



# Article On The Microstructures and Hardness of The Nb-24Ti-18Si-5Al-5Cr-5Ge and Nb-24Ti-18Si-5Al-5Cr-5Ge-5Hf (at.%) Silicide Based Alloys

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**Abstract:** The microstructures and hardness of the as cast and heat treated (1400 °C/100 h) alloys Nb-24Ti-18Si-5Ge-5Cr-5Al (ZF6) and Nb-24Ti-18Si-5Ge-5Cr-5Al-5Hf (ZF9) were studied. Both alloys were compared with refractory metal bcc solid solution + intermetallic High Entropy Alloys (HEAs). There was macrosegregation of Si, Ti, Cr and Al in both alloys. The roles of Ge and Hf on macrosegregation are discussed. In both alloys the primary phase was the  $\beta$ Nb<sub>5</sub>Si<sub>3</sub>. In the as cast alloy ZF6 the Nb<sub>ss</sub>,  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> and C14-NbCr<sub>2</sub> Laves phase and Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectic were formed. The microstructure of the as cast alloy ZF9 consisted of Nb<sub>ss</sub>,  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> and C14-NbCr<sub>2</sub> Laves phase. The heat-treated microstructures of the alloys ZF6 and ZF9 consisted of Nb<sub>ss</sub>,  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> and C14-NbCr<sub>2</sub> Laves phase. The heat-treated microstructures of the alloys ZF6 and ZF9 consisted of Nb<sub>ss</sub>,  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> and C14-NbCr<sub>2</sub> Laves phase. The heat-treated microstructures of the alloys ZF6 and ZF9 consisted of Nb<sub>ss</sub>,  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> and  $\alpha$ Nb<sub>5</sub>Si<sub>3</sub> and Nb<sub>5</sub>Si<sub>3</sub>,  $\alpha$ Nb<sub>5</sub>Si<sub>3</sub> and  $\gamma$ Nb<sub>5</sub>Si<sub>3</sub>, respectively. The surfaces of both alloys were contaminated by oxygen where TiO<sub>2</sub> and HfO<sub>2</sub> formed respectively in the alloys ZF6 and ZF9 and Nb<sub>5</sub>Si<sub>3</sub>. The roles of alloying additions on the hardness of the Nb<sub>ss</sub> and Nb<sub>5</sub>Si<sub>3</sub> and relationships between alloy hardness and alloy parameters VEC (valence electron concentration),  $\delta$  (related to atomic size) and  $\Delta\chi$  (related to electronegativity) were discussed.

**Keywords:** Nb-silicide based alloys; high entropy alloys; complex concentrated alloys; macrosegregation; intermetallics; hardness

## 1. Introduction

The search for new alloys capable of operating in the demanding conditions in gas turbine engines at temperatures higher than those experienced by the currently used Ni-based superalloys has concentrated on refractory metal silicide-based alloys, in particular Nb-Si based alloys and Mo-Si based alloys [1,2]. The former also are known as Nb-silicide based alloys or Nb in situ composites and have microstructures consisting of Nb<sub>ss</sub> and intermetallic(s), such as Nb<sub>5</sub>Si<sub>3</sub>, Nb<sub>3</sub>Si, NbCr<sub>2</sub> and others, depending on alloy chemistry and processing [2,3].

The tetragonal Nb<sub>5</sub>Si<sub>3</sub> silicide is most desirable compared with Nb<sub>3</sub>Si owing to its better creep properties and oxidation resistance. Alloying additions can enhance the stability of the tetragonal  $\beta$ Nb<sub>5</sub>Si<sub>3</sub>, suppress the Nb<sub>ss</sub> + Nb<sub>3</sub>Si eutectic and replace it by the Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectic, enhance the eutectoid decomposition of Nb<sub>3</sub>Si and modify the properties and morphology of the silicides [3,4].

Important alloying additions in Nb-silicide based alloys are Al, Cr, Ti and Hf. These four elements are essential for oxidation resistance [3,5], the first two are expected to confer oxidation resistance with Si, and Hf scavenges oxygen and forms hafnia. The solid solubilities of Al, Cr and Hf in the Nb<sub>ss</sub> increase with increasing Ti content and all four elements strengthen the Nb<sub>ss</sub>. Aluminium and Cr

increase the ductile to brittle transition temperature (DBTT) of bcc Nb. Aluminium has a stronger effect than Cr, and Ti and Hf have a weak effect on DBTT [6]. Alloying with Cr and Ti respectively increases and decreases the shear modulus of Nb. Aluminium and Cr additions in Nb or Nb-Ti increase the Peierls-Nabarro energy and thus reduce dislocation mobility, tensile ductility and thereby decrease the resistance to fracture of the Nb<sub>ss</sub> [7].

Titanium and Hf substitute Nb in Nb<sub>5</sub>Si<sub>3</sub> and can destabilise its desirable tetragonal structure [2,3], and decrease its hardness [8]. In Nb<sub>5</sub>Si<sub>3</sub> the solid solubility of Cr is low, and Si can be substituted by Al [8]. Germanium, like Al, substitutes for Si in Nb<sub>5</sub>Si<sub>3</sub> [9] and in synergy with Ti and Hf improves the fracture toughness of Nb<sub>5</sub>Si<sub>3</sub> from about 3 MPa $\sqrt{m}$  to 8–11 MPa $\sqrt{m}$  for 24 at.% Ti in solution [10,11].

Our research group has shown that in Nb-24Ti-18Si-5Al-5Cr based alloys, the Al and Cr in synergy with Ti stabilise the Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectic, and Al in synergy with Ti enhances the transformation of  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> to  $\alpha$ Nb<sub>5</sub>Si<sub>3</sub> and promotes the precipitation of Nb<sub>ss</sub> via the transformation  $\beta$ Nb<sub>5</sub>Si<sub>3</sub>  $\rightarrow \alpha$ Nb<sub>5</sub>Si<sub>3</sub> + Nb<sub>ss</sub> [12]. The latter transformation was enhanced even in the absence of Ti when Al and Ge were in synergy with Si in the alloy Nb-18Si-5Al-5Ge [13]. Cretegny et al. have attributed the increased strength of MASC (Nb-24Ti-16Si-8.2Hf-2Cr-1.9Al) type alloys (compared with Nb-25Ti-8Hf-xSi alloys) to the propensity of Al and Cr additions to favour eutectic growth compared with the simpler quaternary alloys [14].

The authors have shown that Ge promotes the Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectic and microstructures similar to those achieved in Nb-silicide based alloys where Ti was in synergy with Al and Cr [13,15]. Considering the reported beneficial effect of Ge on the oxidation of Nb-silicide based alloys [5] and the restrictions imposed on the Al, Cr and Hf contents of Nb-24Ti-18Si-zxAl-yCr-zHf based alloys owing to mechanical property targets, we embarked on the work presented in this paper in order to understand how the synergy of Ge with the alloying elements Al, Cr, Ti and Hf at the concentrations of the "base" alloy Nb-24Ti-18Si-5Al-5Cr affects the stability of  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> and NbCr<sub>2</sub>, and the vol.% of Nb<sub>ss</sub> and the shape and size of Nb<sub>5</sub>Si<sub>3</sub>. This knowledge is important for optimising room and elevated temperature mechanical properties and oxidation resistance [3]. The effects of the synergy of the above elements on the oxidation behaviour of alloys based on Nb-24Ti-18Si-5Al-5Cr will be the subject of a separate publication. In this paper we report on two alloys with Ge and Hf additions. The results for each alloy are presented separately. The as solidified and heat-treated microstructures are compared with similar alloys without Ge. The nominal compositions of the latter alloys are given in the Appendix A.

#### 2. Experimental

The nominal compositions of the alloys of this study (all compositions are given in at.%) were Nb-24Ti-18Si-5Al-5Cr-5Ge (alloy ZF6) and Nb-24Ti-18Si-5Al-5Cr-5Ge-5Hf (alloy ZF9). The Ge, Si and Ti concentrations were set respectively at 5, 18 and 24 at.%, as in previous research [12,13]. The DBTT of aero engine materials must not exceed  $-50 \degree C$  [3]. The Al composition was kept at 5 at.% because the DBTT of Nb increases above  $-50 \degree C$  at higher Al concentrations [6] and the Cr composition was kept at 5 at.% as in previous work [12,15]. This choice of alloys makes possible their comparison with non-Ge containing alloys regarding stability of microstructure (this work) and oxidation (separate publication).

We used arc melting and high purity elements (Nb 99.99 wt.%, Ti 99.95 wt.%, Si 99.999 wt.%, Ge 99.999 wt.%, Cr 99.5 wt.%, Hf 99.99% wt.% and Al 99.999% wt.%) to prepare the alloys. The elemental charges were placed in a copper water-cooled crucible and were melted in a high purity argon atmosphere using a tungsten electrode. Each alloy was melted five times to homogenize its composition as much as possible. A tube furnace (Lenton Furnaces, Market Harborough, UK) was used for the heat treatments. The specimens for the latter were cut from the bulk of the buttons, wrapped in Ta foil and placed in an alumina crucible. They were heat treated at 1400 °C for 100 h under an argon atmosphere. Pure Ti sponge was used in the tube furnace as oxygen getter.

