



Article Butenolides from the Coral-Derived Fungus Aspergillius terreus SCSIO41404

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Abstract: Five undescribed butenolides including two pairs of enantiomers, (+)-asperteretal G (1a), (-)-asperteretal G (1b), (+)-asperteretal H (2a), (-)-asperteretal H (2b), asperteretal I (3), and *para*-hydroxybenzaldehyde derivative, (*S*)-3-(2,3-dihydroxy-3-methylbutyl)-4-hydroxybenzaldehyde (14), were isolated together with ten previously reported butenolides 4–13, from the coral-derived fungus *Aspergillus terreus* SCSIO41404. Enantiomers 1a/1b and 2a/2b were successfully purified by high performance liquid chromatography (HPLC) using a chiral column, and the enantiomers 1a and 1b were new natural products. Structures of the unreported compounds, including the absolute configurations, were elucidated by NMR and MS data, optical rotation, experimental and calculated electronic circular dichroism, induced circular dichroism, and X-ray crystal data. The isolated butenolides were evaluated for antibacterial, cytotoxic, and enzyme inhibitory activities. Compounds 7 and 12 displayed weak antibacterial activity, against *Enterococcus faecalis* (IC₅₀ = 25 µg/mL) and *Klebsiella pneumoniae* (IC₅₀ = 50 µg/mL), respectively, whereas 6 showed weak inhibitory effect on acetylcholinesterase. Nevertheless, most of the butenolides showed inhibition against pancreatic lipase (PL) with an inhibition rate of 21.2–73.0% at a concentration of 50 µg/mL.

Keywords: butenolides; enantiomers; Aspergillus terreus; pancreatic lipase

1. Introduction

Marine-derived fungi have been proven to be a valuable and rich source of novel and bioactive natural products [1]. Butenolides or butyrolactones, possessing the α , β -unsaturated γ -butyrolactone skeleton, were frequently isolated from fungi, especially the *Aspergillus* species [2]. Butenolides have been reported to show diverse biological activities, including anti-inflammatory, cytotoxic, antiviral, antioxidant, antimicrobial, antidiabetic, protein kinase-inhibitory, and α -glucosidase-inhibitory activities [2–4]. Furthermore, in vivo pharmacological, pharmacokinetics, and metabolism studies have been recently reported [3,5,6], indicating that they could be potential drug leads. Butenolides have also been reported as mycotoxins, possessing significant toxicity such as hepatic and renal oxidative damage [7]. For naturally occurring butenolides, racemic or scalemic mixtures are common [4]. Therefore, separation of racemic or scalemic mixtures to obtain enantiopure compounds is a very difficult task in natural products research. For a discovery of bioactive secondary metabolites, the coral-derived fungus *Aspergillus terreus* SCSIO41404 was investigated. Consequently, five undescribed butenolides (**1a**, **1b**, **2a**, **2b**, and **3**), ten previously reported compounds (**4–13**), and an undescribed *para*-hydroxybenzaldehyde derivative (**14**) were



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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). isolated and identified from the ethyl acetate (EtOAc) extract of the fungal fermentation (Figure 1). Mixtures of **1a/1b** and **2a/2b** were successfully separated by HPLC using a chiral column to afford two pairs of enantiomers. The first pair of enantiomers (**1a** and **1b**) are considered to be new natural products, since the mixture **1a/1b** has previously been obtained synthetically [8]. Herein, we report the isolation, structure determination, and preliminary antibacterial, cytotoxic, and enzyme inhibitory activity assays of these compounds.



Figure 1. Structures of (1-14).

2. Results and Discussion

The wheat culture of *Aspergillus* sp. SCSIO41404 was extracted with EtOAc. Several chromatographic methods, including silica gel column and semi-preparative HPLC with octadecylsilyl (ODS) column, were used for isolation of the 16 compounds (Figure 1).

