescence spectroscopy, synchronous spectroscopy, ultrafiltration studies, etc. This is the first time evidence that the addition of quercetin can remove AFB<sub>1</sub> from HSA and decrease the bound fraction of AFB<sub>1</sub> has been illustrated. Furthermore, experiments in vivo are necessary in the future to explore the toxicological outcome of the competitive interaction between AFB<sub>1</sub> and quercetin with HSA.

#### 5. Materials and Methods

##### 5.1. Reagents

Aflatoxin B<sub>1</sub> (AFB<sub>1</sub>, from Sigma), human serum albumin (HSA, from Sigma), quercetin (from Aladdin), ketoprofen (from Shanghai chemical Technology Co., Ltd), and ibuprofen (Beijing Bailingwei Technology Co., Ltd) were used as received. Tris solution was obtained from Amersco. Other chemicals were of analytical grade.

##### 5.2. Fluorescence Spectroscopic Measurements

Fluorescence measurements were carried out employing an F-2500 fluorescence spectrophotometer (Shimadzu, Japan) at room temperature (25 °C), with a 1-cm path length quartz cell, using an excitation wavelength of 280 nm. In the AFB<sub>1</sub>-HSA system, the concentration of HSA was fixed at 5 μM, whereas the AFB<sub>1</sub> concentration was varied from 0 to 9 μM. In the quercetin-HSA system, the concentration of HSA was fixed at 5 μM, whereas the quercetin concentration was varied from 0 to 4 μM. In competitive interaction between AFB<sub>1</sub> and quercetin with HSA, the concentration of HSA was fixed at 5 μM, whereas the concentration of AFB<sub>1</sub> and quercetin was varied as above. In competitive interaction studies, the construction methods of different systems were as follows: The system (HSA + AFB<sub>1</sub>) was constructed by the addition of AFB<sub>1</sub> to the HSA solution. The system (HSA + quercetin) was constructed by the addition of quercetin to the HSA solution. For the system (HSA + AFB<sub>1</sub> + quercetin), AFB<sub>1</sub> was added to the HSA solution firstly, and quercetin was then added to the solution. For the system (HSA + quercetin + AFB<sub>1</sub>), quercetin was added to the HSA solution firstly, followed by the addition of AFB<sub>1</sub>. For the HSA + (AFB<sub>1</sub> + quercetin) system, AFB<sub>1</sub> and quercetin were added to the Tris solution firstly, followed by the addition of HSA.

For the sake of eliminating the inner-filter effects, the fluorescence intensities were corrected applying the following equation [26,46]:

$$F_{\text{cor}} = F_{\text{obs}} \times e^{\frac{A_{\text{ex}} + A_{\text{em}}}{2}} \quad (1)$$

where  $F_{\text{cor}}$  and  $F_{\text{obs}}$  represent the corrected and observed fluorescence intensities, respectively; whereas  $A_{\text{ex}}$  and  $A_{\text{em}}$  denote the absorbance values at excitation and emission wavelength, respectively. The corrected fluorescence data were used for further analysis, related to HSA fluorescence quenching.

The fluorescence quenching data are usually evaluated via the Stern–Volmer equation [16,17]:

$$\frac{F_0}{F} = 1 + K_q \tau_0 [Q] = 1 + K_{\text{sv}} [Q] \quad (2)$$

where  $F_0$  and  $F$  are the fluorescence emission intensity without and with the addition of a known concentration  $[Q]$ , respectively.  $K_{\text{sv}}$  is the Stern–Volmer constant;  $K_q$  is the quenching rate of the biomolecule;  $\tau_0$  is the average fluorescence lifetime of HSA without quencher.