The microstructures of the cast and heat-treated alloys were characterized using X-ray diffraction (XRD), scanning electron microscopy (SEM) with micro-analysis and electron microprobe analysis (EPMA). For XRD we used a Siemens D500 X-ray diffractometer (HiltonBrooks Ltd, Crew, UK) with

CuK<sub> $\alpha$ </sub> radiation ( $\lambda$  = 1.540562 Å), an acceleration voltage of 40 kV, a current of 30 mA and a step of 0.02 degrees per second. The phases were identified using ICDD (International Centre for Diffraction Data) database. The XRD experiments used bulk specimens and thus texture effects were likely. The lattice parameters of the Nb<sub>ss</sub> were calculated using the Nelson-Riley extrapolation method to an accuracy of 0.1%. For imaging and micro-analysis we used a Camscan Mk2 SEM (Camscan Electron Optics Ltd., Cambridge, UK) equipped with an energy dispersive X-ray spectrometer (EDS) and a CAMECA SX-51 Electron Microprobe (EPMA) (CAMECA SAS, Gennevilliers Cedex, France) with high purity standards of Nb, Ti, Si, Ge, Cr, Hf and Al elements that were ground and polished to 1  $\mu$ m finish. High purity Co was used for calibration prior to EDS analysis. An accelerating voltage of 20 kV was used and the electron beam size was 1  $\mu$ m. The count rate for the calibration was controlled to be about 3000 cps by adjusting the probe current. The ZAF correction method was used.

Specimens from the top, bulk and bottom of the buttons were used for micro-analysis. In each specimen we did large area (0.5 mm × 0.5 mm) analysis, point (phase) analysis and area analysis of eutectic microstructures. Large area analysis data was used to study macrosegregation, which is common in arc melted Nb-silicide based alloys [16]. At least 10 point analyses were performed on phases with size  $\geq 5 \ \mu m$  and at least 5 large area analyses were taken from each of the top, bulk and bottom areas of the buttons to determine actual compositions. In the results section, the analysis data is given with the average, minimum and maximum values and the standard deviation.

The macrohardness of the alloys was measured using a CV Instruments Vickers hardness tester (CV Instruments, Bowers Group, Camberley, UK) (430 AAT) with a load of 10 kg and holding time of 15 s. For each measurement we ensured that the typical alloy microstructure was sampled by the indenter. The microhardness of Nb<sub>5</sub>Si<sub>3</sub> was measured using a Mitutoyo hardness machine with a load of 0.05 kg and holding time of 15 s. The microhardness of the Nb<sub>ss</sub> was measured with a load of 0.025 kg and holding time of 15 s. At least 10 measurements were taken for the macrohardness of the alloys and the microhardness of the aforementioned phases. Each indent was measured twice. A gas pycnometer was used to measure the densities of the as cast alloys. We used the Zeiss KSRun version 3 high resolution imaging analysis software and the large area backscatter electron images that were used for large area analyses to measure the area fractions of phases.

#### 3. Results

#### 3.1. Nb-24Ti-18Si-5Al-5Cr-5Ge (Alloy ZF6)

The actual composition of the as cast alloy (ZF6-AC) was Nb-25.7Ti-16.8Si-5.1Ge-5.0Cr-5.8Al and its density was  $6.54 \pm 0.01$  g/cm<sup>3</sup> (Table 1). There was macrosegregation of Si, Cr, Ti and Al the concentrations of which were in the range 15.8 to 20.1 at.%, 3.5 to 5.6 at.%, 23.9 to 26.3 at.% and 4.1 to 6.7 at.%, respectively (Table 2). The microstructure of ZF6-AC consisted of the Nb<sub>ss</sub>, βNb<sub>5</sub>Si<sub>3</sub> and C14-Cr<sub>2</sub>Nb phases (Figure 1 and Table 2). Typical images of the microstructure of different parts of ZF6-AC are shown in Figure 2a–c. The  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> grains in the bulk of ZF6-AC were large and faceted (Figure 2b). The Laves phase was formed between Ti-rich  $Nb_{ss}$  dendrites only in the bulk of the button (Figure 2b) where it was present at a small volume fraction and its average composition was 18.4Nb-24.8Ti-6.2Si-1.2Ge-46.3Cr-3.1Al (Table 2). The Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectic was also observed in the top and bulk of ZF6-AC with average composition 45.5Nb-28.4Ti-8.1Si-3.7Ge-8.0Cr-6.3Al. Ti-rich areas were formed at the edges of  $Nb_{ss}$  and  $\beta Nb_5Si_3$ . In the Ti-rich  $Nb_{ss}$  the concentrations of Si, Ge and Al were very close to those in the Nb<sub>ss</sub> but the Cr concentration was higher (Table 2). The  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> was leaner than the Si concentration in unalloyed Nb<sub>5</sub>Si<sub>3</sub>. The microstructure in the bottom of the button was different compared with the top and bulk and consisted only of the  $\beta Nb_5Si_3$  and  $Nb_{ss}$  phases (Figure 2c) with a higher volume fraction of the Nb<sub>ss</sub> (Table 1). The microstructure shown in Figure 2c had evolved from a very fine eutectic that had formed at the very bottom of ZF6-AC where the melt had been in direct contact with the water-cooled copper crucible.

Table 1.

Density % Area Alloy (g/cm<sup>3</sup>) Тор Bulk Bottom  $22.4 \pm 1.8$  $17.1 \pm 3.7$  $29.2 \pm 2.8$ ZF6-AC  $6.54 \pm 0.01$ ZF6-HT  $17.3 \pm 3.5$ --ZF9-AC  $6.96 \pm 0.02$ \_ \_ ZF9-HT \_

Nb-24Ti-18Si-5Ge-5Cr-5Al (ZF6) and Nb-24Ti-18Si-5Ge-5Cr-5Al-5Hf (ZF9).

Density and % areas of  $Nb_{ss}$  in as cast (AC) and heat treated (HT) alloys

Area/Phase	Nb	Ti	Si	Ge	Cr	Al
ZF6-AC	-	-	-	-	-	-
Ton *	$41.6\pm0.4$	$26.2 \pm 0.1$	$16.1 \pm 0.1$	$4.9 \pm 0.1$	$5.4 \pm 0.2$	$5.8 \pm 0.6$
Top	41.4-42.0	26.1-26.2	16.0-16.1	4.8-5.0	5.2-5.5	5.3-6.2
D 11 %	$41.9 \pm 1.0$	$24.9\pm0.8$	$18.0 \pm 1.9$	$5.1 \pm 0.3$	$4.5 \pm 1.0$	$5.6 \pm 1.1$
Bulk *	40.9-43.3	23.9-26.0	16.1-20.1	4.9-5.6	3.5-5.6	4.1-6.6
D - 11 *	$41.1 \pm 0.4$	$26.1 \pm 0.4$	$16.3 \pm 0.4$	$5.4 \pm 0.1$	$5.2 \pm 0.2$	$6.1 \pm 0.5$
Bottom "	40.7-41.6	25.5-26.3	15.8-16.6	5.3-5.5	5.0 - 5.4	5.9-6.7
Nīb	$52.3 \pm 1.6$	$27.9 \pm 0.7$	$2.1 \pm 0.2$	$1.7 \pm 0.3$	$9.2 \pm 1.1$	$6.8 \pm 0.4$
IND <sub>SS</sub>	50.2-53.6	27.0-28.8	2.0 - 2.4	1.4-2.2	8.0-10.6	6.6-7.3
Ti rich Mb	$45.4 \pm 3.1$	$31.3 \pm 1.4$	$1.8 \pm 0.4$	$1.6 \pm 0.3$	$12.9 \pm 1.2$	$7.0 \pm 0.3$
II-IICII IND <sub>SS</sub>	42.0-48.7	29.9-32.7	1.3-2.1	1.3-1.9	11.5 - 14.4	6.7-7.7
BND-Sia	$42.4 \pm 1.1$	$22.1 \pm 0.9$	$23.2 \pm 1.3$	$7.3 \pm 0.4$	$1.4 \pm 0.2$	$3.6 \pm 1.0$
p1ND5513	41.3-43.9	20.9-23.0	22.1-24.8	6.9-7.9	1.1-1.5	2.3-4.8
Ti rich (Nh Ci	$34.1 \pm 0.4$	$29.5 \pm 0.3$	$20.9 \pm 1.4$	$8.4 \pm 0.4$	$2.4 \pm 0.4$	$4.7 \pm 0.7$
11-ficit p105513	33.8-34.5	29.1-29.7	19.3-27.1	8.1-8.9	2.1-2.8	4.1-5.4
C14-Cr <sub>2</sub> Nb +	18.4	24.8	6.2	1.2	46.3	3.1
Enterthe	$45.5 \pm 1.2$	$28.4\pm0.6$	$8.1 \pm 1.5$	$3.7 \pm 0.5$	$8.0 \pm 1.1$	$6.3 \pm 0.4$
Eutectic	44.2-46.9	27.7-29.0	5.8-9.0	3.2-4.3	6.6-9.3	6.0-6.8
ZF6-HT *	$43.1 \pm 1.0$	$24.3 \pm 1.2$	$17.4 \pm 0.8$	$5.8 \pm 0.2$	$4.2 \pm 0.3$	$5.2 \pm 0.8$
(1400 °C/100 h)	41.9-44.6	23.0-25.9	16.4 - 18.7	5.5-6.2	4.0 - 4.7	4.1-6.1
NTL	$52.6 \pm 0.5$	$26.9 \pm 0.3$	$0.9 \pm 0.5$	$1.0 \pm 0.2$	$12.3 \pm 0.2$	$6.3 \pm 0.4$
IND <sub>SS</sub>	52.1-53.0	26.4-27.2	0-1.2	0.8 - 1.2	11.9-12.4	5.7-6.7
NTL C:	$42.2 \pm 0.9$	$21.7 \pm 0.9$	$23.1 \pm 1.2$	$7.4 \pm 0.5$	$1.8 \pm 0.4$	$3.8 \pm 0.6$
105513	41.4-43.4	20.5-22.6	21.7-24.4	7.0-8.1	1.5-2.3	3.1-4.3
Ti wish Mile Ci	$37.6 \pm 1.1$	$26.1 \pm 1.1$	$21.0 \pm 0.3$	$7.8 \pm 0.2$	$2.7 \pm 0.1$	$4.8 \pm 0.3$
11-rich ND5513	35.9_39.1	25.0_27.9	20.8-21.6	76-80	25_28	46-52