Compound 1 was isolated as a yellow powder and its molecular formula $C_{16}H_{12}O_4$ was based on HRESIMS m/z 269.0811 [M + H]⁺ (calcd for $C_{16}H_{13}O_4$, 269.0808), accounting for eleven degrees of unsaturation. The ¹H NMR spectrum (Table 1) showed an A₂B₂ spin system of the *para*-disubstituted benzene ring [δ_H 7.17 (2H, d, *J* = 8.6 Hz, H-2', H-6') and 6.74 (2H, d, *J* = 8.6 Hz, H-3', H-5')], a monosubstituted benzene ring [δ_H 7.63 (2H, d, *J* = 7.4 Hz, H-2", H-6"), 7.28 (2H, t, *J* = 7.4 Hz, H-5") and 7.23 (1H, t, *J* = 7.4 Hz, H-4")], and a singlet at δ_H 6.26 (1H, H-4). Besides the above six corresponding hydrogen-bearing carbon signals [δ_C 130.6 (C-2'/6'), 116.7 (C-3'/5'), 128.9 (C-2"/6"), 129.3 (C-3"/5"), 129.4 (C-4"), 82.4 (C-4)], six carbons remained in the ¹³C NMR spectrum, including an ester carbonyl [δ_C 171.5 (C-1)] and five non-protonated sp² carbons [δ_C 140.2 (C-2), 132.2 (C-3),

128.4 (C-1'), 159.7 (C-4'), and 129.5 (C-1")]. The carbon signals ($\delta_{\rm C}$ 171.5, 140.2, 132.2, 82.4) and the HMBC correlations from H-4 to C-1, C-2 and C-3, indicated the presence of the butenolide skeleton (Figure 2) [9]. These data showed a close similarity to a synthetic racemic furanone derivative (3-hydroxy-5-(4-hydroxyphenyl)-4-phenyl-2(5H)-furanone, **5d** in the reference) [8]. Moreover, the other key HMBC correlations (Figure 2), from H-4 to C-1' and C-1", from H-2'/H-6' to C-4 ($\delta_{\rm C}$ 82.4) and C-4', confirmed a planar structure of **1**, which was assigned the trivial name asperteretal G (**1**).

	1		2		3	
No	$\delta_{ m C}$, Type	$\delta_{ m H}$ (J in Hz)	δ_{C} , Type	$\delta_{ m H}$ (J in Hz)	$\delta_{ m C}$, Type	$\delta_{ m H}$ (J in Hz)
1	171.5, C		172.5, C		175.1, C	
2	140.2, C		138.5, C		125.7, C	
3	132.2, C		141.0, C		158.5, C	
4	82.4, CH	6.26, s	83.0, CH	5.72, s	99.2, CH	6.48, s
5					30.0, CH ₂	3.72, overlapped 3.80, d (15.7)
1'	128.4, C		127.6, C		123.3, C	
2′	130.6, CH	7.17, d (8.6)	130.1, CH	7.09, d (8.6)	131.7, CH	7.48, d (8.8)
3′	116.7, CH	6.74, d (8.6)	116.5, CH	6.79, d (8.6)	116.6, CH	6.83, d (8.8)
4'	159.7, C		159.7, C		160.9, C	
5'	116.7, CH	6.74, d (8.6)	116.5, CH	6.79, d (8.6)	116.6, CH	6.83, d (8.8)
6'	130.6, CH	7.17, d (8.6)	130.1, CH	7.09, d (8.6)	131.7, CH	7.48, d (8.8)
1″	129.5, C		27.4, CH	2.49, hept (7.0)	130.6, C	
2″	128.9, CH	7.63, d (7.4)	21.0, CH ₃	0.99, d (7.0)	130.3, CH	6.92, d (2.3)
3″	129.3, CH	7.28, t (7.4)	20.1, CH ₃	1.08, d (7.0)	121.4, C	
4″	129.4, CH	7.23, t (7,4)			153.0, C	
5″	129.3, CH	7.28, t (7.4)			118.1, CH	6.66, d (8.4)
6″	128.9, CH	7.63, d (7.4)			128.2, CH	6.95, dd (8.4, 2.3)
7″					32.2, CH ₂	2.66, dd (16.6, 7.3) 2.95, dd (16.6, 4.8)
8″					70.5, CH	3.72, overlapped
9″					78.0 <i>,</i> C	11
10″					25.8, CH ₃	1.30, s
11″					21.2, CH ₃	1.23, s

Table 1. ¹H NMR (700 MHz) and ¹³C NMR (175 MHz) Data for 1–3 in CD₃OD.