For static quenching, the modified Stern–Volmer equation analyzes the data [47–49]:

$$\frac{F_0}{\Delta F} = \frac{F_0}{F_0 - F} = \frac{1}{f_a K_a [Q]} + \frac{1}{f_a} \quad (3)$$

where  $\Delta F = F_0 - F$  is the difference in fluorescence intensity before and after the addition of the quencher at concentration  $[Q]$ ,  $f_a$  represents the fraction of accessible fluorescence, and  $K_a$  denotes the effective quenching constant for the accessible fluorophores [50,51].

The binding constant ( $K_a$ ) and the number of binding sites ( $n$ ) can be evaluated using the following equation [52,53]:

$$\log\left(\frac{F_0 - F}{F}\right) = \log K_a + n \log [Q] \quad (4)$$

where  $K_a$  is the binding constant of the interaction between the quencher and HSA and  $n$  is the number of binding sites. Based on a plot of  $\log[(F_0 - F)/F]$  versus  $\log [Q]$ , the  $n$  equal to the slope and  $K_a$  can be obtained.

### 5.3. Synchronous Fluorescence

The synchronous fluorescence spectra were collected by simultaneous scanning of the excitation and emission monochromators on an F-2500 fluorescence spectrophotometer (Shimadzu, Japan). The experimental condition was that the wavelength interval ( $\Delta\lambda$ ) of Thr and Trp were 15 nm and 60 nm, respectively [54,55], at 25 °C. The concentration of HSA was  $4 \times 10^{-6}$  M. The concentrations of AFB<sub>1</sub> and quercetin both were  $0-8 \times 10^{-6}$  M.

### 5.4. Site Marker Competitive Experiments

Site marker competitive experiments were performed with ketoprofen (as the site I marker) and ibuprofen (as the site II marker) [56,57]. First, 1.0 mL of the  $4 \times 10^{-6}$  M HSA solution was added to a 1-cm fluorescence cuvette, with the addition of  $4 \times 10^{-4}$  M quercetin and  $4 \times 10^{-4}$  M AFB<sub>1</sub> solution to make the ratio of HSA to ligand concentration 1:1. Then,  $4 \times 10^{-3}$  M ketoprofen or ibuprofen were added the above solution, and the concentrations of ibuprofen and ketoprofen were 8.0, 16.0, 24.0, 32.0, 40.0, 48.0, 56.0, 64, and  $72.0 \times 10^{-6}$  M.

### 5.5. Ultrafiltration

Quercetin, AFB<sub>1</sub>, and HSA solution was prepared at a concentration ratio of 1:1:1 and reacted at room temperature for 30 min. After the reaction, the solution was centrifuged at 4 °C at 10,000 rpm for 15 min and washed 3 times on the filter with Tris-HCL buffer, collecting the filtrated solution for further analysis of the next term. Methanol/water (50:50; v/v) was added to the filter, and the blended mixture was left standing for 30 min. Then, it was centrifuged at 4 °C at 10,000 rpm for 15 min. Here, the centrifugation operation should be repeated 3 times. Furthermore, the elution solution was collected for the fluorescence measurements. The fluorescence emission spectra in the range of 400–500 nm of free solution and the elution were scanned at an excitation of 365 nm.

### 5.6. Statistical Analyses

All of the data were statistically analyzed [53,58] by the one-way ANOVA test (IBM SPSS Statistics 20), with the level of significance at a minimum of  $p < 0.05$  and a maximum of  $p < 0.01$ .

**Author Contributions:** Conceptualization, L.M. and Y.Z.; methodology, L.M., L.C., and H.T.; validation, T.G., W.L., and Y.Y.; investigation, H.T. and L.C.; data curation, H.D.; writing, original draft preparation, H.T. and S.L.; writing, review and editing, H.T., H.Z., and L.M.; supervision, L.M. and Y.Z.; project administration, L.M. and Y.Z.; funding acquisition, L.M., H.Z., and T.G.

**Funding:** This research was funded by Fundamental Research Funds for the Central Universities of China (Grant Numbers SWU118088 and SWU117005) and the Technology Innovation and Application Demonstration Project related to people's livelihood in Chongqing (Project No. cstc2018jscx-msybX0204).