Table 2. Chemical analysis data (at.%) of as cast (AC) and heat treated (HT) alloy ZF6.

<sup>+</sup> This was the average of three analyses; \* large area analysis.



Figure 1. X-ray diffractograms of the as cast and heat treated (1400 °C/100 h) alloy ZF6.



**Figure 2.** Typical backscatter electron images of the cross section of alloy ZF6: (**a**–**c**) top, bulk and bottom areas of the as cast alloy ZF6-AC; (**d**) the bulk of the heat-treated alloy ZF6-HT (1400  $^{\circ}$ C/100 h).

After the heat treatment the actual composition of the heat-treated specimen (ZF6-HT) was 43.1Nb-24.3Ti-17.4Si-5.8Ge-4.2Cr-5.2Al (Table 2). The XRD data (Figure 1) would suggest that the Nb<sub>ss</sub>,  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> and  $\alpha$ Nb<sub>5</sub>Si<sub>3</sub> phases were present. The microstructure is shown in Figure 2d. The volume fraction of the Nb<sub>ss</sub> did not change significantly (Table 1). Ti-rich areas were still observed in the Nb<sub>5</sub>Si<sub>3</sub>. The C14-Cr<sub>2</sub>Nb Laves phase and the eutectic were absent after the heat treatment. A phase with black contrast was formed at a small volume fraction at the edges of some Nb<sub>ss</sub> and Nb<sub>5</sub>Si<sub>3</sub> grains. X-ray elemental maps taken in EPMA confirmed that this phase was rich in oxygen and Ti (Figure 3).



**Figure 3.** EPMA/Wavelength Dispersive spectroscopy (WDS) X-ray maps from the centre of the cross section of ZF6-HT (1400 °C/100 h) showing oxygen contamination at the edges of some of Nbss grains.

#### 3.2. Nb-24Ti-18Si-5Ge -5Cr-5Al-5Hf (Alloy ZF9)

The actual composition of the as cast alloy (ZF9-AC) was 35.8Nb-26.4Ti-16.4Si-5.3Ge-5.6Cr-5.1Al-5.4Hf and its density was  $6.96 \pm 0.02$  g/cm<sup>3</sup> (Table 1). There was macrosegregation of Ti, Si, Al and Cr the concentrations of which were in the ranges 24.1 to 27.5 at.%, 15.3 to 18.4 at.%, 4.1 to 6.1 at.% and 4.3 to 6.4 at.%, respectively (Table 3). The phases present in the microstructure of ZF9-AC were the Nb<sub>ss</sub>,  $\beta$ Nb<sub>5</sub>Si<sub>3</sub>,  $\gamma$ Nb<sub>5</sub>Si<sub>3</sub> and C14-Cr<sub>2</sub>Nb (Figure 4 and Table 3). Typical images of the microstructure are shown in Figure 5a–c. The partitioning of Hf between the aforementioned phases made the characterisation of the microstructure extremely difficult owing to only slight variations in contrast between different phases under back scatter electron imaging conditions. The phase with grey contrast in Figure 5 was Nb<sub>5</sub>Si<sub>3</sub> with average composition 37.2Nb-21.0Ti-24.2Si-6.6Ge-1.2Cr-5.0Al-4.8Hf and Si + Ge + Al concentration about 35.8 at.%, close to the Si concentration in unalloyed Nb<sub>5</sub>Si<sub>3</sub>. The phase with light grey contrast in Figure 5 was Hf-rich Nb<sub>5</sub>Si<sub>3</sub> in which the Hf concentration was almost double that in the Nb<sub>5</sub>Si<sub>3</sub>. The Nb<sub>ss</sub> was formed between the former two phases and exhibited dark gray contrast. Reliable results of the volume fraction of the  $Nb_{ss}$  in this alloy could not be obtained as the contrast of the solid solution was close to that of the Nb<sub>5</sub>Si<sub>3</sub> and Hf-rich Nb<sub>5</sub>Si<sub>3</sub>. The C14-NbCr<sub>2</sub> Laves phase was formed between the Nb<sub>ss</sub> and Hf-rich Nb<sub>5</sub>Si<sub>3</sub> in the top and bulk of ZF9-AC and its average composition was 20.5Nb-19.7Ti-5.6Si-1.3Ge-43.7Cr-4.0Al-5.2Hf. As was the case in the alloys Nb-24Ti-18Si-5Cr-5Ge [15] and ZF6, the Laves phase was absent in the bottom of the button of ZF9-AC. The  $Nb_{ss} + Nb_5Si_3$  eutectic was not observed in any part of the button of this alloy. A microstructure exhibiting fine features that could be representative of eutectic was observed only once in one area in the top of ZF9-AC. Detailed study of the button did not find any other areas with similar features.

Area/Phase	Nb	Ti	Si	Ge	Cr	Al	Hf
ZF9-AC	-	-	-	-	-	-	-
T*	$35.6 \pm 0.1$	$27.3 \pm 0.2$	$16.0 \pm 0.6$	$5.3 \pm 0.2$	$5.9 \pm 0.3$	$4.4 \pm 0.3$	$5.5 \pm 0.1$
10p <sup>1</sup>	35.7-35.9	27.0-27.5	15.3-16.5	5.0 - 5.4	5.5-6.1	4.1-4.7	5.5-5.6
D 11 *	$36.7 \pm 0.4$	$24.8\pm0.6$	$17.6\pm0.8$	$5.1 \pm 0.2$	$4.6 \pm 0.2$	$5.8 \pm 0.4$	$5.4 \pm 0.2$
Bulk*	36.3-37.3	24.1-25.5	16.4 - 18.4	4.9-5.4	4.3-4.9	5.2-6.1	5.2-5.6
D (1 *	$35.3 \pm 0.1$	$27.1 \pm 0.2$	$15.5 \pm 0.1$	$5.5 \pm 0.5$	$6.2 \pm 0.3$	$5.0 \pm 0.2$	$5.4 \pm 0.2$
Bottom*	35.2-35.4	27.0-27.3	15.4-15.6	5.1-6.0	5.9-6.4	4.9-5.2	5.2-5.5
NI	$40.4\pm3.4$	$31.5 \pm 2.3$	$1.6 \pm 0.2$	$1.0 \pm 0.2$	$15.0 \pm 1.2$	$7.8 \pm 0.3$	$2.7 \pm 0.2$
IND <sub>SS</sub>	35.5-44.4	29.1-35.1	1.5 - 1.8	0.7 - 1.1	13.4-16.1	7.5-8.4	2.5-3.0
ONT C:	$37.2 \pm 0.2$	$21.0 \pm 0.3$	$24.2\pm0.5$	$6.6 \pm 0.1$	$1.2 \pm 0.1$	$5.0 \pm 0.4$	$4.8 \pm 0.1$
BIND5513	36.8-37.3	20.6-21.2	23.8-25.1	6.5-6.8	1.2-1.3	4.4-5.5	4.7-4.9
Hf-rich	$30.9\pm0.7$	$24.0\pm0.3$	$24.5\pm0.6$	$7.0 \pm 0.3$	$0.8 \pm 0.2$	$3.8 \pm 0.6$	$9.0 \pm 0.2$
Nb <sub>5</sub> Si <sub>3</sub>	29.9-31.8	23.6-24.2	23.7-25.1	6.6-7.4	0.5-0.9	3.3-4.8	8.8-9.2
$Cr_2Nb^+$	20.5	19.7	5.6	1.3	43.7	4	5.2
ZEO LIT*	$38.0\pm2.0$	$24.2 \pm 1.5$	$17.1 \pm 1.3$	$5.2 \pm 0.2$	$4.8 \pm 0.6$	$5.5 \pm 0.8$	$5.2 \pm 0.2$
ZF9-H1*	36.4-40.2	22.8-25.7	15.8 - 18.4	5.0 - 5.4	4.1-5.3	4.9-6.4	5.0 - 5.4
NIL	$56.6 \pm 3.3$	$25.3 \pm 1.2$	$0.7 \pm 0.5$	$0.6 \pm 0.6$	$11.8\pm0.8$	$3.8 \pm 1.2$	$1.2 \pm 0.1$
IND <sub>SS</sub>	56.8-59.8	24.3-26.7	0.0 - 1.4	0.0-1.2	10.6-12.6	2.4 - 5.4	1.1–1.3
NIL CI	$37.0 \pm 0.6$	$21.2\pm0.4$	$24.8\pm0.5$	$6.7 \pm 0.3$	$2.3 \pm 0.3$	$3.2 \pm 0.4$	$4.8 \pm 0.2$
105513	36.4-37.9	20.6-21.7	24.2-25.2	6.3-7.2	2.0-2.7	2.7-3.4	4.5-4.9
Hf-rich	$30.2 \pm 1.2$	$24.3\pm0.9$	$25.4\pm0.6$	$7.0 \pm 0.2$	$1.0 \pm 0.3$	$3.3 \pm 0.5$	$8.8\pm0.1$
Nb <sub>5</sub> Si <sub>3</sub>	29.0-31.3	23.6-25.3	25.0-26.1	6.8–7.2	0.7–1.3	2.7-3.6	8.8-8.9