Although 1 has one stereogenic carbon (C-4), its optical rotation was close to zero and the inapparent Cotton effect in the electronic circular dichroism (ECD) spectrum suggested that 1 was not enantiomerically pure. Therefore, 1 was subject to a separation by HPLC using a chiral column CHIRALPAK IC, eluted with isopropanol (IPA)/n-hexane (Hex): 18/82, to obtain two pure enantiomers (1a and 1b, two well-separated peaks in a ratio of approximately 1:1) (Figure 3). The enantiomers **1a** and **1b** possessed opposite Cotton effects in the ECD spectra, and their absolute configurations were determined by comparison of the experimental ECD spectra with those calculated for each enantiomer (Figure 4A,C). The ECD spectra of 1a and 1b showed positive and negative Cotton effects in the regions of $\pi \rightarrow \pi^*$ transition (260–280 nm), respectively, indicating the 4S and 4R absolute configurations for 1a and 1b, respectively, since 1a was obtained as a suitable crystal, and its X-ray analysis was performed. The structure of **1a** and the absolute configuration of C-4 were confirmed as shown in the ortep diagram (Figure 5). The crystal data collected on a XtalLAB PRO single-crystal diffractometer using Cu K α radiation are in Table S1. Therefore, this pair of enantiomers, as new natural products, were obtained for the first time, and named (+)-asperteretal G (1a) and (-)-asperteretal G (1b).



Figure 2. Key COSY and HMBC correlations in (1-3) and (14).



Figure 3. HPLC chromatograms of (\pm) -1 and (\pm) -2.

Compound **2** was obtained as a colorless oil. Its molecular formula was established as $C_{13}H_{14}O_4$ by HRESIMS m/z 235.0969 [M + H]⁺ (calcd for $C_{13}H_{15}O_4$, 235.0965), accounting for seven degrees of unsaturation. In the ¹H NMR spectrum (Table 1) of **2**, the proton signals of a *para*-disubstituted benzene ring [δ_H 7.09 (2H, d, *J* = 8.6 Hz, H-2', H-6') and 6.79 (2H, d, *J* = 8.6 Hz, H-3', H-5')] and an isopropyl group [2.49 (1H, hept, *J* = 7.0 Hz, H-1"), 0.99 (3H, d, *J* = 7.0 Hz, H-2"), and 1.08 (3H, d, *J* = 7.0 Hz, H-3")] were clearly observed. Analysis of the NMR data (Table 1) indicated that **2** had a similar skeleton to that of **1**, except for the phenyl group attached to C-3 in **1** was replaced by an isopropyl group, which was confirmed by HMBC correlations from H₃-2" and H₃-3" to C-3 (Figure 2). HMBC correlations from H-4 (δ_H 5.72, s) to C-1 (δ_C 172.5), C-3 (δ_C 141.0) and C-2 (δ_C 138.5), from H-2'/H-6' to C-4 (δ_C 83.0) and C-4' (δ_C 159.7), and COSY correlations from H-1" to H₃-2",



H-1" and H₃-3", confirmed the structure of **2**. Therefore, the planar structure of **2**, named asperteretal H (**2**), was elucidated as shown.

Figure 4. Experimental and calculated ECD spectra of 1a, 1b (A,C), 2a, and 2b (B,D).



Figure 5. The ortep view of 1a.

Similarly, the optical rotation and the inapparent ECD Cotton effect also suggested that **2** was not enantiomerically pure. The enantiomers **2a** and **2b** were also successfully separated by HPLC with a chiral column CHIRALPAK IC, eluted with IPA/Hex: 12/88 (two well-separated peaks in a ratio of approximately 1:1) (Figure 3). Enantiomers **2a** and **2b** showed opposite Cotton effects in the ECD spectra, and their absolute configurations were also determined by comparison of the experimental with the calculated ECD spectra for each enantiomer (Figure 4B,D). The positive and negative Cotton effects of **2a** and **2b** showed opposite Cotton effects of **2a** and **2b** showed by comparison of the experimental with the calculated ECD spectra for each enantiomer (Figure 4B,D).

in the region near 250 nm, respectively, supporting the 4R and 4S absolute configurations for **2a** and **2b**, respectively [10,11]. Thus, this pair of enantiomers were identified as (+)-asperteretal H (**2a**) and (-)-asperteretal H (**2b**).