**Conflicts of Interest:** The funders had no role in the design of the study; in the collection, analyses, or interpretation of the data; in the writing of the manuscript; nor in the decision to publish the results.

### References

1. Rushing, B.R.; Selim, M.I. Aflatoxin B<sub>1</sub>: A review on metabolism, toxicity, occurrence in food, occupational exposure, and detoxification methods. *Chem. Toxicol.* **2019**, *124*, 81–100. [[CrossRef](#)]
2. Dai, Y.Q.; Huang, K.L.; Zhang, B.Y.; Zhu, L.Y.; Xu, W.T. Aflatoxin B<sub>1</sub>-induced epigenetic alterations: An overview. *Food Chem. Toxicol.* **2017**, *109*, 683–689. [[CrossRef](#)] [[PubMed](#)]
3. Hamid, A.S.; Tesfamariam, I.G.; Zhang, Y.C.; Zhang, Z.G. Aflatoxin B<sub>1</sub>-induced hepatocellular carcinoma in developing countries: Geographical distribution, mechanism of action and prevention (Review). *Oncol. Lett.* **2013**, *5*, 1087–1092.
4. Su, C.; Hu, Y.; Gao, D.; Luo, Y.; Chen, A.J.; Jiao, X.; Gao, W. Occurrence of Toxicogenic Fungi and Mycotoxins on Root Herbs from Chinese Markets. *J. Prot.* **2018**, *81*, 754–761. [[CrossRef](#)] [[PubMed](#)]
5. Ma, R.; Zhang, L.; Liu, M.; Su, Y.-T.; Xie, W.-M.; Zhang, N.-Y.; Dai, J.-F.; Wang, Y.; Rajput, S.A.; Qi, D.-S.; et al. Individual and Combined Occurrence of Mycotoxins in Feed Ingredients and Complete Feeds in China. *Toxins* **2018**, *10*, 113. [[CrossRef](#)] [[PubMed](#)]
6. Yilmaz, S.; Kaya, E.; Comakli, S. Vitamin E (alpha tocopherol) attenuates toxicity and oxidative stress induced by aflatoxin in rats. *Adv. Clin. Exp. Med.* **2017**, *26*, 907–917. [[CrossRef](#)] [[PubMed](#)]
7. Madrigal-Bujaidar, E.; Morales-Gonzalez, J.A.; Sanchez-Gutierrez, M.; Izquierdo-Vega, J.A.; Reyes-Arellano, A.; Alvarez-Gonzalez, I.; Perez-Pasten, R.; Madrigal-Santillan, E. Prevention of Aflatoxin B<sub>1</sub>-induced dna breaks by beta-D-glucan. *Toxins* **2015**, *7*, 2145–2158. [[CrossRef](#)]
8. Mannaa, F.A.; Abdel-Wahhab, K.G.; Abdel-Wahhab, M.A. Prevention of cardiotoxicity of aflatoxin B-1 via dietary supplementation of papaya fruit extracts in rats. *Cytotechnology* **2014**, *66*, 327–334. [[PubMed](#)]
9. Selim, K.M.; El-hofy, H.; Khalil, R.H. The efficacy of three mycotoxin adsorbents to alleviate aflatoxin B-1-induced toxicity in *Oreochromis niloticus*. *Aquac. Int.* **2014**, *22*, 523–540. [[CrossRef](#)]
10. Jaynes, W.F.; Zartman, R.E. Aflatoxin Toxicity Reduction in Feed by Enhanced Binding to Surface-Modified Clay Additives. *Toxins* **2011**, *3*, 551–565. [[PubMed](#)]
11. Jenkins, T.P.; Fryer, T.; Dehli, R.I.; Jürgensen, J.A.; Fuglsang-Madsen, A.; Føns, S.; Laustsen, A.H. Toxin Neutralization Using Alternative Binding Proteins. *Toxins* **2019**, *11*, 53. [[CrossRef](#)] [[PubMed](#)]
12. Rawal, S.; Kim, J.E.; Coulombe, R., Jr. Aflatoxin B<sub>1</sub> in poultry: Toxicology, metabolism and prevention. *Res. Sci.* **2010**, *89*, 325–331. [[CrossRef](#)]
13. Ma, L.; Maragos, C.M.; Zhang, Y. Interaction of zearalenone with bovine serum albumin as determined by fluorescence quenching. *Mycotoxin Res.* **2017**, *34*, 39–48. [[CrossRef](#)] [[PubMed](#)]
14. Bonomo, S.; Jørgensen, F.S.; Olsen, L. Dissecting the Cytochrome P450 1A2- and 3A4-Mediated Metabolism of Aflatoxin B<sub>1</sub> in Ligand and Protein Contributions. *Chem. A Eur. J.* **2017**, *23*, 2884–2893. [[CrossRef](#)] [[PubMed](#)]
15. Galtier, P.; Camguilhem, R.; Bodin, G. Evidence for in vitro and in vivo interaction between ochratoxin A and three acidic drugs. *Cosmet. Toxicol.* **1980**, *18*, 493–496. [[CrossRef](#)]
16. Kumagai, S. Ochratoxin A: Plasma concentration and excretion into bile and urine in albumin-deficient rats. *Chem. Toxicol.* **1985**, *23*, 941–943. [[CrossRef](#)]
17. Baudrimont, I.; Murn, M.; Betbeder, A.; Guilcher, J.; Creppy, E. Effect of piroxicam on the nephrotoxicity induced by ochratoxin A in rats. *Toxicology* **1995**, *95*, 147–154. [[CrossRef](#)]