+ This was the average data of three measurements, \* large area analysis.



Figure 4. X-ray diffractograms of the as cast (AC) and heat treated (HT) (1400 °C/100 h) alloy ZF9.





**Figure 5.** Typical backscatter electron images of the alloy ZF9: (**a**–**c**) top, bulk and bottom areas of the as cast alloy ZF9-AC; (**d**) the bulk area and (**e**) the top area of the heat-treated alloy ZF9-HT (1400 °C/100 h). In (**e**) the phase exhibiting white contrast is HfO<sub>2</sub>.

After the heat treatment the actual composition of the heat-treated specimen (ZF9-HT) was 38.0Nb-24.2Ti-17.1Si-5.2Ge-4.8Cr-5.5Al-5.2Hf (Table 3). The XRD data (Figure 4) would suggest that the phases Nb<sub>ss</sub>,  $\beta$ Nb<sub>5</sub>Si<sub>3</sub>,  $\gamma$ Nb<sub>5</sub>Si<sub>3</sub>,  $\alpha$ Nb<sub>5</sub>Si<sub>3</sub> and HfO<sub>2</sub> were present. The C14-NbCr<sub>2</sub> Laves phase was absent. The microstructure is shown in Figure 5d. The Hf-rich Nb<sub>5</sub>Si<sub>3</sub> was still observed. Hafnia particles exhibiting white contrast were found only near the surface of ZF9-HT (Figure 5e). Their average composition (EPMA/WDS) was 34.7Hf-65.3O.

The lattice parameter of the Nb<sub>ss</sub> in the alloys ZF6 and ZF9, and hardness data are given in the Tables 4 and 5, respectively. The lattice parameter of Nb<sub>ss</sub> increased after the heat treatment. The addition of Hf in the alloy ZF9 caused a decrease in hardness compared with the alloy ZF6.

Alloy	ZF6-AC	ZF6-HT	ZF9-AC	ZF9-HT	•
Parameter	3.252	3.260	3.256	3.264	

Table 5. Vickers hardness (HV) of the as cast and heat-treated alloys ZF6 and ZF9.

Table 4. Lattice parameter (Å) of the Nb<sub>ss</sub> in the alloys ZF6 and ZF9.

Allov	Hardness	Microhardness		
	interiess	Nb <sub>ss</sub>	Nb <sub>5</sub> Si <sub>3</sub>	
ZF6-AC	$840 \pm 34$	+	$1644 \pm 89$	
ZF6-HT	$854 \pm 31$	$691 \pm 37$	$1576 \pm 60$	
ZF9-AC	$801 \pm 37$	+	$1495 \pm 88$	
ZF9-HT	$779 \pm 11$	+	$1391 \pm 21$	

<sup>+</sup> Not measured owing to the small size of the Nb<sub>ss</sub> in the alloy.

## 4. Discussion

#### 4.1. Macrosegregation

Macrosegregation is a common phenomenon in as cast Nb-silicide based alloys that are produced in water cooled copper crucibles using non-consumable W electrodes for arc melting [16]. It is also observed in Nb-silicide based alloys that are prepared using vacuum induction melting and induction skull melting [17].

In the alloy ZF6 there was macrosegregation of Si. The latter was low when Ge was in synergy only with Si and increased after alloying with Ti (see Table 6 and compare alloys ZF1 [9] and ZF3 [18]) and with Ti, Al and Cr (Table 6, alloy ZF6). In the alloy ZF9, the addition of Hf reduced the macrosegregation

of Si and Al (Tables 6 and 7), increased that of Ti (Table 7) and did not change the macrosegregation of Cr (Table 7).

Elements in Synergy	Alloy Code* & Reference	$C_{max}^{Si} - C_{min}^{Si}$	Macrosegregation
Ti + Cr + Ge	ZF4 [15]	5.3	<b>A</b>
Ti + Cr + Al + Ge	ZF6	4.3	
Ti + Ge	ZF3 [18]	3.2	
Ti + Cr + Al + Ge + Hf	ZF9	3.1	
Ti + Al + Ge	ZF5 [13]	2.9	
Al + Ge	ZF8 [13]	2.5	
Ti + Al	KZ7 [12]	2.3	
Ge	ZF1 [9]	2.0	
Ti + Cr	KZ4 [12]	1.9	
Ti	KZ3 [12]	1.6	
Cr + Ge	ZF7 [15]	1.5	
Ti + Cr + Al	KZ5 [12]	1.3	

 Table 6.
 Synergistic effect of alloying elements on macrosegregation of Si in as cast Nb-18Si silicide-based alloys.

\* see Appendix A.

Table 7. Effect of alloying on macrosegregation of Al, Cr and Ti in as cast Nb-18Si silicide-based alloys.

Alloving Additions	Alloy Code* &	$C_{max}^{i} - C_{min}^{i}$ (i = Ti, Cr, Al)			
	Reference	Ti	Cr	Al	
Ti + Ge + Cr + Al + Hf	ZF9	3.4	2.1	2	
Ti + Ge + Cr	ZF4 [15]	3.3	2.2	-	
Ti + Ge + Cr + Al	ZF6	2.4	2.1	2.6	
Ti + Al	KZ7 [12]	2.3	-	-	
Ti + Cr + Al	KZ5 [12]	1.4	-	-	
Ti + Ge	ZF3 [18]	1.4	-	-	
Ti + Cr	KZ4 [12]	1.4	1.8	-	
Ti	KZ3 [12]	3.1	-	-	

\* see Appendix A.

Long range transport due to the flow of melt and movement of solid during solidification play a key role in the formation of macrosegregation [19]. The latter also is affected by the composition of the alloy and the viscosity of its melt [19]. Complex interrelationships of heat transfer, solute transport, solid movement and fluid flow control macrosegregation [16]. The macrosegregation of Si in Nb-silicide based alloys has been attributed to such interrelationships that affect the buoyancy and thermo-capillary forces that arise by solute partitioning in these alloys and their effects on the densities of phases and the sign and value of  $d\gamma_L/dT$  ( $\gamma_L$  is the liquid surface tension) [16].

The effect of the synergy of Ge with Hf and other alloying elements on the macrosegregation of Si, Ti and Cr can be understood by studying the role played by solute partitioning in the solidification of Nb-silicide based alloys. The chemical analysis data for the phases present in the alloys for which macrosegregation data is summarised in the Tables 6 and 7 can help one to find out whether there is dependence of macrosegregation on elemental ratios or sums in the Nb<sub>ss</sub>, Nb<sub>5</sub>Si<sub>3</sub>, NbCr<sub>2</sub> phases and the Nb<sub>ss</sub> + Nb<sub>5</sub>Si<sub>3</sub> eutectic. Trends for the Si + Ge + Al and Si + Ge concentrations in Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectics are shown in the Tables 8 and 9, respectively. The Figures 6 and 7 show the trends in the macro-segregation of Si and Ti for different elemental ratios and sums.

**Table 8.** Concentration of Si + Ge + Al in the Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectic in the as cast alloys ZF5, ZF6 and ZF8.

Alloy Code	Nominal Composition (at.%)	Si + Ge +Al in The Eutectic (at.%)
ZF8-AC [13]	Nb-18Si-5Ge-5Al	22.6
ZF5-AC [13]	Nb-24Ti-18Si-5Ge-5Al	18.8
ZF6-AC	Nb-24Ti-18Si-5Ge-5Cr-5Al	18.1

**Table 9.** Concentration of Si + Ge in the Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectic in as cast ZF series alloys.