Compound **3** was obtained as a yellow oil. Its molecular formula $C_{22}H_{22}O_6$ was determined based on HRESIMS m/z 381.1340 [M – H]⁻ (calcd for C₂₂H₂₁O₆, 381.1344), accounting for twelve degrees of unsaturation. The ¹H NMR spectrum (Table 1) revealed diagnostic signals of a *para*-disubstituted [δ_H 7.48 (2H, d, J = 8.8 Hz, H-2', H-6') and 6.83 (2H, d, J = 8.8 Hz, H-3', H-5') and a 1,3,4-trisubstituted benzene rings [$\delta_H 6.92$ (d, J = 2.3 Hz, H-2"), 6.66 (d, J = 8.4 Hz, H-5") and 6.95 (dd, J = 8.4, 2.3 Hz, H-6")], two oxygen-bearing methines [$\delta_{\rm H}$ 6.48 (s, H-4), 3.72 (H-8")], two methylenes [$\delta_{\rm H}$ 3.80 (d, J = 15.7 Hz H-5a), 3.72 (H-5b), 2.95 (dd, *J* = 16.6, 4.8 Hz, H-7"a) and 2.66 (dd, *J* = 16.6, 7.3 Hz, H-7"b)], and two methyls [$\delta_{\rm H}$ 1.30 (3H, s, H-10"), 1.23 (3H, s, H-11")]. The ¹³C NMR spectrum (Table 1), in combination with DEPT and HSQC spectra, revealed the presence of 20 carbon resonances including two methyls [$\delta_{\rm C}$ 25.8 (C-10"), 21.2 (C-11")], two sp³ methylenes [$\delta_{\rm C}$ 32.2 (C-7"), 30.0 (C-5)], eight protonated sp² ($\delta_{\rm C}$ 131.7, 131.7, 130.3, 128.2, 118.1, 116.6, 116.6, 99.2), one sp³ oxygenated methine ($\delta_{\rm C}$ 70.5) and one ester carbonyl ($\delta_{\rm C}$ 175.1), seven on-protonated sp^2 (δ_C 160.9, 153.0, 158.5, 130.6, 125.7, 123.3, 121.4), and one sp^3 oxygenated methine ($\delta_{\rm C}$ 78.0). Its 1D-NMR data were similar to those of asperteretal E [4], except that H₂-8" of asperteretal E was replaced by a hydroxyl group (C-8", $\delta_{H/C}$ 3.72/70.5). COSY correlation from H_2 -7" to H-8" and HMBC correlations from H-8" to C-3" and H_2 -7", H_3 -10", H_3 -11" to C-8" confirmed a planar structure of **3**. Compound **3** was named asperteretal I. Since there are two stereogenic carbons (C-4 and C-8"), there are four possible stereomers for 3. The inapparent ECD Cotton effect (Figure S1) spectra suggested that 3 was not enantiomerically pure [4]. Our effort to separate the isomers of 3 with various chiral columns and mobile phase systems was not successful. Therefore, the absolute configurations of C-4 and C-8" in 3 were still undetermined.

Compound **14** was isolated as a yellow oil and had a molecular formula $C_{12}H_{16}O_4$ as determined by HRESIMS m/z 223.0976 $[M - H]^-$ (calcd for $C_{12}H_{15}O_4$, 223.0976), accounting for five degrees of unsaturation. Analysis of the NMR data indicated that **14** possessed an aldehyde group (δ_H 9.75, s, δ_C 193.1, CHO-7), a 1,2,4-trisubstituted benzene ring (δ_H 7.75, d, J = 2.0 Hz, δ_C 134.8, CH-2; δ_H 7.64, dd, J = 8.3, 2.0 Hz, δ_C 131.5, CH-6; δ_H 6.91, d, J = 8.3 Hz, δ_C 116.7, CH-5), an oxygen-bearing sp³ methine (δ_H 3.63, dd, J = 10.4, 1.8 Hz, δ_C 79.4, CH-9), a sp³ methylene (δ_H 3.07, dd, J = 14.1, 1.8 Hz, 2.60, dd, J = 14.1, 10.4 Hz, δ_C 33.8, CH₂-8), and two methyls (δ_H 1.25, 6H, s, δ_C 25.7, 25.1, CH₃-11 and CH₃-12). These data were similar to those of 3-(2,3-dihydroxy-isopentyl)-4-hydroxy-acetophenone [12] except for the replacement of the acetyl group on C-1 of 3-(2,3-dihydroxy-isopentyl)-4-hydroxy-acetophenone with an aldehyde group (CHO-7) in **14** [12]. Moreover, the HMBC correlations from H-6 to C-7 and H-7 to C-2 further determined the structure (Figure 1). The absolute configuration of C-9 was determined to be *S* by using the Mo₂(OAc)₄-induced circular dichroism method (a positive cotton effect at 314 nm) (Figure 6) [13]. Thus, **14** was identified as (*S*)-3-(2,3-dihydroxy-3-methylbutyl)-4-hydroxybenzaldehyde.