18. Poór, M.; Kunsági-Máté, S.; Bencsik, T.; Petrik, J.; Vladimir-Knežević, S.; Kőszegi, T. Flavonoid aglycones can compete with Ochratoxin A for human serum albumin: A new possible mode of action. *Int. J. Biol. Macromol.* **2012**, *51*, 279–283. [[CrossRef](#)]
19. Poór, M.; Boda, G.; Needs, P.W.; Kroon, P.A.; Lemli, B.; Bencsik, T. Interaction of quercetin and its metabolites with warfarin: Displacement of warfarin from serum albumin and inhibition of CYP2C9 enzyme. *Biomed. Pharmacother.* **2017**, *88*, 574–581. [[CrossRef](#)]
20. Di Bari, L.; Ripoli, S.; Pradhan, S.; Salvadori, P. Interactions between quercetin and warfarin for albumin binding: A New Eye on Food/Drug Interference. *Chirality* **2010**, *22*, 593–596. [[CrossRef](#)]
21. Jing, J.-J.; Liu, B.; Wang, X.; He, L.-L.; Guo, X.-Y.; Xu, M.-L.; Li, Q.-Y.; Gao, B.; Dong, B.-Y. Binding of fluphenazine with human serum albumin in the presence of rutin and quercetin: An evaluation of food-drug interaction by spectroscopic techniques. *Luminescence* **2017**, *32*, 1056–1065. [[CrossRef](#)]
22. Poór, M.; Boda, G.; Kunsági-Máté, S.; Needs, P.W.; Kroon, P.A.; Lemli, B. Fluorescence spectroscopic evaluation of the interactions of quercetin, isorhamnetin, and quercetin-3'-sulfate with different albumins. *J. Lumin.* **2018**, *194*, 156–163. [[CrossRef](#)]
23. Das, P.; Chaudhari, S.K.; Das, A.; Kundu, S.; Saha, C. Interaction of flavonols with human serum albumin: A biophysical study showing structure–activity relationship and enhancement when coated on silver nanoparticles. *J. Biomol. Struct. Dyn.* **2018**, *37*, 1414–1426. [[CrossRef](#)]
24. Poor, M.; Kunsagi-Mate, S.; Czibulya, Z.; Li, Y.; Peles-Lemli, B.; Petrik, J.; Vladimir-Knezevic, S.; Koszegi, T. Fluorescence spectroscopic investigation of competitive interactions between ochratoxin A and 13 drug molecules for binding to human serum albumin. *Luminescence* **2013**, *28*, 726–733. [[CrossRef](#)]
25. Sengupta, B.; Sengupta, P.K. Binding of quercetin with human serum albumin: A critical spectroscopic study. *Biopolymers* **2003**, *72*, 427–434. [[CrossRef](#)]
26. Mohseni-Shahri, F.S.; Housaindokht, M.R.; Bozorgmehr, M.R.; Moosavi-Movahedi, A.A. The influence of the flavonoid quercetin on the interaction of propranolol with human serum albumin: Experimental and theoretical approaches. *J. Lumin.* **2014**, *154*, 229–240. [[CrossRef](#)]
27. Poór, M.; Bálint, M.; Hetényi, C.; Gődér, B.; Kunsági-Máté, S.; Kőszegi, T.; Lemli, B.; Moretti, A. Investigation of Non-Covalent Interactions of Aflatoxins (B1, B2, G1, G2, and M1) with Serum Albumin. *Toxins* **2017**, *9*, 339. [[CrossRef](#)]