Alloy Code & Reference	Nominal Composition (at.%)	Si + Ge in The Eutectic (at.%)	♠
ZF1-AC [9]	Nb-18Si-5Ge	17.7	
ZF2-AC [9]	Nb-18Si-10Ge	17.6	
ZF8-AC [13]	Nb-18Si-5Ge-5Al	16.3	
ZF7-AC [15]	Nb-18Si-5Ge-5Cr	15.2	
ZF4-AC [15]	Nb-24Ti-18Si-5Ge-5Cr	14.8	
ZF5-AC [13]	Nb-24Ti-18Si-5Ge-5Al	12.6	
ZF6-AC	Nb-24Ti-18Si-5Ge-5Cr-5Al	11.8	



**Figure 6.** Schematic diagram showing the dependence of the macrosegregation of element X (vertical axis) on the alloying elements in specific phases in Nb silicide-based alloys. The ranges of X values are given in Tables 6 and 7. Chemical analysis data for the Nb<sub>ss</sub> + Nb<sub>5</sub>Si<sub>3</sub> eutectic in Ge containing Nb silicide based alloys is given in the Tables 8 and 9. The chemical analysis data for specific phases in Nb silicide based alloys can be found in the references given in Tables 6–9. The <u>thick line</u> shows (a) how the Si macrosegregation varies with (Si + Ge)<sub>eutectic</sub> (range 11.8 to 17.7, nose ZF6), (Si/Al)<sub>Nb5Si3</sub> (range 4.8 to 9.5, nose ZF9), (Si/Cr)<sub>Nb5Si3</sub> (range 16.6 to 24, nose ZF6), (Si/Al)<sub>Nbss</sub> (range 0.21 to 0.49, nose ZF6), (TM/SM)<sub>Nbss</sub> (range 2.7 to 3.8, nose ZF5), (TM/SM)<sub>Ti rich Nb5si</sub> (range 3.4 to 5.6, nose ZF5), (Si/Cr)<sub>Nb5Si3</sub> (range 16.6 to 24, nose ZF6) and (Si/Al)<sub>Ti rich Nb5Si3</sub> (range 4.4 to 8.7, nose ZF6), (Ti/Cr)<sub>Nb5Si3</sub> (range 16.6 to 24, nose ZF6) and (Si/Al)<sub>Ti rich Nb5Si3</sub> (range 4.4 to 8.7, nose ZF6), (Ti/Cr)<sub>Nb5Si3</sub> (range 14.9 to 17.5, nose ZF4) and (Ti/Cr)<sub>Ti rich Nb5Si3</sub> (range 9.8 to 30, nose ZF4) and (c) how the Cr macrosegregation varies with (Si/Cr)<sub>Nb5si3</sub> (range 0.1 to 0.28, nose ZF4). The <u>dashed line</u> shows (d) how the macrosegregation of Si varies with (Si/Al)<sub>Ti rich Nb5Si3</sub> (range 4.3 to 8.7) and (Si + Al + Ge)<sub>eutectic</sub> (range 18.1 to 22.6) and (e) how the Ti macrosegregation varies with (Si/Al)<sub>Nb5Si3</sub> (range 4.8 to 9.5) and (Si/Al)<sub>Nbss</sub> (range 0.21 to 0.33), X = Si, Ti.



**Figure 7.** Schematic diagram showing the dependence of the macrosegregation of element X (vertical axis) on alloying elements in specific phases in Nb silicide-based alloys. The ranges of X values are given in Tables 6 and 7. The chemical analysis data for specific phases in Nb silicide-based alloys can be found in the references given in the Tables 6–9. The <u>thick line</u> shows (a) how the Si macrosegregation varies with ([Si + Ge]/Cr)<sub>Laves</sub> (range 2.3 to 4.4, nose ZF6), (TM/[Si + Ge])<sub>Nbss</sub> (range 6.8 to 18.9, nose ZF9), (Ti/Cr)<sub>Nb5Si3</sub> (range 12.7 to 17.5, nose ZF9), (Ti/Cr)<sub>Ti rich Nb5Si3</sub> (range 9.8 to 30, nose ZF9), (TM/SM)<sub>Nb5Si3</sub> (range 0.5 to 0.8, nose ZF9), (TM/SM)<sub>Ti rich Nb5Si3</sub> (range 0.6 to 1, nose ZF9) and (TM/[Si + Ge])<sub>Nb5Si3</sub> (range 0.6 to 0.9, nose ZF9) (b) how the Ti macrosegregation varies with (Si/Cr)<sub>Nb5Si3</sub> (range 16.6 to 20.2, nose ZF9). X = Si, Ti, Cr, TM = Ti + Cr + Hf and SM = Si + Ge + Al. The <u>dashed line</u> shows (d) how the macrosegregation of Si varies with ([Si + Ge + Al]/Cr)<sub>Laves</sub> (range 1.4 to 5.5) and (TM/[Si + Ge])<sub>Ti rich Nbss</sub> (range 9 to 13.2) and (e) how the Ti macrosegregation varies with (Ti/Cr)<sub>Ti rich Nbss</sub> (range 2.1 to 2.6). X = Si, Ti and TM = Ti + Cr + Hf.

The Figures 6 and 7 show that Si macrosegregation in Nb-silicide based alloys is linked with solute partitioning during solidification, as evidenced by the variation of Si macrosegregation with the ratios and sums of different elements. Figure 6 shows that the ratios  $(Si/Al)_{Nbss}$ ,  $(Si/Cr)_{Nbss}$ ,  $(Ti/Cr)_{Nbss}$ ,  $(TM/SM)_{Ti rich Nbss}$  (where TM = Ti + Cr + Hf and SM = Si + Ge + Al),  $(Si/Al)_{Nb5Si3}$ ,  $(Si/Cr)_{Nb5Si3}$ , and the  $(Si+Ge)_{eutrectic}$  all exhibit the same trend with increasing Si macrosegregation. The "nose" of the curves corresponds to the alloys ZF6 or ZF9. In Figure 6 the trend of the  $(Si/Al)_{Ti rich Nb5Si3}$  and  $(Si + Al + Ge)_{eutectic}$  is shown by the lower part of the curves. The trend in Figure 6 is opposite to the trend shown in Figure 7 for the ratios  $(Ti/Cr)_{Nb5Si3}$ ,  $(Ti/Cr)_{Ti rich Nb5Si3}$ ,  $(TM/SM)_{Ti rich Nb5Si3}$ ,  $(TM/SM)_{Ti rich Nb5Si3}$ ,  $(TM/SM)_{Ti rich Nb5Si3}$ ,  $(TM/SM)_{Ti rich Nb5Si3}$ ,  $(TM/SM)_{Nb5Si3}$ ,  $(TM/SM)_{Ti rich Nb5Si3}$ ,  $(TM/SI)_{Nb5Si3}$ ,  $(TM/SI)_{Ti rich Nb5Si3}$ , (

that are present in these alloys. It should be noted that all the aforementioned elements are required for achieving a balance between environmental and mechanical properties in Nb-silicide based alloys [3].

The Figures 6 and 7 also show that Ti macrosegregation in Nb-silicide based alloys is linked with solute partitioning during solidification as evidenced by the variation of Ti macrosegregation with different element ratios and sums. The Figure 6 shows that the ratios  $(Si/Cr)_{Nb5Si3}$ ,  $(Si/Al)_{Ti rich Nb5Si3}$ ,  $(Ti/Cr)_{Nb5Si3}$  and  $(Ti/Cr)_{Ti rich Nb5Si3}$  exhibit the same trend with increasing Ti macrosegregation. The "nose" of the curves corresponds to the alloys ZF6 or ZF4. In Figure 6 the trend of  $(Si/Al)_{Nb5Si3}$  and  $(Si/Al)_{Nb5Si3}$  is shown by the lower part of the curves. The trend in Figure 6 is opposite to the trend shown in Figure 7 for the ratios  $(Si/Cr)_{Nb5S}$  and  $(Ti/Cr)_{Nb5S}$  where the "nose" corresponds to the alloys ZF4. The trend of  $(Ti/Cr)_{Ti rich Nb5S}$  in Figure 7 is shown by the lower part of the figure and is opposite to that shown in Figure 6. The data would thus suggest that in Ge containing Nb-silicide based alloys the Cr plays the important role in the macrosegregation of Ti owing to partitioning between the Nb<sub>5</sub>si and the Nb<sub>5</sub>Si<sub>3</sub> silicide.

Figures 6 and 7 also show that Cr macrosegregation in Nb-silicide based alloys is linked with solute partitioning during solidification as evidenced by the variation of Cr macrosegregation with different elemental ratios and sums. These two figures show that the ratios (Si/Cr)<sub>Nb5Si3</sub> and (Si/Cr)<sub>Nb5s</sub> exhibit opposite trends with the "nose" of the curves corresponding to the alloy ZF9. This would suggest that Hf plays a role in the macro-segregation of Cr. The latter is not strong in Nb-silicide based alloys with Cr  $\leq$  8 at.% compared with Si or Ti (see Table 7). The Hf is also known not to affect significantly the partitioning of Cr between the Nb<sub>ss</sub> and the Nb<sub>5</sub>Si<sub>3</sub> silicide. Thus, the effect of the synergy of Ge and Hf on the macrosegregation of Cr is expected to be weak.