The previously reported compounds were identified as butyrolactone II (4) [14], methyl I-2-benzyl-4-hydroxy-3-(4-hydroxyphenyl)-5-oxo-2,5-dihydrofuran-2-carboxylate (5) [8,15], butyrolactone I (6) [16], versicolactone B (7) [17], aspernolide D (8) [18], aspernolide A (9) [19], butyrolacton V (10) [16], terrelactone (11) [20], butyrolactone VI (12) [21], and butyrolactone IV (13) [16] by comparison of their spectroscopic data (Supporting Information, Figures S26–S35) with those reported in the literature.

Compounds 1–14 were evaluated for their cytotoxic activity against human lung carcinoma (A549) and human hepatocellular carcinoma (HepG2) cell lines, but none showed cytotoxicity at a concentration of 50 μ M. Most of the compounds were screened for their enzyme inhibitory effects against pancreatic lipase (PL) and acetylcholinesterase (AChE) in vitro, and also antibacterial activity against five pathogenic bacteria, *Staphylococcus aureus* ATCC 29213, *Enterococcus faecalis* ATCC 29212, *Klebsiella pneumoniae* ATCC 13883, methicillin-resistant *S. aureus* (MRSA, clinical strain), and methicillin-resistant *S. epidermidis*

(MRSE, clinical strain) (Table 2). Only 7 and 12 displayed weak antibacterial activity against *E. faecalis* (IC₅₀ value of 25 μ g/mL) and *K. pneumoniae* (IC₅₀ value of 50 μ g/mL), respectively. Compound 6 displayed weak inhibitory effect on AChE (35.2%, at 50 μ g/mL concentration). Nevertheless, most of the butenolides (1a/1b, 2a/2b, 3, 6, 7, 10, 12, and 13) showed inhibition against PL with inhibition rate of 21.2–73.0% at a concentration of 50 μ g/mL (Table 2).



Figure 6. Mo₂(Oac)₄-induced CD spectrum of 14.

Comp.	Enzyme Inhibition	Rate at 50 µg/mL (%)	Antibacterial Activities (MIC, µg/mL)		
	PL	AChE	E. faecalis	K. pneumoniae	
1a/1b	58.8	<10	>100	>100	
2a/2b	67.2	<10	>100	>100	
3	35.5	<10	>100	>100	
4	<10	<10	>100	>100	
5	<10	<10	>100	>100	
6	37.6	35.2	>100	100	
7	73.0	<10	25	>100	
10	54.1	<10	>100	>100	
11	<10	<10	>100	>100	
12	21.2	<10	>100	50	
13	66.8	<10	>100	>100	
Control	86.5 ^a	83.7 ^b	4 ^c	0.5 ^c	

Table 2. The enzyme inhibitory and antibacterial activities of 1–7 and 10–13.

^a Orlistat, ^b tacrine, and ^c ampicillin were used as positive controls.