28. Pacheco, M.E.; Bruzzone, L. Interactions between imazethapyr and bovine serum albumin: Spectrofluorimetric study. *J. Lumin.* **2012**, *132*, 2730–2735. [[CrossRef](#)]
29. Neamtu, S.; Tosa, N.; Bogdan, M. Spectroscopic investigation of tolmetin interaction with human serum albumin. *J. Pharm. Biomed. Anal.* **2013**, *85*, 277–282. [[CrossRef](#)]
30. Nunes, N.M.; Pacheco, A.F.C.; Agudelo, Á.J.P.; Da Silva, L.H.M.; Pinto, M.S.; Hespanhol, M.D.C.; Pires, A.C.D.S. Interaction of cinnamic acid and methyl cinnamate with bovine serum albumin: A thermodynamic approach. *Food Chem.* **2017**, *237*, 525–531. [[CrossRef](#)]
31. Kőszegi, T.; Poór, M.; Manderville, R.A.; Pfohl-Leszkowicz, A. Ochratoxin A: Molecular Interactions, Mechanisms of Toxicity and Prevention at the Molecular Level. *Toxins* **2016**, *8*, 111. [[CrossRef](#)]
32. Faisal, Z.; Lemli, B.; Szerencses, D.; Kunsagi-Mate, S.; Balint, M.; Hetenyi, C.; Kuzma, M.; Mayer, M.; Poor, M. Interactions of zearalenone and its reduced metabolites alpha-zearalenol and beta-zearalenol with serum albumins: Species differences, binding sites, and thermodynamics. *Mycotoxin Res.* **2018**, *34*, 269–278. [[CrossRef](#)]
33. Fanali, G.; Di Masi, A.; Trezza, V.; Marino, M.; Fasano, M.; Ascenzi, P. Human serum albumin: From bench to bedside. *Mol. Asp. Med.* **2012**, *33*, 209–290. [[CrossRef](#)]
34. Poór, M.; Li, Y.; Matisz, G.; Kiss, L.; Kunsági-Máté, S.; Kőszegi, T. Quantitation of species differences in albumin–ligand interactions for bovine, human and rat serum albumins using fluorescence spectroscopy: A test case with some Sudlow's site I ligands. *J. Lumin.* **2014**, *145*, 767–773. [[CrossRef](#)]
35. Faisal, Z.; Derdak, D.; Lemli, B.; Kunsagi-Mate, S.; Balint, M.; Hetenyi, C.; Csepregi, R.; Koszegi, T.; Sueck, F.; Cramer, B.; et al. Interaction of 2R-ochratoxin A with Serum Albumins: Binding Site, Effects of Site Markers, Thermodynamics, Species Differences of Albumin-binding, and Influence of Albumin on Its Toxicity in MDCK Cells. *Toxins* **2018**, *10*, 353. [[CrossRef](#)]
36. Hagelberg, S.; Hult, K.; Fuchs, R. Toxicokinetics of ochratoxin A in several species and its plasma-binding properties. *J. Appl. Toxicol.* **1989**, *9*, 91–96. [[CrossRef](#)]
37. Studer-Rohr, I.; Schlatter, J.; Dietrich, D.R. Kinetic parameters and intraindividual fluctuations of ochratoxin A plasma levels in humans. *Arch. Toxicol.* **2000**, *74*, 499–510. [[CrossRef](#)]