## 4.2. Microstructures

#### 4.2.1. Cast Alloys

In both alloys the Nb<sub>5</sub>Si<sub>3</sub> was the primary phase. The solidification paths in the bulk of the ingots were  $L \rightarrow L + \beta Nb_5Si_3 \rightarrow L + \beta Nb_5Si_3 + (Nb_{ss} + \beta Nb_5Si_3)_{eutectic} \rightarrow L + \beta Nb_5Si_3 + (Nb_{ss} + \beta Nb_5Si_3)_{eutectic} + C14-NbCr_2 and L \rightarrow L + Nb_5Si_3 \rightarrow L + Nb_5Si_3 + Nb_{ss} \rightarrow L + Nb_5Si_3 + Nb_{ss} + C14-NbCr_2$ , respectively for the alloys ZF6 and ZF9 with the C14-NbCr<sub>2</sub> forming from the last to solidify liquid. The addition of Hf suppressed the Nb<sub>ss</sub> +  $\beta Nb_5Si_3$  eutectic but not the Laves phase.

The formation of the Laves phase is linked with Cr redistribution during solidification [12]. This would explain its absence in the bottom of the buttons owing to the effect of cooling rate on Cr partitioning [14,15]. The addition of Hf made possible the formation of the C14-NbCr<sub>2</sub> Laves phase at higher cooling rates in ZF9-AC, compared with ZF6-AC.

Compared with the alloy KZ5-AC (Nb-24Ti-18Si-5Al-5Cr [12]) the addition of Ge in ZF6-AC decreased the solid solubility of Cr in the Nb<sub>ss</sub>. Compared with the alloy ZF6-AC the addition of Hf in ZF9-AC (i) increased the solid solubilities of Cr and Al and decreased the solid solubility of Ge in the Nb<sub>ss</sub>.

There are three types of 5-3 silicides in the binary Nb-Si system [20], namely the  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> (t/32, I4/mcm, prototype W<sub>5</sub>Si<sub>3</sub>),  $\alpha$ Nb<sub>5</sub>Si<sub>3</sub> (t/32, I4/mcm, prototype Cr<sub>5</sub>B<sub>3</sub>) and  $\gamma$ Nb<sub>5</sub>Si<sub>3</sub> (hP16, P6<sub>3</sub>/mcm, prototype Mn<sub>5</sub>Si<sub>3</sub>), the former two are the equilibrium phases and the third is metastable and is stabilised by interstitial impurities [20]. The latter silicide has the same structure with the Ti<sub>5</sub>Si<sub>3</sub> and Hf<sub>5</sub>Si<sub>3</sub> [21] silicides. Thus, it is expected that alloying the Nb<sub>5</sub>Si<sub>3</sub> with Ti or/and Hf would tend to increase the stability of the  $\gamma$ Nb<sub>5</sub>Si<sub>3</sub>. The XRD data of ZF9-AC (Figure 4) contained characteristic peaks corresponding to the  $\gamma$ Nb<sub>5</sub>Si<sub>3</sub>. Bewlay et al [22] suggested that the Nb<sub>5</sub>Si<sub>3</sub> had hexagonal rather than tetragonal structure when the Nb/(Ti+Hf) ratio was less than one in the alloy and silicide. For the Nb<sub>5</sub>Si<sub>3</sub> and Hf rich Nb<sub>5</sub>Si<sub>3</sub> silicides in the present work the above ratio was in the range 1.4 to 1.5 (Table 3).

The Al content of the Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectic in the alloy ZF6 (Table 8) was the same with the eutectic in the cast alloy ZF5 (Nb-24Ti-18Si-5Al-5Ge [13]) but not the Si + Ge content (Table 9). This

was attributed to the synergy of Al and Ti, which is stronger than that of Cr and Ti (Table 9, compare the alloys ZF8-AC and ZF5-AC with the alloys ZF7-AC and ZF4-AC).

The C14-NbCr<sub>2</sub> Laves phase was observed in the bulk of the buttons of the alloys ZF6-AC, ZF9-AC and in ZF4 (Nb-24Ti-18Si-5Cr-5Ge [15]), and was absent in the bottom of the buttons of all three alloys but present in the top only of the latter two alloys. If we were to use the ratio  $\Lambda = [Si + Ge (+ Al)]/Cr$  as a criterion for the sensitivity of C14-NbCr<sub>2</sub> Laves phase formation on alloy composition and consider the large area analysis data for the top, bulk and bottom of the above alloys given in Tables 2 and 3 and [15], it is concluded that the lowest value of  $\Lambda$  for Laves phase formation in Ge containing alloys is about 3.7 at.% when Al is not present in the alloy and about 4.3 at.% when Al is in synergy with Cr.

In the alloys ZF6-AC and ZF9-AC the Laves phase was the hexagonal C14-Cr<sub>2</sub>Nb (hP12) (Figures 1 and 4). In the Laves phase the Al and Si atoms substitute for Cr atoms [23]. Furthermore, alloying with Si [24] or Al [25] stabilises the C14-NbCr<sub>2</sub> to low temperatures. The Si + Ge + Al + Cr concentration in the Laves phase (or B element content in  $AB_2 = NbCr_2$ ) in the alloys ZF6 and ZF9 was respectively 56.8 at.% and 54.6 at.% and the Ge concentrations were 1.2 at.% and 1.3 at.%, respectively. Thus, the B element content in the C14-NbCr2 was similar to that reported in the alloys KZ4 (Nb-24Ti-18Si-5Cr) and KZ7 (Nb-24Ti-18Si-5Al) [12] and JG3 (Nb-24Ti-18Si-5Al-5Cr-5Hf-2Mo [26]) where the (Cr + Si + X)/(Nb + Y) ratio (X = Al, Y = Hf, Ti) was in the range 1.1 to 1.5 compared with (Cr + Si + X)/(Nb + Y) ratios of 1.3 and 1.2, respectively for the alloys ZF6 and ZF9 (X = Al, Ge, Y = Hf, Ti). The off-stoichiometric and lean in B element content NbCr<sub>2</sub> has also been reported in alloys of the Nb-Cr-Ti [27] and Nb-Cr-Hf [28] ternary systems and was attributed to the anti-site defect substitution with Ti or Hf partitioning to A site and thus displacing some Nb atoms to the B site in  $AB_2 = NbCr_2$ . The Ge concentration in the Laves phase was in agreement with [10]. Considering the data for the C14-NbCr<sub>2</sub> Laves phase in [15] it is suggested (a) that Al does not affect the solid solubility of Ge in this phase, which is about 1.3 at.% and (b) that in the presence of Ge the solid solubility of Si in the Laves phase is in the range 5.5 to 6.5 at.%.

#### 4.2.2. Heat Treated Alloys

After the heat treatment no Ti rich areas were observed in the Nb<sub>ss</sub>, meaning the latter was "homogenized". The Si and Ge concentrations in the Nb<sub>ss</sub> were low (Tables 2 and 3) and in agreement with those in the binary Nb-Si and Nb-Ge systems [20,21]. Thus, the synergy of Cr and Al with/out Hf did not change significantly the solid solubilities of both Si and Ge in the Nb<sub>ss</sub> and did not affect the "homogenization" of the solid solution. In the Nb<sub>ss</sub> in ZF6-HT the solid solubilities of Cr and Al did not change but in ZF9-HT the solid solubility of both Cr and Al decreased. This was attributed to the synergy of Hf and Ge. The Laves phase was absent in both alloys, as was also the case in the heat-treated alloys KZ5 (Nb-24Ti-18Si-5Al-5Cr [12]) and JG3 (Nb-24Ti-5Al-5Cr-5Hf-2Mo [26]).

According to the XRD data (Figures 1 and 4) the  $\beta Nb_5Si_3$  had partially transformed to the  $\alpha Nb_5Si_3$ in both alloys. Thus, equilibrium was not reached in both alloys after 100 h at 1400 °C. This and the data in [13,15,18] for heat treated Ti and Ge containing alloys would suggest that alloying with Ge makes the transformation of  $\beta Nb_5Si_3$  to  $\alpha Nb_5Si_3$  more sluggish.

In the Nb<sub>5</sub>Si<sub>3</sub> the solid solubilities of Ti, Si, Cr, Al and Ge were essentially the same in both alloys. The Hf-rich areas in Nb<sub>5</sub>Si<sub>3</sub> were still present in ZF9-HT with no significant change in the solubility of Hf. Compared with the data for Hf containing Nb-silicide based alloys in [29,30] it is concluded that the solubility of Hf in Nb<sub>5</sub>Si<sub>3</sub> is not affected by the presence of Ge in the alloy.

In the alloy ZF6-HT, a very small volume fraction of Ti rich oxide was formed at the edges of some of its Nb<sub>ss</sub> and Nb<sub>5</sub>Si<sub>3</sub> grains (Figures 2d and 3). This oxide was also observed in heat treated Ti and Ge containing alloys with no Hf present [13,15,18], but was absent in Ti free but Ge containing heat treated Nb-silicide based alloys [9,13,18]. This phase formed because of the contamination of the alloy by O during heat treatment. In the ZF9-HT, HfO<sub>2</sub> was formed instead of titanium oxide. This is in agreement with data for heat treated Hf and Ti-containing Nb-silicide based alloys [29,30] and confirmed the scavenging of oxygen by Hf.