3. Materials and Methods

3.1. General Experimental Procedures

UV spectra were measured on a Shimadzu UV-2600 PC spectrophotometer (Shimadzu, Kyoto, Japan). Optical rotations were recorded on a PerkinElmer MPC 500 (Waltham, MA, USA) polarimeter. For ECD spectra, Chirascan circular dichroism spectrometer (Applied Photophysics, Leatherhead Surrey, UK) was used. NMR spectra were acquired by a Bruker Avance spectrometer (Bruker, Billerica, MA, USA) at 700 MHz for ¹H and 175 MHz for ¹³C. HRESIMS spectra were recorded on a Bruker miXis TOF-QII mass spectrometer (Bruker, Billerica, MA, USA). X-ray diffraction intensity data were measured on an Agilent Xcalibur Nova single-crystal diffractometer (Santa Clara, CA, USA) using Cu K α radiation. TLC and column chromatography were performed on plates precoated with silica gel GF₂₅₄ (10–40 µm) and over silica gel (200–300 mesh) (Qingdao Marine Chemical Factory, Qingdao, China), respectively. Spots were detected on TLC (Qingdao Marine Chemical Factory, Qingdao, China) under 254 nm UV light. Semi-preparative HPLC was performed using an ODS column (YMC-pack ODS-A, YMC Co., Ltd., Kyoto, Japan, 10 mm × 250 mm, 5 µm).

3.2. Fungal Material

The fungal strain *Aspergillus* sp. SCSIO41404 was isolated from a soft coral *Sinularia* sp., collected in the Luhuitou waters ($109^{\circ}29'37.3''$ E, $18^{\circ}11'33.4''$ N) of Sanya Bay in the South China Sea, in June 2019. The strain was stored on Muller Hinton broth (MB) agar (malt extract 15 g, sea salt 10 g, agar 15 g, H₂O 1 L, pH 7.4–7.8) at 4 °C and deposited in the CAS Key Laboratory of Tropical Marine Bio-resources and Ecology, South China Sea Institute of Oceanology, Chinese Academy of Sciences, Guangzhou, China. The strain was identified as *Aspergillus terreus* based on the ITS region of the rDNA (GenBank accession No. KU866665.1) (Table S2).

3.3. Fermentation and Extraction

The seed medium (malt extract 15 g, sea salt 10 g, H_2O 1 L, pH 7.4–7.8) in 500 mL Erlenmeyer flasks (150 mL/flask) was incubated at 28 °C for 3 days on a rotating shaker (180 rpm). The seed medium was added to the wheat fermentation medium (wheat 200 g, sea salt 10g, H_2O 200 mL) in a 1000 mL Erlenmeyer flask. In total, 40 Erlenmeyer flasks were incubated for 30 days at 25 °C without shaking. The whole wheat cultures were crushed and extracted with EtOAc three times to afford an organic extract.

3.4. Isolation and Purification

The EtOAc extract (638.1 g) was subjected to silica gel column chromatography eluted with PE-EtOAc-MeOH (50:1:0 to 0:0:1, v/v) in gradient to yield seven fractions (Frs.1–7). Fr.1 was subjected to HPLC with an ODS column, eluted with MeOH/H₂O (5–100%) to afford six subfractions (Frs.1-1–1-6). Fr.1-4 was purified by semipreparative HPLC (47% MeOH/H₂O, 2 mL/min) to yield **5** (4.8 mg). Fr.2 was subjected to MPLC with an ODS column, eluted with MeOH/H₂O (10–100%) to afford eight subfractions (Frs.2-1–2-8). Fr.2-3 was purified by semipreparative HPLC (60% MeOH/H₂O, 2 mL/min) to yield **1** (5.9 mg) and **2** (8.5 mg). Fr.3 was purified by semipreparative HPLC (60% MeOH/H₂O, 2 mL/min) to yield **7** (10.2 mg). Fr.4 gave predominantly **6** (25 g). Fr.5 was purified by semipreparative HPLC (65% MeOH/H₂O, 2 mL/min) to give **4** (8.9 mg), **9** (3.2 mg), **10** (17.3 mg), and **13** (5.3 mg). Fr.6 was subjected to MPLC with an ODS column, eluting with MeOH/H₂O (5%100%) to give eight subfractions (Frs.6-1–6-8). Fr.6-4 was purified by semipreparative HPLC (42% MeOH/H₂O, 2 mL/min) to yield **14** (2.5 mg). Fr.6-6 was purified by semipreparative HPLC (42% MeOH/H₂O, 2 mL/min) to yield **3** (1.6 mg), **8** (0.81 mg), **11** (5.4 mg), and **12** (10.2 mg).