38. Rimac, H.; Debeljak, Ž.; Bojić, M.; Miller, L.; Rimac, Ž.D.H. Displacement of Drugs from Human Serum Albumin: From Molecular Interactions to Clinical Significance. *Curr. Med. Chem.* **2017**, *24*, 1. [[CrossRef](#)]
39. Yamasaki, K.; Chuang, V.T.G.; Maruyama, T.; Otagiri, M. Albumin–drug interaction and its clinical implication. *Biochim. Biophys. (BBA) Gen. Subj.* **2013**, *1830*, 5435–5443. [[CrossRef](#)]
40. Perry, J.L.; Il'ichev, Y.V.; Kempf, V.R.; McClendon, J.; Park, G.; Manderville, R.A.; Ruker, F.; Dockal, M.; Simon, J.D. Binding of ochratoxin A derivatives to human serum albumin. *J. Phys. Chem. B* **2003**, *107*, 6644–6647. [[CrossRef](#)]
41. Xiao, J.B.; Kai, G.Y. A Review of dietary polyphenol-plasma protein interactions: Characterization, influence on the bioactivity, and structure-affinity relationship. *Crit. Rev. Food Sci. Nutr.* **2012**, *52*, 85–101. [[CrossRef](#)]
42. Tang, F.; Xie, Y.; Cao, H.; Yang, H.; Chen, X.; Xiao, J. Fetal bovine serum influences the stability and bioactivity of resveratrol analogues: A polyphenol-protein interaction approach. *Food Chem.* **2017**, *219*, 321–328. [[CrossRef](#)]
43. Xiao, J.B.; Zhao, Y.R.; Wang, H.; Yuan, Y.M.; Yang, F.; Zhang, C.; Yamamoto, K. Noncovalent interaction of dietary polyphenols with common human plasma proteins. *J. Agric. Food Chem.* **2017**, *59*, 10747–10754. [[CrossRef](#)]
44. Choi, K.-C.; Chung, W.-T.; Kwon, J.-K.; Yu, J.-Y.; Jang, Y.-S.; Park, S.-M.; Lee, S.-Y.; Lee, J.-C. Inhibitory effects of quercetin on aflatoxin B<sub>1</sub>-induced hepatic damage in mice. *Chem. Toxicol.* **2010**, *48*, 2747–2753. [[CrossRef](#)]
45. El-Nekeety, A.A.; Abdel-Azeim, S.H.; Hassan, A.M.; Hassan, N.S.; Aly, S.E.; Abdel-Wahhab, M.A.; Abdel-Wahhab, P.D.M.A. Quercetin inhibits the cytotoxicity and oxidative stress in liver of rats fed aflatoxin-contaminated diet. *Toxicol. Rep.* **2014**, *1*, 319–329. [[CrossRef](#)]
46. Panda, D.; Datta, A. Evidence for covalent binding of epicocconone with proteins from synchronous fluorescence spectra and fluorescence lifetimes. *J. Chem. Sci.* **2007**, *119*, 99–104. [[CrossRef](#)]
47. Maurya, N.; Parray, M.U.D.; Maurya, J.K.; Kumar, A.; Patel, R. Interaction of promethazine and adiphenine to human hemoglobin: A comparative spectroscopic and computational analysis. *Spectrochim. A Mol. Biomol. Spectrosc.* **2018**, *199*, 32–42. [[CrossRef](#)]