#### 4.2.3. Lattice Parameter of The $Nb_{ss}$

The atomic radii of Nb, Ti, Si, Ge, Cr, Al and Hf are 1.47 Å, 1.47 Å, 1.17 Å, 1.39 Å, 1.30 Å, 1.43 Å and 1.59 Å, respectively [31]. Compared with the alloys ZF4 (Nb-24Ti-18Si-5Cr-5Ge [15]) and ZF5 (Nb-24Ti-18Si-5Al-5Ge [13]), the lattice parameter of the Nb<sub>ss</sub> in the alloy ZF6 respectively decreased by about 0.40% and 0.55% in the as cast condition, and by about 0.28% and 0.73% in the heat treated condition (Table 4). The contraction of the lattice of Nb<sub>ss</sub> individually by Cr and Al was discussed in [13,15]. The data for the alloy ZF6 would thus suggest that the synergy of Ge with Al and Cr led to further contraction of the Nb<sub>ss</sub> lattice. The increase of the lattice parameter of the Nb<sub>ss</sub> in ZF6-HT was attributed to the reduction in the concentrations of Ti, Si, Ge and Al.

The addition of Hf to the Nb<sub>ss</sub> was reported to increase the lattice parameter owing to the larger atomic radius of Hf compared with that of Nb [32,33]. In this work the lattice parameter of the Nb<sub>ss</sub> in the alloy ZF9 was 0.12% higher than that in the alloy ZF6 in the as cast and heat-treated conditions. The increase of the Nb<sub>ss</sub> lattice parameter in ZF9-HT, compared with the as cast condition, was attributed to the reduction of the concentrations of all solute elements.

#### 4.3. Comparison with High Entropy Alloys

Some Nb-silicide based alloys, some bcc Nb solid solutions in Nb-silicide based alloys and some of the eutectics with bcc Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> in Nb-silicide based alloys meet the "accepted" definition of High Entropy Alloys (HEAs) [3]. According to this definition, which asserts that HEAs have principal elements with the concentration of each element being between 35 and 5 at.% [34], the alloy ZF9 is a HEA. However, the alloy ZF6 does not comply with this description of HEAs. More specifically, the alloy ZF9 is a refractory metal (RM) based bcc solid solution + intermetallics HEA or a RM complex concentrated alloy (CCA). The parameters  $\Delta$ H<sup>chem</sup>,  $\Delta$ S<sub>mix</sub>, VEC,  $\delta$ ,  $\Delta\chi$  and  $\Omega$  have been considered for solid solution(s) and solid solution(s) + intermetallic(s) HEAs [34] and Nb-silicide based alloys [3,35,36]. The data in Table 10 shows that the above parameters of the alloys ZF6 and ZF9 do not differ significantly, even though only one of them (ZF9) meets the "criterion" for HEAs. The data in Table 10 shows that the solid solutions are in the ranges for Nb-silicide based alloys [35] and their solid solutions [36].

Condition	Alloy	Nb <sub>ss</sub>			Parame	eter		
-	-	-	∆H <sup>chem</sup> (Kj·mol <sup>-1</sup> )	$\Delta S_{mix}$ (J·mol <sup>-1</sup> ·K)	VEC	δ	$\Delta\chi$	Ω
		Nb <sub>ss</sub>	-39.4	12.31	4.458	8.42	0.146	0.672
AC	ZF6	Ti-rich	-12.87	10.38	4.639	5.12	0.078	1.855
		Nb <sub>ss</sub>	-13	10.9	4.642	5.52	0.077	1.888
HT ZF6	NIL	-40.4	12.16	4.463	8.49	0.149	0.650	
	IND <sub>SS</sub>	-9.7	10.07	4.709	5.02	0.063	2.411	
	o Nh	-40.09	13.65	4.419	8.97	0.167	0.730	
AC	ZF9	IND <sub>SS</sub>	-12.29	11.83	4.626	5.95	0.084	2.146
IIT	750	NIL	-40.73	13.52	4.421	8.99	0.167	0.714
пі	ZF9	IND <sub>SS</sub>	-6.87	9.68	4.764	4.96	0.063	3.367

**Table 10.** The parameters  $\Delta H^{chem}$ ,  $\Delta S_{mix}$ , VEC,  $\delta$ ,  $\Delta \chi$  and  $\Omega$  of the alloys ZF6 and ZF9 and their solid solutions in the as cast (AC) and heat treated (HT) conditions. For calculation of parameters see [35,36]. Note that the symbol Q was used instead of  $\Omega$  in [36].

Research on HEAs has indicated ranges for the above parameters for bcc solid solution(s) and bcc solid solution(s) + intermetallic(s) HEAs. Both alloys of this study have  $\Delta H^{chem}$  more negative than bcc solid solution + intermetallic HEAs, their parameters  $\Delta S_{mix}$  and  $\delta$  are in the range of bcc solid solution + intermetallic HEAs, and the parameters VEC and  $\Omega$  are smaller than those of bcc solid solution + intermetallic HEAs [35]. The bcc solid solutions in the alloys ZF6 and ZF9 (a) have  $\Delta H^{chem}$ 

and  $\Delta S_{mix}$ ,  $\delta$  and  $\Omega$  in the range of RM based bcc solid solution HEAs [37], (b) their  $\Omega$  values fall in the lower range of  $\Omega$  values for the RM based bcc solid solution HEAs [37] and (c) their parameters VEC,  $\delta$ ,  $\Omega$ ,  $\Delta \chi$  and  $\Delta H$  are in the range of bcc solid solution HEAs [38–40].

The density of RM HEAs (or RM CCAs) is in the range 5.59 g/cm<sup>3</sup> (for bcc AlNbTiV) to 13.75 g/cm<sup>3</sup> (for bcc MoNbTaW) [34]. The majority (74%) of the twenty-three RM HEAs reported in [34] have densities higher than those of the alloys ZF6 and ZF9. More specifically, only the NbTiV<sub>2</sub>Zr (bcc, 6.38 g/cm<sup>3</sup>) and CrNbTiVZr (bcc + Laves, 6.52 g/cm<sup>3</sup>) HEAs have densities lower than the alloy ZF6 and the latter alloys and the NbTiVZr (bcc, 6.5 g/cm<sup>3</sup>), CrNbTiZr (bcc + Laves, 6.7 g/cm<sup>3</sup>) and AlNb<sub>1.5</sub>Ta<sub>0.5</sub>Ti<sub>1.5</sub>Zr<sub>0.5</sub> (bcc, 6.88 g/cm<sup>3</sup>) HEAs have densities lower than the alloy ZF9.

The hardness of RM bcc HEAs or CCAs is in the range 306 HV (NbTiV<sub>2</sub>Zr, bcc) to 591 HV (AlMo<sub>0.5</sub>NbTa<sub>0.5</sub>TiZr, bcc + B2) [34]. The hardness of the Nb<sub>ss</sub> in ZF6-HT was higher than the aforementioned range and higher than that of the bcc solid solution RM based HEAs AlNb<sub>1.5</sub>Ta<sub>0.5</sub>Ti<sub>1.5</sub>Zr<sub>0.5</sub>, Al<sub>0.4</sub>Hf<sub>0.6</sub>NbTaTiZr, and Al<sub>0.3</sub>NbTa<sub>0.8</sub>Ti<sub>1.4</sub>V<sub>0.2</sub>Zr<sub>1.3</sub> reported in [37].

## 4.4. Hardness

The hardness of the alloy ZF6 was higher by 11.9% and 8.4% compared with the alloy ZF4 (Nb-24Ti-18Si-5Al-5Ge [13]), and lower by 3.6% and 8.2% compared with the alloy ZF5 (Nb-24Ti-18Si-5Cr-5Ge [15]) in the as cast and heat treated conditions, respectively. The alloy ZF6 exhibited good retention of its hardness after the heat treatment at 1400 °C, as was the case for the alloys ZF4 and ZF5. The hardness data for these three alloys would suggest that in the synergy of Ge with Al and Cr the latter is the key element. Compared with the alloy ZF6, the addition of Hf in the alloy ZF9 resulted to a reduction of hardness by 4.6% and 8.8% in the as cast and heat-treated conditions, respectively. The alloy ZF9 exhibited good retention of its hardness after the heat treatment as the hardness after the heat of ZF9-AC.

The synergy of Cr and Al in the alloy ZF6 cancelled out the negative effect of Al on the hardness of Nb<sub>ss</sub> [13,15] and would suggest that the synergy of Cr and Ge is key to strengthening the Nb<sub>ss</sub>. The hardness of Nb<sub>5</sub>Si<sub>3</sub> in the alloy ZF6 was almost the same as in ZF4 [13] in both the as cast and heat-treated conditions, and lower than that in the alloy ZF5 [15]. This would suggest that in the synergy of Al and Cr with Ge the Al "controls" the hardness of Nb<sub>5</sub>Si<sub>3</sub>. The hardness of Nb<sub>5</sub>Si<sub>3</sub> in ZF9 was lower in both the as cast and heat-treated conditions compared with ZF6. This would suggest that the synergy of Hf with Al, Cr and Ge has a softening effect on Nb<sub>5</sub>Si<sub>3</sub>. The hardness of Nb<sub>5</sub>Si<sub>3</sub> in ZF9-HT was the same as that of unalloyed Nb<sub>5</sub>Si<sub>3</sub>.