Asperteretal G (1): yellow powder; UV (MeOH) λ_{max} (log ε) = 203 (4.37), 221 (4.21), 226 (4.20), 286 (4.17), 327 (3.80) nm; ¹H and ¹³C NMR data, Table 1; HRESIMS *m*/*z* 269.0811 [M + H]⁺ (calcd for C₁₆H₁₃O₄, 269.0808) (Figures S2–S9).

(+)-asperteretal G (1a): $[\alpha]_D^{25}$ +36.9 (c 0.1, MeOH); ECD (0.2 mg/mL, IPA) λ_{max} ($\Delta \varepsilon$) 223 (+5.56), 237 (-4.04), 265 (+3.67), 307 (-2.51) nm.

(-)-asperteretal G (1b): $[\alpha]_D^{25}$ -23.7 (c 0.1, MeOH); ECD (0.2 mg/mL, IPA) λ_{max} ($\Delta \varepsilon$) 223 (-3.57), 239 (+3.40), 263 (-4.23), 308 (+2.64) nm.

Asperteretal H (2): colorless oil; UV (MeOH) λ_{max} (log ε) = 202 (4.28), 230 (4.33), 277 (3.53), 283 (3.51), 310 (3.42) nm; ¹H and ¹³C NMR data, Table 1; HRESIMS *m*/*z* 235.0969 [M + H]⁺ (calcd for C₁₃H₁₅O₄, 235.0965) (Figures S10–S17).

(+)-asperteretal H (2a): $[\alpha]_D^{25}$ +68.2 (c 0.1, MeOH); ECD (0.2 mg/mL, IPA) λ_{max} ($\Delta \varepsilon$) 210 (-3.59), 247 (+8.08), 280 (-1.76) nm.

(-)-asperteretal H (**2b**): $[\alpha]_D^{25}$ -39.7 (c 0.1, MeOH); ECD (0.2 mg/mL, IPA) λ_{max} ($\Delta \varepsilon$) 210 (+4.10), 243 (-10.27), 281 (+1.24) nm.

Asperteretal I (3): yellow oil; $[\alpha]_D^{25}$ +8.0 (*c* 0.1, MeOH); UV (MeOH) λ_{max} (log ε) = 203 (4.45), 305 (4.09) nm; ¹H and ¹³C NMR data, Table 1; HRESIMS *m*/*z* 381.1340 [M - H]⁻ (calcd for C₂₂H₂₁O₆, 381.1344) (Figures S18–S25).

(S)-3-(2,3-Dihydroxy-3-methylbutyl)-4-hydroxybenzaldehyde (14): yellow oil; $[\alpha]_D^{25}$ +7.7 (c 0.10, MeOH); UV (MeOH) λ_{max} (log ε) = 203 (3.88), 224 (3.49), 276 (3.13) nm; ¹H NMR (700 MHz, CD₃OD) δ_H 9.75 (1H, s, H-7), 7.75 (1H, d, *J* = 2.0 Hz, H-2), 7.64 (1H, dd, *J* = 8.3,

2.0 Hz, H-6), 6.91 (1H, d, J = 8.3 Hz, H-5), 3.63 (1H, dd, J = 10.4, 1.8 Hz, H-9), 3.07 (1H, dd, J = 14.1, 1.8 Hz, H_a-8), 2.60 (1H, dd, J = 14.1, 10.4 Hz, H_b-8), 1.25 (6H, s, H-11, H-12); ¹³C NMR (175 MHz, CD₃OD) $\delta_{\rm C}$ 193.1 (CH, C-7), 164.0 (qC, C-4), 134.8 (CH, C-2), 131.5 (CH, C-6), 130.1 (qC, C-1), 129.1 (qC, C-3), 116.7 (CH, C-5), 79.4 (CH, C-9), 73.9 (qC, C-10), 33.8 (CH₂, C-8), 25.7 (CH₃, C-12), 25.1 (CH₃, C-11); HRESIMS m/z 223.0976 [M – H]⁻ (calcd for C₁₂H₁₅O₄, 223.0976) (Figures S36–S43).