48. Poór, M.; Lemli, B.; Bálint, M.; Hetenyi, C.; Sali, N.; Kószegi, T.; Kunsági-Mate, S.; Van Der Fels-Klerx, H. (Ine) Interaction of Citrinin with Human Serum Albumin. *Toxins* **2015**, *7*, 5155–5166. [[CrossRef](#)]
49. Kameníková, M.; Furtmüller, P.G.; Klacsová, M.; Lopez-Guzman, A.; Toca-Herrera, J.L.; Vitkovská, A.; Devínsky, F.; Mučaji, P.; Nagy, M. Influence of quercetin on the interaction of gliclazide with human serum albumin-spectroscopic and docking approaches. *Luminescence* **2017**, *32*, 1203–1211. [[CrossRef](#)]
50. Bagheri, M.; Fatemi, M.H. Fluorescence spectroscopy, molecular docking and molecular dynamic simulation studies of HSA-Aflatoxin B<sub>1</sub> and G<sub>1</sub> interactions. *J. Lumin.* **2018**, *202*, 345–353. [[CrossRef](#)]
51. Poór, M.; Kunsági-Máté, S.; Bálint, M.; Hetényi, C.; Gerner, Z.; Lemli, B. Interaction of mycotoxin zearalenone with human serum albumin. *J. Photochem. Photobiol. B Biol.* **2017**, *170*, 16–24. [[CrossRef](#)]
52. Wang, J.; Ma, L.; Zhang, Y.; Jiang, T. Investigation of the interaction of deltamethrin (DM) with human serum albumin by multi-spectroscopic method. *J. Mol. Struct.* **2017**, *1129*, 160–168. [[CrossRef](#)]
53. Shiri, F.; Rahimi-Nasrabadi, M.; Ahmadi, F.; Ehrlich, H. Multispectroscopic and molecular modeling studies on the interaction of copper-ibuprofenate complex with bovine serum albumin (BSA). *Spectrochim. Acta A Mol. Biomol. Spectrosc.* **2018**, *203*, 510–521. [[CrossRef](#)]
54. Nafisi, S.; Vishkaee, T.S. Study on the interaction of tamiflu and oseltamivir carboxylate with human serum albumin. *J. Photochem. Photobiol. B Biol.* **2011**, *105*, 34–39. [[CrossRef](#)]
55. Hao, C.; Xu, G.; Feng, Y.; Lu, L.; Sun, W.; Sun, R. Fluorescence quenching study on the interaction of ferrocene oxide nanoparticles with bovine serum albumin. *Spectrochim. A Mol. Biomol. Spectrosc.* **2017**, *184*, 191–197. [[CrossRef](#)]
56. Östlund, J.; Zlabek, V.; Zamaratskaia, G. In vitro inhibition of human CYP2E1 and CYP3A by quercetin and myricetin in hepatic microsomes is not gender dependent. *Toxicology* **2017**, *381*, 10–18. [[CrossRef](#)]
57. Sun, M.; Tang, Y.; Ding, T.; Liu, M.; Wang, X. Inhibitory effects of celastrol on rat liver cytochrome P450 1A2, 2C11, 2D6, 2E1 and 3A2 activity. *Fitoterapia* **2014**, *92*, 1–8. [[CrossRef](#)]
58. Li, Y.; Czibulya, Z.; Poór, M.; LeComte, S.; Kiss, L.; Harté, E.; Kószegi, T.; Kunsági-Máté, S. Thermodynamic study of the effects of ethanol on the interaction of ochratoxin A with human serum albumin. *J. Lumin.* **2014**, *148*, 18–25. [[CrossRef](#)]