The application of the rule of mixtures (ROM) to calculate the strength of composites assumes the matrix of the composite to be unaffected by the reinforcement. The hardness of ZF6-HT calculated using ROM is 1423 HV and the measured value was 854 HV (Table 5). In the Ti free alloys ZF1 (Nb-18Si-5Ge) and ZF2 (Nb-18Si-10Ge) with microstructures consisting of Nb<sub>ss</sub> and Nb<sub>5</sub>Si<sub>3</sub>, the same as ZF6-HT, best agreement with the experimental value was given by a Pythagorean type addition rule [9]. If we assume that the Pythagorean rule [41] applies in the case of ZF6-HT a correction of 0.652 would be necessary to match the two values. This correction is essentially the same with the average correction of 0.638 for all alternative addition rules of calculating HV [9,42].

The hardness of the alloys ZF1-HT15 and ZF2-HT15 respectively was 812 HV and 819 HV and the microhardness of Nb<sub>5</sub>Si<sub>3</sub> in the two alloys was 1532 and 1526 HV, respectively [9]. In other words, the hardness values of these alloys and their Nb<sub>5</sub>Si<sub>3</sub> silicides were not significantly different than those of the alloy ZF6-HT. Why the Pythagorean rule addition did not work in the case ZF6-HT? The alloys ZF6 and ZF2 essentially were intermetallic matrix composites, ZF6 is different from ZF2 (see below) and ZF1 is a metal matrix composite. We would like to suggest that the difference is attributed to the microstructures of the three alloys and in particular the constraint effects on hardness.

The strength of in-situ composites is strongly affected by the size, shape and distribution of the matrix and reinforcement phase(s). The presence of sharp corners in the latter, which can act as stress concentrators, must be taken into account in composite strengthening [43]. The constraint imposed

by the brittle intermetallic phase on the ductile matrix can contribute significantly to the composite strength. Research on composites has considered different strengthening mechanisms which either independently or simultaneously can be responsible for the strength of a composite. The proposed mechanisms include (a) strengthening due to constrained plastic flow in the ductile matrix owing to the presence of brittle intermetallics [44] and (b) cavitation at the interfaces and the matrix in the high stressed regions near sharp corners in the reinforcement [45].

The microstructure of the as cast alloy ZF6 had a very high (>0.8) volume fraction of strongly faceted large Nb<sub>5</sub>Si<sub>3</sub> grains with thinner layers of Nb<sub>ss</sub> and Nb<sub>ss</sub> + Nb<sub>5</sub>Si<sub>3</sub> eutectic confined in between the Nb<sub>5</sub>Si<sub>3</sub> grains (Figure 2b). After the heat treatment the prior eutectic was not observed in between the large silicide grains. The as cast alloys ZF1 and ZF2 had microstructures consisting of primary Nb<sub>5</sub>Si<sub>3</sub> and Nb<sub>ss</sub> + Nb<sub>5</sub>Si<sub>3</sub> eutectic. The vol.% of primary Nb<sub>5</sub>Si<sub>3</sub> was slightly more than double in the alloy ZF2 [9]. In both alloys the primary Nb<sub>5</sub>Si<sub>3</sub> was faceted. The ZF1-HT15 had Nb<sub>5</sub>Si<sub>3</sub> with fine prior eutectic in between Nb<sub>5</sub>Si<sub>3</sub> grains (constraint different from ZF6-HT), the ZF2-HT15 had similar microstructure with ZF1-HT5 but with higher vol.% of Nb<sub>5</sub>Si<sub>3</sub> and less fine prior eutectic between the non-facetted Nb<sub>5</sub>Si<sub>3</sub> grains (constraint different from ZF6-HT).

The alloying behaviour of Nb-silicide based alloys and of the phases that can be present in their microstructures, and properties of the alloys and their phases can be studied using the parameters VEC,  $\delta$  and  $\Delta\chi$  [3,4,8,35,36]. In Figure 8 we show how the hardness of the alloys KZ5 (Nb-24Ti-18Si-5Al-5Cr), JN1 (Nb-24Ti-18Si-5Al-5Cr-5Hf [46]), ZF6 and ZF9 varies with the alloy parameters VEC,  $\delta$  and  $\Delta\chi$ . Data for the alloys KZ5 and JN1 has been included in order to show how the additions of Hf and/or Ge to the "basis" alloy KZ5 affect hardness. The Nb<sub>5</sub>Si<sub>3</sub> was facetted in the alloys KZ5, JN1 and ZF6. The Hf-rich Nb<sub>5</sub>Si<sub>3</sub> in ZF9 was also faceted (Figure 5b). The hardness increased as the alloy VEC value decreased and the values of the alloy parameters  $\delta$  and  $\Delta\chi$  increased. The alloy KZ5 had 48–55 vol.% Nb<sub>ss</sub> and 45-52 vol.% Nb<sub>5</sub>Si<sub>3</sub> [12], the alloy JN1 had 42 vol.% Nb<sub>ss</sub> and 58 vol.% Nb<sub>5</sub>Si<sub>3</sub> [46] and the alloy ZF6-HT had 17.3 vol.% Nb<sub>ss</sub> and 82.7 vol.% Nb<sub>5</sub>Si<sub>3</sub>. Constraint effects on hardness owing to the faceting of Nb<sub>5</sub>Si<sub>3</sub> apply to all alloys. The hardness increased as the vol.% of Nb<sub>5</sub>Si<sub>3</sub> increased compared with the alloy KZ5. All the alloys fall on the same curve with high R<sup>2</sup> values in all cases in the HV versus VEC,  $\delta$  and  $\Delta\chi$  plots in Figure 8. Remarkably, the alloy ZF9 falls in the same curve with the alloys KZ5, JN1 and ZF6 even though only its Hf-rich Nb<sub>5</sub>Si<sub>3</sub> was facetted. The alloy ZF9 also had a higher vol.% of Nb<sub>5</sub>Si<sub>3</sub>.



Figure 8. Cont.



(c)

**Figure 8.** Vickers hardness (HV) versus VEC (a) or  $\delta$  (b) or  $\Delta \chi$  (c) of the alloys KZ5 = Nb-24Ti-18Si-5Al-5Cr [12] (blue), JN1 = Nb-24Ti-18Si-5Al-5Cr-5Hf [46] (green), ZF6 (red) and ZF9 (purple).

#### 5. Conclusions

In this work we studied the effects of the synergy of Ti, Ge, Cr, Al and Hf additions on the microstructure and hardness of the as cast and heat-treated alloys Nb-24Ti-18Si-5Al-5Cr-Ge (ZF6) and Nb-24Ti-18Si-5Al-5Cr-5Ge-5Hf (ZF9). We also compared these alloys with RM bcc solid solution + intermetallic HEAs. The partitioning of Al, Cr, Ge, Hf and Ti between the phases that were present in the microstructures contributed to the macrosegregation of Si, Ti, Cr and Al in both alloys. Alloying with Hf reduced but did not eliminate the macrosegregation of Si. Chromium was the key element in the macrosegregation of Ti. The microstructures of the cast alloys consisted of the bcc Nb<sub>ss</sub>, Nb<sub>5</sub>Si<sub>3</sub> silicide and C14-NbCr2 Laves phase and the  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> was the primary phase. The Nb<sub>ss</sub> +  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> eutectic was formed only in ZF6. The microstructures of the heat-treated alloys consisted of the Nb<sub>ss</sub>, Nb<sub>5</sub>Si<sub>3</sub> phases. The transformation of  $\beta$ Nb<sub>5</sub>Si<sub>3</sub> to  $\alpha$ Nb<sub>5</sub>Si<sub>3</sub> was not completed after 100 h at 1400 °C.

Contamination of the alloys by oxygen during heat treatment led to the formation of Ti and Hf oxides respectively in ZF6 and ZF9. Alloying with Hf decreased the hardness of the alloy ZF9. The hardness of the Nb<sub>5</sub>Si<sub>3</sub> silicide decreased to that of the unalloyed silicide after 100 h at 1400  $^{\circ}$ C in the alloy ZF9.

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## Appendix A

Alloy Code	Nominal Composition (at.%)	Reference
KZ3	Nb-24Ti-18Si	[12]
KZ4	Nb-24Ti-18Si-5Cr	[12]
KZ7	Nb-24Ti-18Si-5Al	[12]
KZ5	Nb-24Ti-18Si-5Al-5Cr	[12]
ZF1	Nb-18Si-5Ge	[9]
ZF2	Nb-18Si-10Ge	[9]
ZF3	Nb-24Ti-18Si-5Ge	[18]
ZF7	Nb-18Si-5Cr-5Ge	[15]
ZF4	Nb-24Ti-18Si-5Cr-5Ge	[15]
ZF8	Nb-18Si-5Al-5Ge	[13]
ZF5	Nb-24Ti-18Si-5Al-5Ge	[13]

Table A1. Nominal compositions (at.%) of alloys that are referred to in this paper

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