3.5. X-ray Crystallographic Analysis

Colorless crystals of **1a** were obtained in MeOH/CHCl₃ (1:2) followed by slow evaporation at 4 °C and the crystals' data were collected on a XtalLAB PRO single-crystal diffractometer using Cu K α radiation. The X-ray crystal structure of **1a** was solved using SHELXS97, expanded by difference Fourier techniques, and refined by full-matrix least-squares calculation finally. All non-hydrogen atoms were refined anisotropically, and hydrogen atoms were fixed at calculated positions. Crystallographic data of **1a** (Table S1) have been deposited in the Cambridge Crystallographic Data Centre (deposition number: CCDC 2077455). These data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB21EZ, UK [fax: +44(0)-1223-336033 or e-mail: deposit@ccdc.cam.ac.uk].

3.6. Mo₂(AcO)₄-Induced Circular Dichroism

 $Mo_2(AcO)_4$ (1.0 mg) and 14 (1.0 mg) were dissolved in dimethyl sulfoxide (DMSO) (1 mL) to use as stock solutions, which were mixed by 1:2 v/v. After mixing, the CD spectrum was recorded immediately and scanned every 5 min, until a stationary ICD spectrum (the CD of 14 in $Mo_2(AcO)_4$ solution subtracted from inherent CD of 14 in DMSO) was observed. The sign of the diagnostic band at around 310 nm in the ICD spectrum was correlated to the absolute configuration of C-9 of 14 [22].

3.7. ECD Calculation

The structures of **1a**, **1b**, **2a**, and **2b** were subjected to random conformational searches using the Spartan'14 software with the MMFF method, as used previously [23]. The conformers with a Boltzmann population of over 5% were chosen for ECD calculations using the Gaussian 09 software [24], and the stable conformers were initially optimized at the B3LYP/6-31+G(d,p) level in MeOH using the CPCM model. The overall theoretical calculation of ECD was achieved in MeOH using time-dependent density functional theory at the B3LYP/6-31+G (d, p) level. The ECD spectra were generated using the SpecDis 1.6 (University of Würzburg, Würzburg, Germany) and Prism 5.0 (GraphPad Software Inc., San Diego, CA, USA) software with a half-bandwidth of 0.3–0.4 eV, according to the Boltzmann-calculated contribution of each conformer after UV correction.

3.8. Bioassay

The cytotoxic activity of the obtained compounds was evaluated by the MTT method as reported in our previous study [25]. The antibacterial activity against five bacterial strains, *Staphylococcus aureus* ATCC 29213, *Enterococcus faecalis* ATCC 29212, *Klebsiella pneumoniae* ATCC 13883, methicillin-resistant *S. aureus* (MRSA, clinical strain), and methicillin-resistant *S. epidermidis* (MRSE, clinical strain) was evaluated using a modified broth microdilution method [25,26]. The acetylcholinesterase and PL inhibitory activities were evaluated according to the methods used in our previous study [26,27].

4. Conclusions

We describe here the isolation and structure elucidation of five undescribed and ten previously reported butenolides from the coral-derived fungus *Aspergillus terreus* SCSIO41404, together with an undescribed *para*-hydroxybenzaldehyde derivative (*S*)-3-(2,3-dihydroxy-3-methylbutyl)-4-hydroxybenzaldehyde. Two pairs of enantiomers were separated by HPLC using a chiral column, and enantiomers (+)-asperteretal G (**1a**) and (–)-asperteretal G (**1b**)

were new natural products. After preliminary antibacterial, cytotoxic and enzyme inhibitory bioassays of these compounds, several natural butenolides showed inhibition against PL. This is the first report of the PL inhibitor activity of butenolides, and the effects of these butenolides on the regulation of lipid metabolism deserve further study [28].

Supplementary Materials: The following supporting information can be downloaded at: https: //www.mdpi.com/article/10.3390/md20030212/s1, ITS sequence data of the strain; physicochemical data of **4–13**; Table S1: X-ray crystallographic data of **1**; Table S2: Primer sequences for the genes; Figure S1: Experimental ECD spectrum of **3**; Figures S2–S25: NMR, HRESIMS and UV spectroscopic data of **1–3**; Figures S26–S35: ¹H NMR spectra of **4–13**; Figures S36–S43: NMR, HRESIMS and UV spectroscopic data of **14** (PDF); Crystallographic data of **1** (CIF).

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